

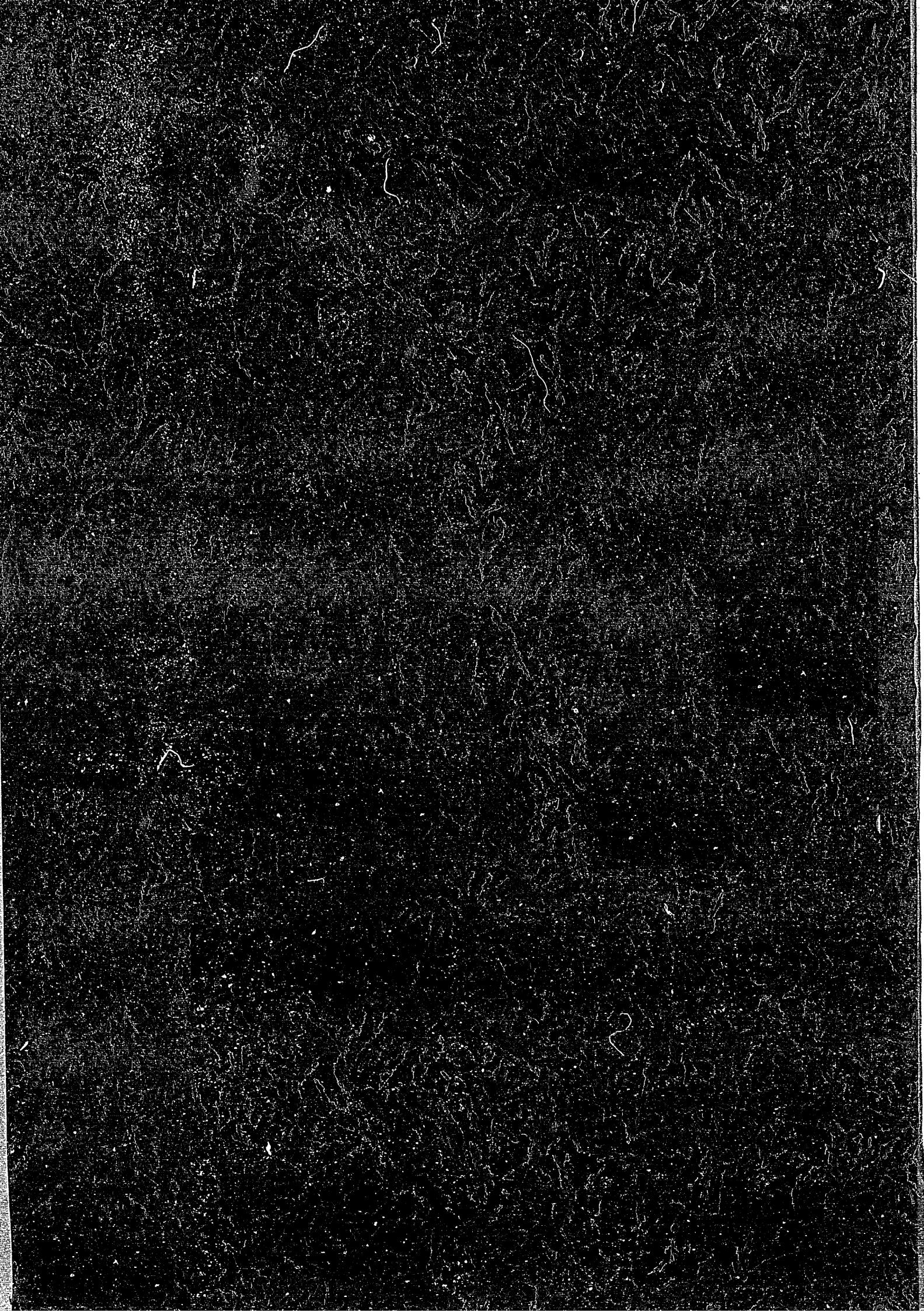
IPPJ-AM-37

IONIZATION CROSS SECTIONS  
OF ATOMS AND IONS  
BY ELECTRON IMPACT

H. TAWARA, T. KATO AND M. OHNISHI

INSTITUTE OF PLASMA PHYSICS  
NAGOYA UNIVERSITY

NAGOYA, JAPAN



**IPPJ-AM-37**

**IONIZATION CROSS SECTIONS OF ATOMS AND IONS  
BY ELECTRON IMPACT**

Hiroyuki TAWARA, Takako KATO and Mari OHNISHI

Institute of Plasma Physics, Nagoya University  
Chikusa-ku, Nagoya 464, Japan

February 1985

This document is prepared as a preprint of compilation of atomic data for fusion research sponsored fully or partly by the IPP/Nagoya University. This is intended for future publication in a journal or will be included in a data book after some evaluations or rearrangements of its contents. This document should not be referred without the agreement of the authors. Enquiries about copyright and reproduction should be addressed to Research Information Center, IPP/Nagoya University, Nagoya, Japan.

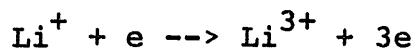
## **Abstract**

Experimental data of the ionization cross sections for atoms and ions ranging from H to U in electron impact are compiled and given, in graphical forms, as a function of the electron impact energy. Some theoretical data are also included. The literature is surveyed up to the end of 1984. Some comments on the compiled data are also given.

## Contents

INTRODUCTION .....	1
REFERENCES FOR INTRODUCTION .....	5
LIST OF PROCESSES OF COMPILED DATA.....	7
FIGURES OF IONIZATION CROSS SECTIONS PLOTTED AS A FUNCTION OF ELECTRON IMPACT ENERGY (Fig. 1-294) .....	13

The process



indicates that singly charged  $\text{Li}^+$  ions are doubly ionized in electron impact, resulting in triply charged  $\text{Li}^{3+}$  ions.

REFERENCES FOR FIGURES .....	307
------------------------------	-----

## INTRODUCTION

The ionization processes of atoms and ions by electron impact are important not only in atomic physics but also in detailed understanding of a number of processes in, for example discharge, radiation physics, laboratory plasmas as well as astrophysical plasmas. In as early as 1920s, the relative ionization probabilities for rare gas atoms by electrons were measured<sup>1)</sup>. Then, in 1930s, total absolute ionization efficiencies defined as "numbers of positive charge per electron per cm path per mm Hg pressure at 0°C" were determined<sup>2)</sup> for rare gas atoms and common molecules. The partial absolute ionization efficiencies for production of Ne<sup>3+</sup>, Ar<sup>5+</sup> and Hg<sup>5+</sup> ions were also determined.<sup>3)</sup> Since then, a number of experimental and theoretical investigations have been made. Kieffer and Dunn<sup>4)</sup> and Kieffer<sup>5)</sup> summarized and reviewed these results on ionization cross sections. Recently much attention has been paid to these ionization processes involving highly ionized ions in high temperature plasmas. Also it has been realized that the ionization processes play a key role even in low temperature plasmas such as those in the periphery of high temperature plasmas and in divertor plasmas<sup>6)</sup>. To handle the ionization processes more easily, a number of the empirical formulas for calculating the ionization cross sections had been proposed. In particular, the formula derived by Lotz<sup>7)</sup> is widely used which has the following

forms:

$$\sigma = \sum_{i=1} \alpha_i q_i \frac{\ln(E/P_i)}{E P_i} \{1 - b_i \exp [-c_i(E/P_i - 1)]\} \quad (1)$$

where  $\sigma$  is the total ionization cross section,  $E$  the electron impact energy,  $P_i$  the binding energy of electrons in the  $i$ -th shell,  $q_i$  the number of electrons in the  $i$ -th shell,  $\alpha_i$ ,  $b_i$  and  $c_i$  are constants given in his papers<sup>7)</sup>. The summation  $\Sigma$  is made over all shells. More recently, a review on such empirical formulas of the ionization cross sections has been made<sup>8)</sup>. Also a critical review of the ionization cross sections for target atoms and ions up to  $Z=10$  has been made by Belfast group<sup>9)</sup>. A list of the bibliographical references on the ionizations of ions by electron impact up to 1983 can be found<sup>10)</sup>.

In this Report, experimental ionization cross sections for atoms and ions ranging from H to U in elemental forms published up to the end of 1984 are compiled and given in two graphical forms which are hoped to be convenient for different users of these results. In these graphs are also shown the curves calculated by the Lotz formula (1) summed over three outermost subshells (for targets with the atomic number  $Z \leq 30$ ) and selected theoretical data since 1970. Theoretical results analytically fitted to the calculated values and those calculated by the simple plane-wave Born approximation are not included. Data for molecules are not included except for the common molecules such as  $H_2$ ,  $D_2$ ,  $N_2$  and  $O_2$ . The numerical tables of these cross sections, which

are stored as one of the databases (AMDIS: Atomic and Molecular Data Interactive System) at our Institute, are available upon request.

Remarks on compiled data

1. The cross sections are given per collision. That is, those for molecular targets are per molecule.
2. The cross sections indicated as  $\sum_{n=1}^{\infty} \{A^{n+} + (n+1)e\}$  are defined as  $\Sigma n \sigma_n$  where  $\sigma_n$  is the partial ionization cross section for production of  $A^{n+}$  ions and are measured in a condenser plate method.
3. It should be kept in mind that highly ionized ions produced in ion sources usually contain a significant fraction of ions in the metastable states which have the ionization cross sections different from those in the ground state. Some discussion on this problem is given<sup>11)</sup>.
4. When the original data are given for unknown low and high fractions of the metastable ions in the beams, those for low fractions are shown in figures.
5. Experimental data for neutral single atoms such as C, N, O and S which are not commonly available can be obtained through the developed crossed-beam technique<sup>12)</sup> recently.
6. Some data obtained in trapped ion beam technique should be borne in mind to be model-dependent<sup>13)</sup>.

7. Most of the theoretical calculations give only the direct ionization cross sections. The discrepancy between theories and experiments may well be due to the exclusion of the excitation-autoionization for theoretical data.
8. We have tried to include the available recent theoretical calculations on the excitation-autoionization cross sections (e.g.,  $\text{Ca}^+$ ,  $\text{Ti}^{3+}$ ,  $\text{Fe}^{15+}$ , etc.).

#### Remarks on the future investigations

It is found through the present compilation that most of the total ionization cross sections for gases seem to be in agreement with each other. However, some problems, which should be solved, arise as follows:

1. Some cross sections for production of highly ionized ions from neutral gaseous targets are sometimes in significant disagreement with each other. This is partly due to the incomplete extraction and collection of these ions and partly due to the detection efficiencies for these ions. The measurement of most of the cross sections, in particular for ions with  $n \geq 3$ , should be repeated.
2. Similar disagreement is also seen even in those for low charge ions from non-gaseous targets. This is mostly due to the lack of proper methods for determining precise target densities.

A method should be developed for accurate determination of the densities of such targets.

3. The experimental data for highly ionized ions, particularly for Fe ions, are urgently required in the investigations of high temperature plasmas.
4. The contribution of the excitation-autoionization to total ionization is quite large for some ions such as Na-like and Mg-like ions. The theoretical studies to find some scaling of this contribution would be useful.

#### Acknowledgements

A number of people contributed to this compilation of the ionization cross section data. In particular, the authors would like to thank Profs. K. Takayanagi, H. Suzuki, T. Iwai, Y. Itikawa, S. Ohtani and the member of the Study Group of Atomic Processes in Plasmas at Institute of Plasma Physics, Nagoya University. Dr. D. C. Gregory, Dr. K. Okuno and Prof. E. Salzborn kindly provided their experimental data in tabular forms. The authors acknowledge Dr. K. Takasugi for the computer program to draw graphs and lists of references in the retrieval and display system of AMDIS.

#### REFERENCES FOR INTRODUCTION

1. K. T. Compton and C. C. van Voorhis, Phys. Rev. 26 (1925) 436.

2. P. T. Smith, Phys. Rev. 36 (1930) 1293.
3. W. Bleakney. Phys. Rev. 35 (1930) 139; Phys. Rev. 36 (1930) 1303.
4. L. J. Kieffer and G. H. Dunn, Rev. Mod. Phys. 38 (1966) 1.
5. L. J. Kieffer, Atomic Data 1 (1969) 19.
6. D. E. Post and C. E. Singer, Proc. 6-th Int. Conf. Plasma Surface Interactions in Controlled Fusion Devices (Ed. A. Miyahara, H. Tawara, N. Itoh and K. Kamada, North-Holland, 1984) p.78.
7. W. Lotz, Z. Phys. 206 (1967) 205; Z. Phys. 216 (1968) 241.
8. Y. Itikawa and T. Kato, IPPJ-AM-17 (Institute of Plasma Physics, Nagoya Univ., 1981).
9. K. L. Bell, H. B. Gilbody, J. G. Hughes, A. E. Kingston and F. J. Smith, J. Phys. Chem. Ref. Data 12 (1983) 891.
10. Y. Itikawa, K. Takayanagi and T. Iwai, At. Data and Nucl. Data Tables 31 (1984) 215; Y. Itikawa and T. Iwai, IPPJ-AM-7 (Institute of Plasma Physics, Nagoya Univ., 1977); Y. Itikawa, IPPJ-AM-24 (Institute of Plasma Physics, Nagoya Univ., 1982).
11. R. G. Montague and M. F. A. Harrison, J. Phys. B 16 (1983) 3045.
12. E. Brook, M. F. A. Harrison and A. C. H. Smith, J. Phys. B 11 (1978) 3115.
13. E. D. Donets and V. P. Ovsyannikov, Soviet Phys. - JETP 53 (1981) 466.

LIST OF PROCESSES OF COMPILED DATA

Figure Number	Process	Figure Number	Process
1.	$H_2 + e \rightarrow \text{sum}$	31.	$N_2 + e \rightarrow N^+$
2.	$\rightarrow H_2^+ + 2e$	32.	$\rightarrow N^{2+}$
3.	$\rightarrow H^+$	33.	$\rightarrow N^{3+}$
4.	$H + e \rightarrow H^+ + 2e$	34.	$N + e \rightarrow \text{sum}$
5.	$H(2S) + e \rightarrow H^+ + 2e$	35.	$\rightarrow N^+ + 2e$
6.	$D_2 + e \rightarrow \text{sum}$	36.	$N^+ + e \rightarrow N^{2+} + 2e$
7.	$He + e \rightarrow \text{sum}$	37.	$N^{2+} + e \rightarrow N^{3+} + 2e$
8.	$\rightarrow He^+ + 2e$	38.	$N^{3+} + e \rightarrow N^{4+} + 2e$
9.	$\rightarrow He^{2+} + 3e$	39.	$N^{4+} + e \rightarrow N^{5+} + 2e$
10.	$He(2S) + e \rightarrow He^+ + 2e$	40.	$N^{5+} + e \rightarrow N^{6+} + 2e$
11.	$He^+ + e \rightarrow He^{2+} + 2e$	41.	$N^{6+} + e \rightarrow N^{7+} + 2e$
12.	$Li + e \rightarrow \text{sum}$	42.	$O_2 + e \rightarrow \text{sum}$
13.	$\rightarrow Li^+ + 2e$	43.	$\rightarrow O_2^+ + 2e$
14.	$\rightarrow Li^{2+} + 3e$	44.	$\rightarrow O_2^{2+} + 3e$
15.	$Li^+ + e \rightarrow Li^{2+} + 2e$	45.	$\rightarrow O^+$
16.	$\rightarrow Li^{3+} + 3e$	46.	$O + e \rightarrow \text{sum}$
17.	$Li^{2+} + e \rightarrow Li^{3+} + 2e$	47.	$\rightarrow O^+ + 2e$
18.	$Be^+ + e \rightarrow Be^{2+} + 2e$	48.	$\rightarrow O^{2+} + 3e$
19.	$B^+ + e \rightarrow B^{2+} + 2e$	49.	$O^+ + e \rightarrow O^{2+} + 2e$
20.	$B^{2+} + e \rightarrow B^{3+} + 2e$	50.	$O^{2+} + e \rightarrow O^{3+} + 2e$
21.	$B^{3+} + e \rightarrow B^{4+} + 2e$	51.	$O^{3+} + e \rightarrow O^{4+} + 2e$
22.	$C + e \rightarrow C^+ + 2e$	52.	$O^{4+} + e \rightarrow O^{5+} + 2e$
23.	$C^+ + e \rightarrow C^{2+} + 2e$	53.	$O^{5+} + e \rightarrow O^{6+} + 2e$
24.	$C^{2+} + e \rightarrow C^{3+} + 2e$	54.	$O^{6+} + e \rightarrow O^{7+} + 2e$
25.	$C^{3+} + e \rightarrow C^{4+} + 2e$	55.	$O^{7+} + e \rightarrow O^{8+} + 2e$
26.	$C^{4+} + e \rightarrow C^{5+} + 2e$	56.	$F^{2+} + e \rightarrow F^{3+} + 2e$
27.	$C^{5+} + e \rightarrow C^{6+} + 2e$	57.	$F^{5+} + e \rightarrow F^{6+} + 2e$
28.	$N_2 + e \rightarrow \text{sum}$	58.	$Ne + e \rightarrow \text{sum}$
29.	$\rightarrow N_2^+ + 2e$	59.	$\rightarrow Ne^+ + 2e$
30.	$\rightarrow N_2^{2+} + 3e$	60.	$\rightarrow Ne^{2+} + 3e$

Figure Number	Process	Figure Number	Process
61.	$\text{Ne} + e \rightarrow \text{Ne}^{3+} + 4e$	91.	$\text{Al}^{5+} + e \rightarrow \text{Al}^{6+} + 2e$
62.	$\rightarrow \text{Ne}^{4+} + 5e$	92.	$\text{Al}^{6+} + e \rightarrow \text{Al}^{7+} + 2e$
63.	$\rightarrow \text{Ne}^{5+} + 6e$	93.	$\text{Al}^{7+} + e \rightarrow \text{Al}^{8+} + 2e$
64.	$\text{Ne}^+ + e \rightarrow \text{Ne}^{2+} + 2e$	94.	$\text{Al}^{8+} + e \rightarrow \text{Al}^{9+} + 2e$
65.	$\text{Ne}^{2+} + e \rightarrow \text{Ne}^{3+} + 2e$	95.	$\text{Al}^{9+} + e \rightarrow \text{Al}^{10+} + 2e$
66.	$\text{Ne}^{3+} + e \rightarrow \text{Ne}^{4+} + 2e$	96.	$\text{Al}^{10+} + e \rightarrow \text{Al}^{11+} + 2e$
67.	$\text{Ne}^{4+} + e \rightarrow \text{Ne}^{5+} + 2e$	97.	$\text{Al}^{11+} + e \rightarrow \text{Al}^{12+} + 2e$
68.	$\text{Ne}^{5+} + e \rightarrow \text{Ne}^{6+} + 2e$	98.	$\text{Si}^{3+} + e \rightarrow \text{Si}^{4+} + 2e$
69.	$\text{Ne}^{6+} + e \rightarrow \text{Ne}^{7+} + 2e$	99.	$\text{P}^{4+} + e \rightarrow \text{P}^{5+} + 2e$
70.	$\text{Ne}^{7+} + e \rightarrow \text{Ne}^{8+} + 2e$	100.	$\text{S} + e \rightarrow \text{S}^+ + 2e$
71.	$\text{Ne}^{8+} + e \rightarrow \text{Ne}^{9+} + 2e$	101.	$\rightarrow \text{S}^{2+} + 3e$
72.	$\text{Ne}^{9+} + e \rightarrow \text{Ne}^{10+} + 2e$	102.	$\rightarrow \text{S}^{3+} + 4e$
73.	$\text{Na} + e \rightarrow \text{sum}$	103.	$\rightarrow \text{S}^{4+} + 5e$
74.	$\rightarrow \text{Na}^+ + 2e$	104.	$\text{Cl}^{2+} + e \rightarrow \text{Cl}^{3+} + 2e$
75.	$\rightarrow \text{Na}^{2+} + 3e$	105.	$\text{Ar} + e \rightarrow \text{sum}$
76.	$\text{Na}^+ + e \rightarrow \text{Na}^{2+} + 2e$	106.	$\rightarrow \text{Ar}^+ + 2e$
77.	$\text{Na}^{2+} + e \rightarrow \text{Na}^{3+} + 2e$	107.	$\rightarrow \text{Ar}^{2+} + 3e$
78.	$\text{Na}^{9+} + e \rightarrow \text{Na}^{10+} + 2e$	108.	$\rightarrow \text{Ar}^{3+} + 4e$
79.	$\text{Mg} + e \rightarrow \text{sum}$	109.	$\rightarrow \text{Ar}^{4+} + 5e$
80.	$\rightarrow \text{Mg}^+ + 2e$	110.	$\rightarrow \text{Ar}^{5+} + 6e$
81.	$\rightarrow \text{Mg}^{2+} + 3e$	111.	$\rightarrow \text{Ar}^{6+} + 7e$
82.	$\text{Mg}^+ + e \rightarrow \text{Mg}^{2+} + 3e$	112.	$\rightarrow \text{Ar}^{7+} + 8e$
83.	$\text{Mg}^{2+} + e \rightarrow \text{Mg}^{3+} + 2e$	113.	$\text{Ar}^+ + e \rightarrow \text{Ar}^{2+} + 2e$
84.	$\text{Mg}^{9+} + e \rightarrow \text{Mg}^{10+} + 2e$	114.	$\rightarrow \text{Ar}^{3+} + 3e$
85.	$\text{Al} + e \rightarrow \text{sum}$	115.	$\rightarrow \text{Ar}^{4+} + 4e$
86.	$\rightarrow \text{Al}^+ + 2e$	116.	$\rightarrow \text{Ar}^{5+} + 5e$
87.	$\text{Al}^+ + e \rightarrow \text{Al}^{2+} + 2e$	117.	$\text{Ar}^{2+} + e \rightarrow \text{Ar}^{3+} + 2e$
88.	$\text{Al}^{2+} + e \rightarrow \text{Al}^{3+} + 2e$	118.	$\rightarrow \text{Ar}^{4+} + 3e$
89.	$\text{Al}^{3+} + e \rightarrow \text{Al}^{4+} + 2e$	119.	$\rightarrow \text{Ar}^{5+} + 4e$
90.	$\text{Al}^{4+} + e \rightarrow \text{Al}^{5+} + 2e$	120.	$\text{Ar}^{3+} + e \rightarrow \text{Ar}^{4+} + 2e$

**Figure      Process**  
Number

121.	$\text{Ar}^{3+} + e \rightarrow \text{Ar}^{5+} + 3e$	151.	$\text{Fe}^+ + e \rightarrow \text{Fe}^{2+} + 2e$
122.	$\text{Ar}^{4\text{OM}} + e \rightarrow \text{Ar}^{5\text{OM}} + 2e$	152.	$\text{Fe}^{2+} + e \rightarrow \text{Fe}^{3+} + 2e$
123.	$\text{Ar}^{5\text{OM}} + e \rightarrow \text{Ar}^{6\text{OM}} + 3e$	153.	$\text{Fe}^{3+} + e \rightarrow \text{Fe}^{4+} + 2e$
124.	$\text{Ar}^{5\text{OM}} + e \rightarrow \text{Ar}^{6\text{OM}} + 2e$	154.	$\text{Fe}^{4+} + e \rightarrow \text{Fe}^{5+} + 2e$
125.	$\text{Ar}^{6\text{fS}} + e \rightarrow \text{Ar}^{7\text{fA}} + 2e$	155.	$\text{Fe}^{5+} + e \rightarrow \text{Fe}^{6+} + 2e$
126.	$\text{Ar}^{7\text{fA}} + e \rightarrow \text{Ar}^{8+} + 2e$	156.	$\text{Fe}^{6+} + e \rightarrow \text{Fe}^{7+} + 2e$
127.	$\text{Ar}^{8+} + e \rightarrow \text{Ar}^{9+} + 2e$	157.	$\text{Fe}^{7+} + e \rightarrow \text{Fe}^{8+} + 2e$
128.	$\text{Ar}^{10+} + e \rightarrow \text{Ar}^{11+} + 2e$	158.	$\text{Fe}^{8+} + e \rightarrow \text{Fe}^{9+} + 2e$
129.	$\text{Ar}^{11+} + e \rightarrow \text{Ar}^{12+} + 2e$	159.	$\text{Fe}^{9+} + e \rightarrow \text{Fe}^{10+} + 2e$
130.	$\text{Ar}^{12+} + e \rightarrow \text{Ar}^{13+} + 2e$	160.	$\text{Fe}^{10+} + e \rightarrow \text{Fe}^{11+} + 2e$
131.	$\text{Ar}^{14+} + e \rightarrow \text{Ar}^{15+} + 2e$	161.	$\text{Fe}^{11+} + e \rightarrow \text{Fe}^{12+} + 2e$
132.	$\text{Ar}^{15+} + e \rightarrow \text{Ar}^{16+} + 2e$	162.	$\text{Fe}^{12+} + e \rightarrow \text{Fe}^{13+} + 2e$
133.	$\text{Ar}^{16+} + e \rightarrow \text{Ar}^{17+} + 2e$	163.	$\text{Fe}^{13+} + e \rightarrow \text{Fe}^{14+} + 2e$
134.	$\text{Ar}^{17+} + e \rightarrow \text{Ar}^{18+} + 2e$	164.	$\text{Fe}^{14+} + e \rightarrow \text{Fe}^{15+} + 2e$
135.	$\text{K}^+ + e \rightarrow \text{sum}$	165.	$\text{Fe}^{15+} + e \rightarrow \text{Fe}^{16+} + 2e$
136.	$\text{K}^{2+} + e \rightarrow \text{K}^{3+} + 2e$	166.	$\text{Fe}^{16+} + e \rightarrow \text{Fe}^{17+} + 2e$
137.	$\text{K}^{2+} + e \rightarrow \text{K}^3 + 2e$	167.	$\text{Fe}^{17+} + e \rightarrow \text{Fe}^{18+} + 2e$
138.	$\text{Ca}^{3X} + e \rightarrow \text{sum}$	168.	$\text{Fe}^{18+} + e \rightarrow \text{Fe}^{19+} + 2e$
139.	$\text{Ca}^{4X} + e \rightarrow \text{Ca}^{2+} + 2e$	169.	$\text{Fe}^{19+} + e \rightarrow \text{Fe}^{20+} + 2e$
140.	$\text{Ca}^{2+} + e \rightarrow \text{Ca}^{3+} + 2e$	170.	$\text{Fe}^{20+} + e \rightarrow \text{Fe}^{21+} + 2e$
141.	$\text{Sc}^{3X} + e \rightarrow \text{Sc}^{4+} + 2e$	171.	$\text{Fe}^{21+} + e \rightarrow \text{Fe}^{22+} + 2e$
142.	$\text{Sc}^{4X} + e \rightarrow \text{Sc}^{5+} + 2e$	172.	$\text{Fe}^{22+} + e \rightarrow \text{Fe}^{23+} + 2e$
143.	$\text{Sc}^{5X} + e \rightarrow \text{Sc}^{6+} + 2e$	173.	$\text{Fe}^{23+} + e \rightarrow \text{Fe}^{24+} + 2e$
144.	$\text{Sc}^{6X} + e \rightarrow \text{Sc}^{7+} + 2e$	174.	$\text{Fe}^{24+} + e \rightarrow \text{Fe}^{25+} + 2e$
145.	$\text{Sc}^{7X} + e \rightarrow \text{Sc}^{8+} + 2e$	175.	$\text{Fe}^{25+} + e \rightarrow \text{Fe}^{26+} + 2e$
146.	$\text{Sc}^{8X} + e \rightarrow \text{Sc}^{9+} + 2e$	176.	$\text{Ni}^{3+} + e \rightarrow \text{Ni}^{4+} + 2e$
147.	$\text{Sc}^{9X} + e \rightarrow \text{Sc}^{10+} + 2e$	177.	$\text{Ni}^{4+} + e \rightarrow \text{Ni}^{18+} + 2e$
148.	$\text{Ti}^{2X} + e \rightarrow \text{Ti}^{3+} + 2e$	178.	$\text{Cu}^{+} + e \rightarrow \text{sum} .80$
149.	$\text{Ti}^{3X} + e \rightarrow \text{Ti}^{4+} + 2e$	179.	$\text{Cu}^{2+} + e \rightarrow \text{Cu}^{3+} + 2e$
150.	$\text{V}^{5X} + e \rightarrow \text{V}^{6+} + 2e$	180.	$\text{Cu}^{3+} + e \rightarrow \text{Cu}^{4+} + 2e$

**Figure      Process**  
Number

Figure Number	Process	Figure Number	Process
181.	$Zn + e \rightarrow$ sum	211.	$Mo^{5+} + e \rightarrow Mo^{6+} + 2e$
182.	$Zn^+ + e \rightarrow Zn^{2+} + 2e$	212.	$Mo^{14+} + e \rightarrow Mo^{15+} + 2e$
183.	$Zn^{19+} + e \rightarrow Zn^{20+} + 2e$	213.	$Mo^{24+} + e \rightarrow Mo^{25+} + 2e$
184.	$Ga^+ + e \rightarrow Ga^{2+} + 2e$	214.	$Mo^{32+} + e \rightarrow Mo^{33+} + 2e$
185.	$Kr + e \rightarrow$ sum	215.	$Ag + e \rightarrow$ sum
186.	$\rightarrow Kr^+ + 2e$	216.	$\rightarrow Ag^+ + 2e$
187.	$\rightarrow Kr^{2+} + 3e$	217.	$\rightarrow Ag^{2+} + 3e$
188.	$\rightarrow Kr^{3+} + 4e$	218.	$Cd + e \rightarrow$ sum
189.	$\rightarrow Kr^{4+} + 5e$	219.	$In + e \rightarrow$ sum
190.	$\rightarrow Kr^{5+} + 6e$	220.	$Sb^+ + e \rightarrow Sb^{2+} + 2e$
191.	$\rightarrow Kr^{6+} + 7e$	221.	$\rightarrow Sb^{3+} + 3e$
192.	$\rightarrow Kr^{7+} + 8e$	222.	$Sb^{2+} + e \rightarrow Sb^{3+} + 2e$
193.	$\rightarrow Kr^{8+} + 9e$	223.	$Sb^{3+} + e \rightarrow Sb^{4+} + 2e$
194.	$\rightarrow Kr^{9+} + 10e$	224.	$Te + e \rightarrow$ sum
195.	$Kr^+ + e \rightarrow Kr^{2+} + 2e$	225.	$I^+ + e \rightarrow I^{2+} + 2e$
196.	$\rightarrow Kr^{3+} + 3e$	226.	$Xe + e \rightarrow$ sum
197.	$Kr^{2+} + e \rightarrow Kr^{3+} + 2e$	227.	$\rightarrow Xe^+ + 2e$
198.	$\rightarrow Kr^{4+} + 3e$	228.	$\rightarrow Xe^{2+} + 3e$
199.	$Kr^{3+} + e \rightarrow Kr^{4+} + 2e$	229.	$\rightarrow Xe^{3+} + 4e$
200.	$\rightarrow Kr^{5+} + 3e$	230.	$\rightarrow Xe^{4+} + 5e$
201.	$Kr^{4+} + e \rightarrow Kr^{6+} + 3e$	231.	$\rightarrow Xe^{5+} + 6e$
202.	$Kr^{18+} + e \rightarrow Kr^{19+} + 2e$	232.	$\rightarrow Xe^{6+} + 7e$
203.	$Rb + e \rightarrow$ sum	233.	$\rightarrow Xe^{7+} + 8e$
204.	$Rb^+ + e \rightarrow Rb^{2+} + 2e$	234.	$\rightarrow Xe^{8+} + 9e$
205.	$\rightarrow Rb^{3+} + 3e$	235.	$\rightarrow Xe^{9+} + 10e$
206.	$\rightarrow Rb^{4+} + 4e$	236.	$\rightarrow Xe^{10+} + 11e$
207.	$\rightarrow Rb^{5+} + 5e$	237.	$\rightarrow Xe^{11+} + 12e$
208.	$Sr + e \rightarrow$ sum	238.	$\rightarrow Xe^{12+} + 13e$
209.	$Sr^+ + e \rightarrow Sr^{2+} + 2e$	239.	$\rightarrow Xe^{13+} + 14e$
210.	$Zr^{3+} + e \rightarrow Zr^{4+} + 2e$	240.	$Xe^+ + e \rightarrow Xe^{2+} + 2e$

Figure Number	Process	Figure Number	Process
241.	$Xe^+ + e \rightarrow Xe^{3+} + 3e$	271.	$W^+ + e \rightarrow W^{2+} + 2e$
242.	$\rightarrow Xe^{4+} + 4e$	272.	$Hg + e \rightarrow \text{sum}$
243.	$\rightarrow Xe^{5+} + 5e$	273.	$\rightarrow Hg^+ + 2e$
244.	$Xe^{2+} + e \rightarrow Xe^{3+} + 2e$	274.	$\rightarrow Hg^{2+} + 3e$
245.	$\rightarrow Xe^{4+} + 3e$	275.	$\rightarrow Hg^{3+} + 4e$
246.	$\rightarrow Xe^{5+} + 4e$	276.	$\rightarrow Hg^{4+} + 5e$
247.	$\rightarrow Xe^{6+} + 5e$	277.	$\rightarrow Hg^{5+} + 6e$
248.	$Xe^{3+} + e \rightarrow Xe^{4+} + 2e$	278.	$Hg^+ + e \rightarrow Hg^{2+} + 2e$
249.	$\rightarrow Xe^{5+} + 3e$	279.	$Pb + e \rightarrow \text{sum}$
250.	$\rightarrow Xe^{6+} + 4e$	280.	$\rightarrow Pb^+ + 2e$
251.	$Xe^{4+} + e \rightarrow Xe^{5+} + 2e$	281.	$\rightarrow Pb^{2+} + 3e$
252.	$\rightarrow Xe^{6+} + 3e$	282.	$\rightarrow Pb^{3+} + 4e$
253.	$Xe^{5+} + e \rightarrow Xe^{6+} + 2e$	283.	$Bi^+ + e \rightarrow Bi^{2+} + 2e$
254.	$Xe^{6+} + e \rightarrow Xe^{7+} + 2e$	284.	$\rightarrow Bi^{3+} + 3e$
255.	$\rightarrow Xe^{8+} + 3e$	285.	$Bi^{2+} + e \rightarrow Bi^{3+} + 2e$
256.	$\rightarrow Xe^{9+} + 4e$	286.	$\rightarrow Bi^{4+} + 3e$
257.	$Cs + e \rightarrow \text{sum}$	287.	$Bi^{3+} + e \rightarrow Bi^{4+} + 3e$
258.	$Cs^+ + e \rightarrow Cs^{2+} + 2e$	288.	$Tl + e \rightarrow \text{sum}$
259.	$\rightarrow Cs^{3+} + 3e$	289.	$Tl^+ + e \rightarrow Tl^{2+} + 2e$
260.	$\rightarrow Cs^{4+} + 4e$	290.	$U + e \rightarrow \text{sum}$
261.	$\rightarrow Cs^{5+} + 5e$	291.	$\rightarrow U^+ + 2e$
262.	$Ba + e \rightarrow \text{sum}$	292.	$\rightarrow U^{2+} + 3e$
263.	$\rightarrow Ba^+ + 2e$	293.	$\rightarrow U^{3+} + 4e$
264.	$\rightarrow Ba^{2+} + 3e$	294.	$\rightarrow U^{4+} + 5e$
265.	$\rightarrow Ba^{3+} + 4e$		
266.	$\rightarrow Ba^{4+} + 5e$		
267.	$Ba^+ + e \rightarrow Ba^{2+} + 2e$		
268.	$Au + e \rightarrow \text{sum}$		
269.	$Hf^{3+} + e \rightarrow Hf^{4+} + 2e$		
270.	$Ta^{3+} + e \rightarrow Ta^{4+} + 2e$		



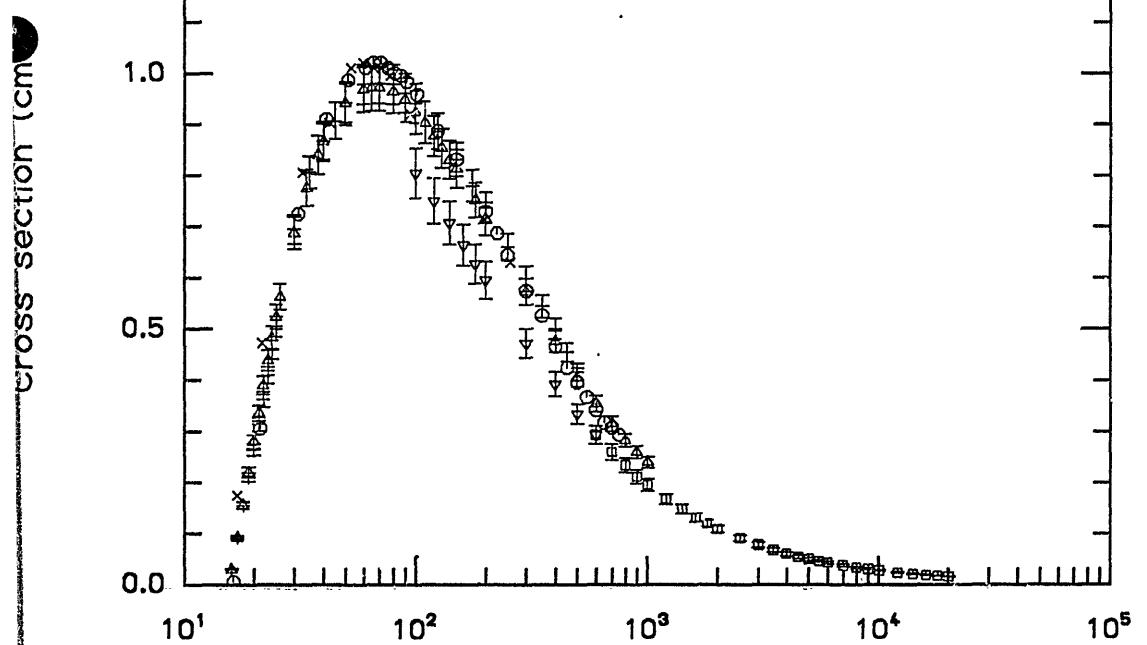
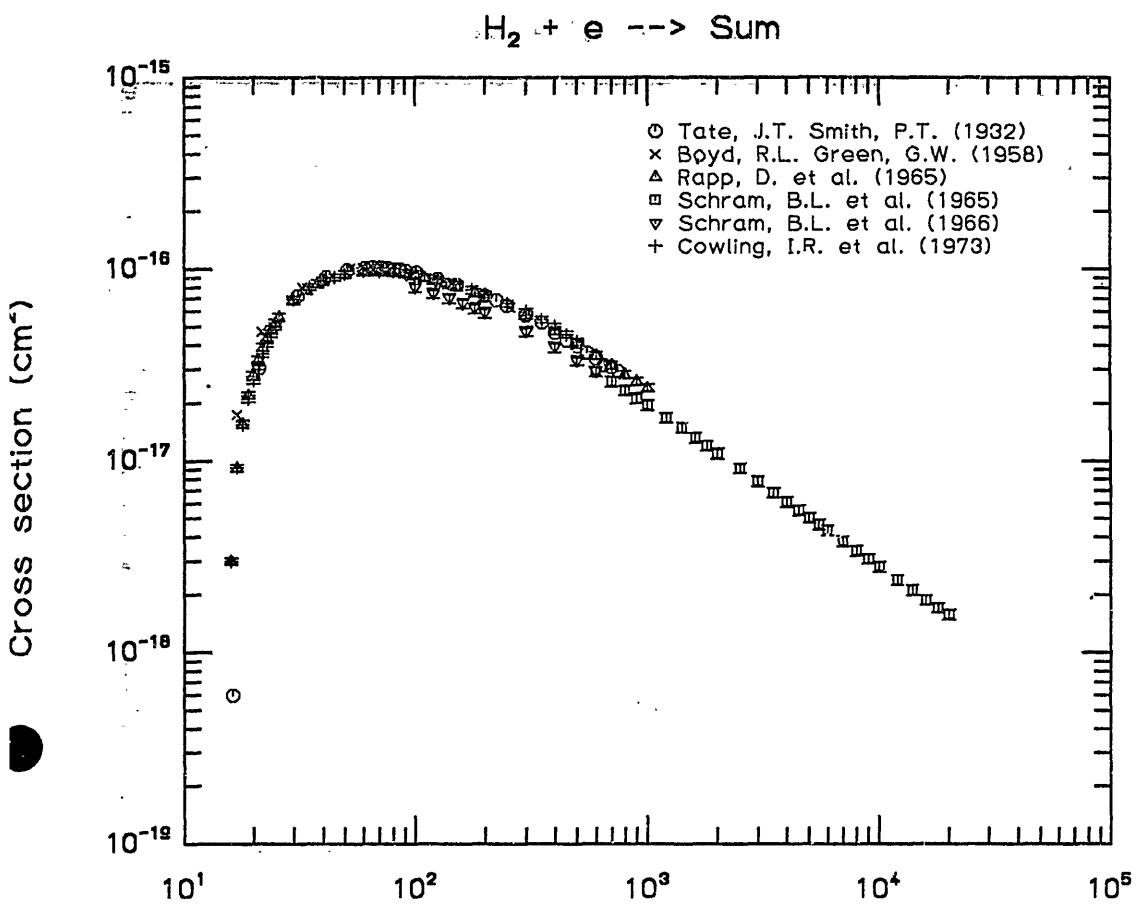


Fig. 1

Electron energy (eV)

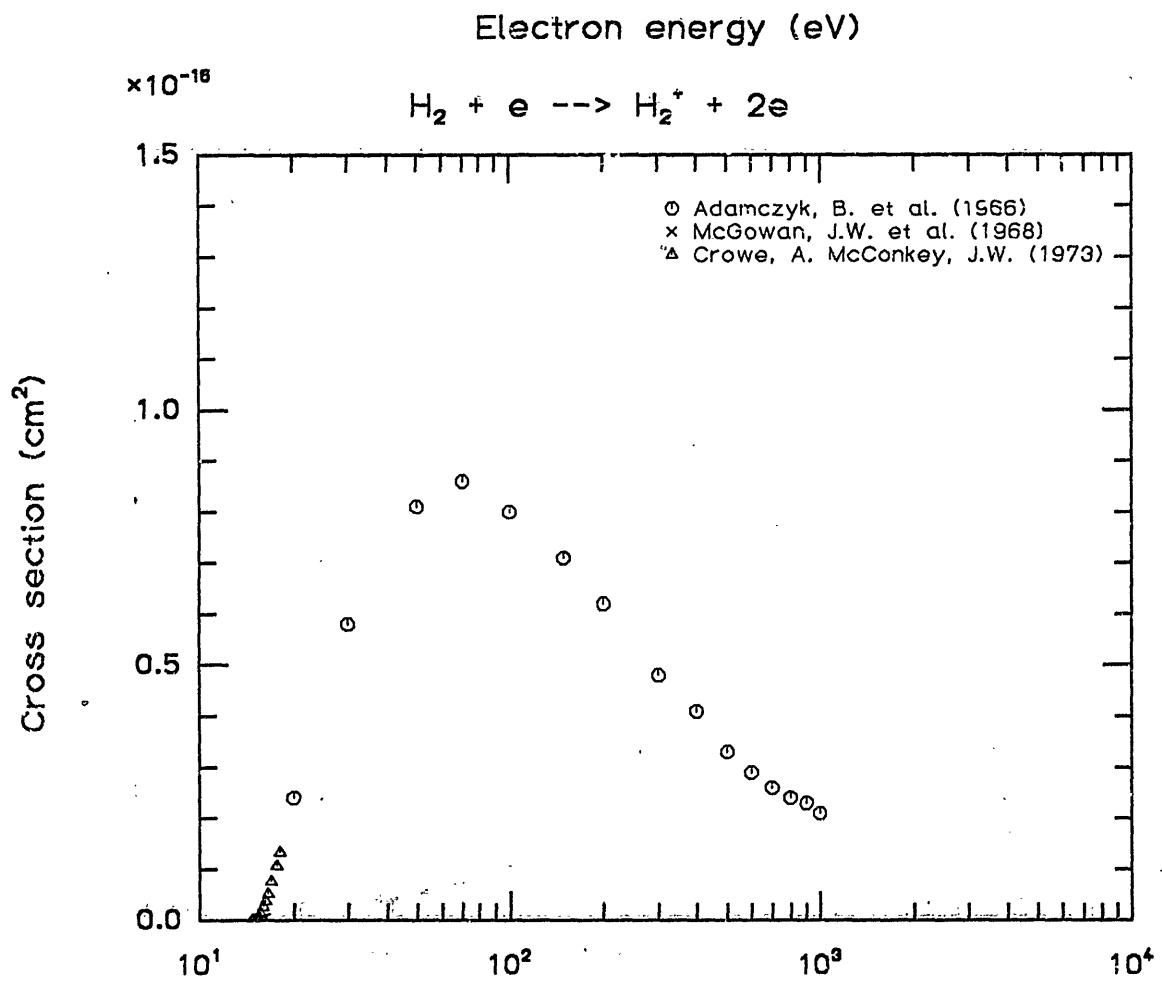
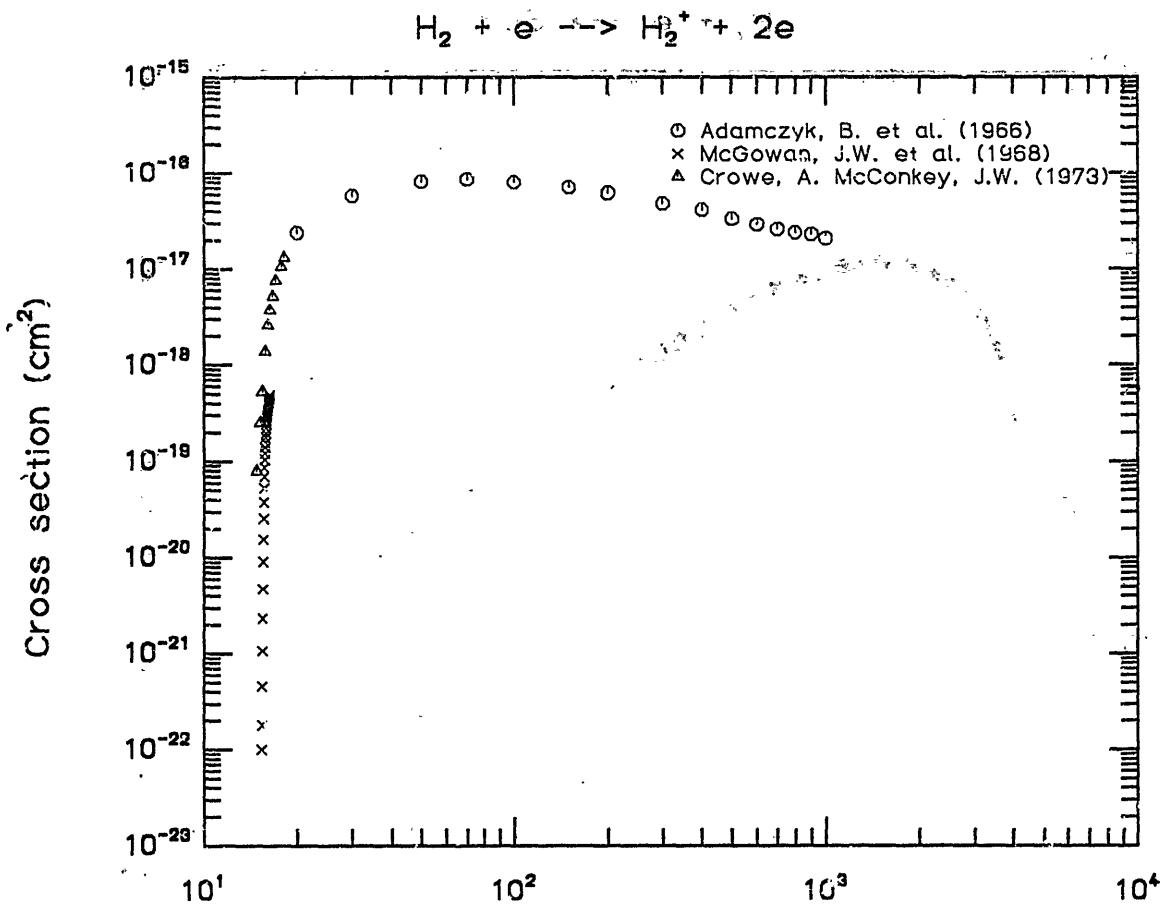
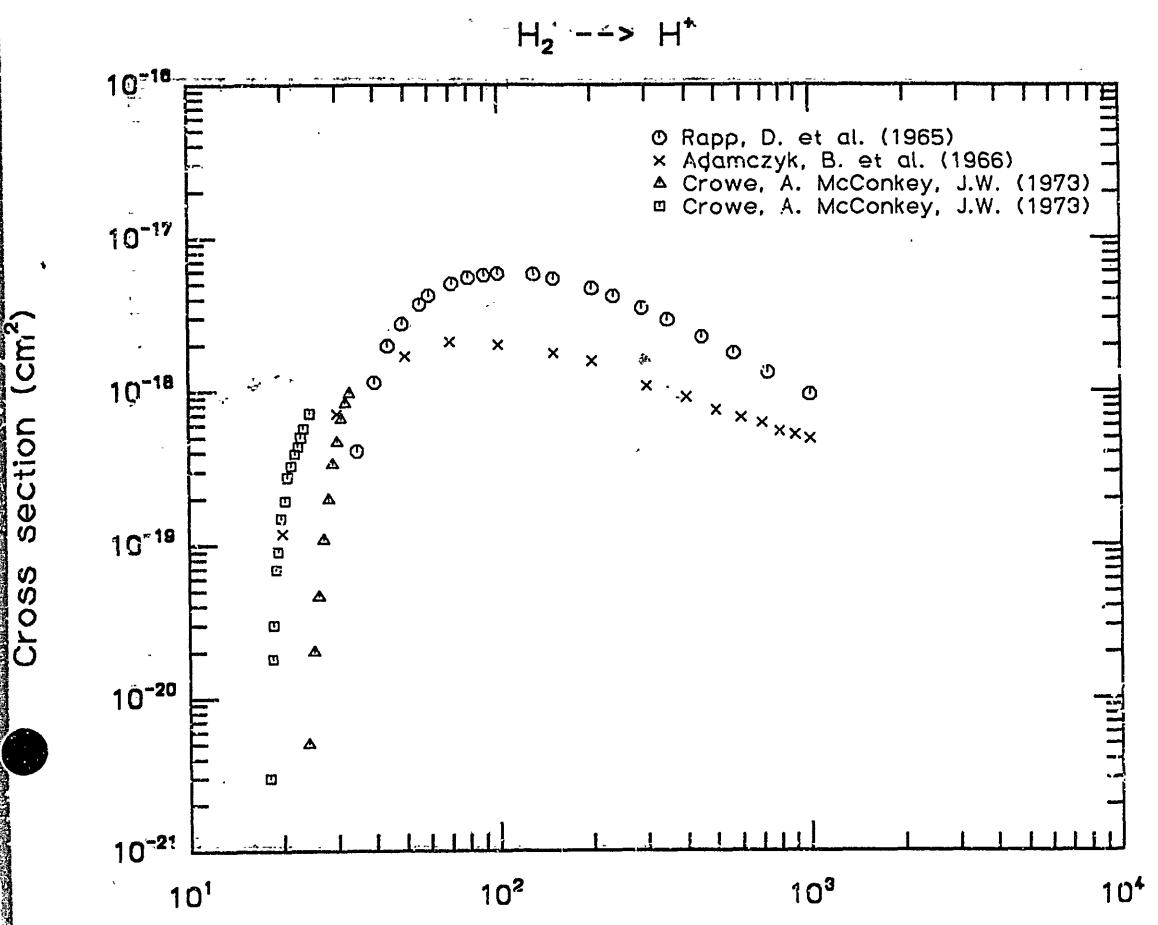


Fig. 2      Electron energy (eV)



Electron energy (eV)

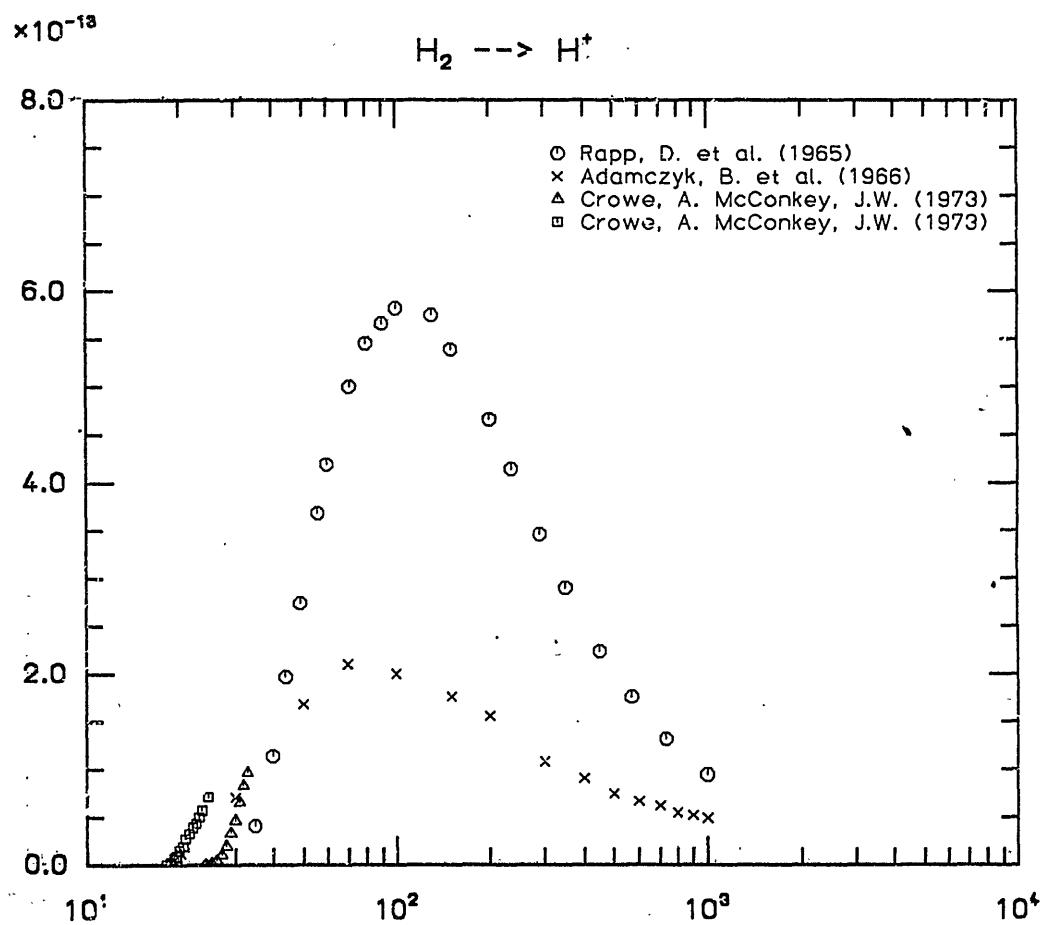


Fig. 3

Electron energy (eV)

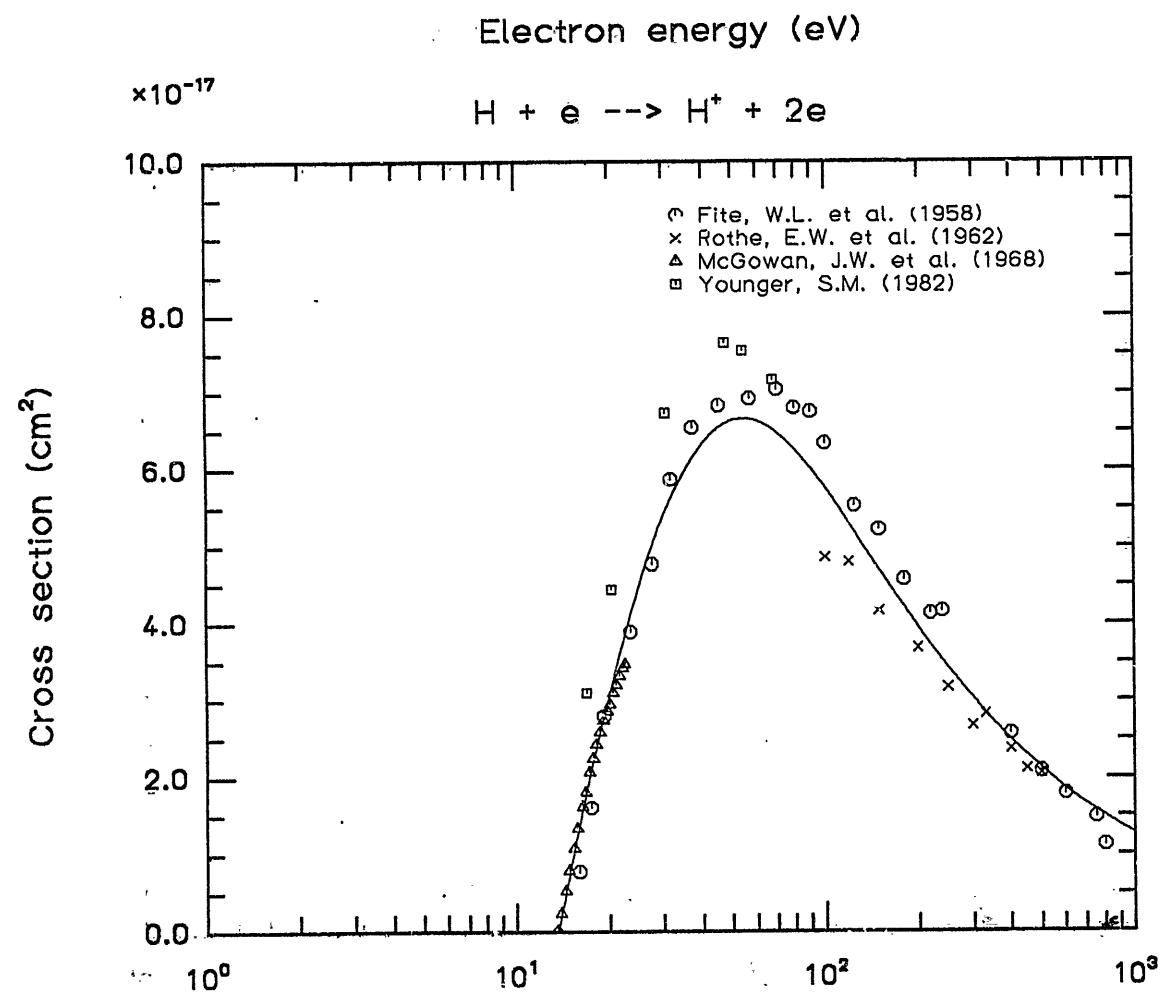
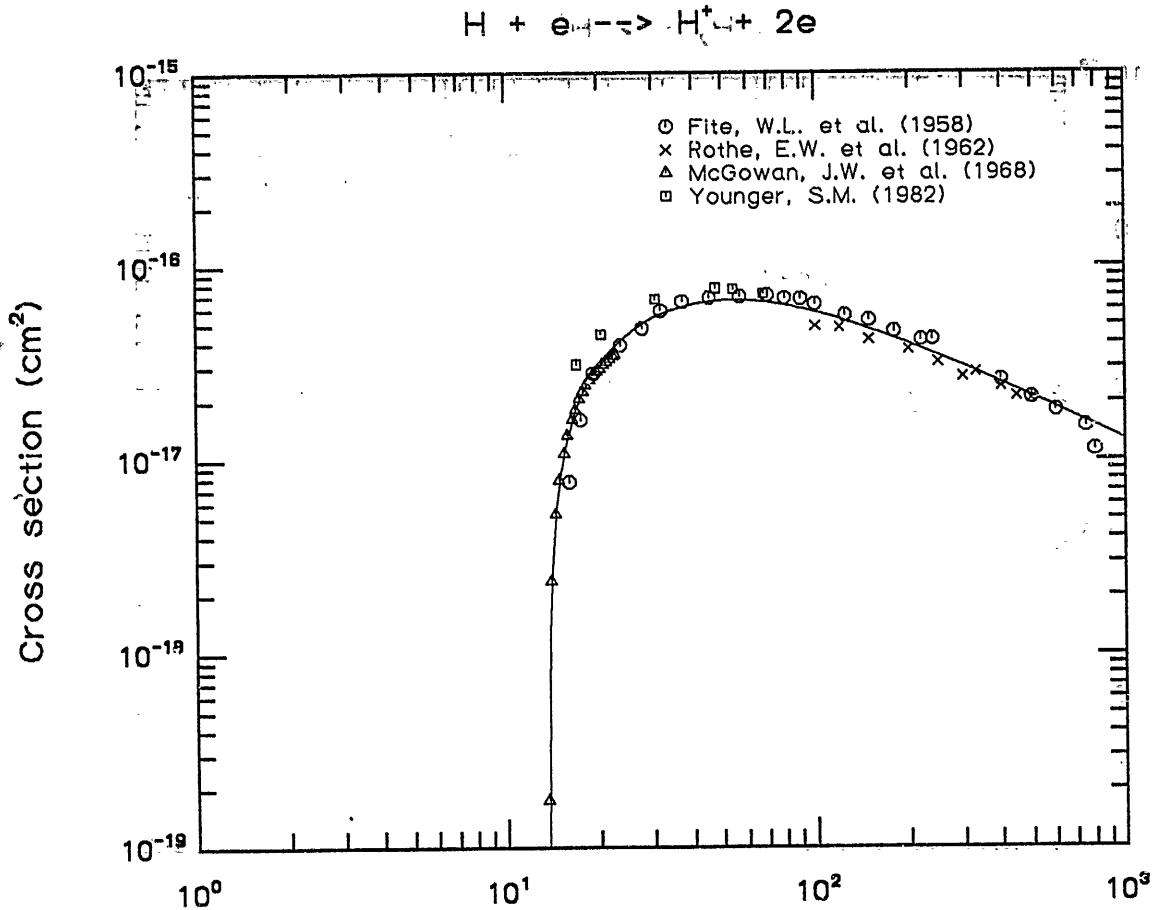
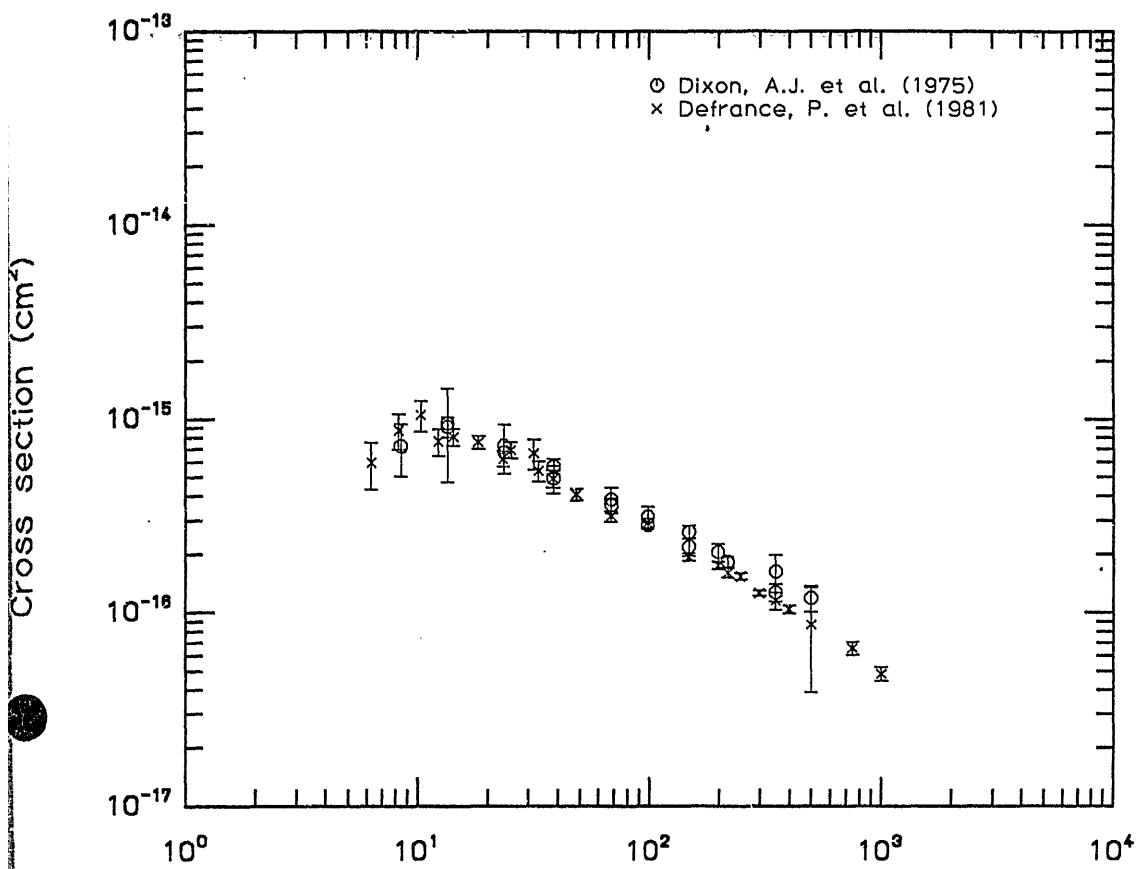
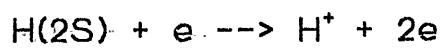


Fig. 4



Electron energy (eV)

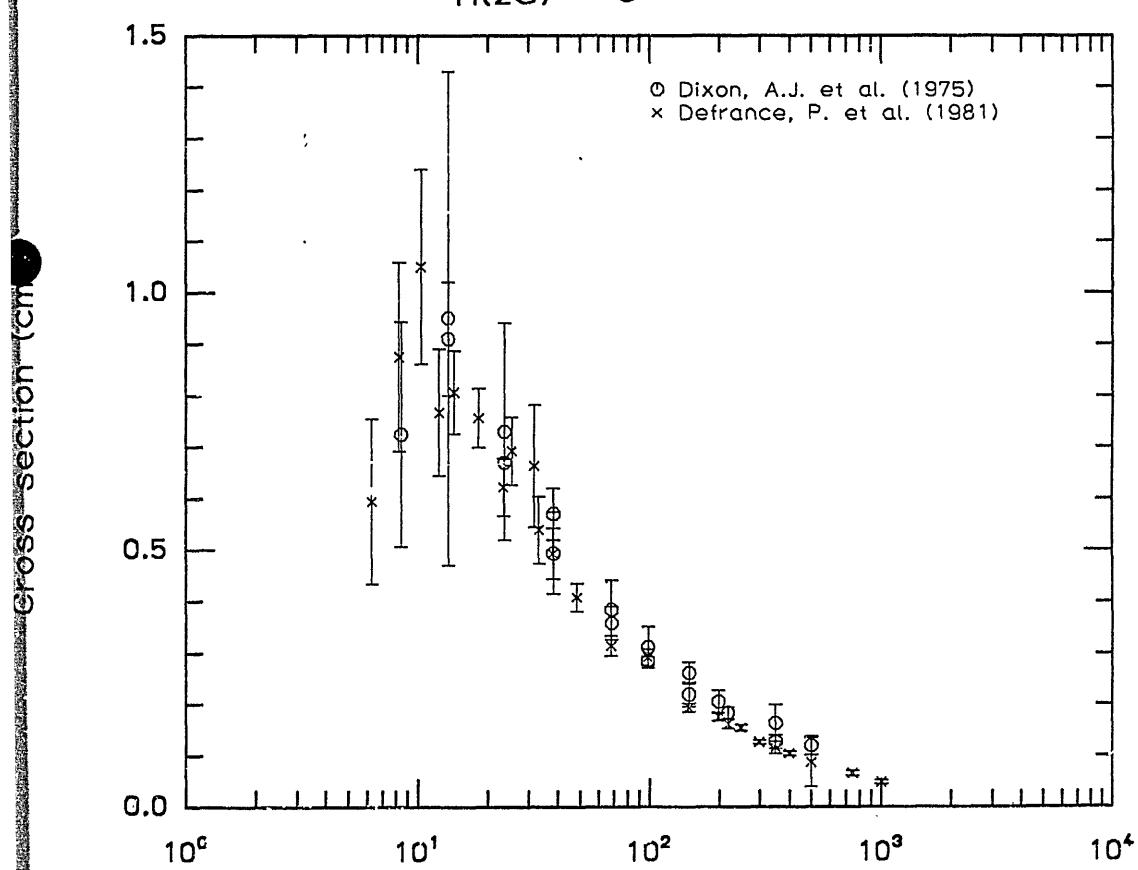


Fig. 5

Electron energy (eV)

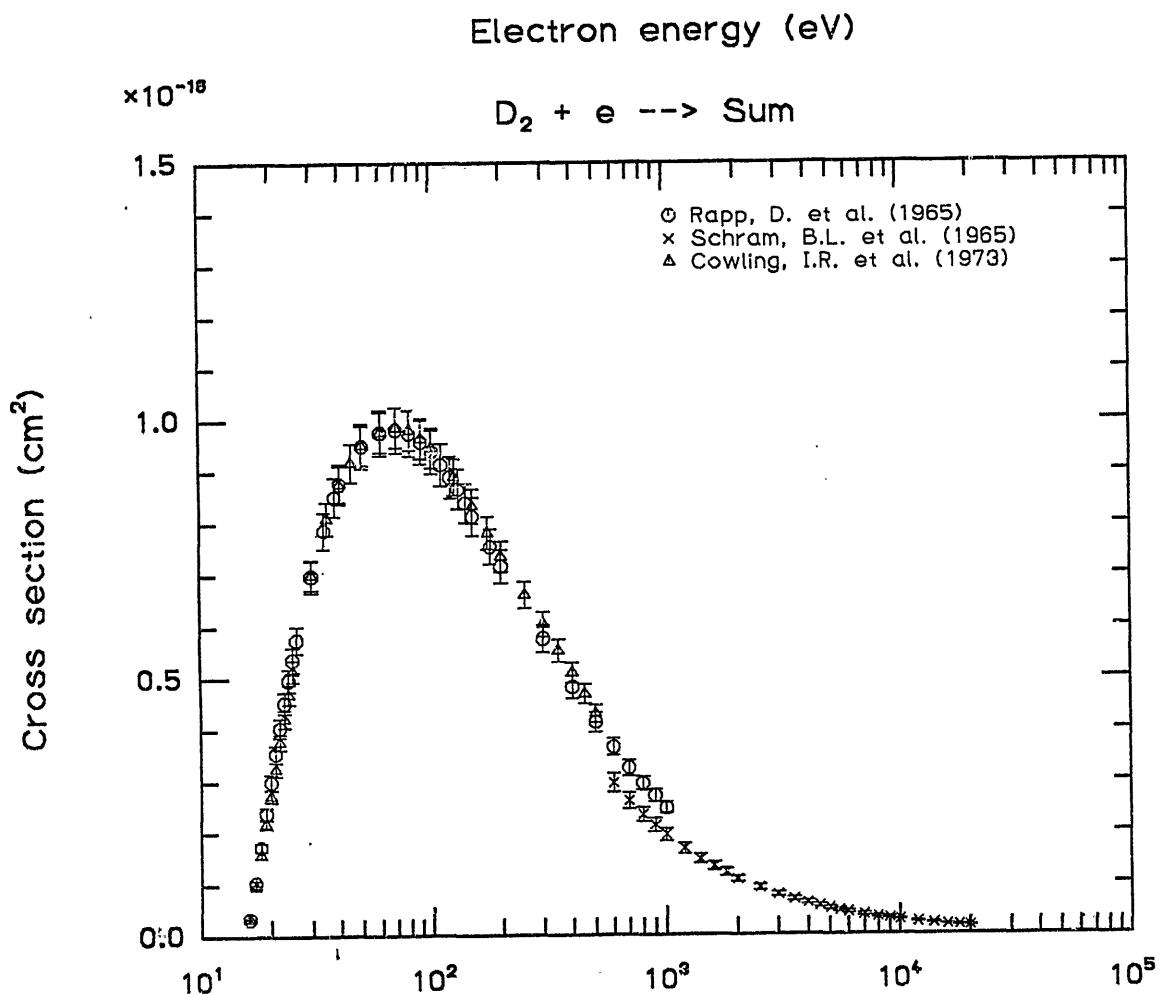
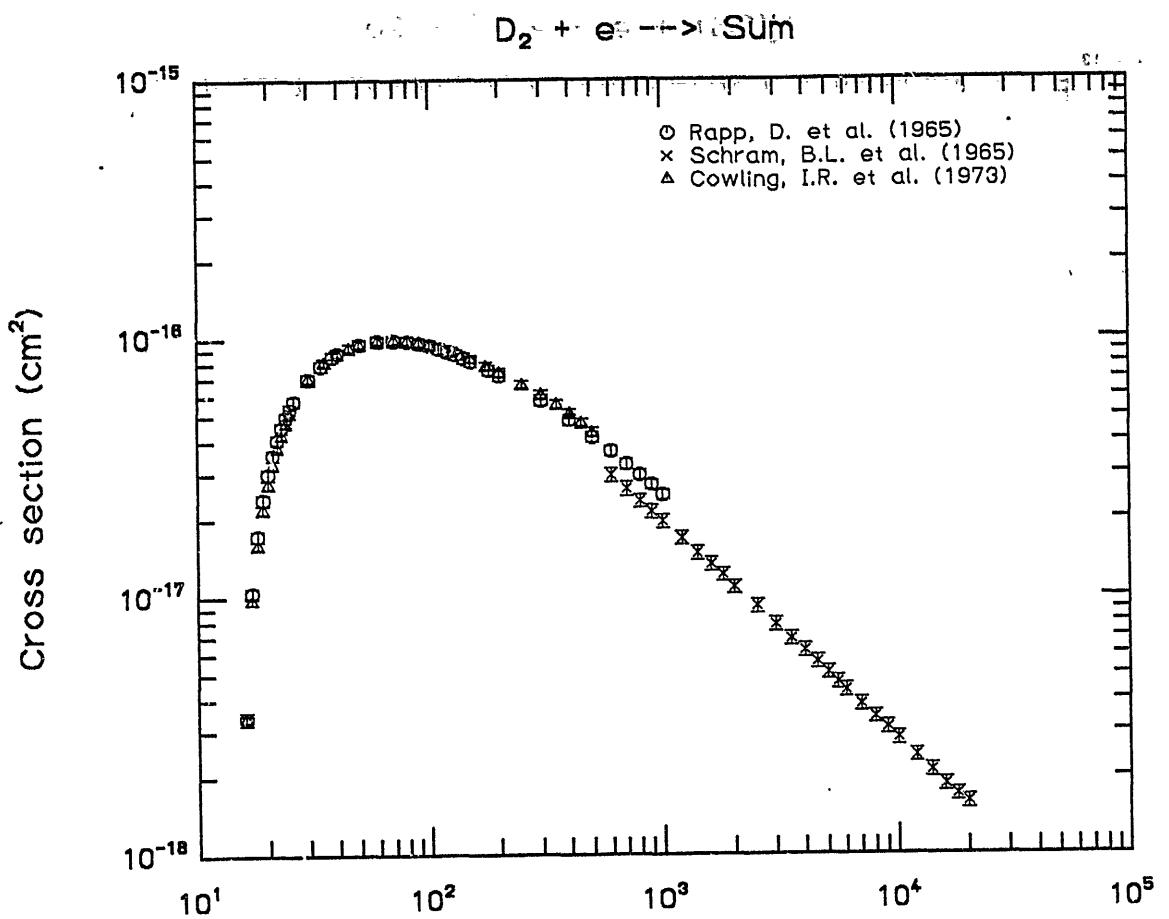
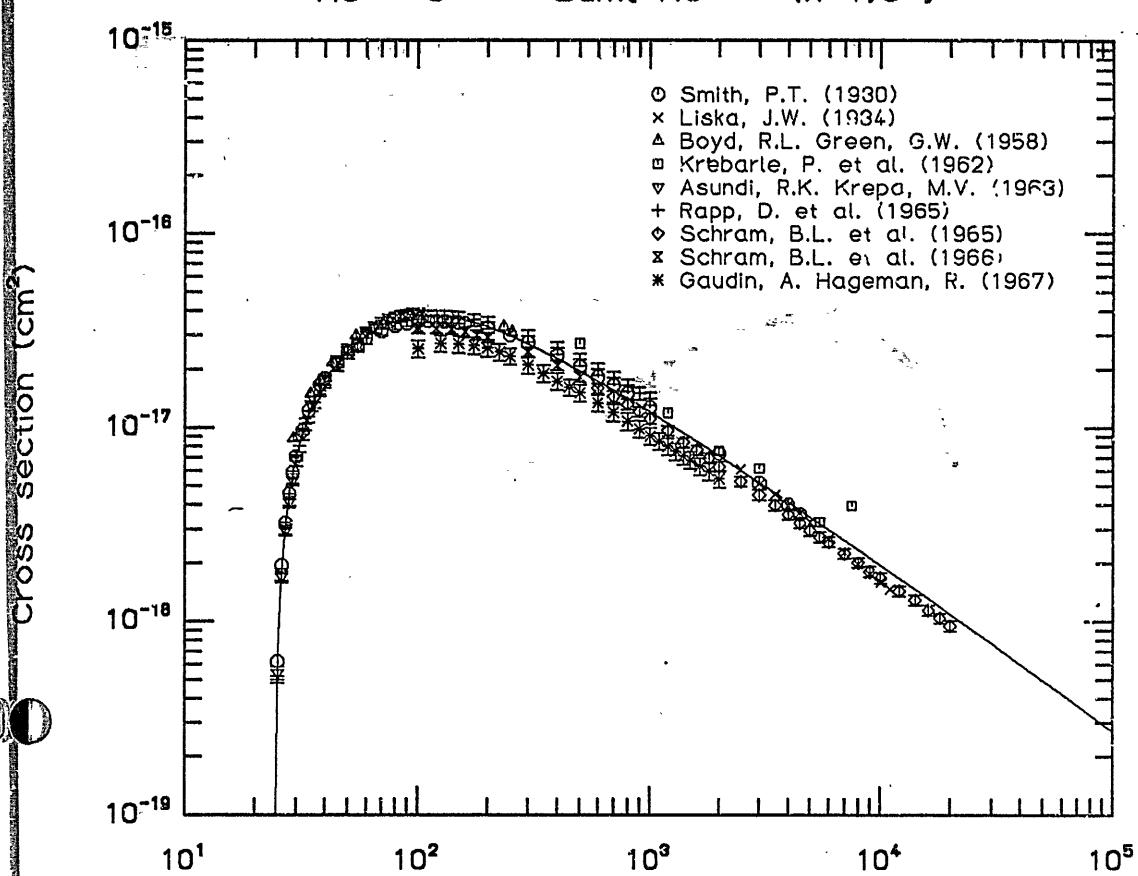


Fig. 6

Electron energy (eV)



Electron energy (eV)

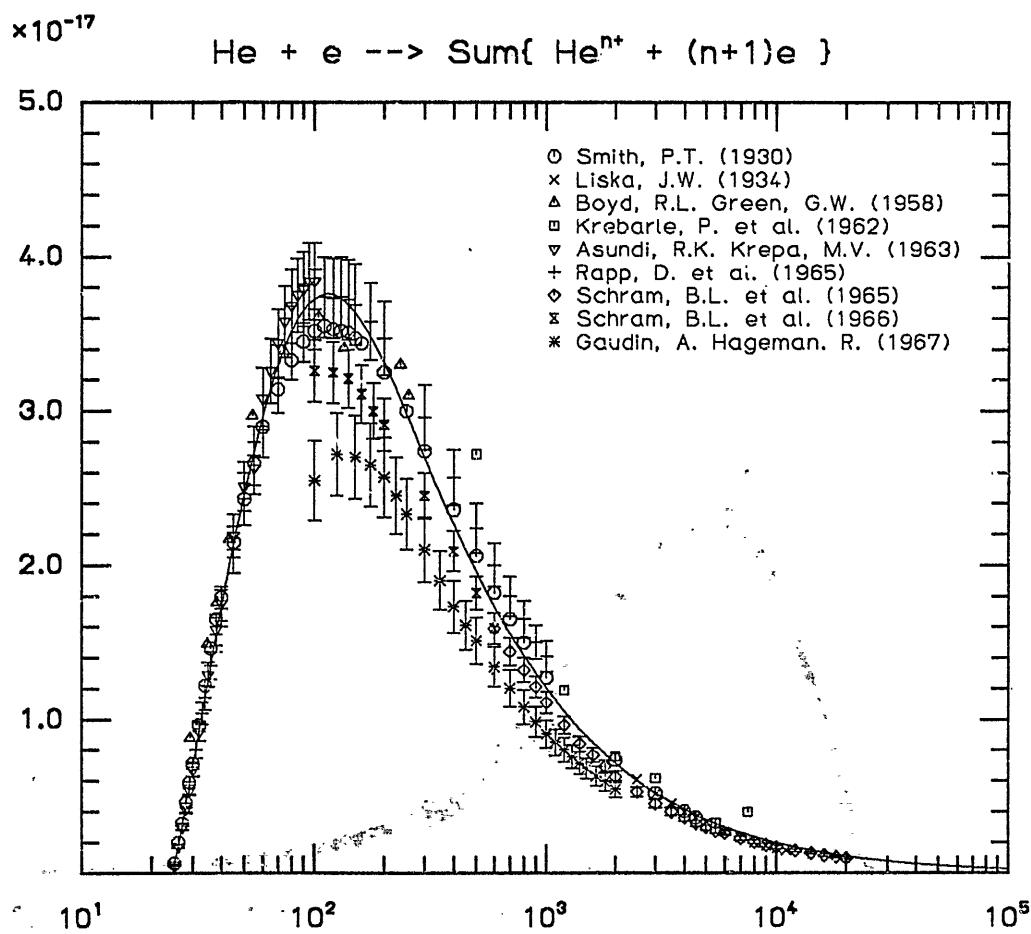


Fig. 7

Electron energy (eV)

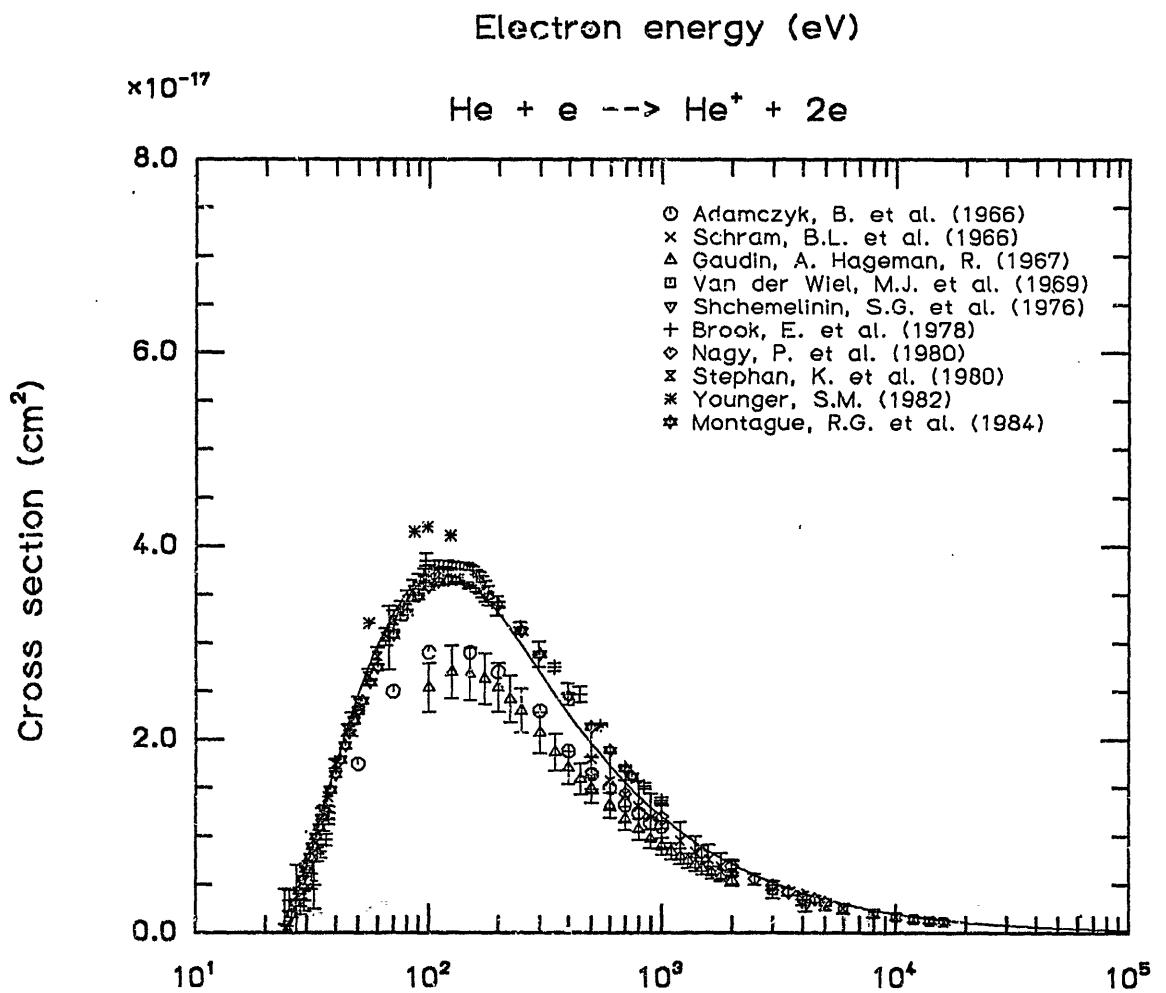
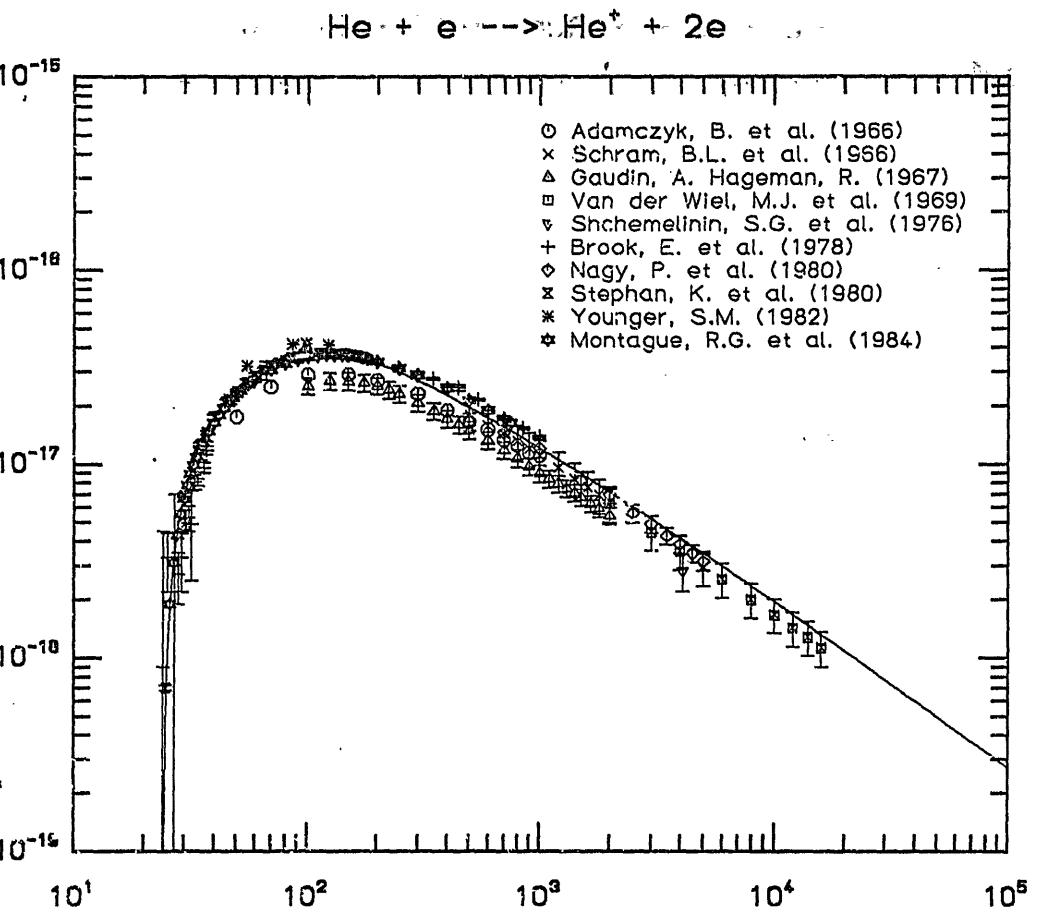


Fig. 8

Electron energy (eV)

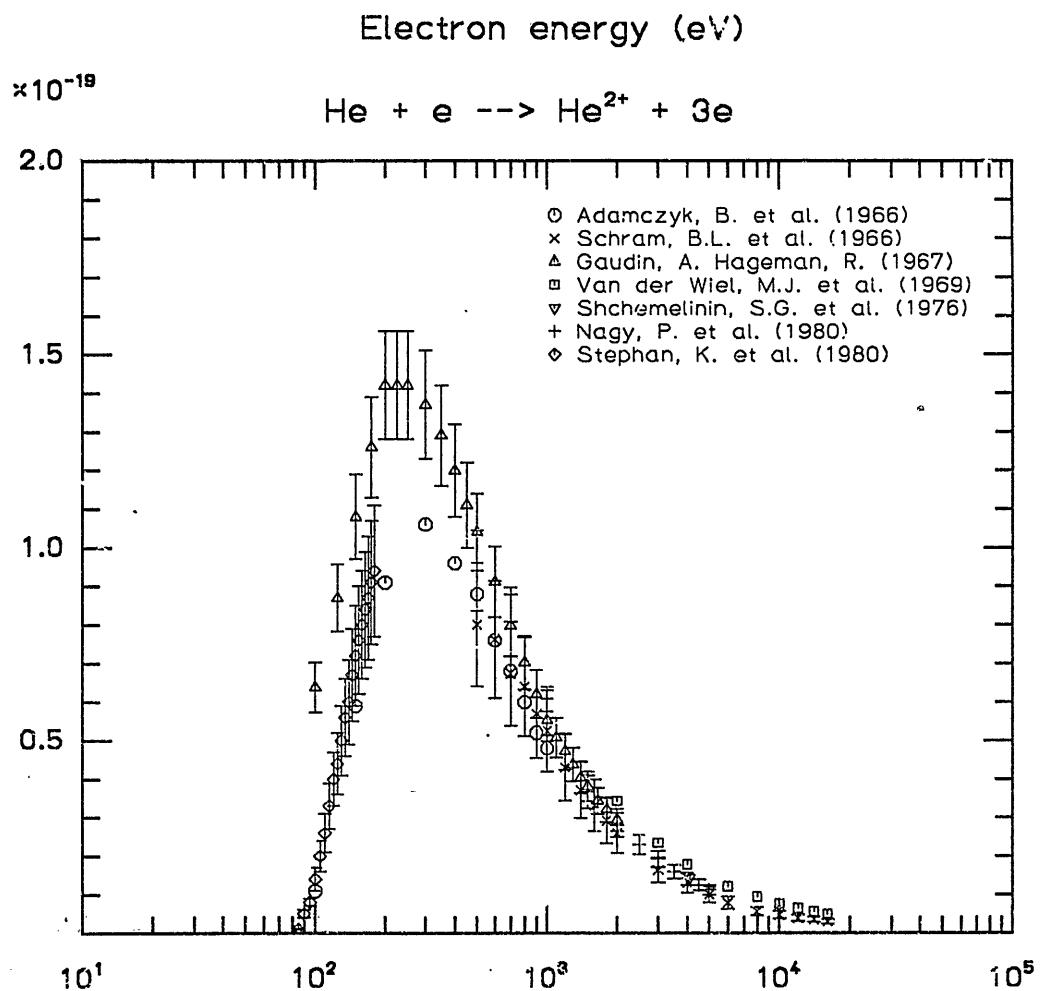
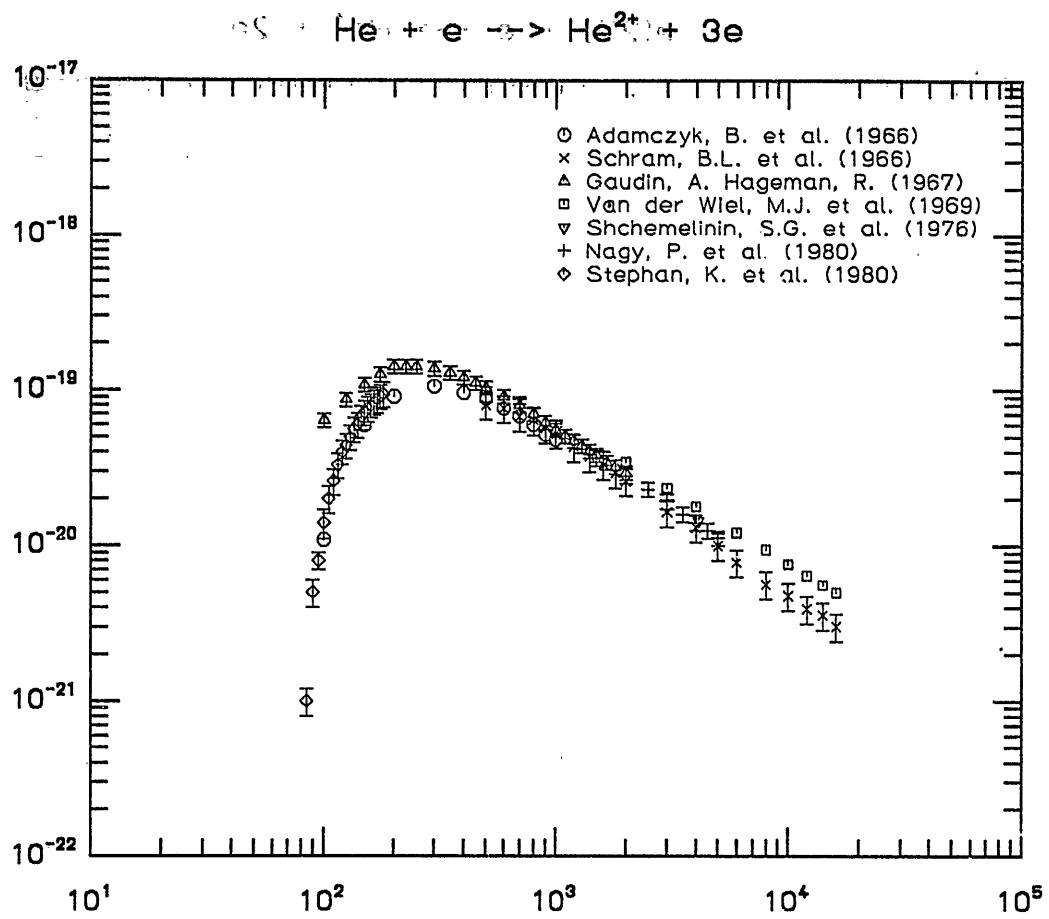


Fig. 9

Electron energy (eV)

- Adamczyk, B. et al. (1966)
- × Schram, B.L. et al. (1966)
- △ Gaudin, A. Hageman, R. (1967)
- Van der Wiel, M.J. et al. (1969)
- ▽ Shchemelinin, S.G. et al. (1976)
- + Nagy, P. et al. (1980)
- ◊ Stephan, K. et al. (1980)

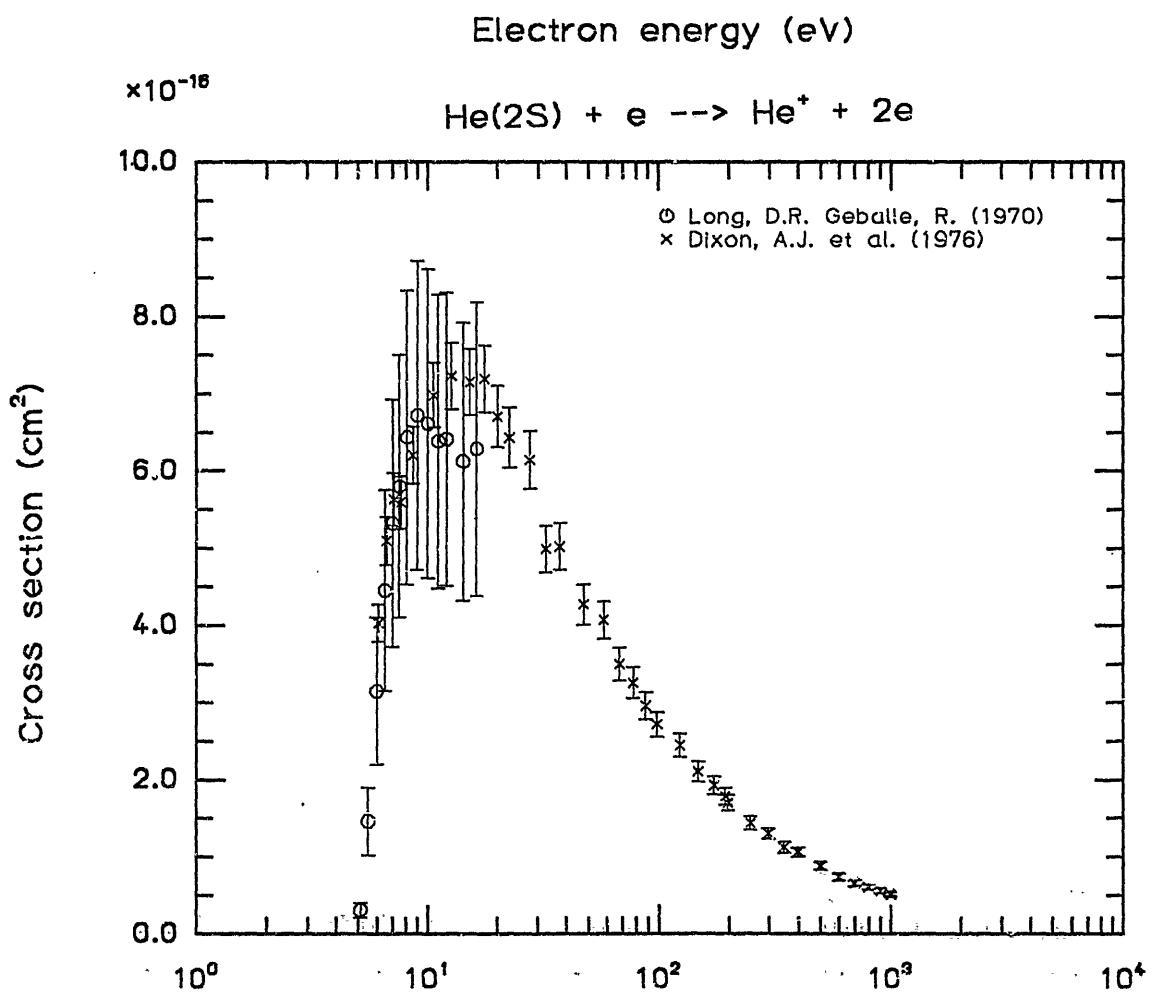
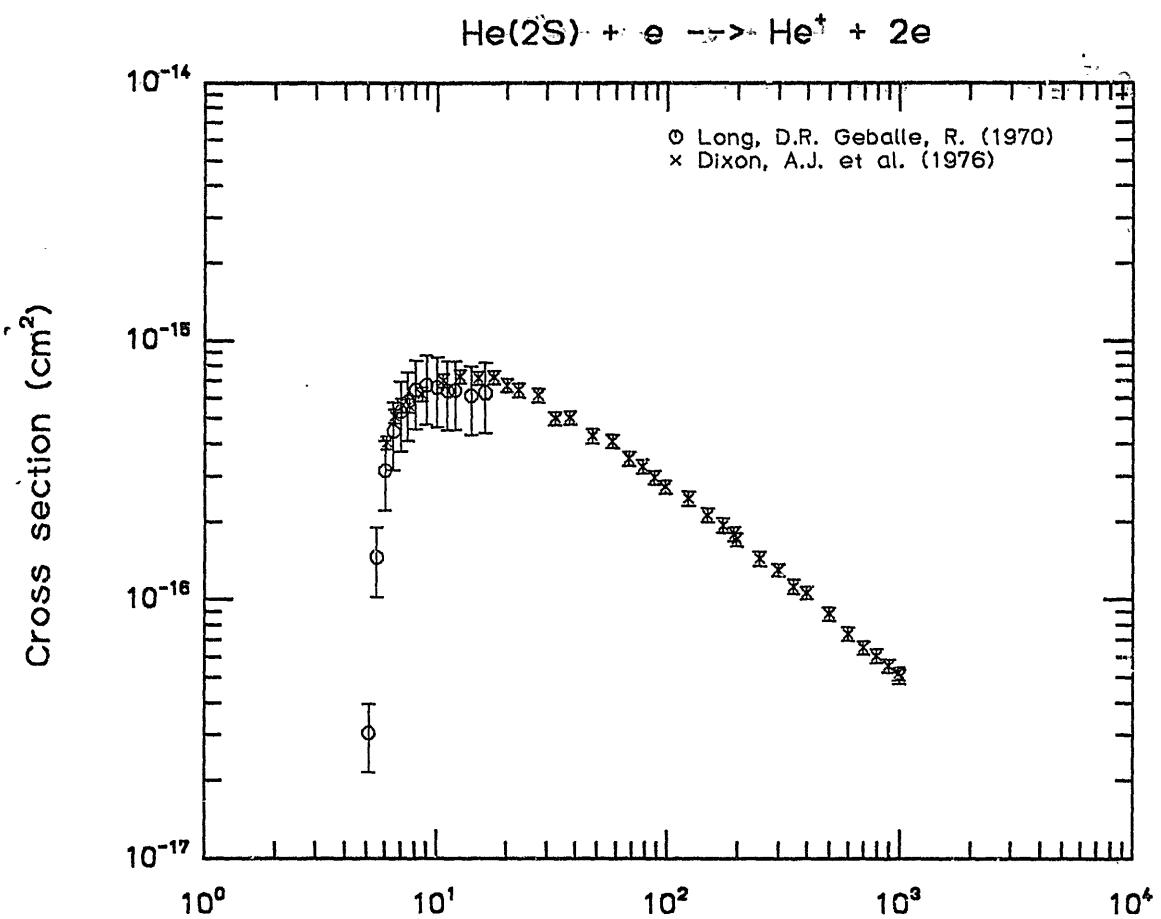
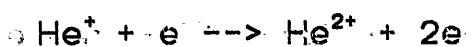


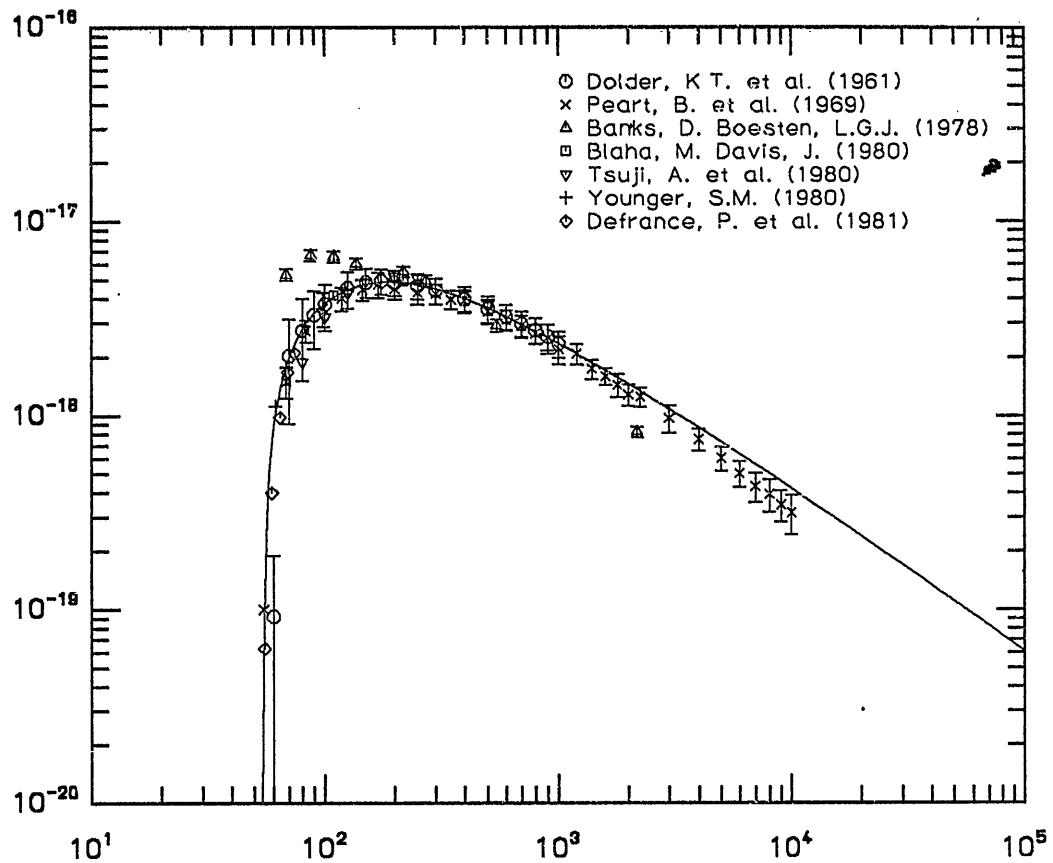
Fig. 10

Electron energy (eV)

C. P. J.



Cross section ( $\text{cm}^2$ )



Cross section ( $\text{cm}^2$ )

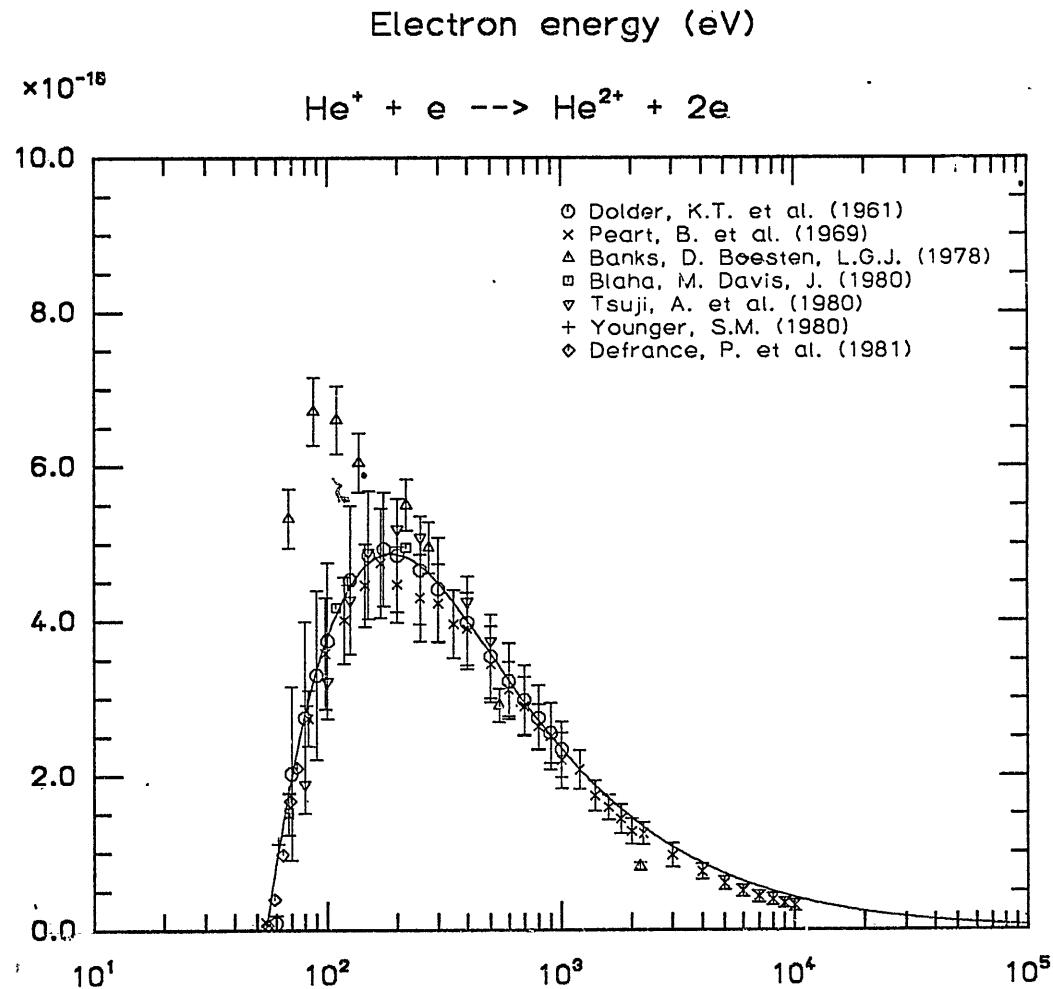


Fig. 11

Electron energy (eV)

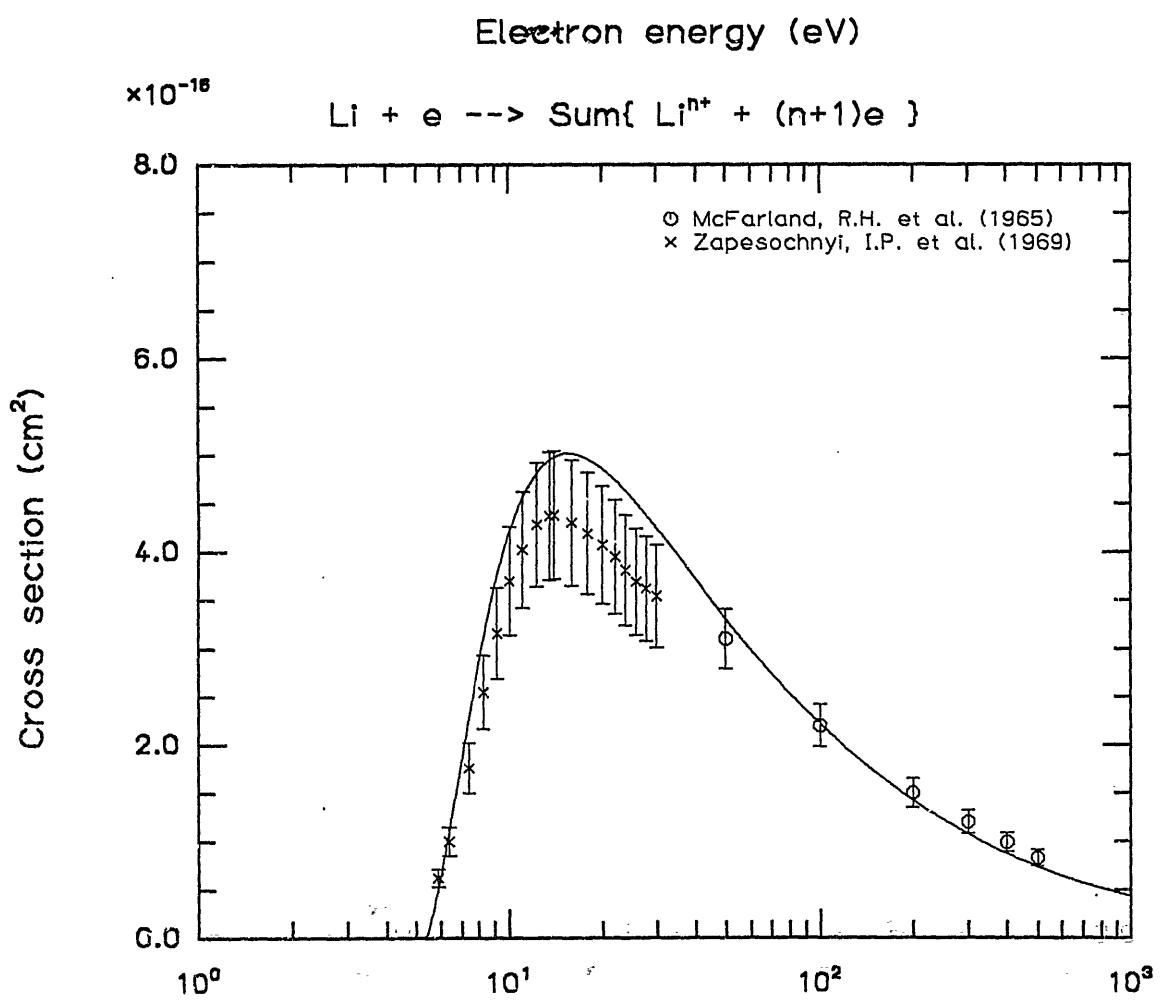
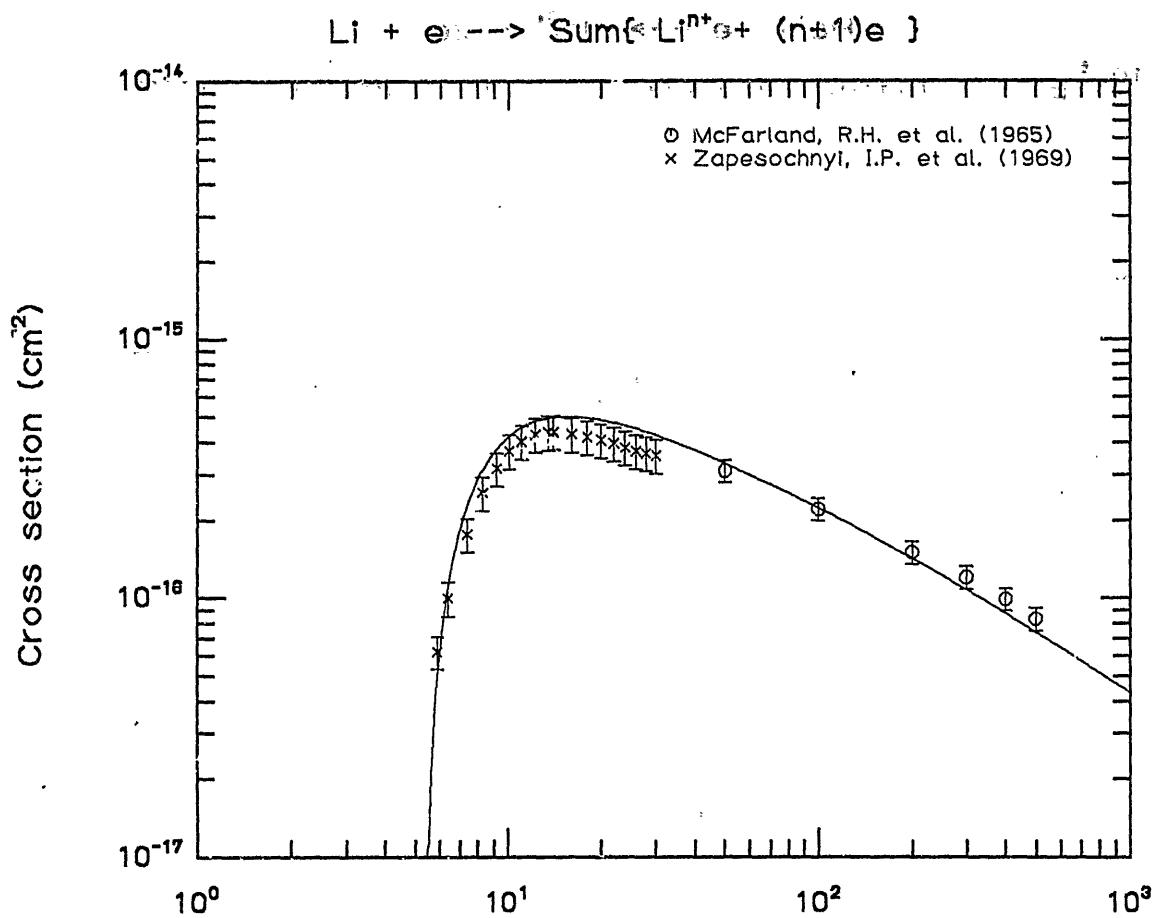


Fig. 12

Electron energy (eV)

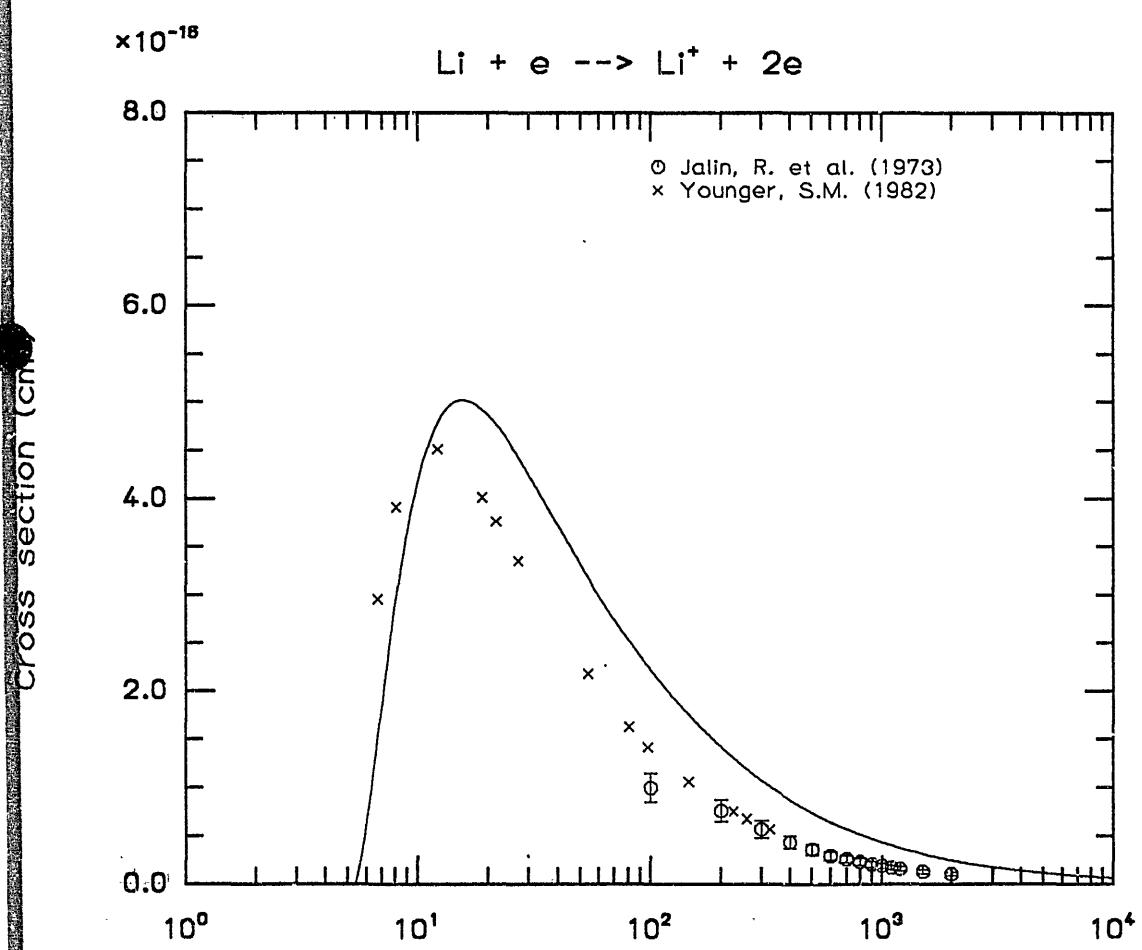
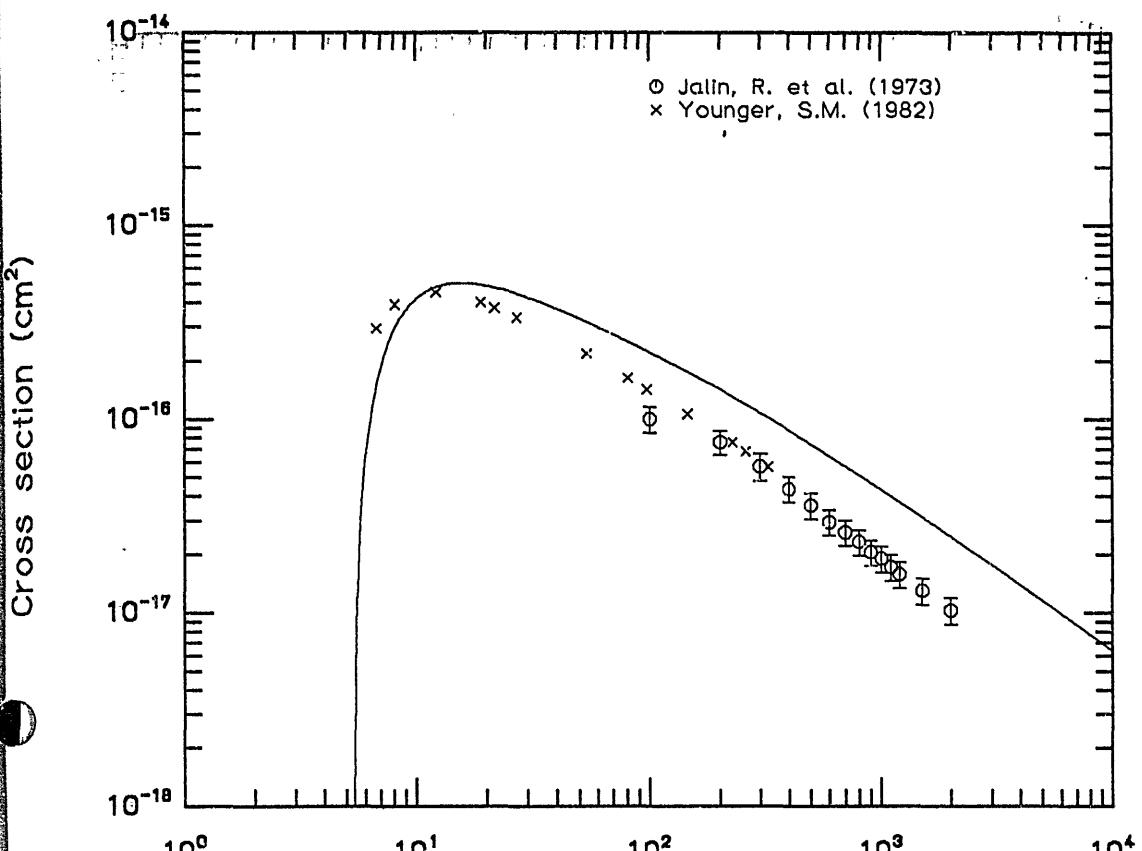
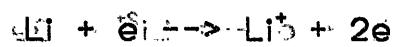


Fig. 13

Electron energy (eV)

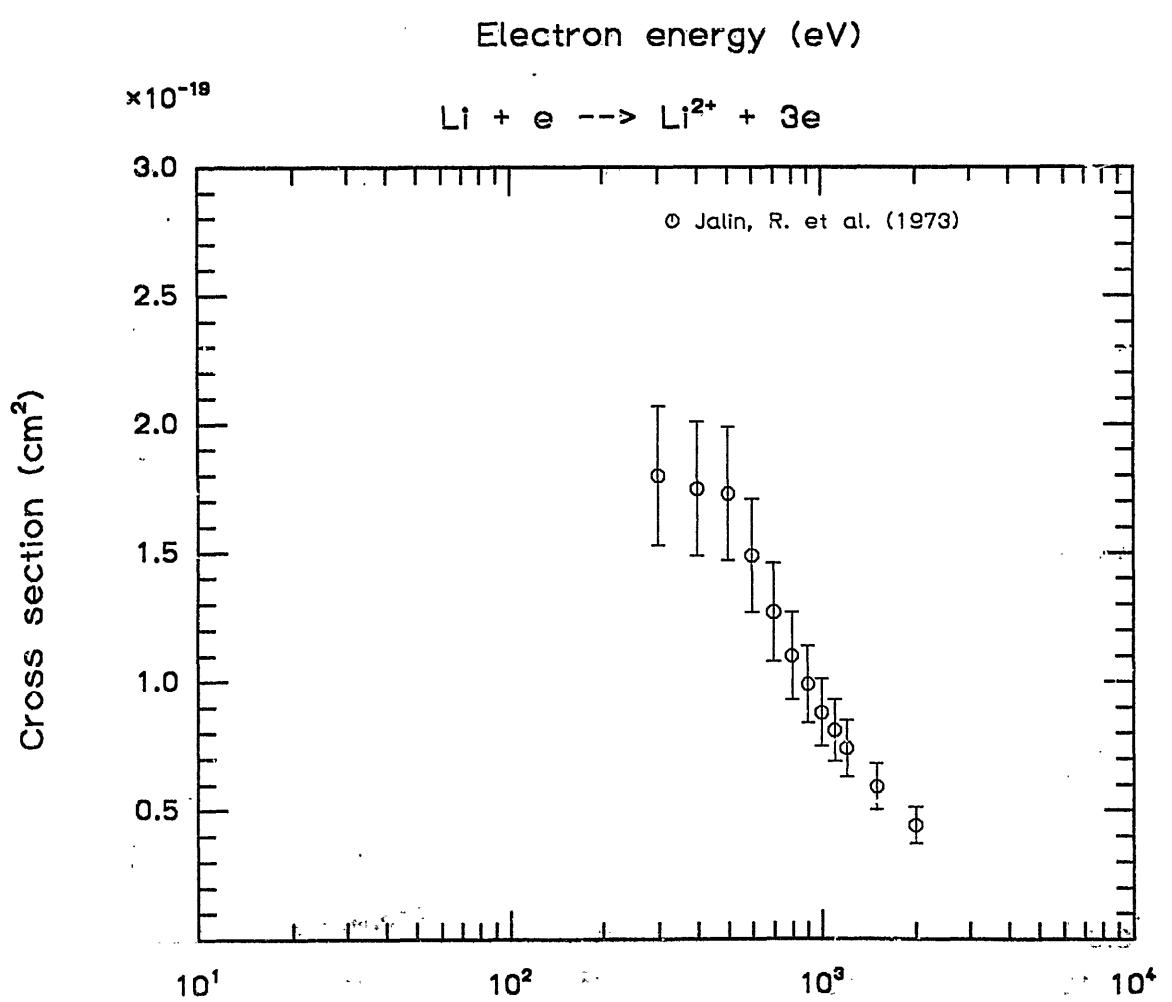
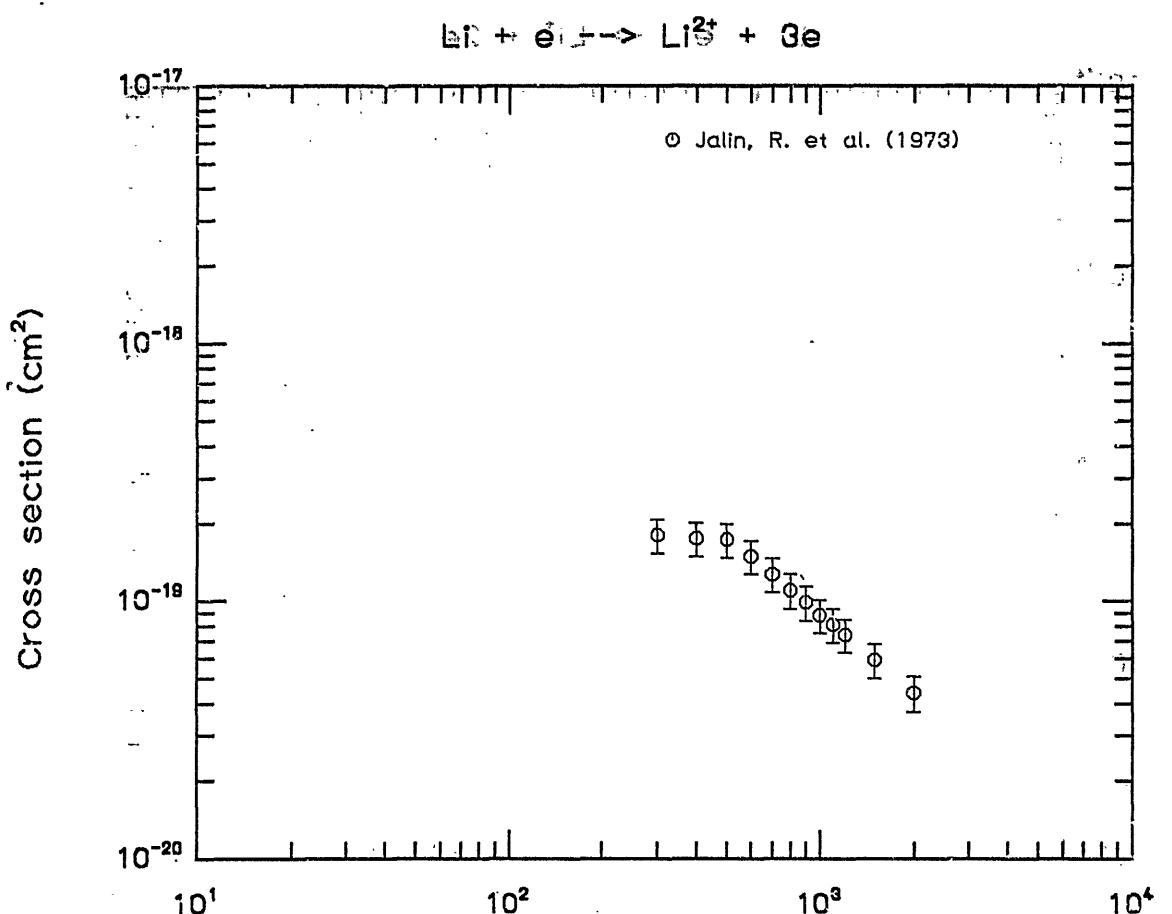


Fig. 14

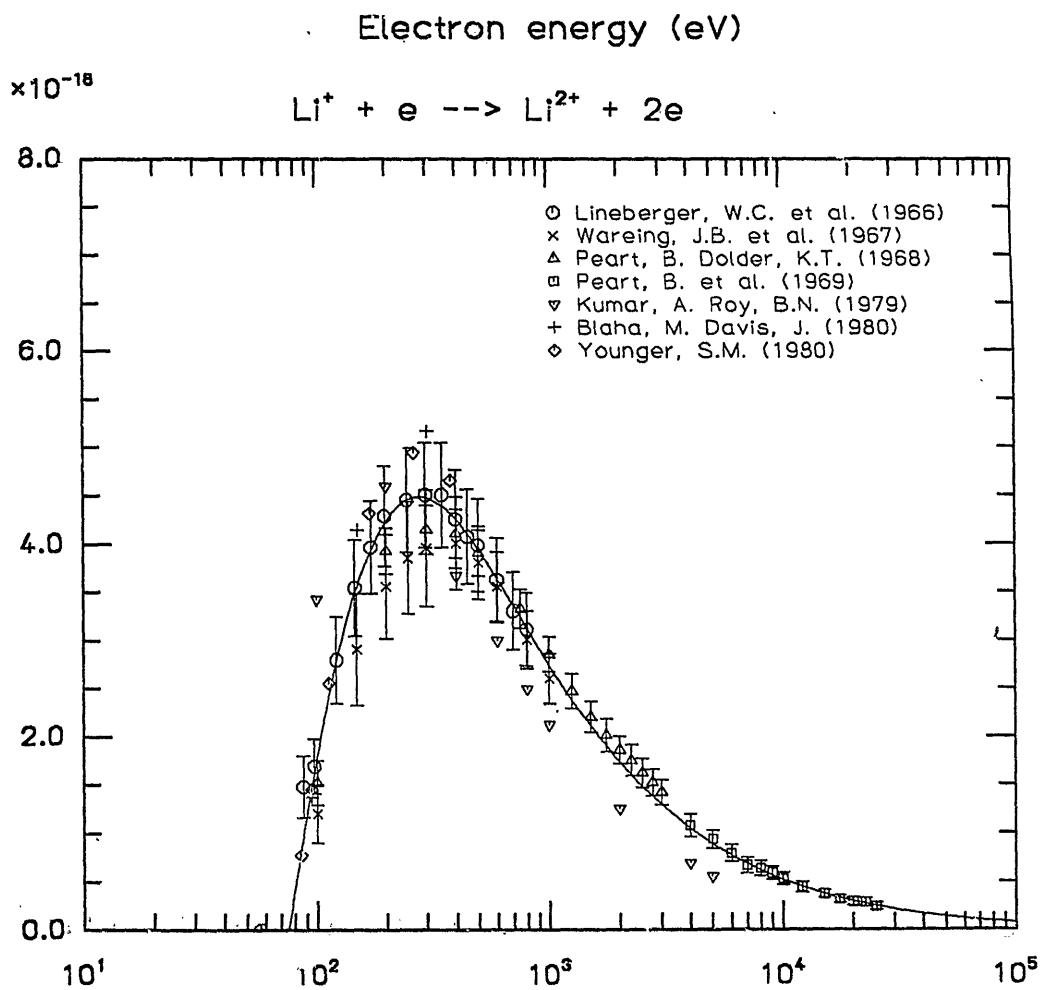
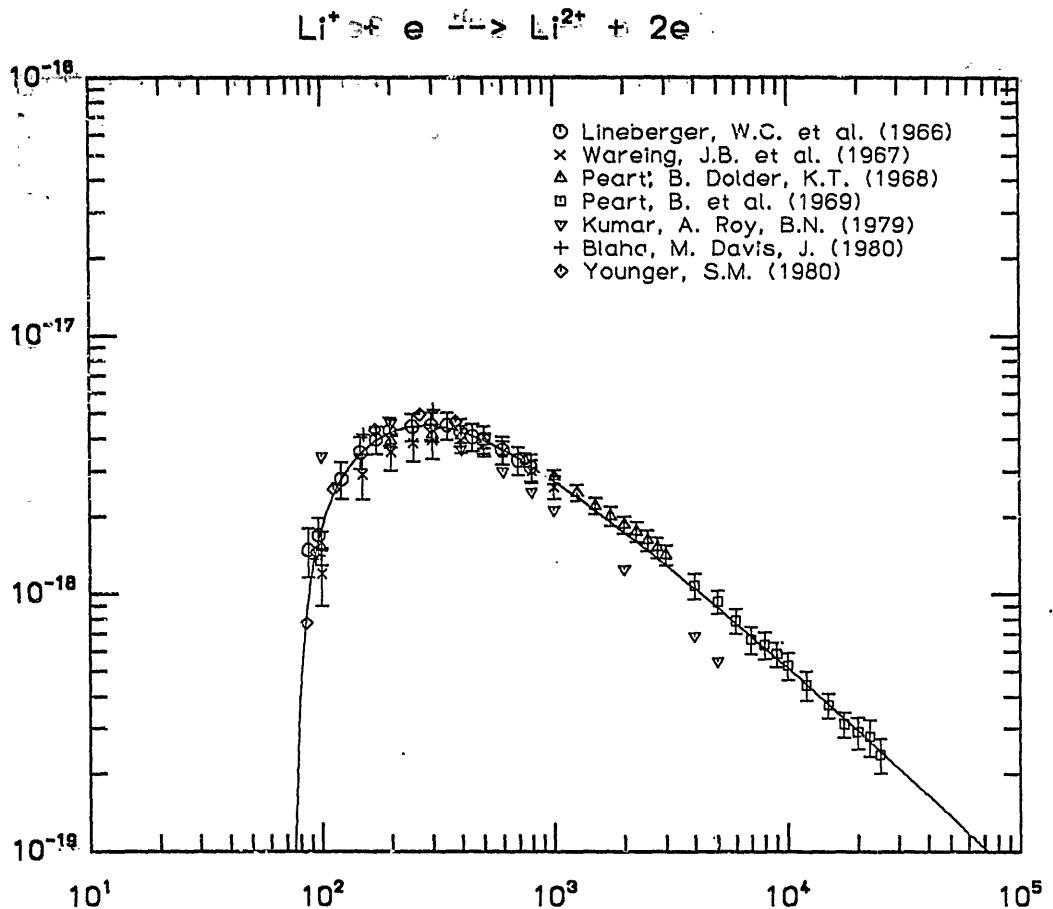


Fig. 15      Electron energy (eV)

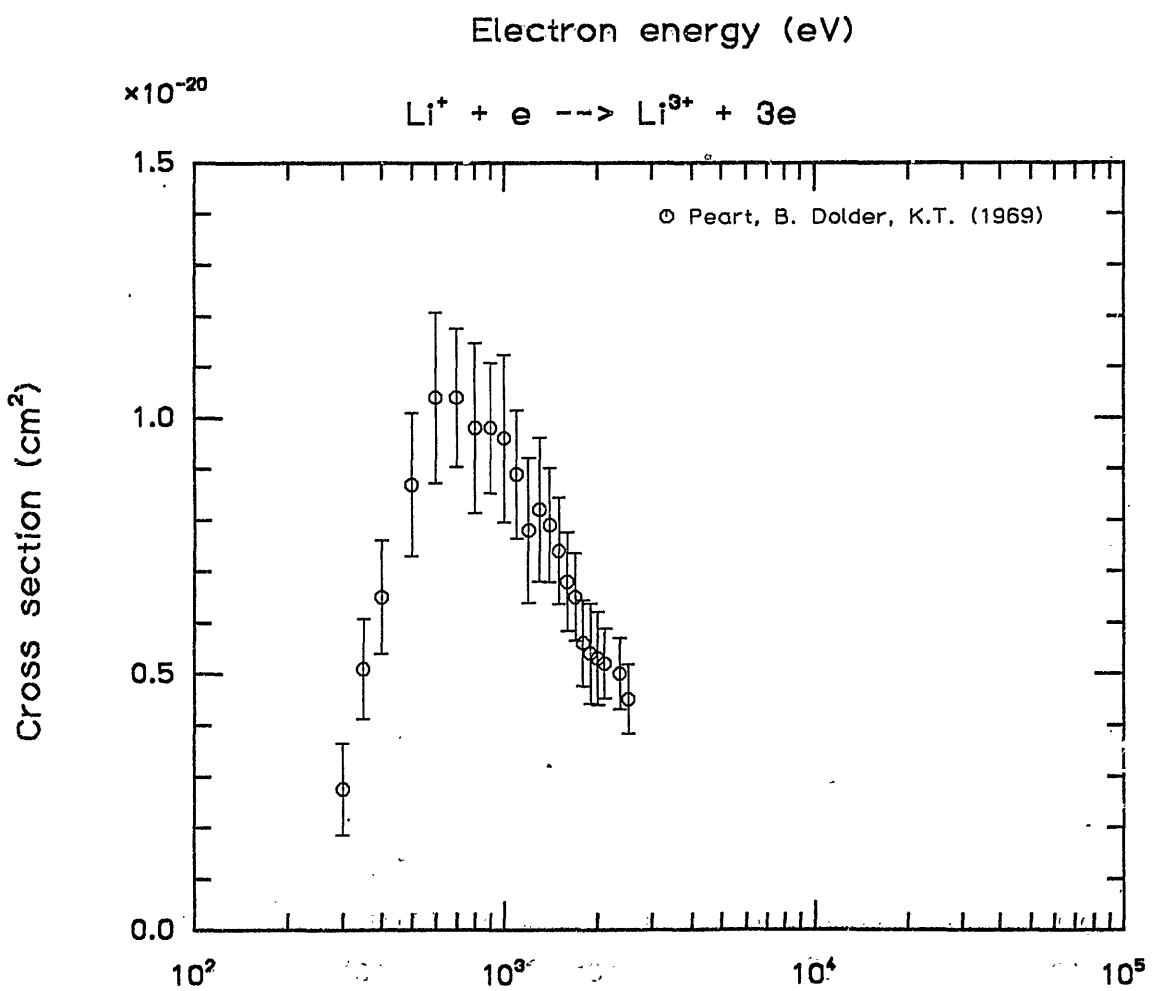
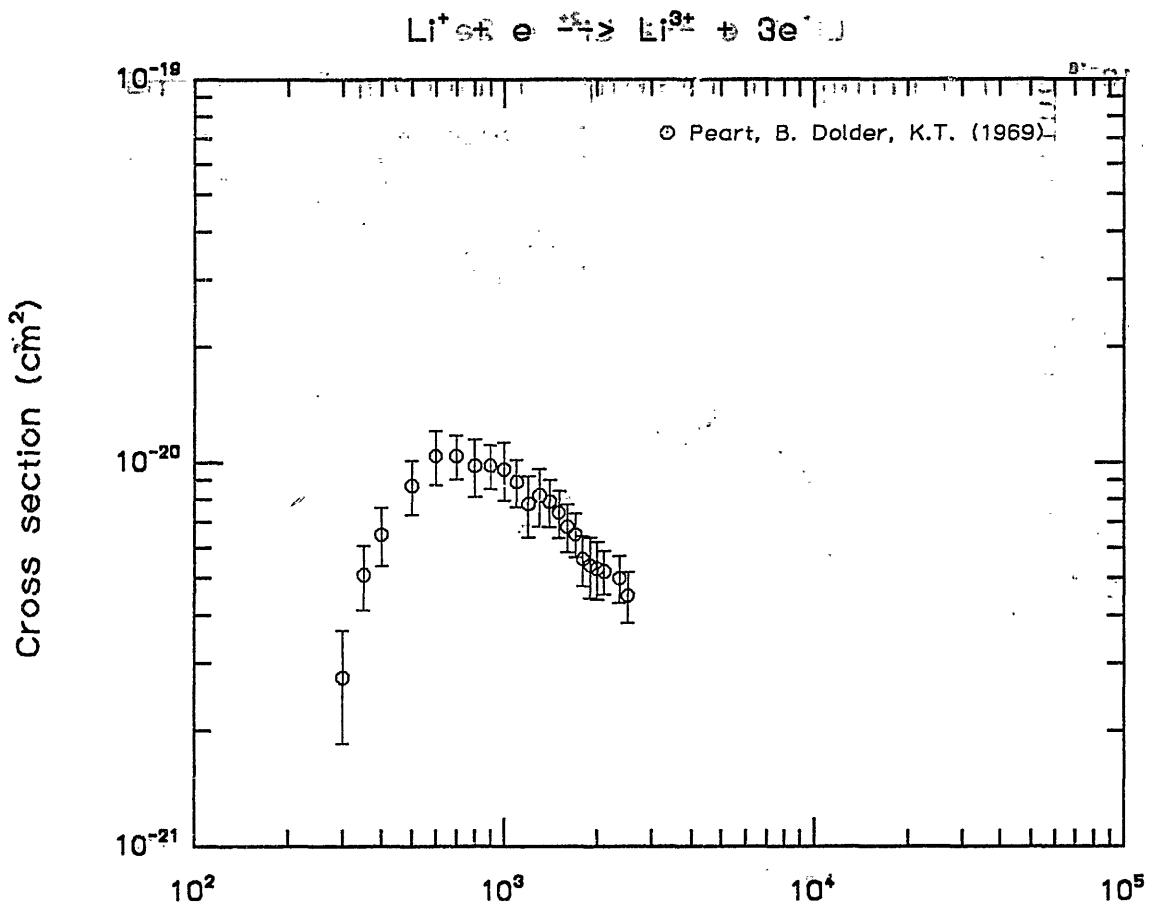


Fig. 16

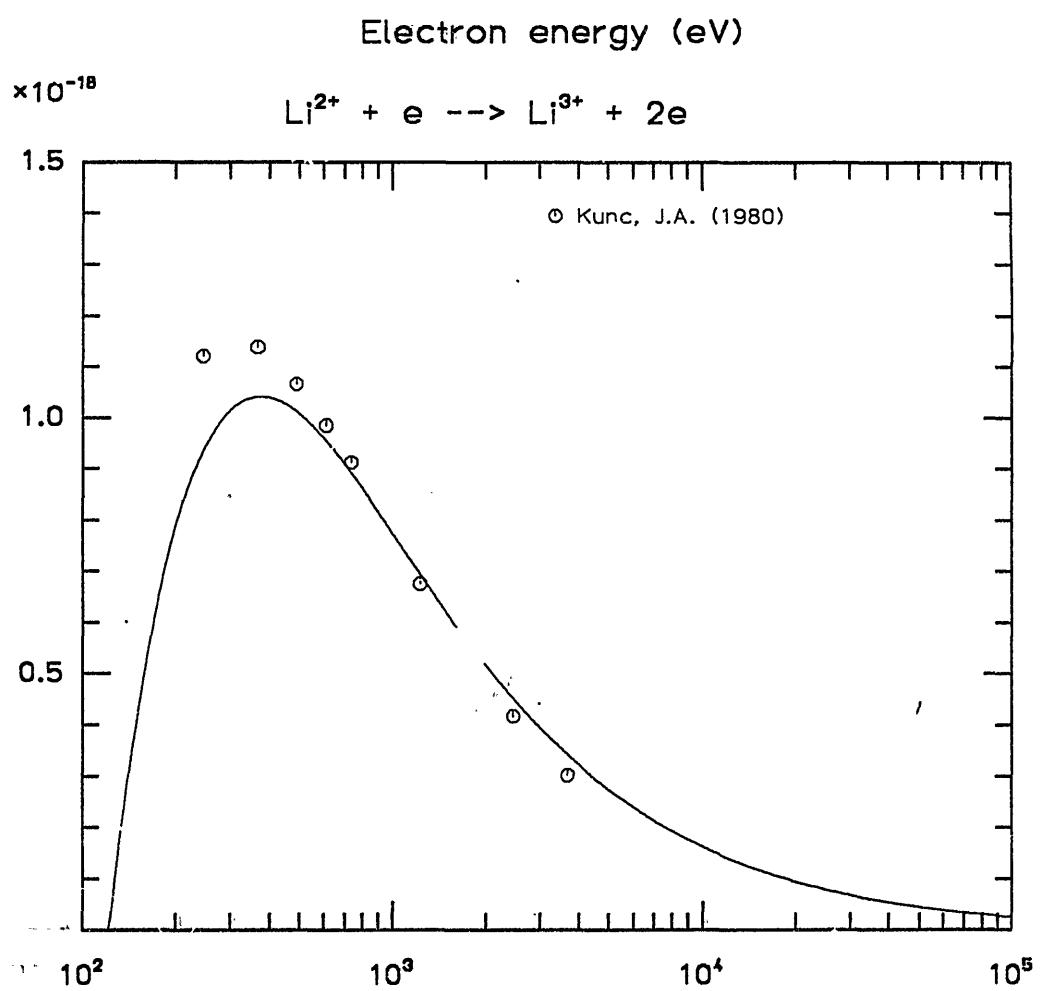
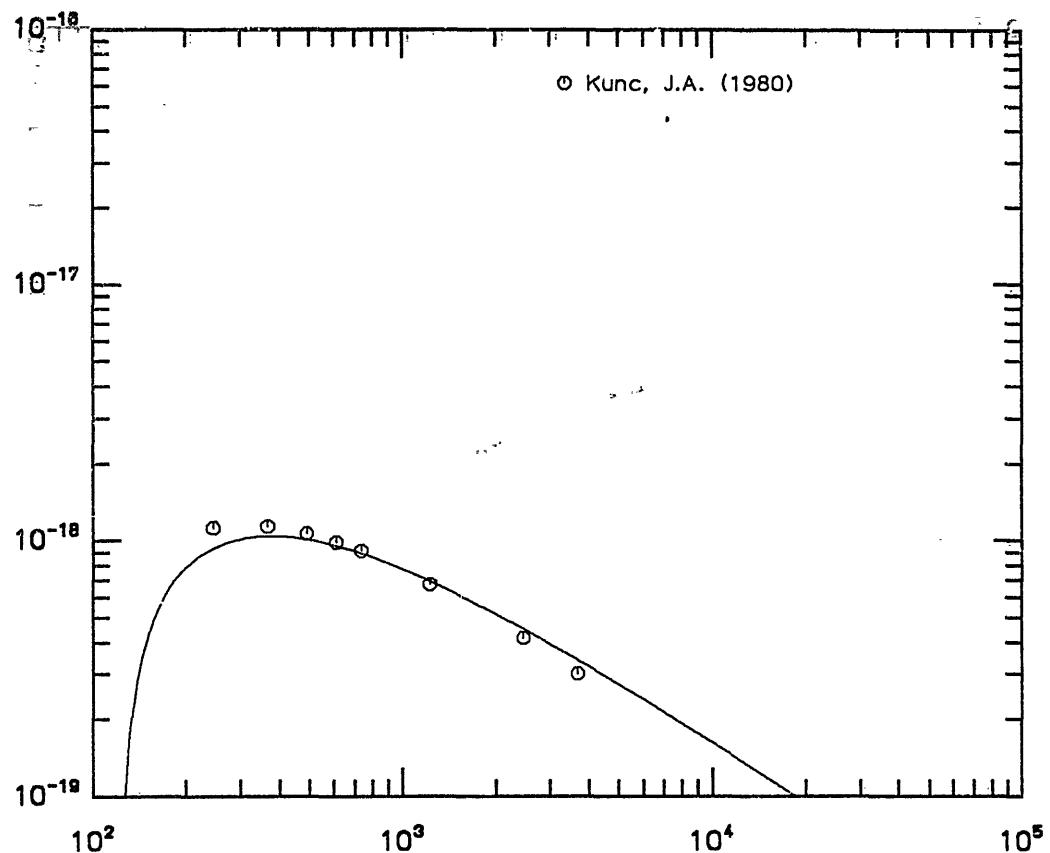
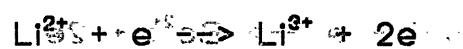


Fig. 17      Electron energy (eV)

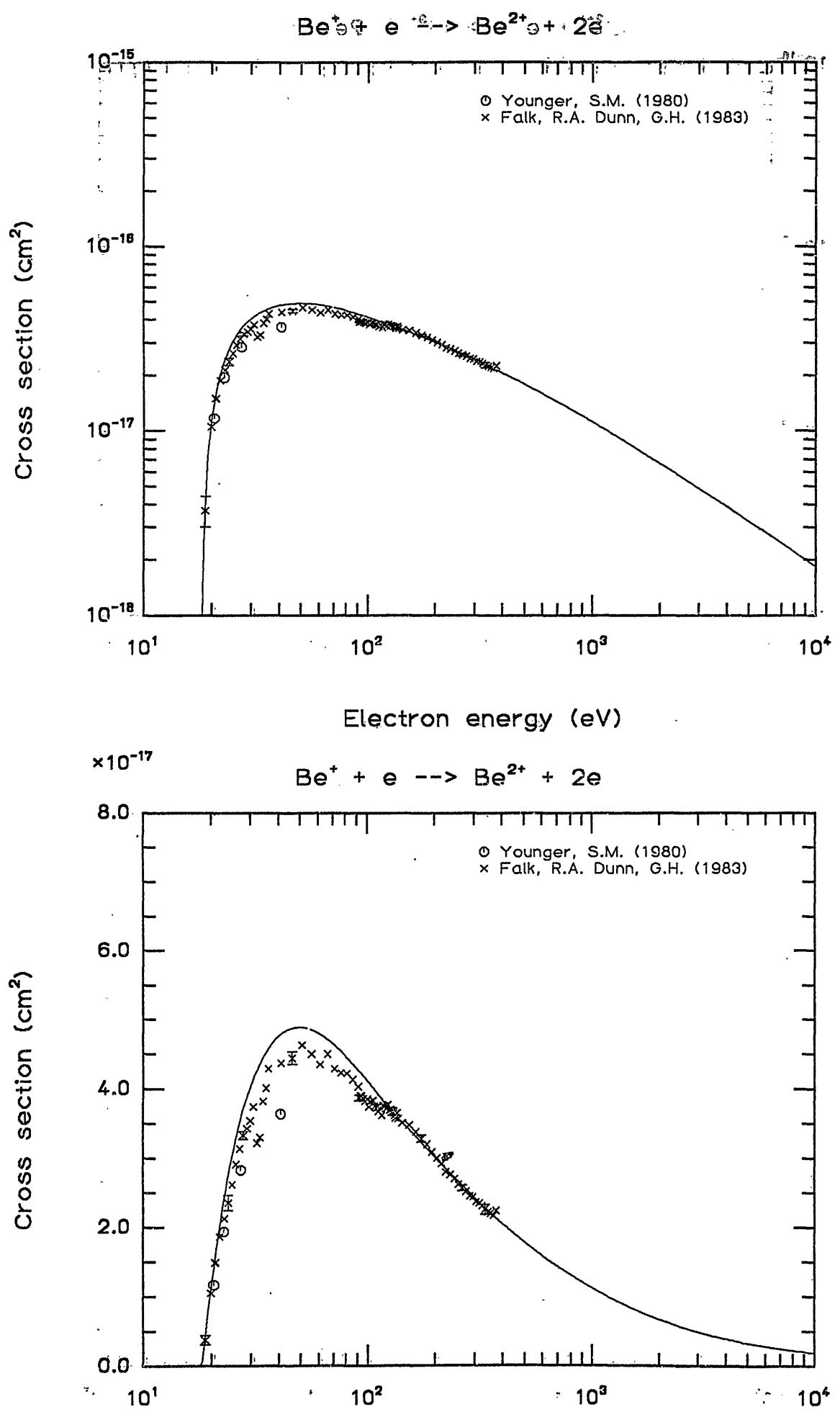
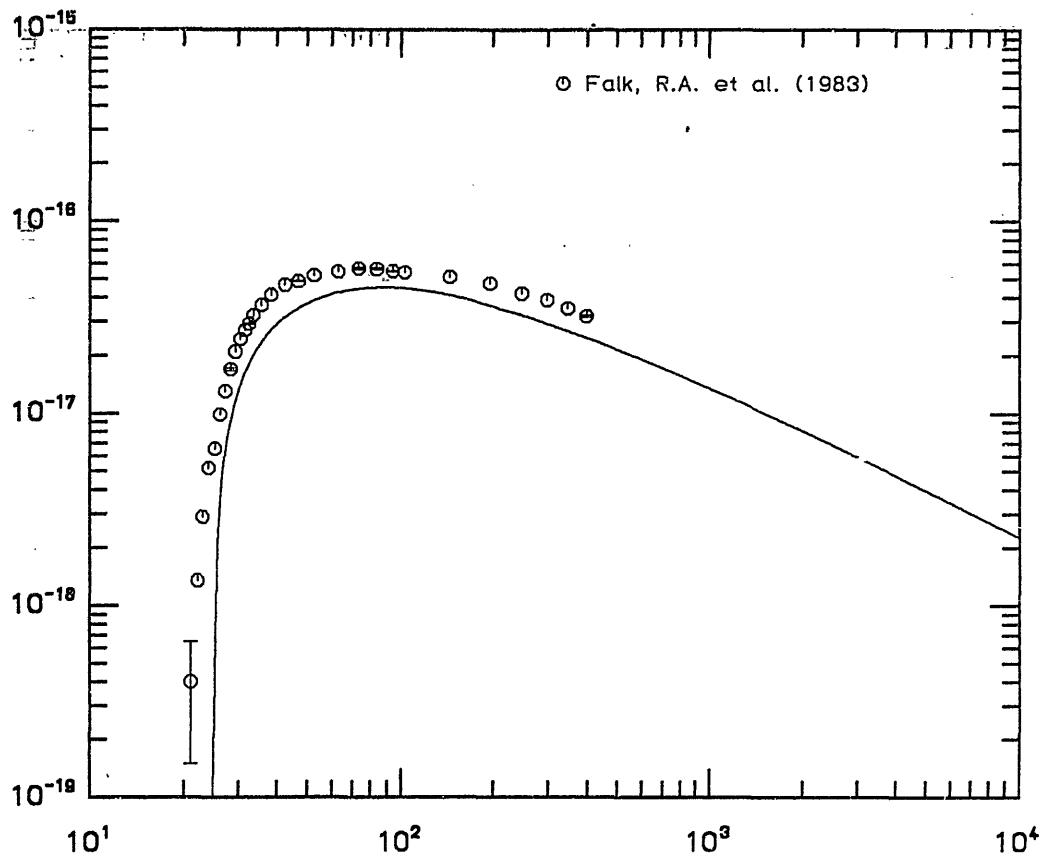
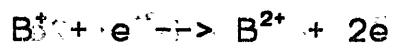


Fig. 18

Electron energy (eV)

VI E7



Electron energy (eV)

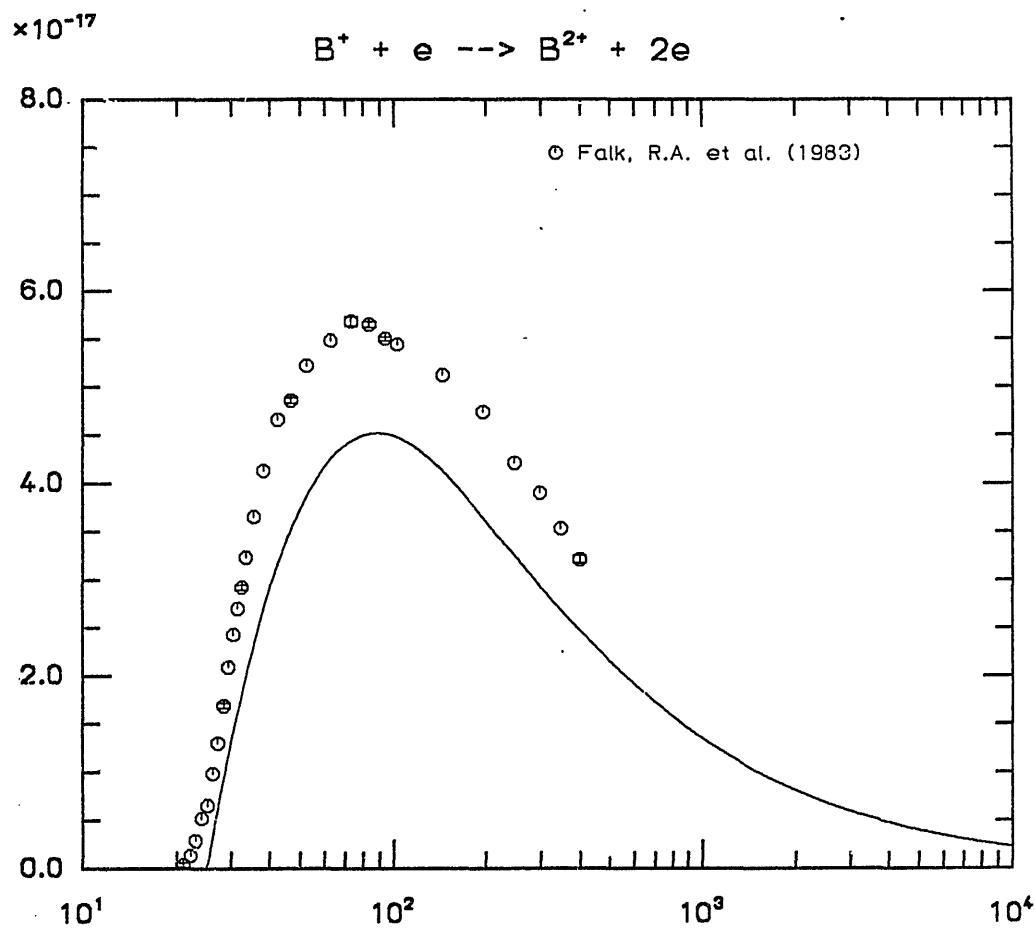


Fig. 19      Electron energy (eV)

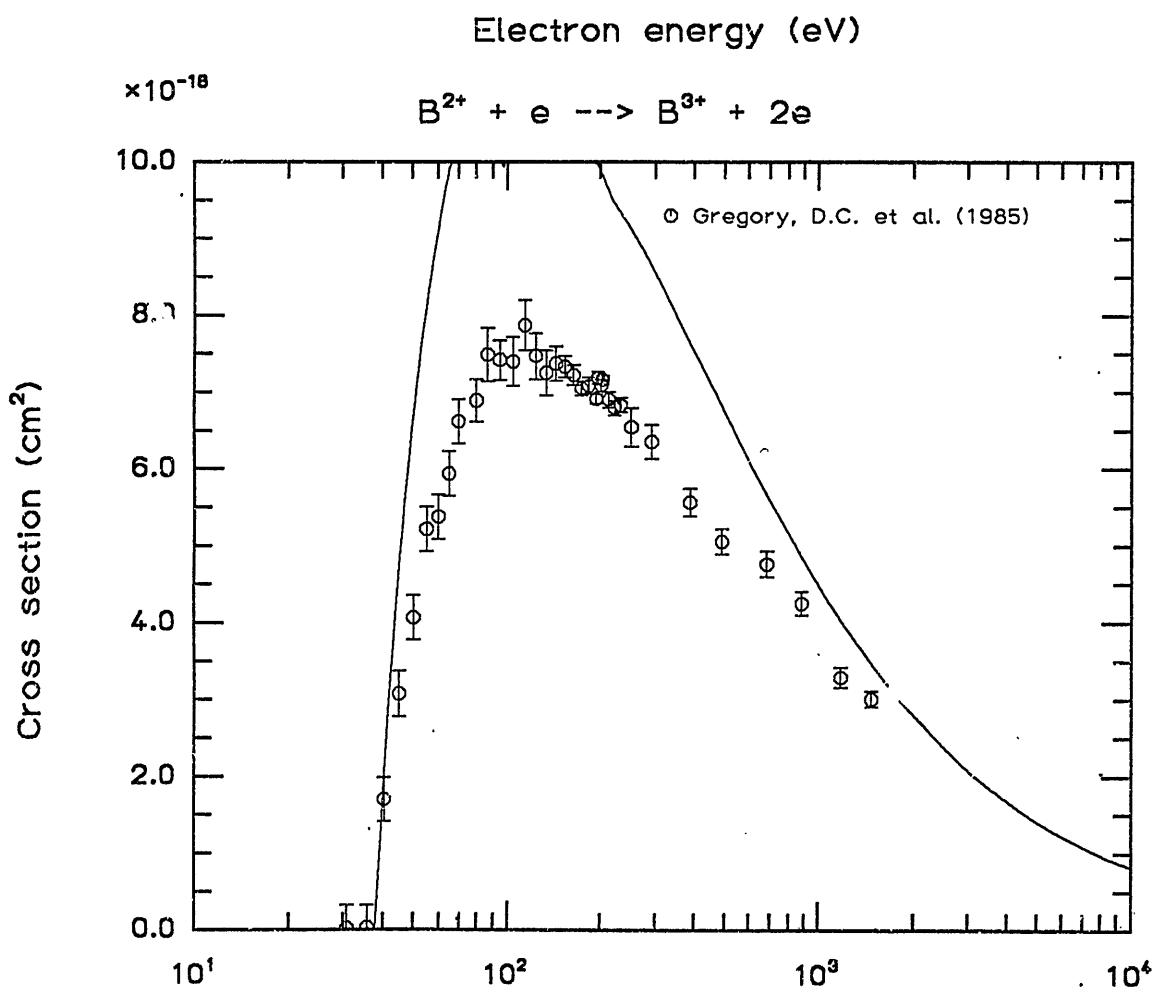
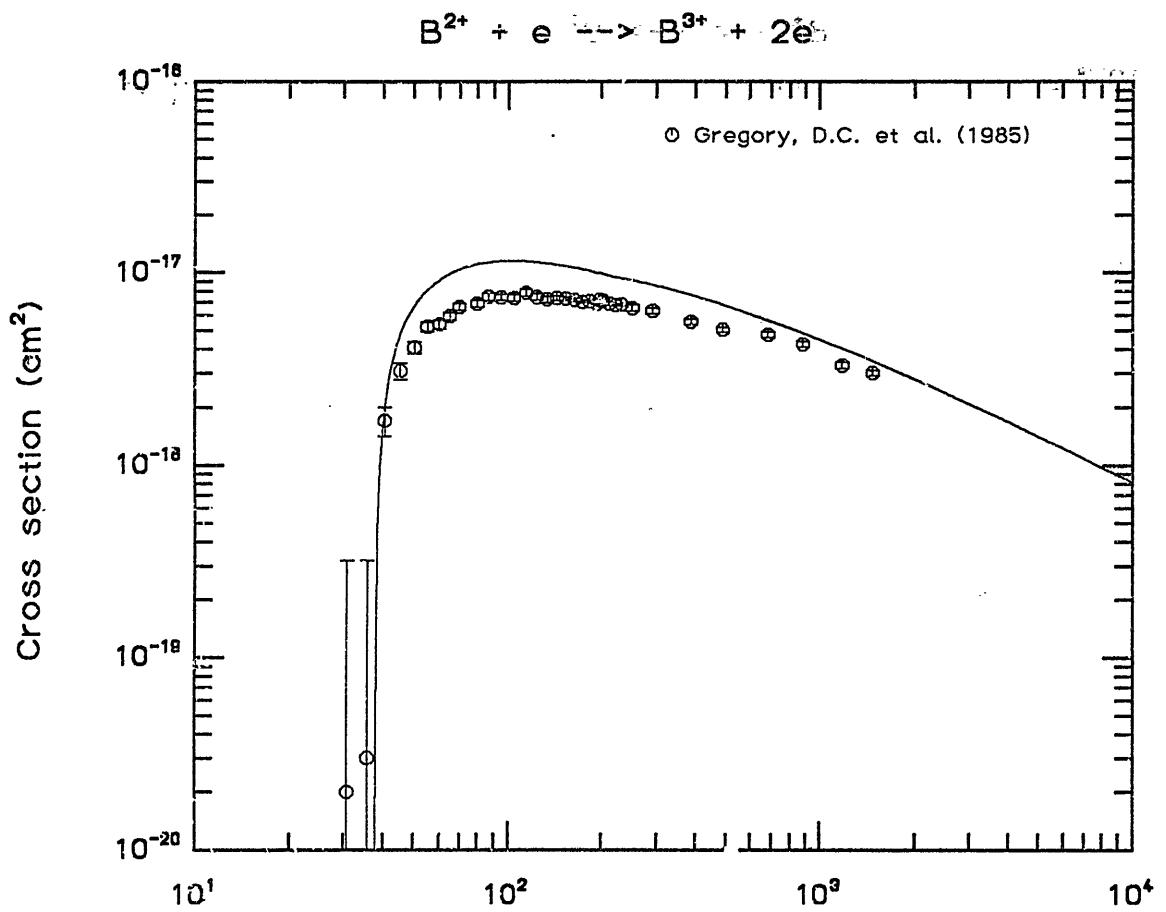


Fig. 20

Electron energy (eV)

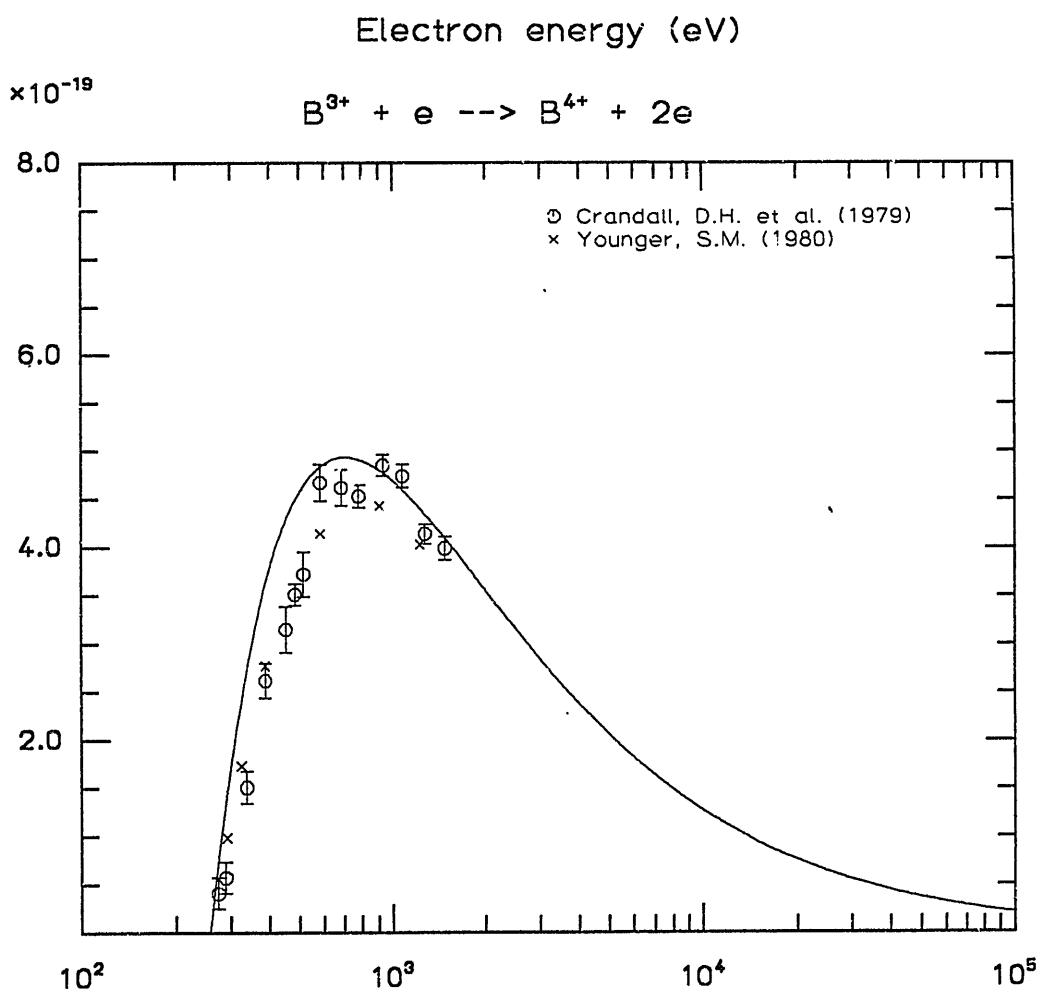
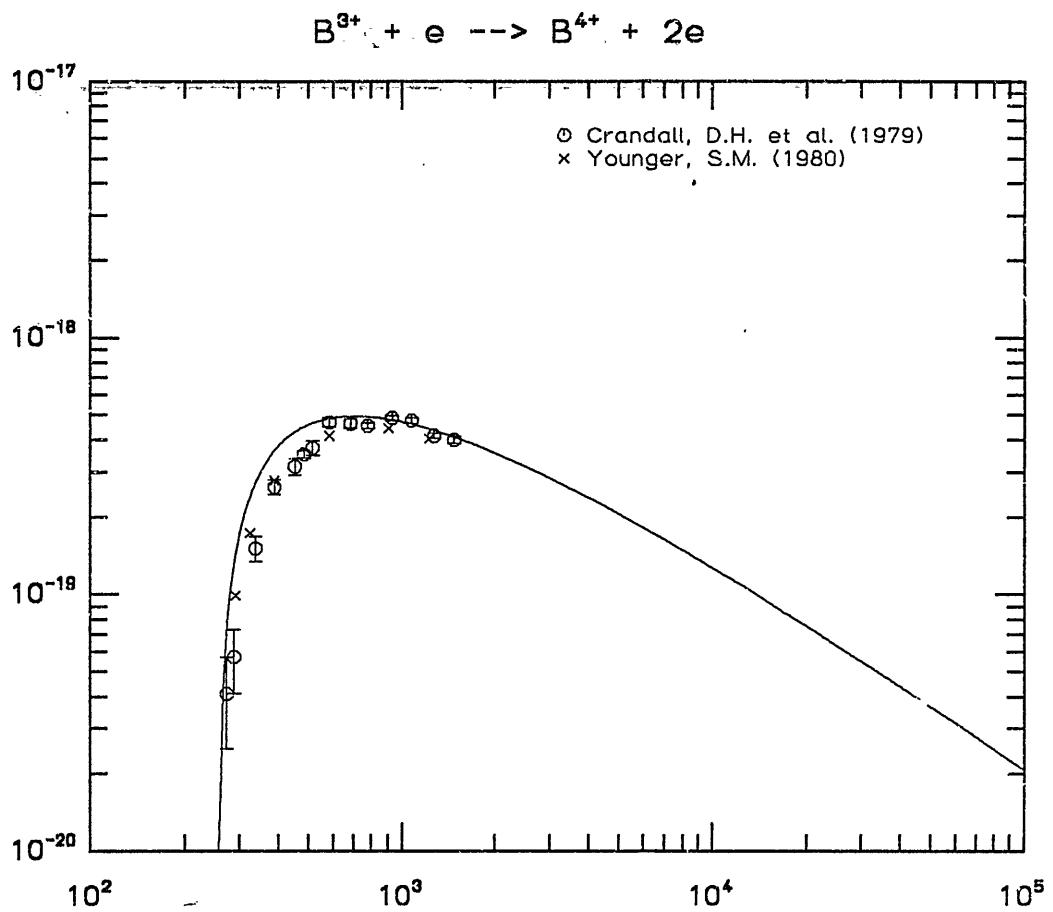


Fig. 2.1 Electron energy (eV)

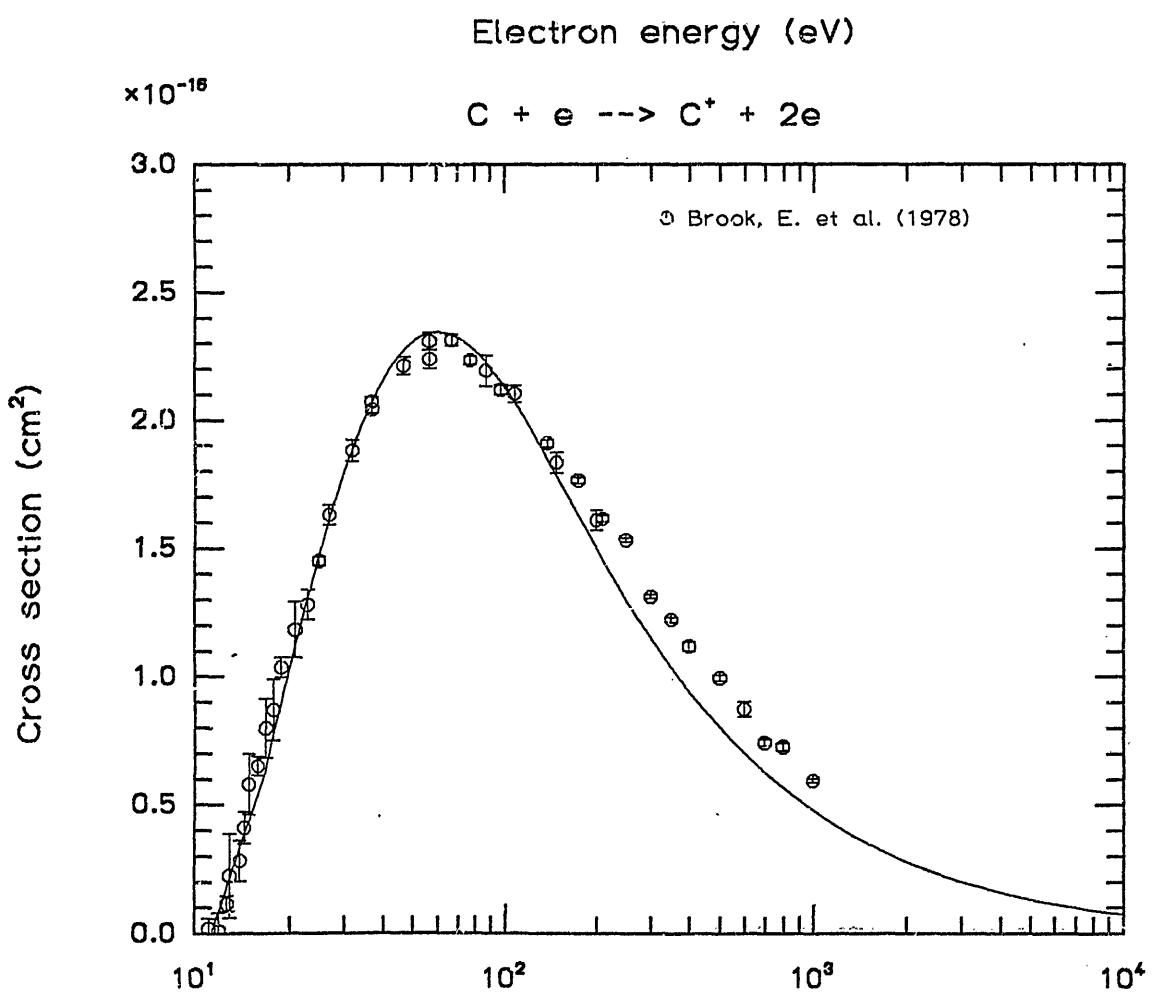
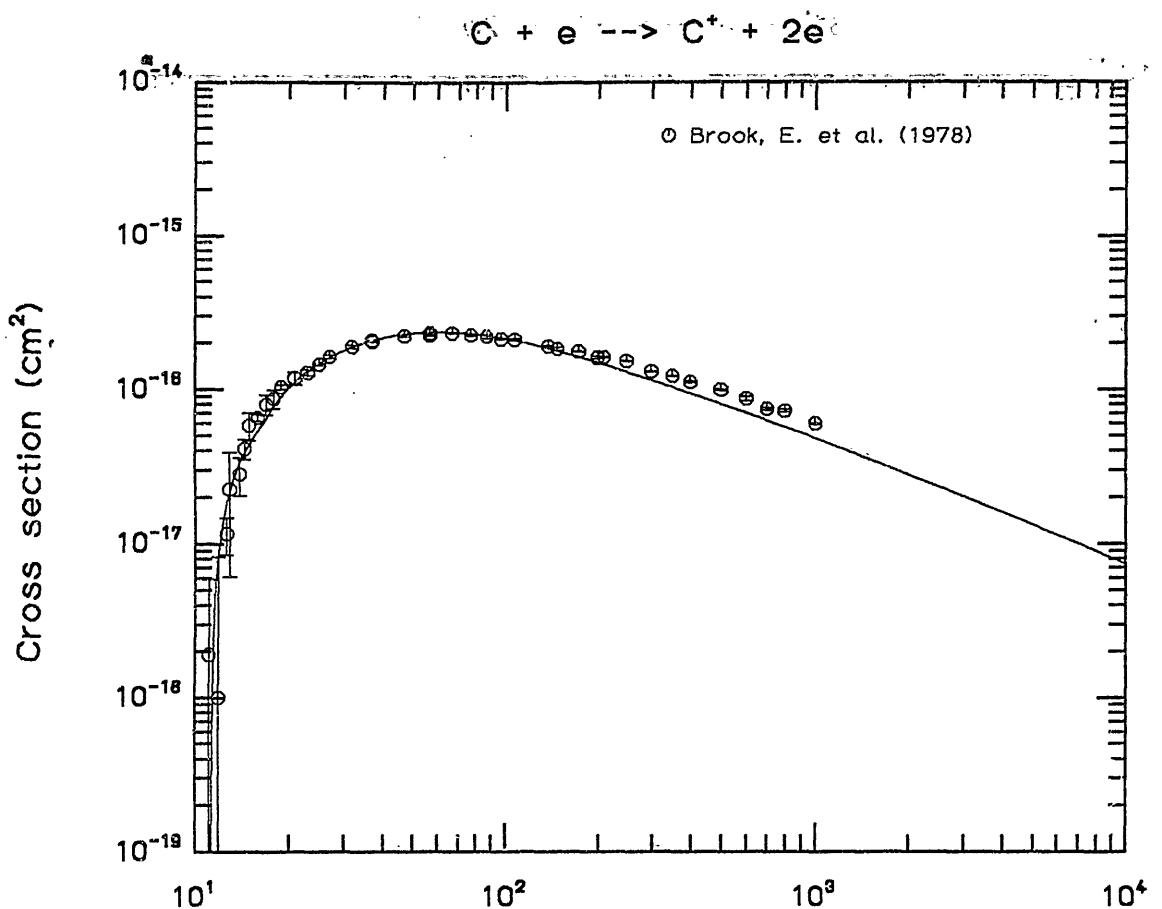
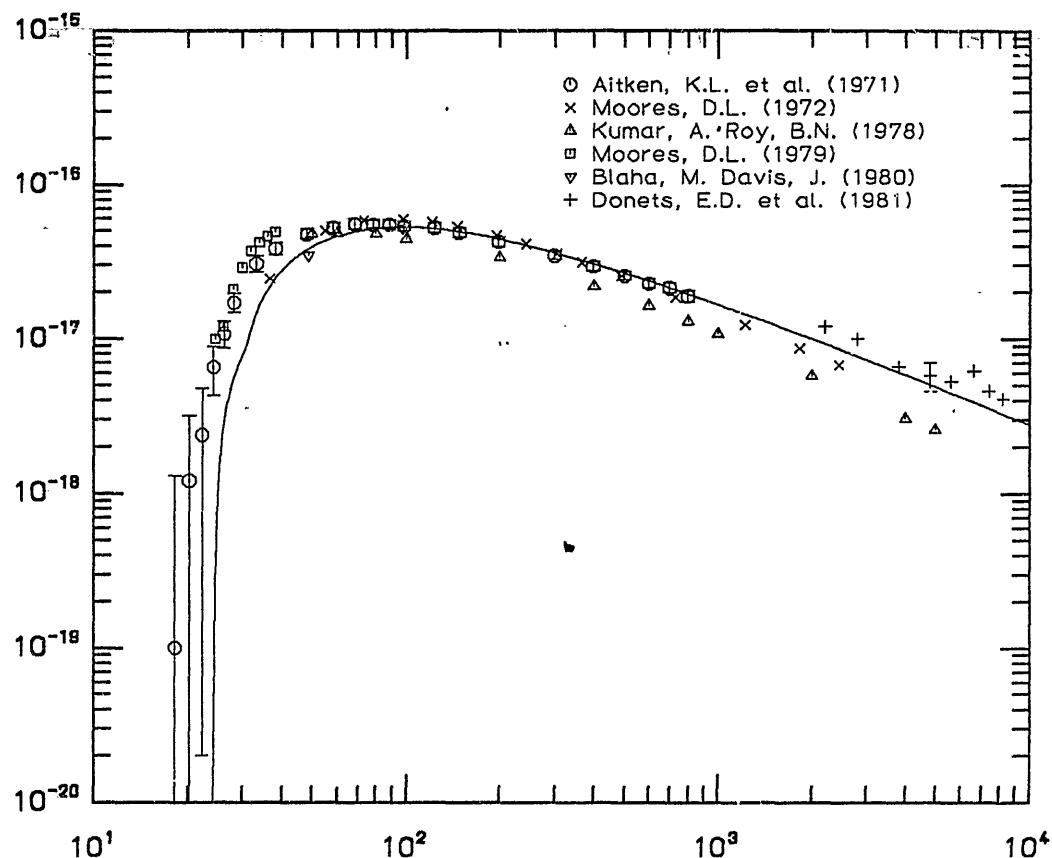
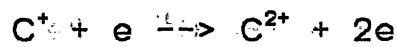


Fig. 22

Electron energy (eV)



Electron energy (eV)

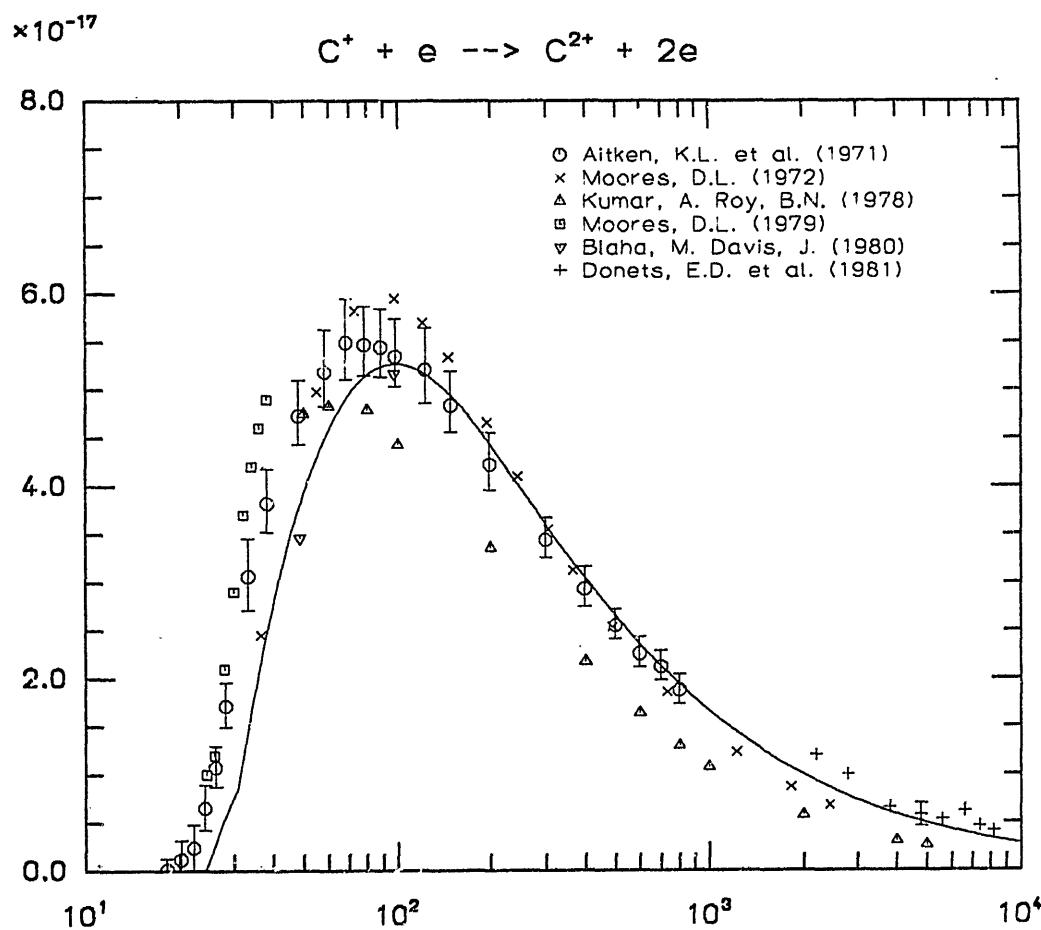


Fig. 23

Electron energy (eV)

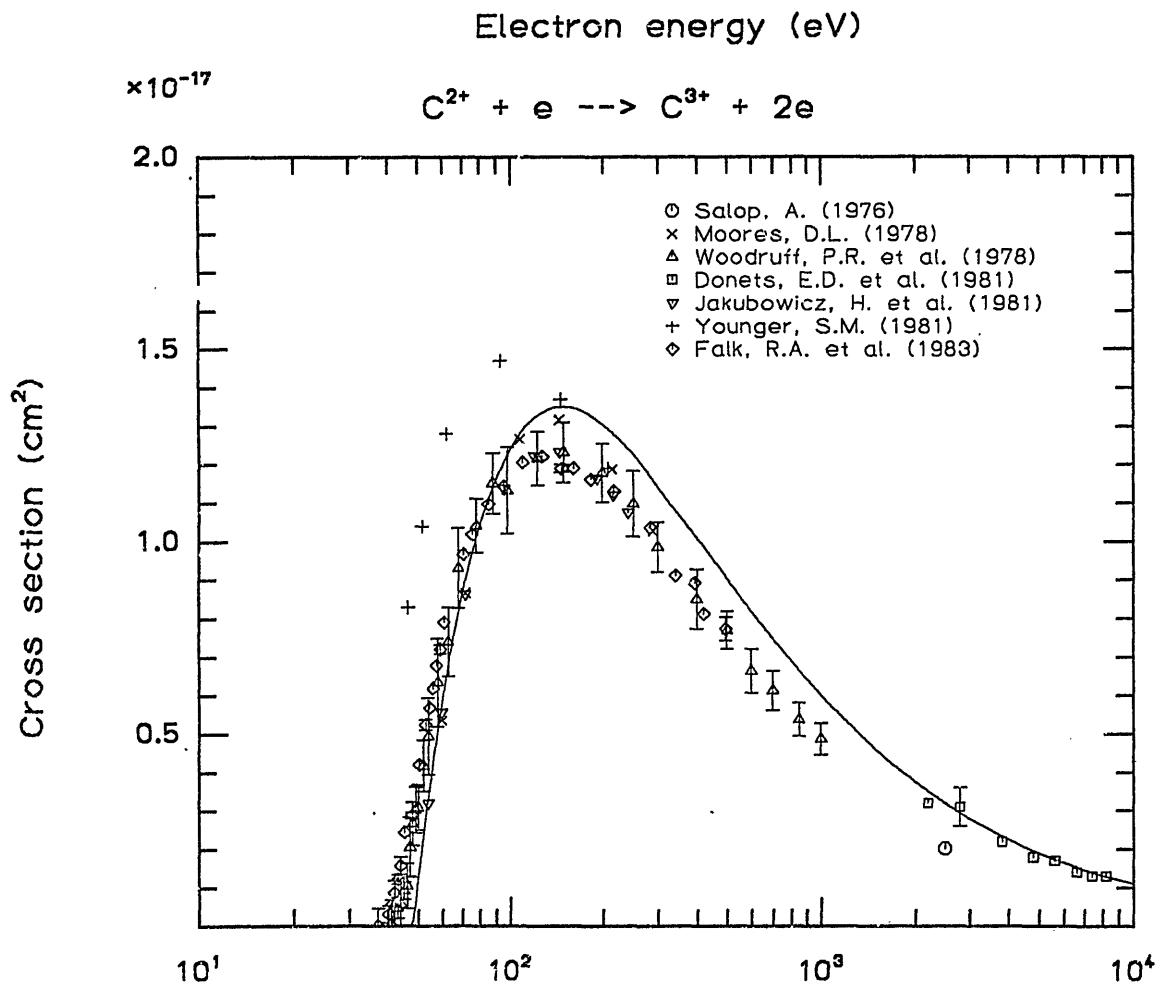
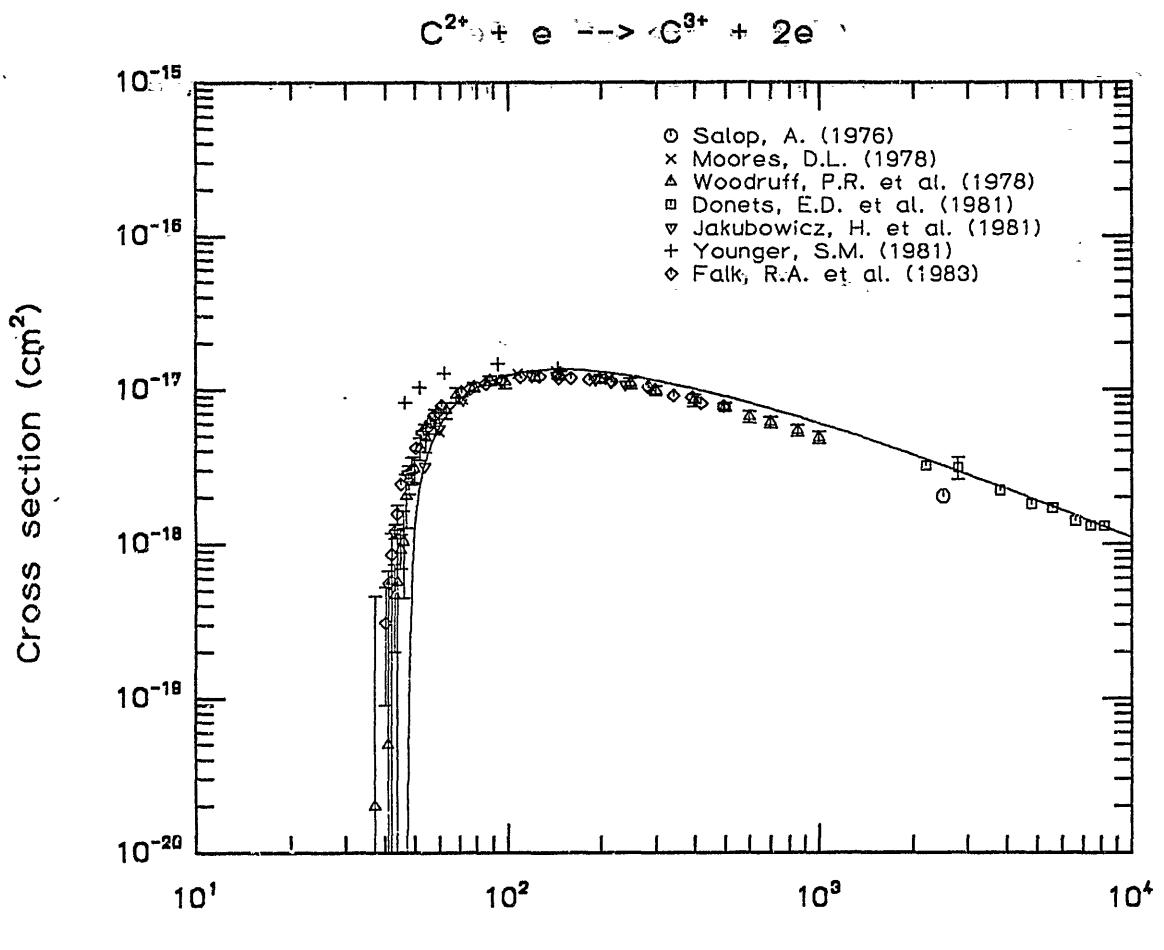


Fig. 24

Electron energy (eV)

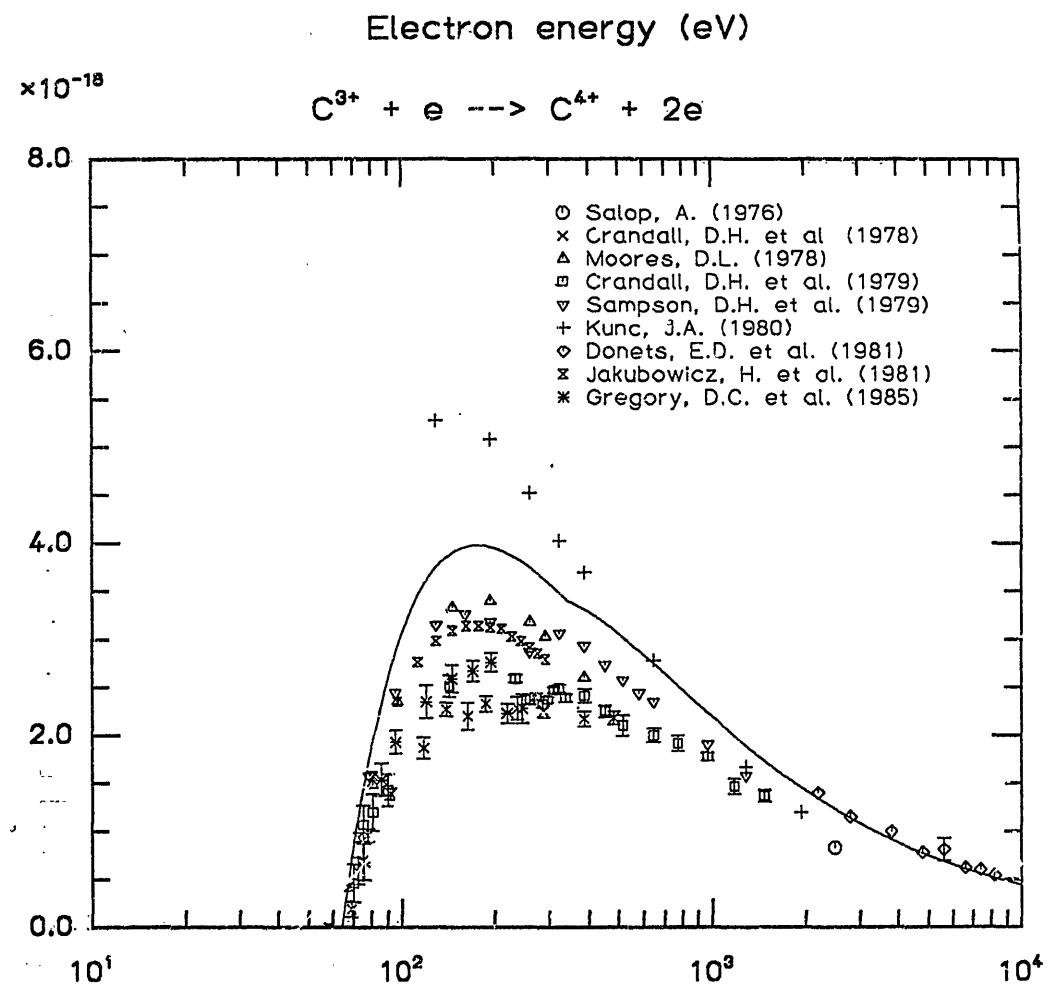
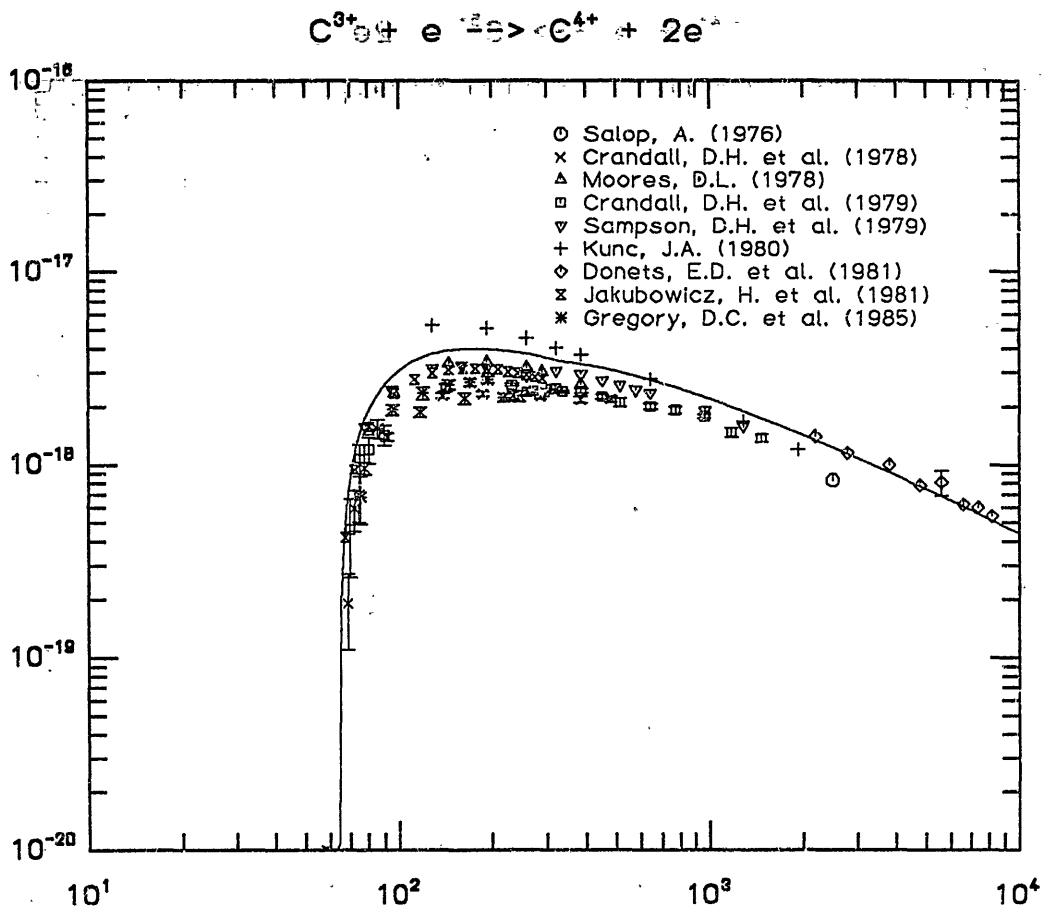


Fig. 25      Electron energy (eV)

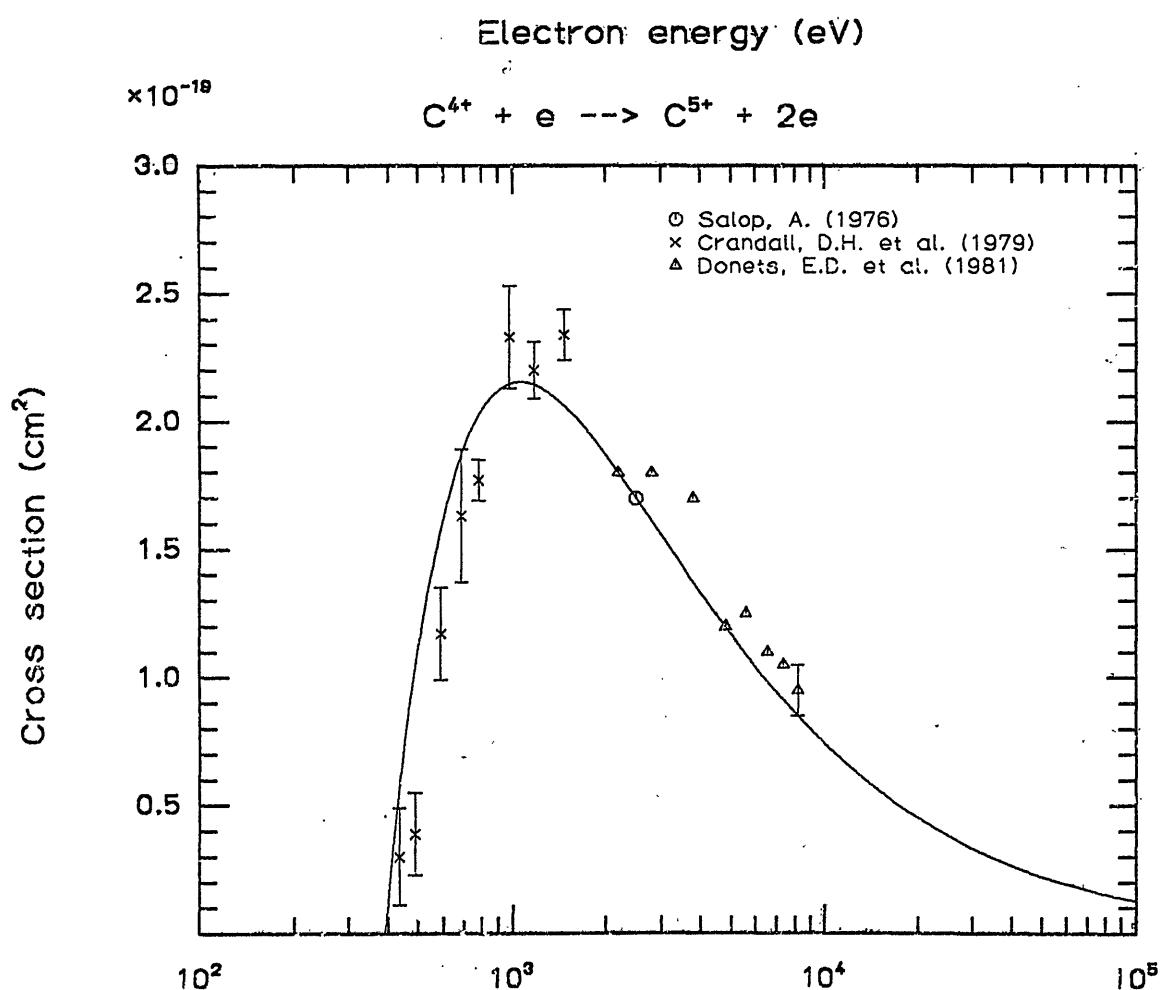
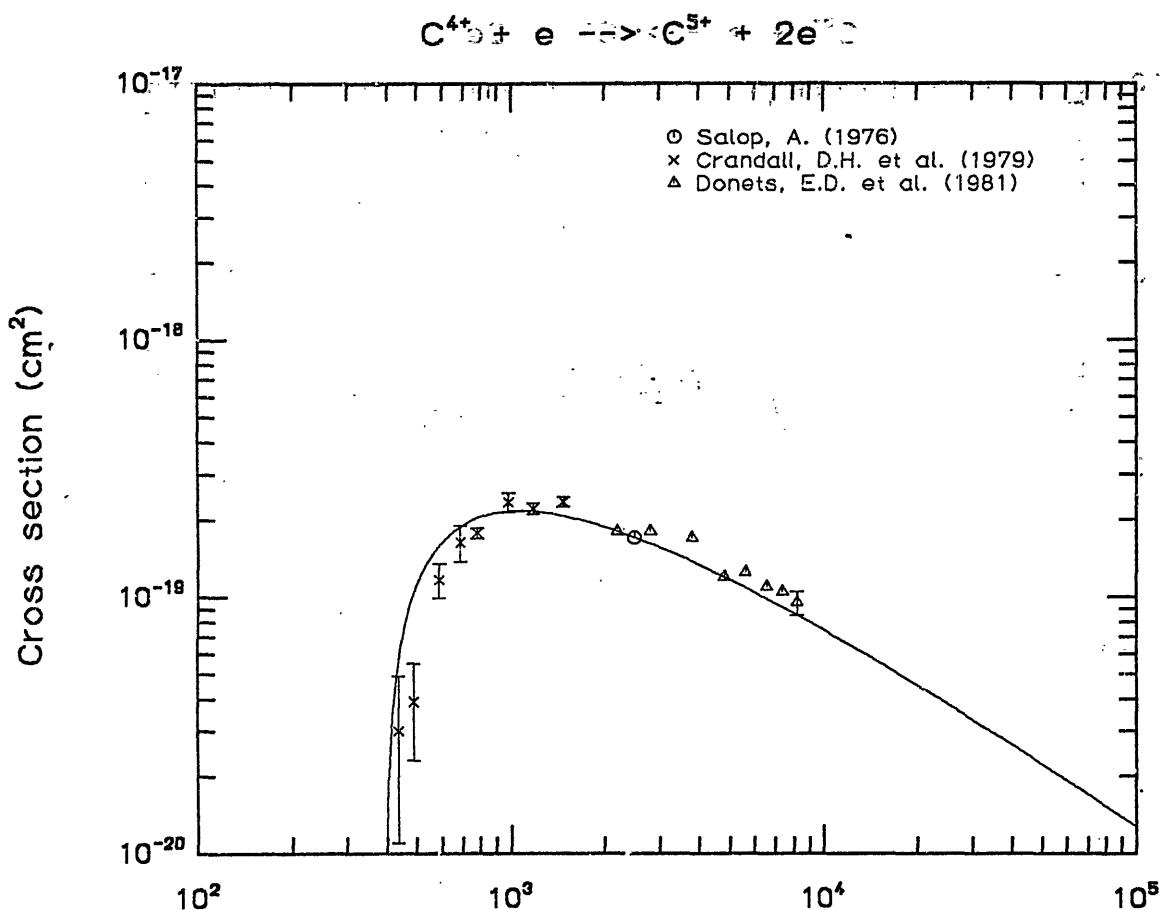


Fig. 26

Electron energy (eV)

CS (cm)

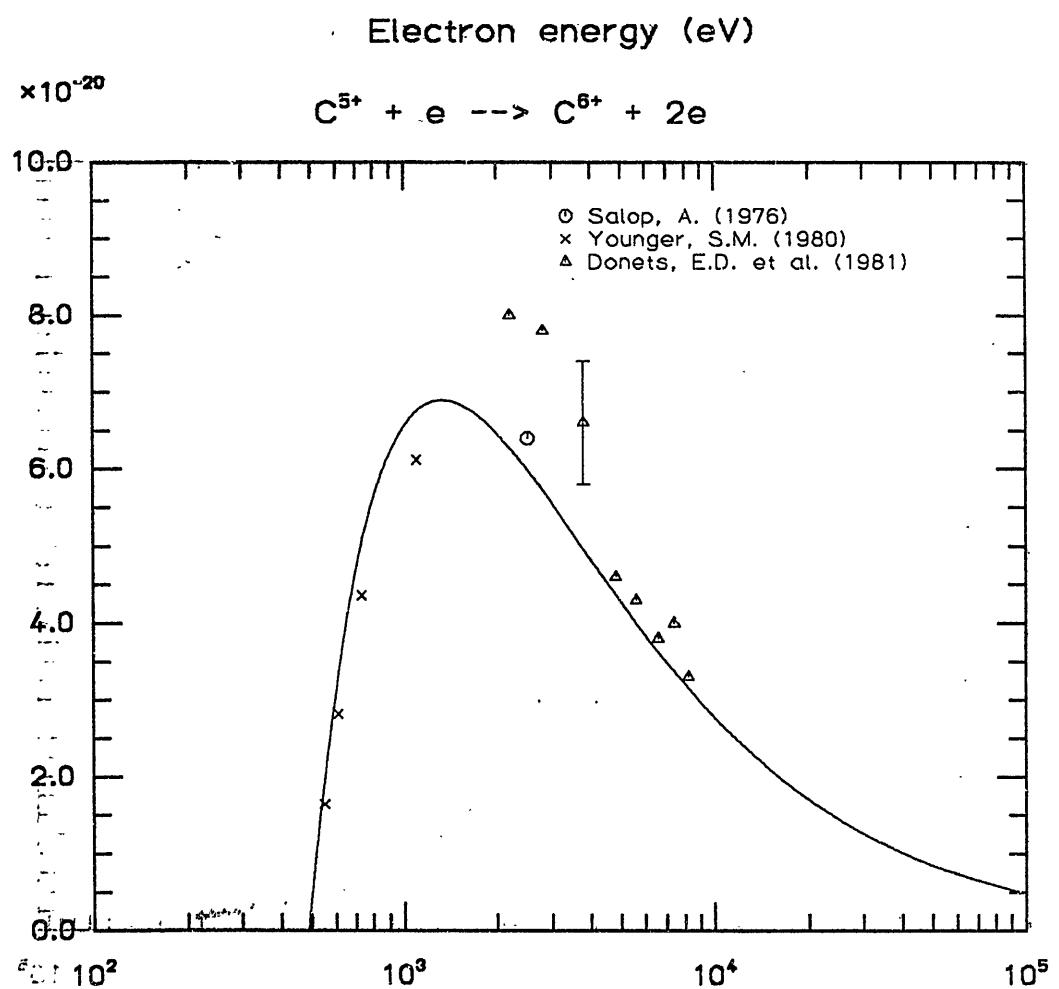
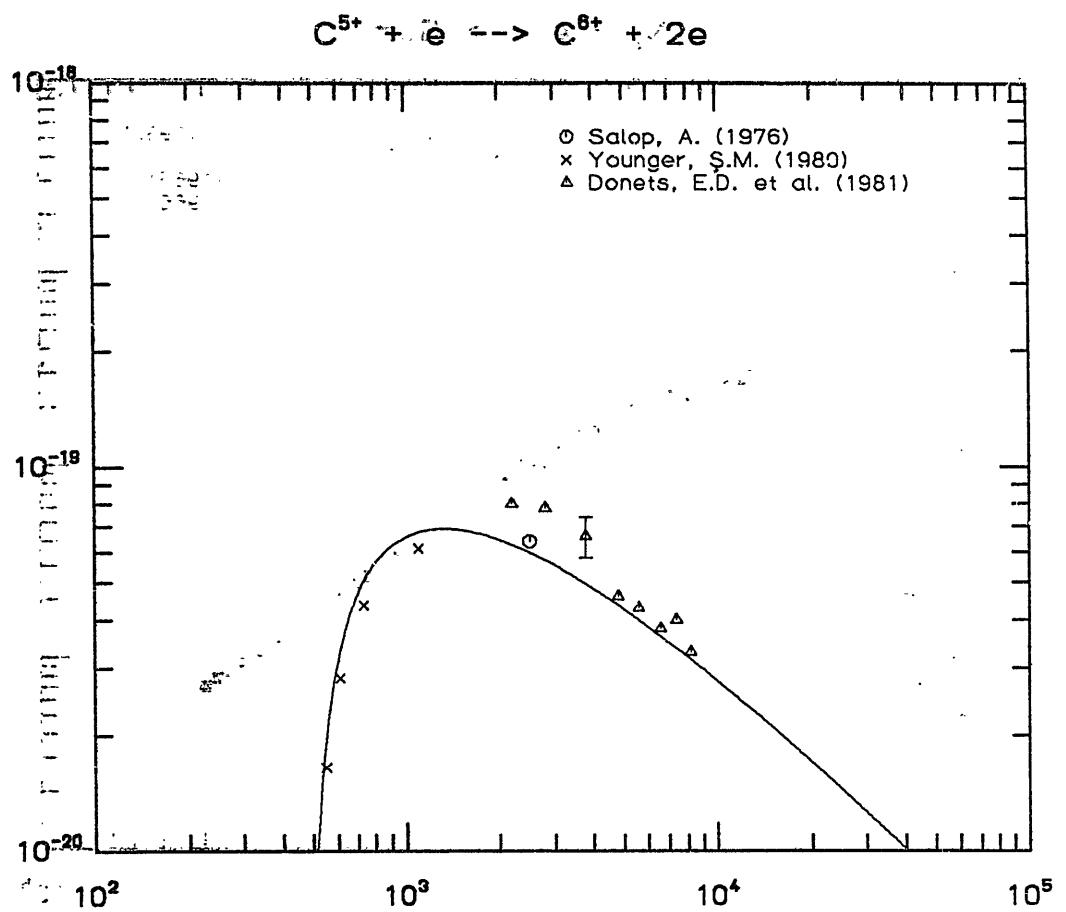


Fig. 27

Electron energy (eV)

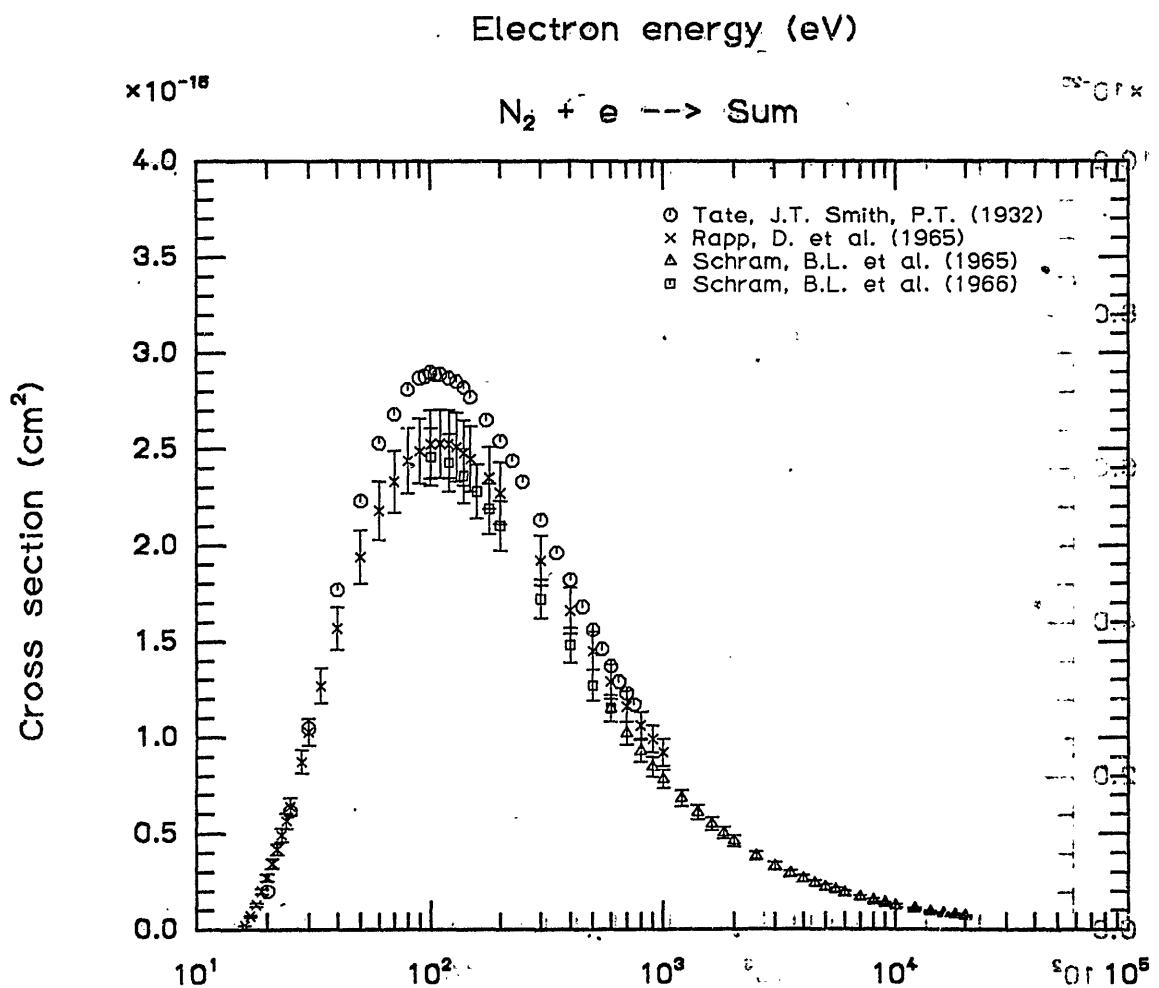
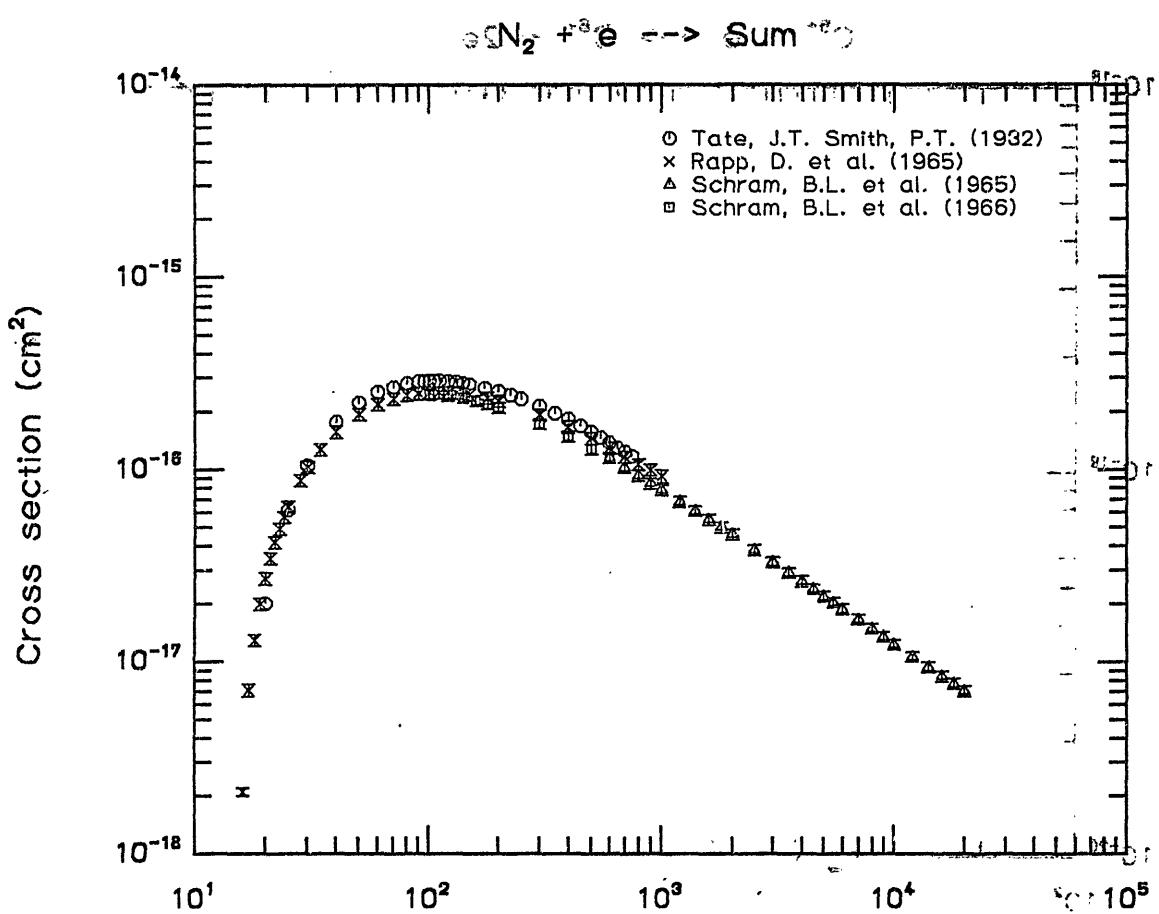


Fig. 28

(a) Electron energy (eV)

Fig. 28

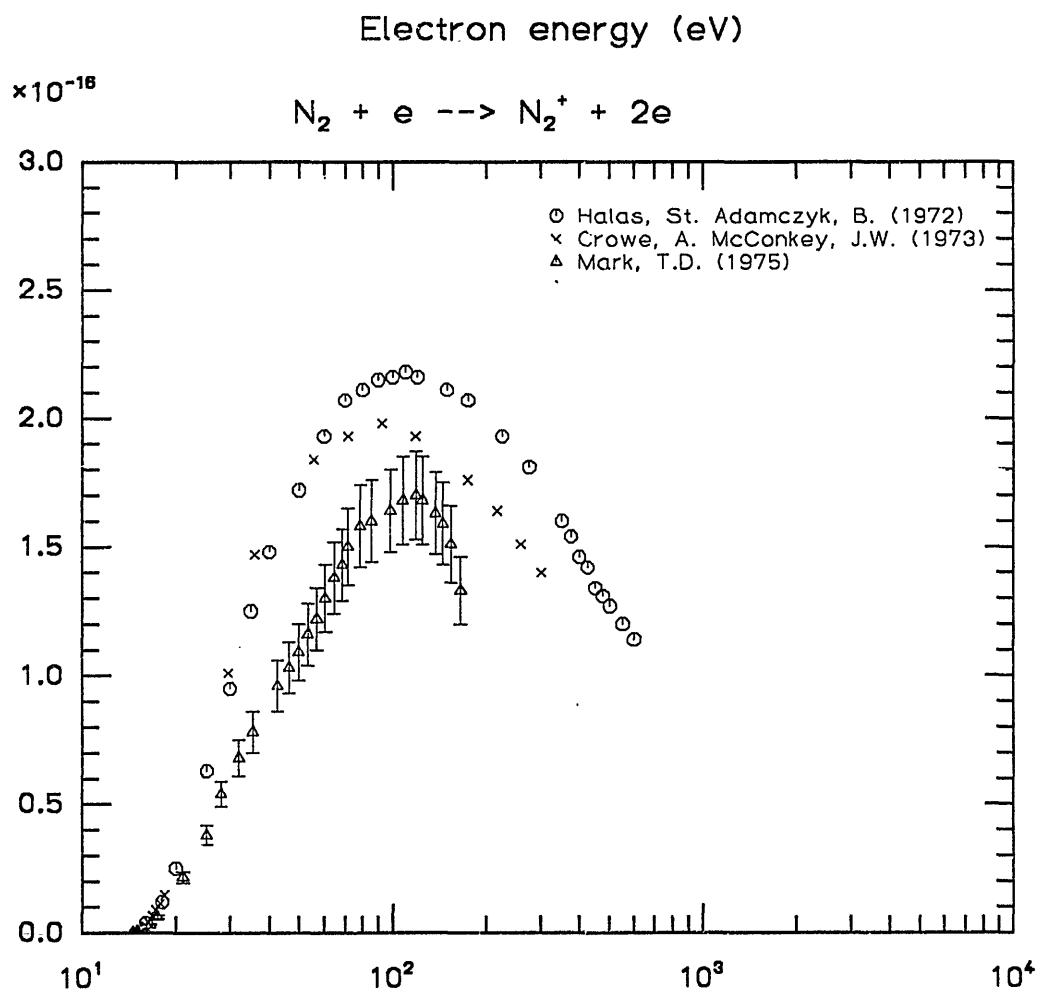
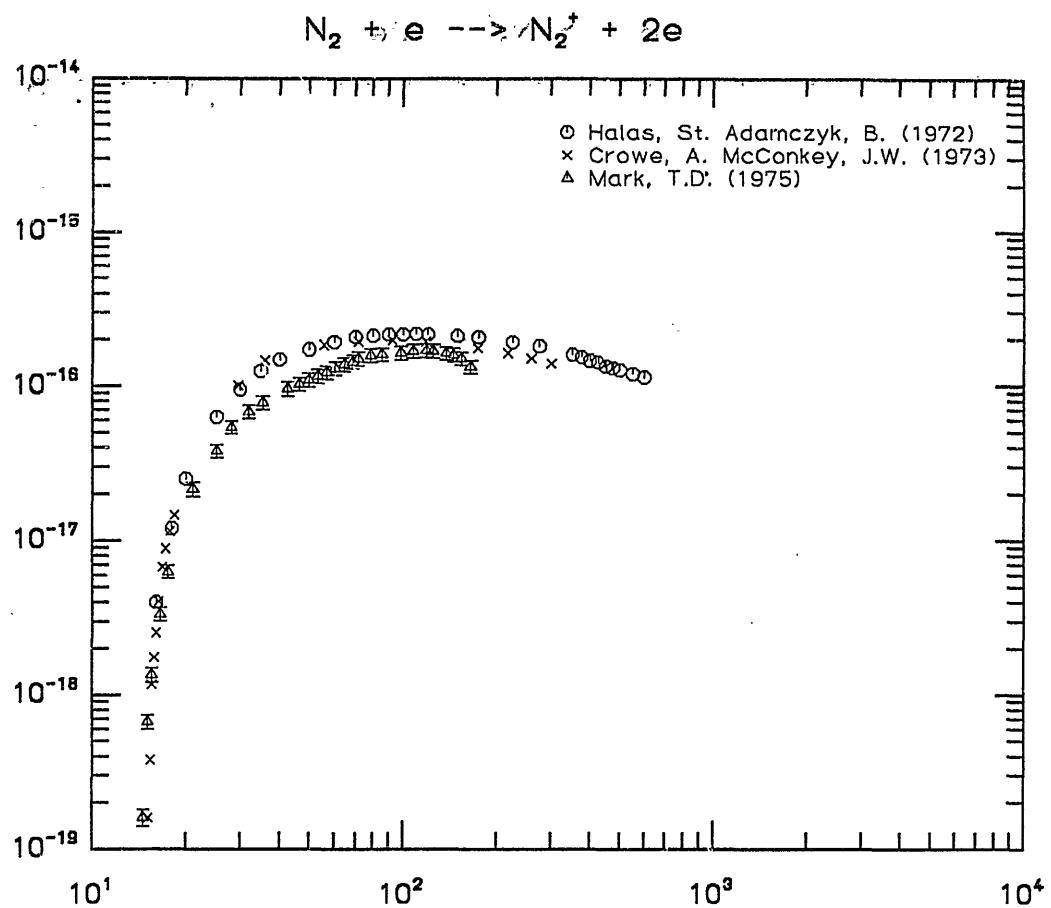


Fig. 29

Electron energy (eV)

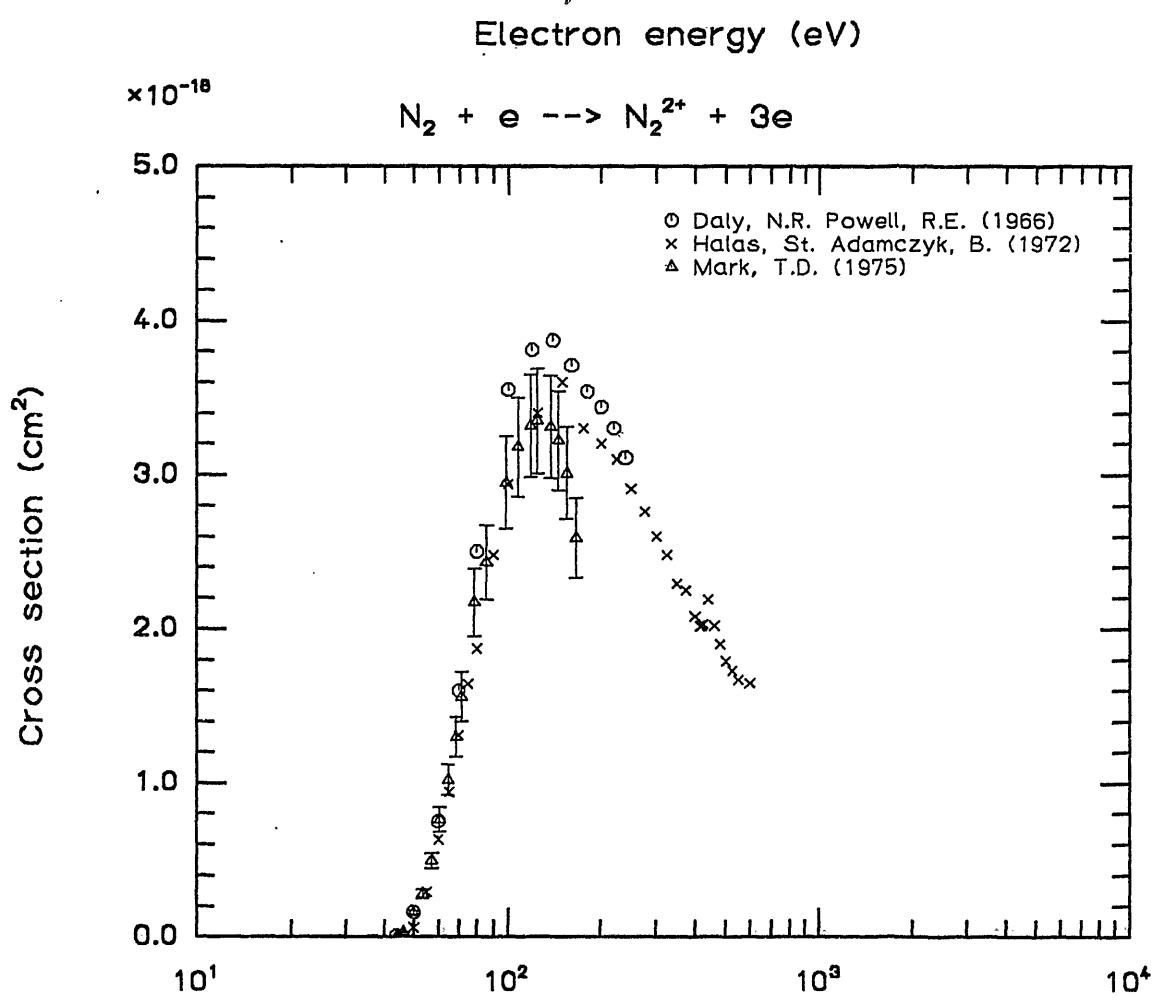
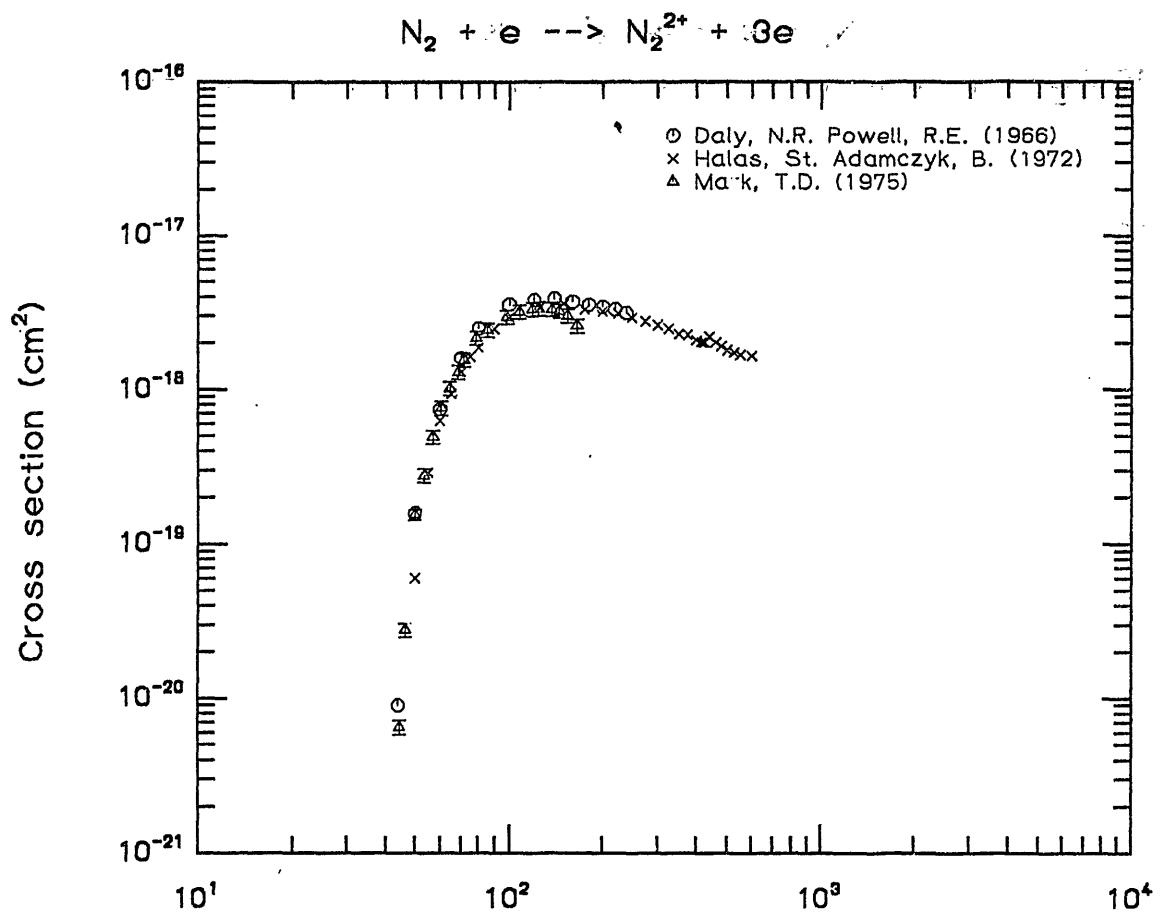
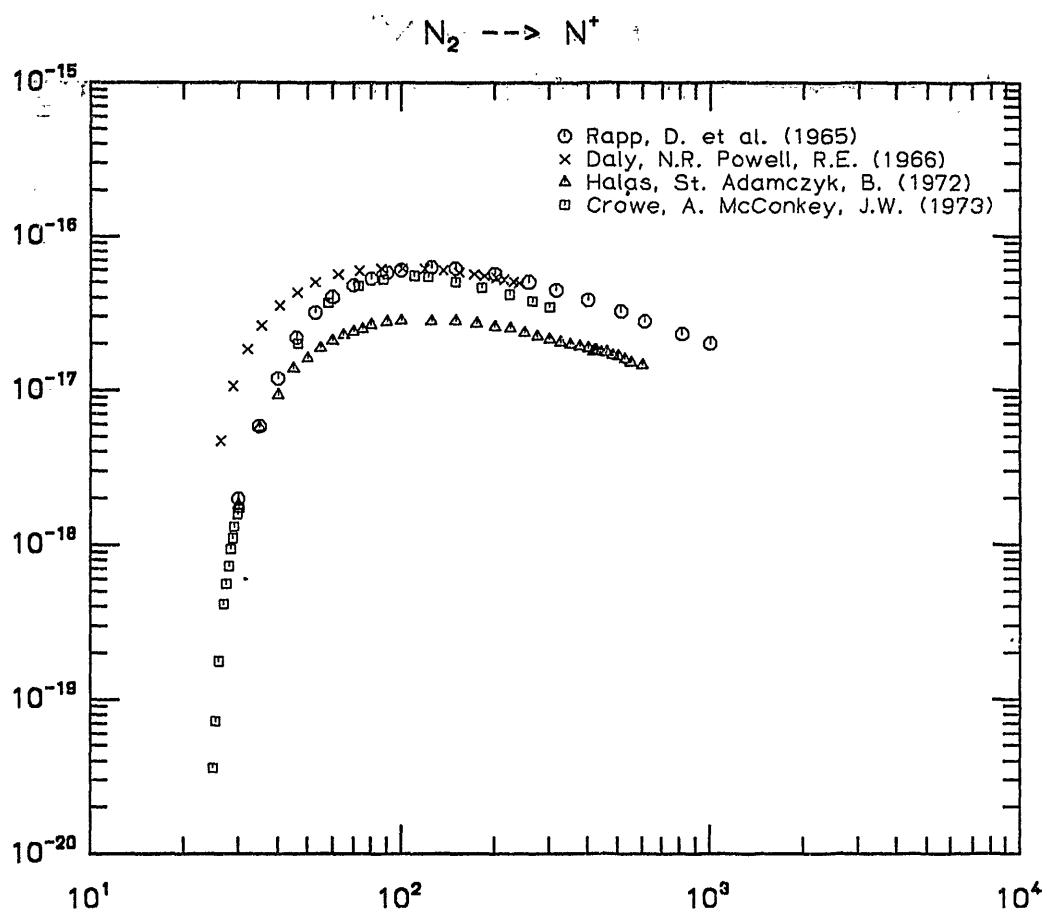


Fig. 30

Electron energy (eV)



Electron energy (eV)

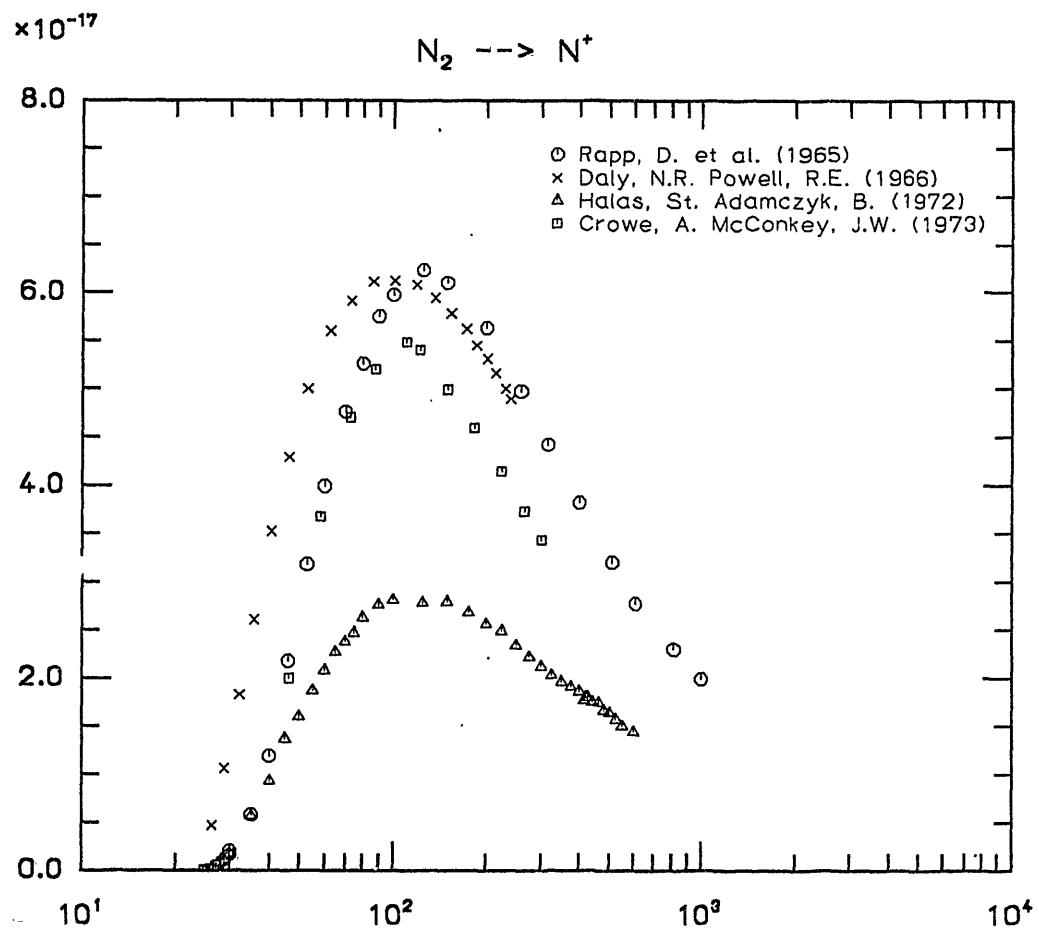


Fig. 31

Electron energy (eV)

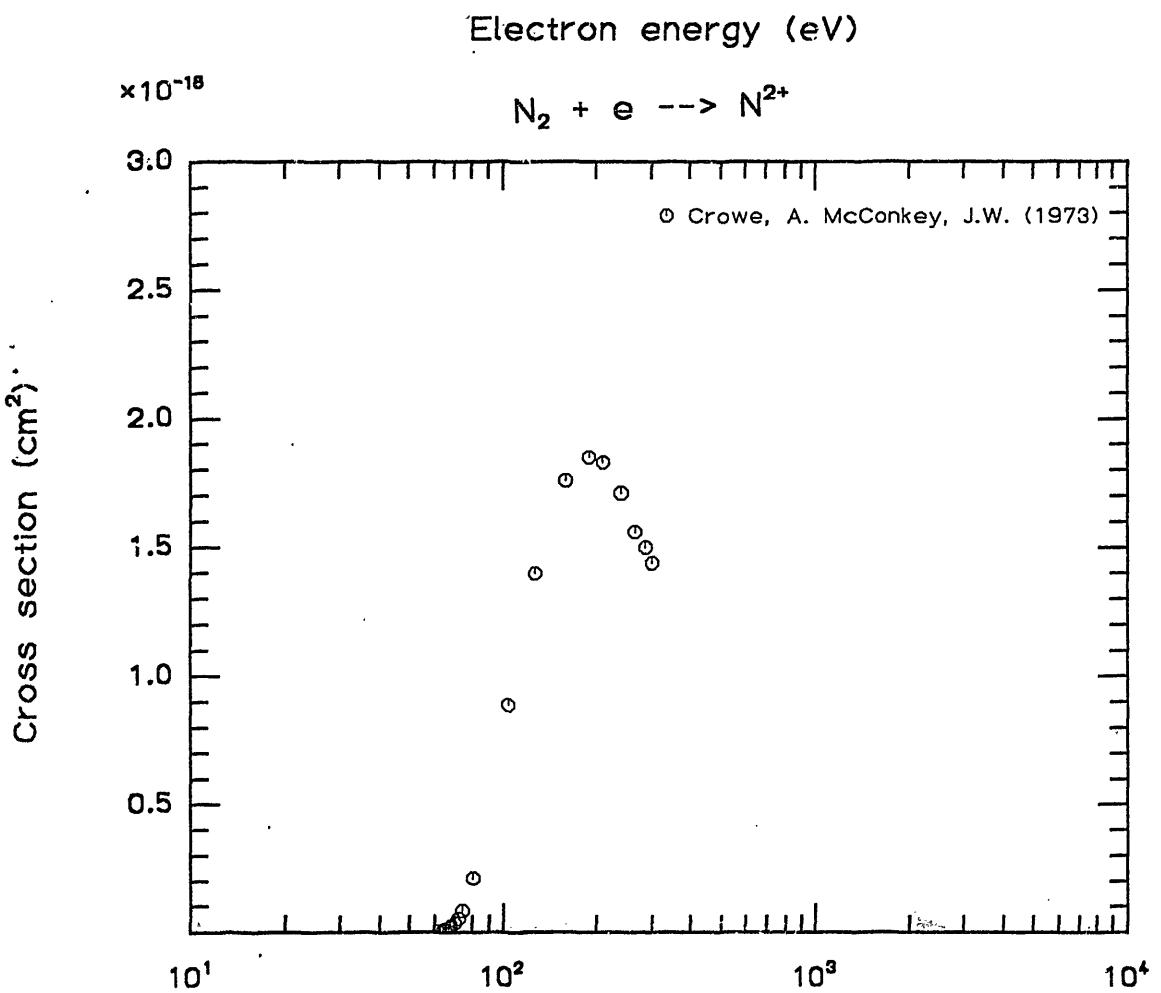
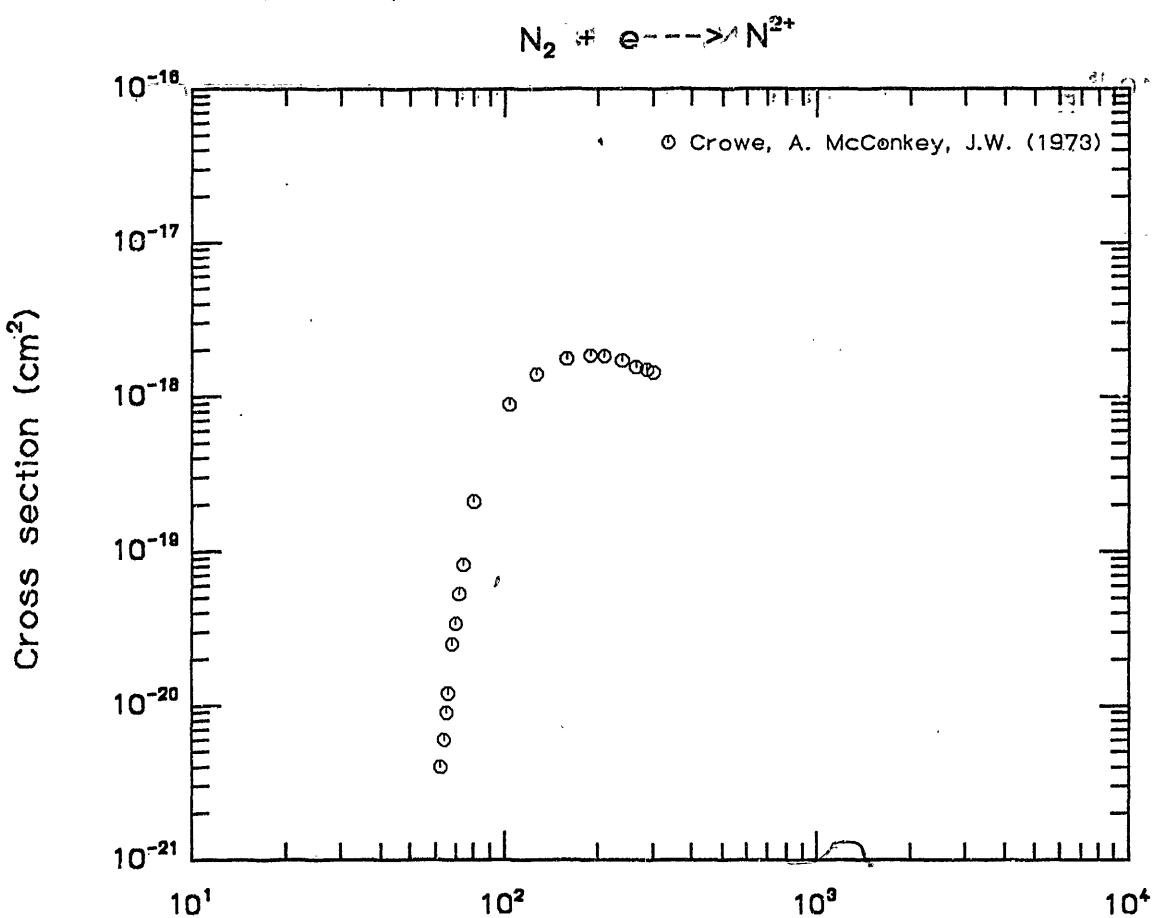
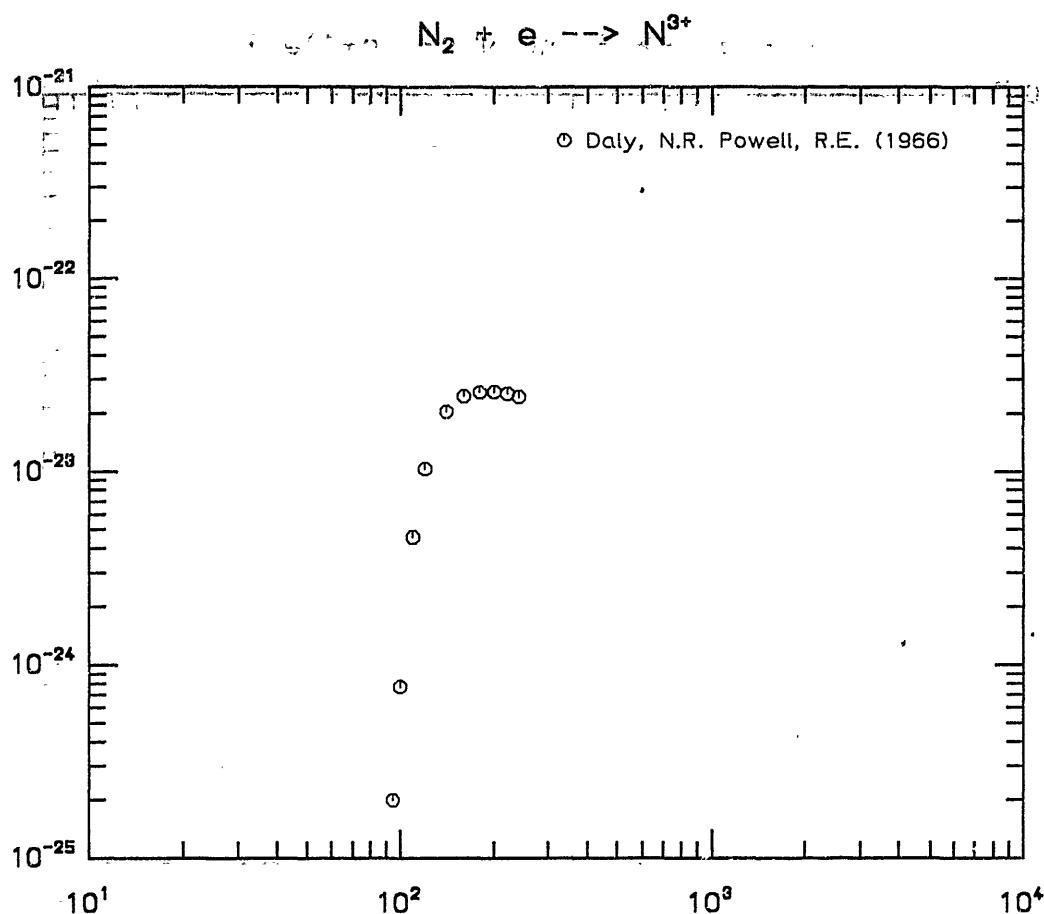


Fig. 32

Electron energy (eV)



Electron energy (eV)

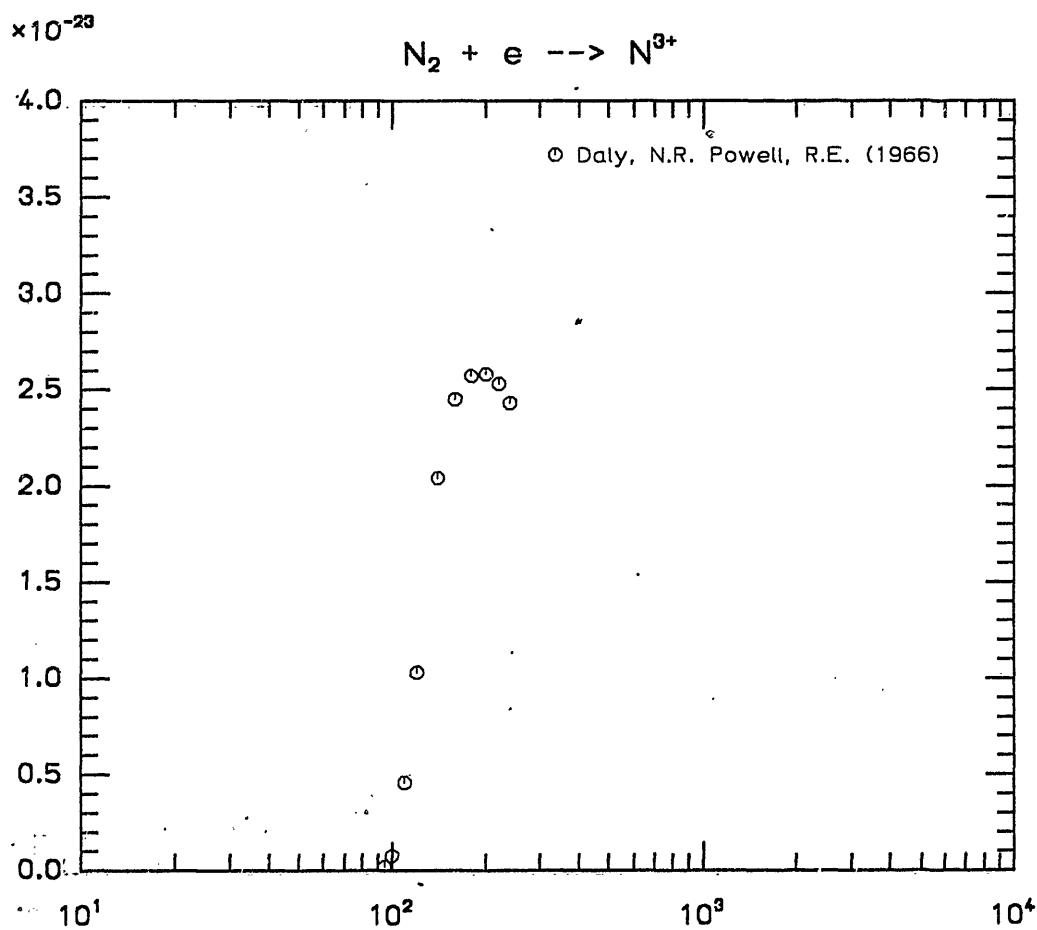


Fig. 33      Electron energy (eV)

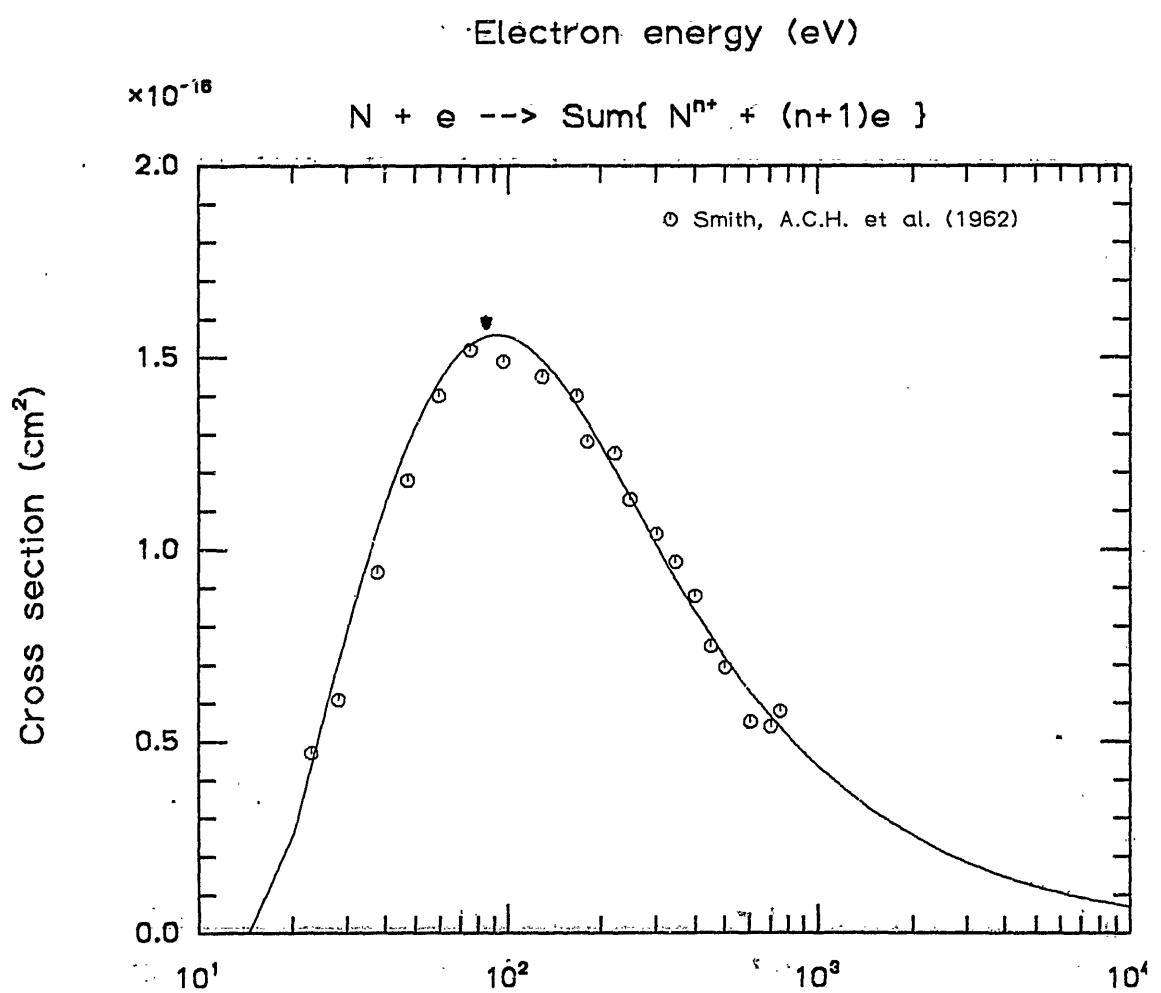
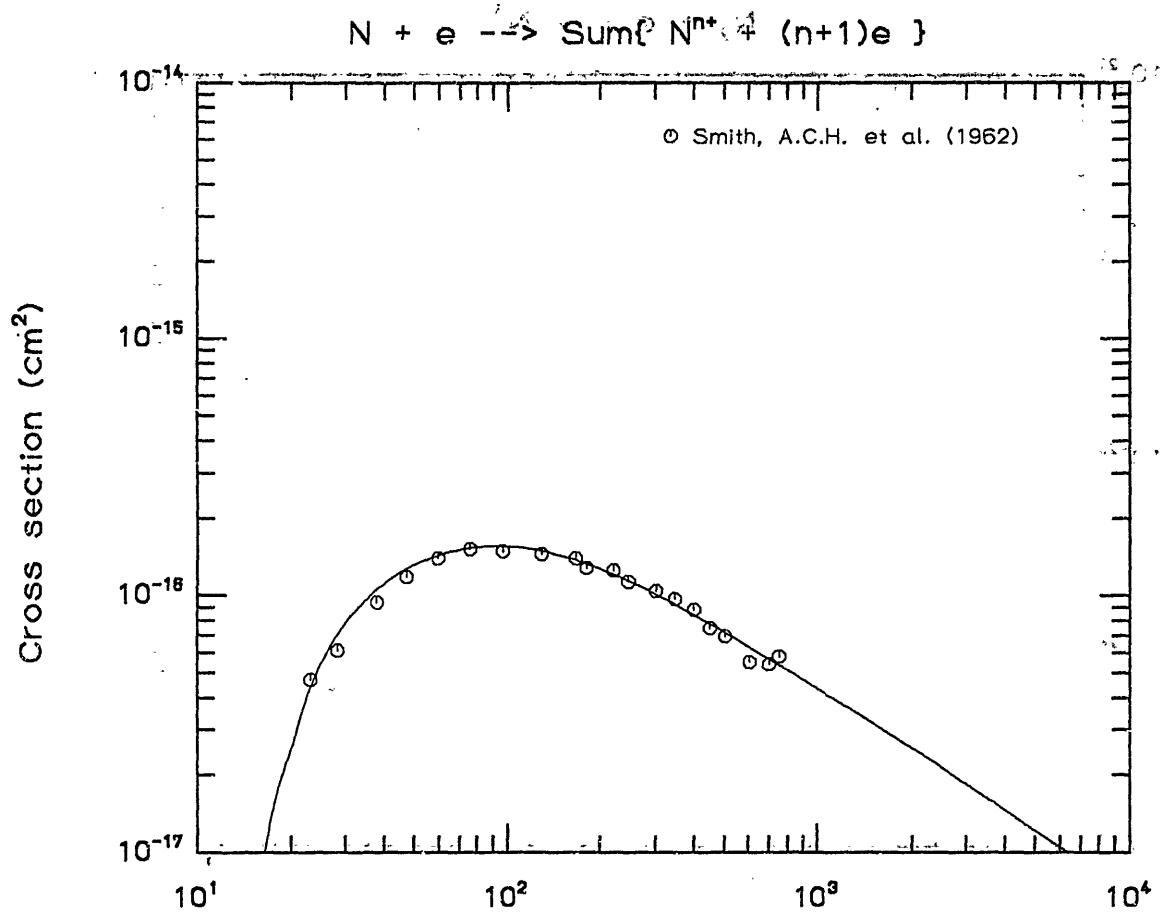


Fig. 34

Electron energy (eV)

EE PH

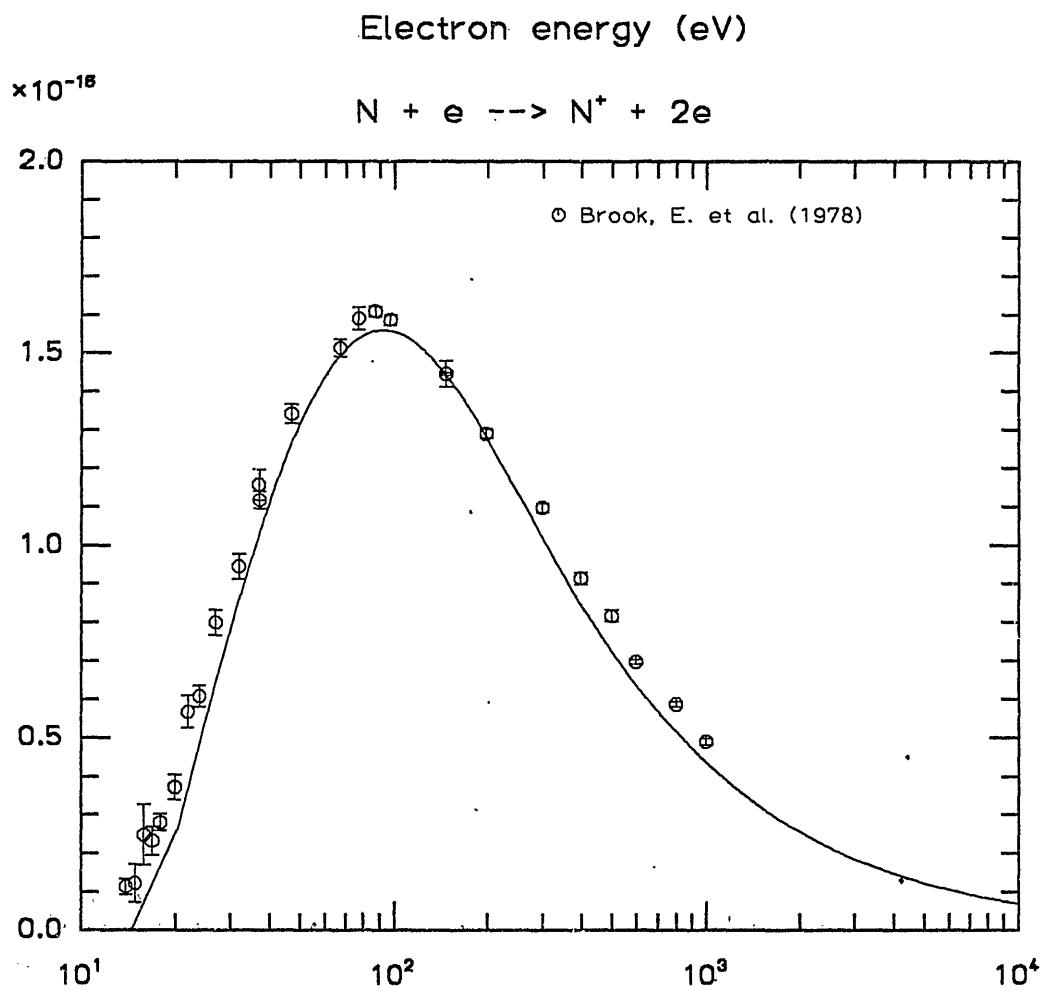
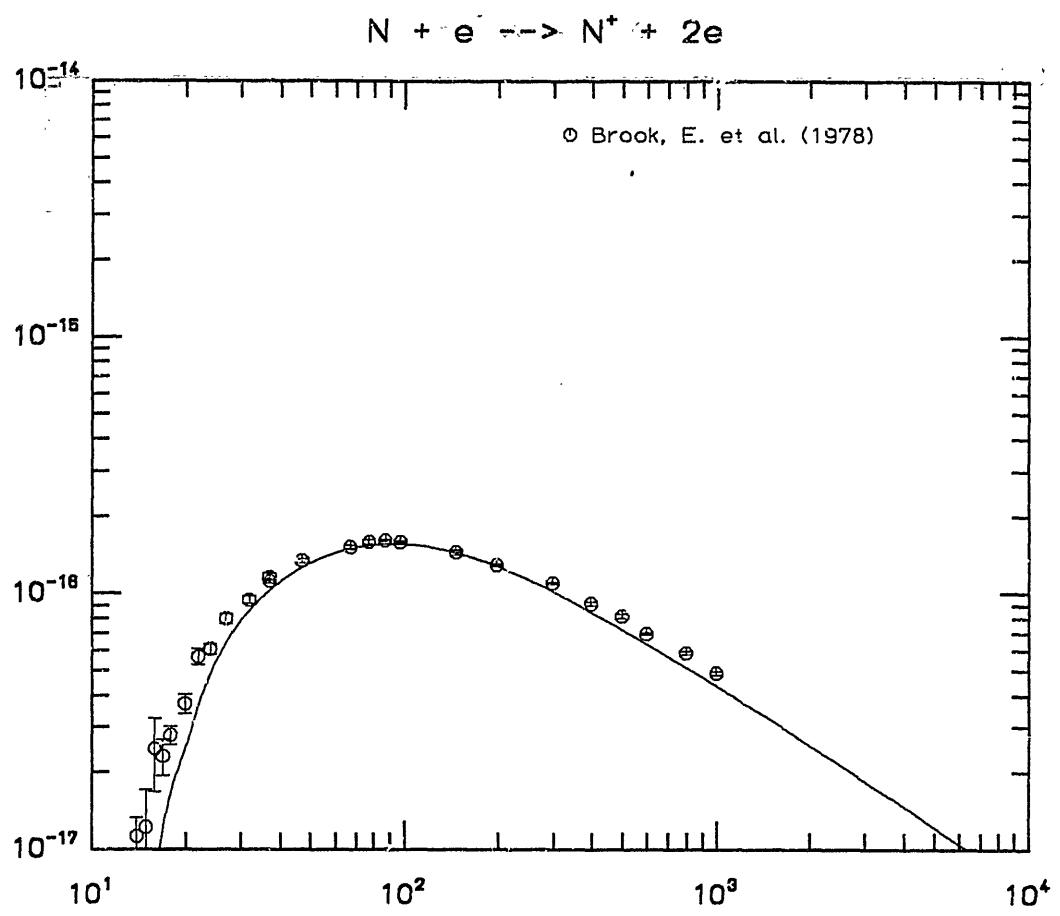


Fig. 35      Electron energy (eV)

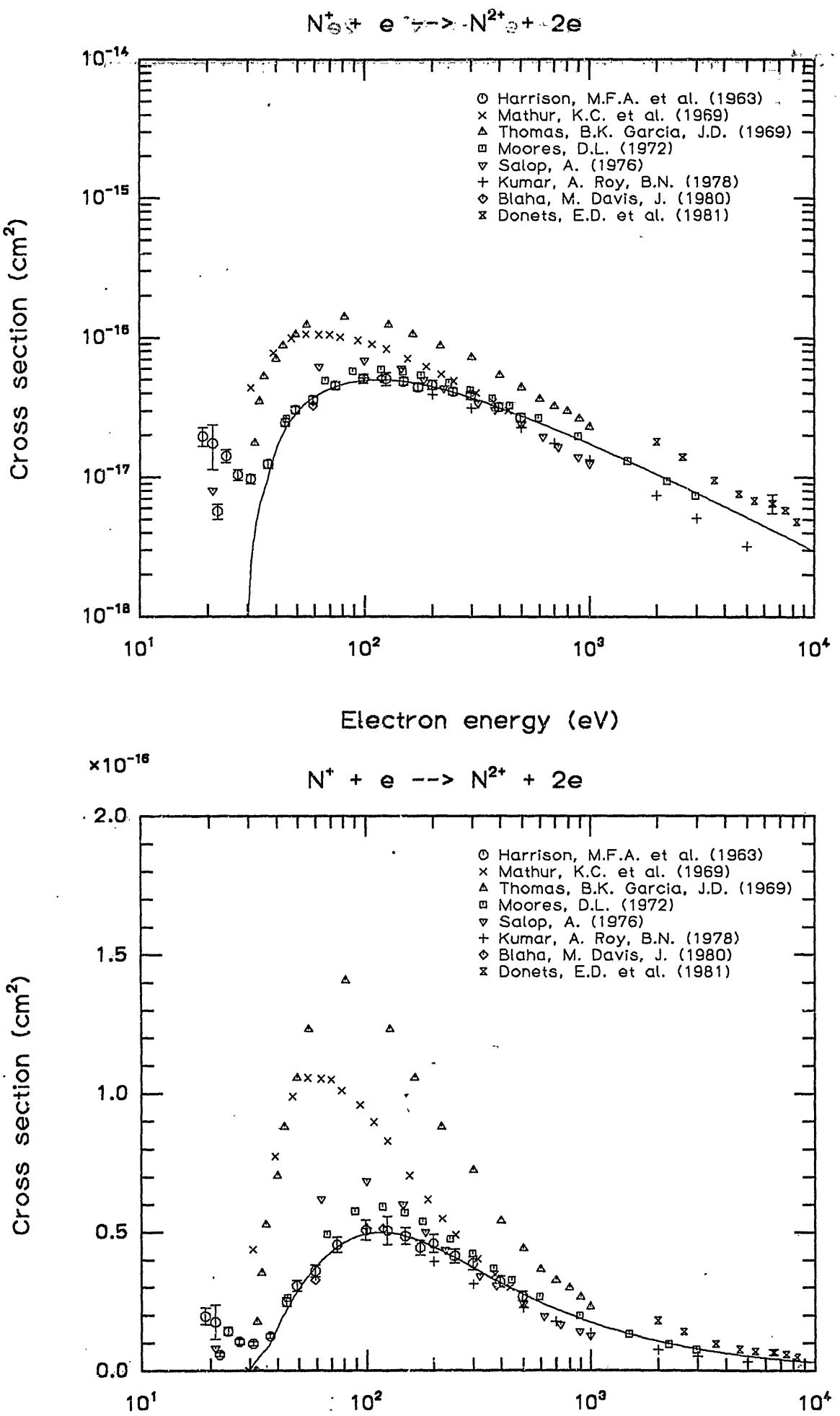
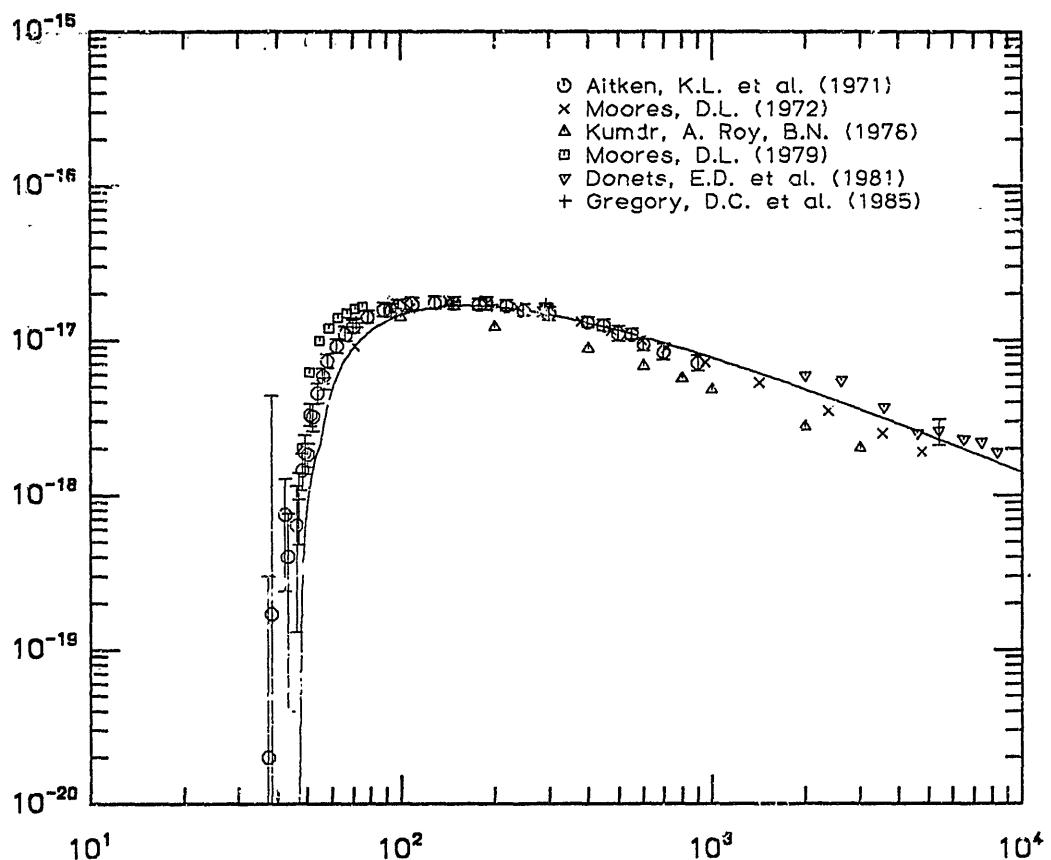
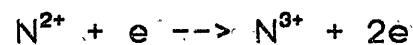


Fig. 36

Electron energy (eV)



Electron energy (eV)

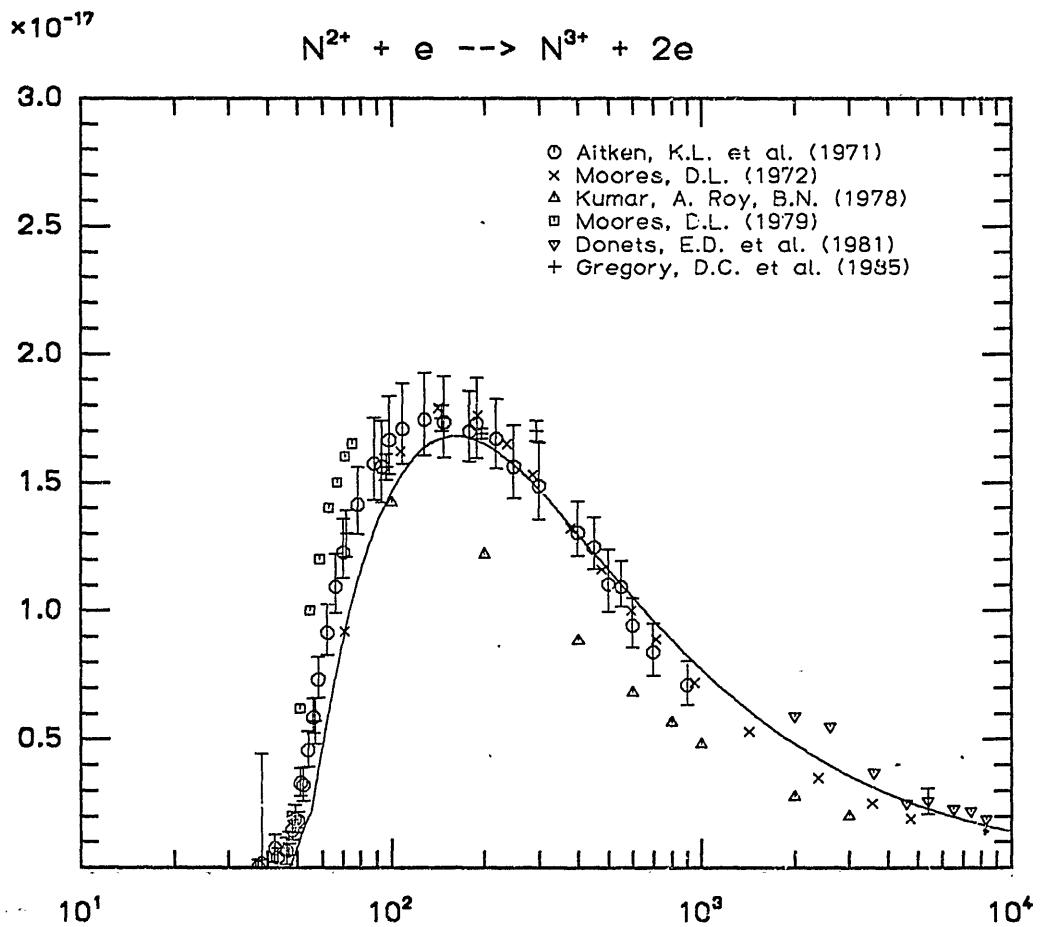


Fig. 37

Electron energy (eV)

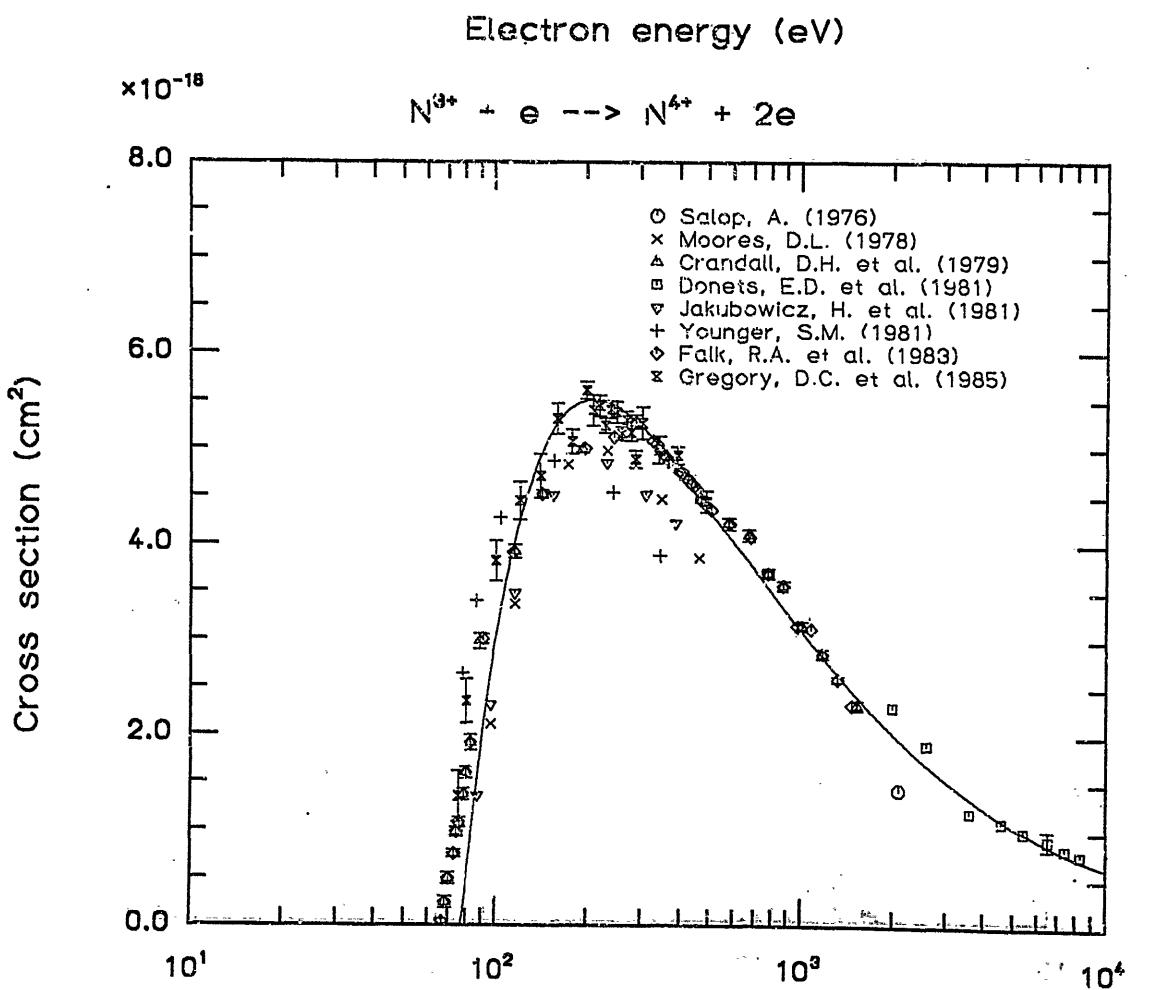
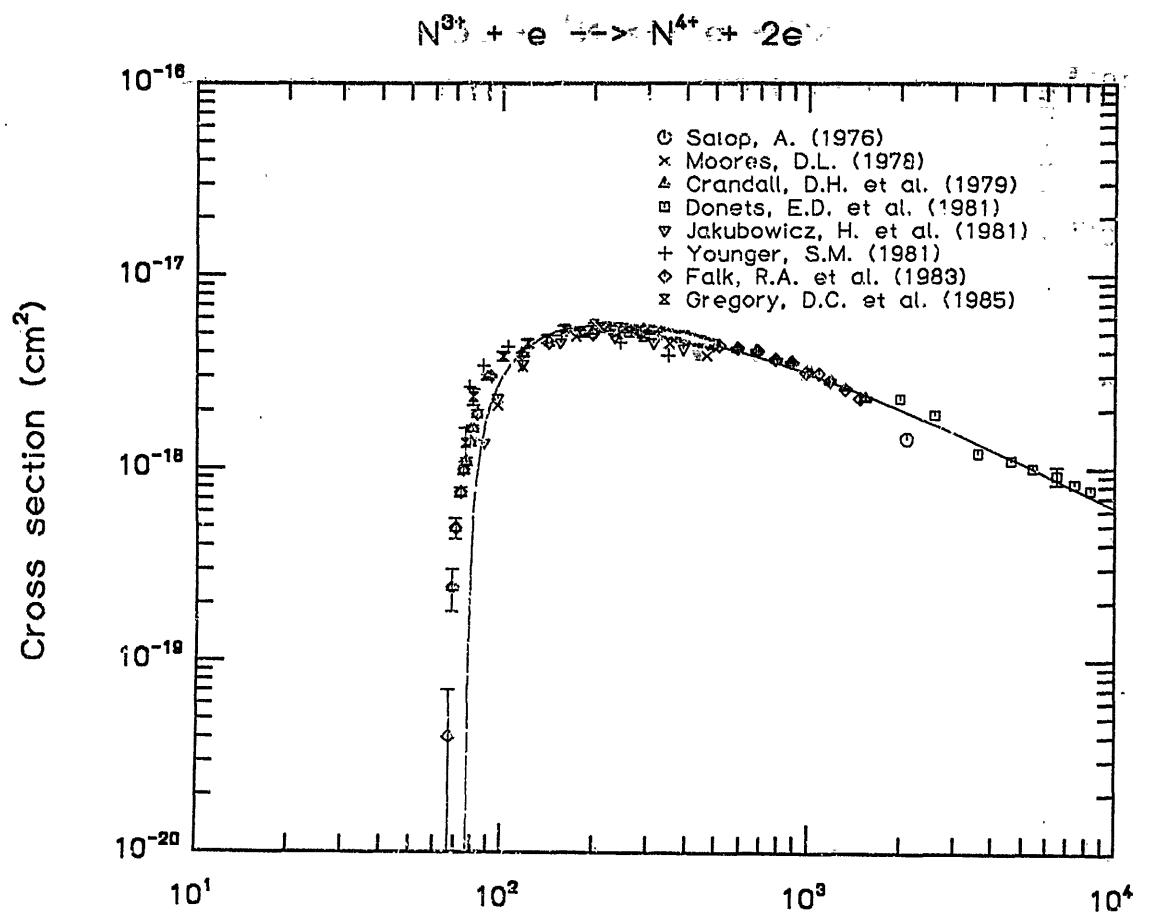


Fig. 38

Electron energy (eV)

Fig. 38

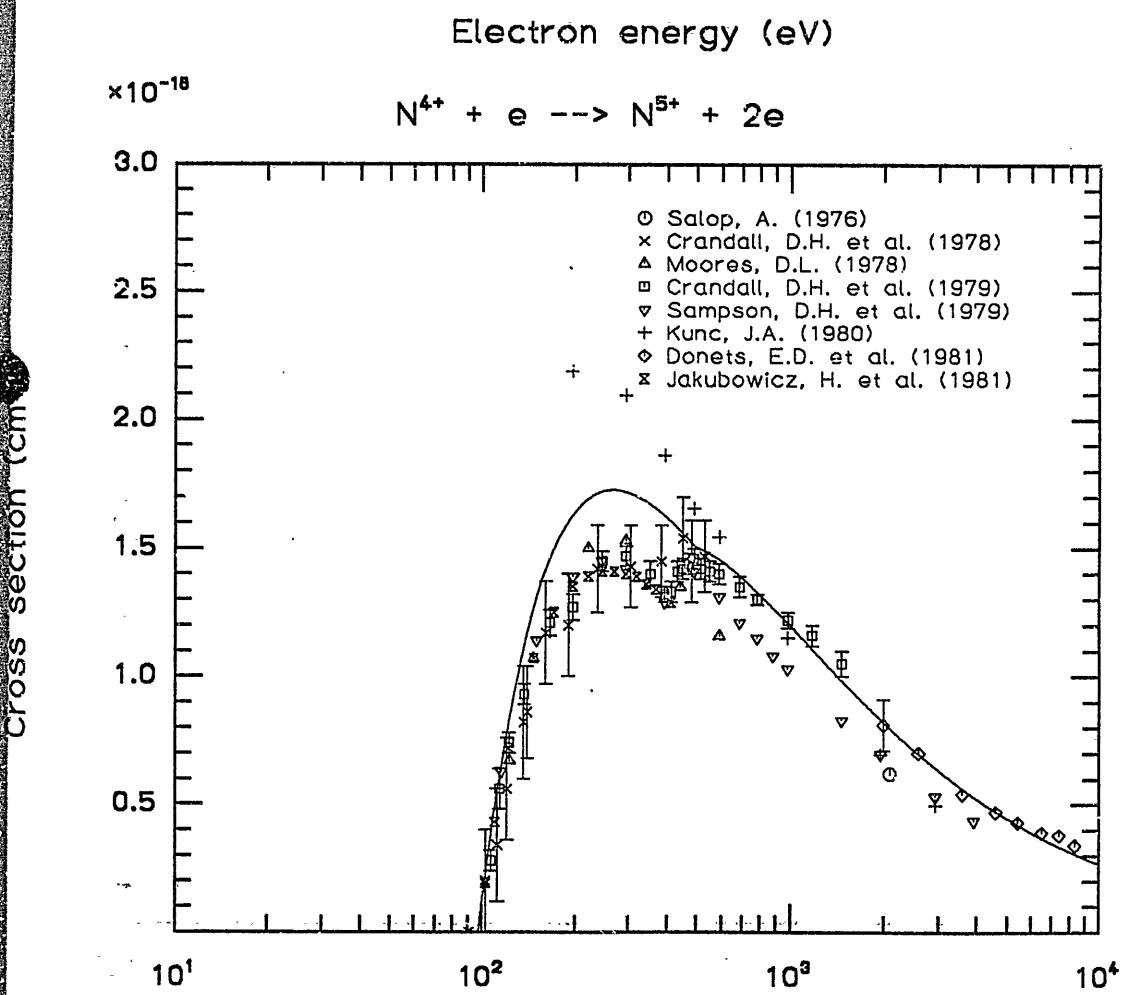
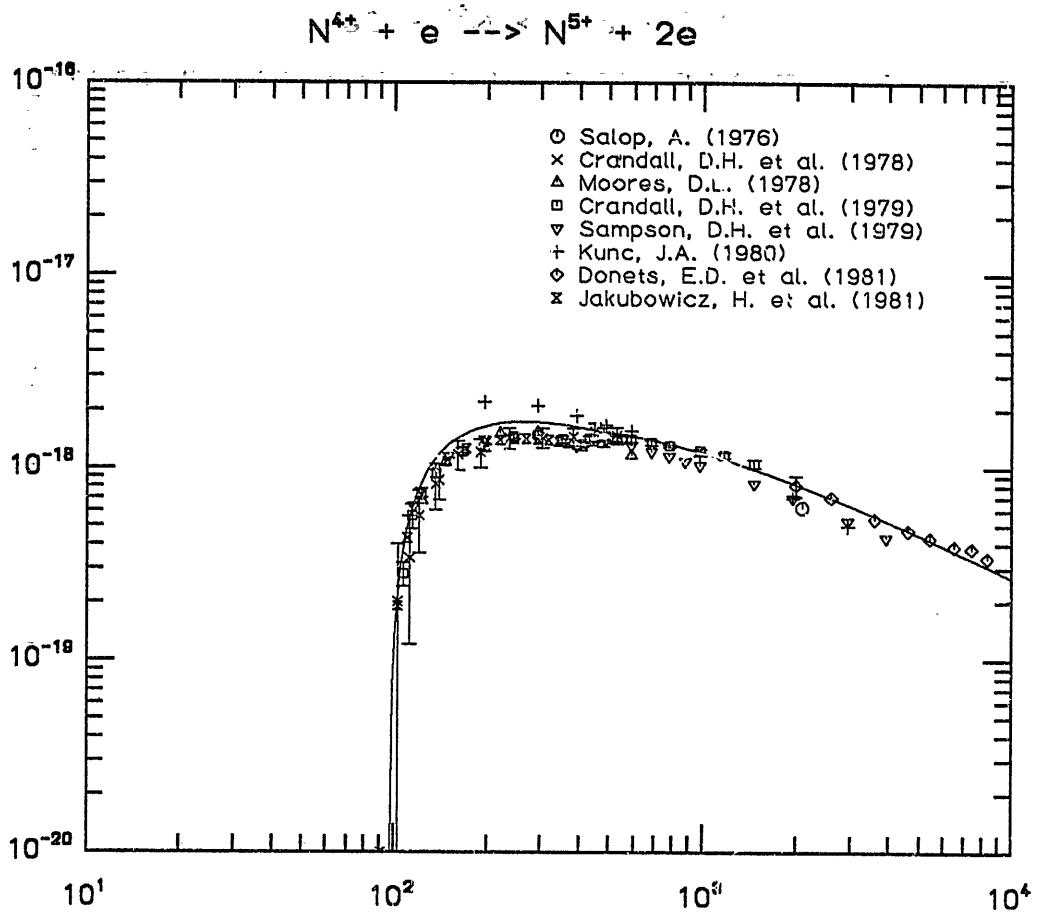


Fig. 39

Electron energy (eV)

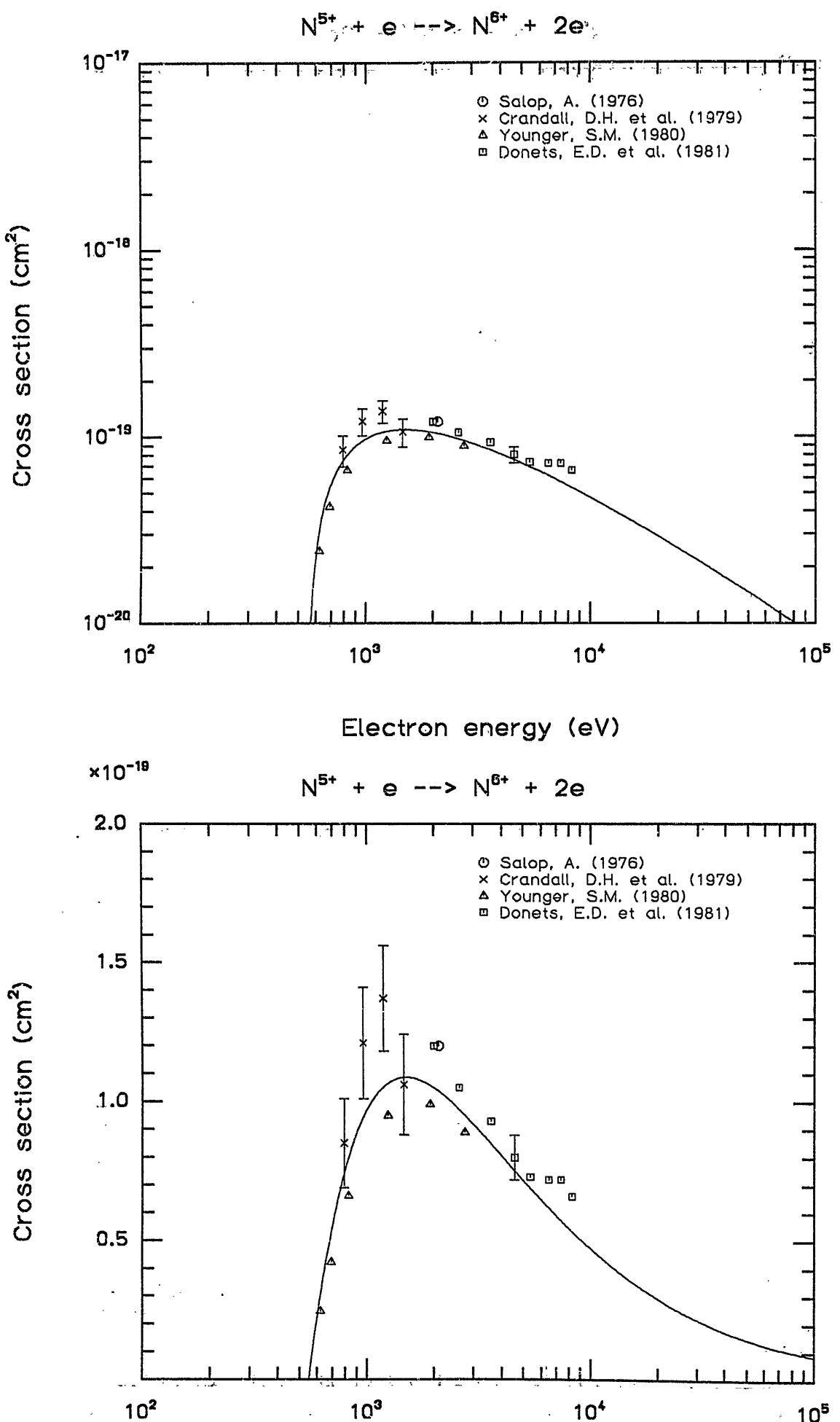
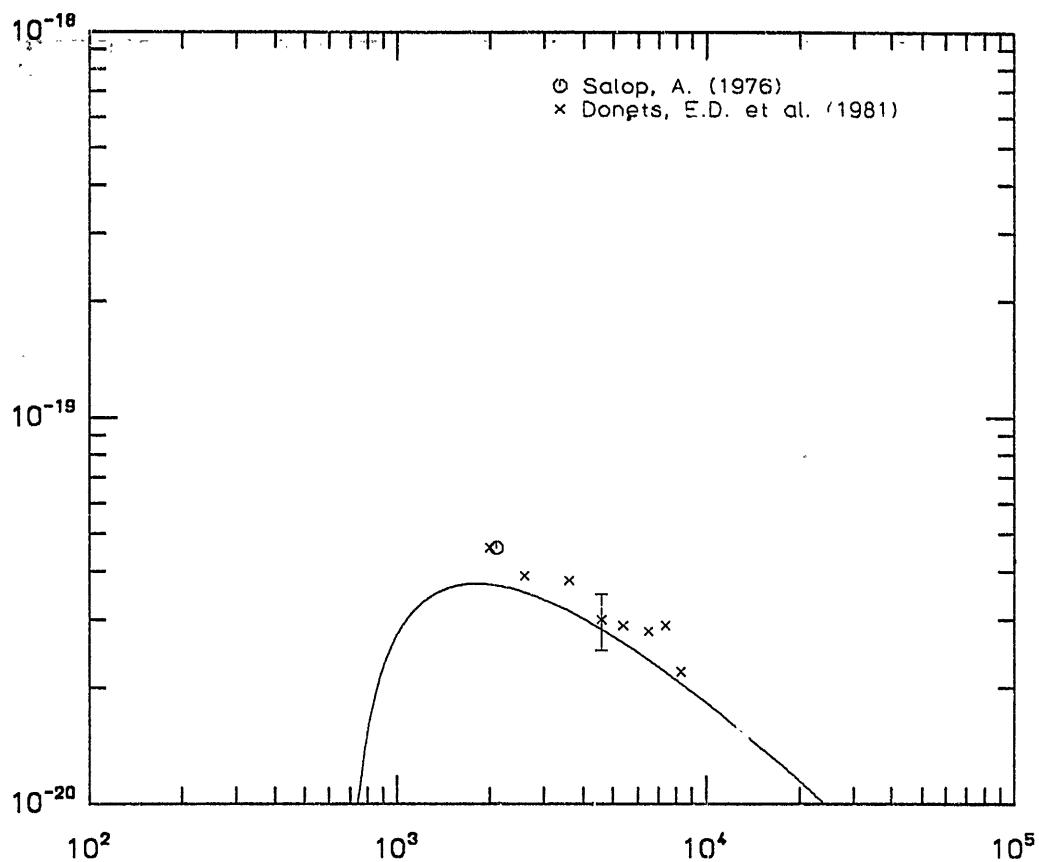
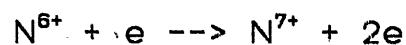


Fig. 40 Cross section vs. Electron energy (eV)

EE 0.4



Electron energy (eV)

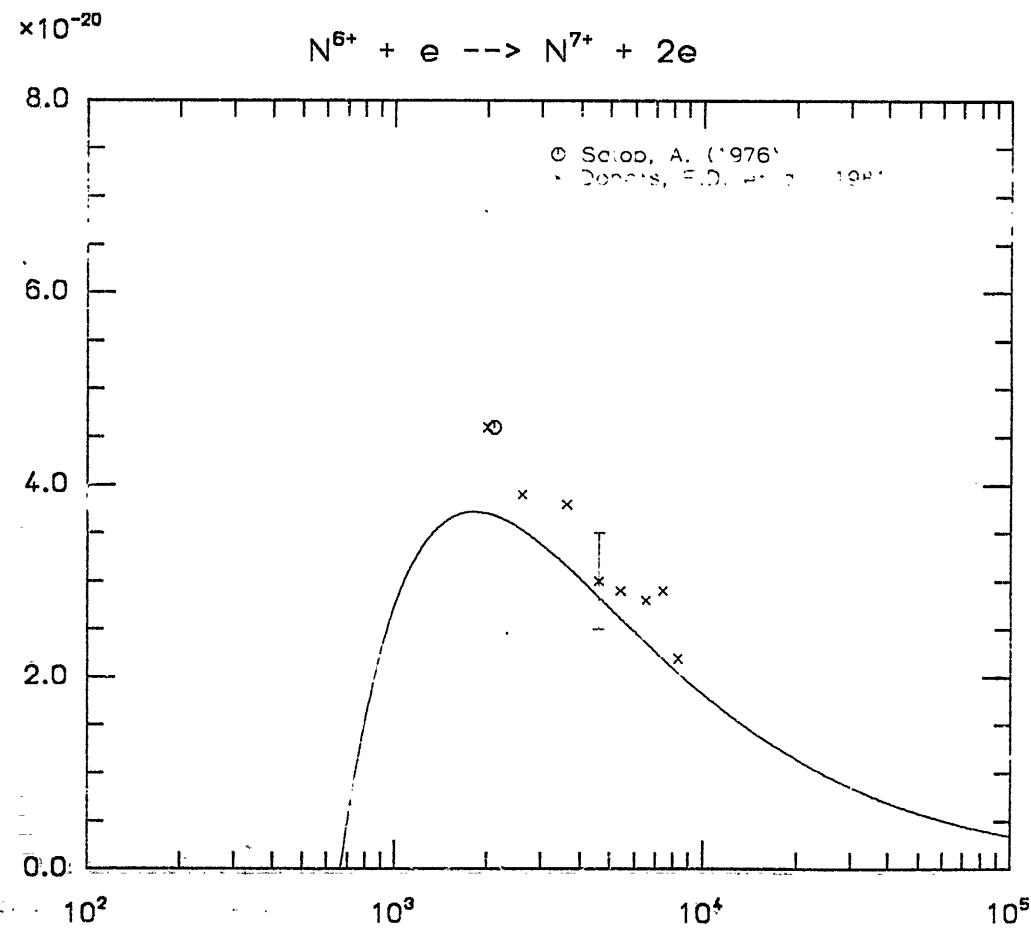


Fig. 41

Electron energy (eV)

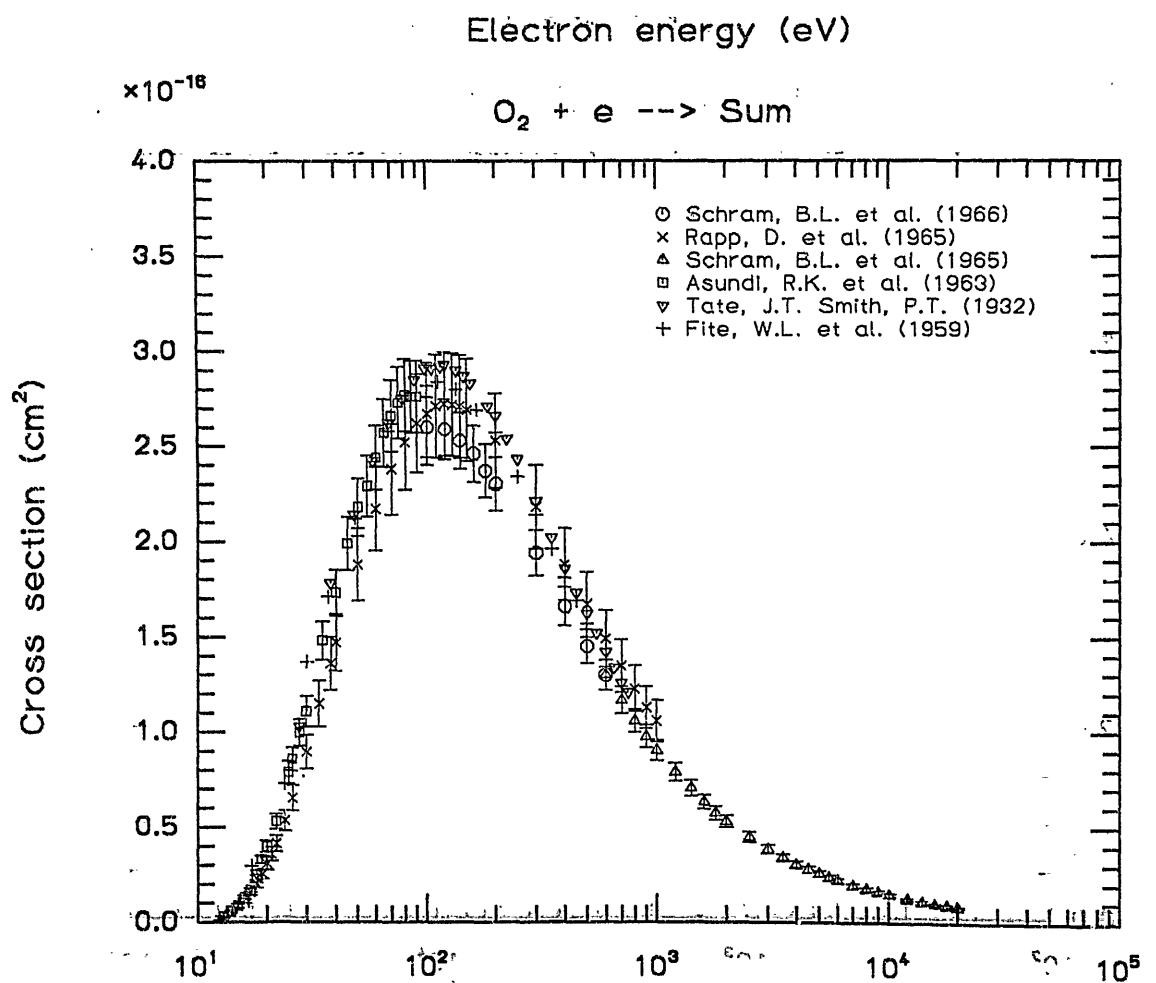
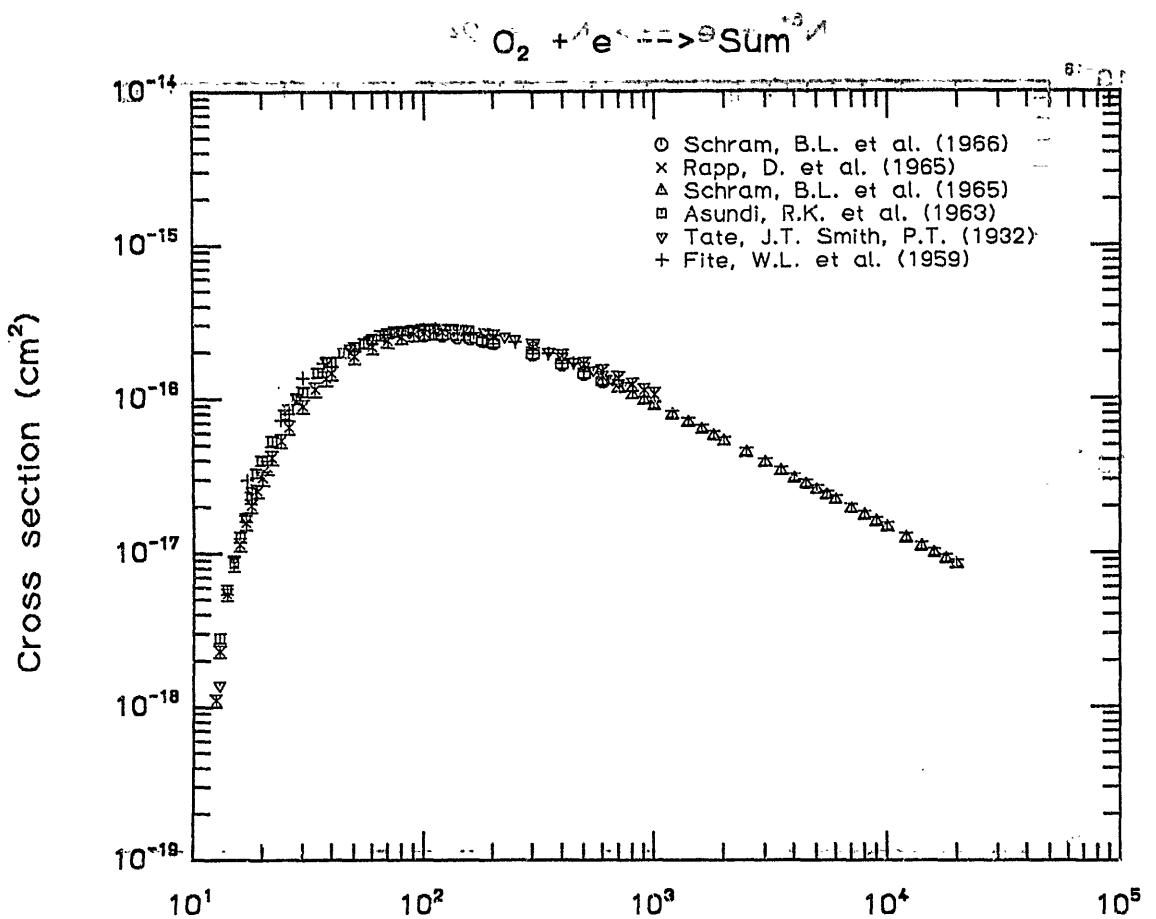
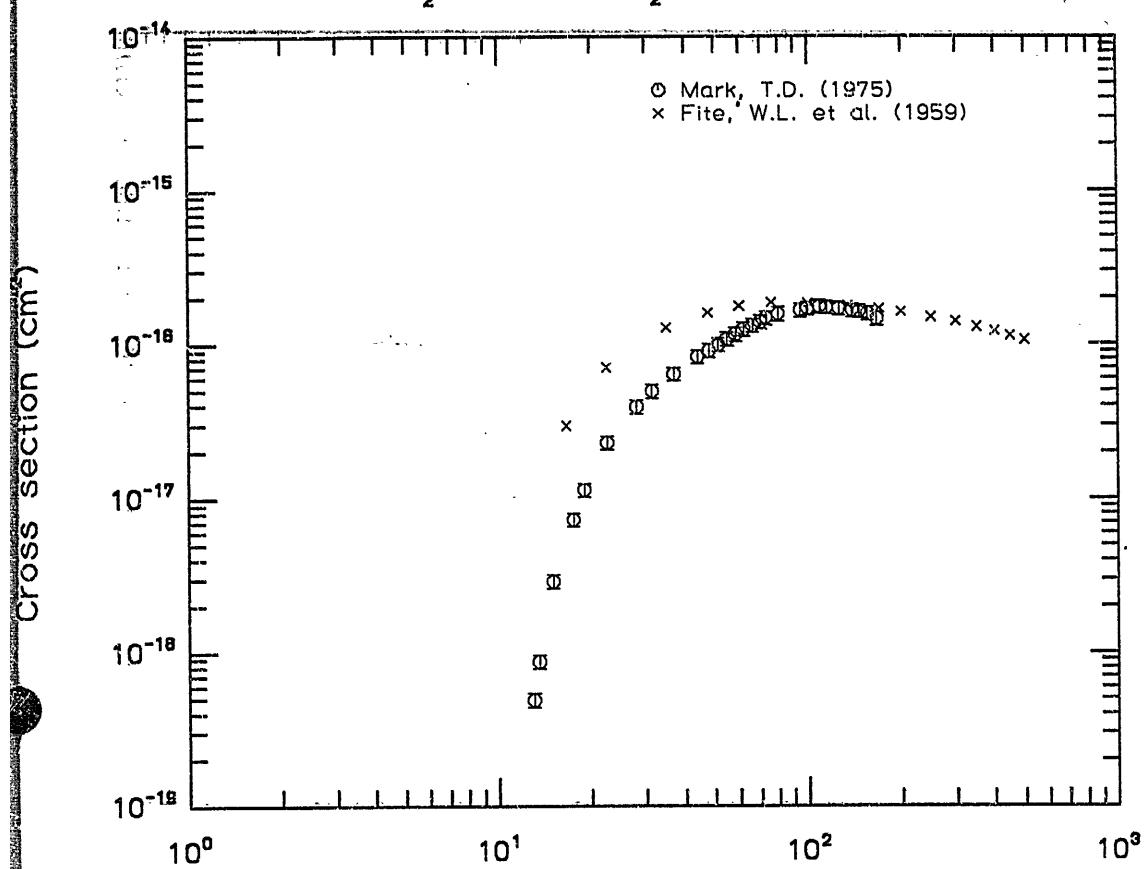
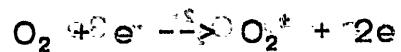


Fig. 42 (Ve) Electron energy (eV) 14. pi?



Electron energy (eV)

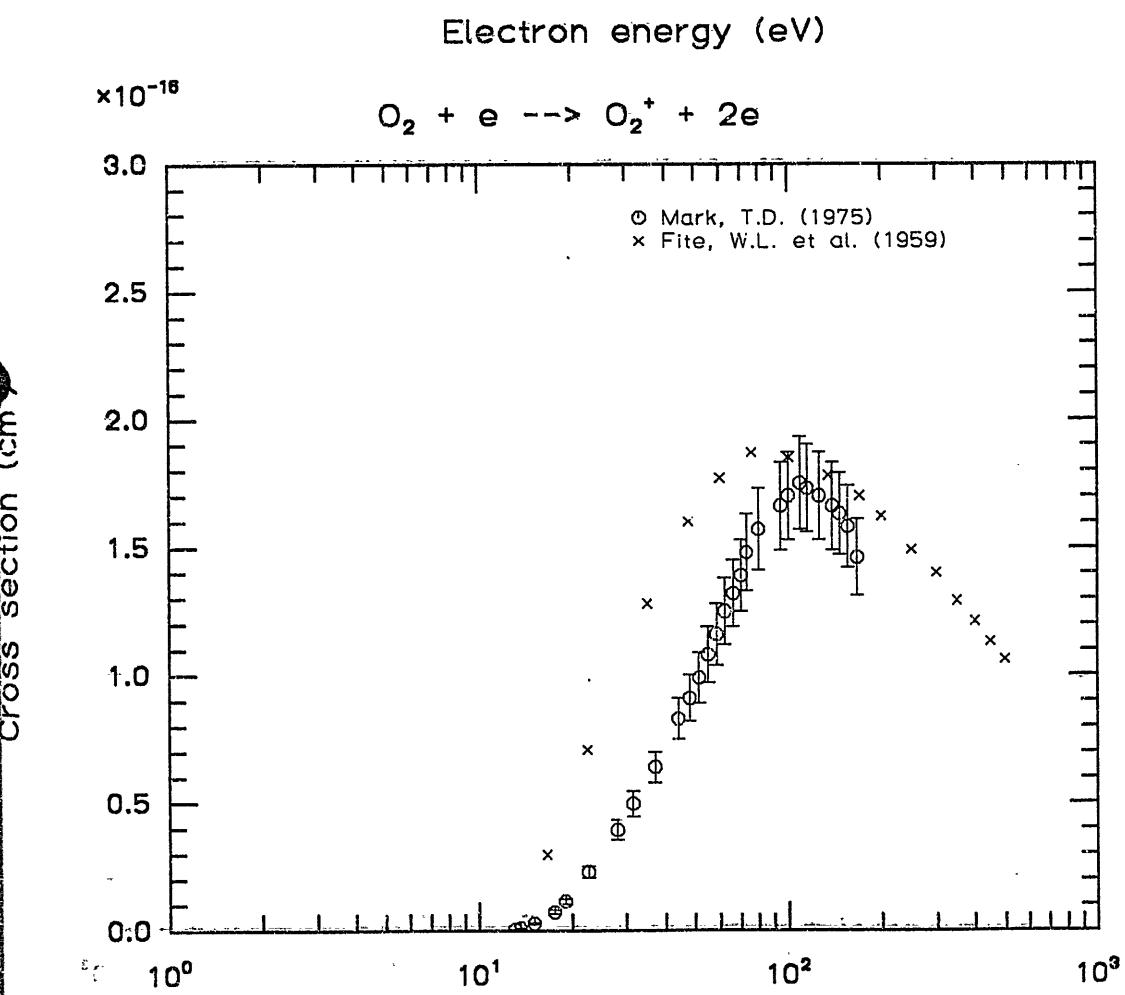
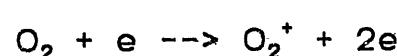


Fig. 43

Electron energy (eV)

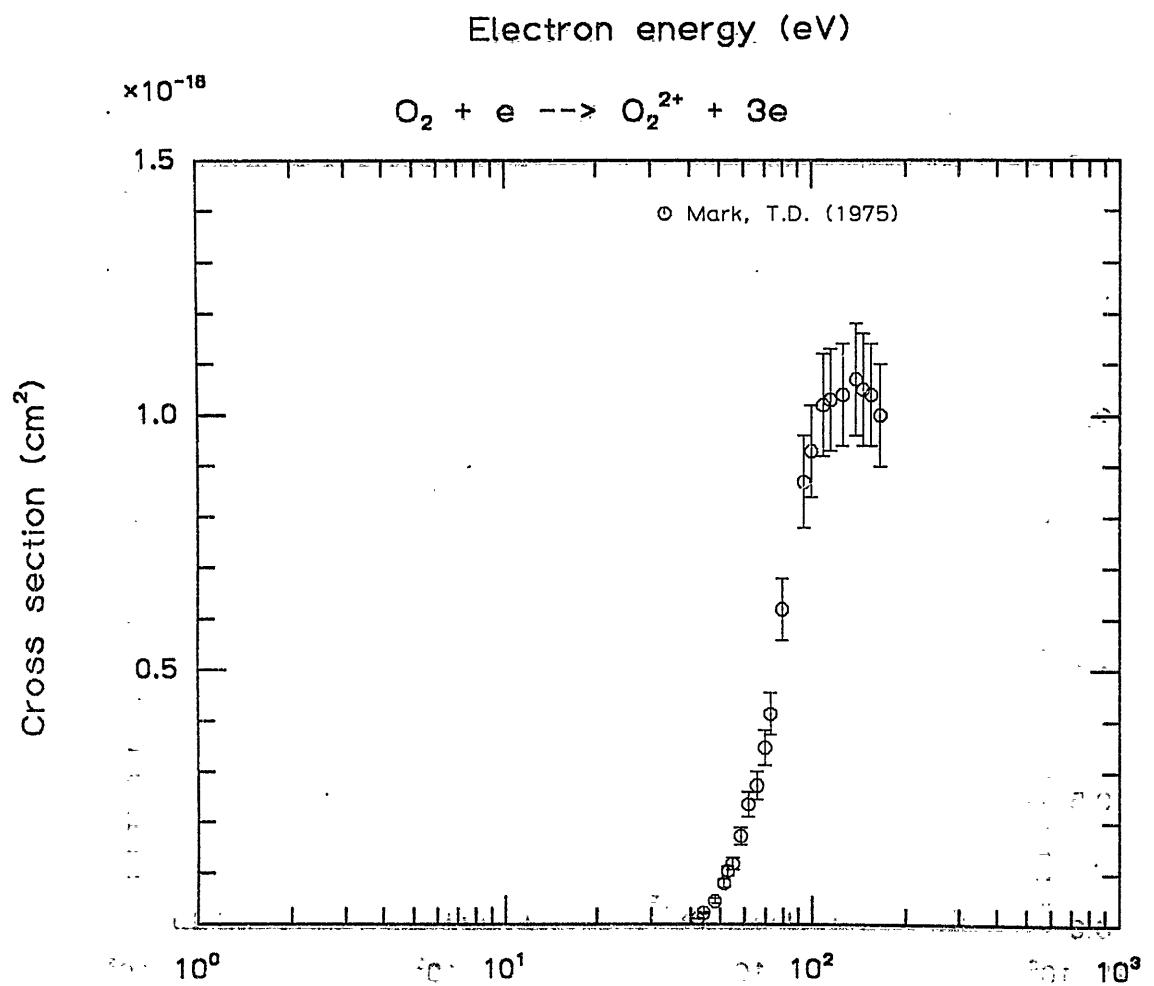
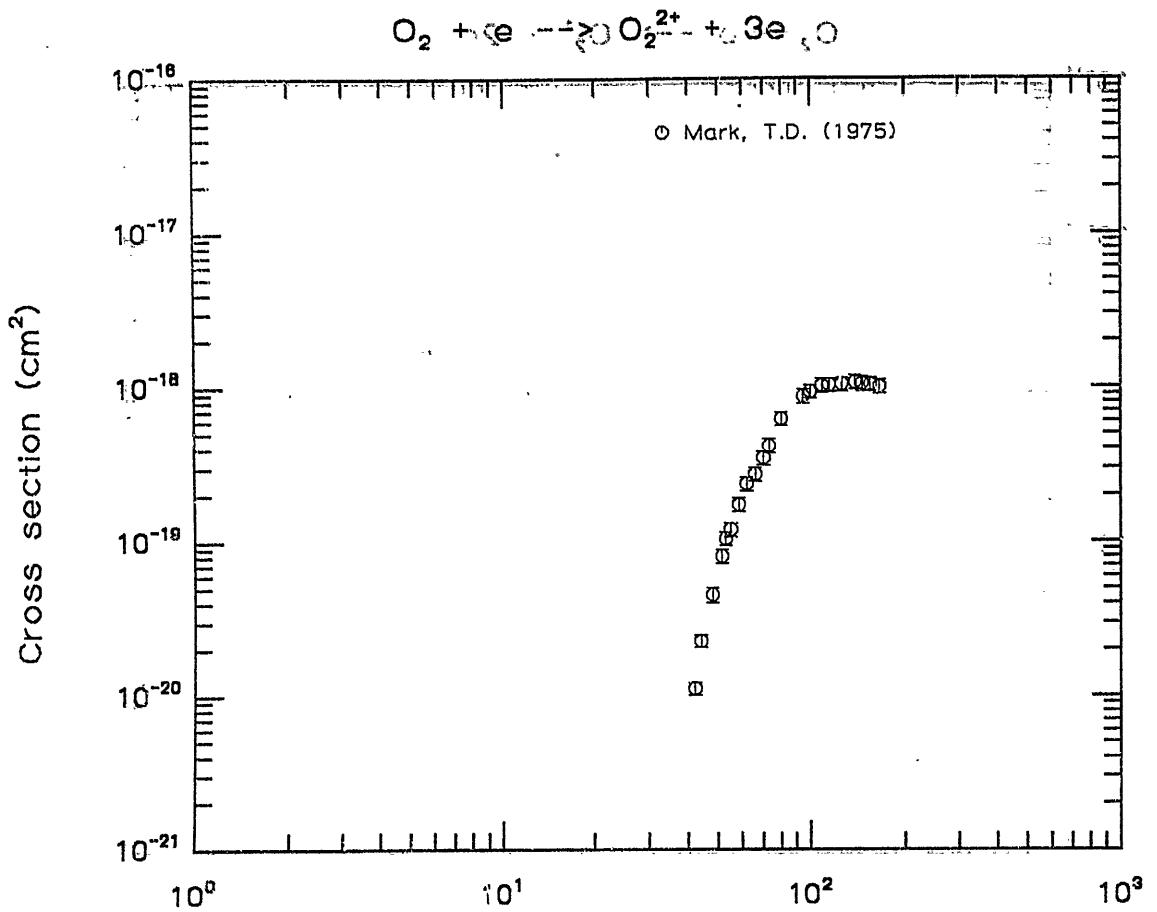


Fig. 44 (a) Electron energy (eV)

Fig. 44 (b)

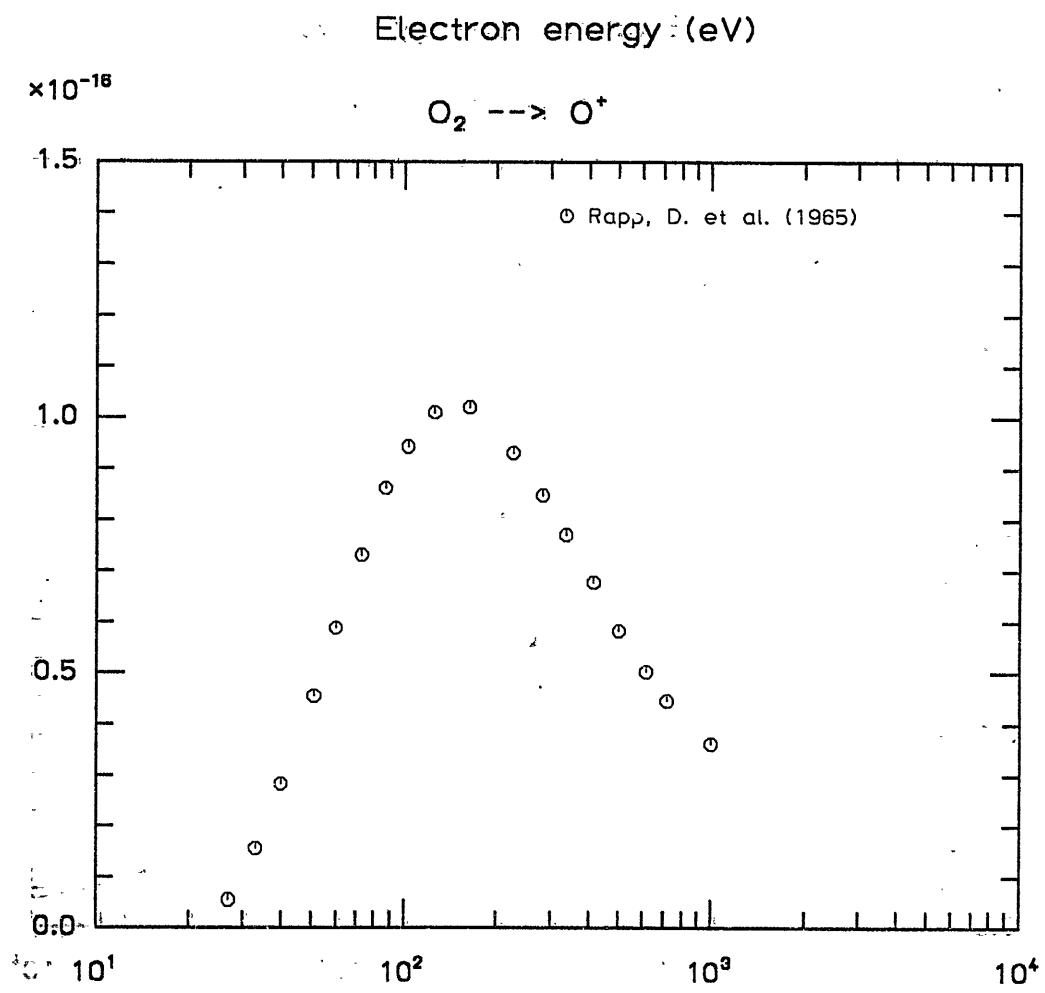
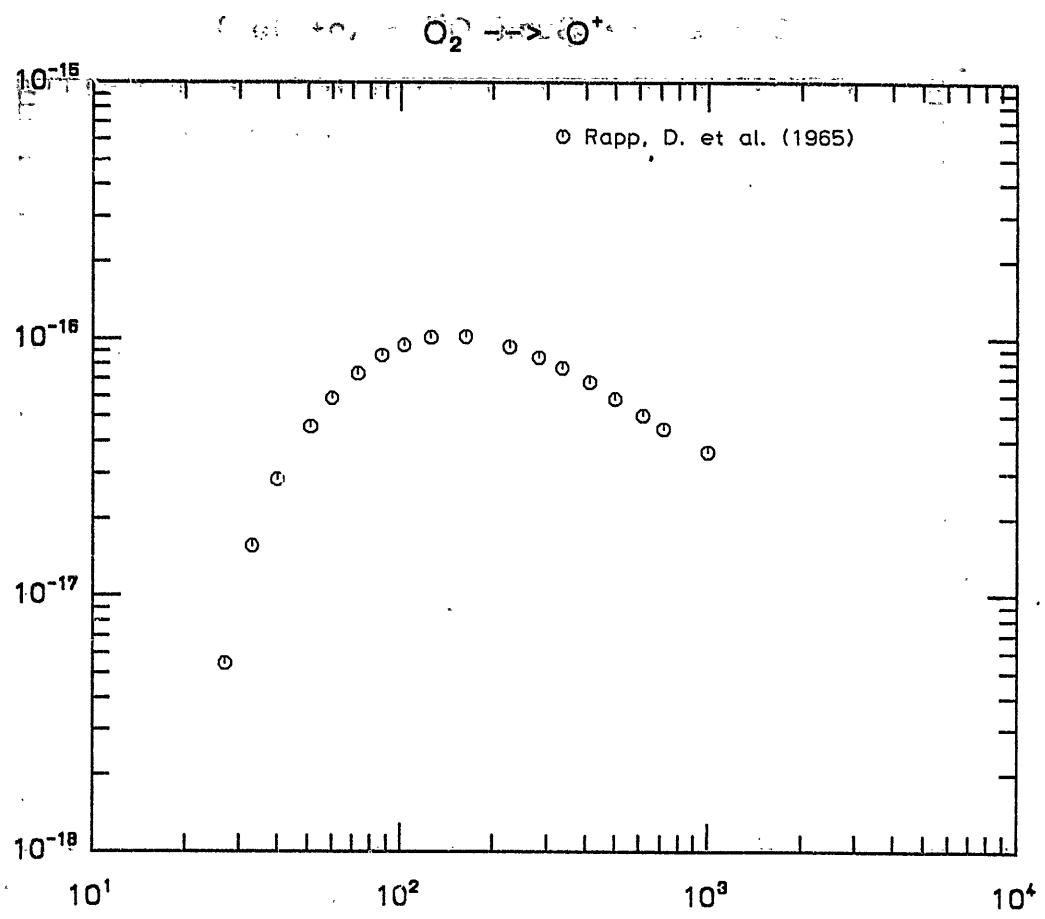


Fig. 45 Electron energy (eV) 64 64

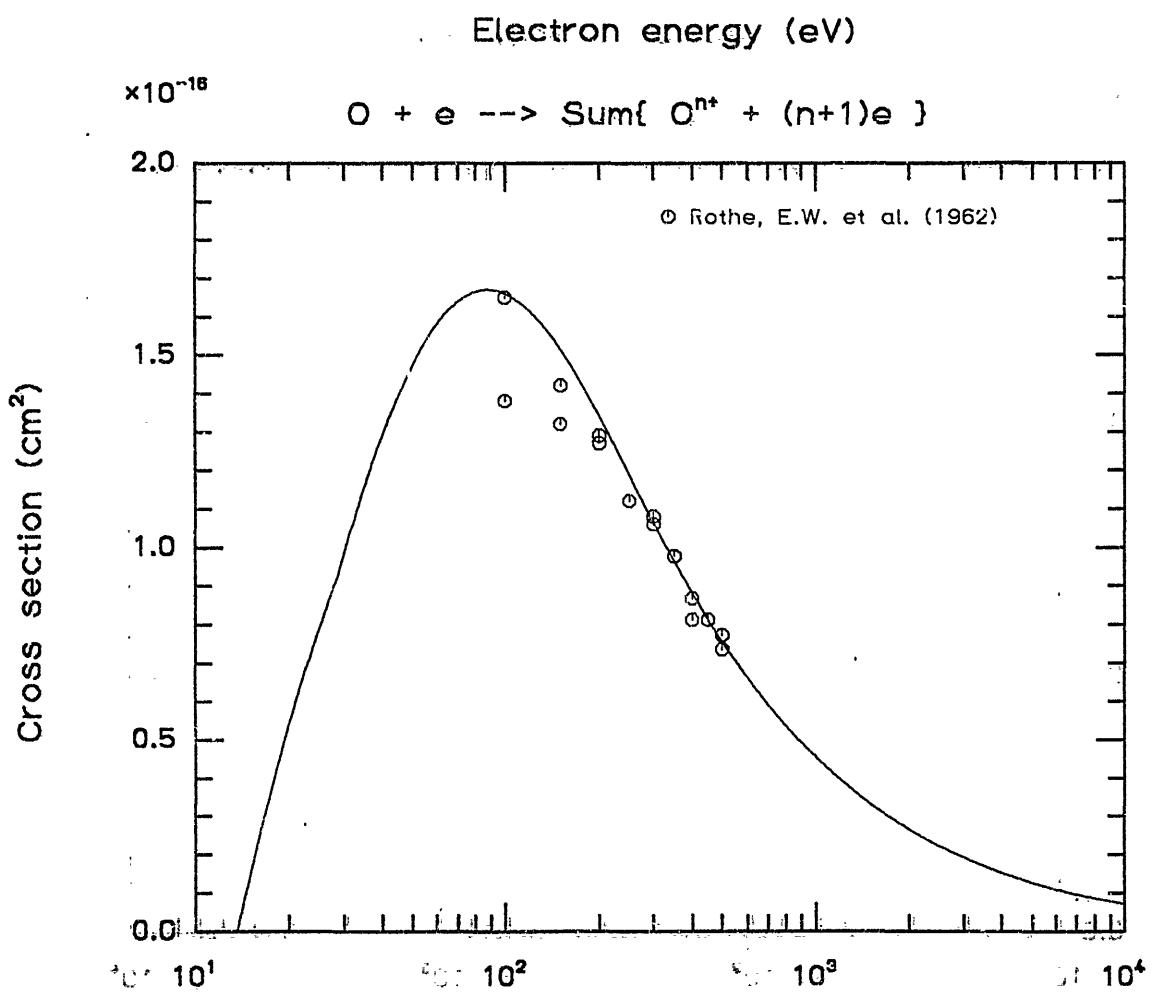
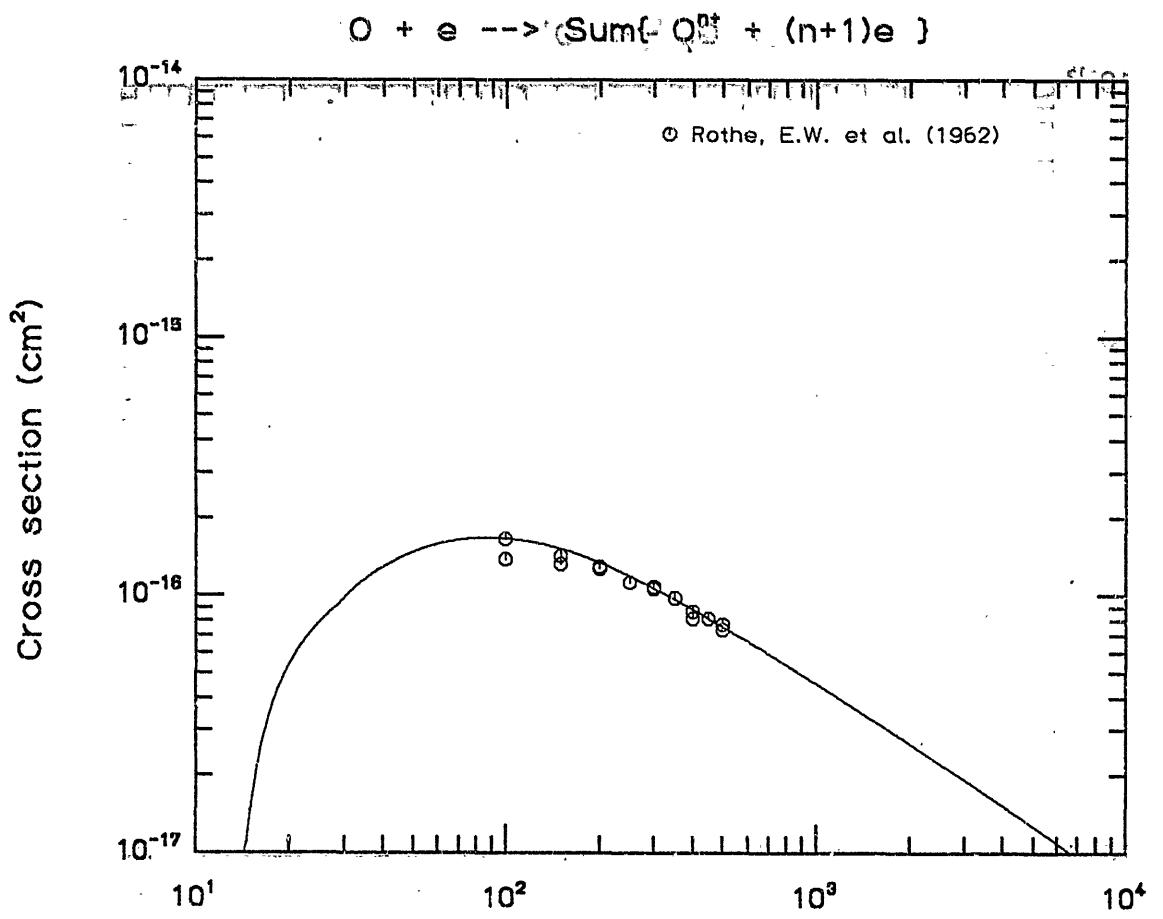
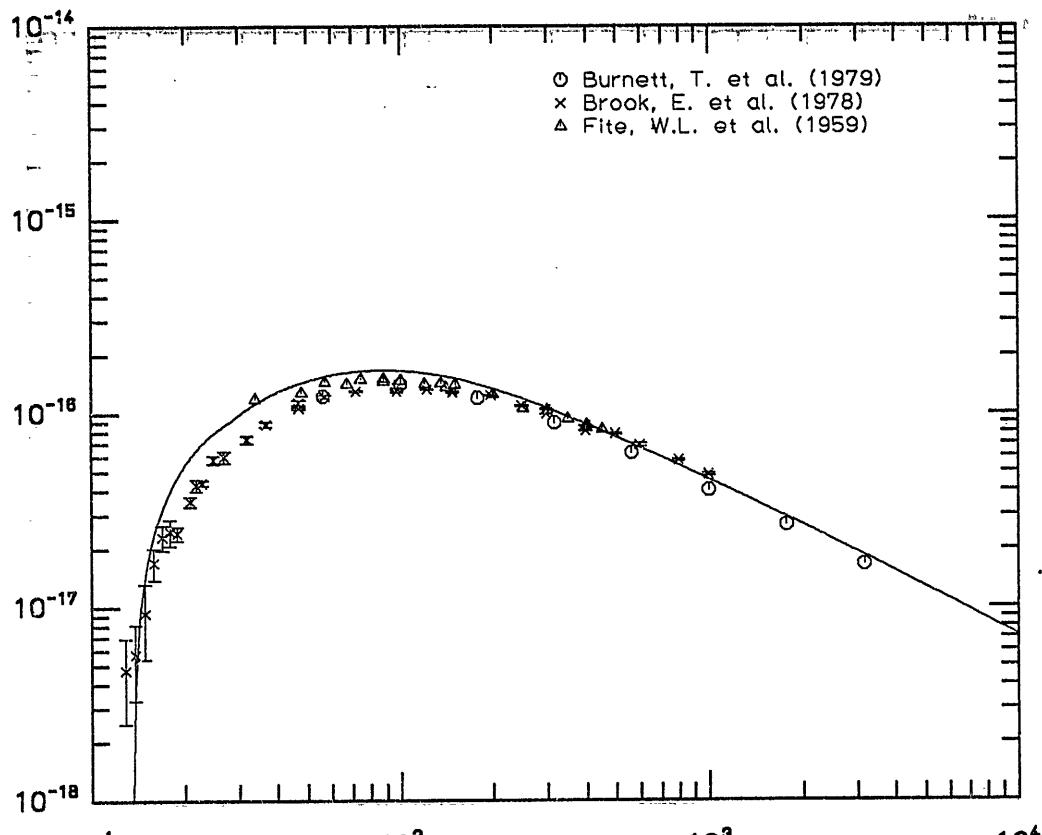
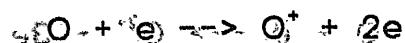


Fig. 46 (V) Electron energy (eV) 24. fig



Electron energy (eV)

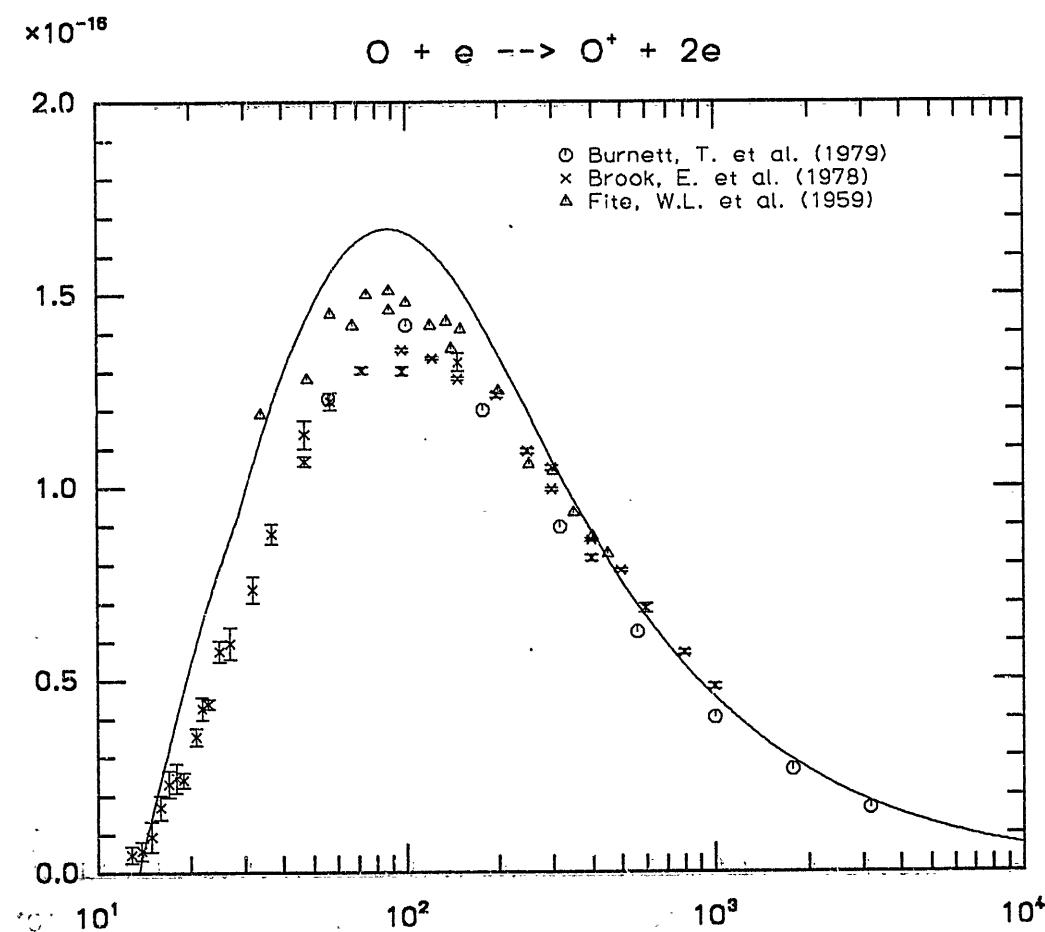


Fig. 47 Electron energy (eV) 3.4.4

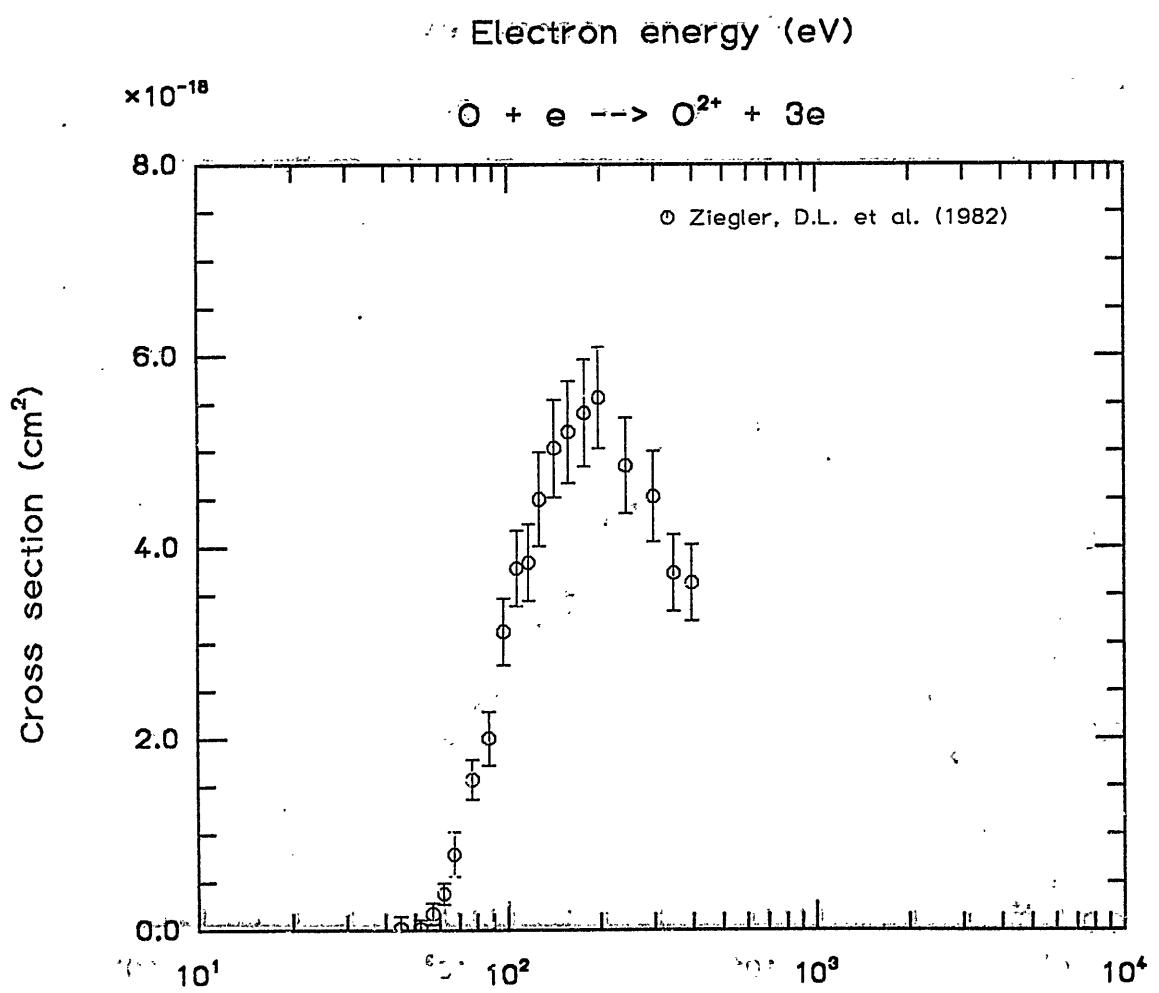
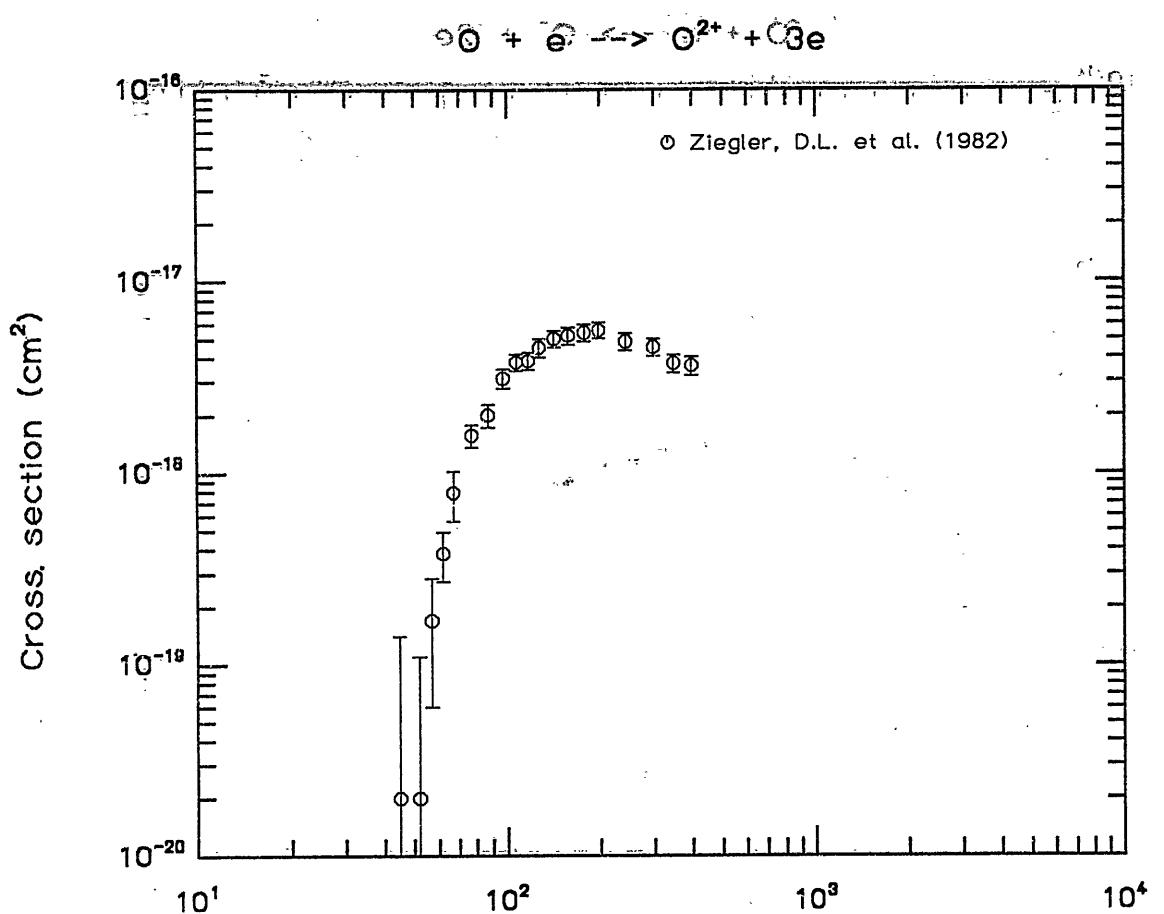


Fig. 48

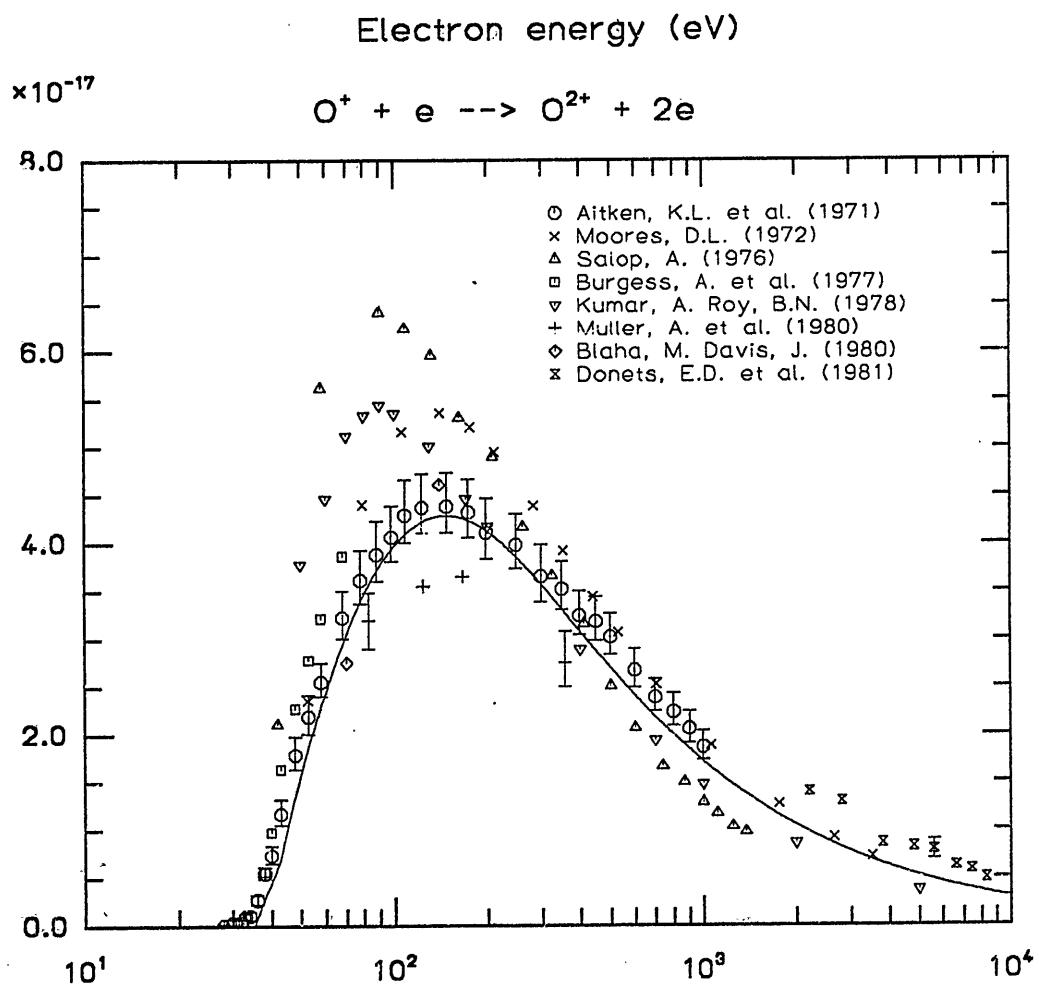
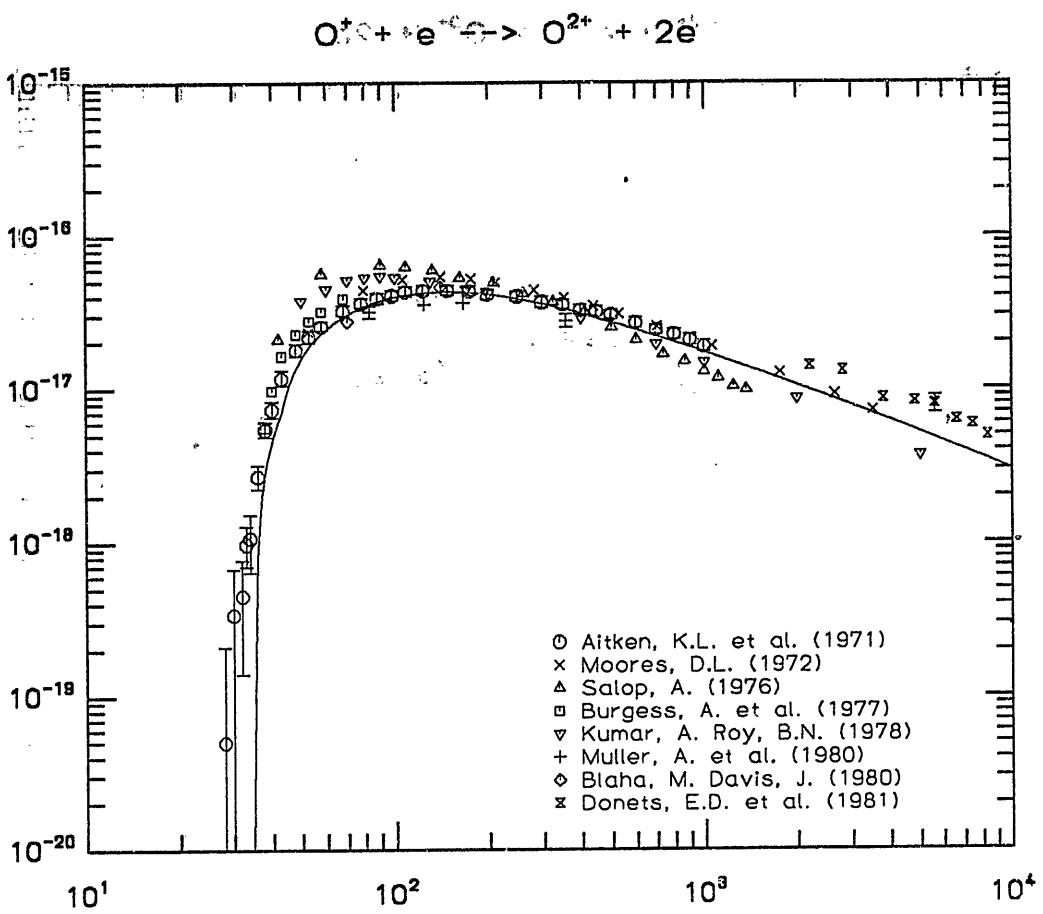


Fig. 49

Electron energy (eV)

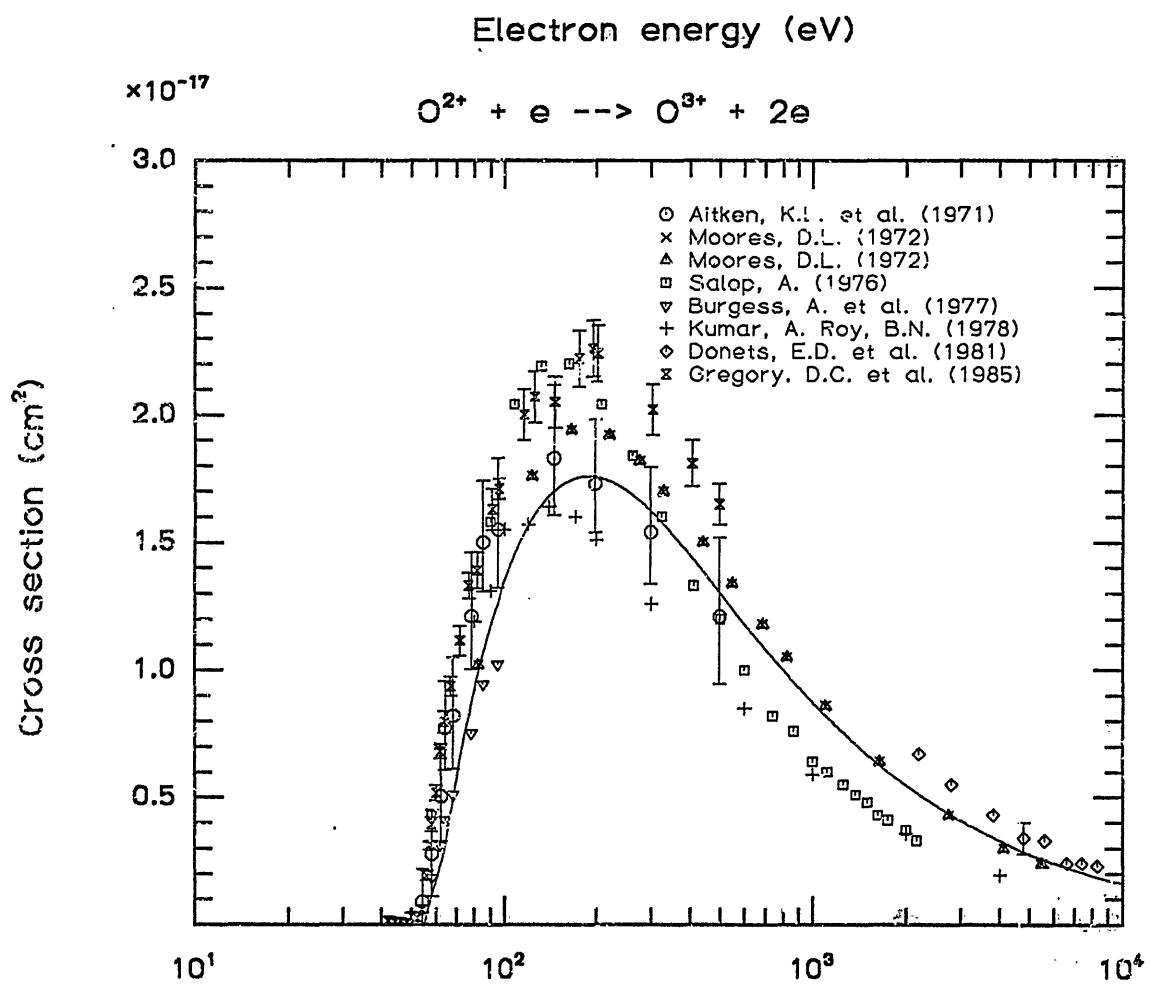
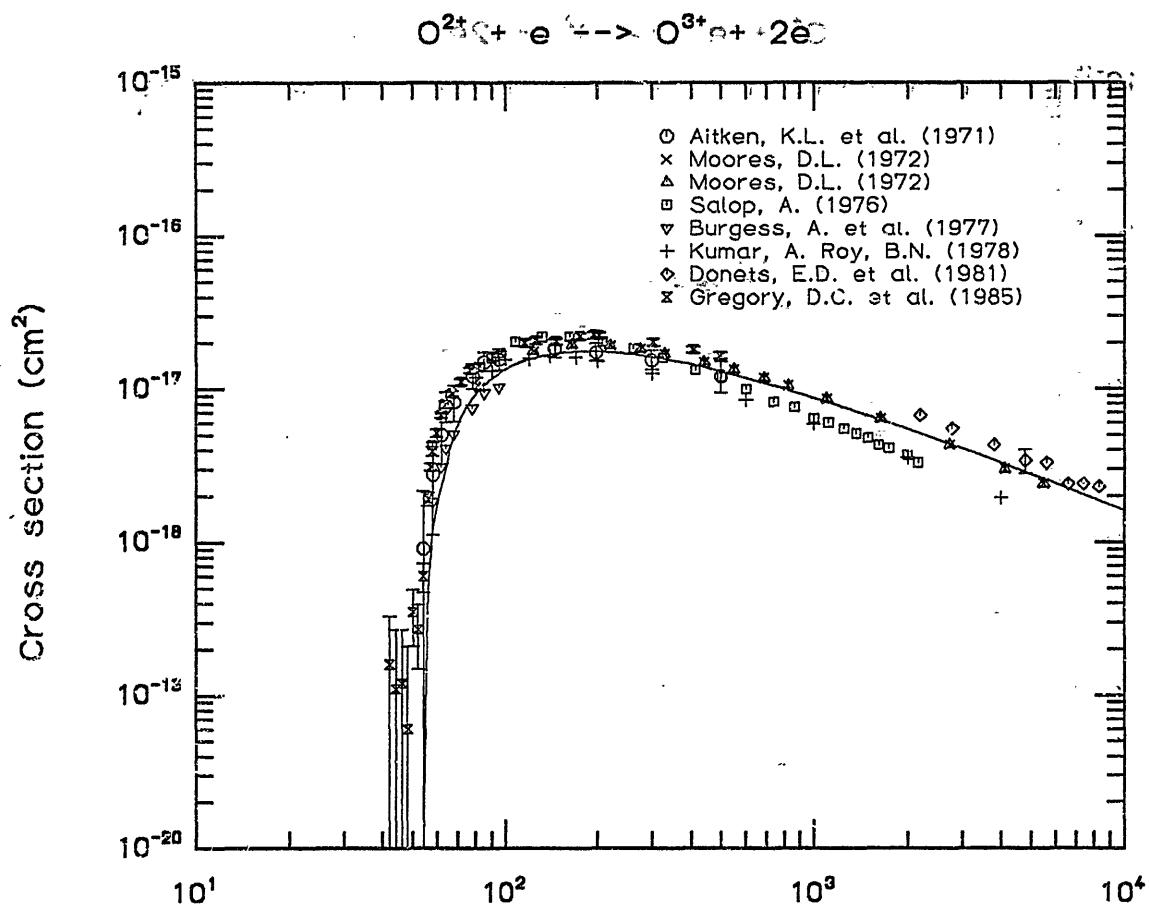


Fig. 50

Electron energy (eV)

Op. 17

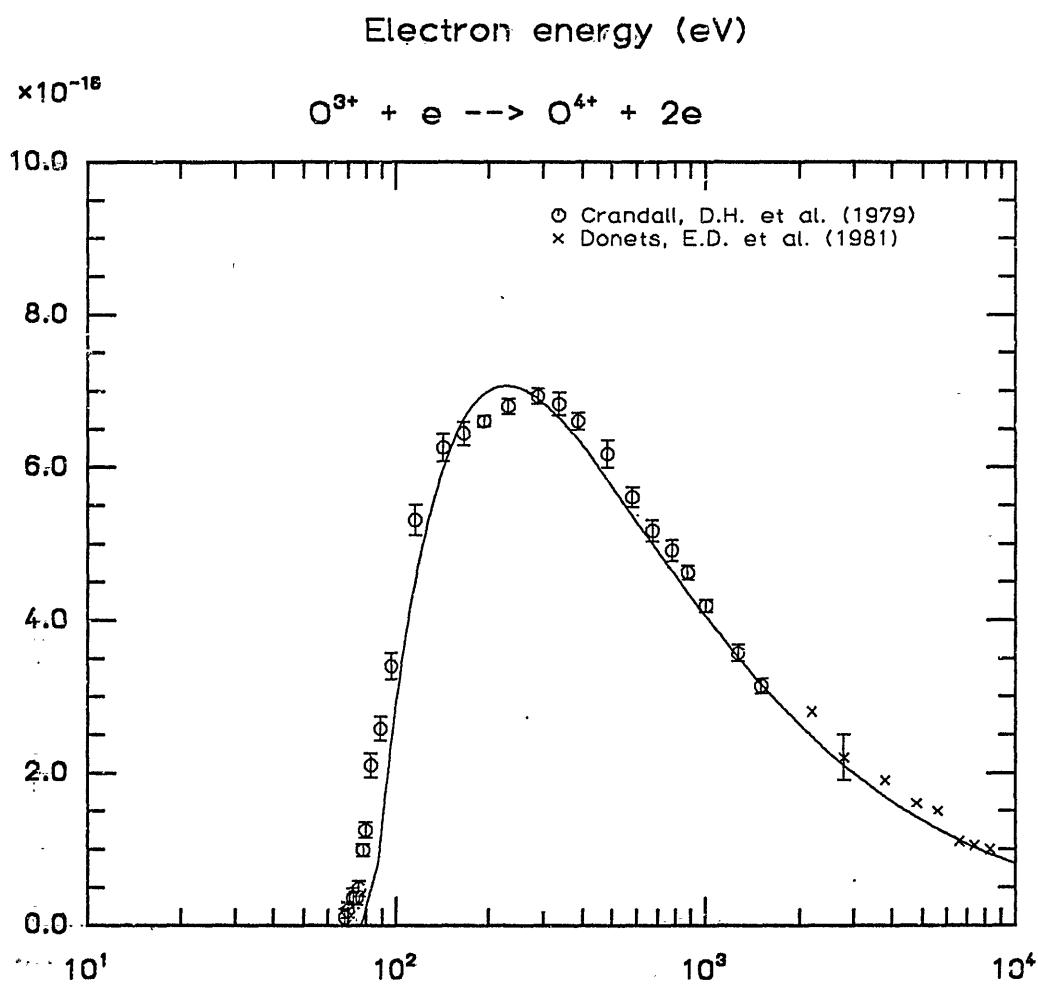
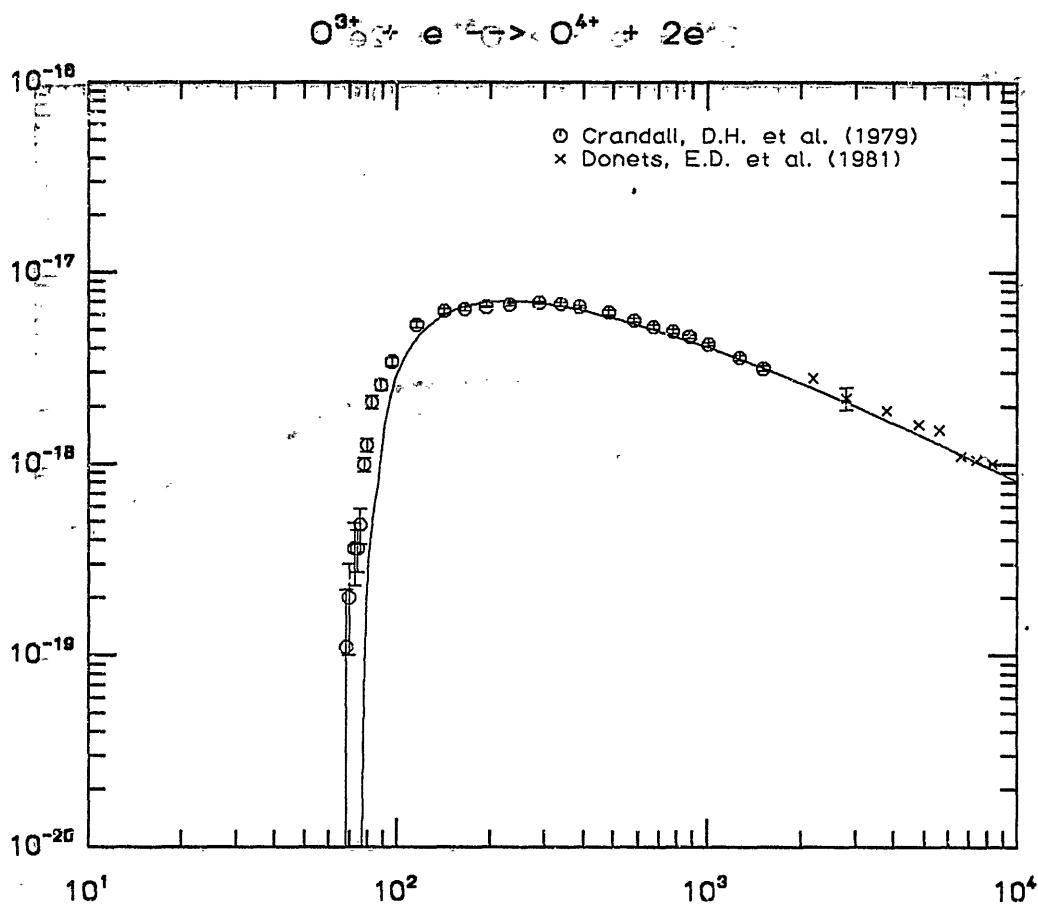


Fig. 51.

Electron energy (eV)

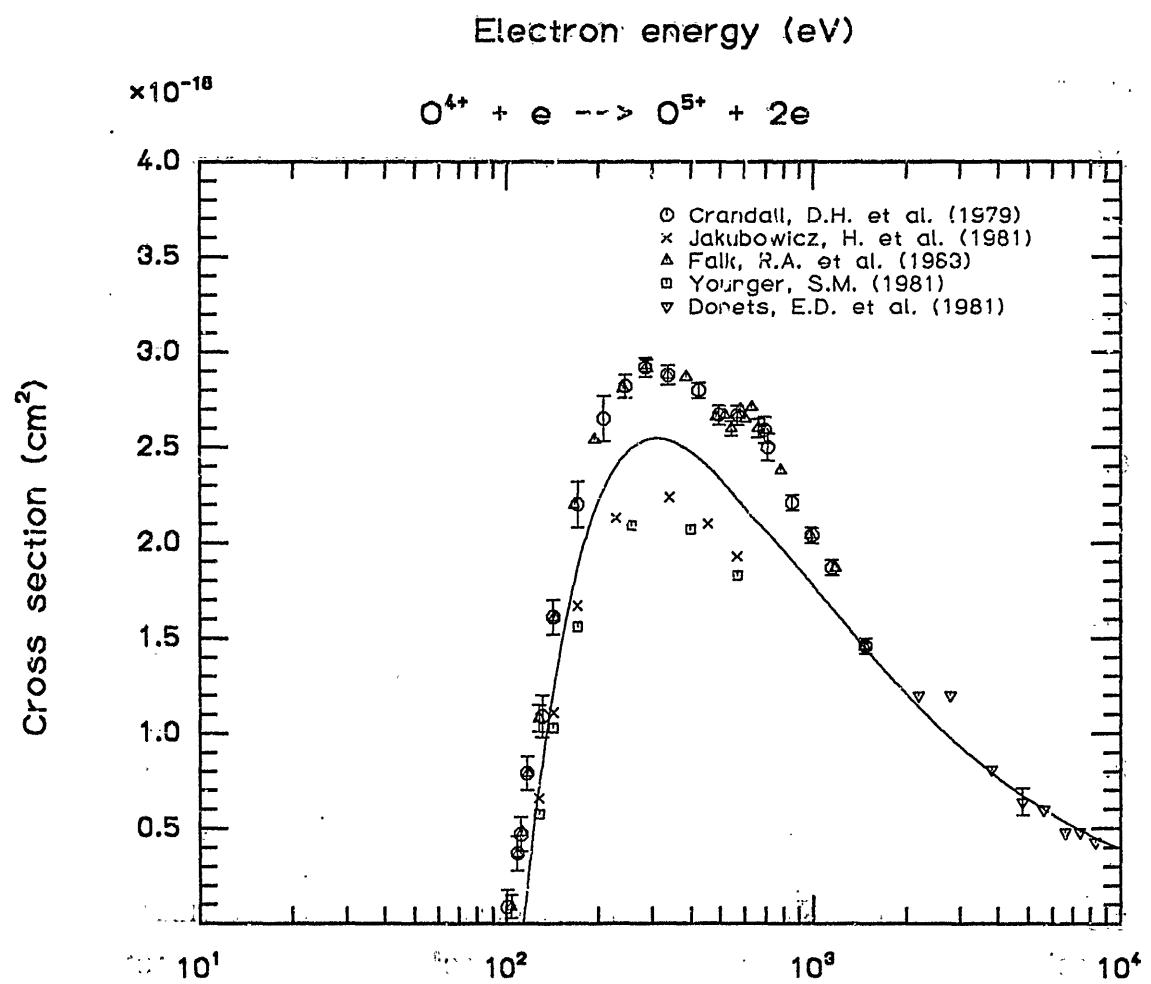
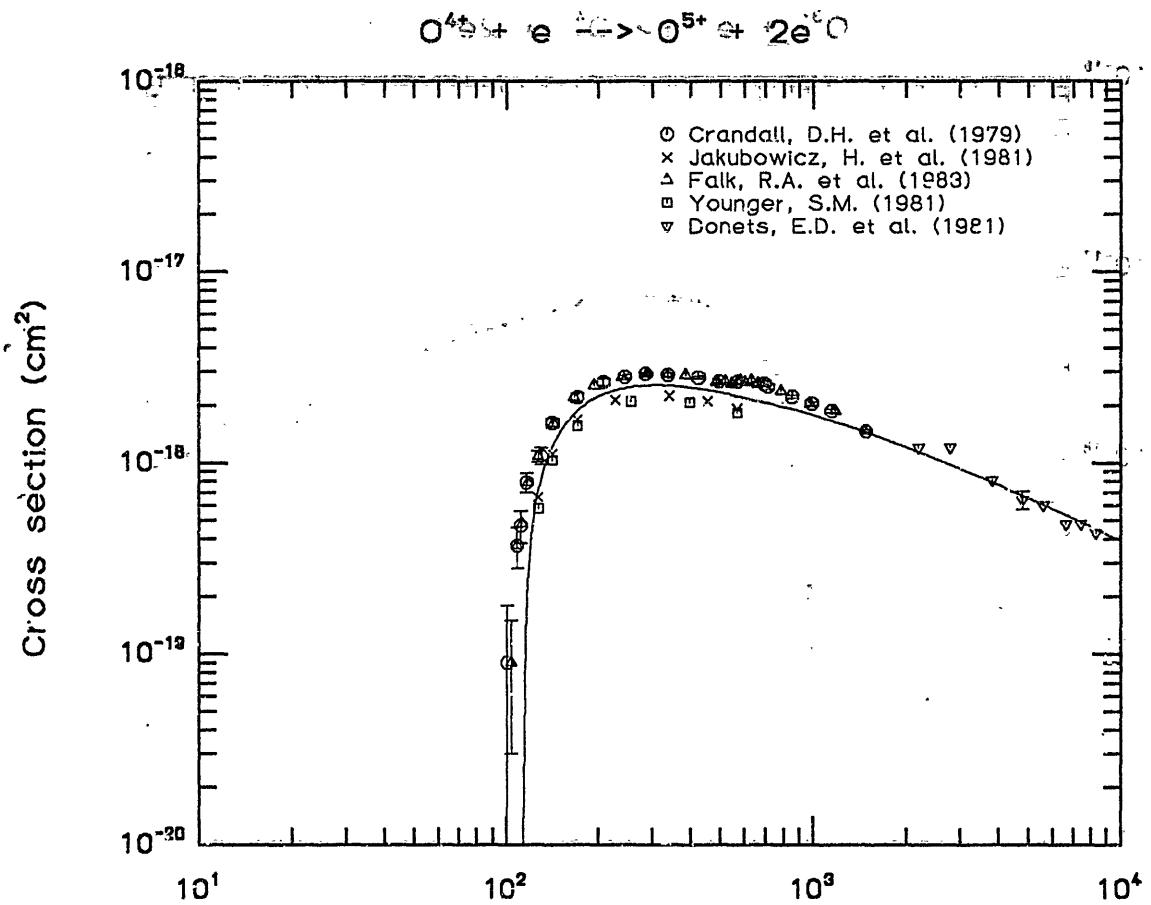
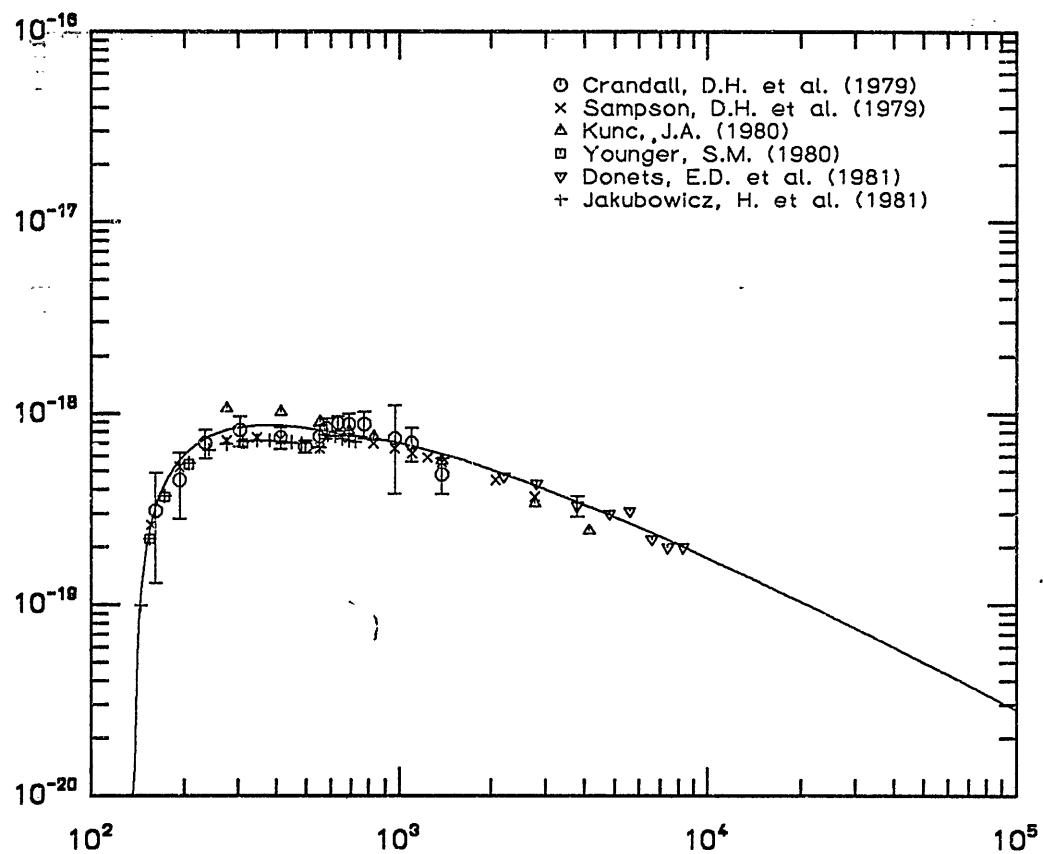
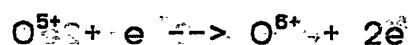


Fig. 52 (a) Electron energy (eV) (b)  $O^{4+} + e^- \rightarrow O^{5+} + 2e^-$



Electron energy (eV)

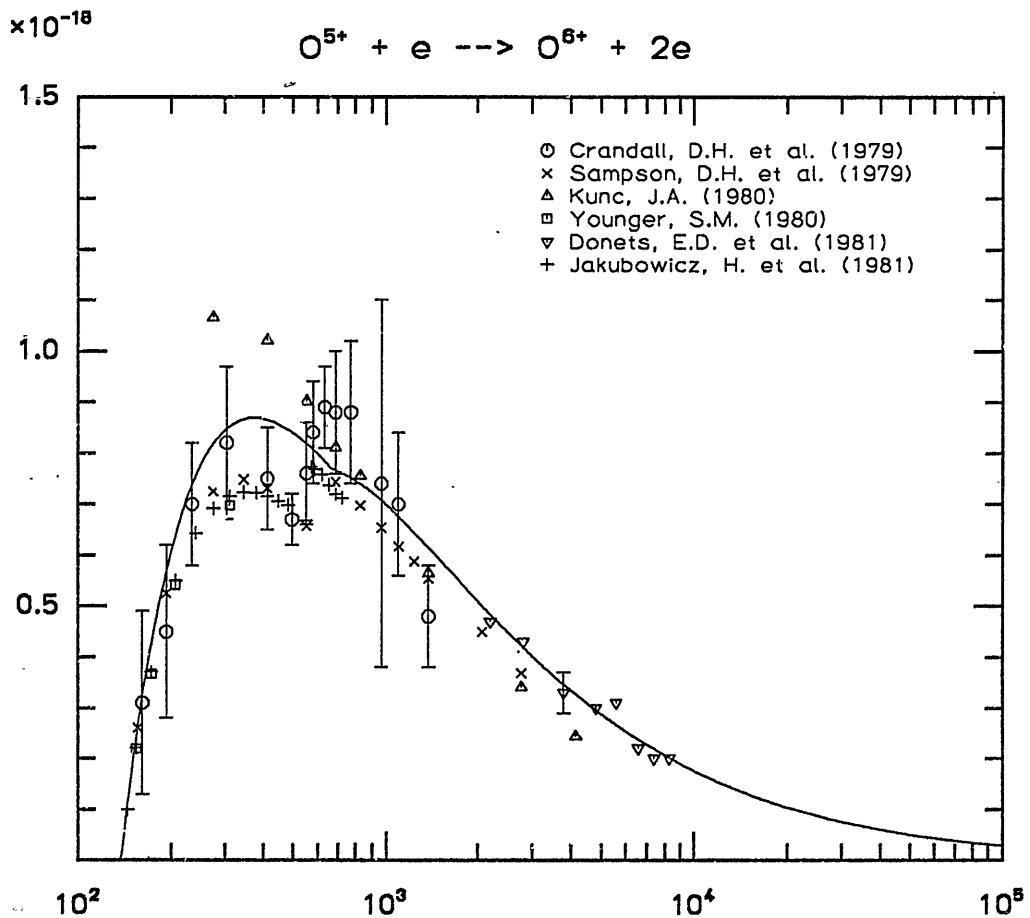


Fig. 53

Electron energy (eV)

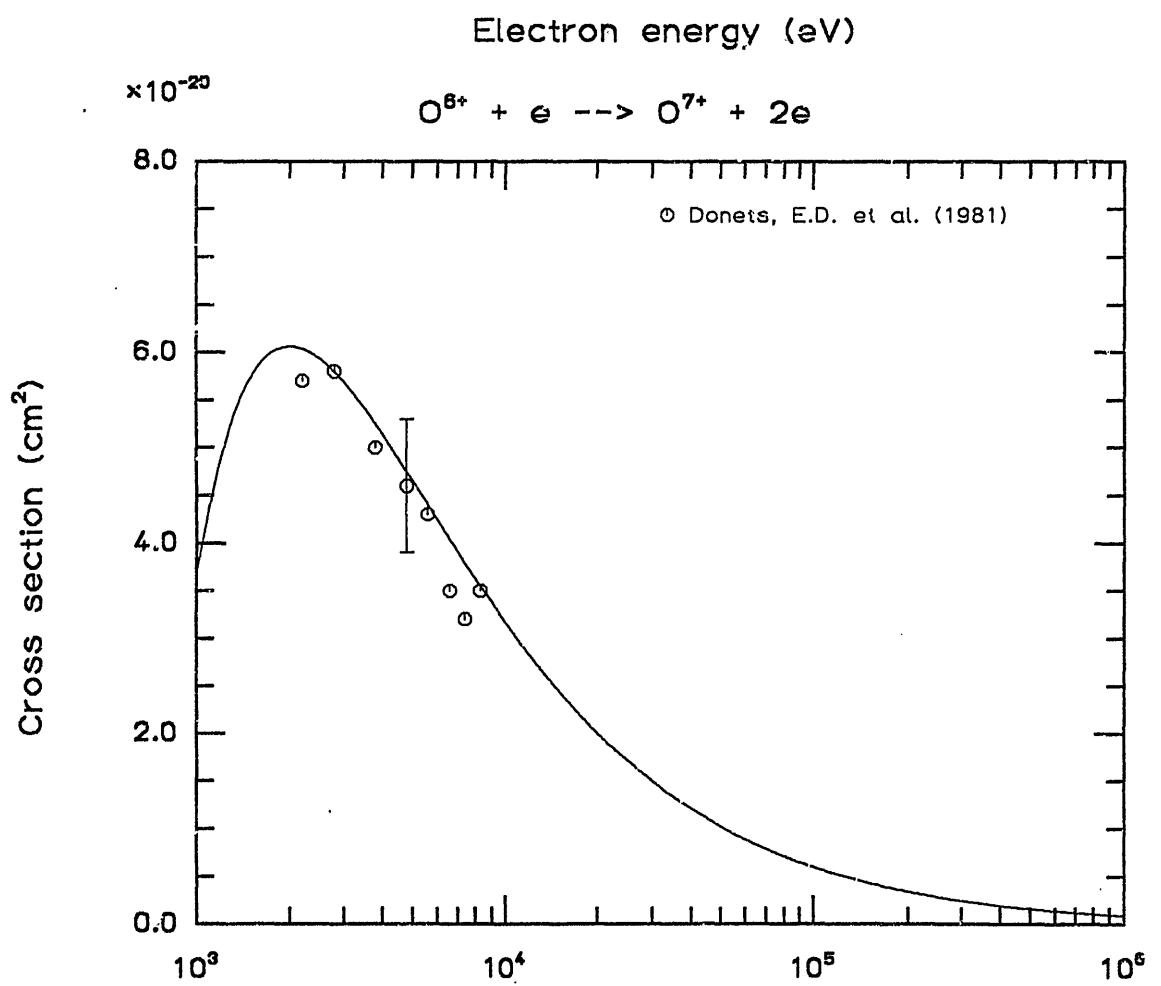
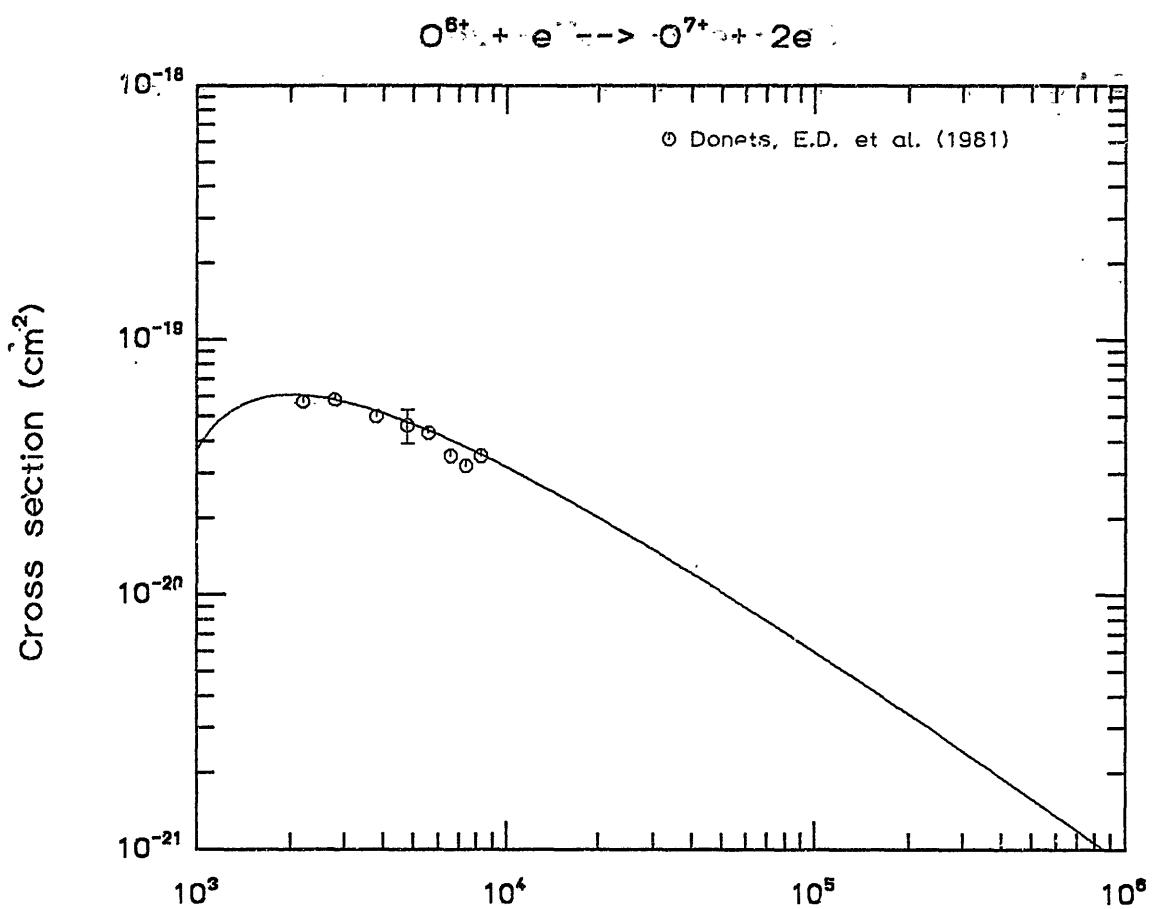
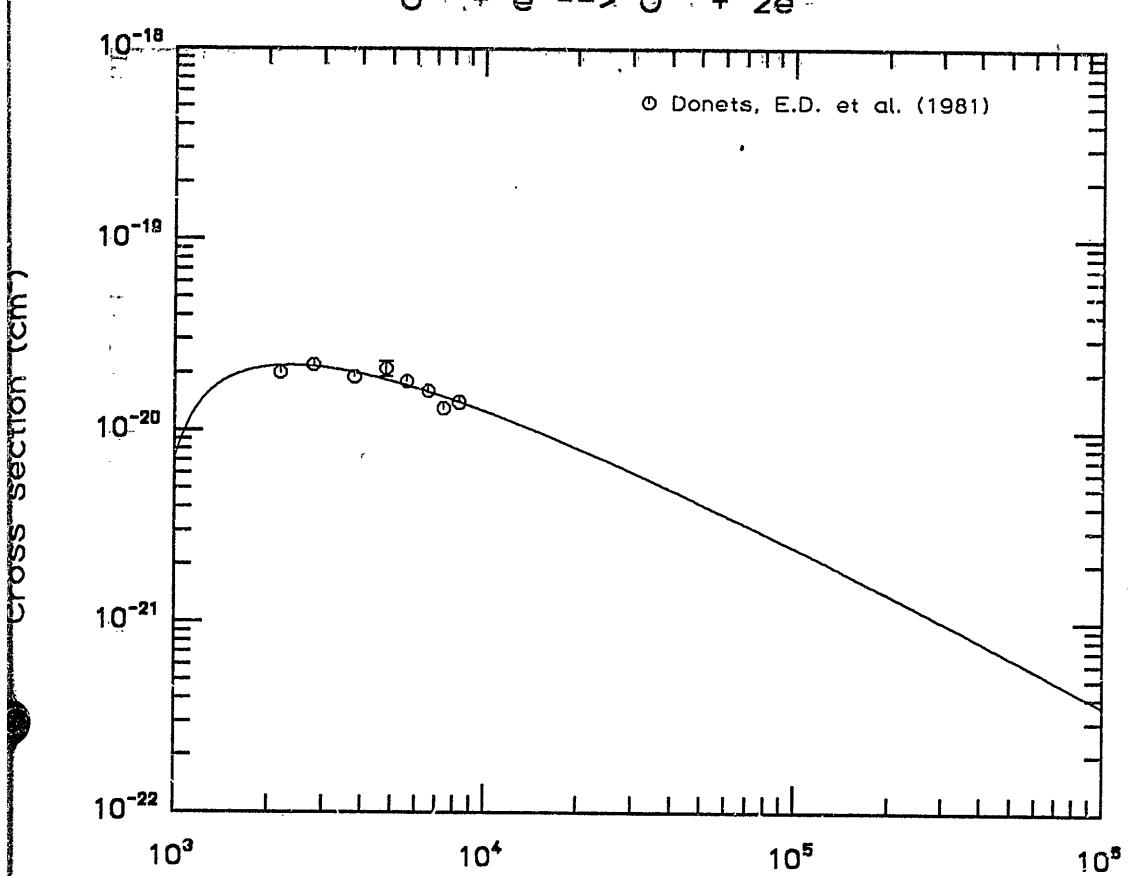


Fig. 54

Electron energy (eV)



Electron energy (eV)

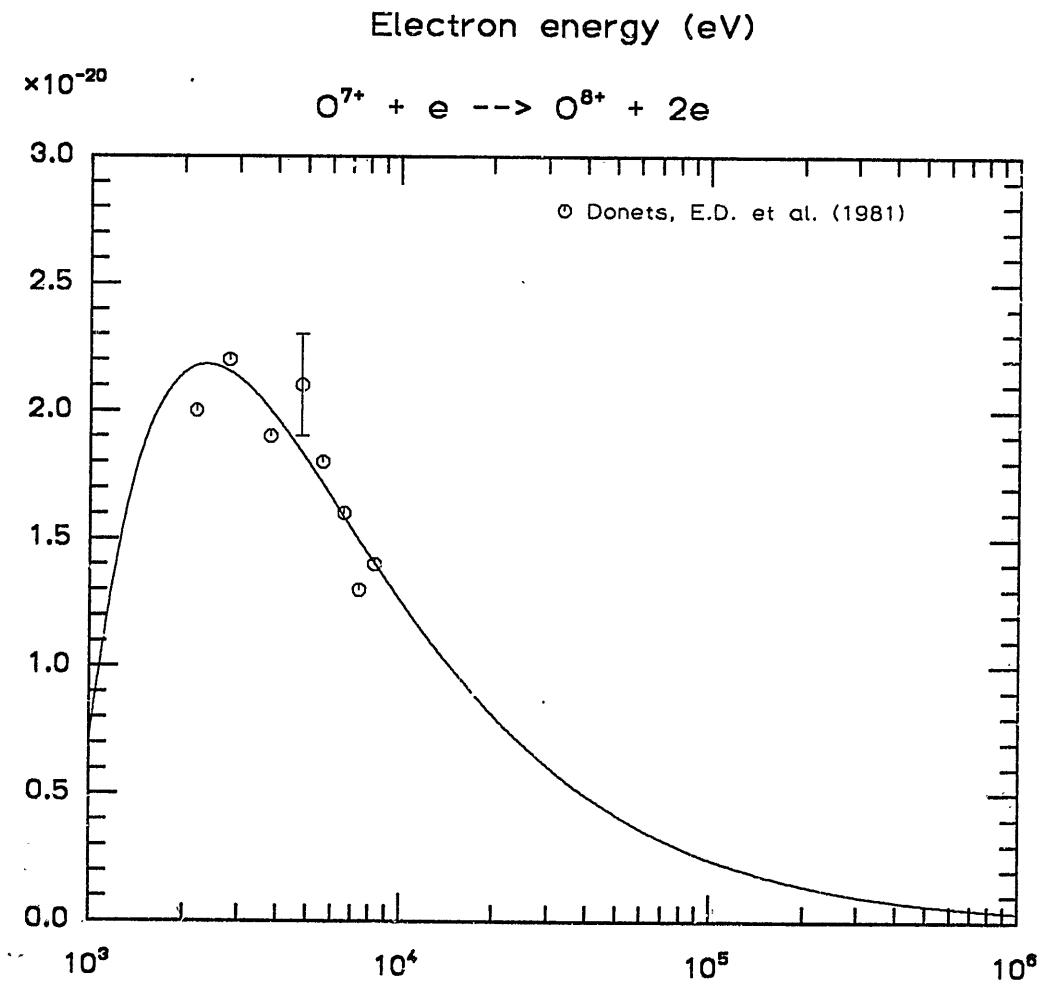
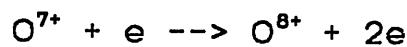
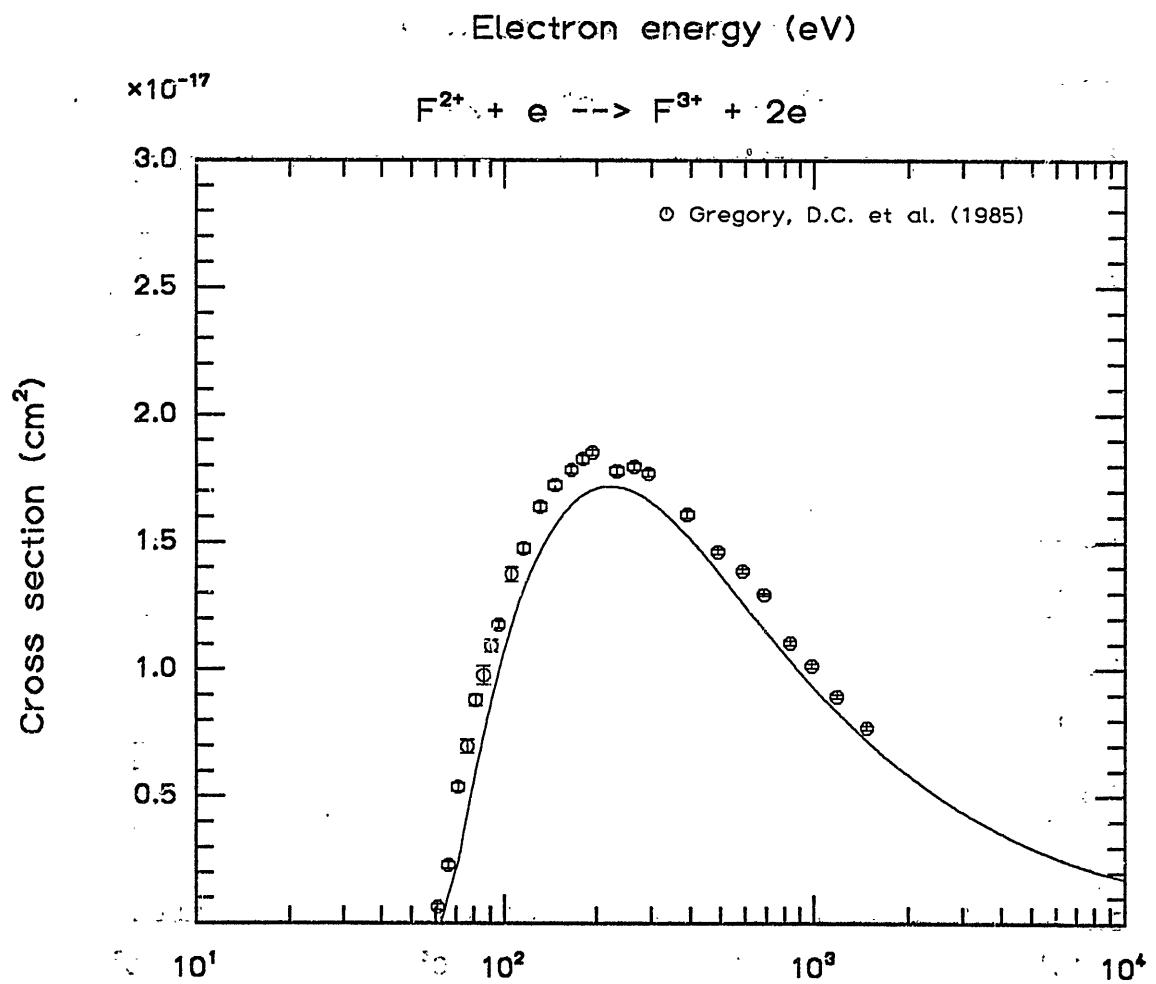
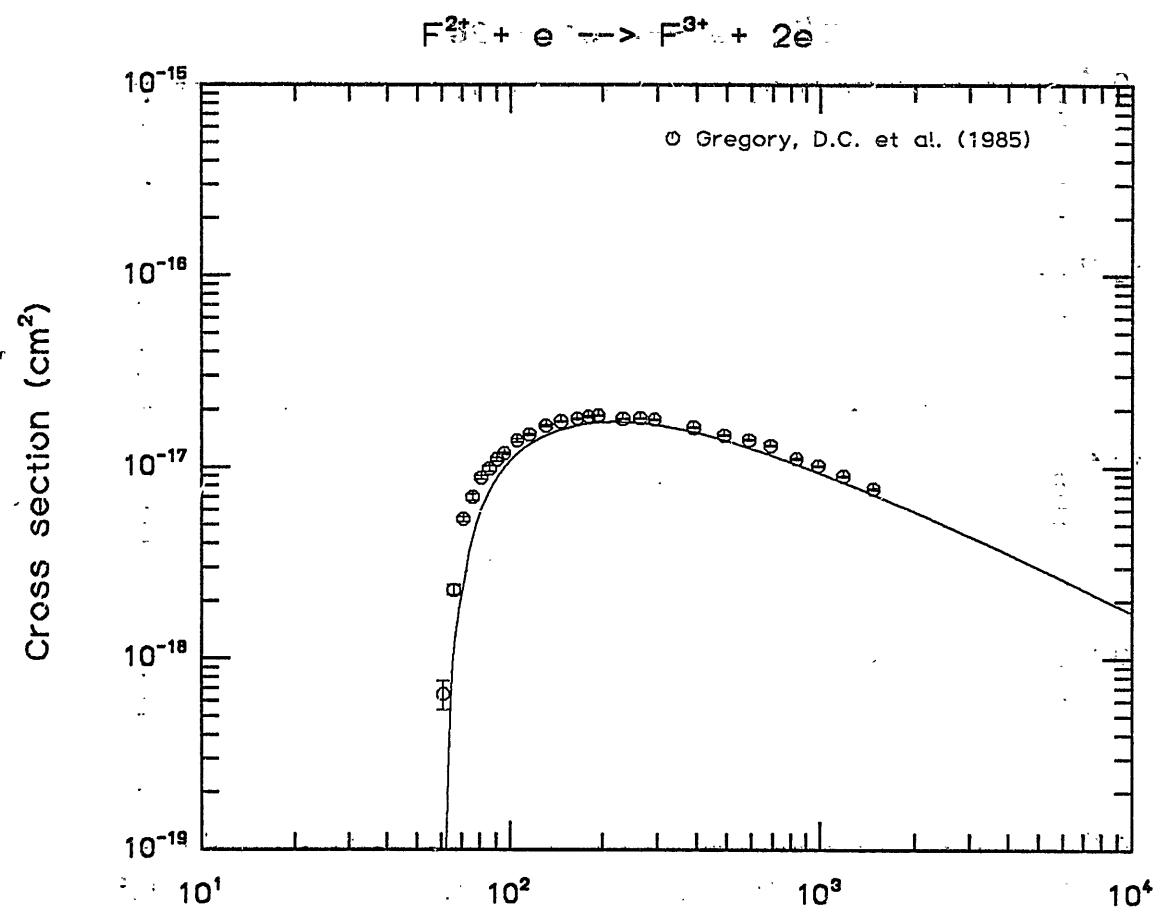


Fig. 55

Electron energy (eV)



**Fig. 56** Cross section vs Electron energy (eV)  $\text{F}^{2+} \rightarrow \text{F}^{3+}$

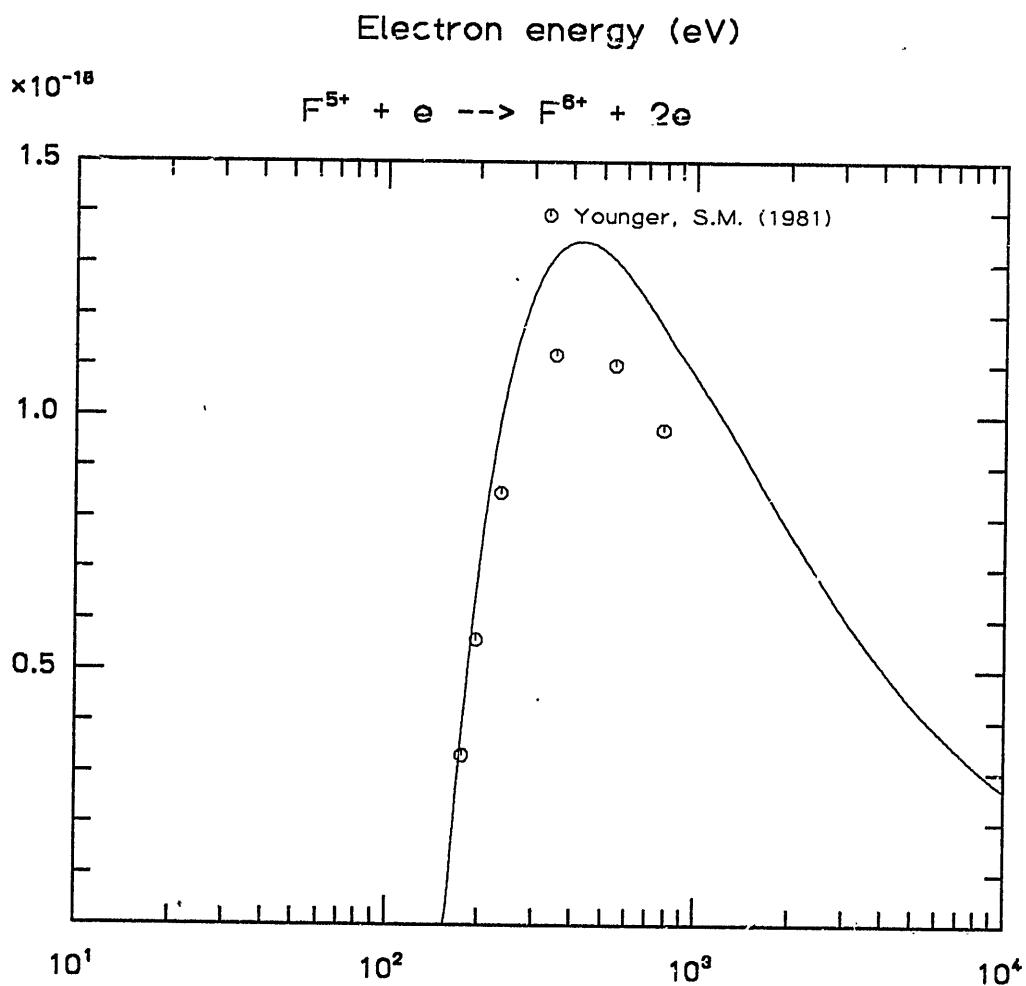
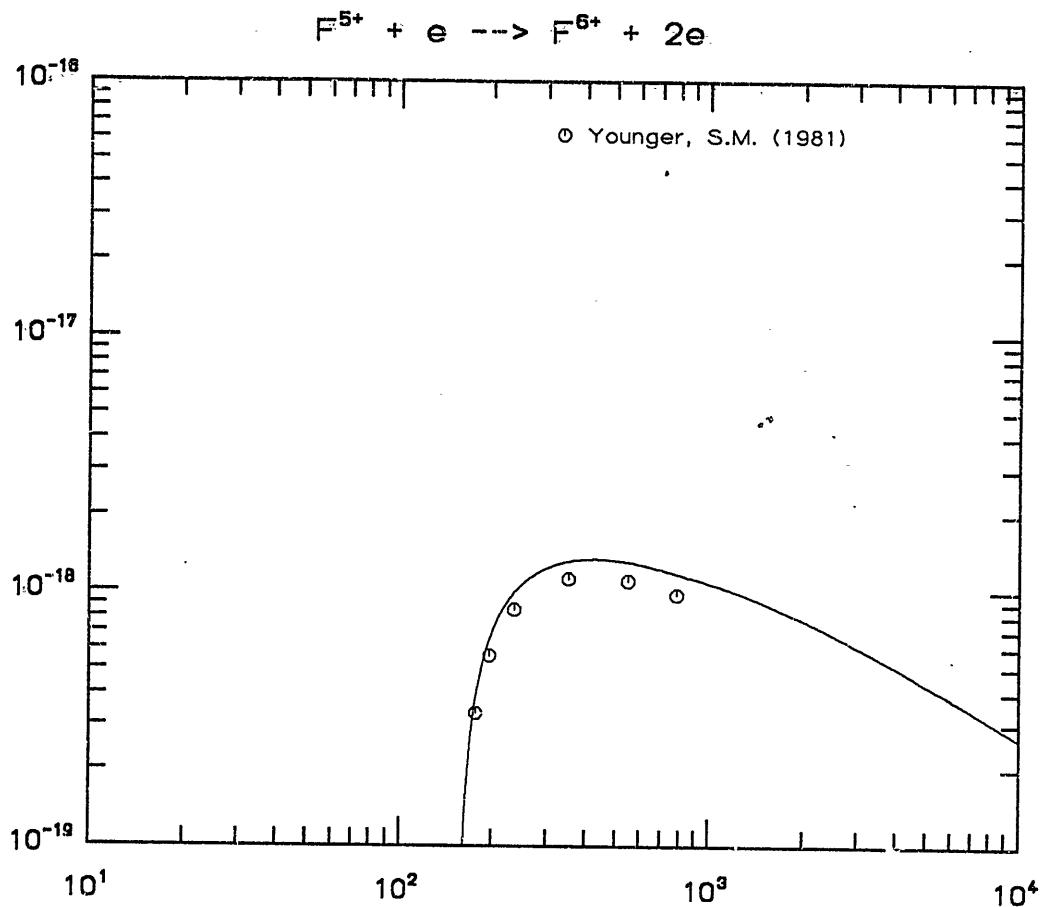


Fig. 57

Electron energy (eV)

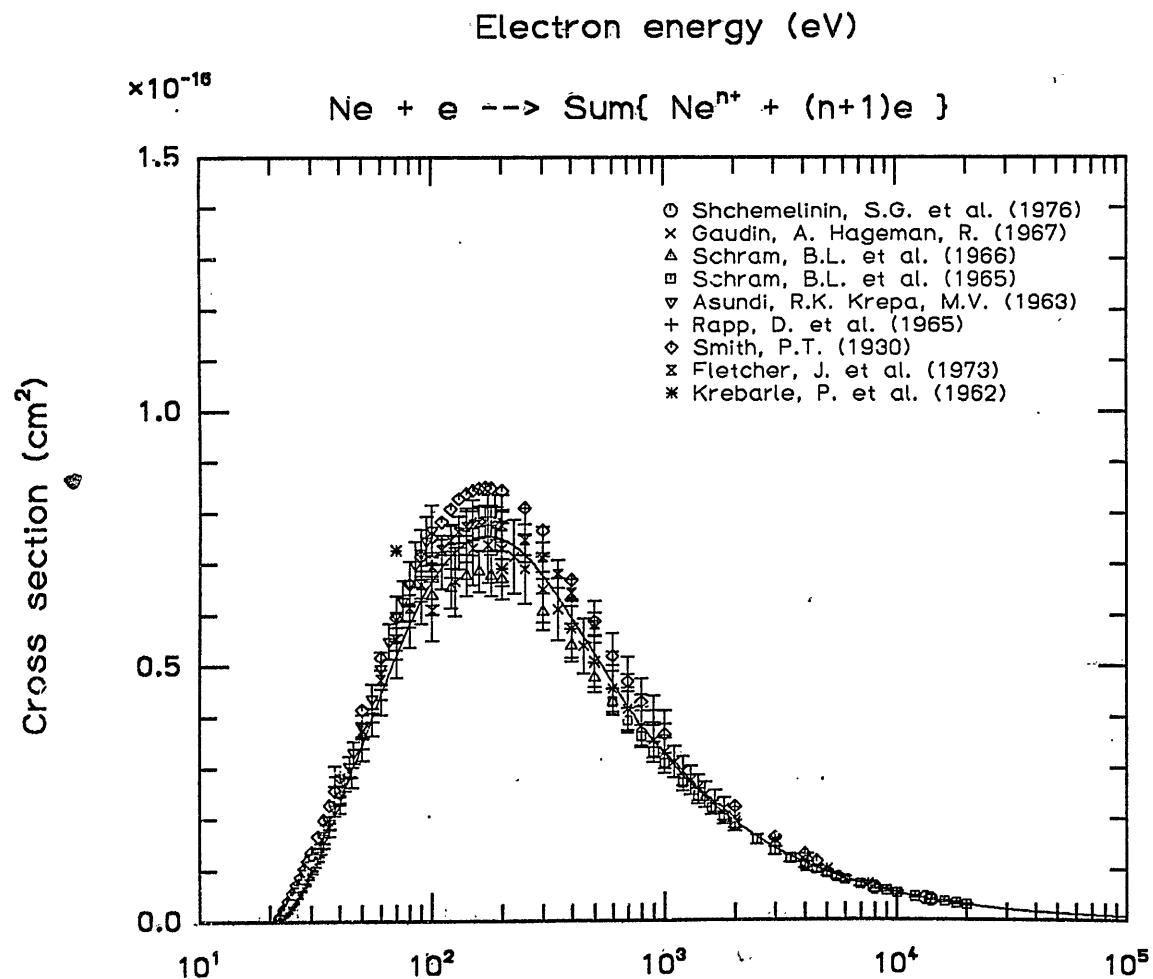
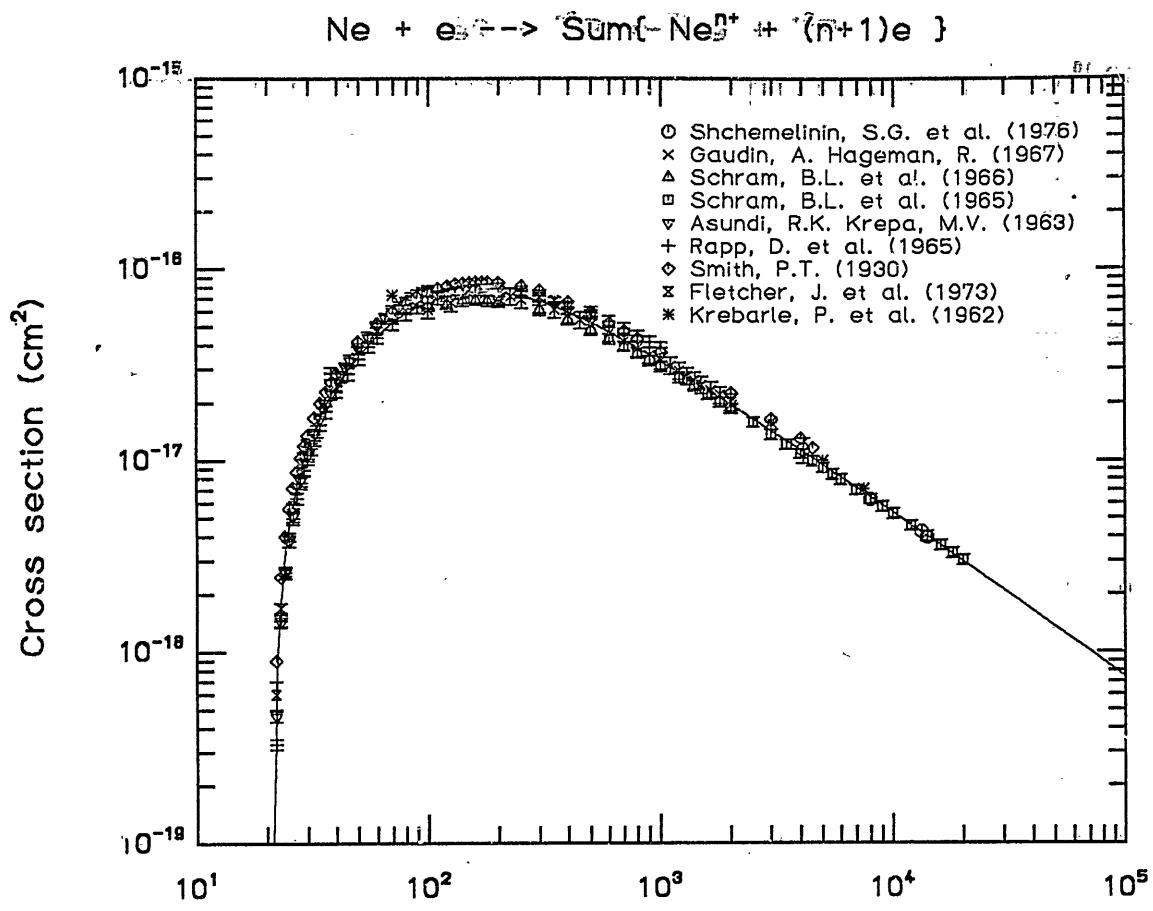
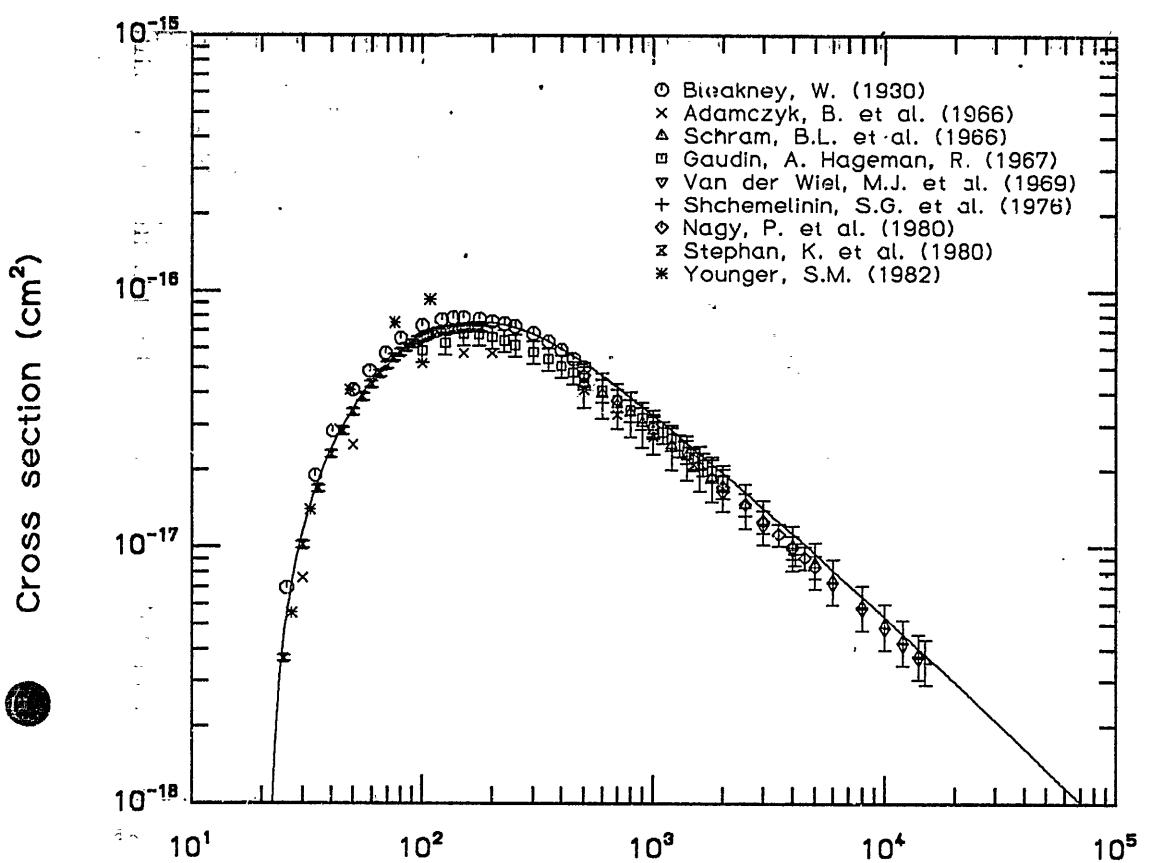
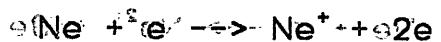


Fig. 58 (v) Electron energy(eV)

VC DIC



Electron energy (eV)

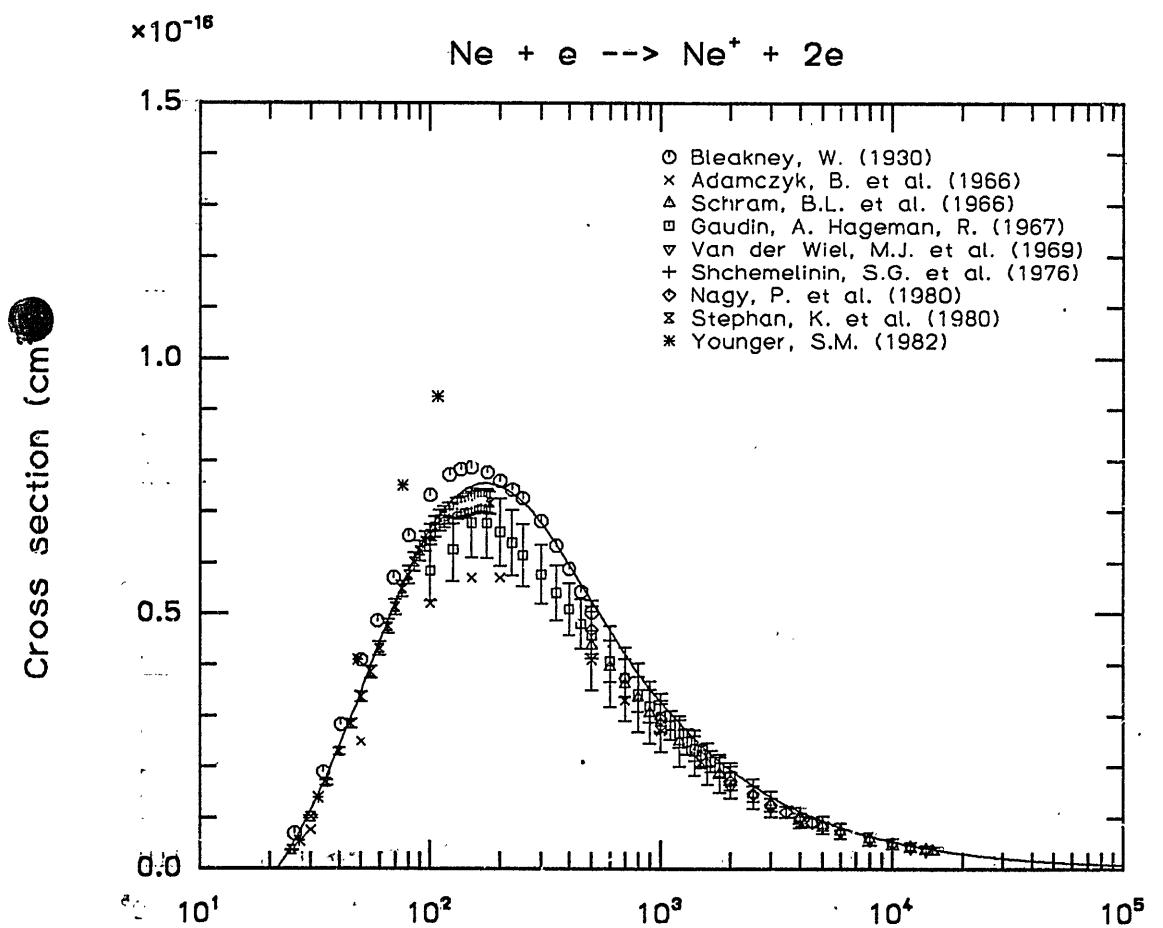


Fig. 59      Electron energy (eV)

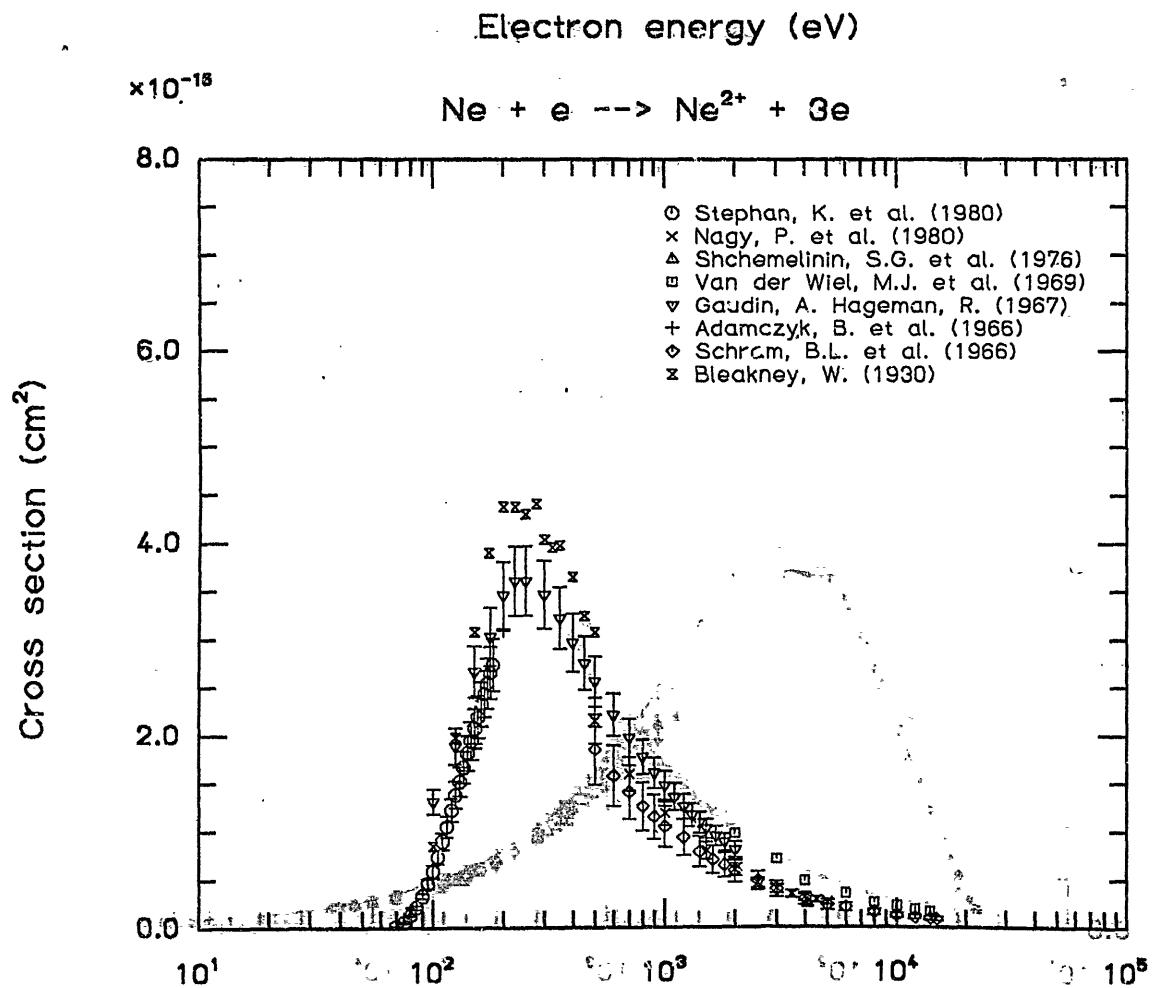
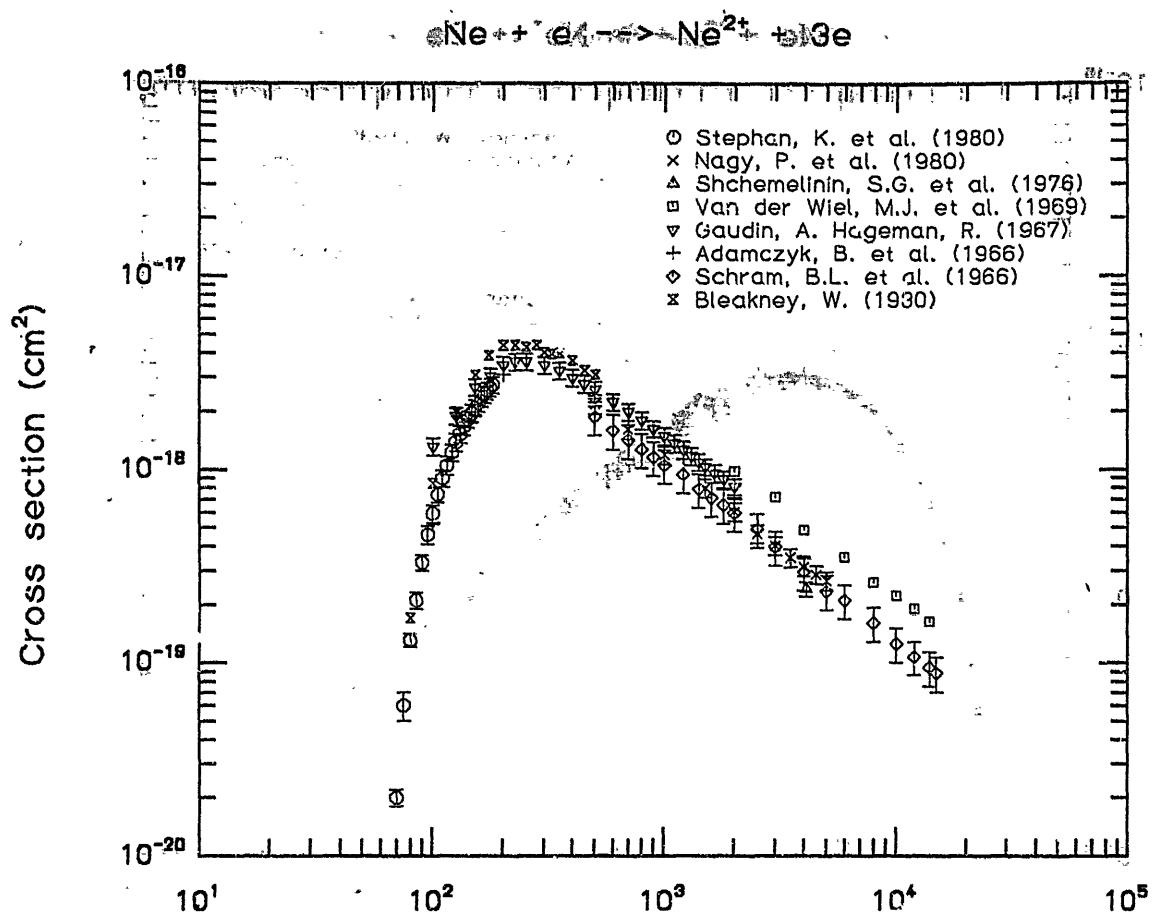
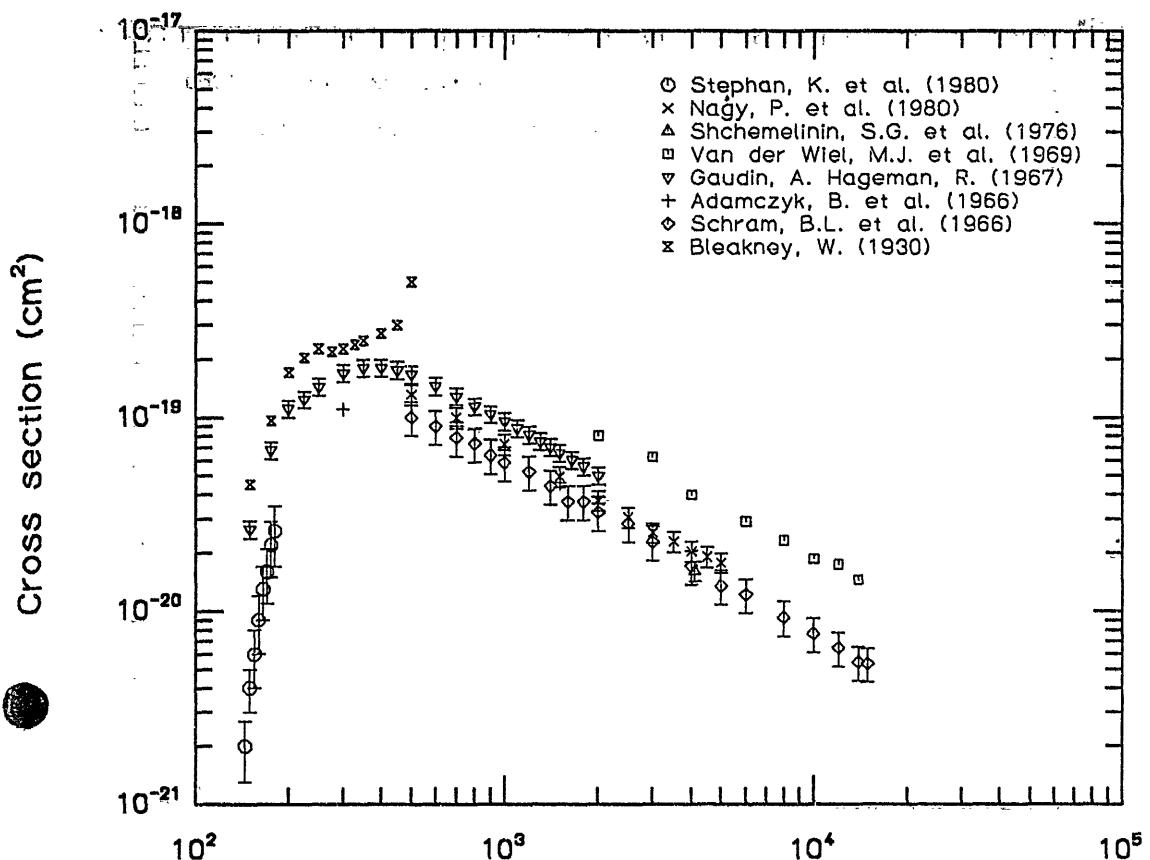


Fig. 60 (v) Electron energy (eV) Fig. 61



Electron energy (eV)

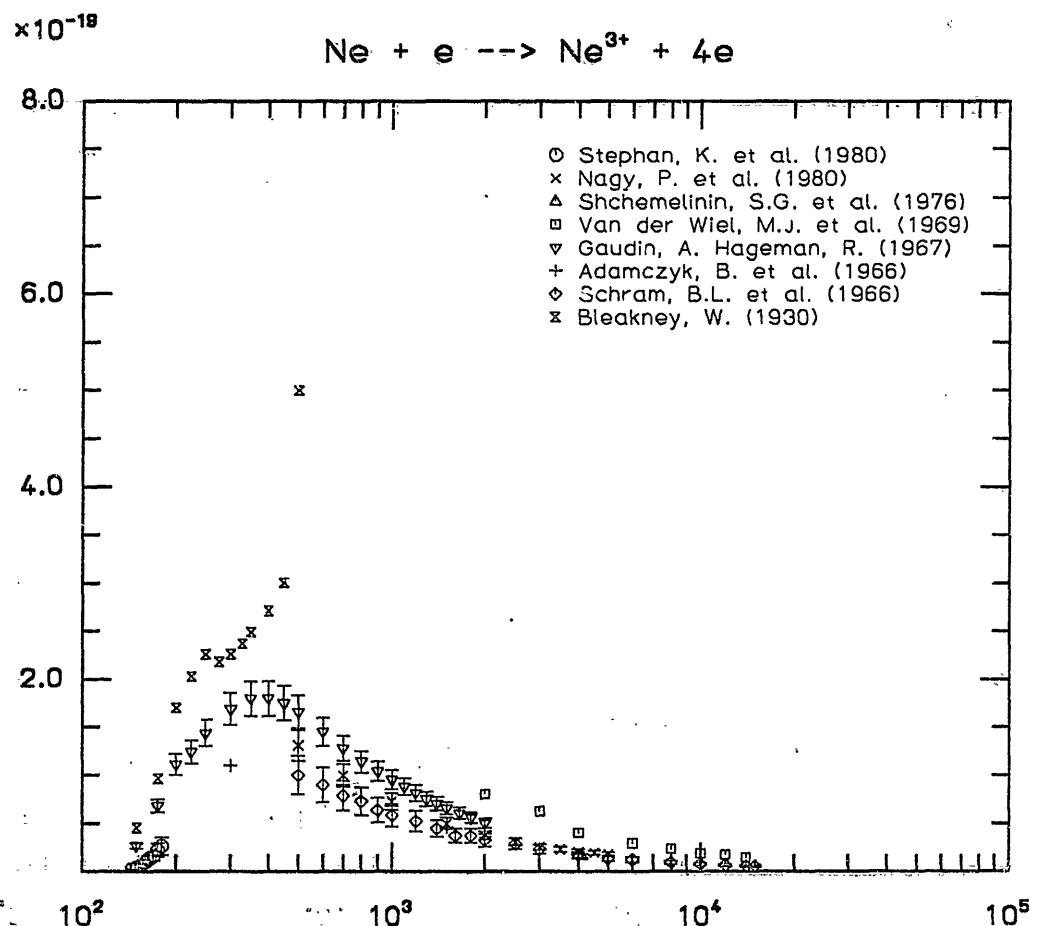


Fig. 61 (a) Electron energy (eV) 50-10<sup>5</sup>

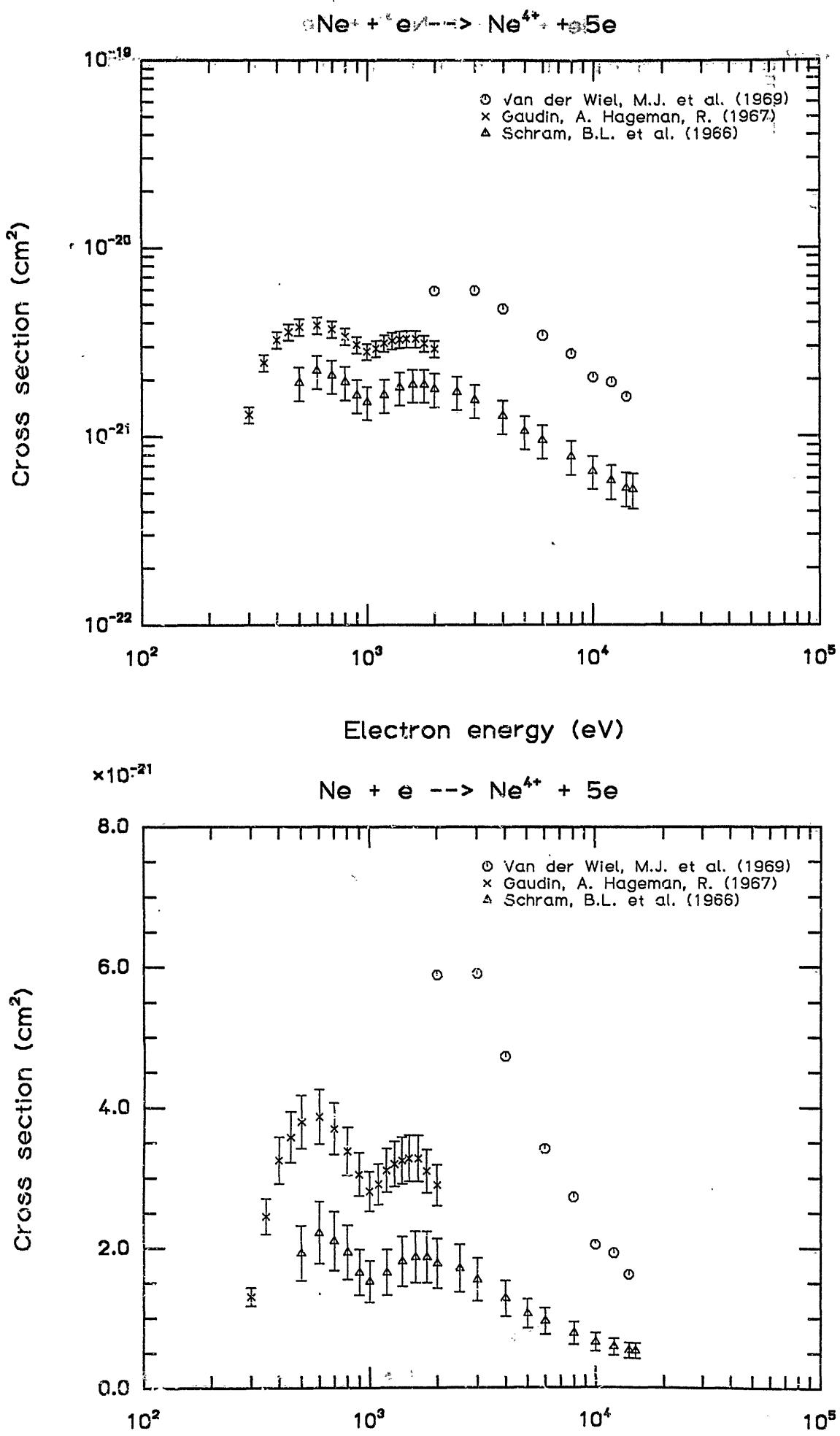


Fig. 62

Electron energy (eV)

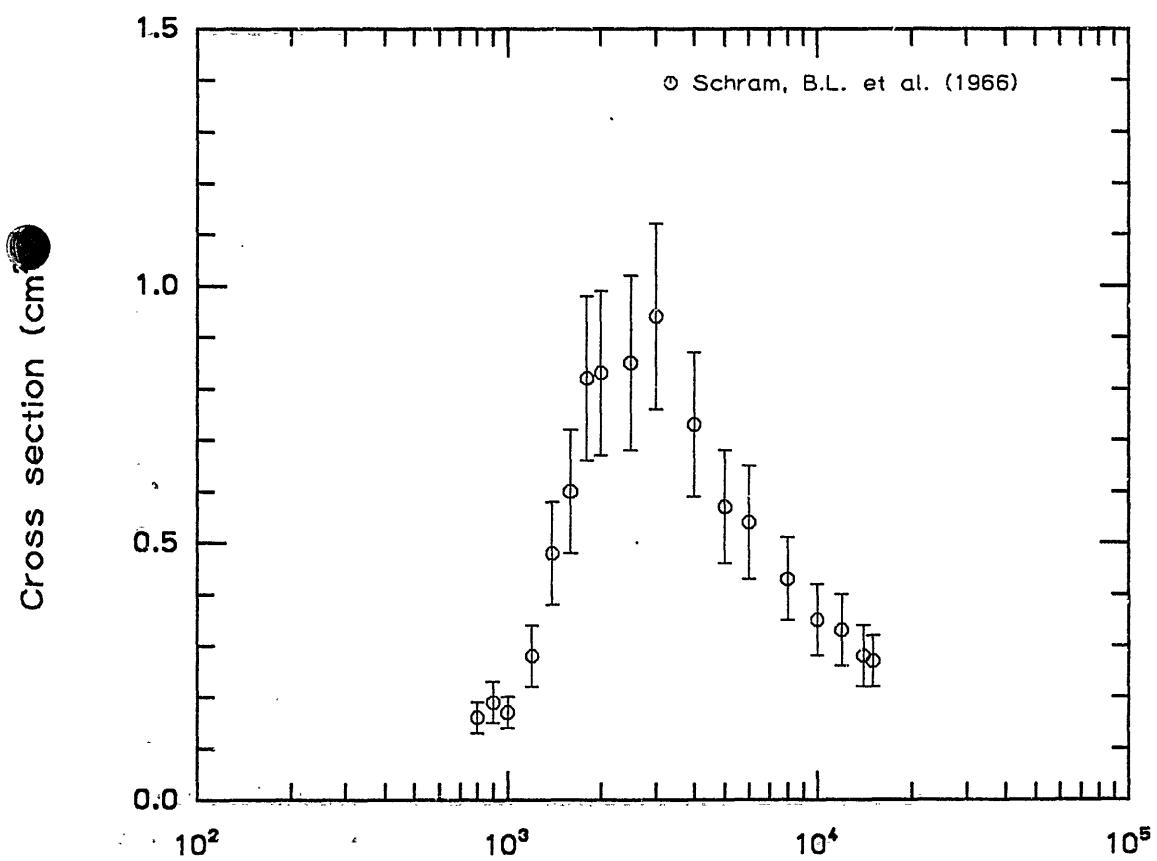
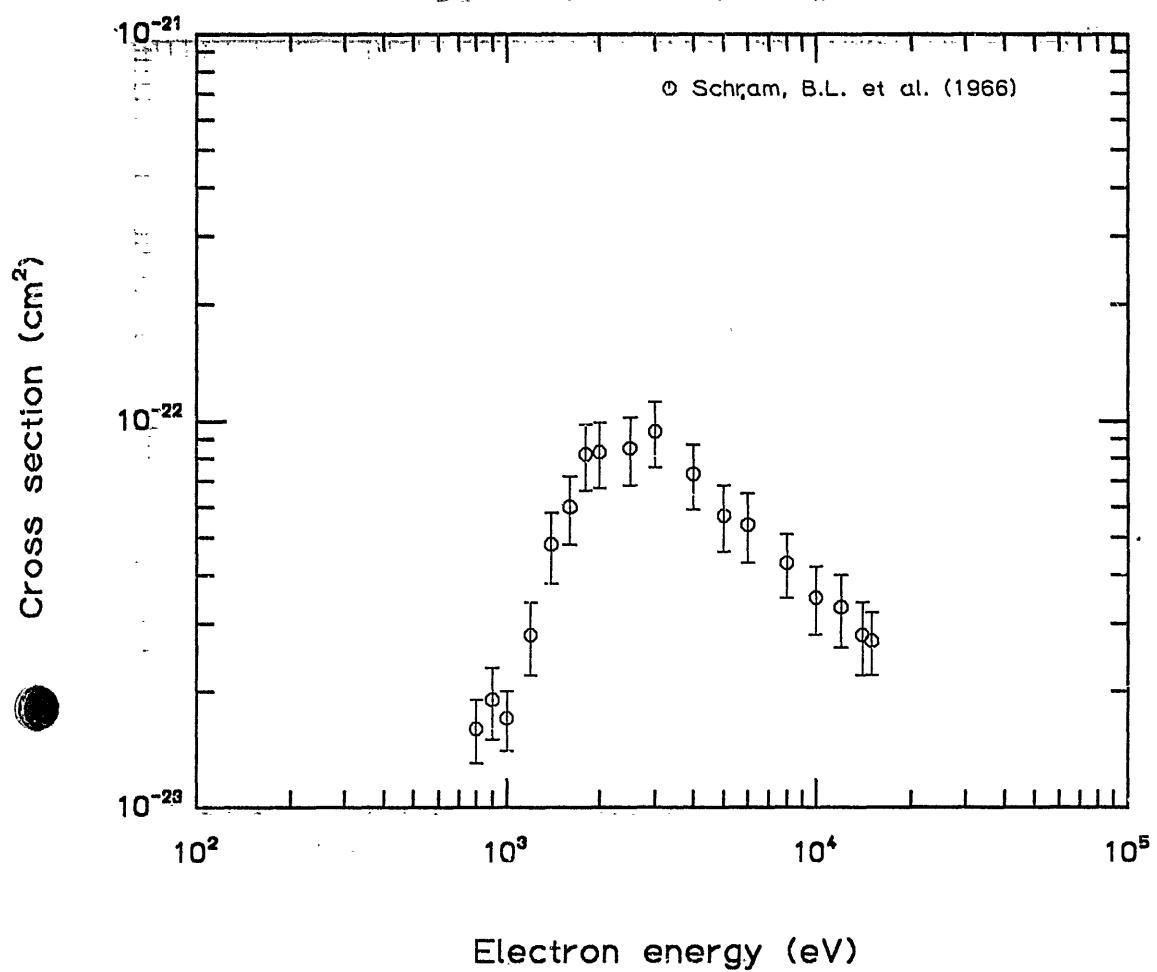
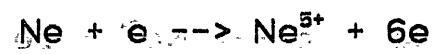


Fig. 63      Electron energy (eV)

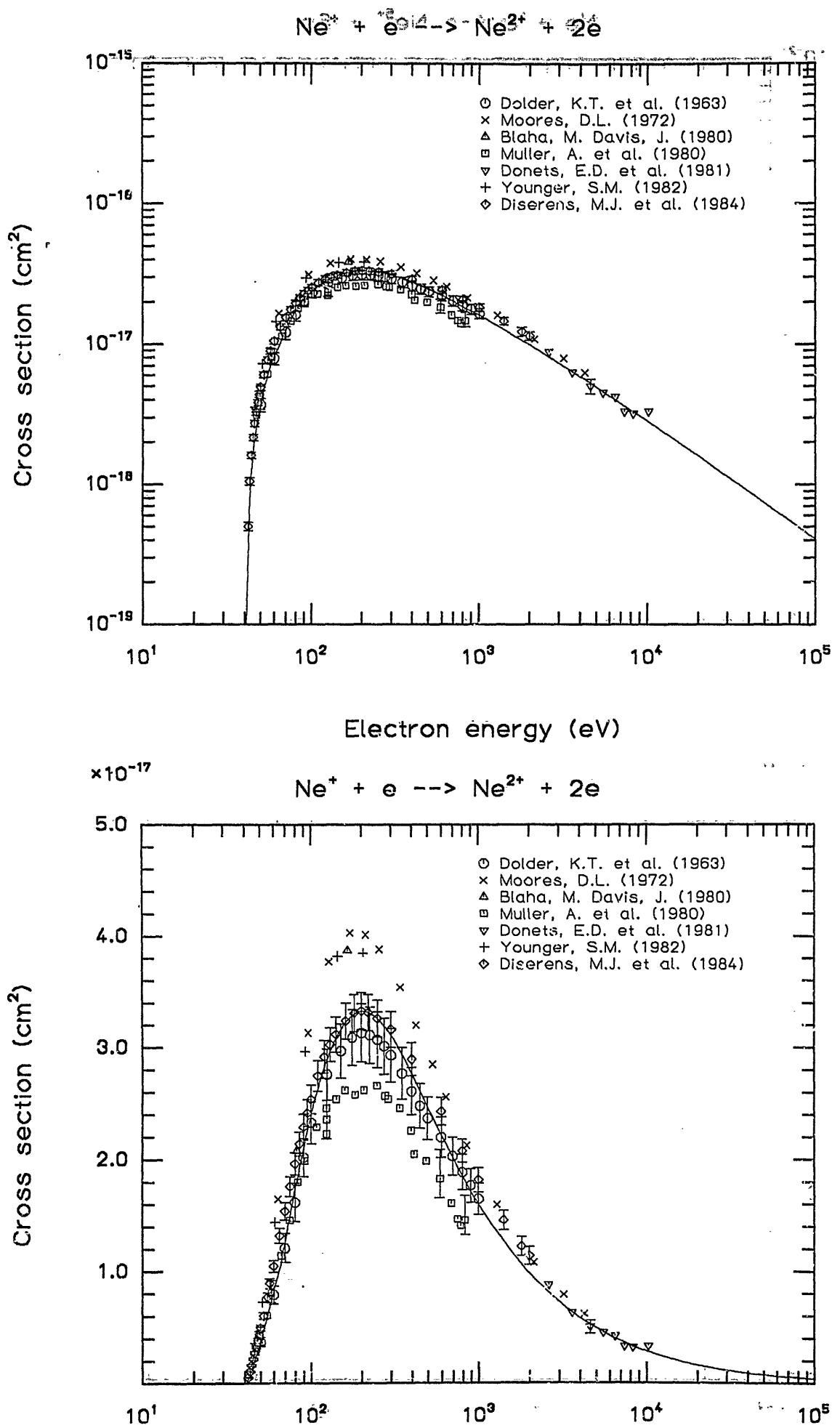


Fig. 64

Electron energy (eV)

88 p.F

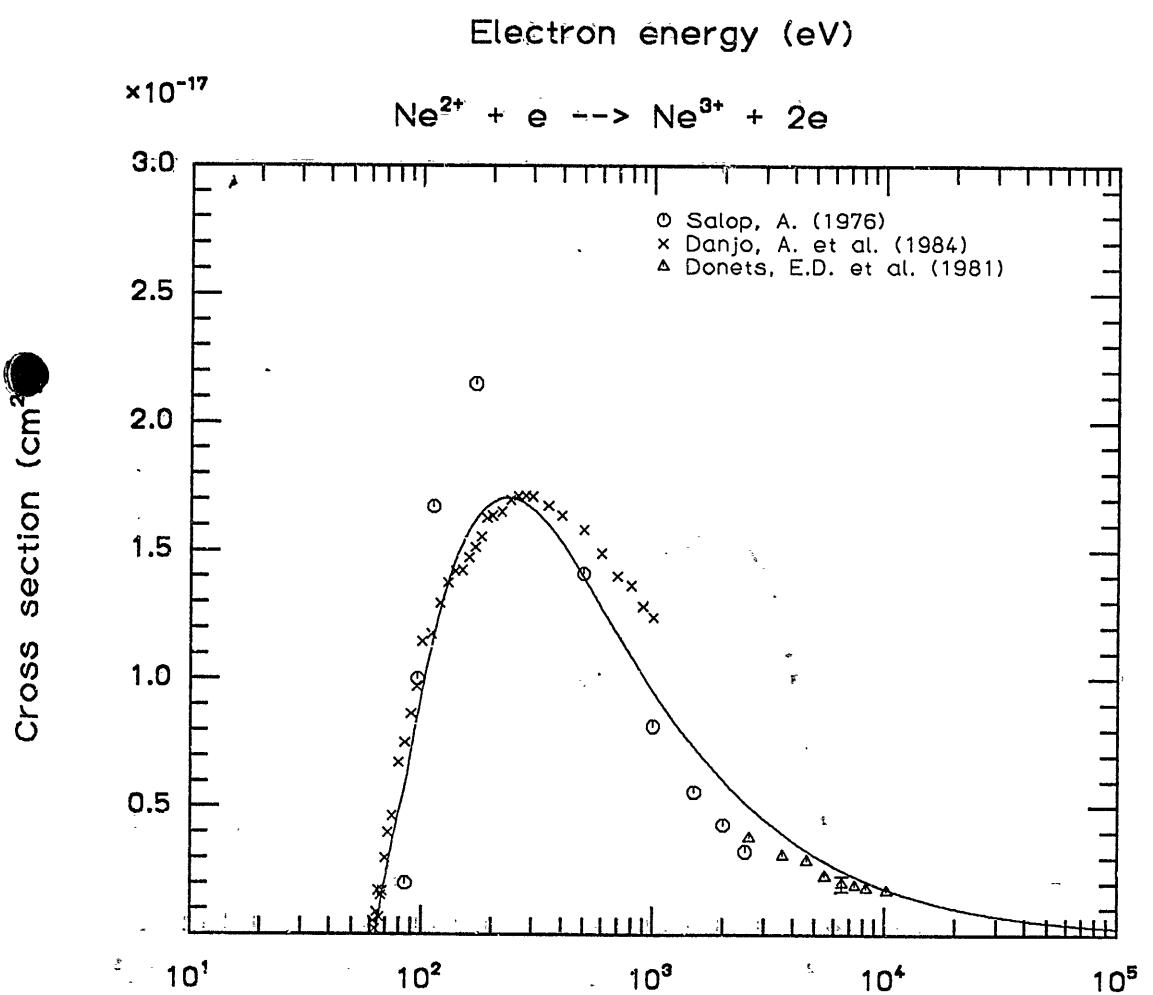
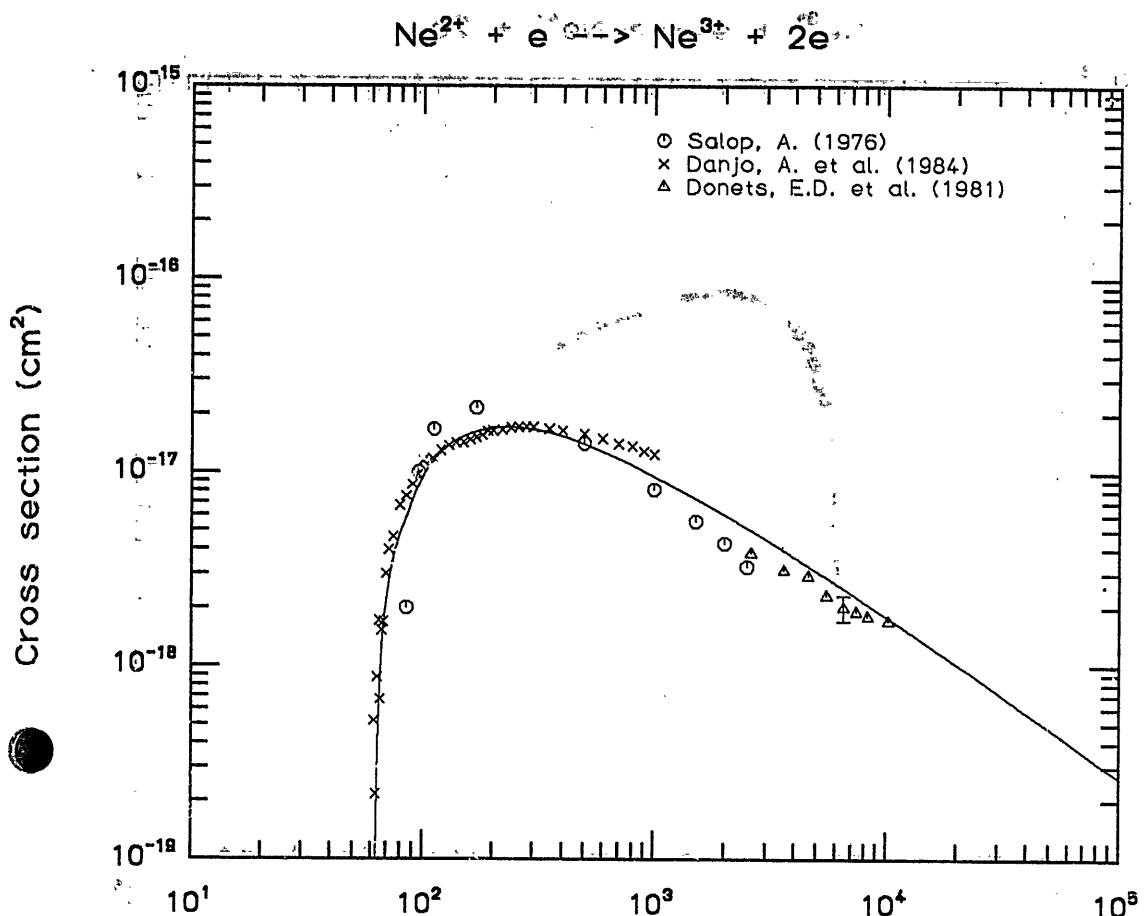


Fig. 65

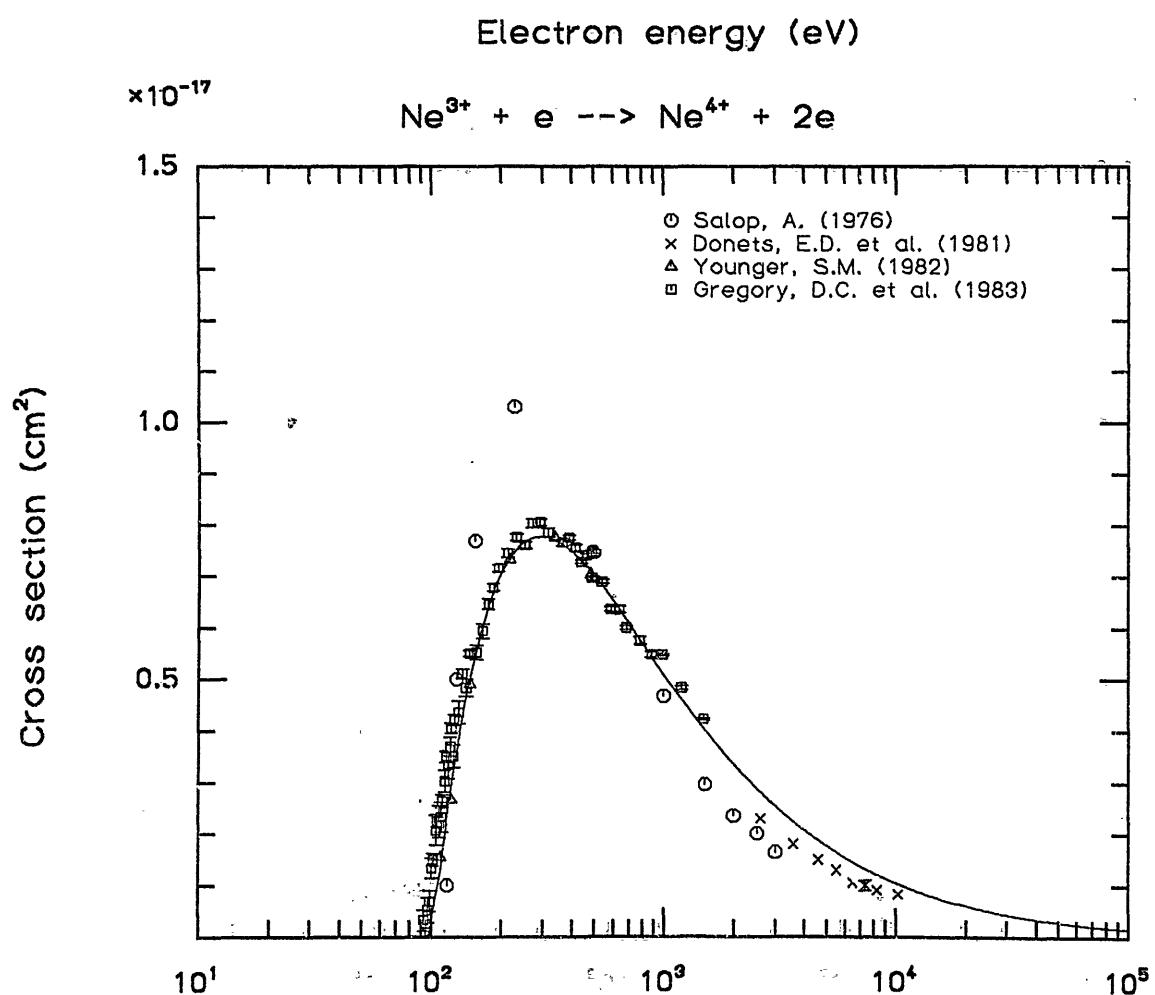
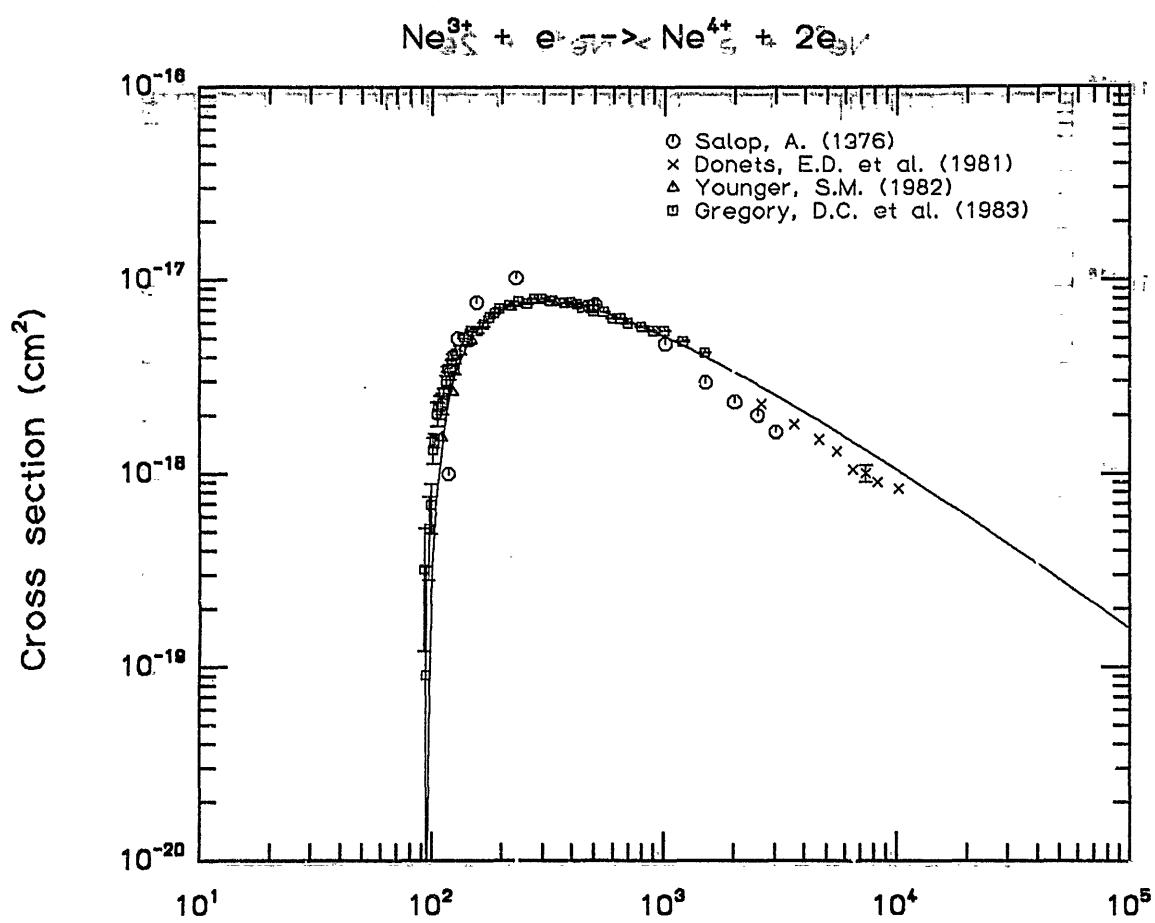
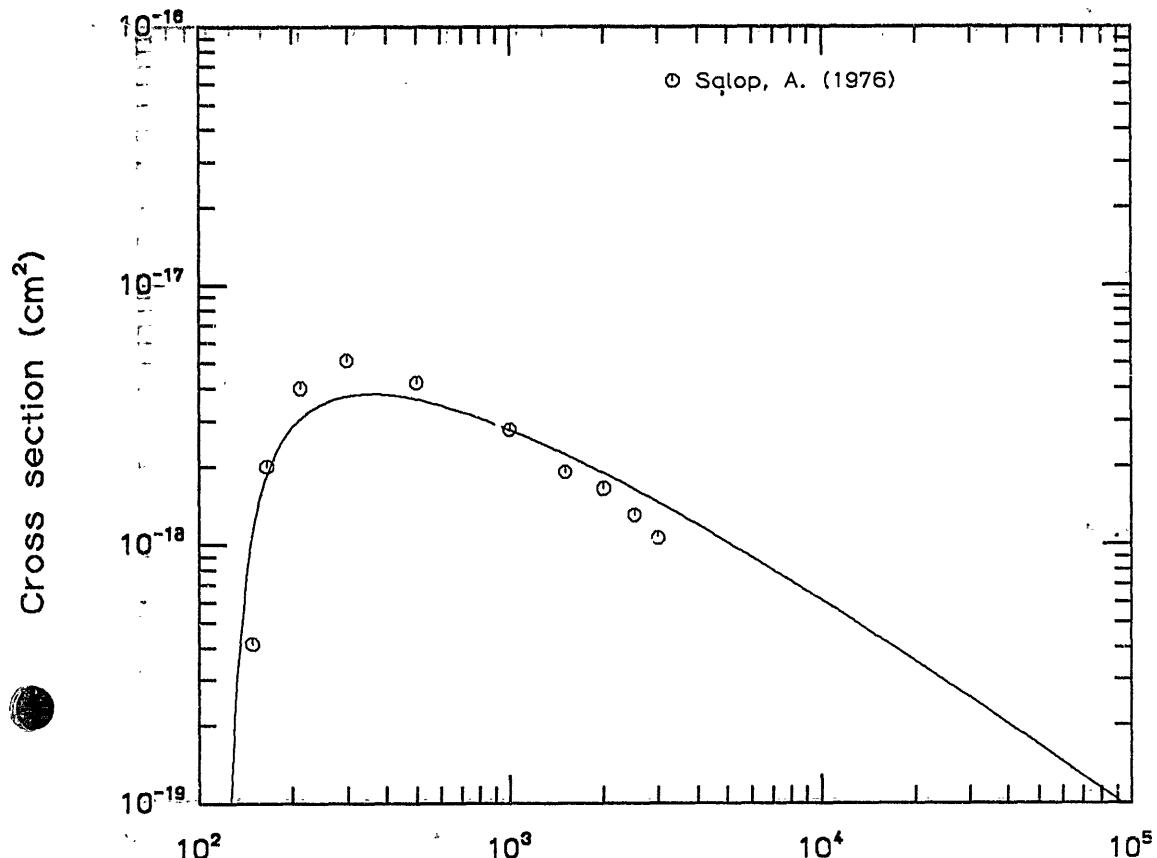
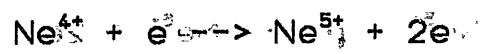


Fig. 66 (Ne) Electron energy (eV) 20-10<sup>5</sup>



Electron energy (eV)

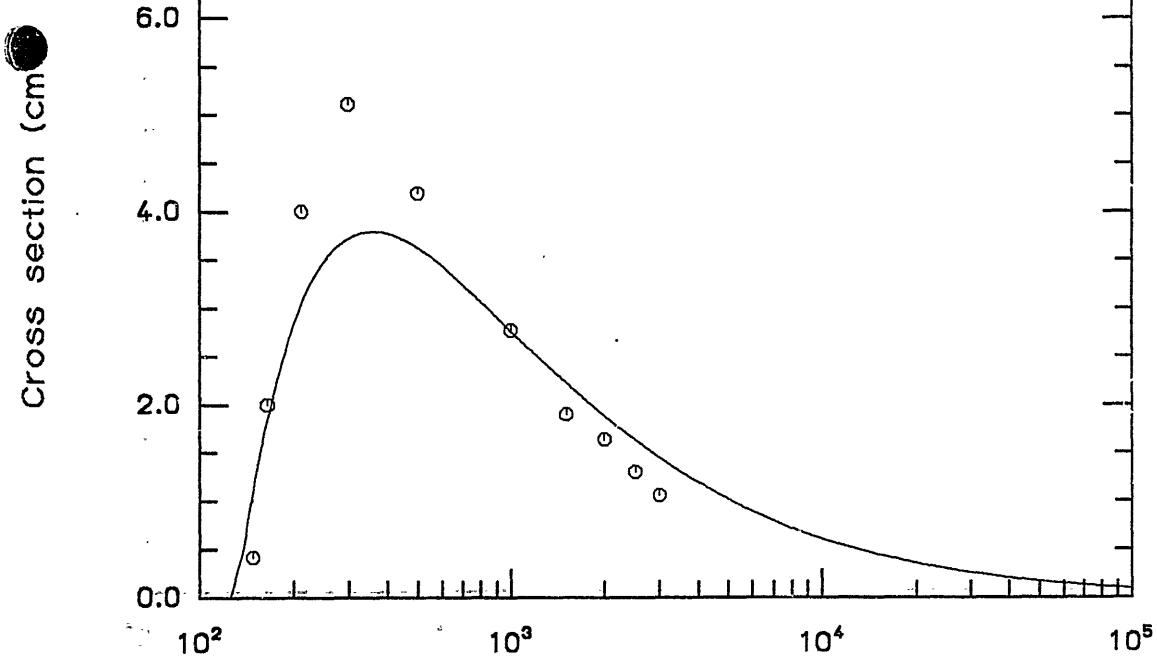
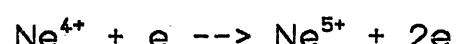


Fig. 67

Electron energy (eV)

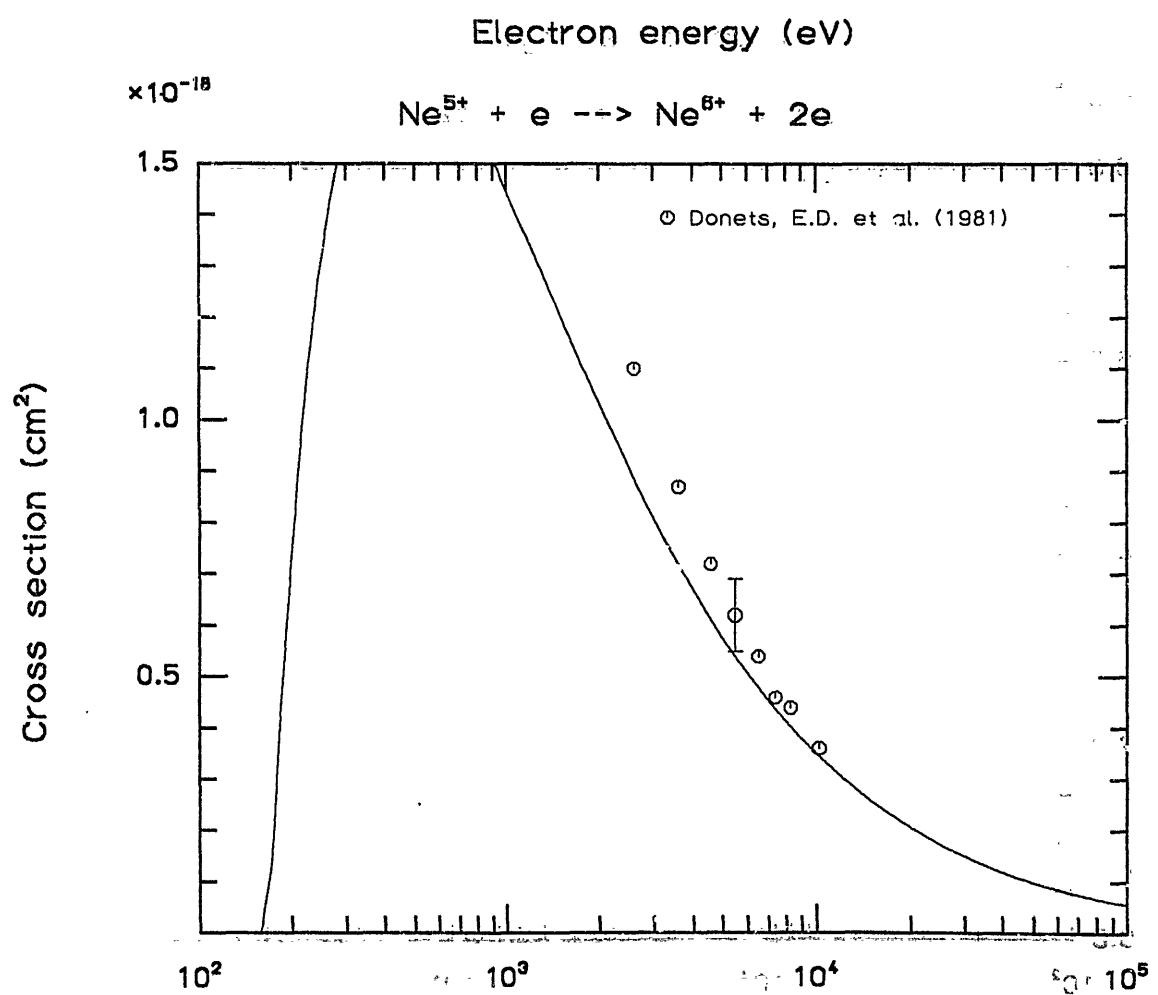
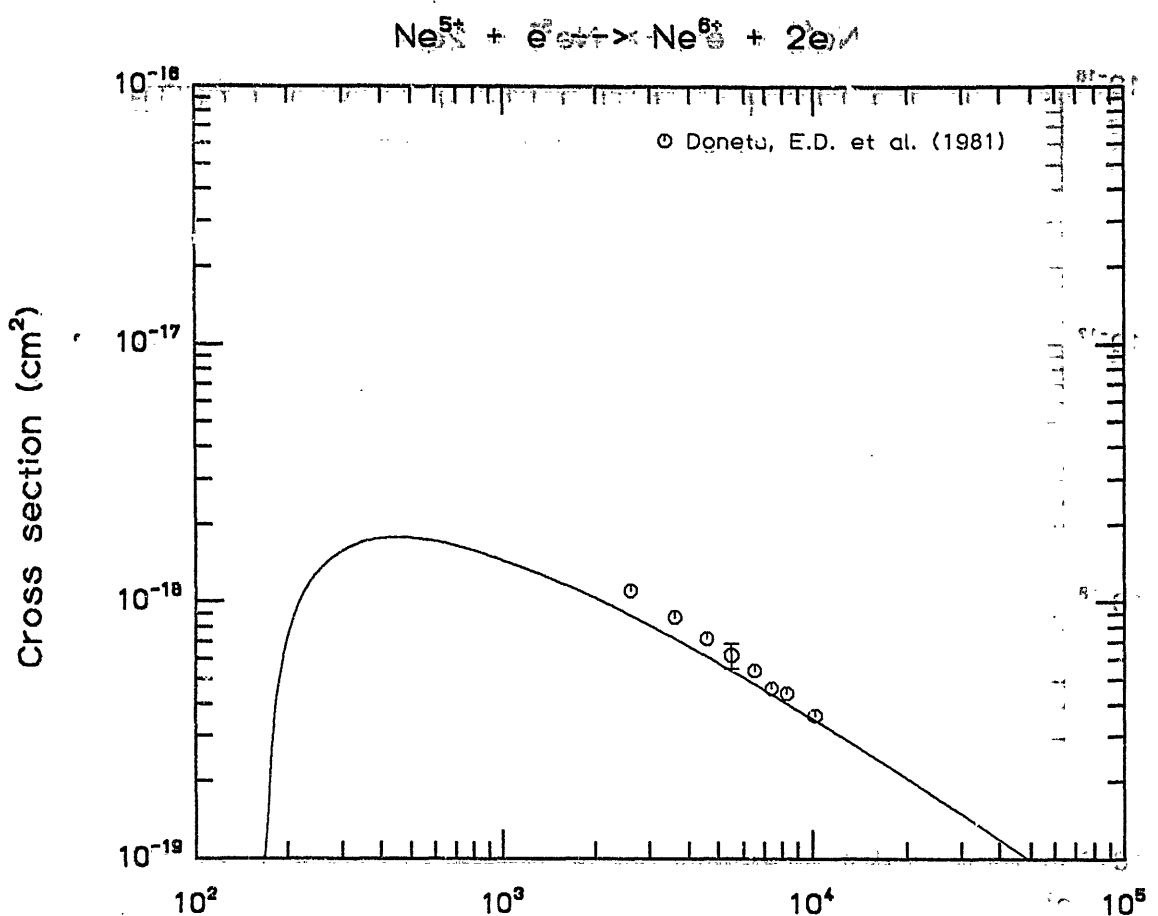


Fig. 68       $\text{Ne}^{5+} + e^- \rightarrow \text{Ne}^{6+} + 2e^-$

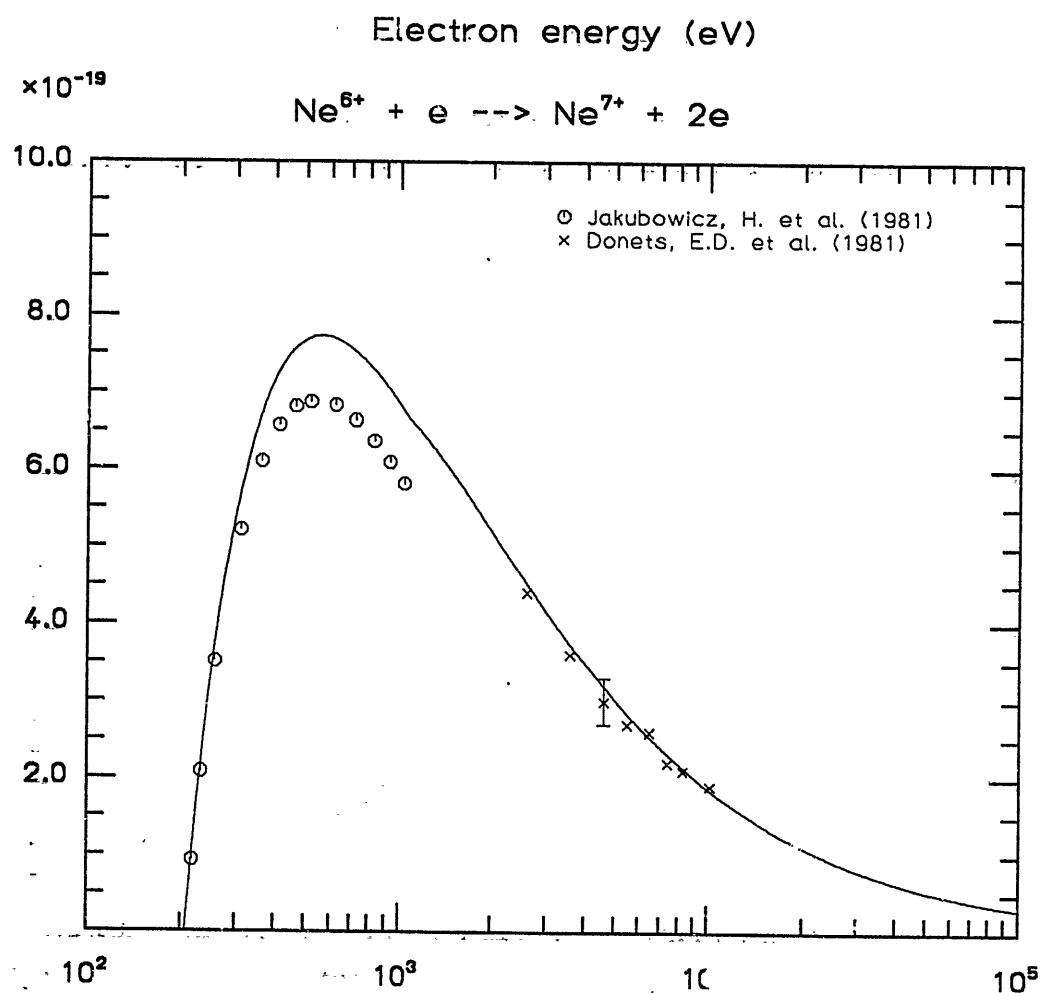
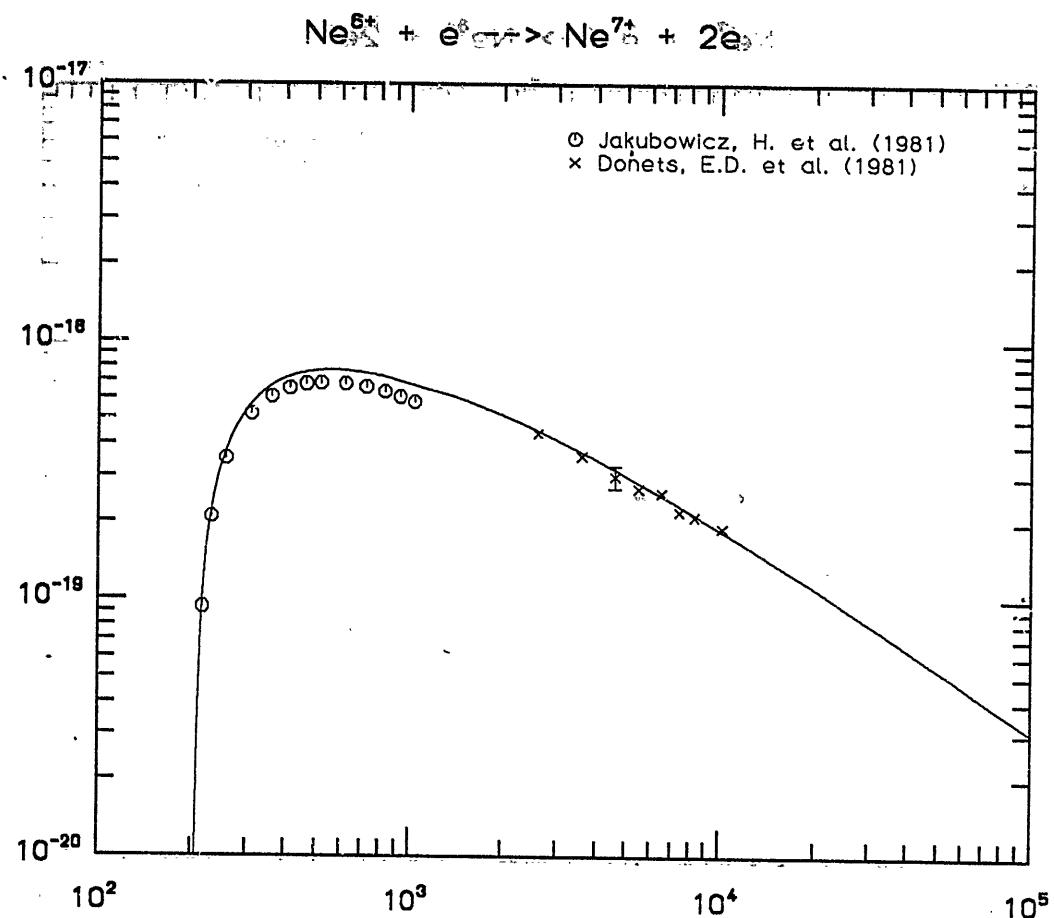


Fig. 69 (a) Electron energy (eV)

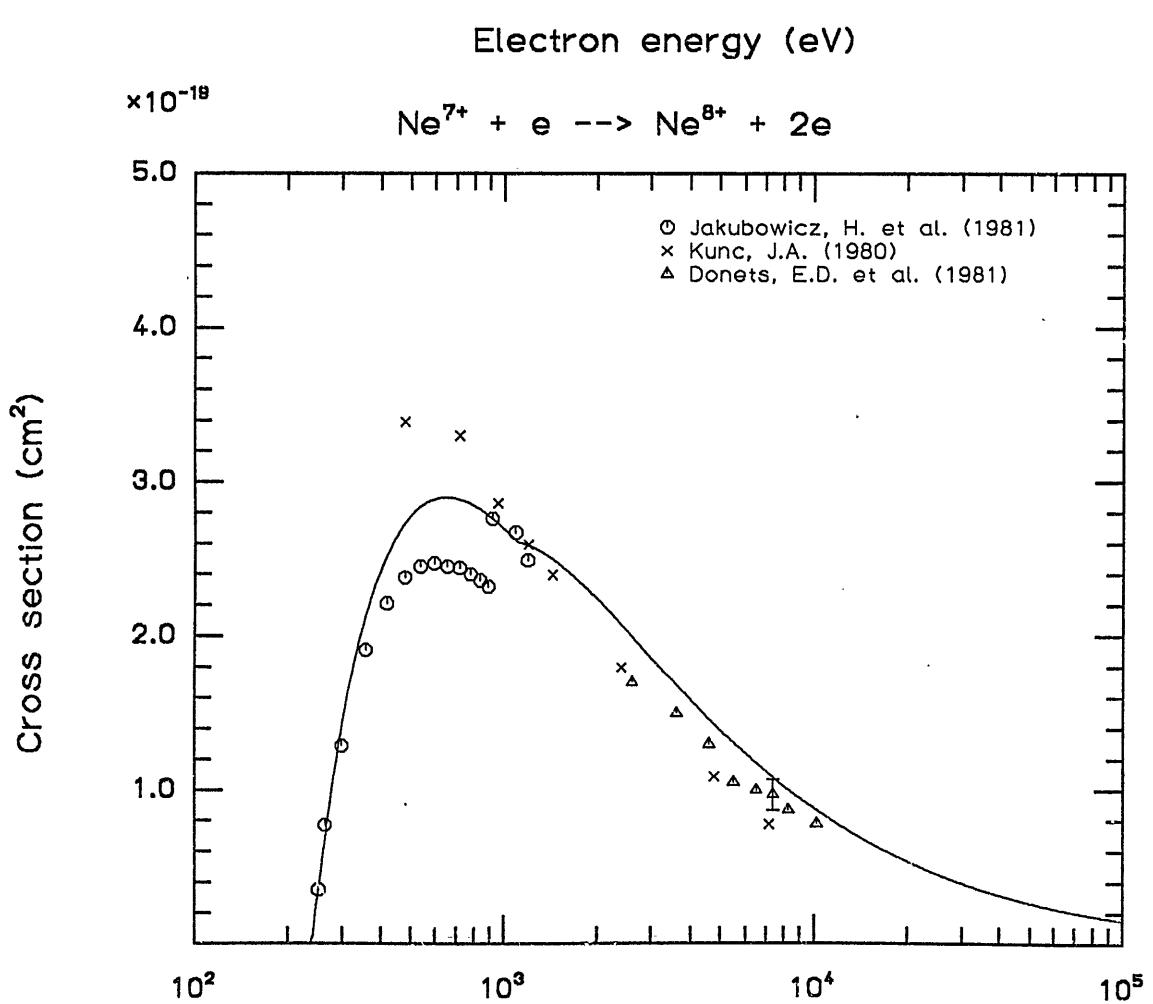
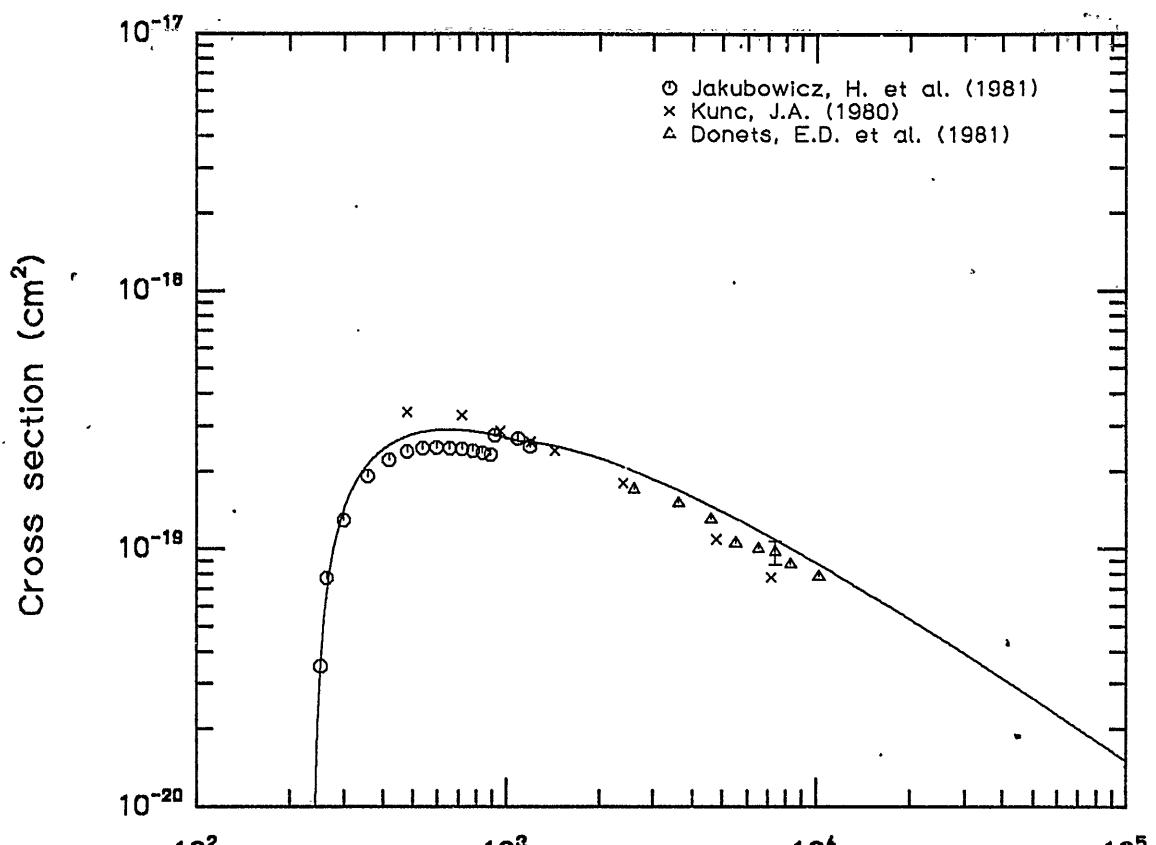
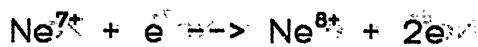


Fig. 70

Electron energy (eV)

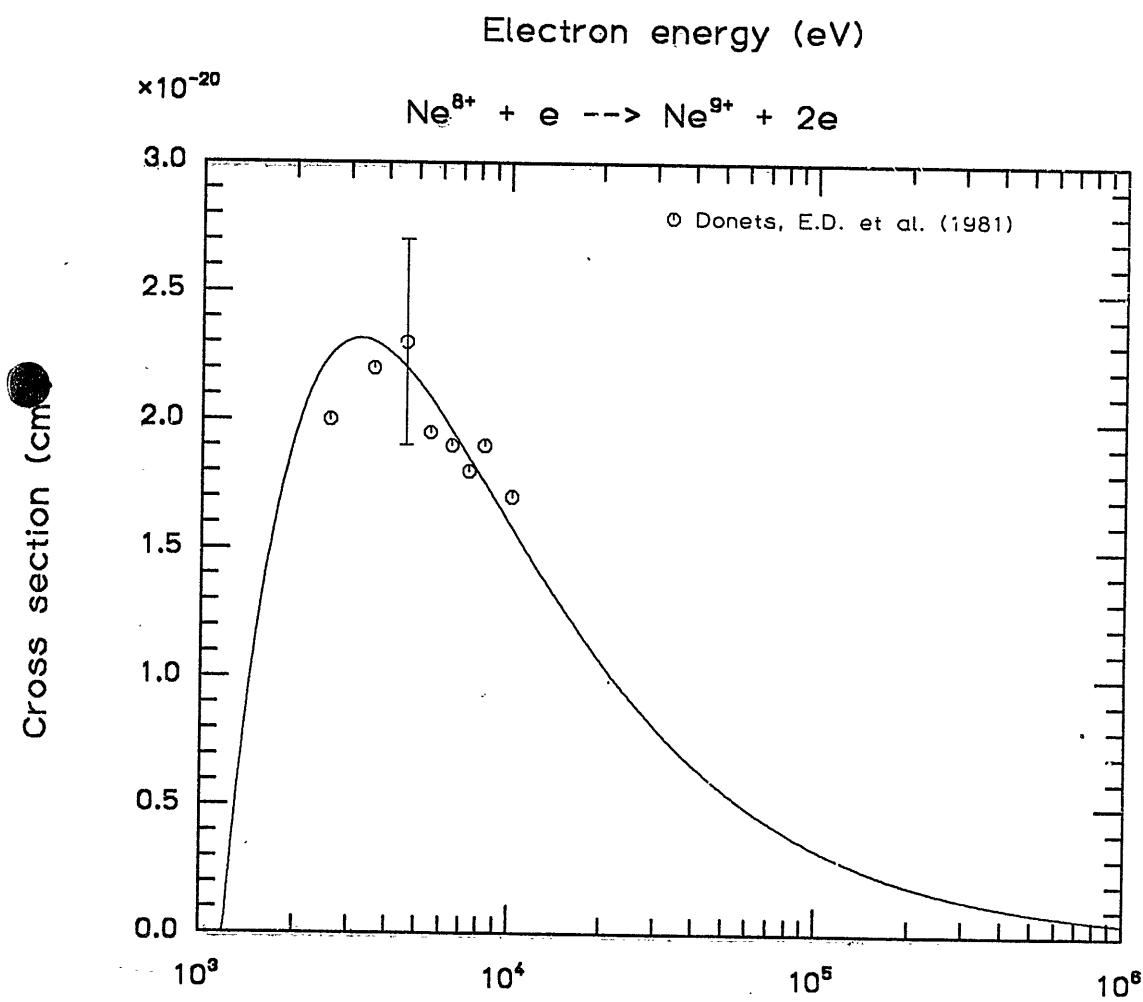
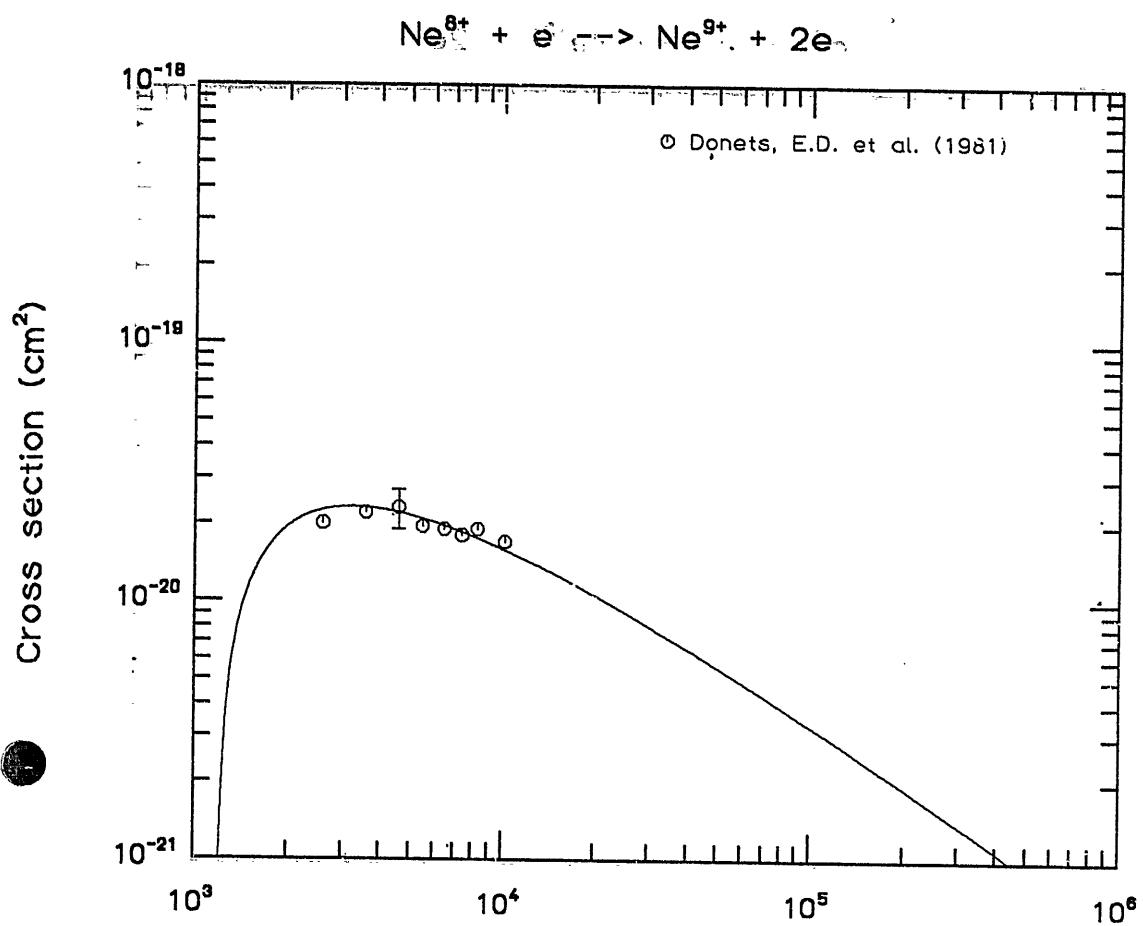


Fig. 71      Electron energy (eV)

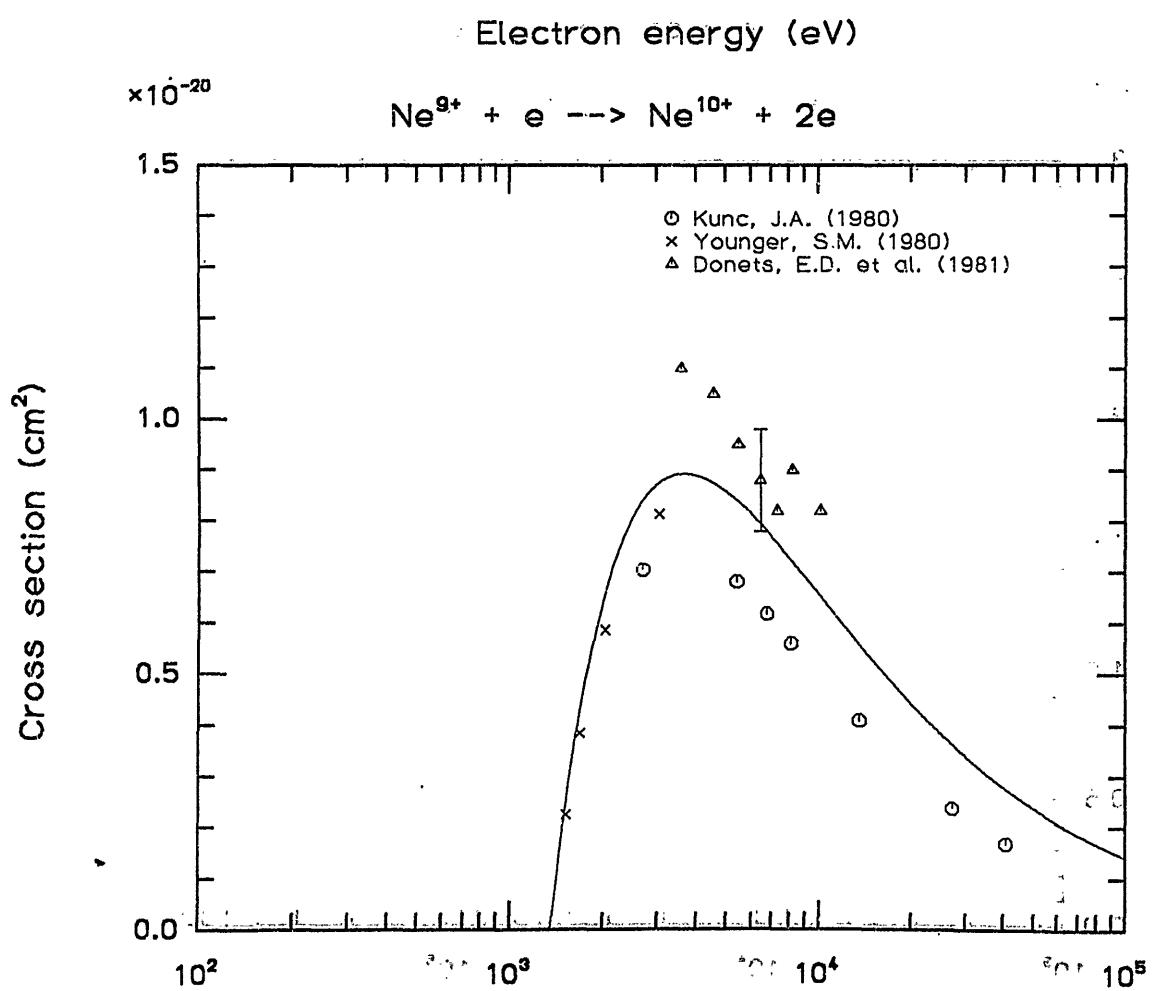
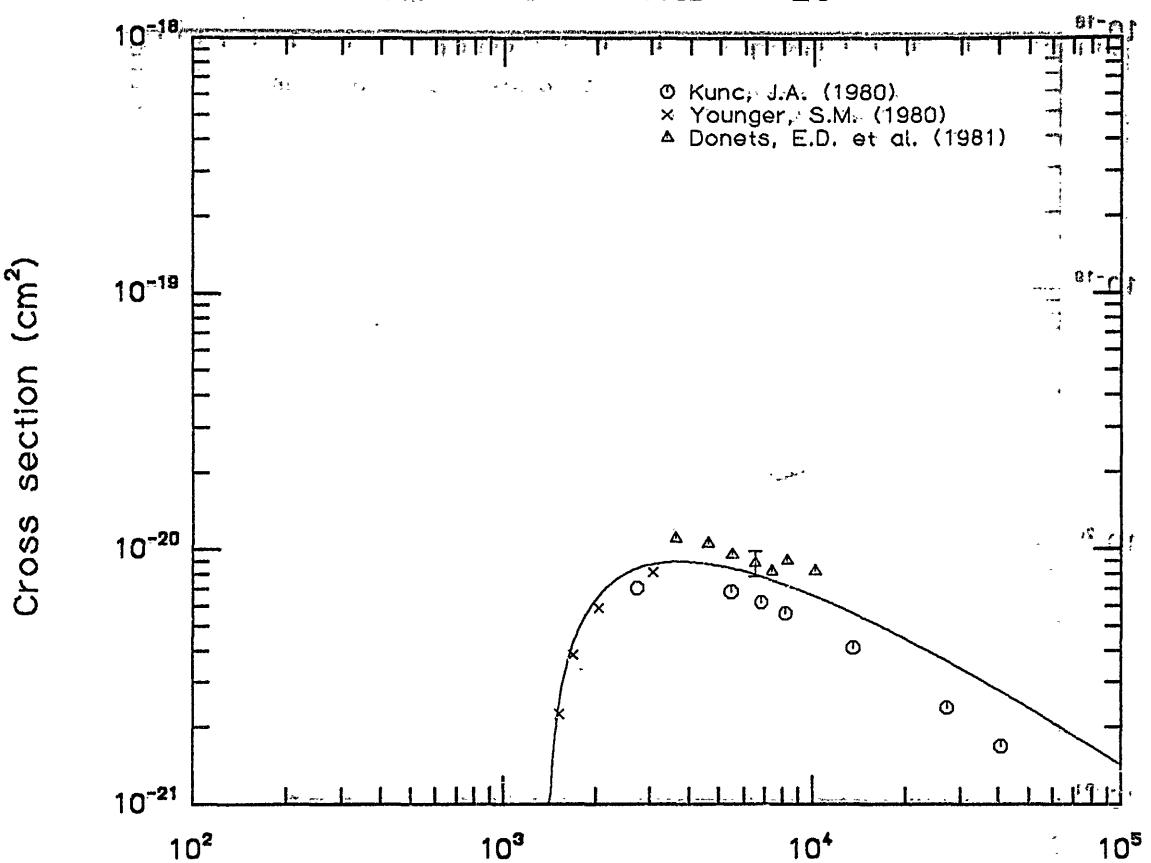
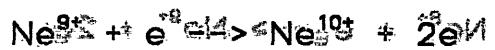
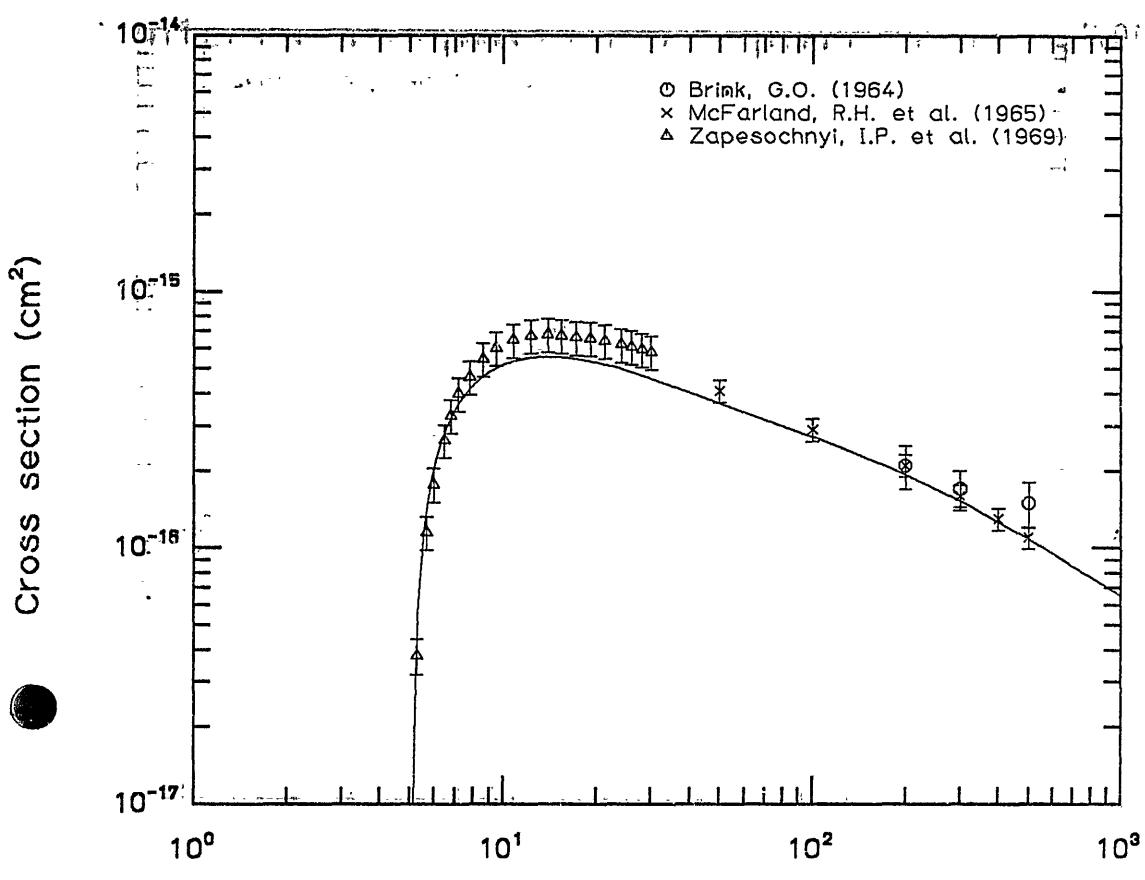
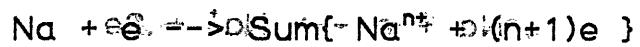


Fig. 72 (v) Electron energy (eV)



Electron energy (eV)

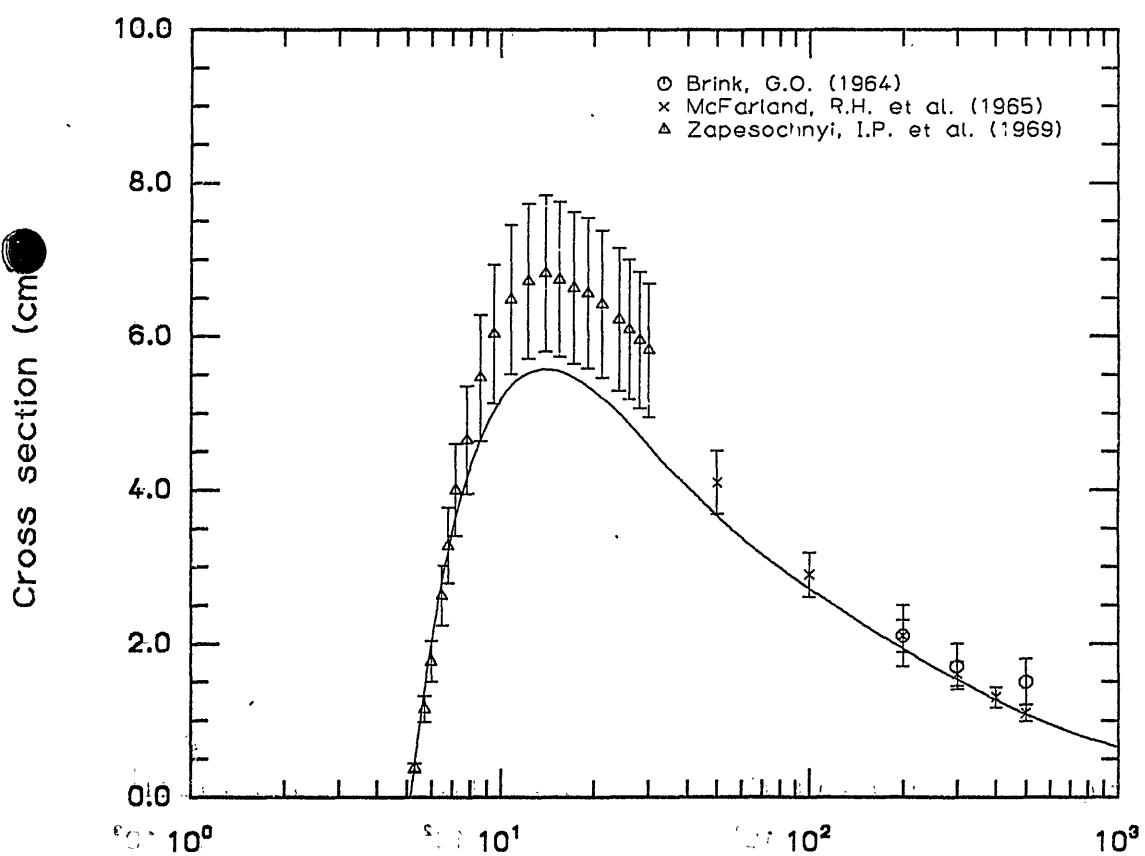
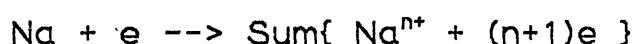


Fig. 73

Electron energy (eV)

$\Delta V = 0.1$

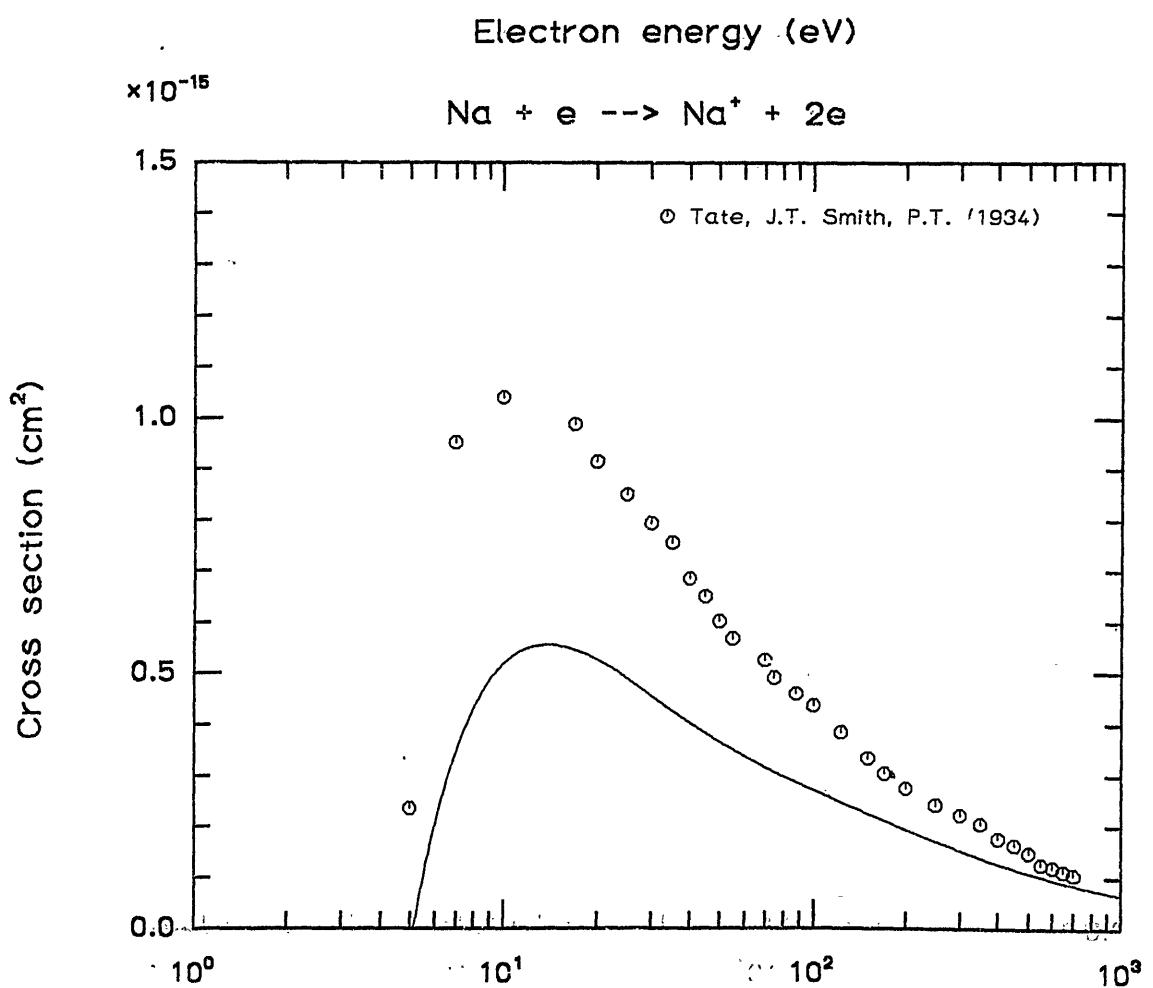
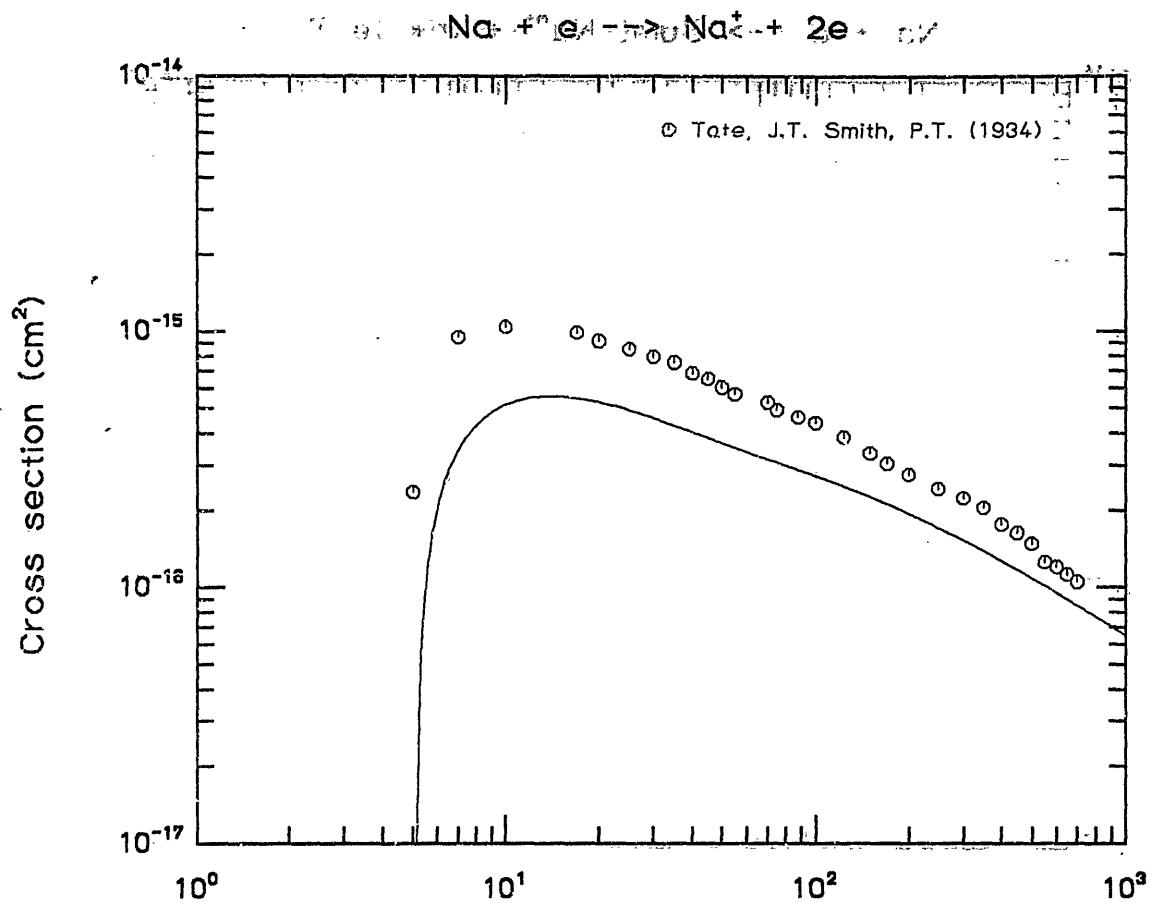
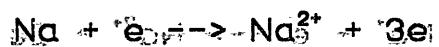
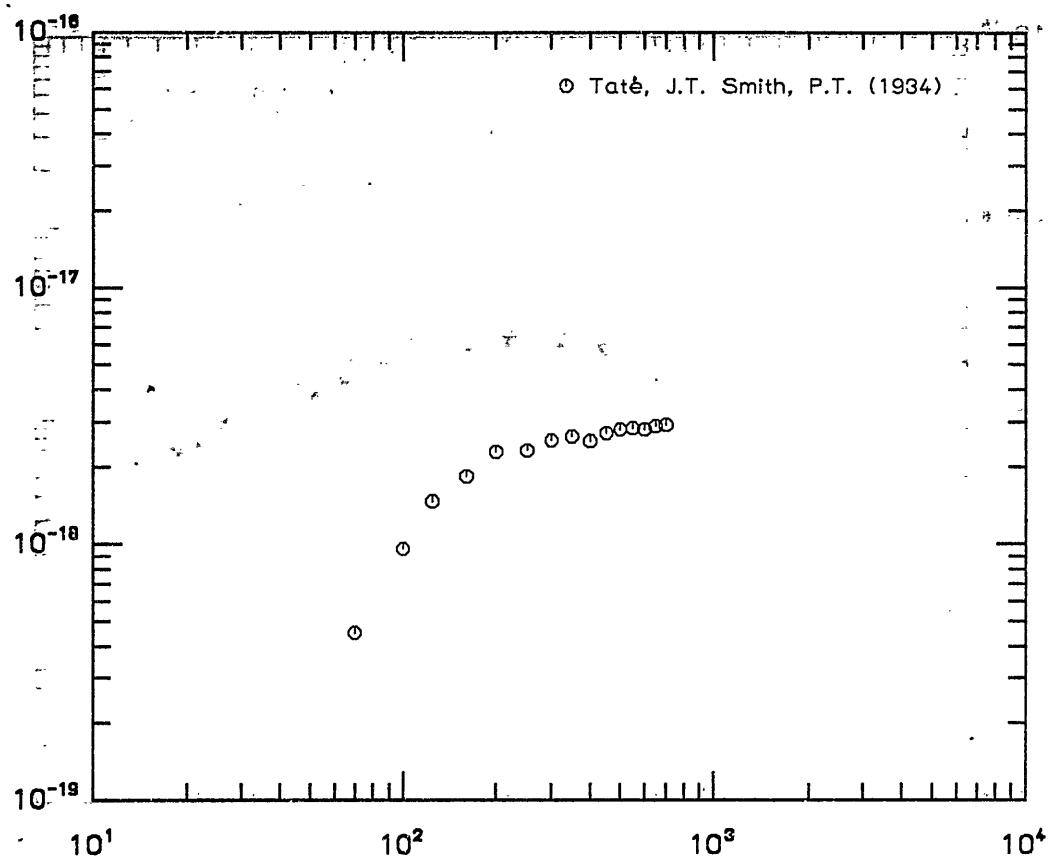


Fig. 74 (v) Electron energy (eV) ET. pF

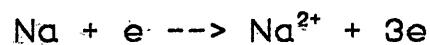


○ Tate, J.T. Smith, P.T. (1934)

Cross section ( $\text{cm}^2$ )



Electron energy (eV)



○ Tate, J.T. Smith, P.T. (1934)

Cross section ( $\text{cm}^2$ )

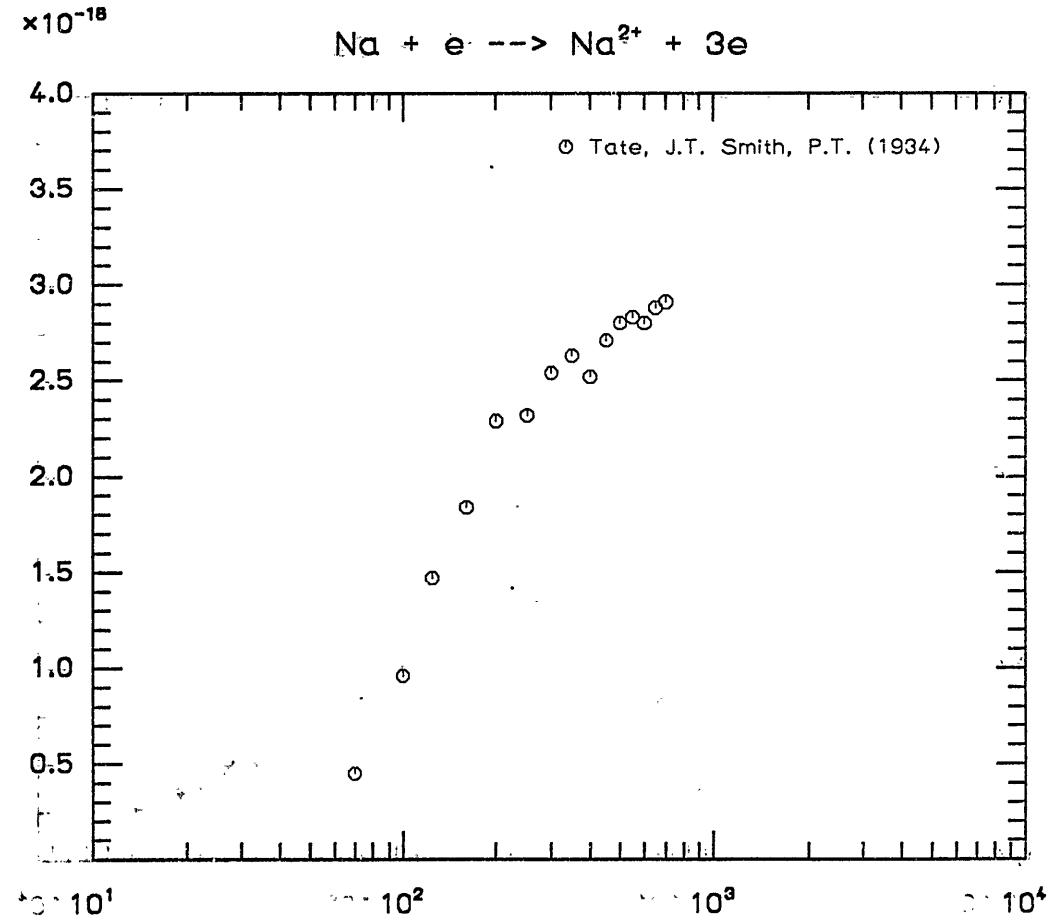


Fig. 75

(○) Electron energy (eV)

θγ ♂F

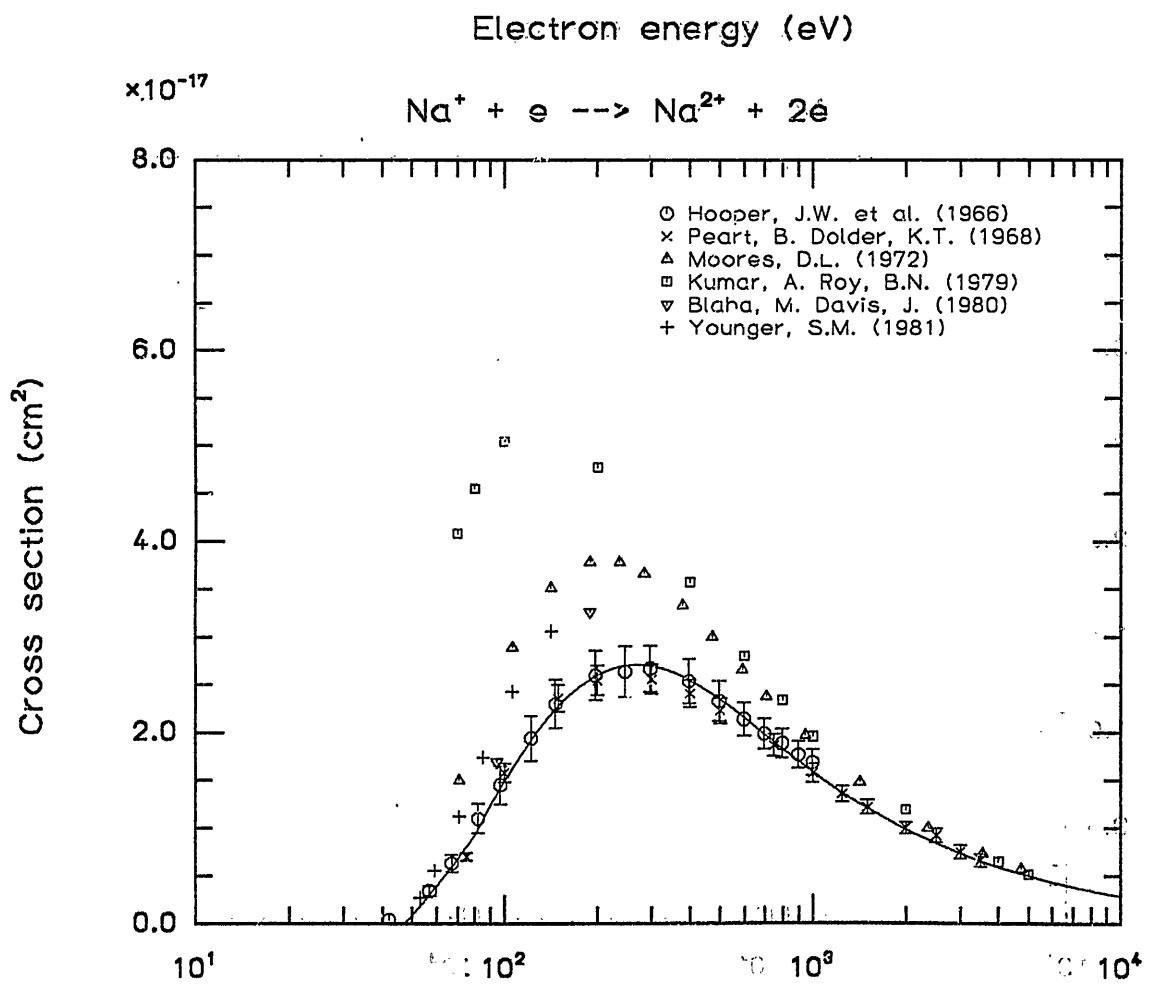
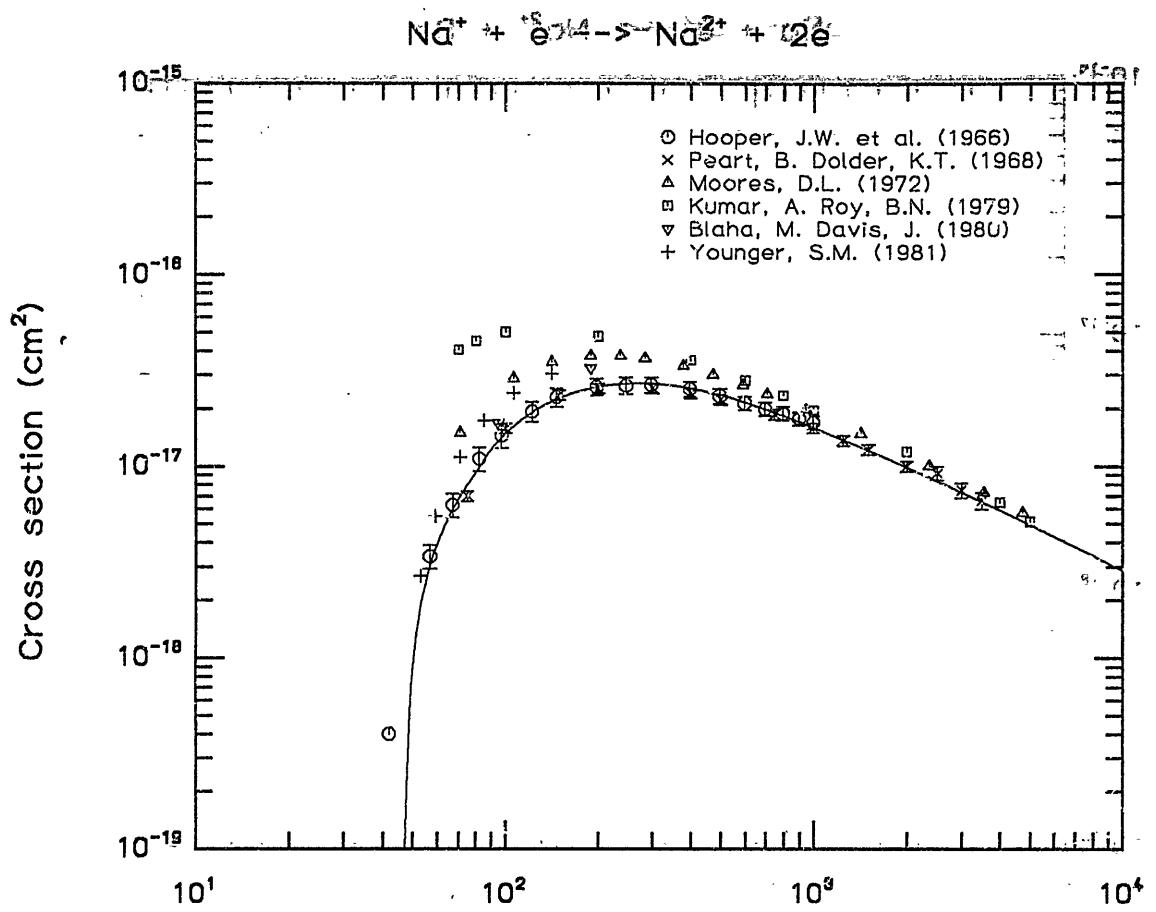
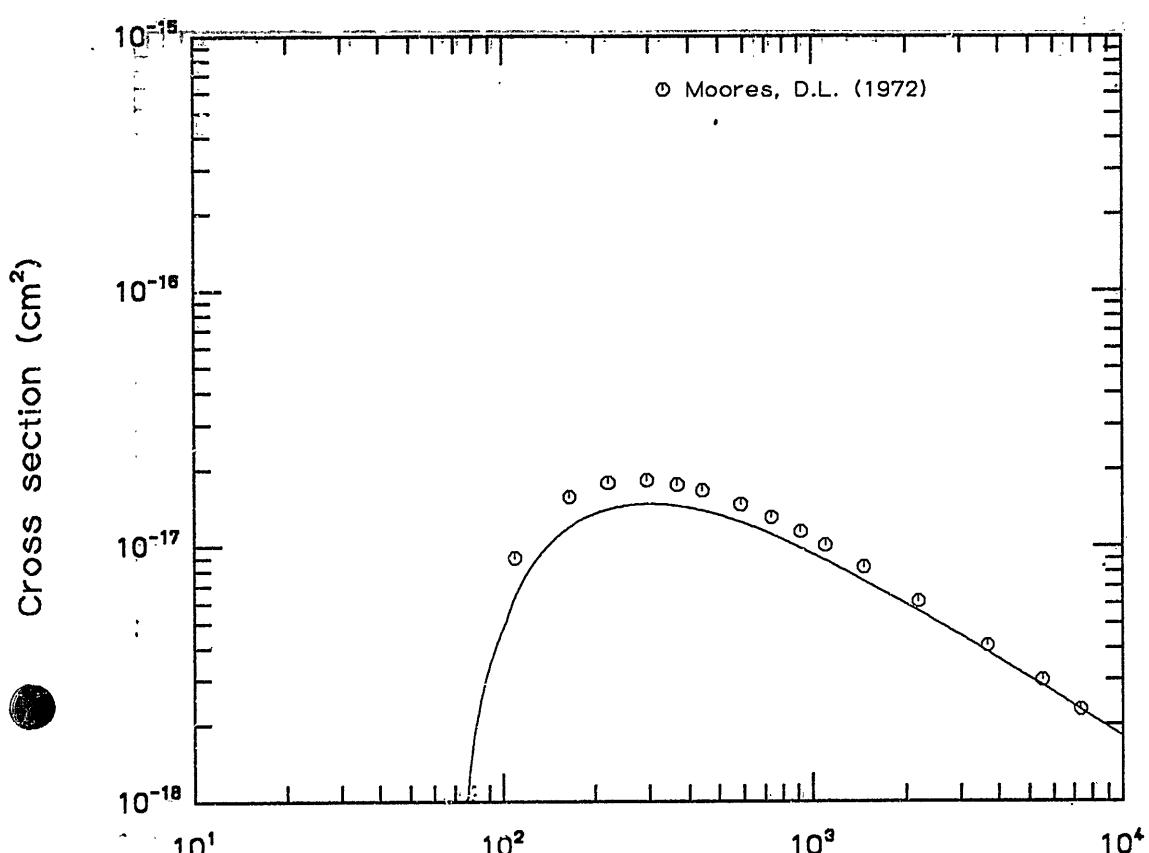
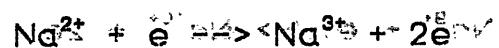


Fig. 76

( $\vee$  Electron energy (eV))

Fig. 77



Electron energy (eV)

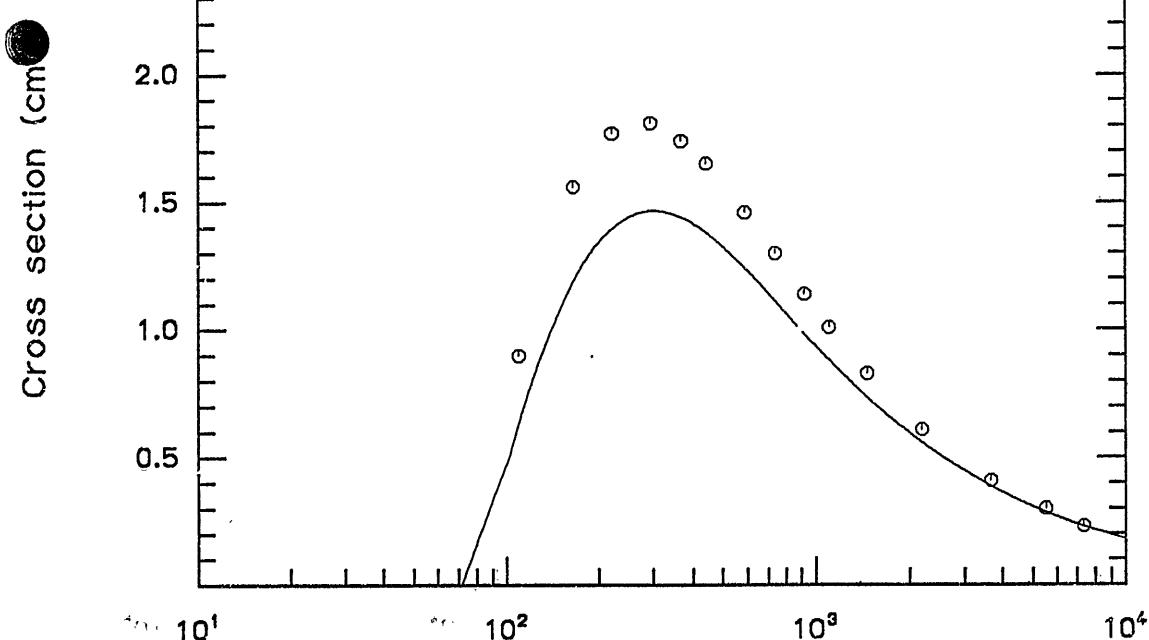
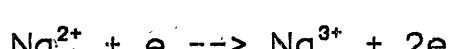


Fig. 77

Electron energy (eV)

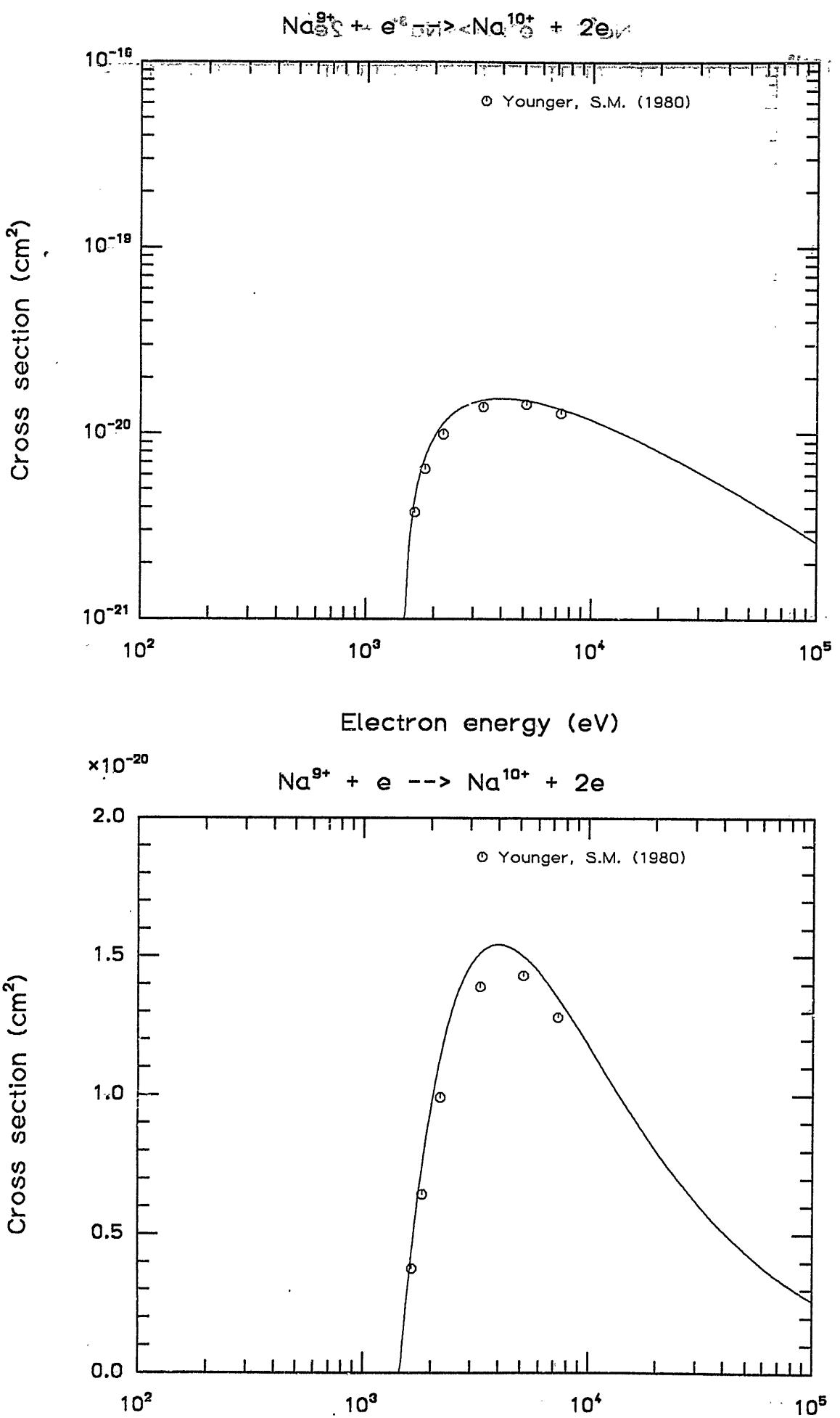
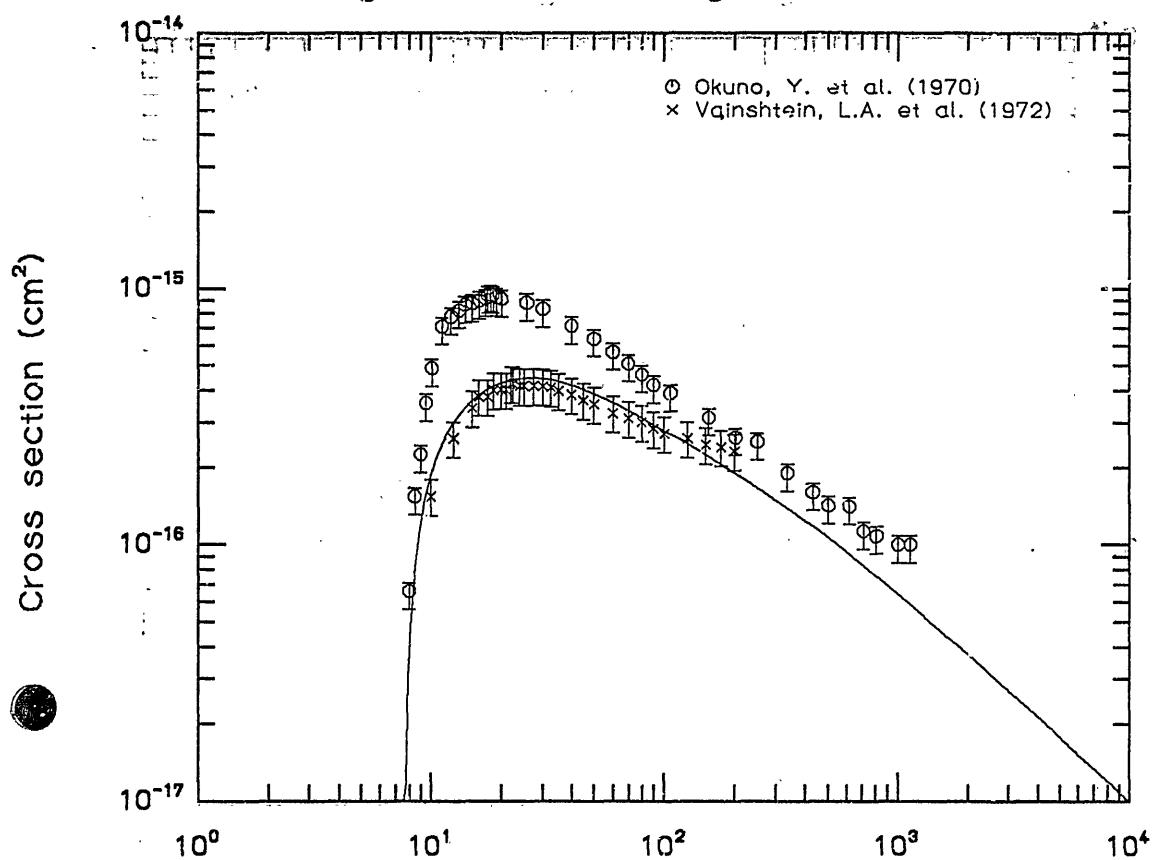
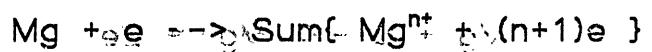


Fig. 78



Electron energy (eV)

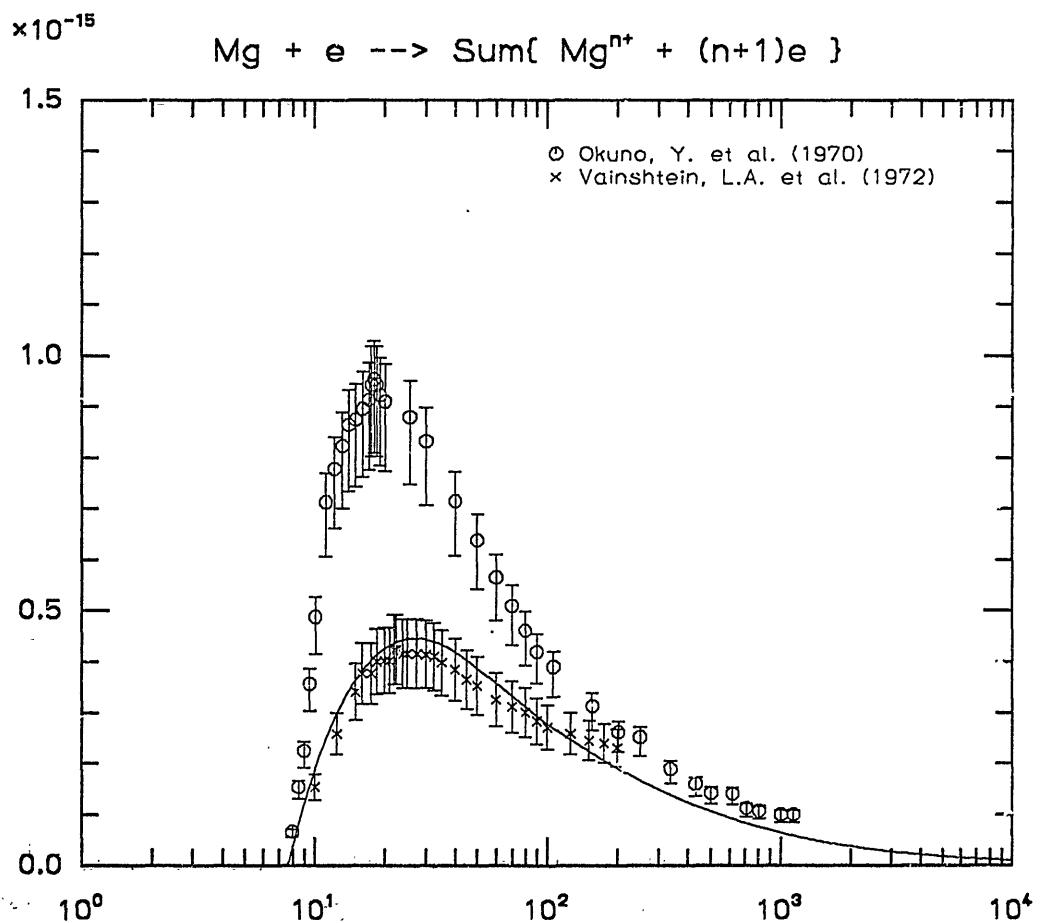


Fig. 79

Electron energy (eV)

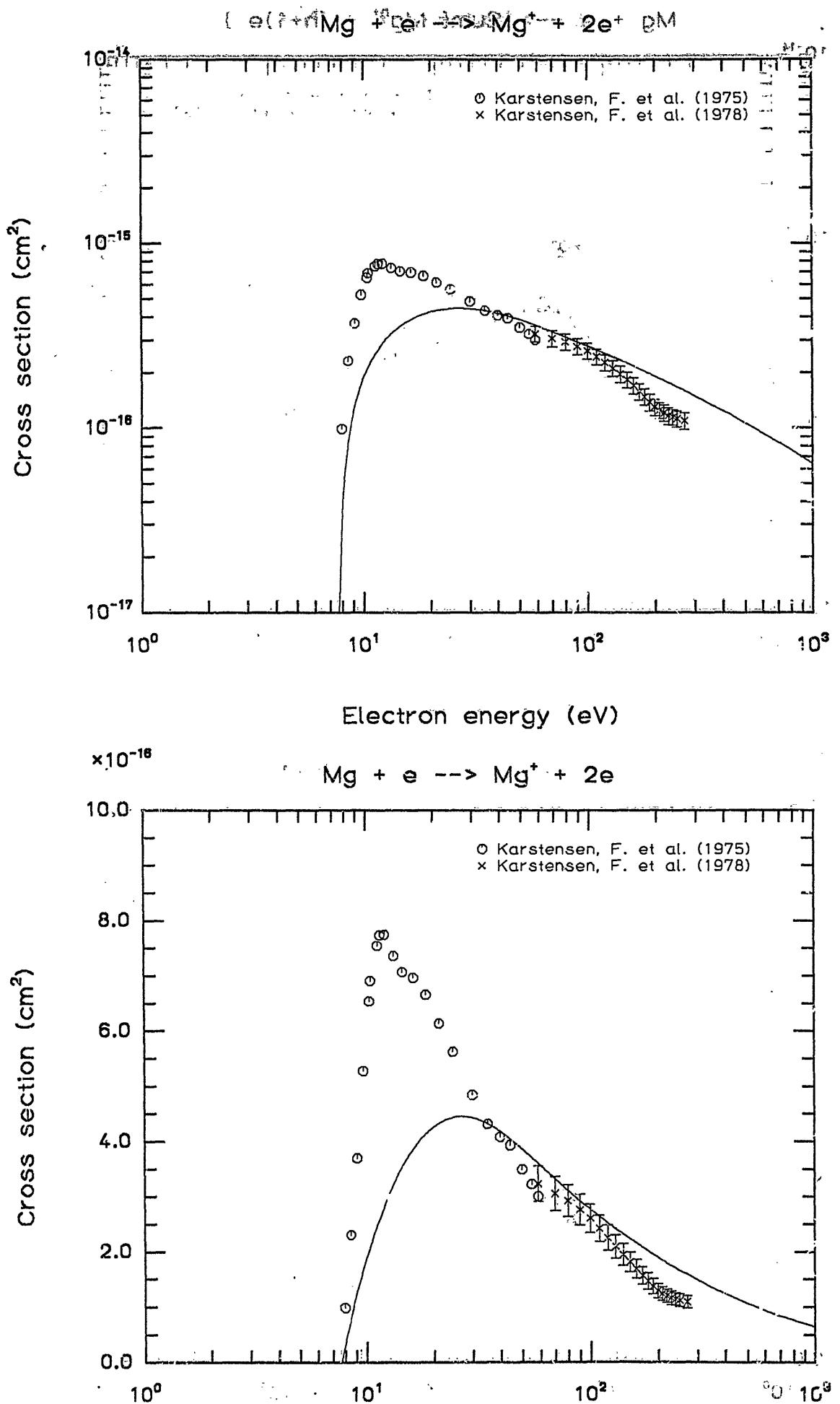
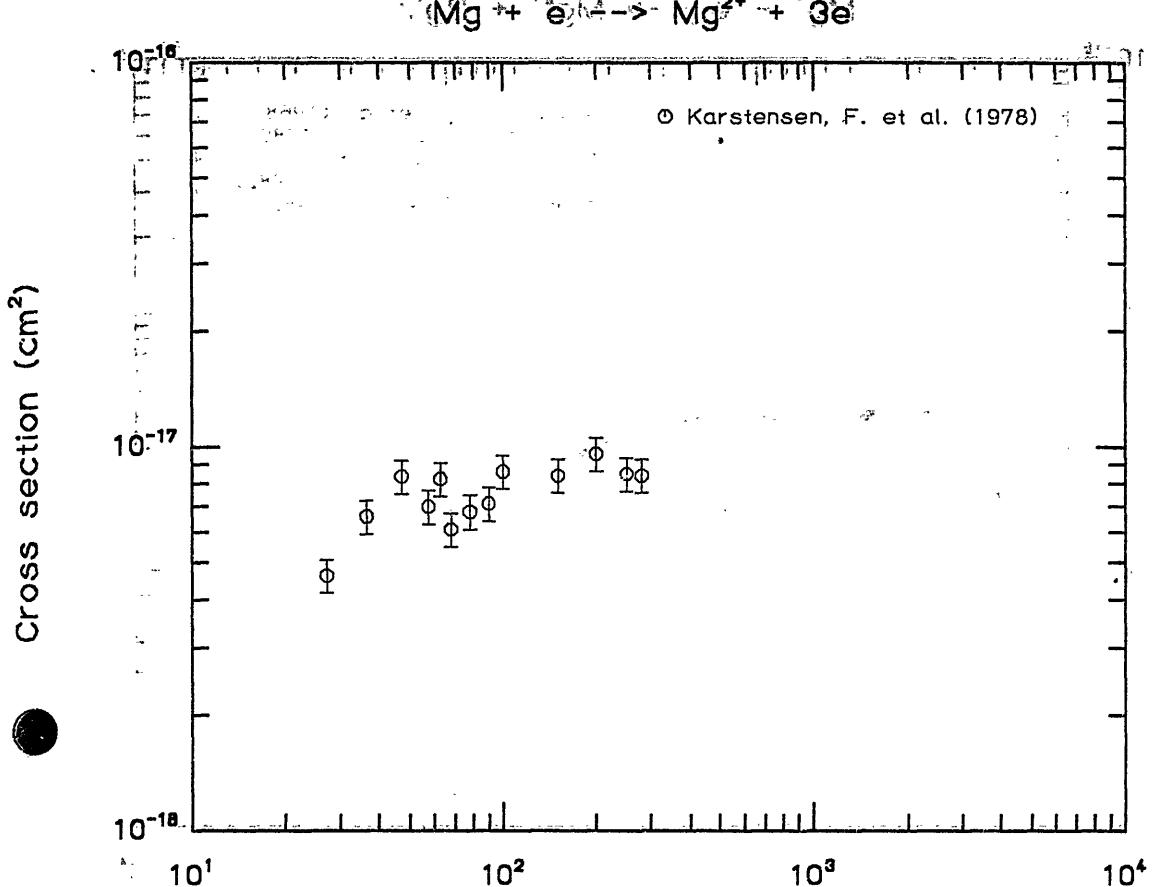
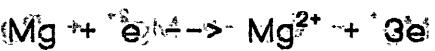


Fig. 80

(V) Electron energy (eV)

87.017



Electron energy (eV)

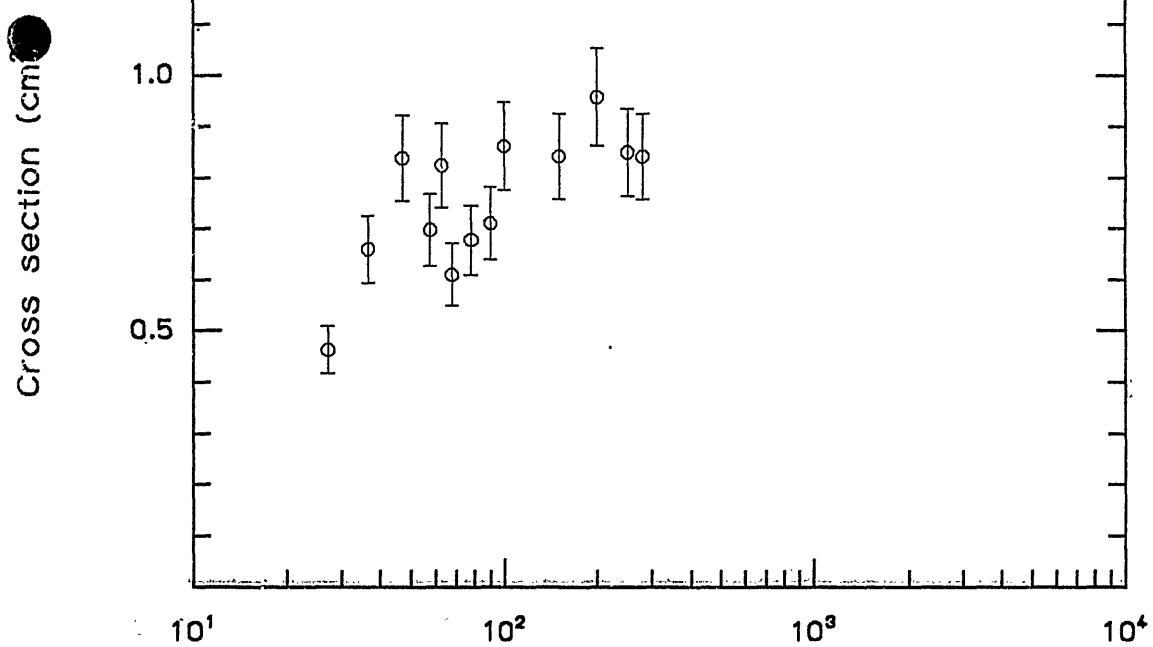
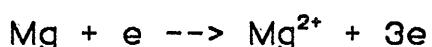


Fig. 81

Electron energy (eV)

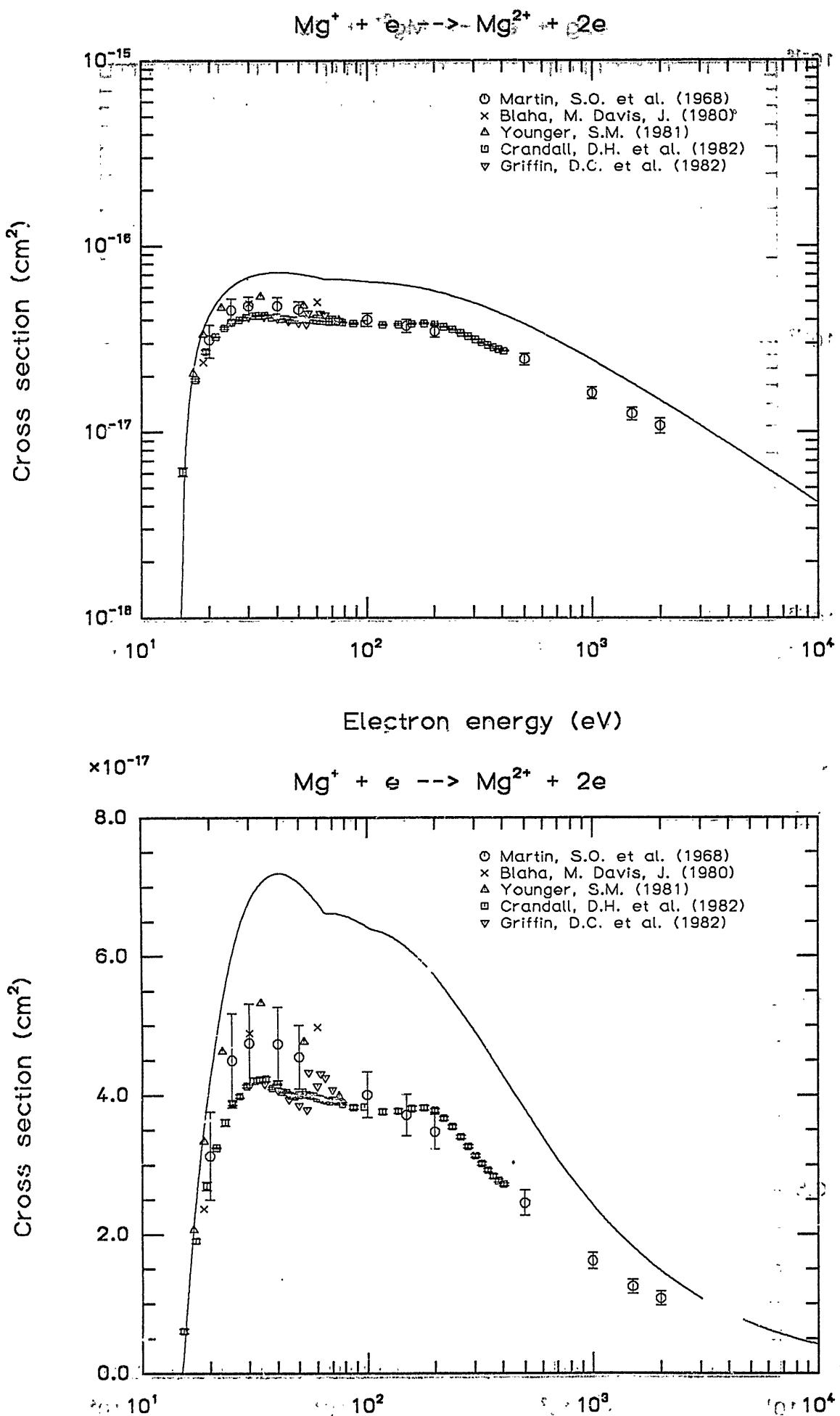
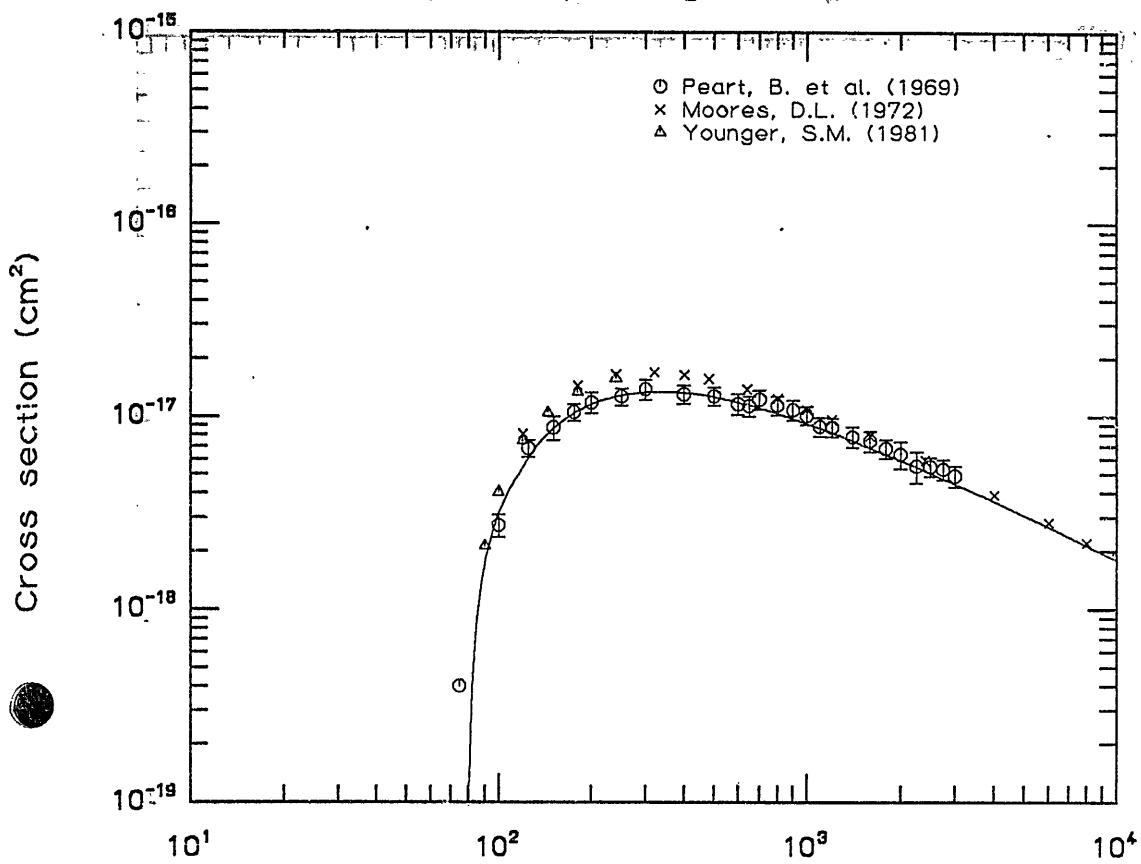
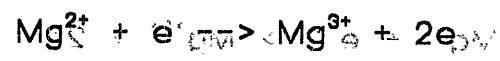


Fig. 82

(v) Electron energy (eV)

18.pif



Electron energy (eV)

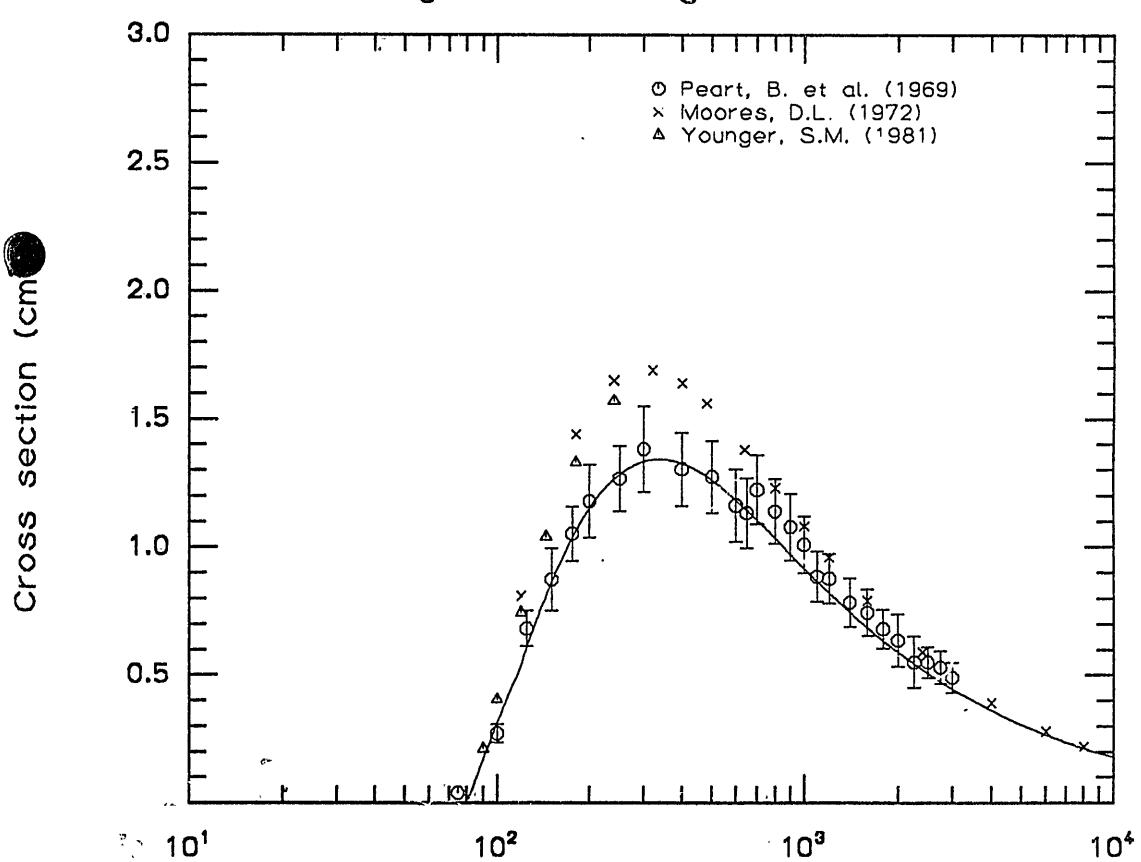
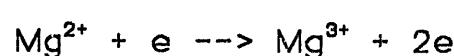


Fig. 83

Electron energy (eV)

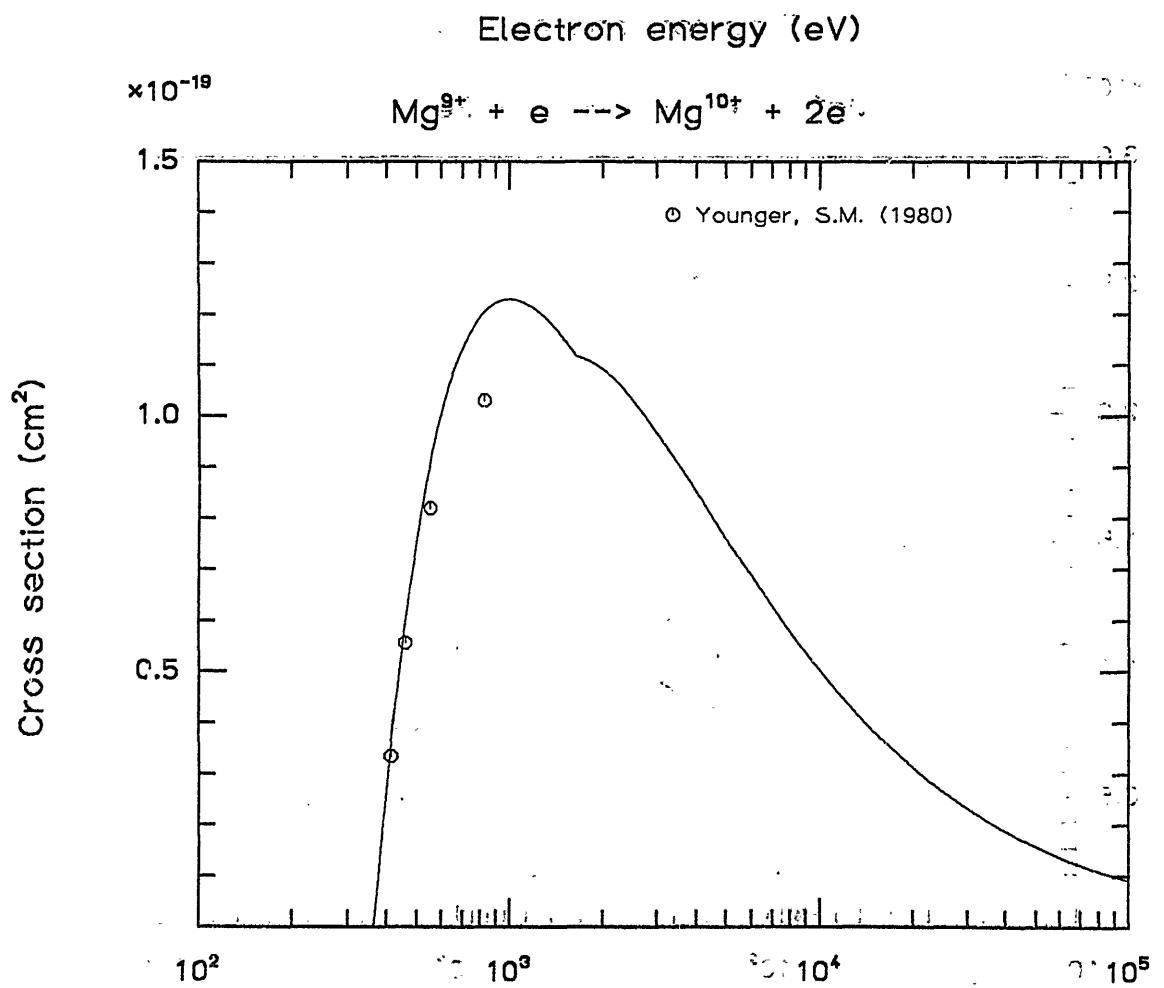
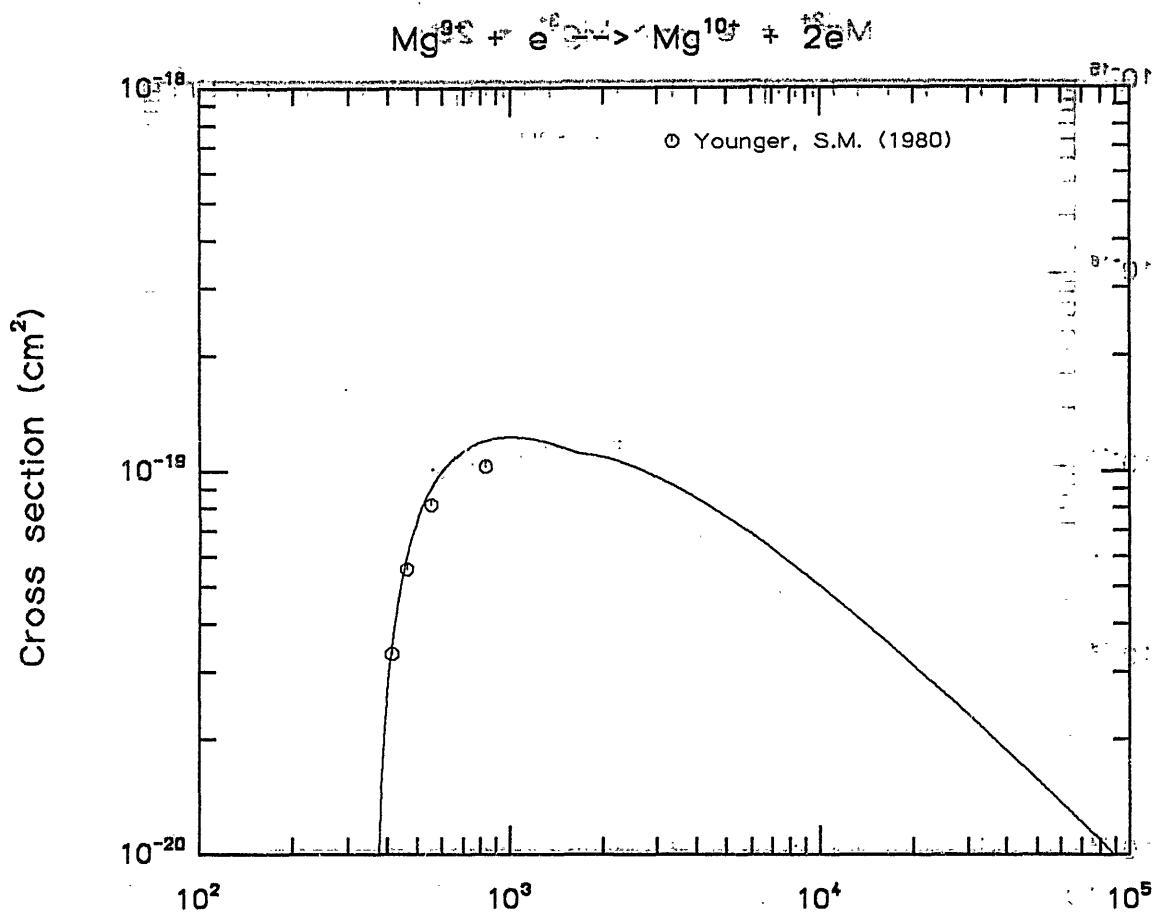


Fig. 84 (v Electron energy (eV)) E8 piR

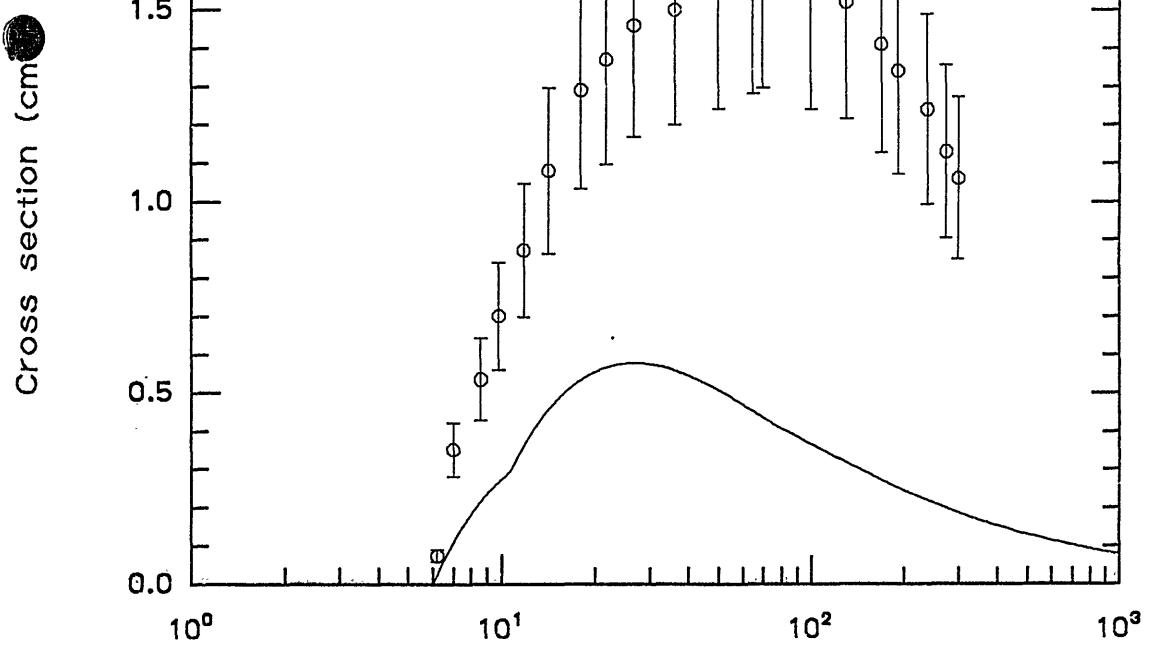
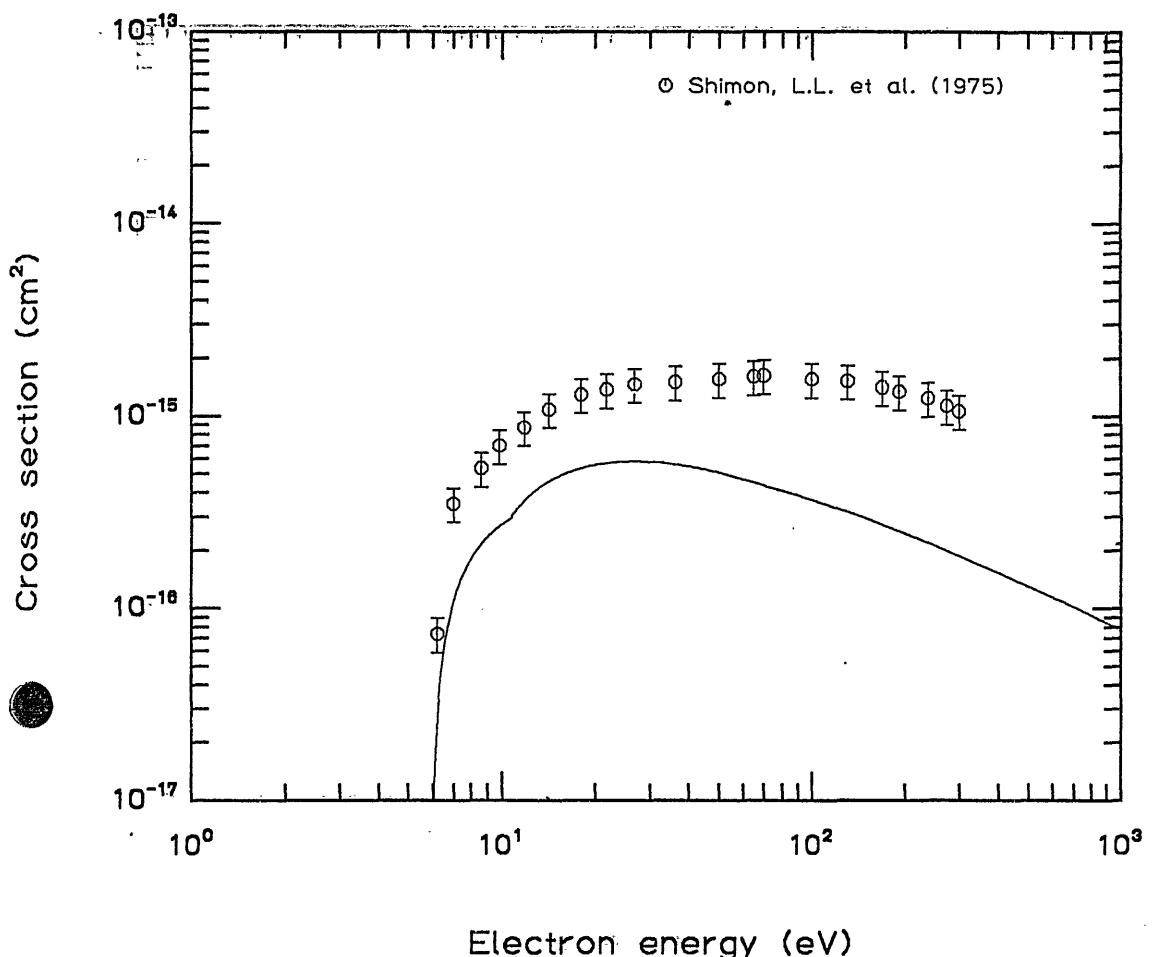
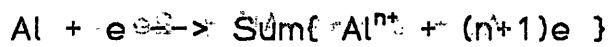


Fig. 85

Electron energy (eV)

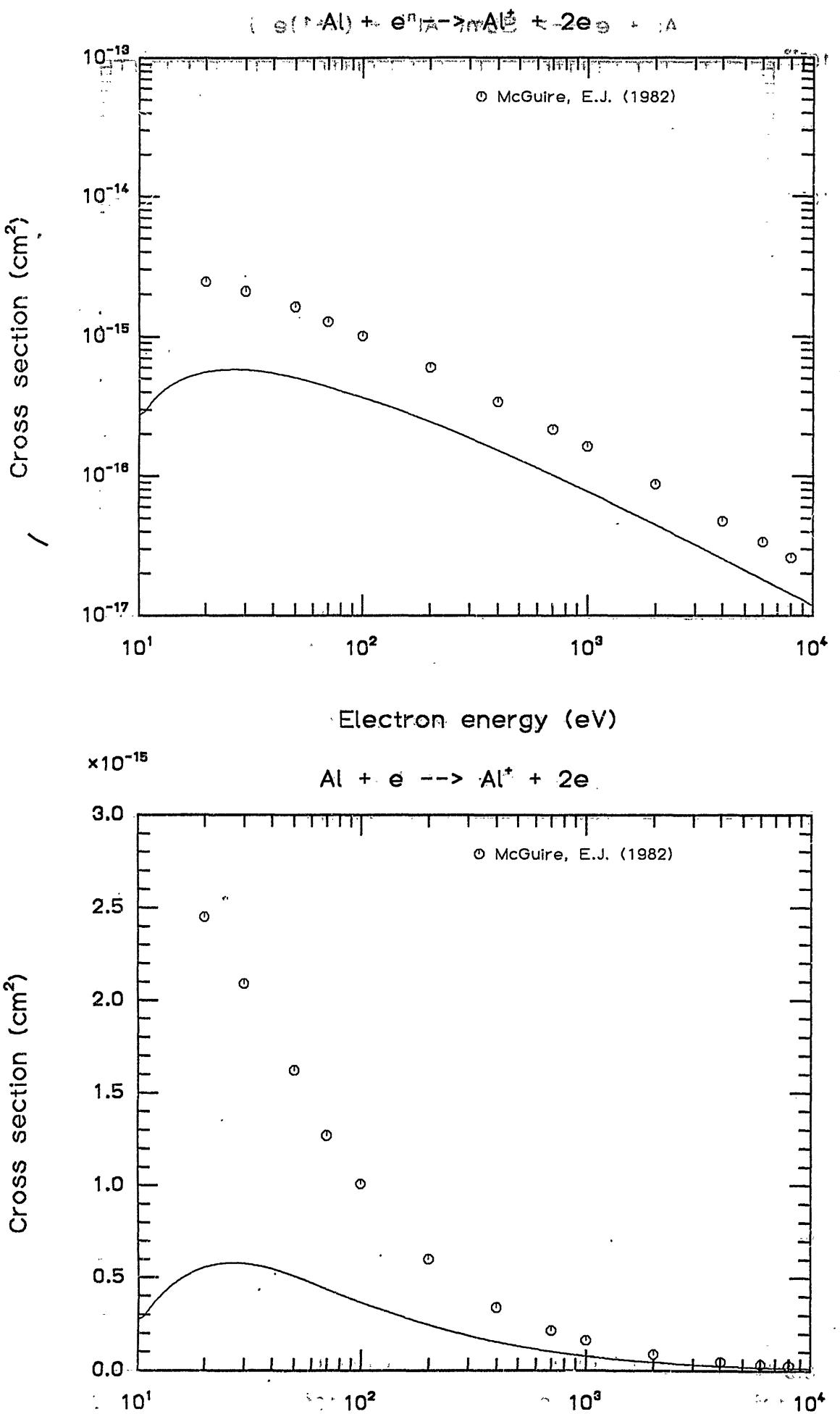


Fig. 86      Electron energy (eV)      Fig. 86

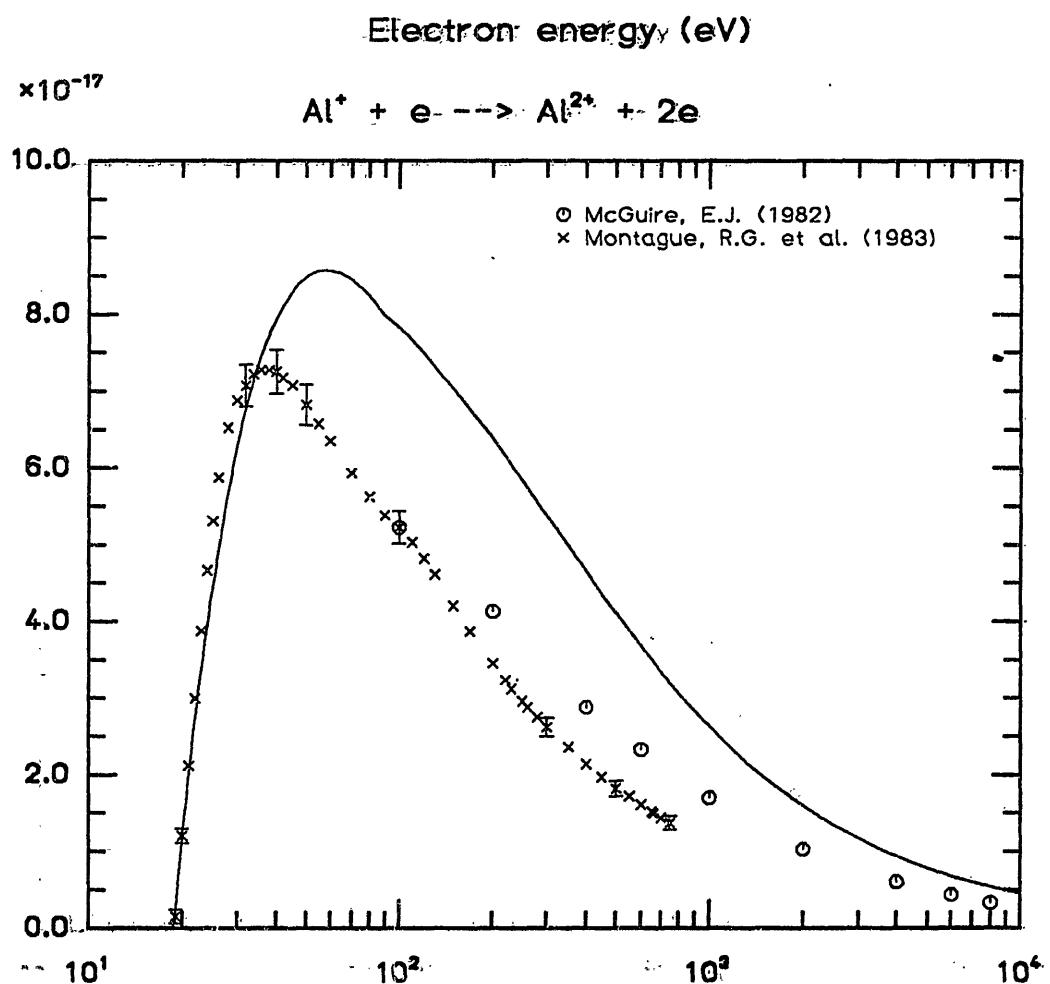
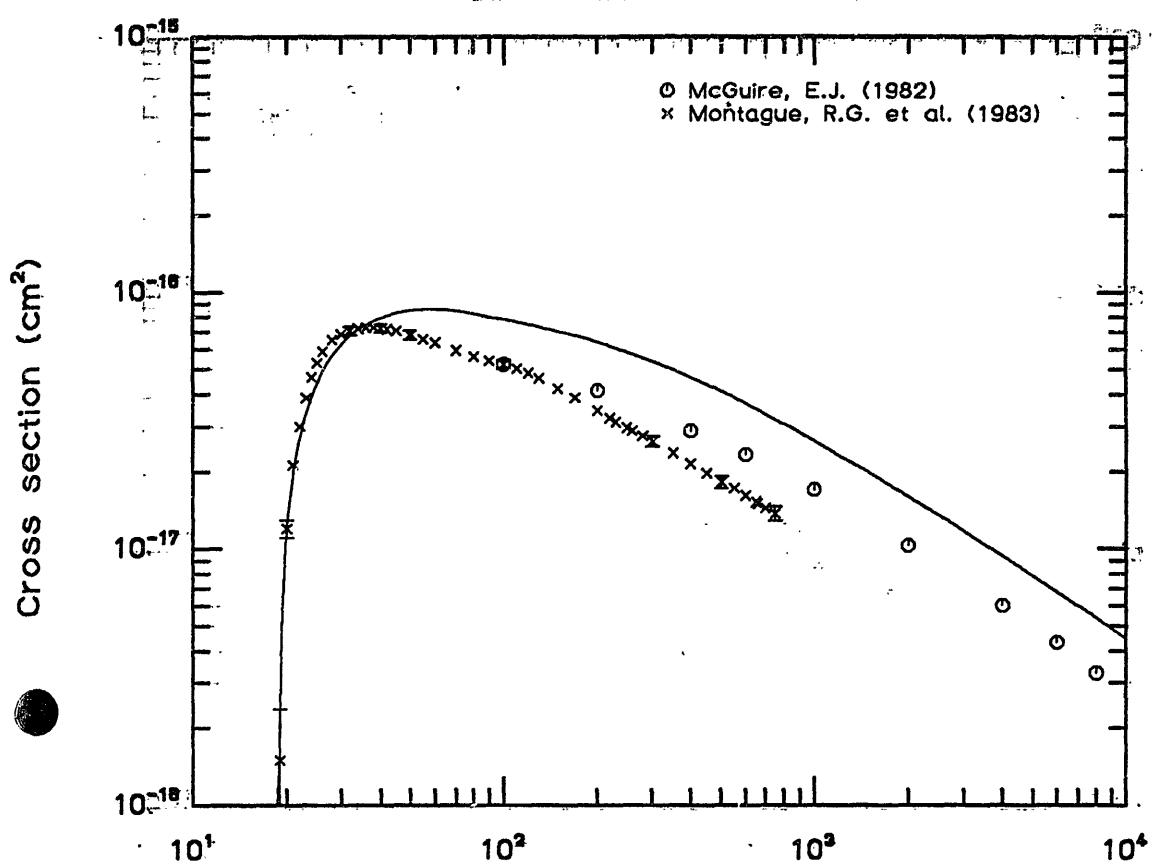


Fig. 87

Electron energy (eV)

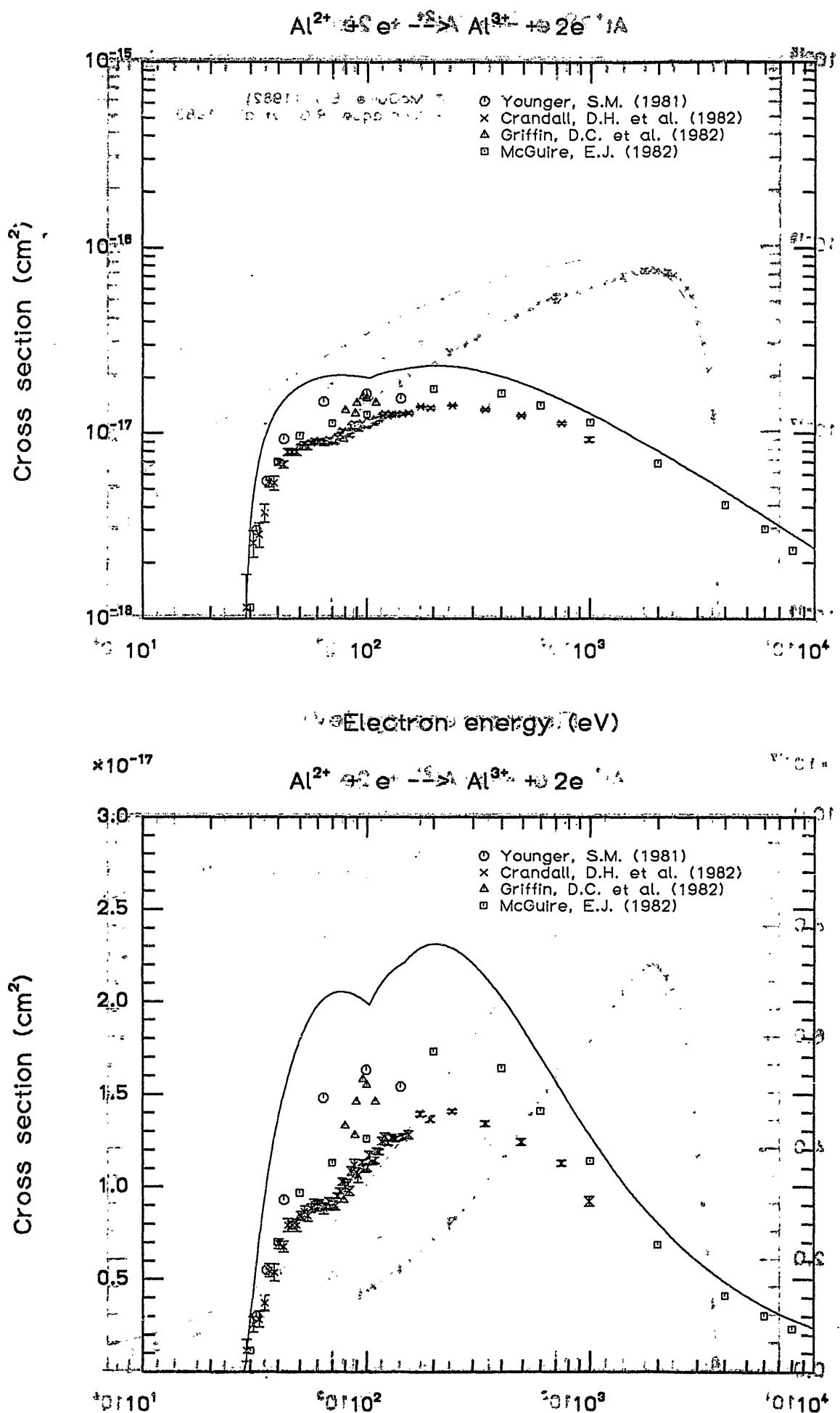
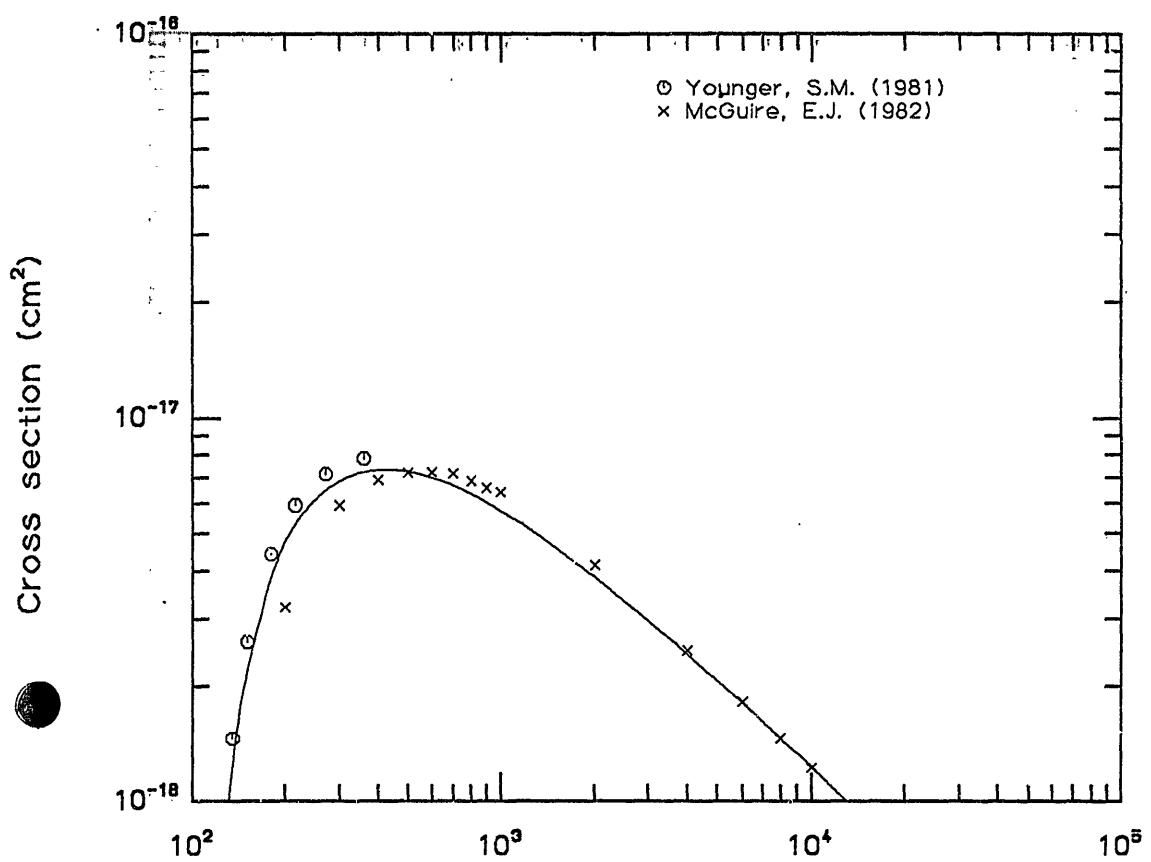
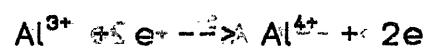


Fig. 88

(vs Electron energy) (eV)

Fig. 88



Electron energy (eV)

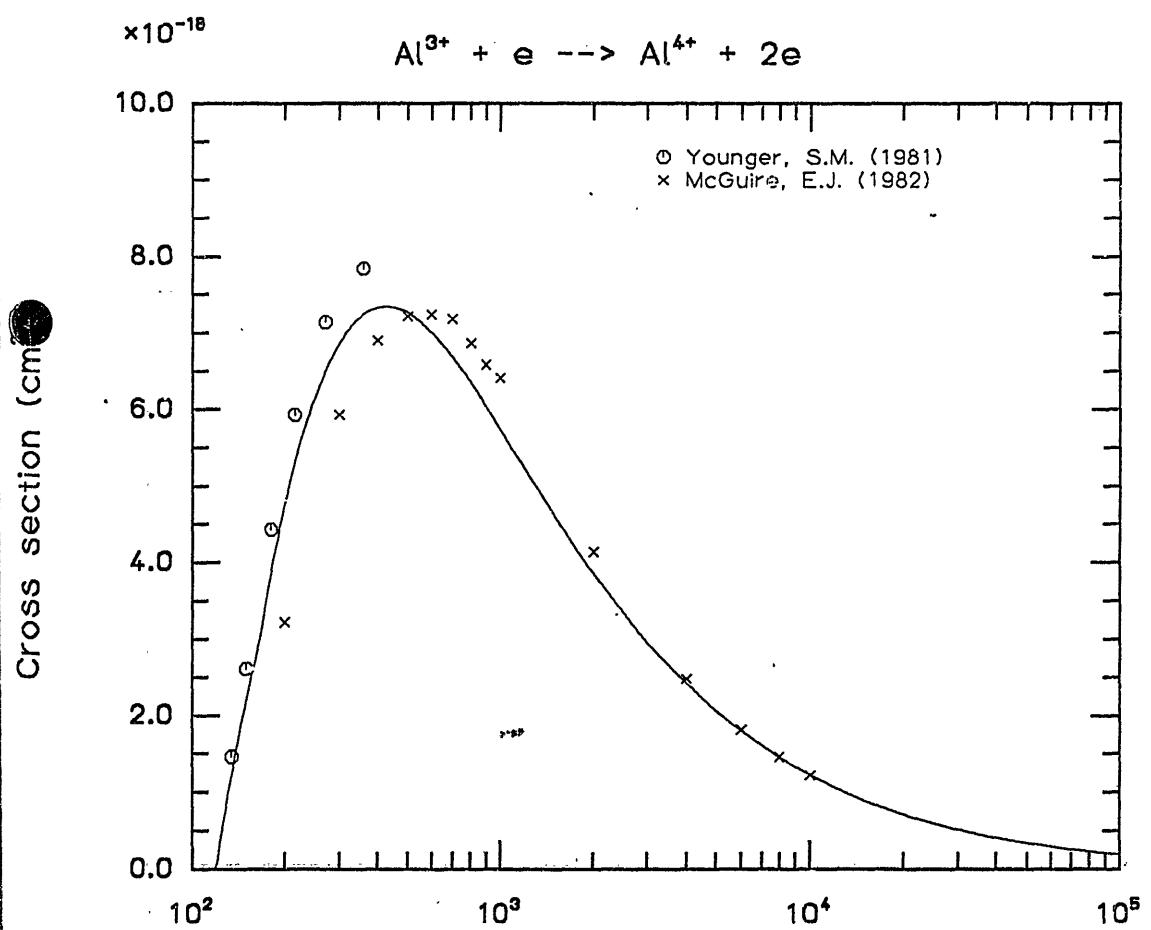


Fig. 89

Electron energy (eV)

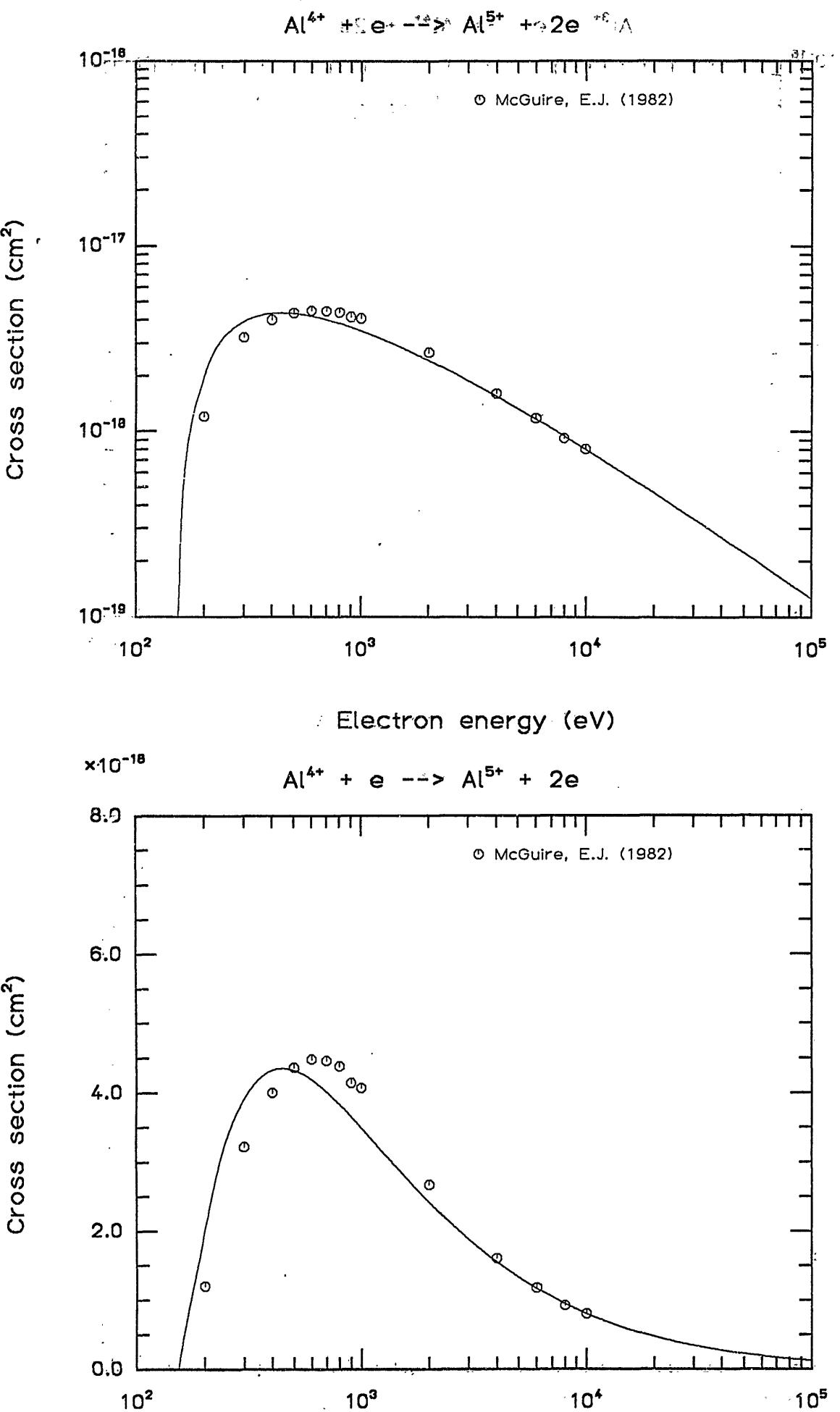
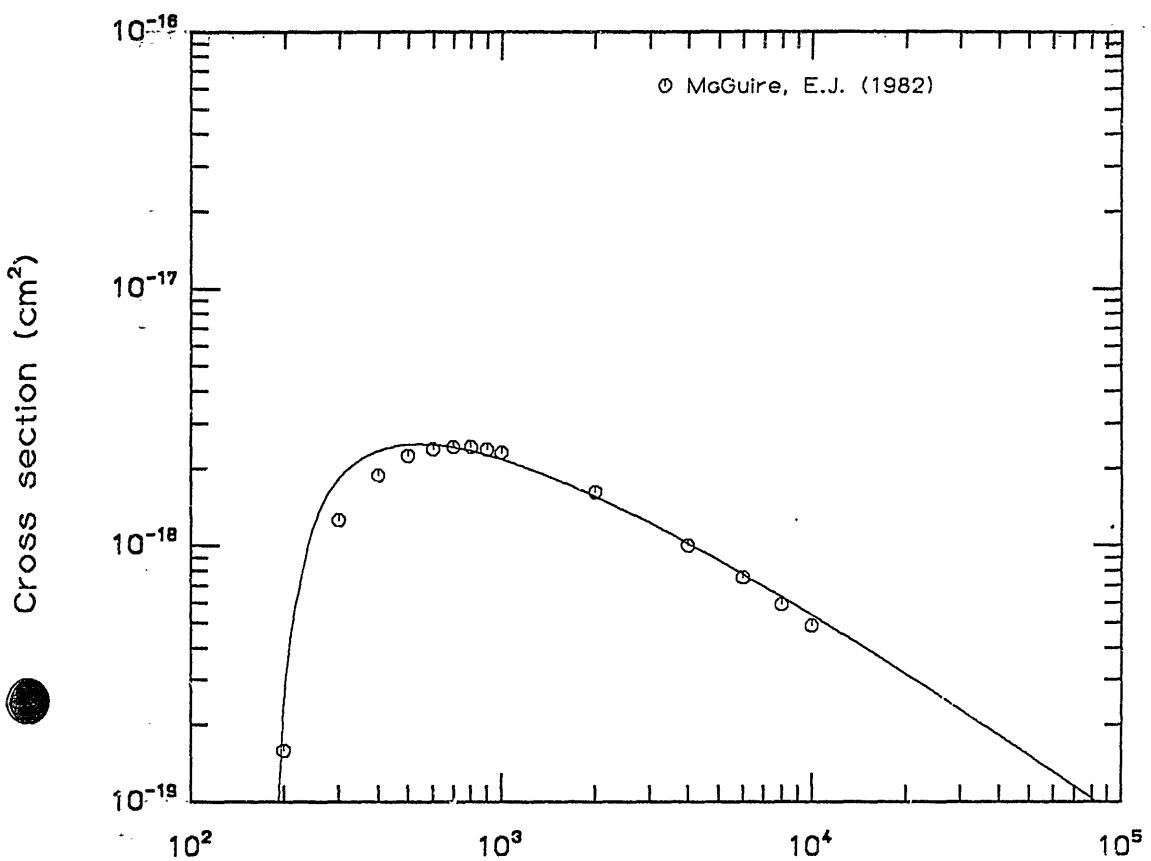
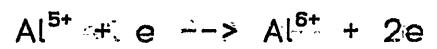


Fig. 90

Electron energy (eV)



Electron energy (eV)

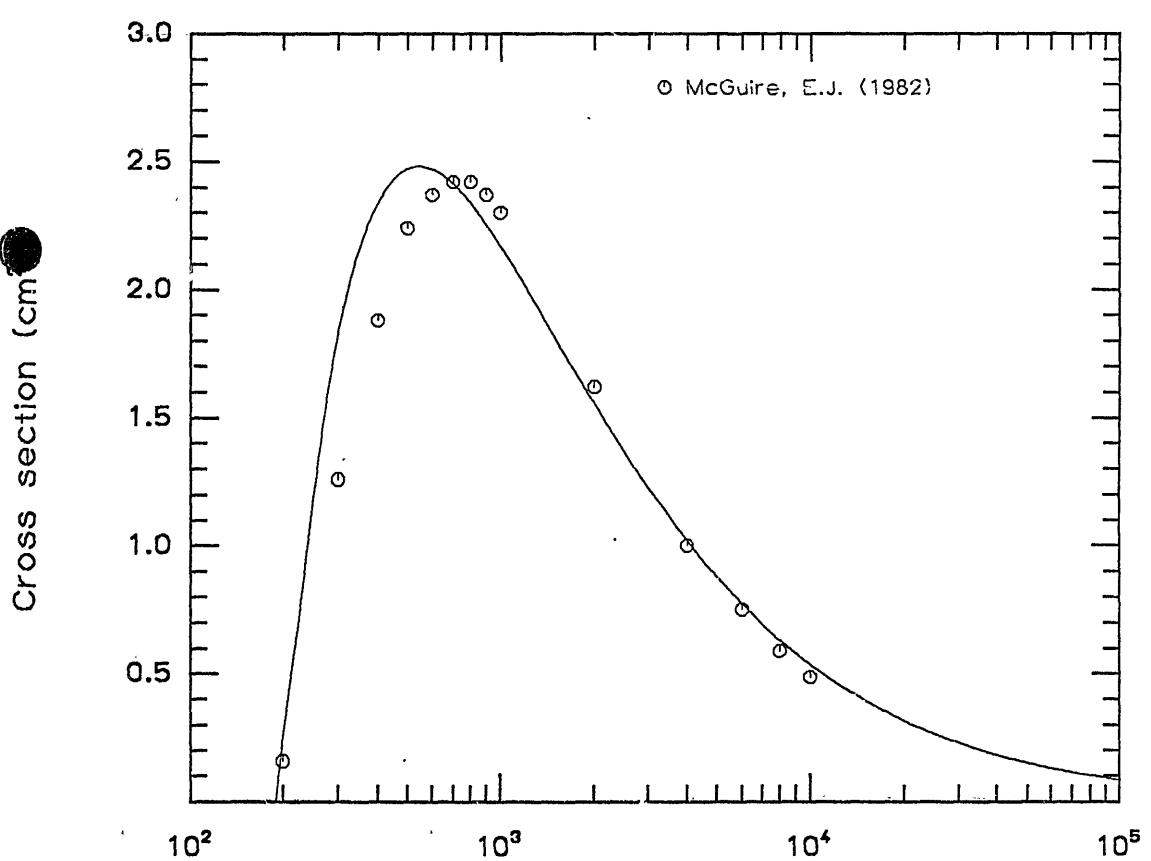
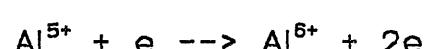


Fig. 91

Electron energy (eV)

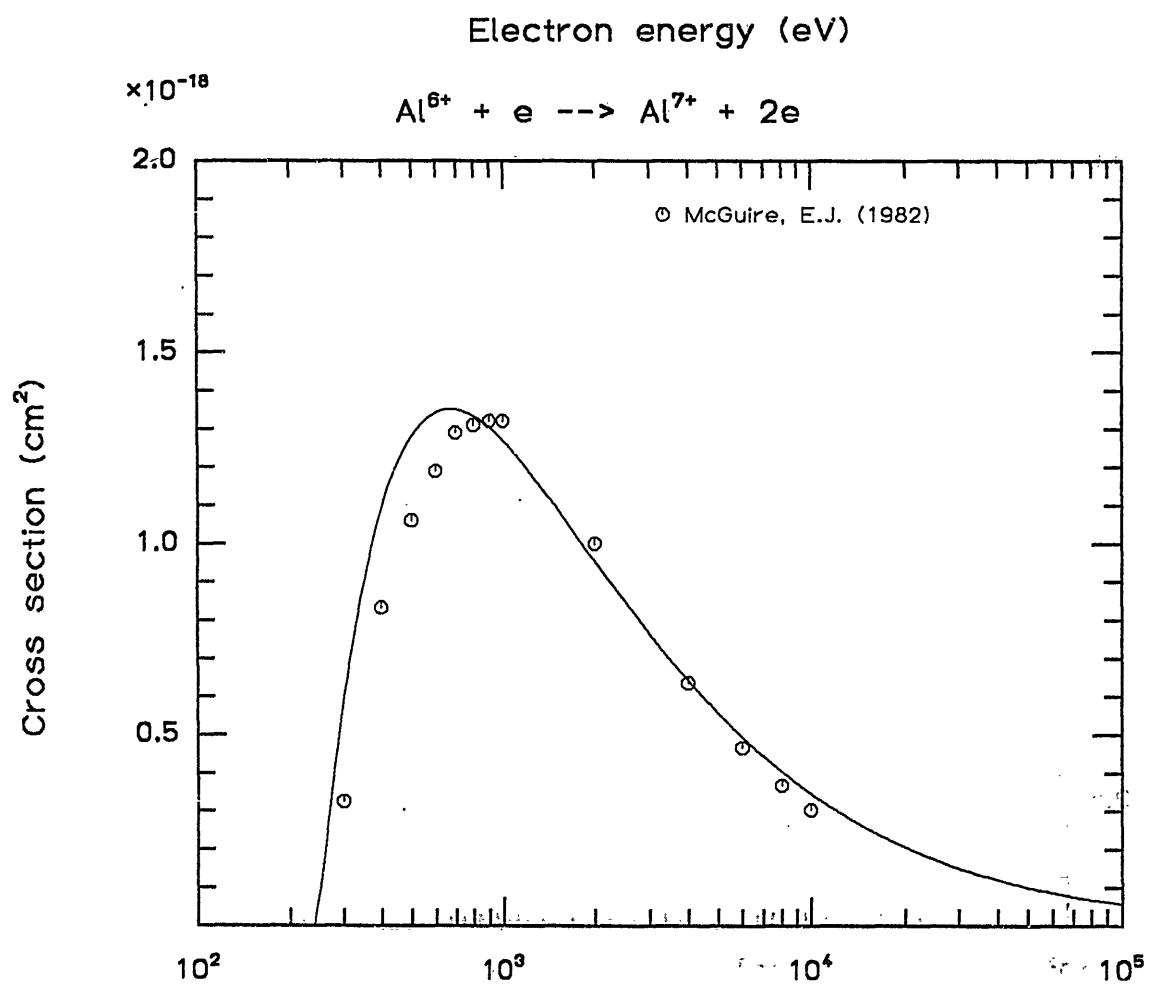
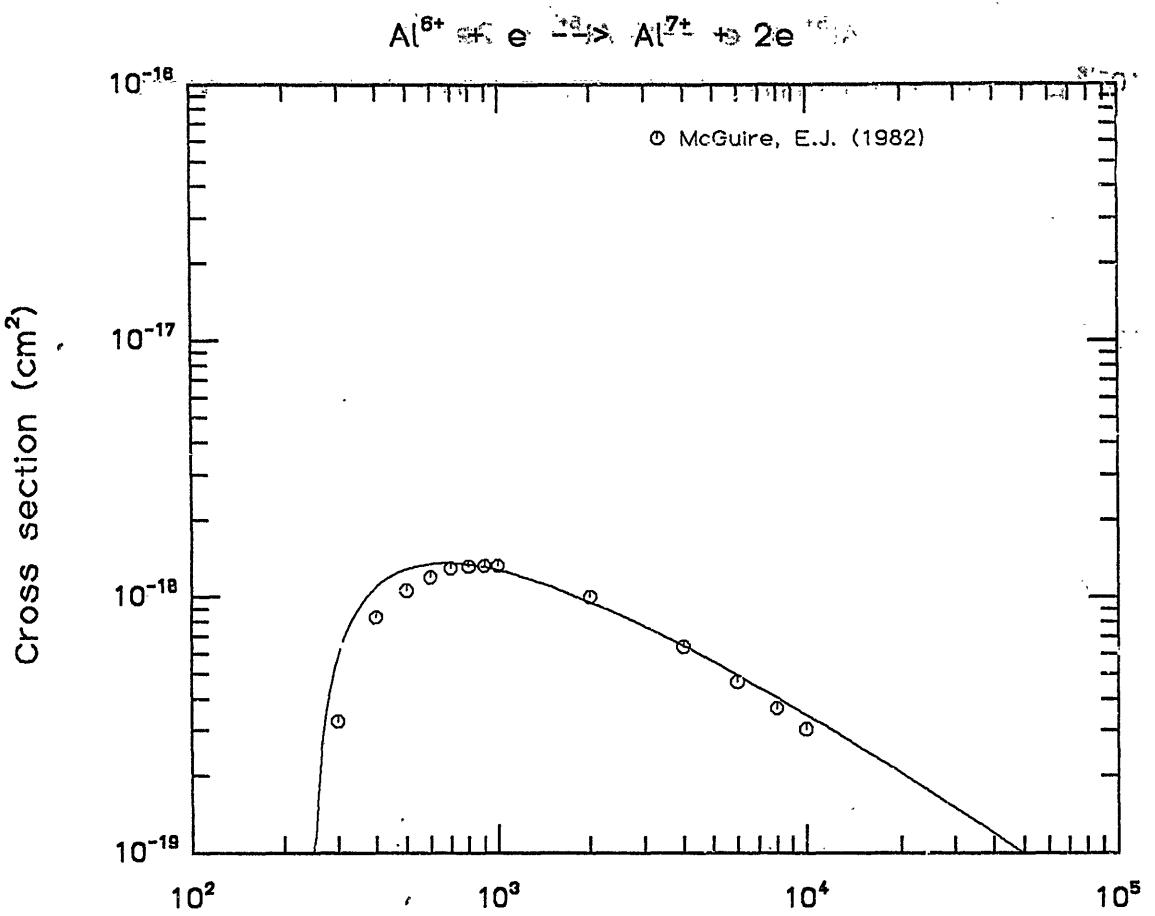


Fig. 92      Cross section vs. Electron energy (eV)      re p. 3

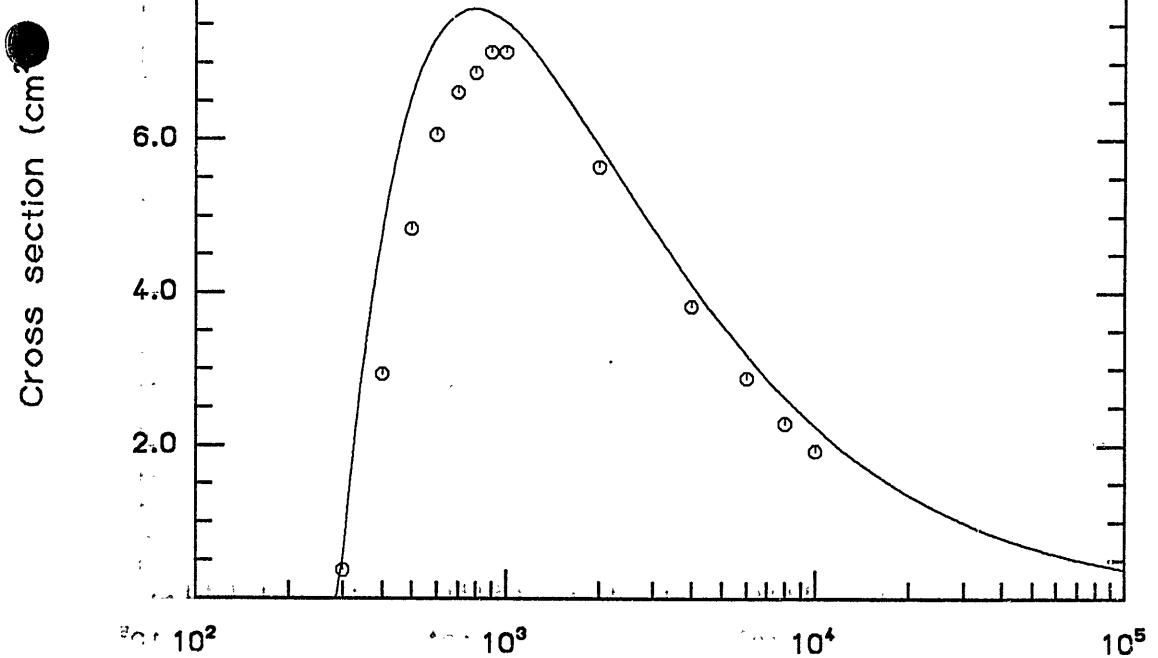
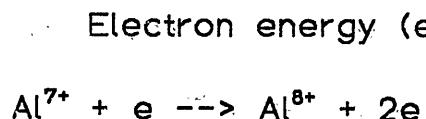
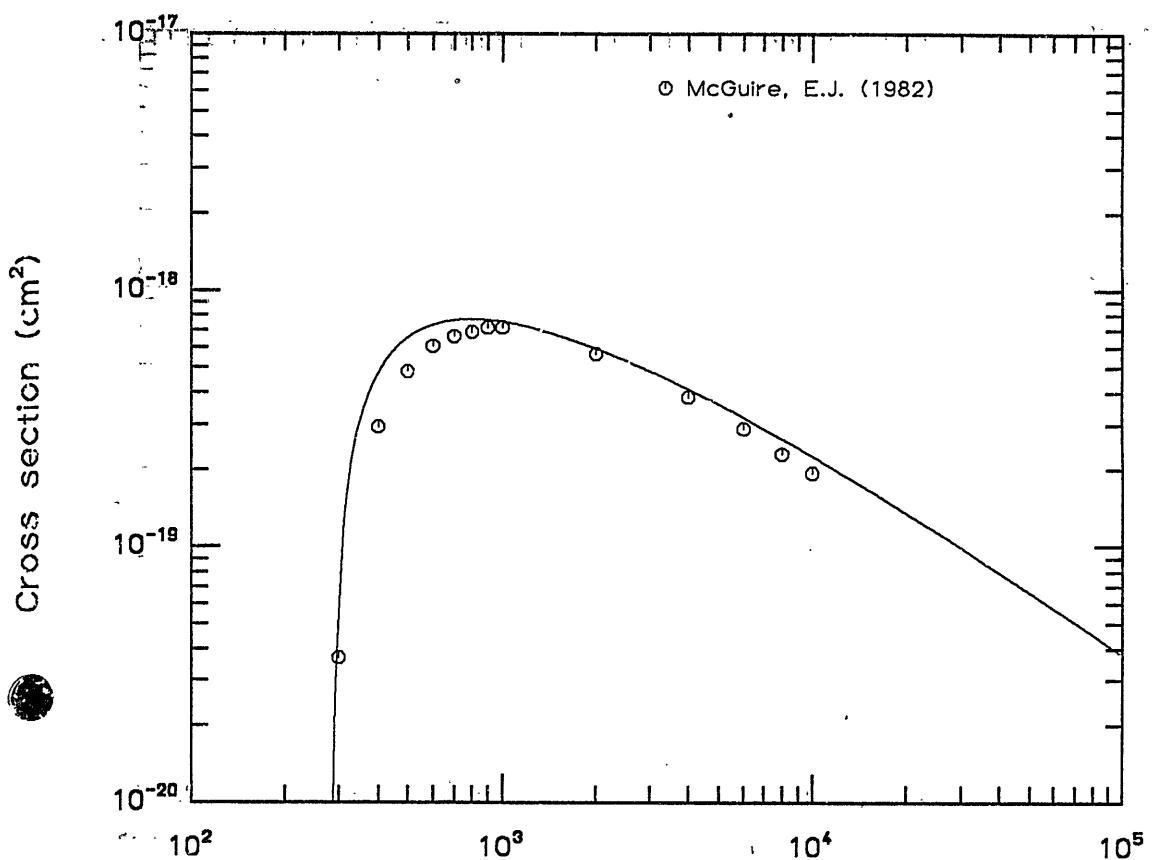


Fig. 93 (a) Electron energy (eV)  $\text{Al}^{7+} + e^- \rightleftharpoons \text{Al}^{8+} + 2e^-$

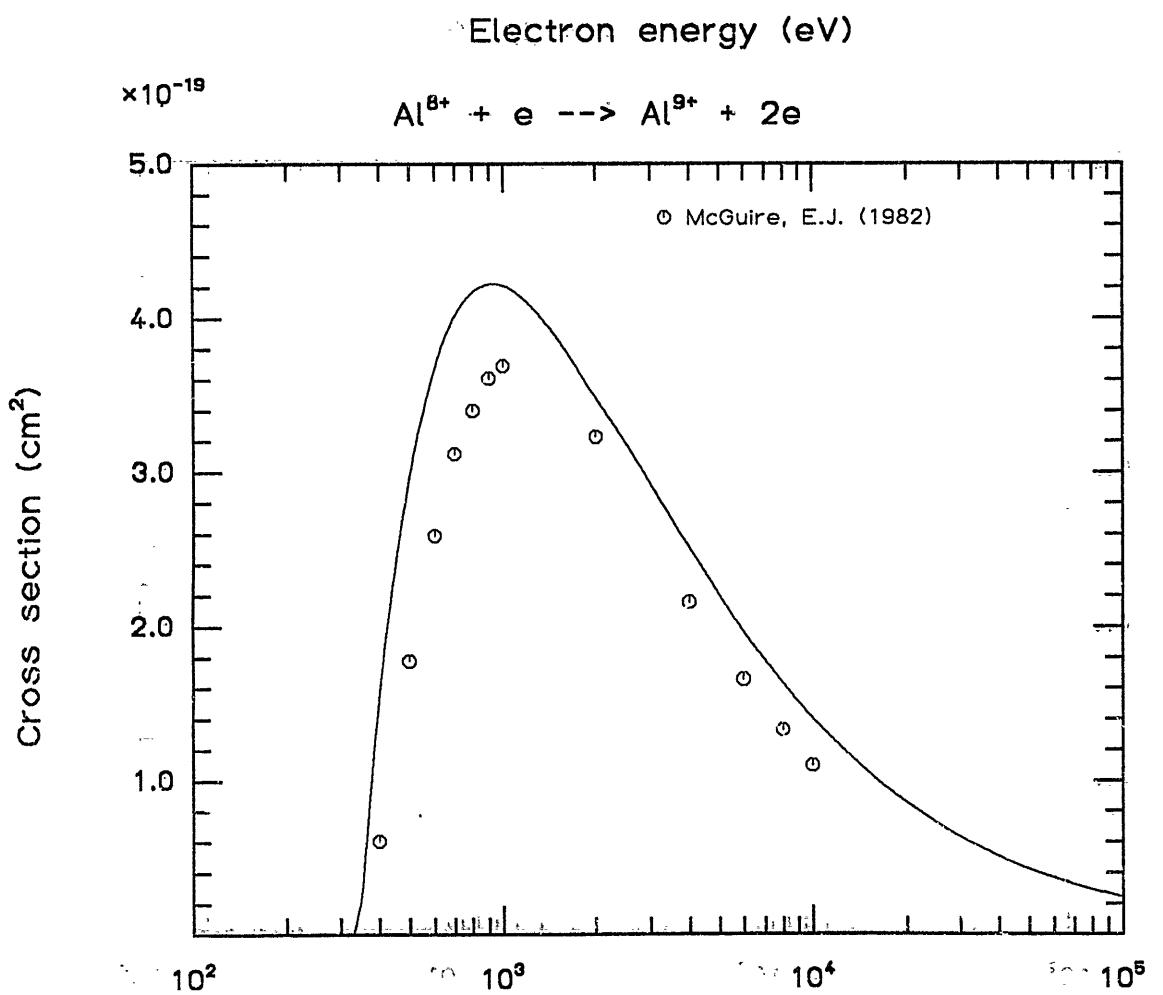
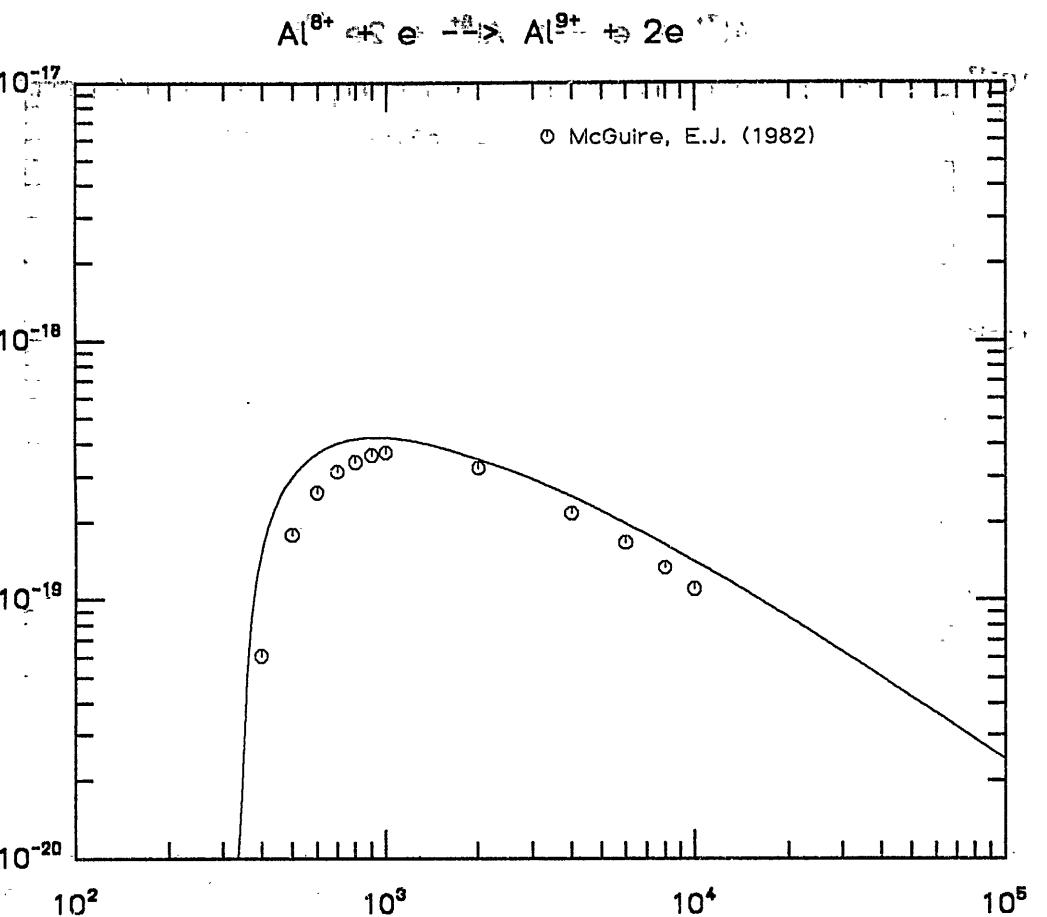
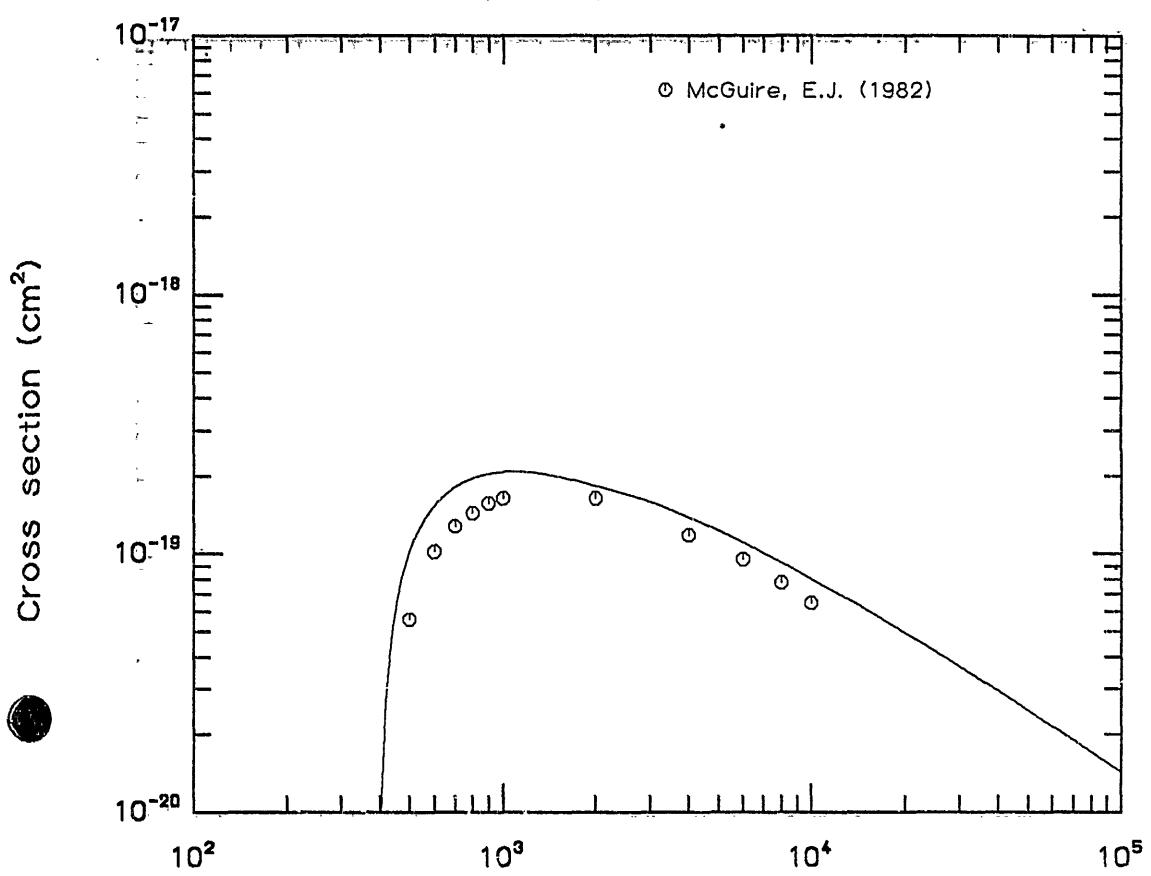
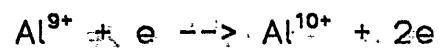


Fig. 94

(a) Electron energy (eV)

$\text{ee} \rightarrow \text{dH}$



Electron energy (eV)

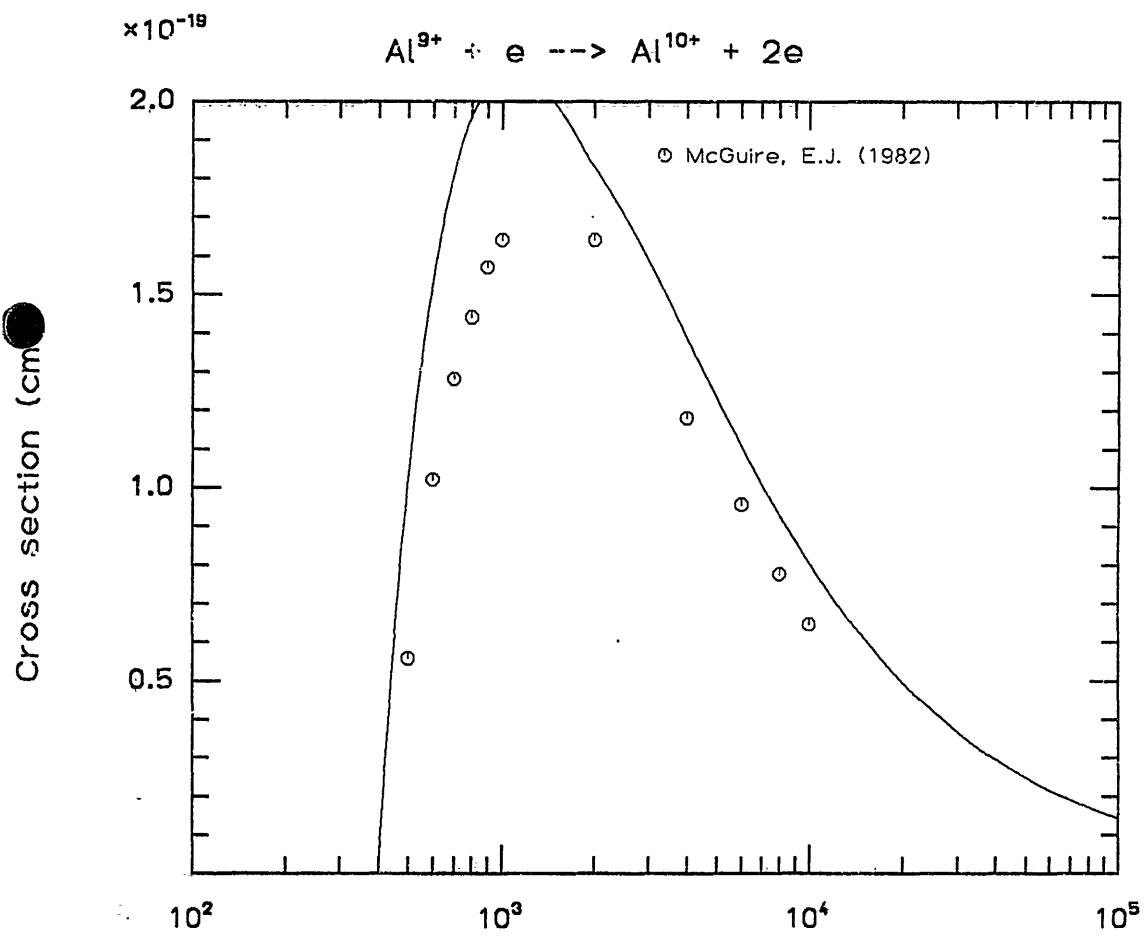


Fig. 95

Electron energy (eV)

80

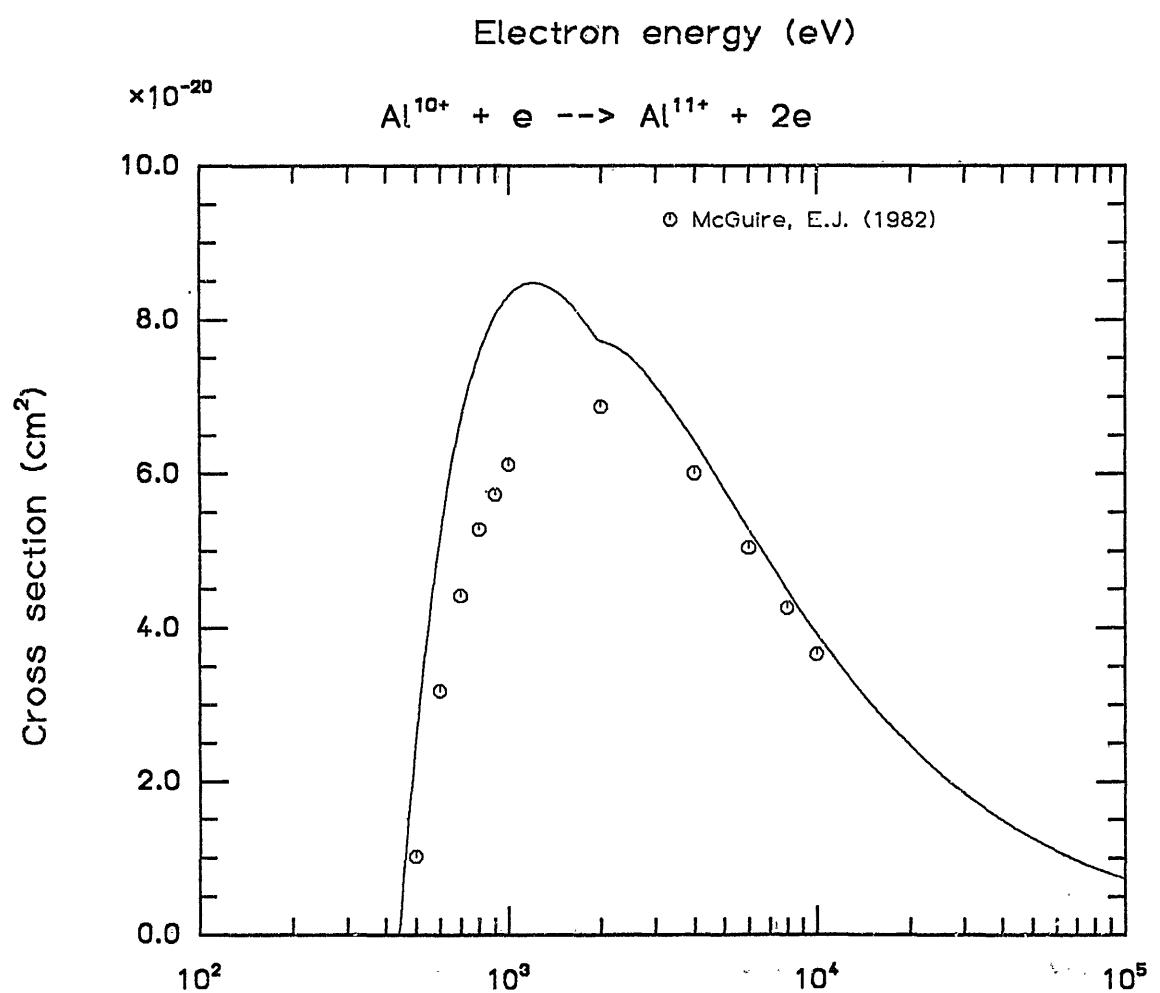
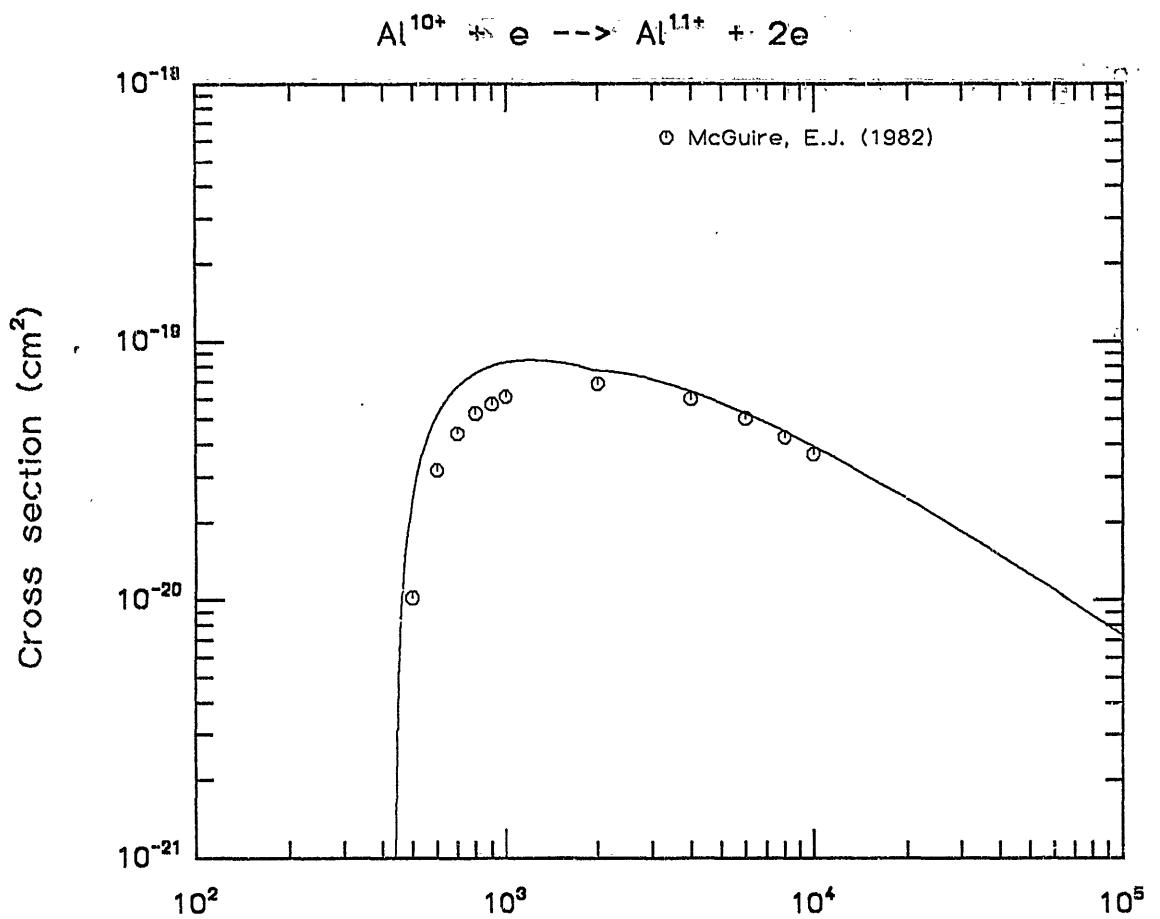
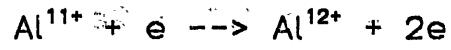


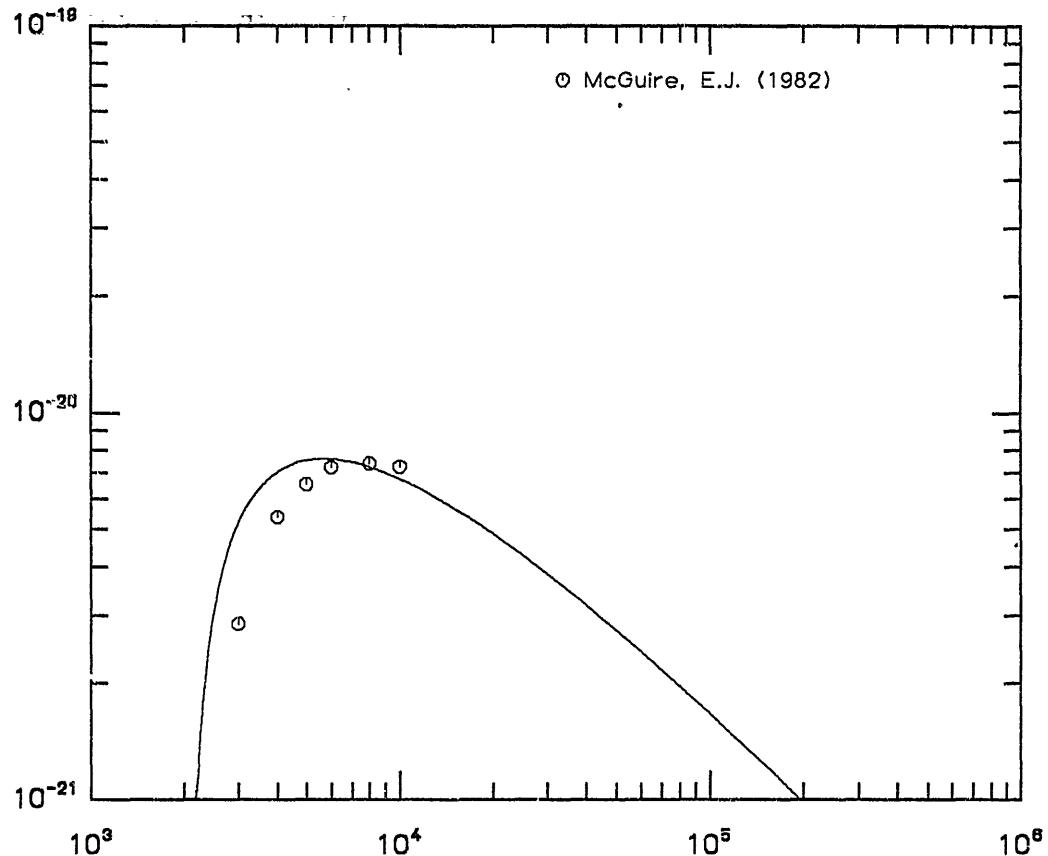
Fig. 96

Electron energy (eV)

McGuire, E.J. (1982)



● Cross section ( $\text{cm}^2$ )



Electron energy (eV)

● Cross section ( $\text{cm}^2$ )

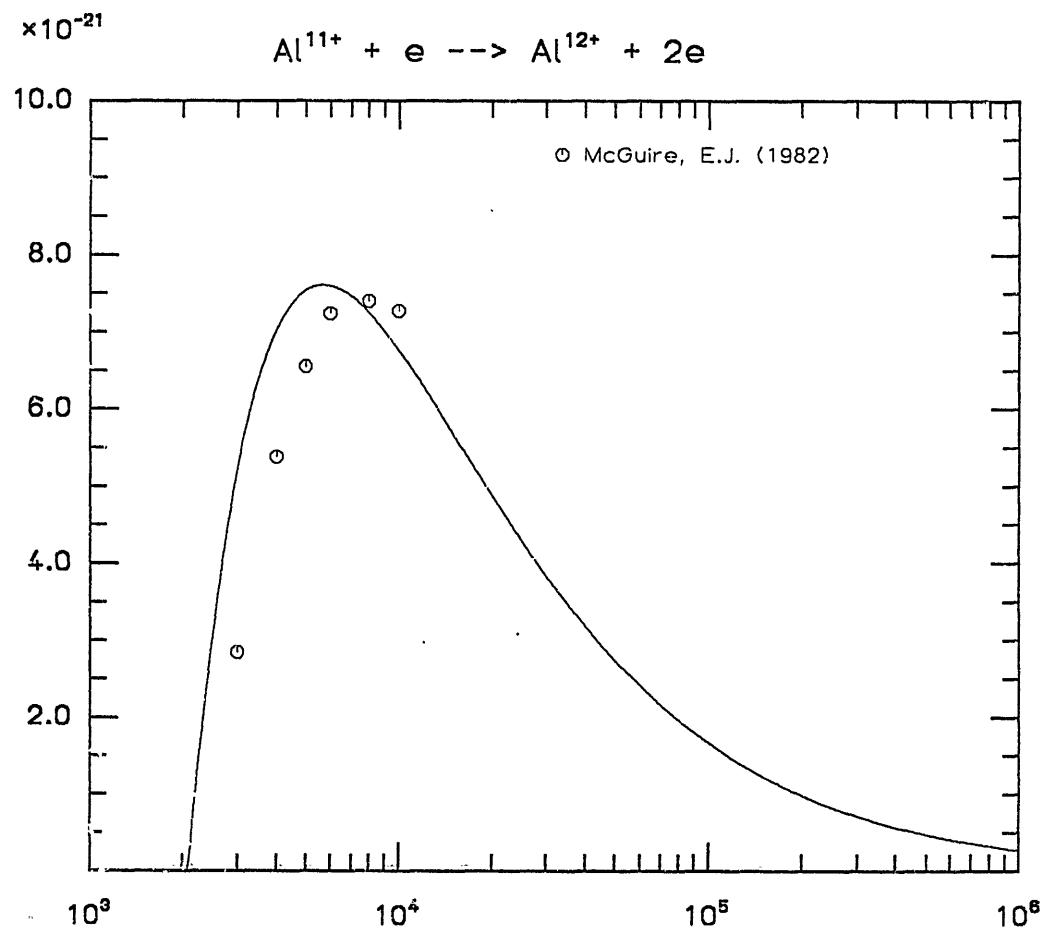


Fig. 97

Electron energy (eV)

8.0

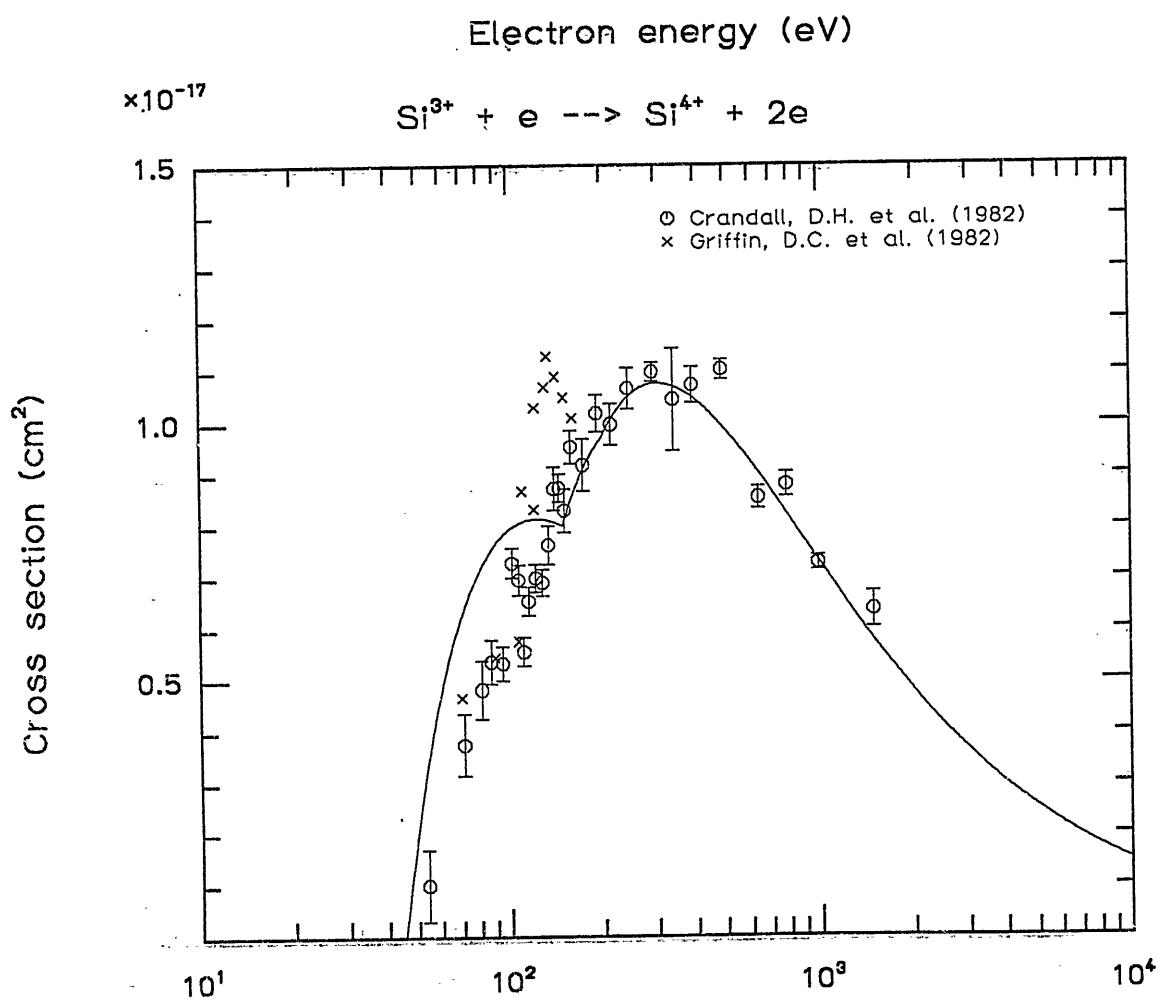
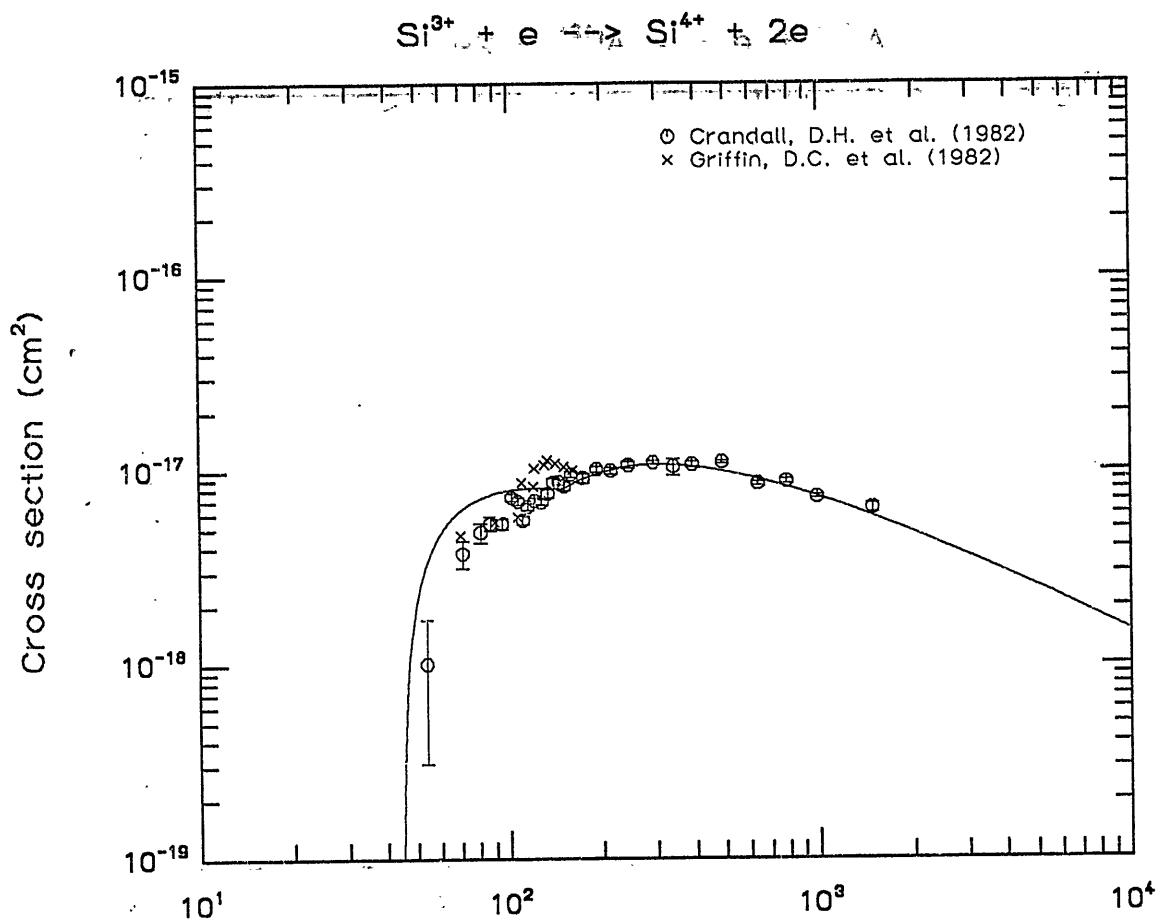
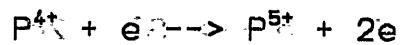


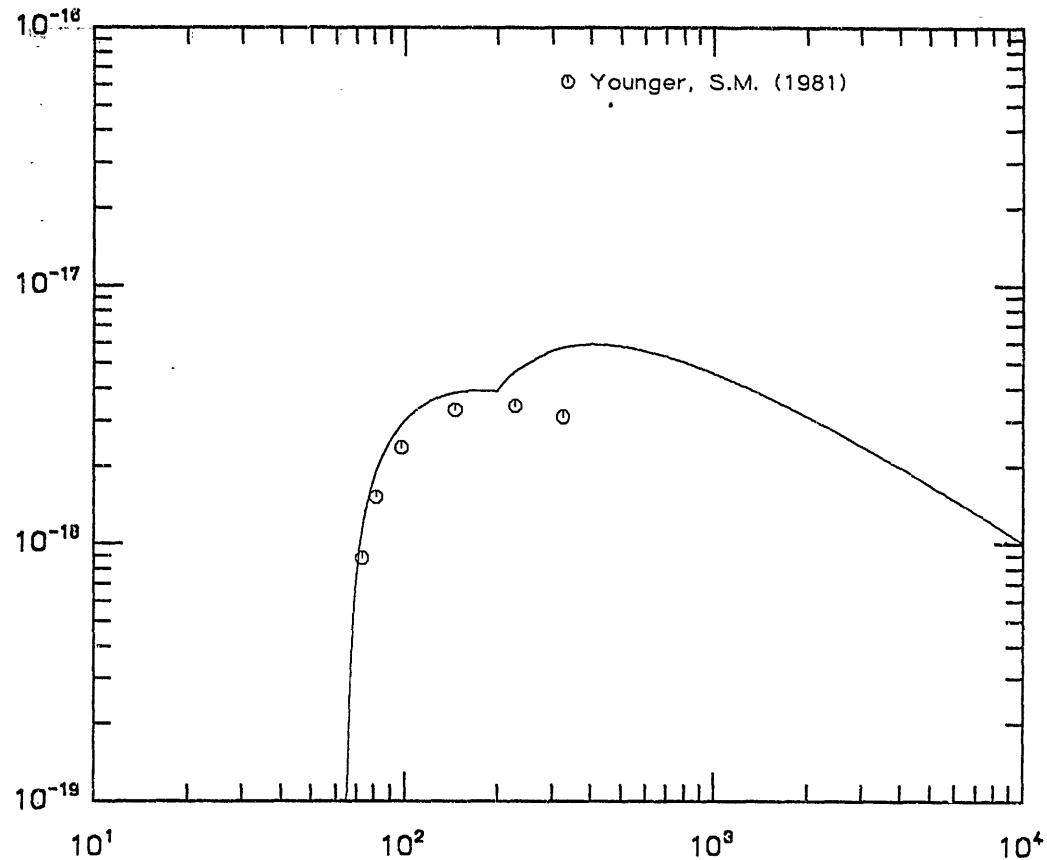
Fig. 98

Electron energy (eV)

$\text{e}^- \text{ erg}$



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

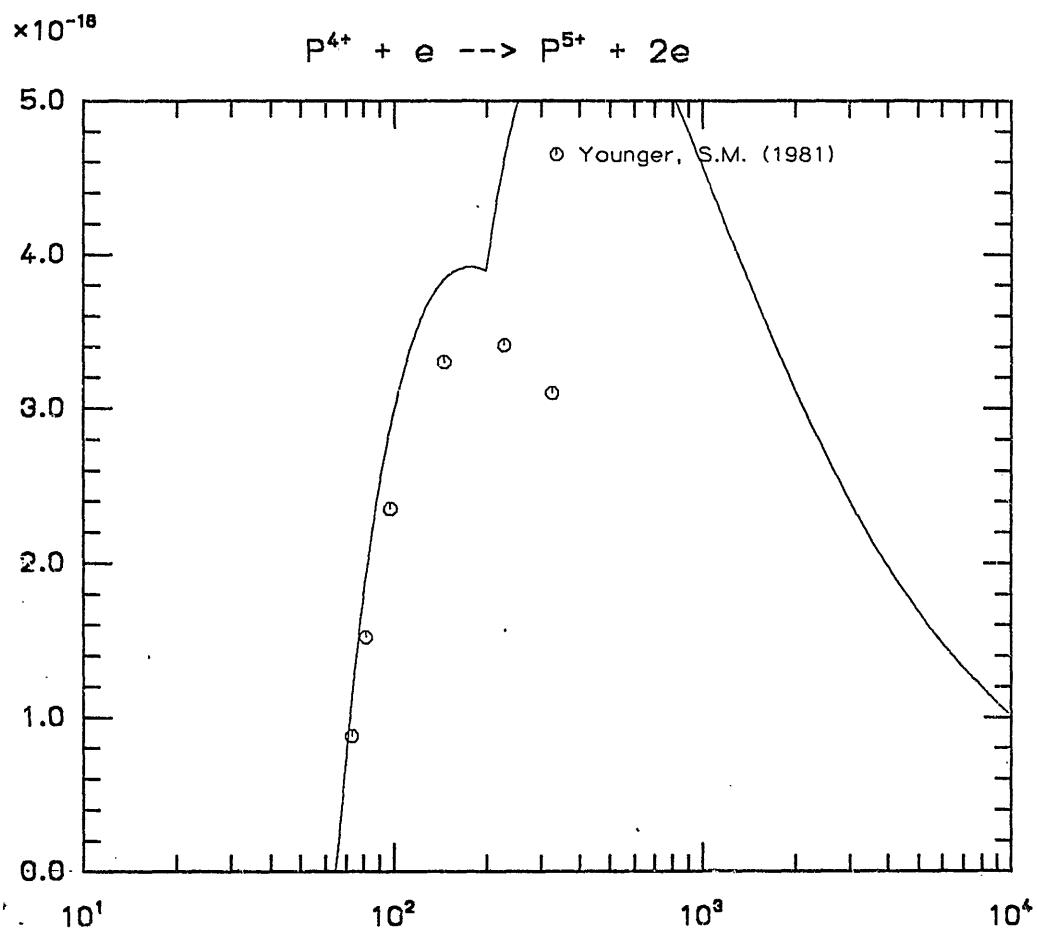


Fig. 99

Electron energy (eV)

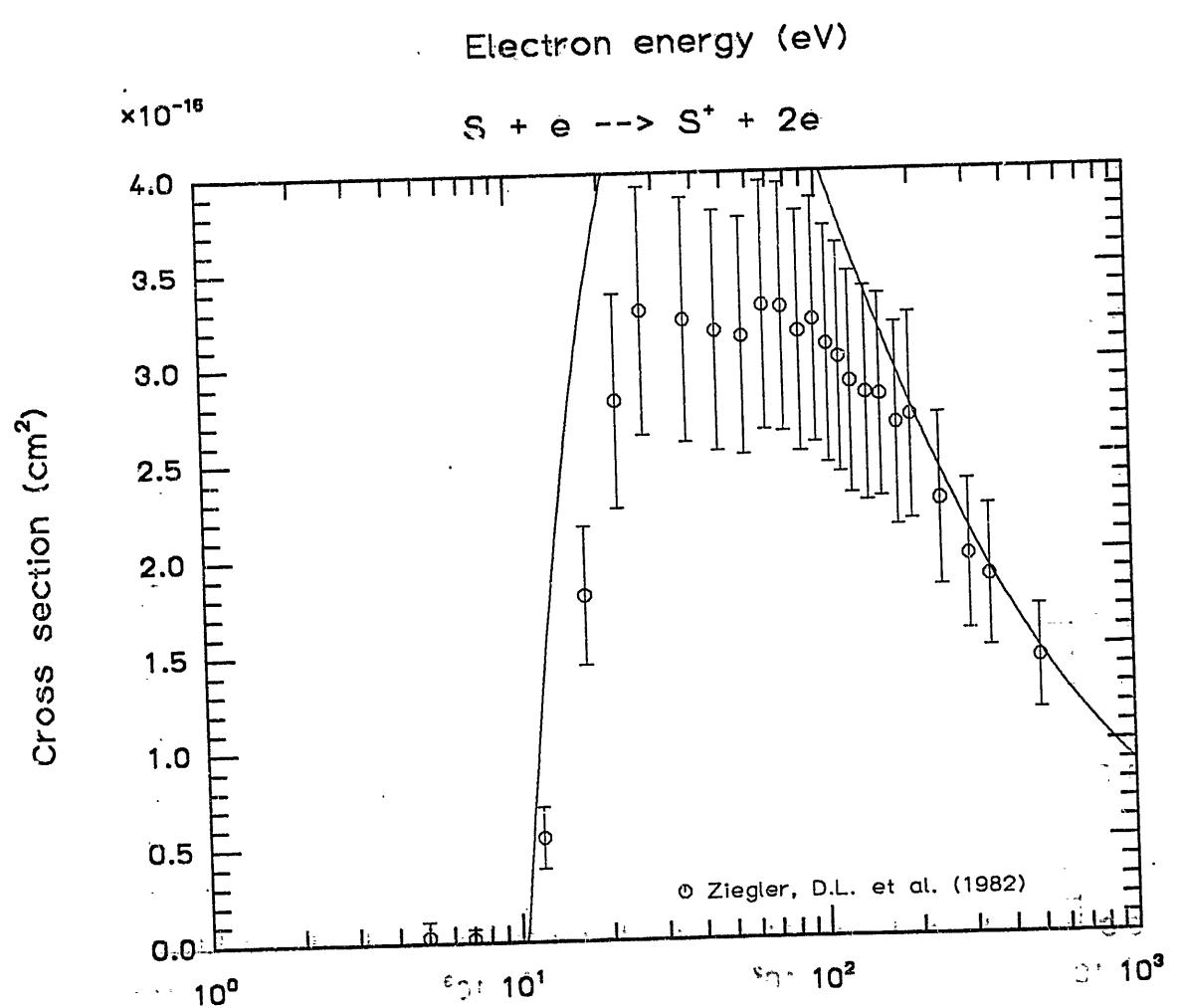
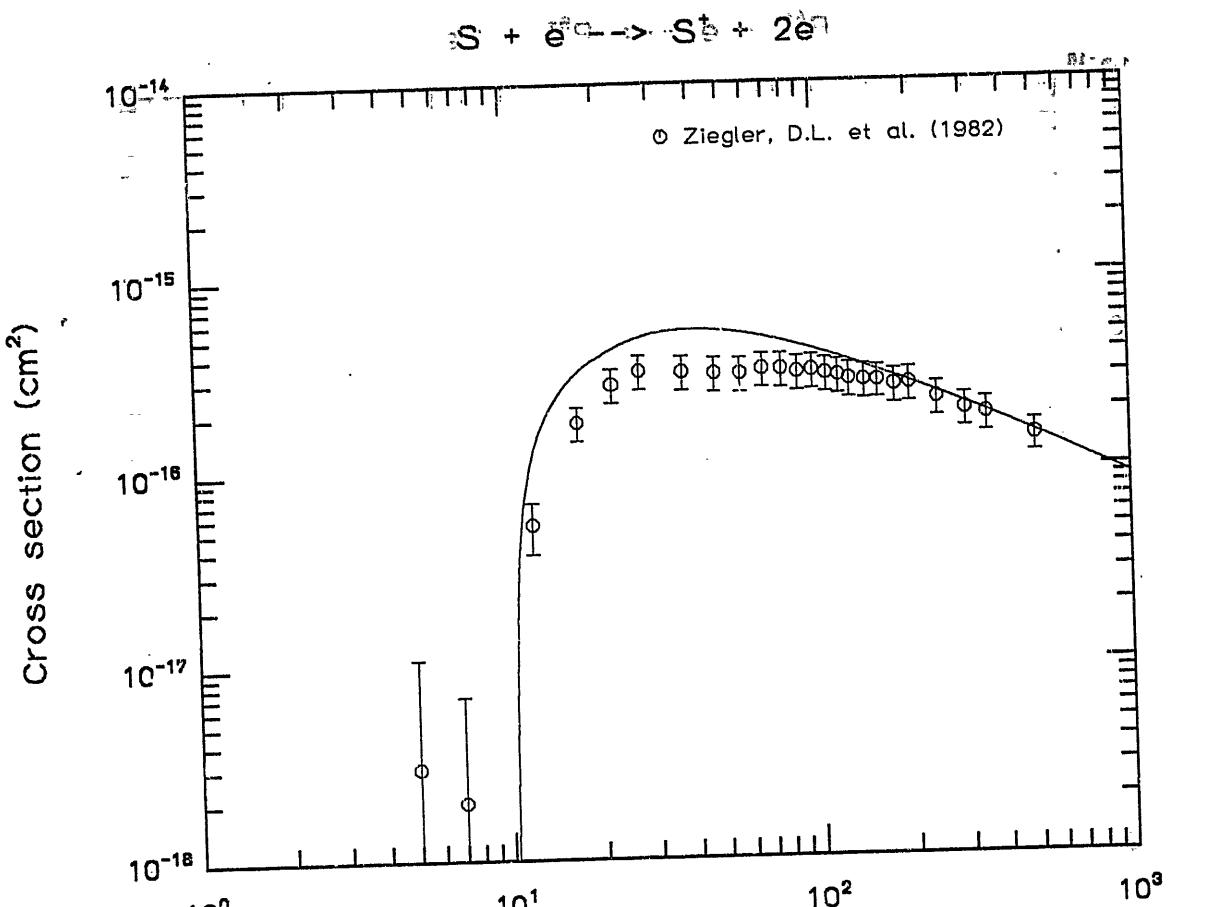
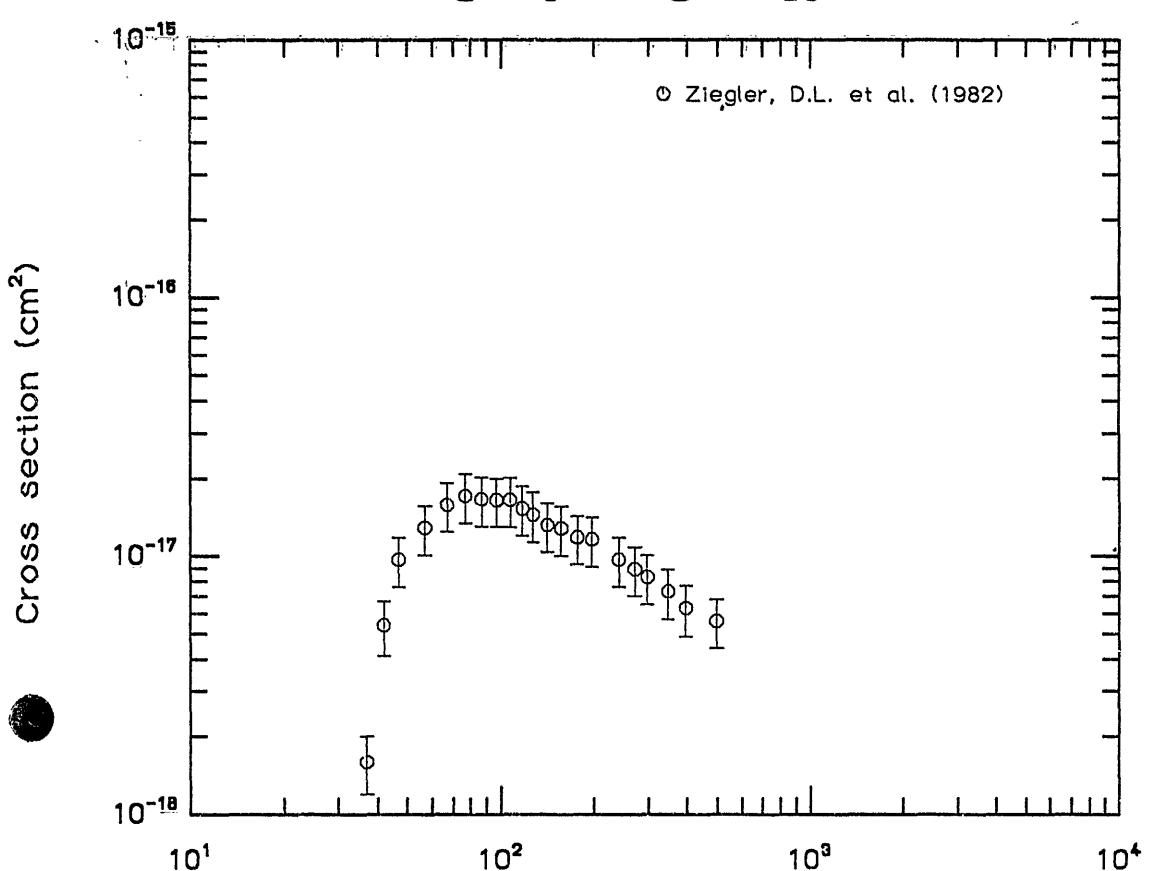
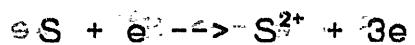


Fig. 100 (ve) Electron energy (eV)  $\text{ee} \cdot \text{pF}$



Electron energy (eV)

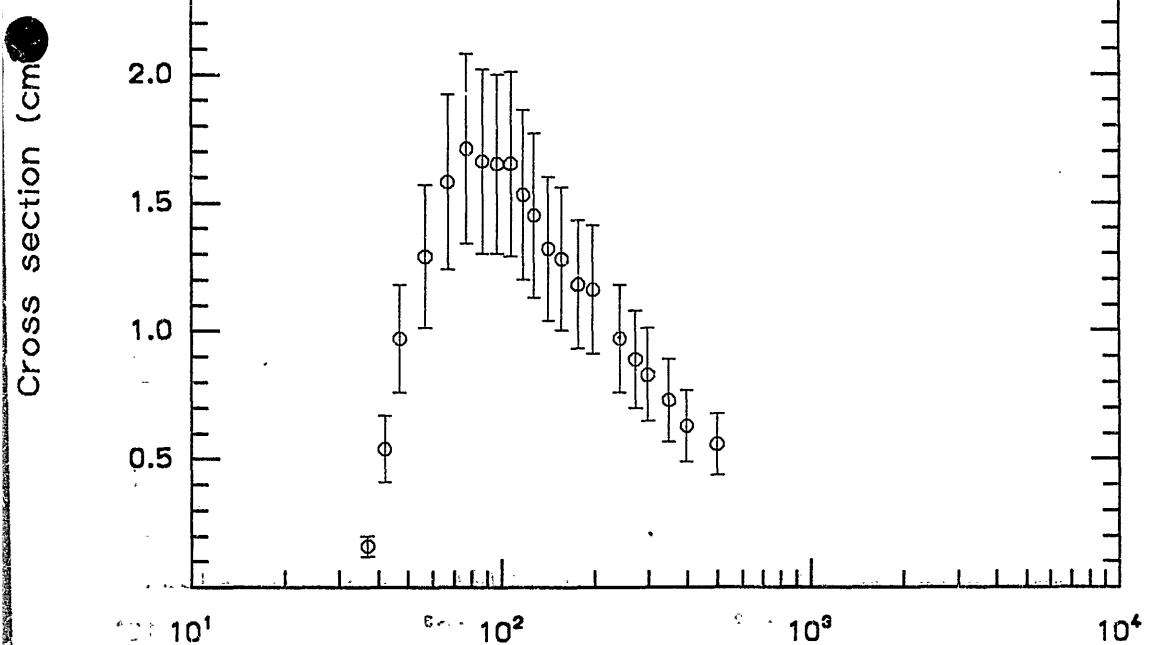
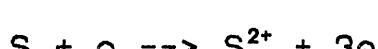


Fig. 101 (a) Electron energy (eV) (b) 10<sup>1</sup> - 10<sup>4</sup>

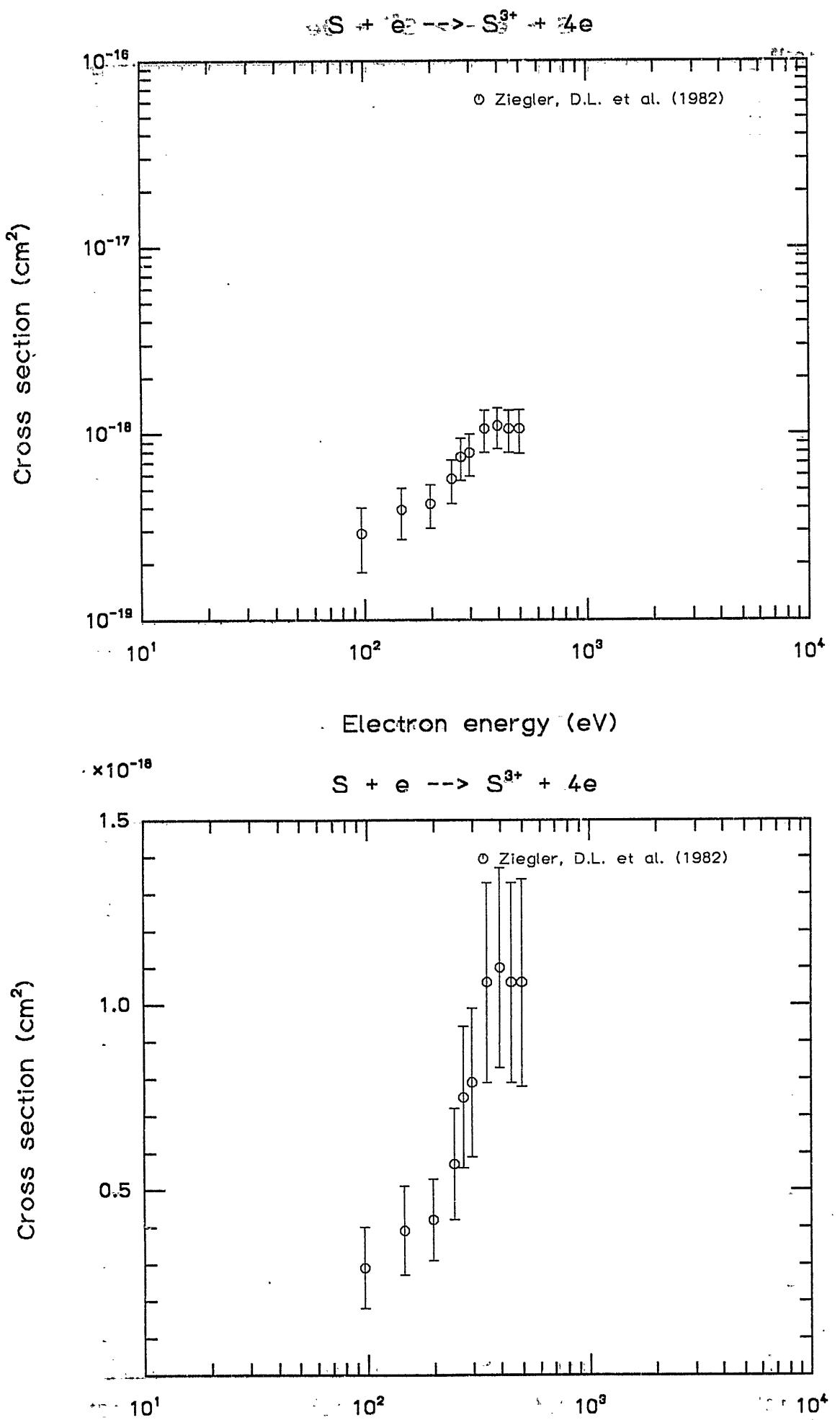
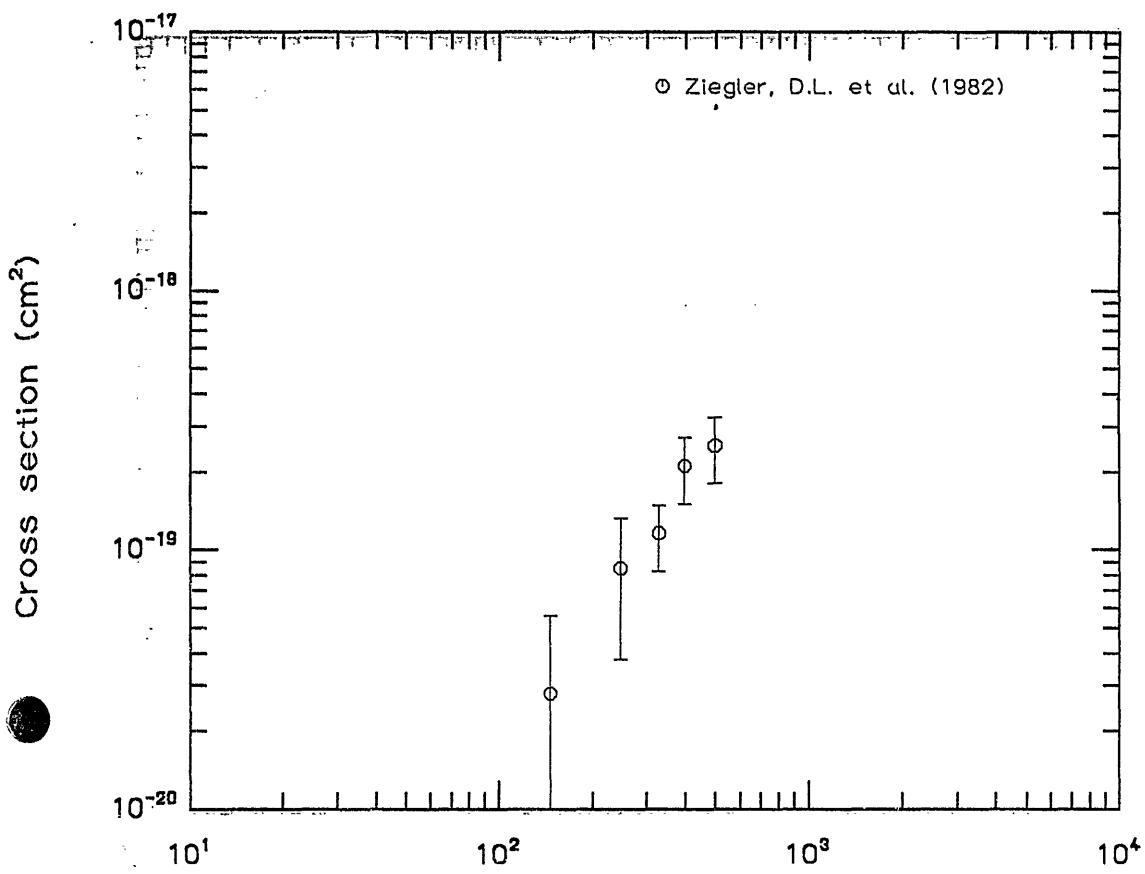
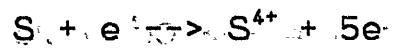


Fig. 102. Cross section vs Electron energy (eV) for the reactions



Electron energy (eV)

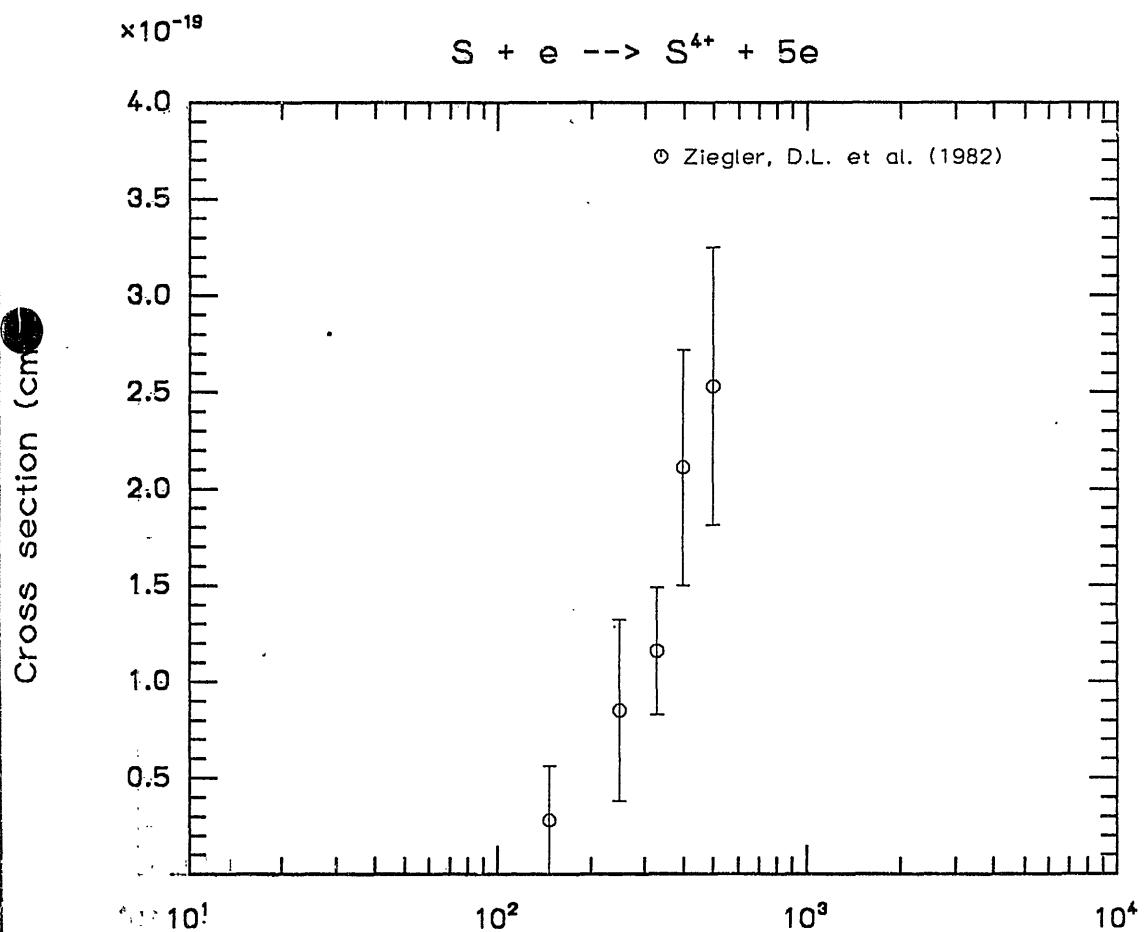


Fig. 103

Electron energy (eV)

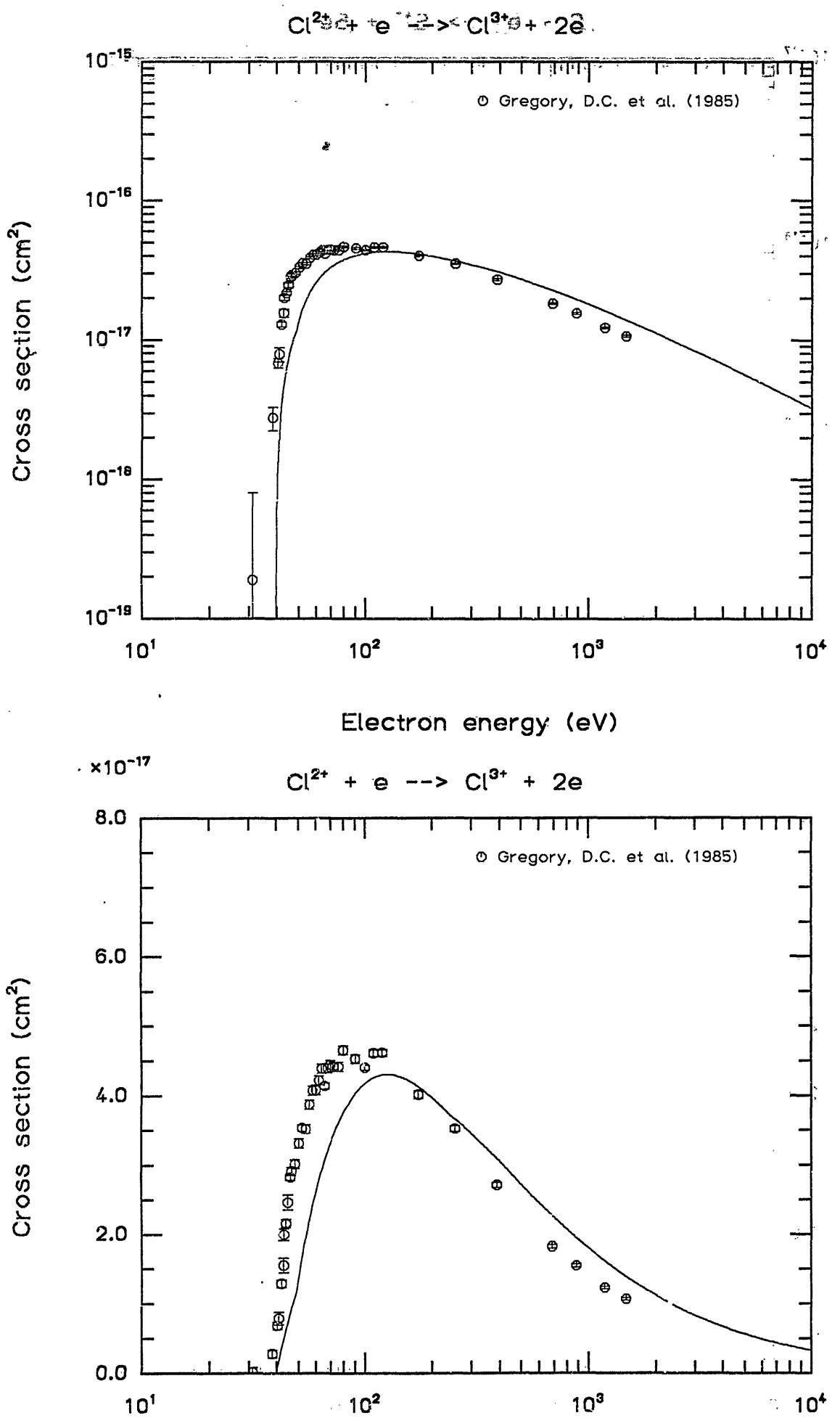
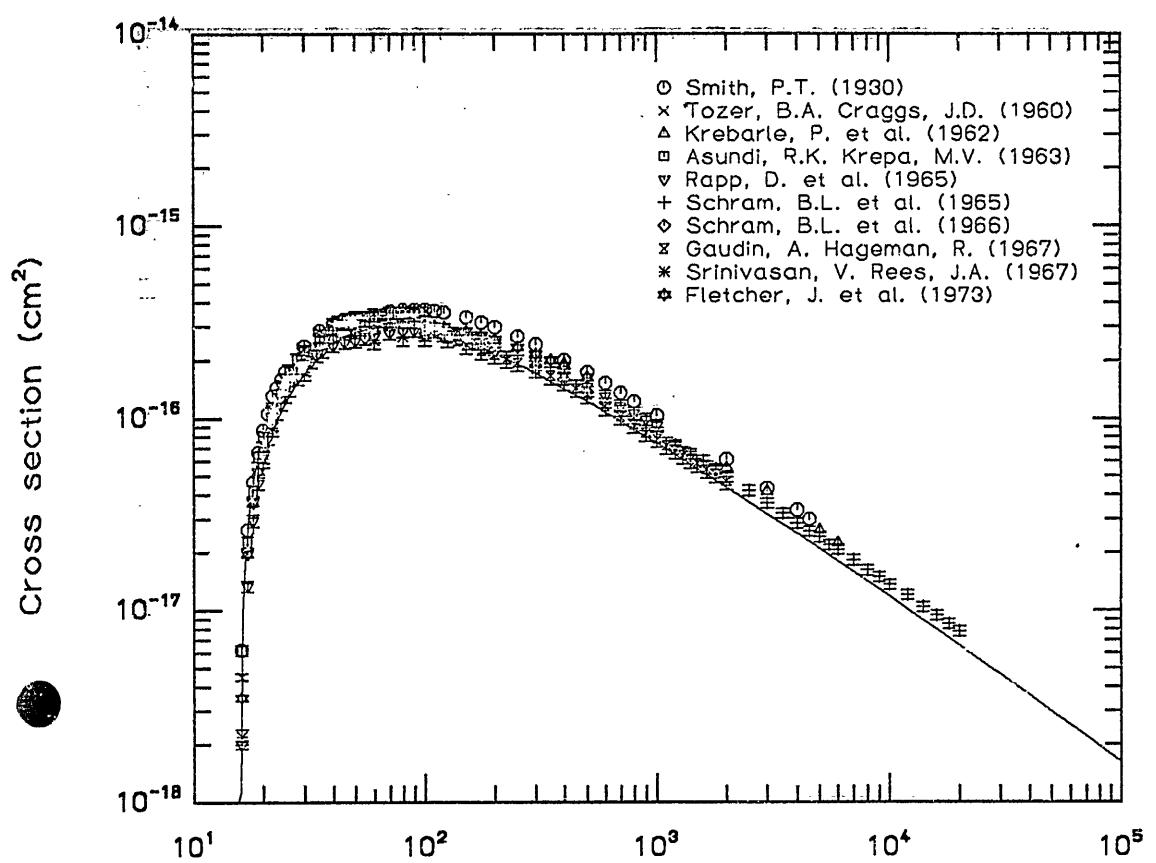
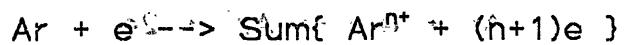


Fig. 104      Electron energy (eV)      8011 p.1



Electron energy (eV)

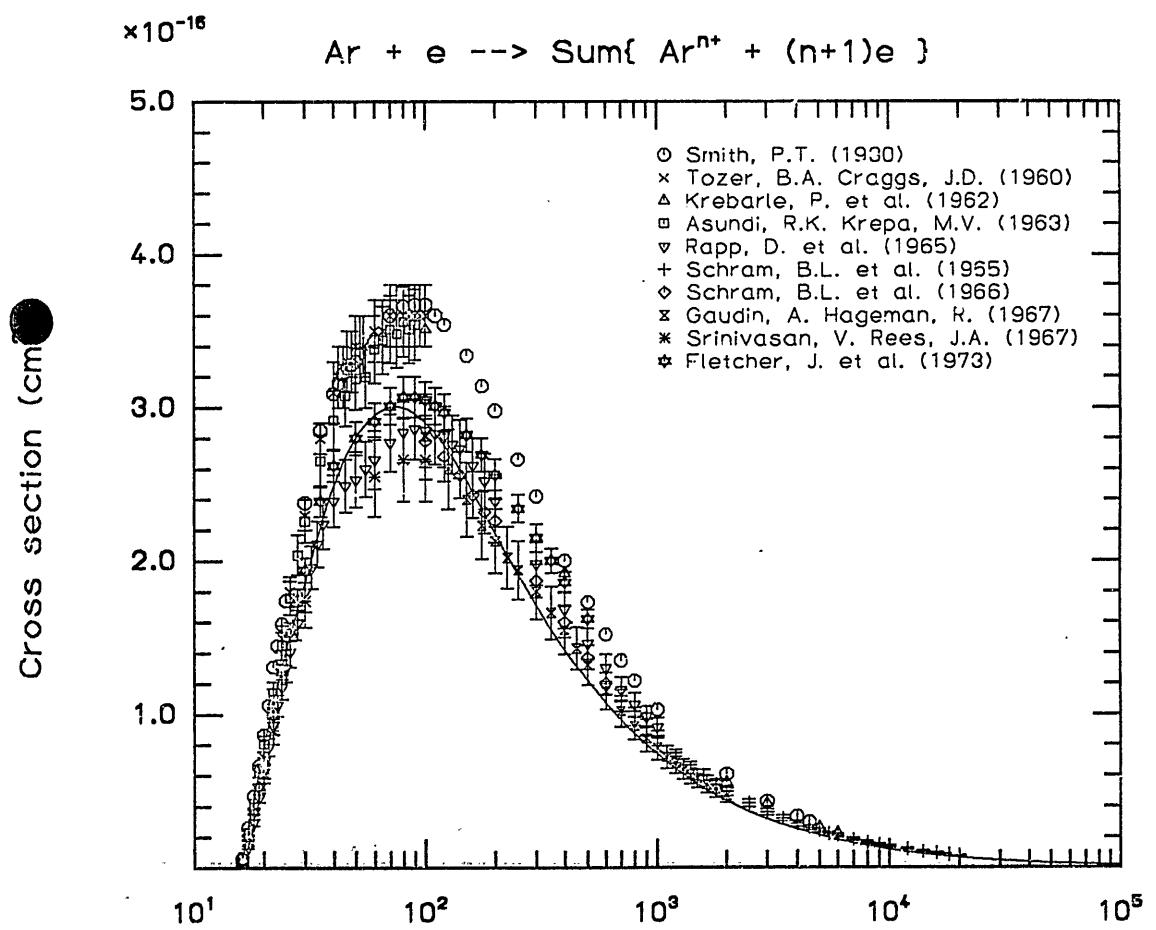


Fig. 105      Electron energy (eV)

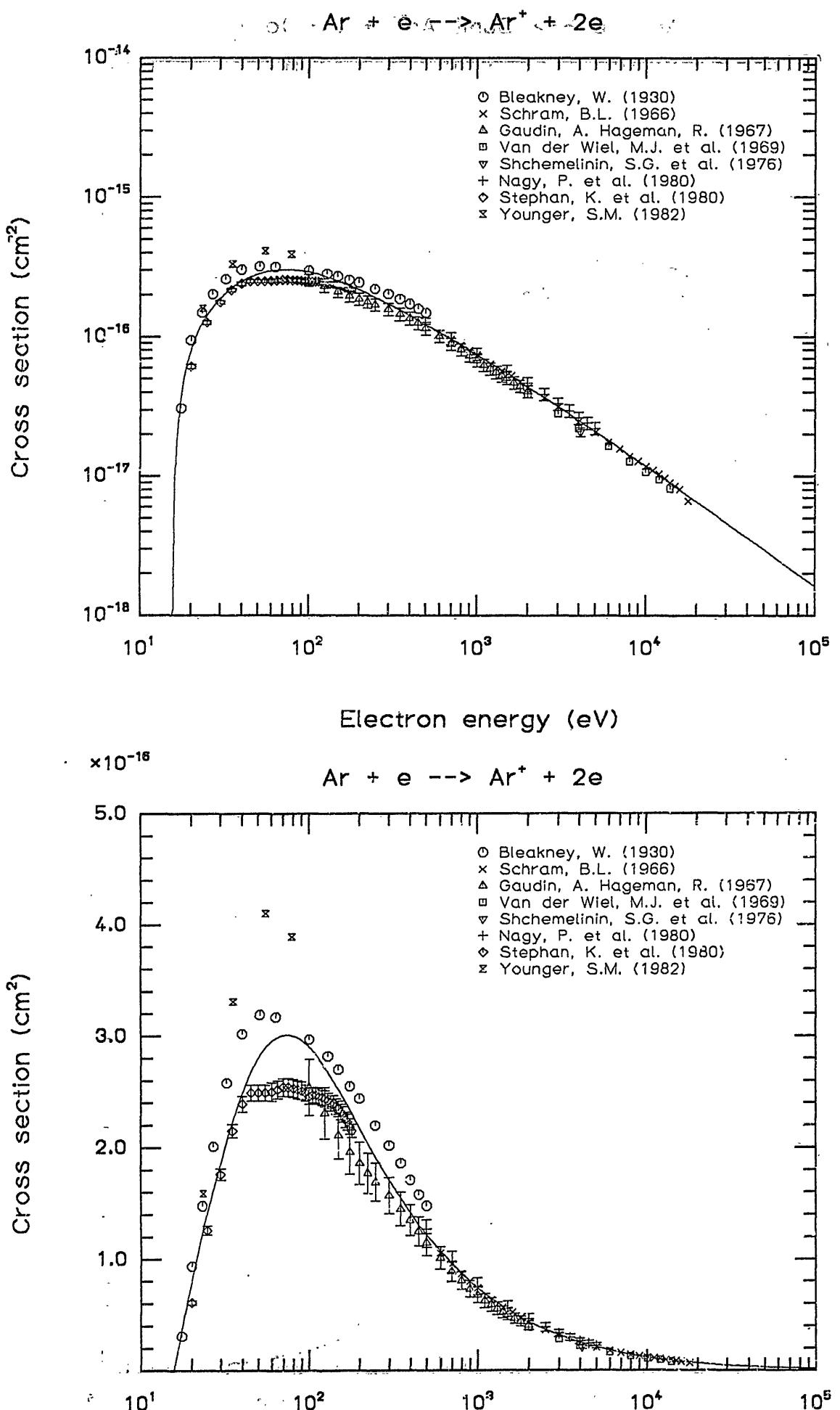
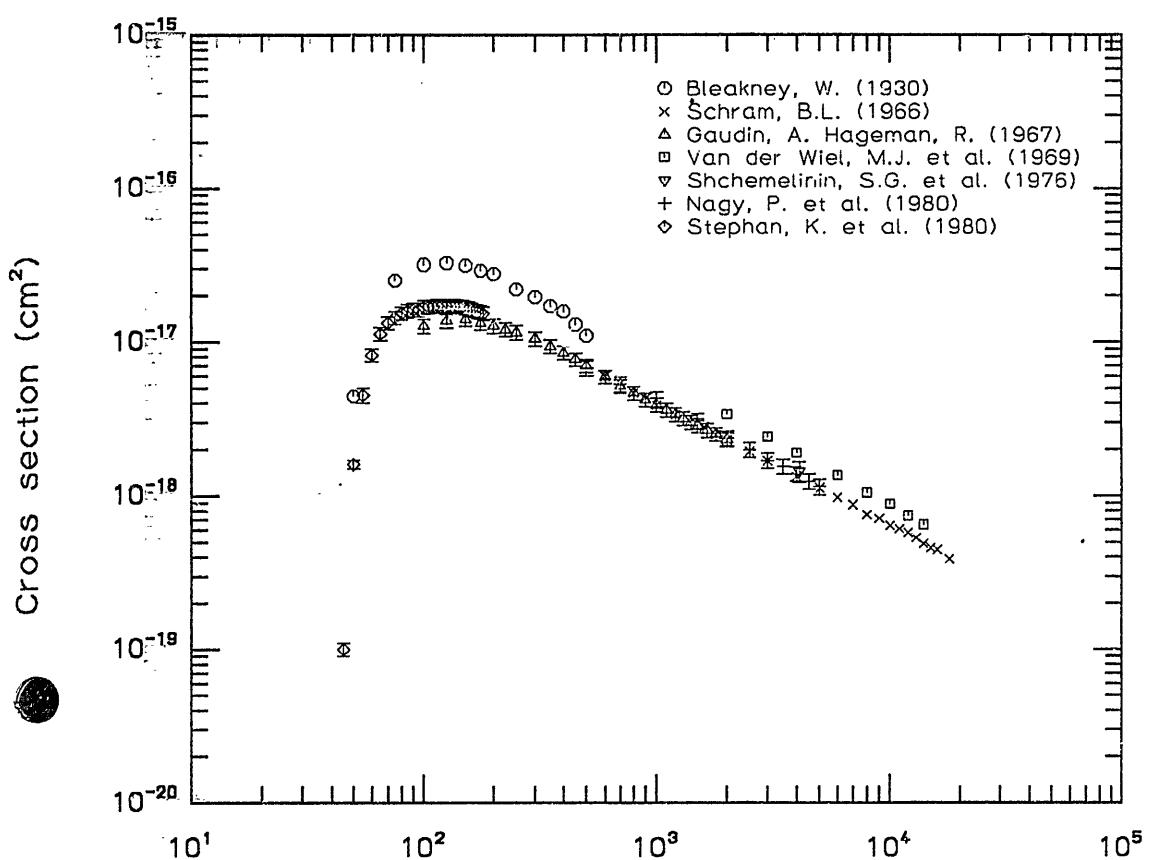
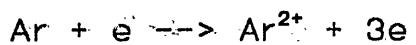


Fig. 106 Electron energy (eV) 201 pF



Electron energy (eV)

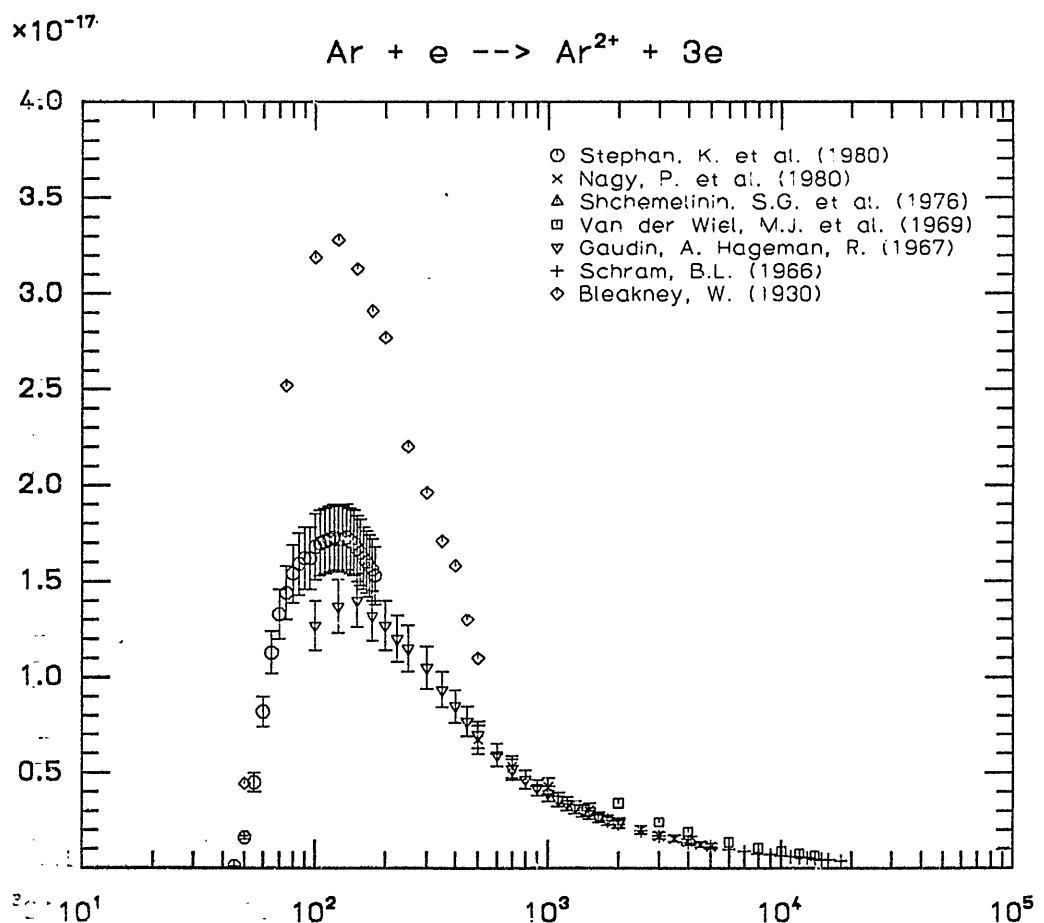


Fig. 107

Electron energy (eV)

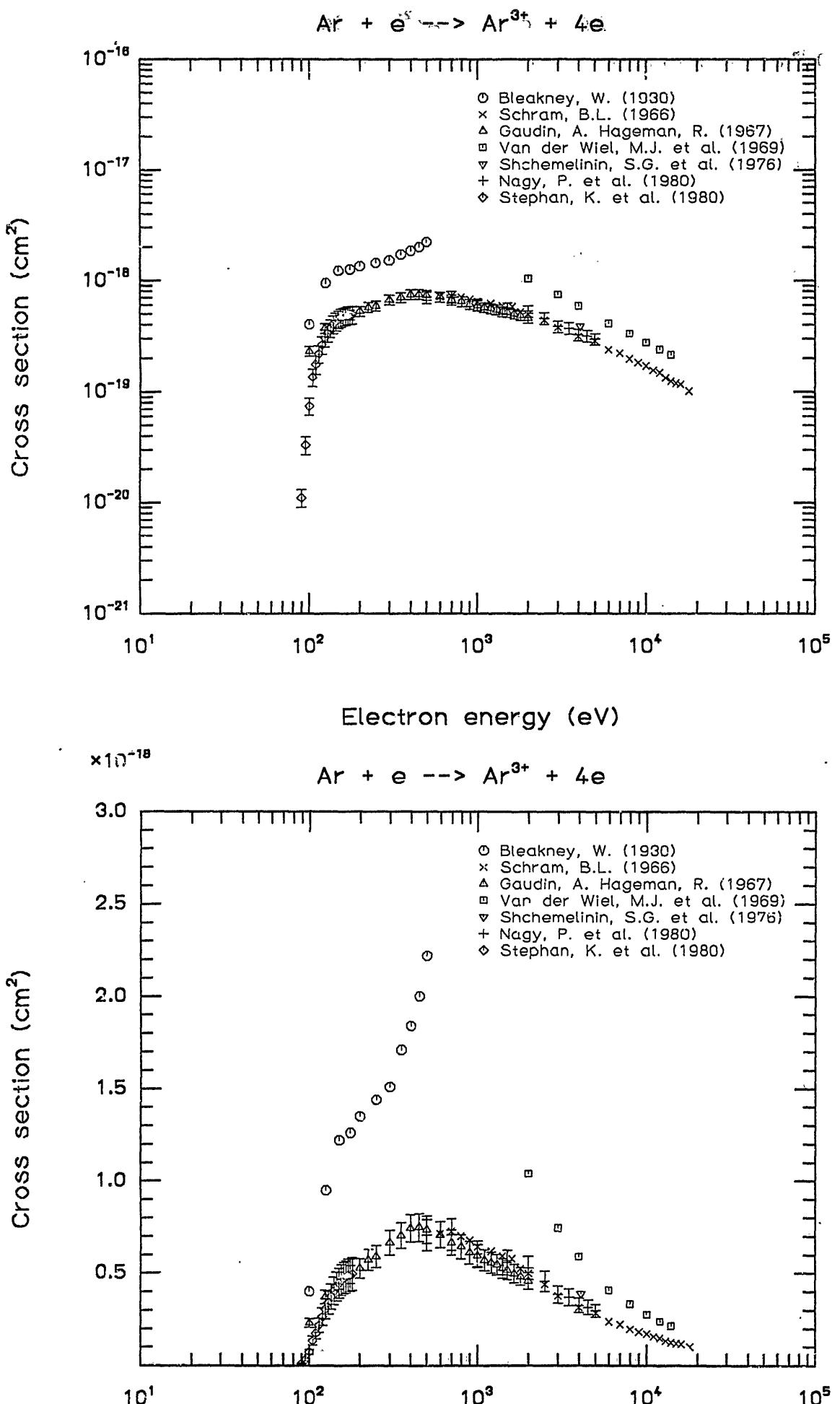
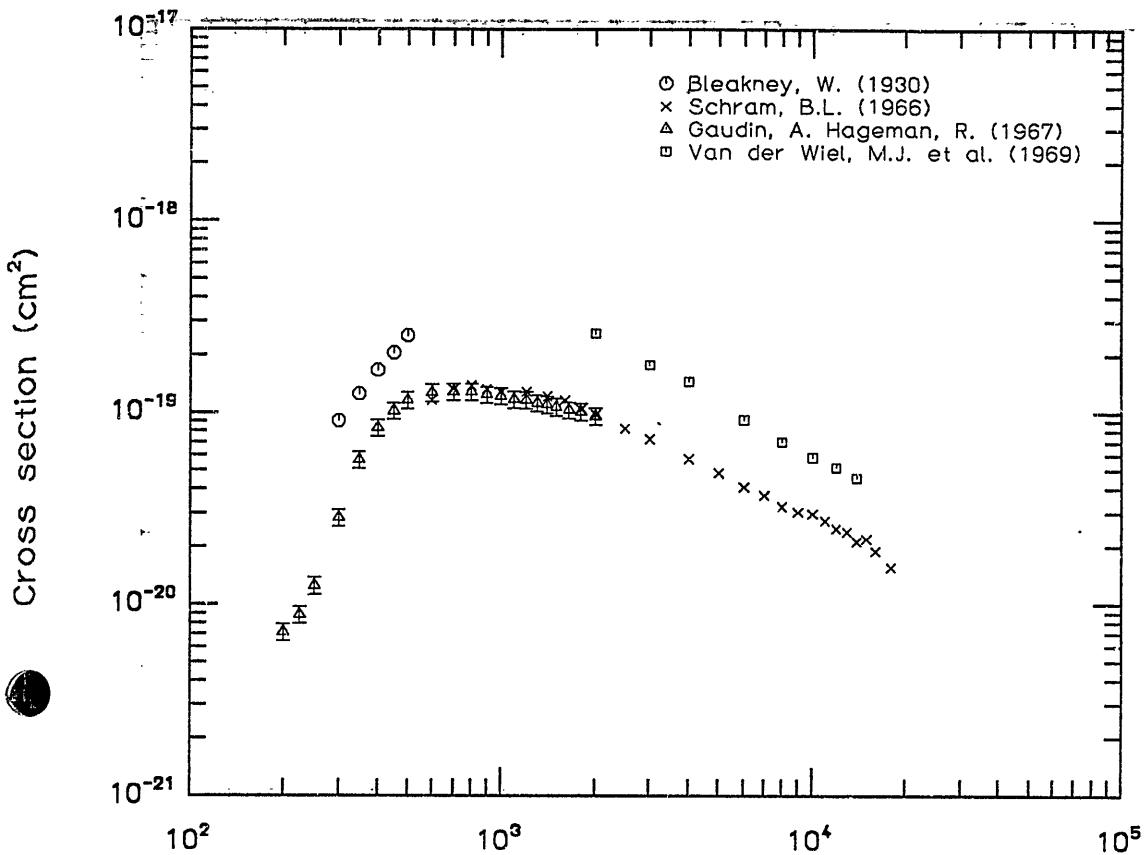
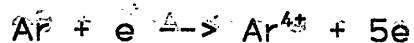


Fig. 108      Electron energy (eV)



Electron energy (eV)

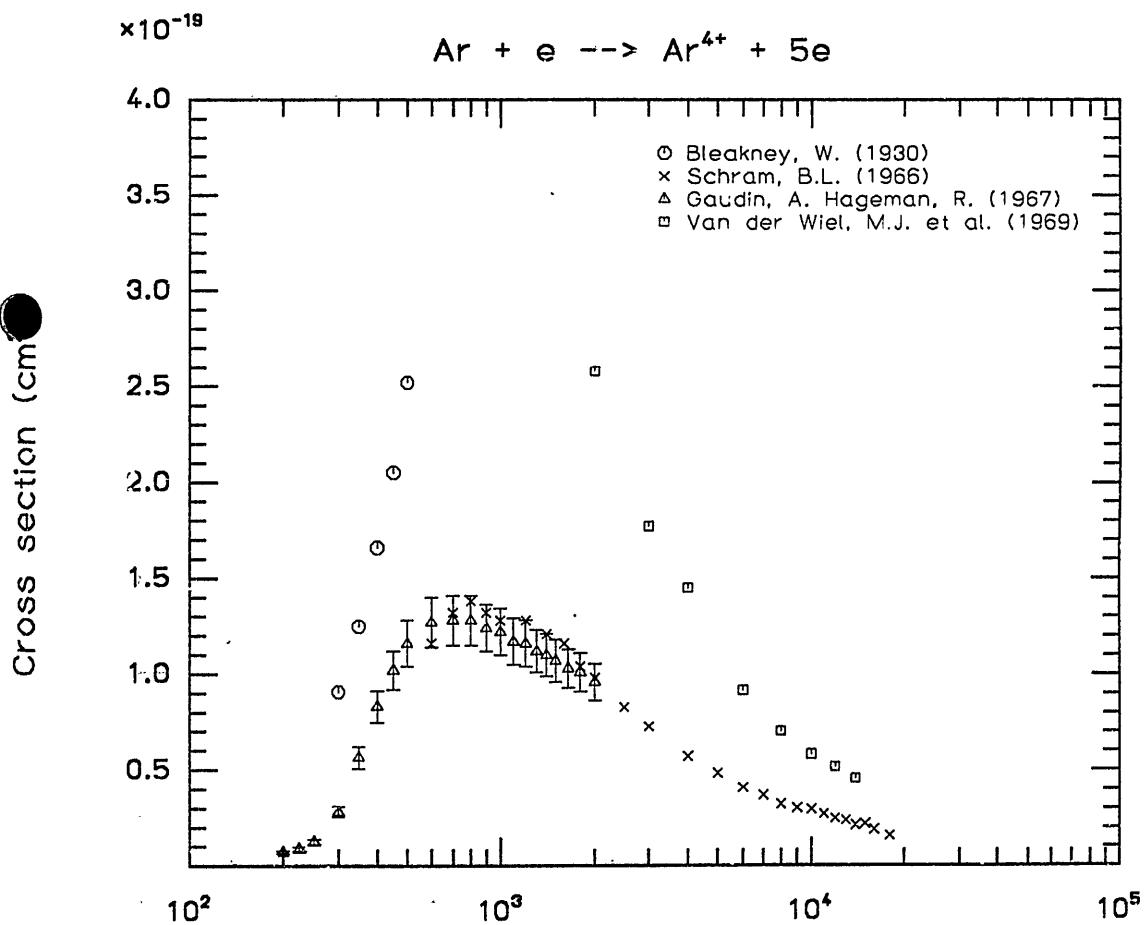


Fig. 109

Electron energy (eV)

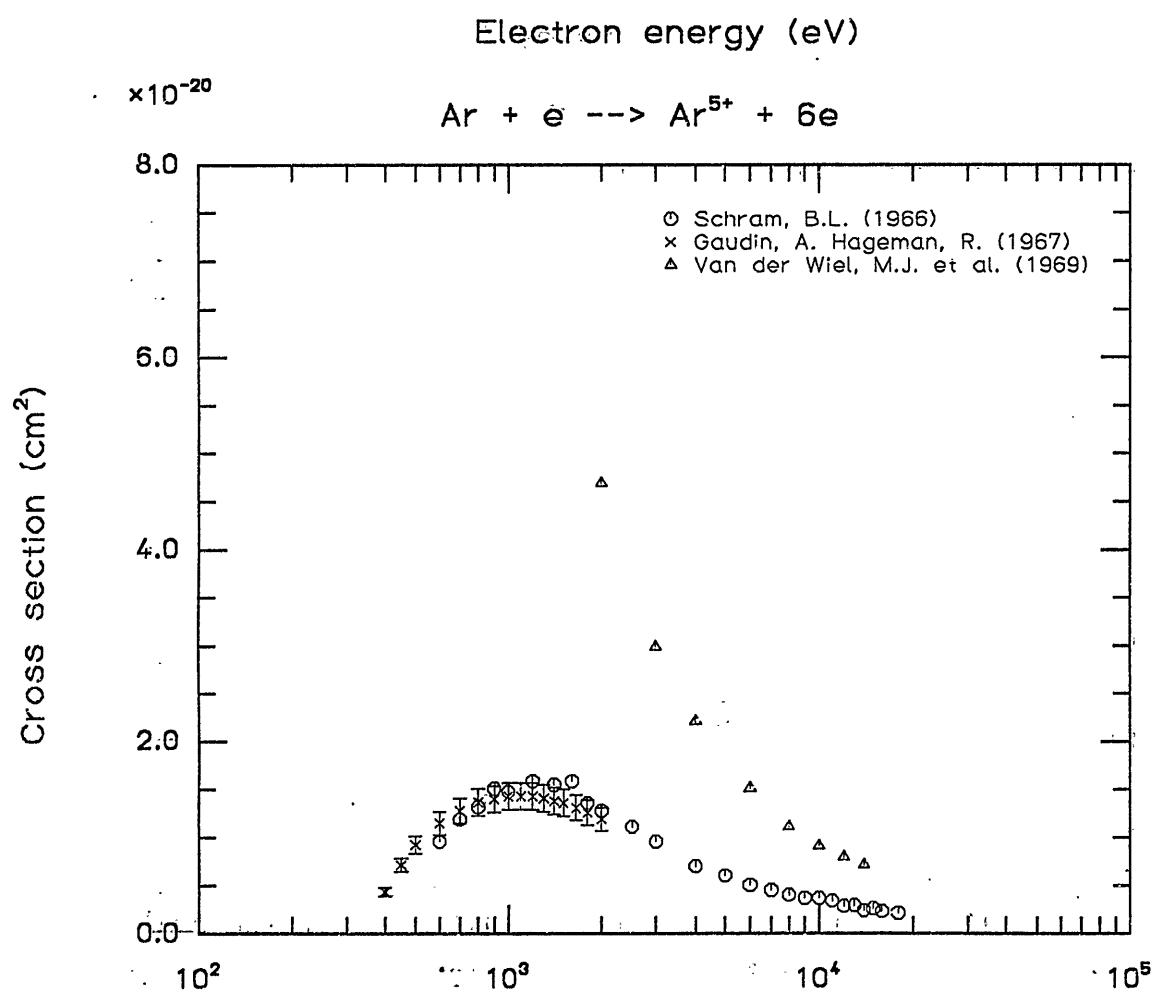
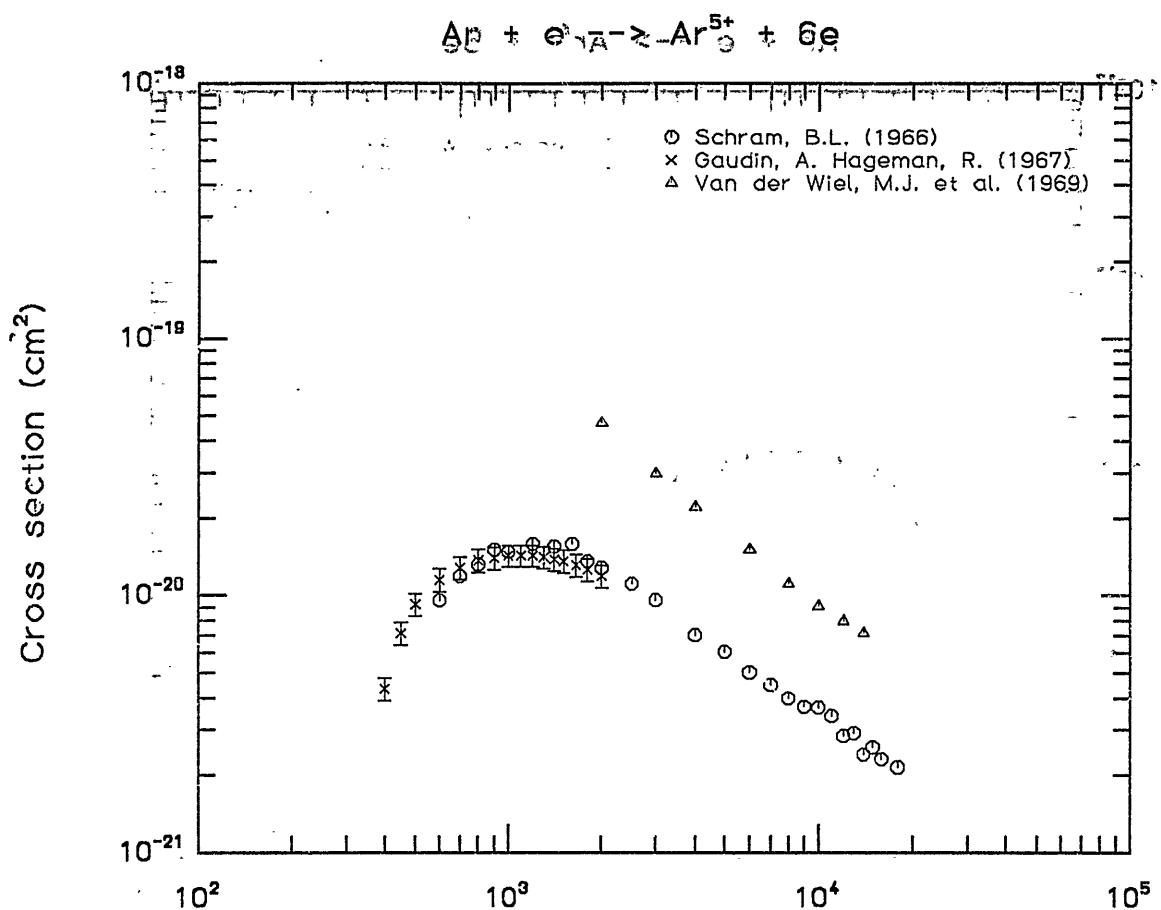
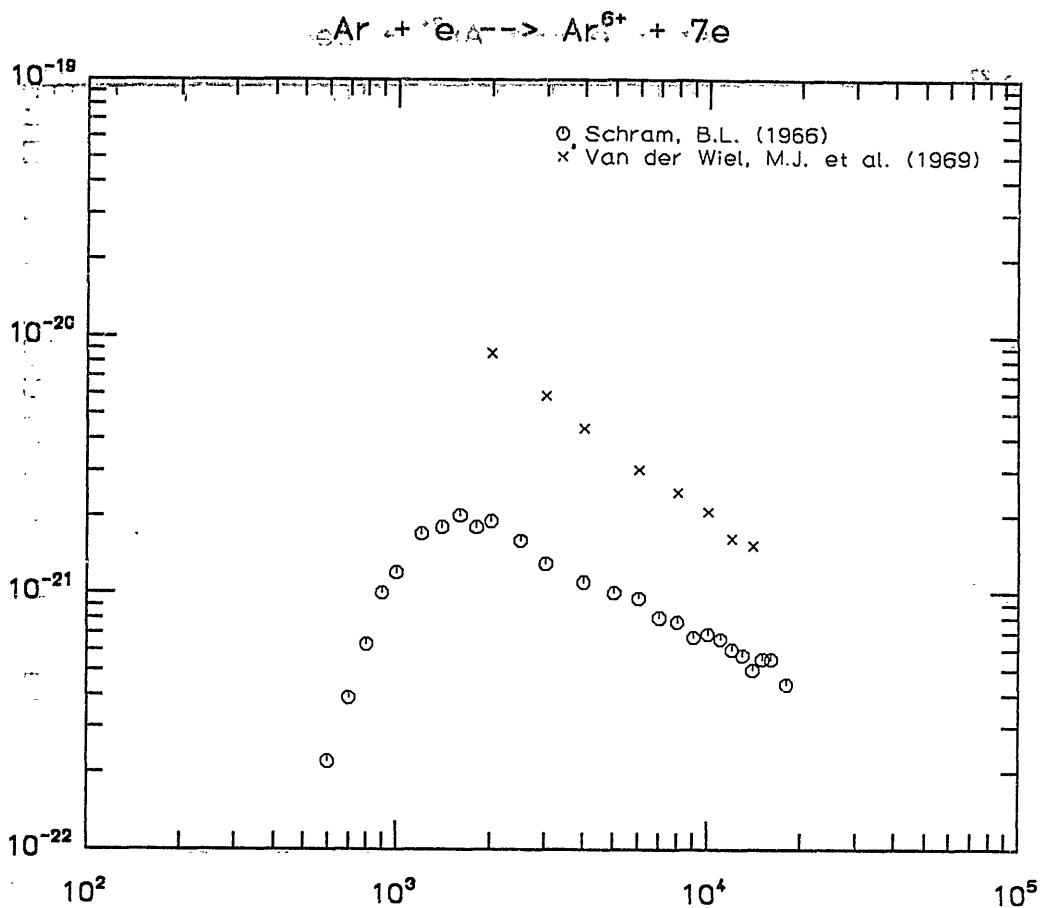


Fig. 110 (v) Electron energy (eV) CCT p17



Electron energy (eV)

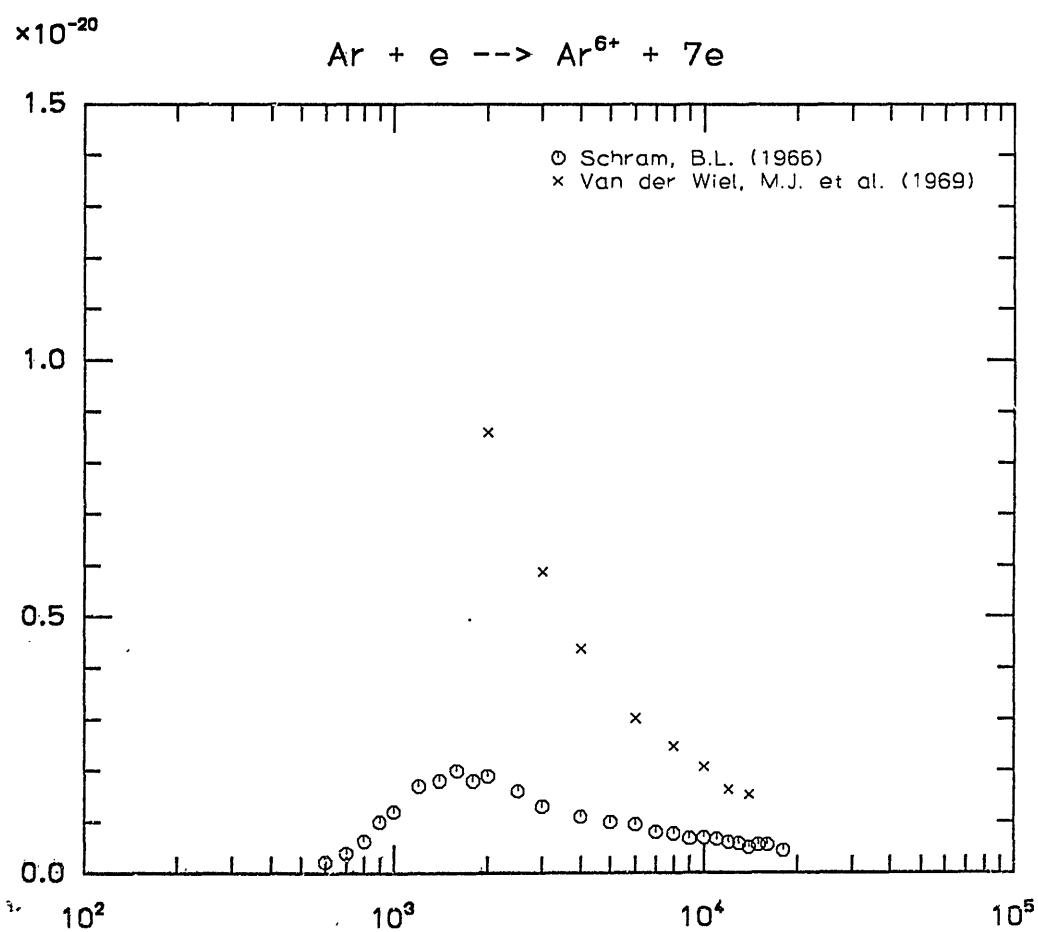


Fig. 111      Electron energy (eV)

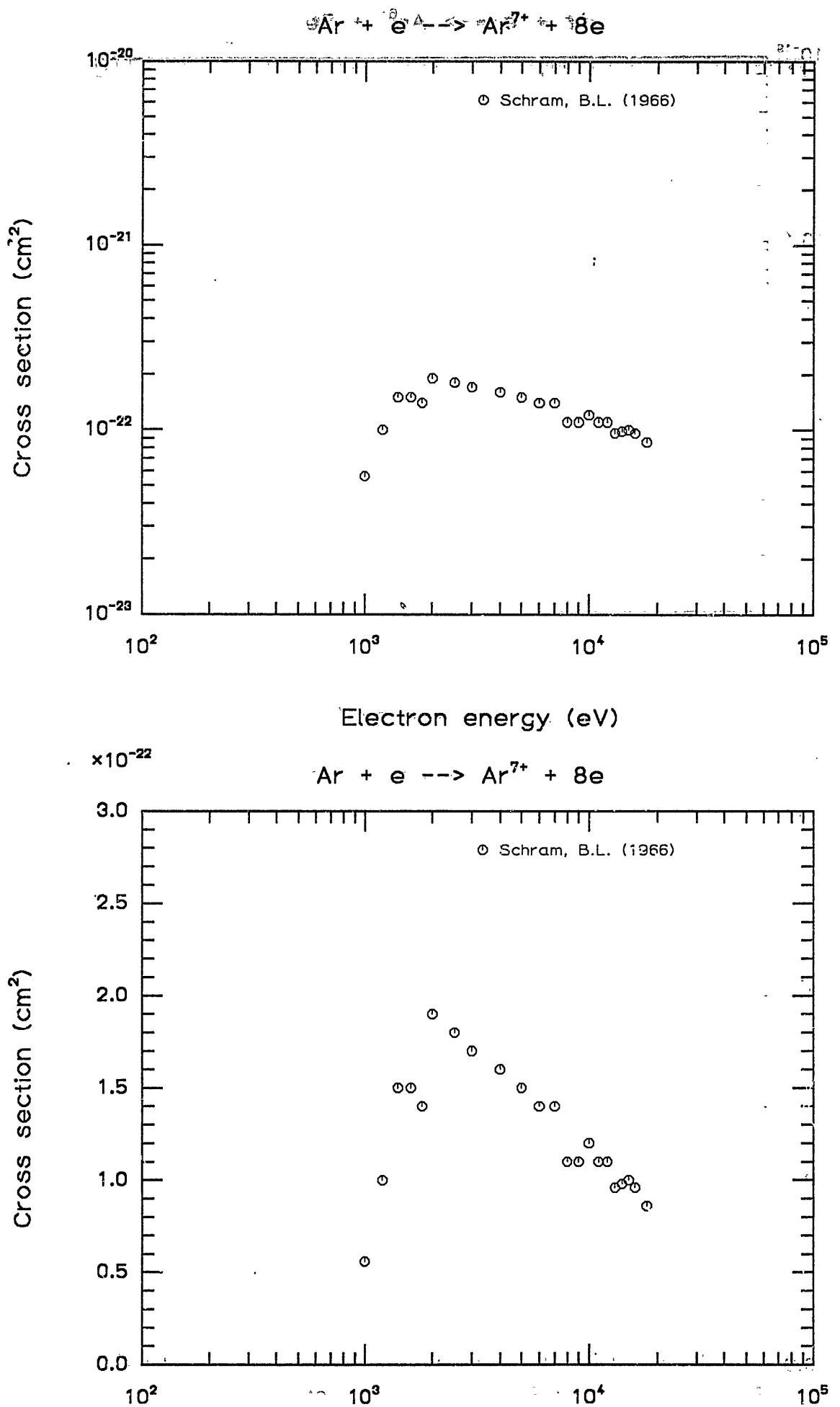


Fig. 112. Cross section vs. Electron energy (eV) for the ionization of argon.

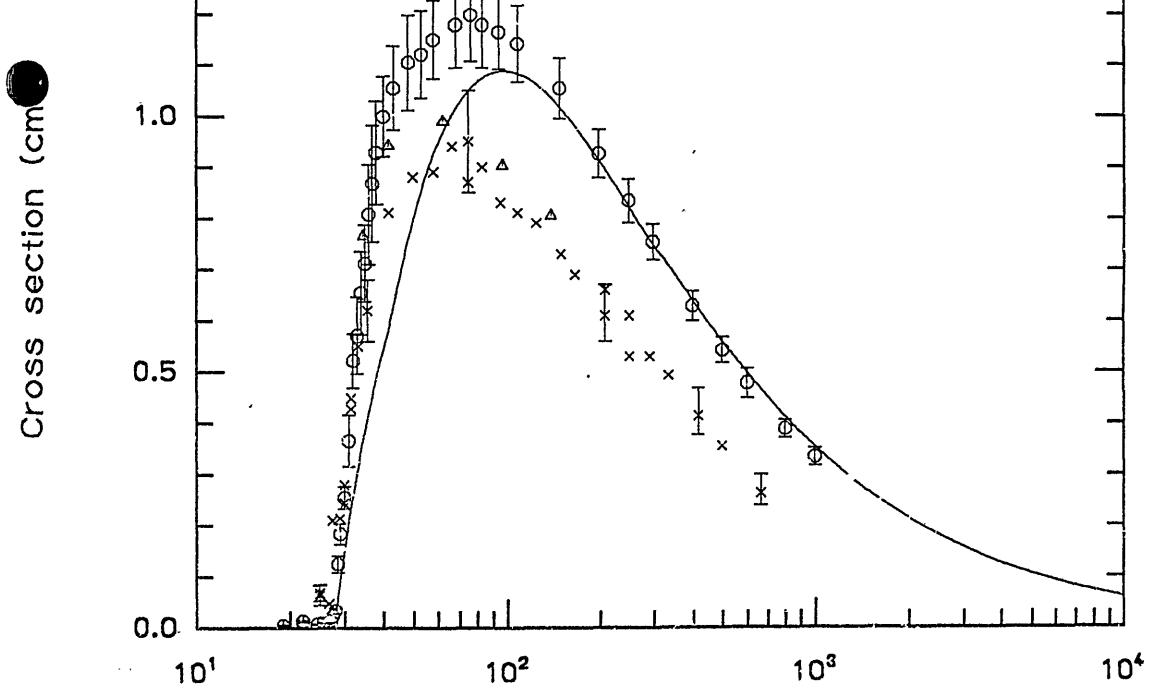
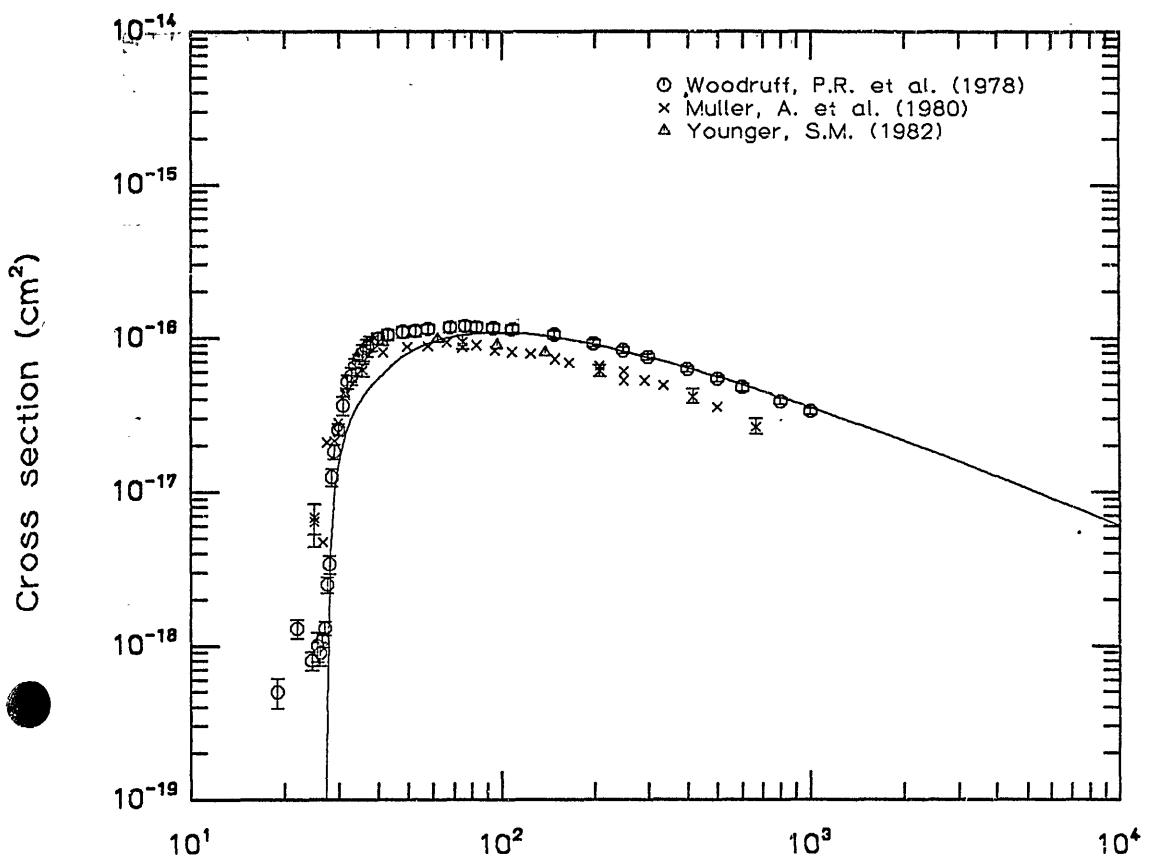
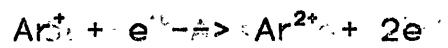


Fig. 113

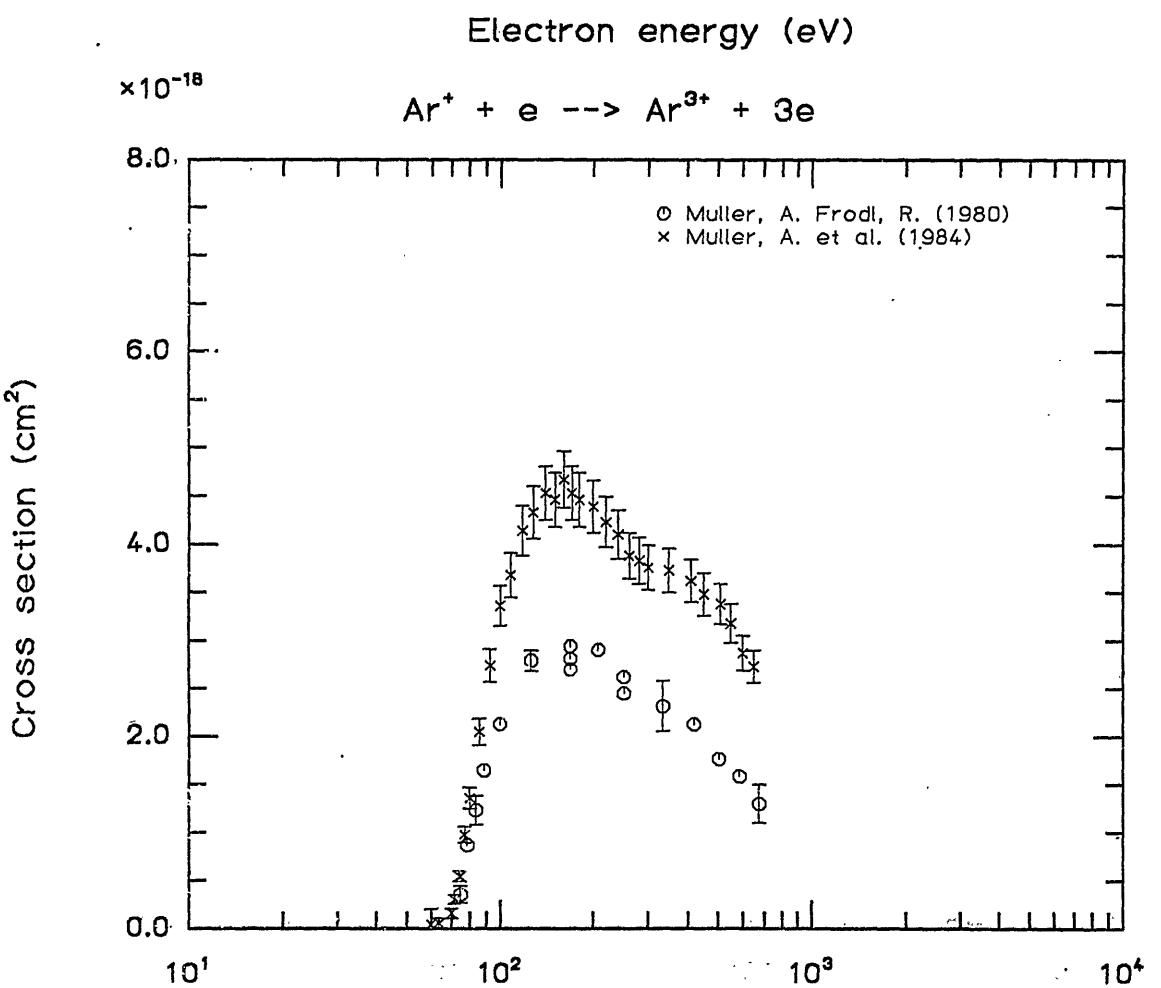
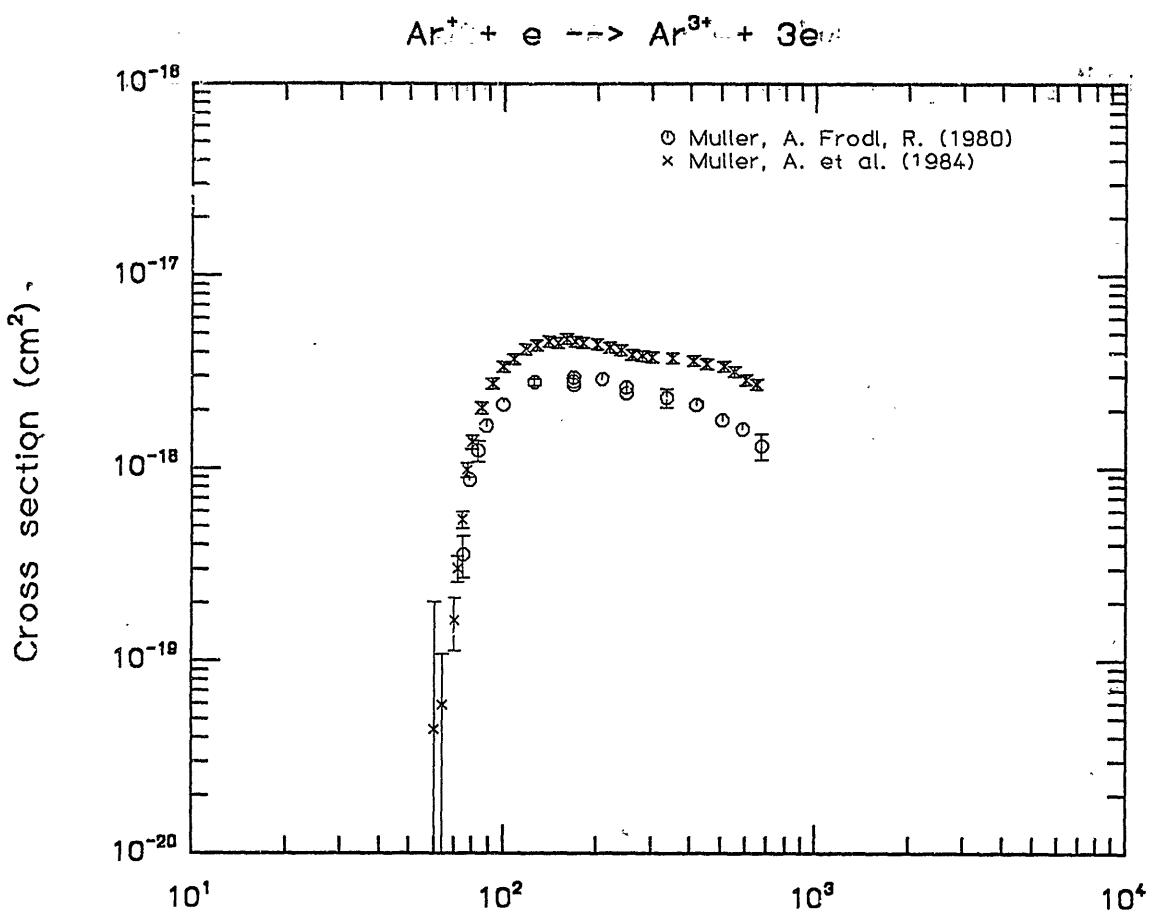
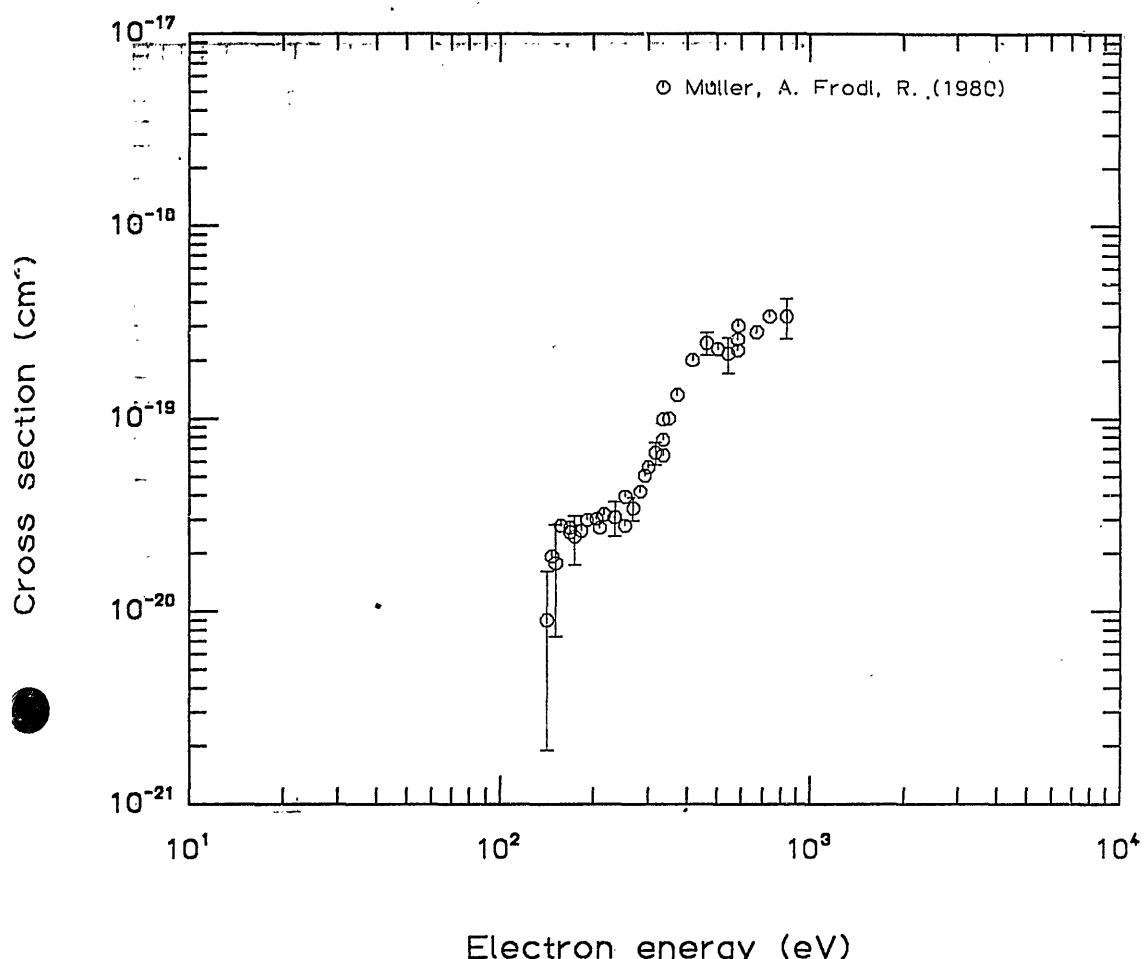
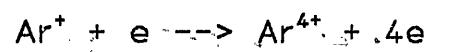


Fig. 114 Electron energy (eV)  $\text{cm}^2$   $\text{cm}^2$



Electron energy (eV)

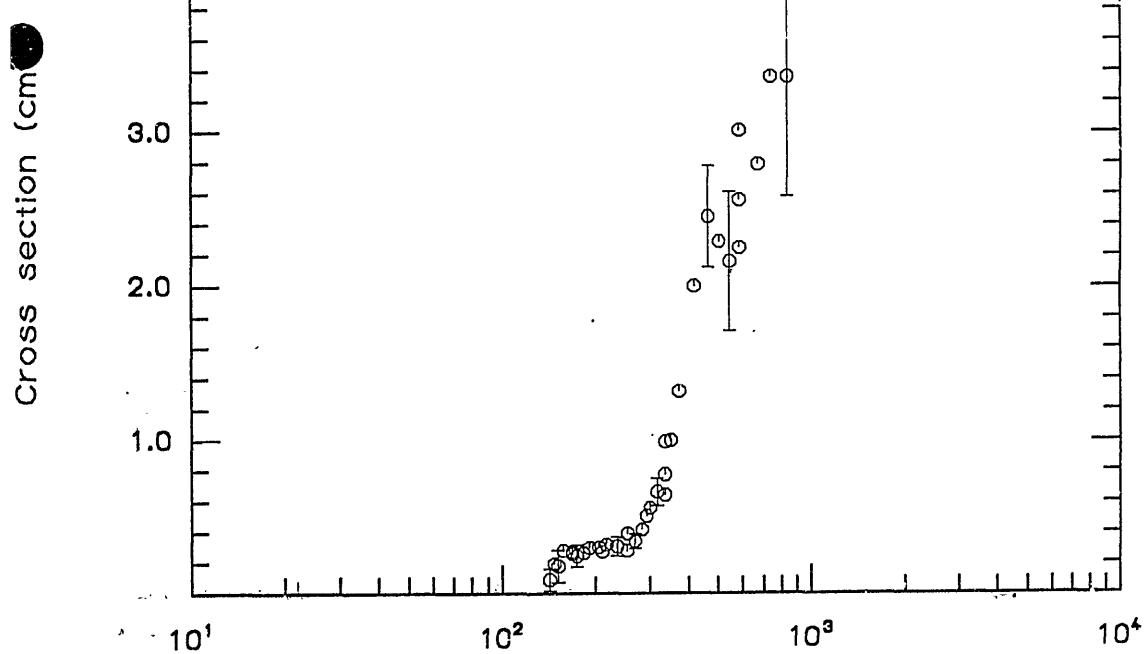


Fig. 115

Electron energy (eV)

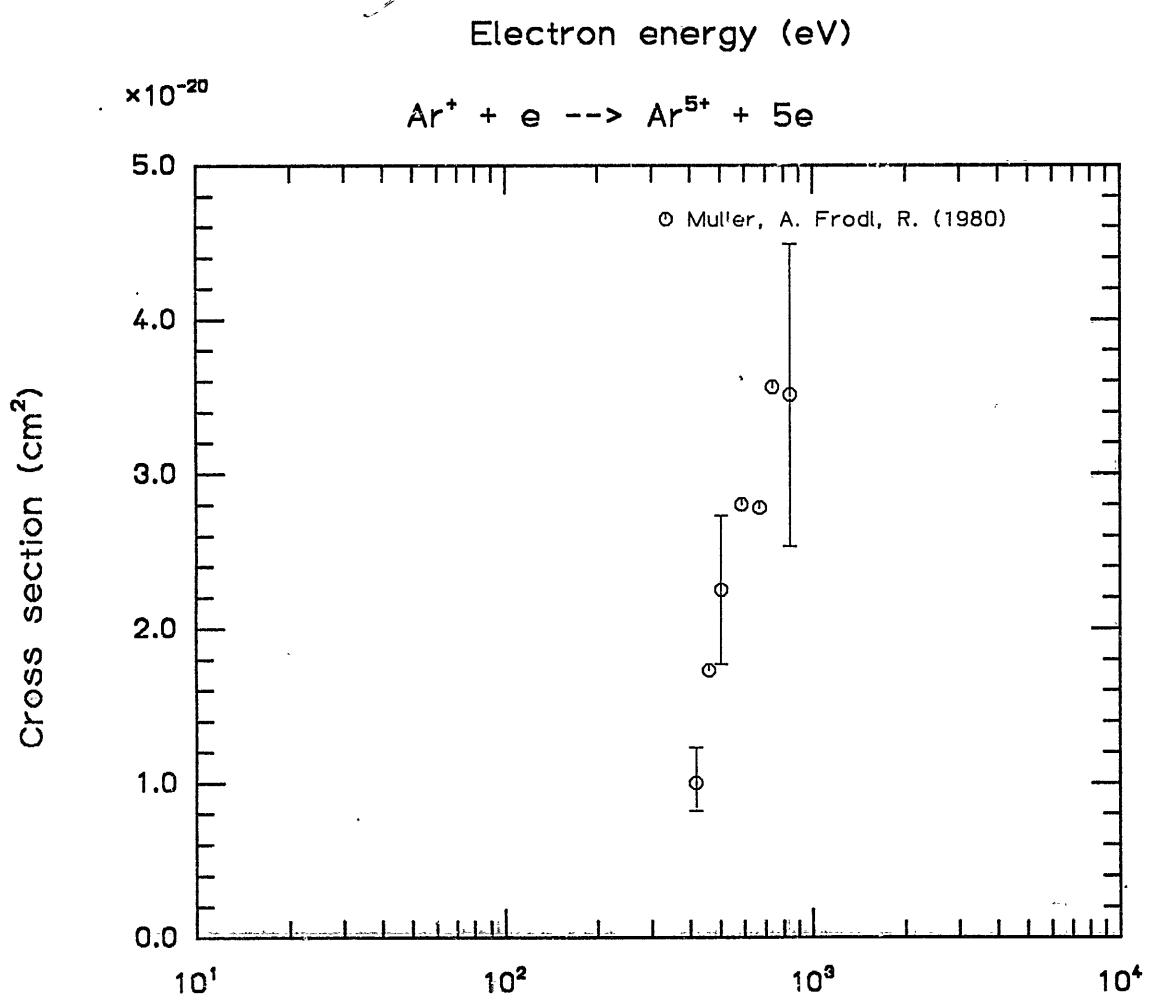
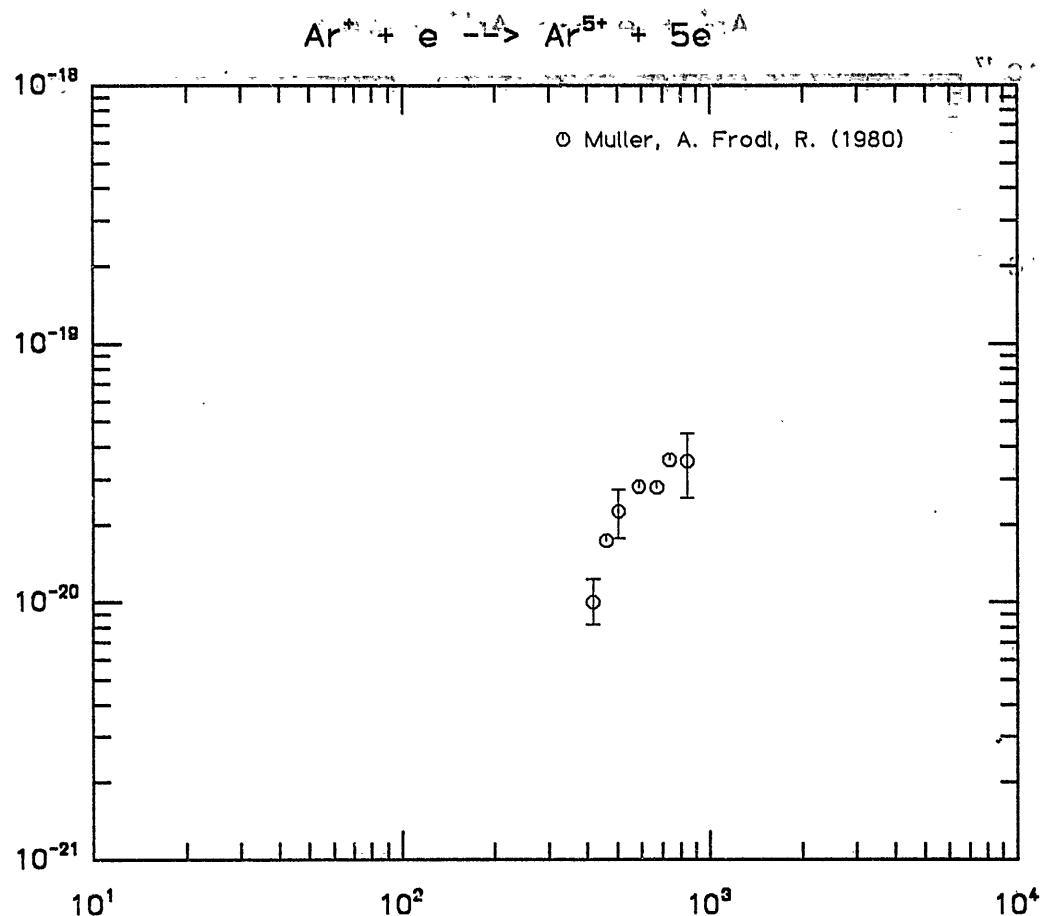
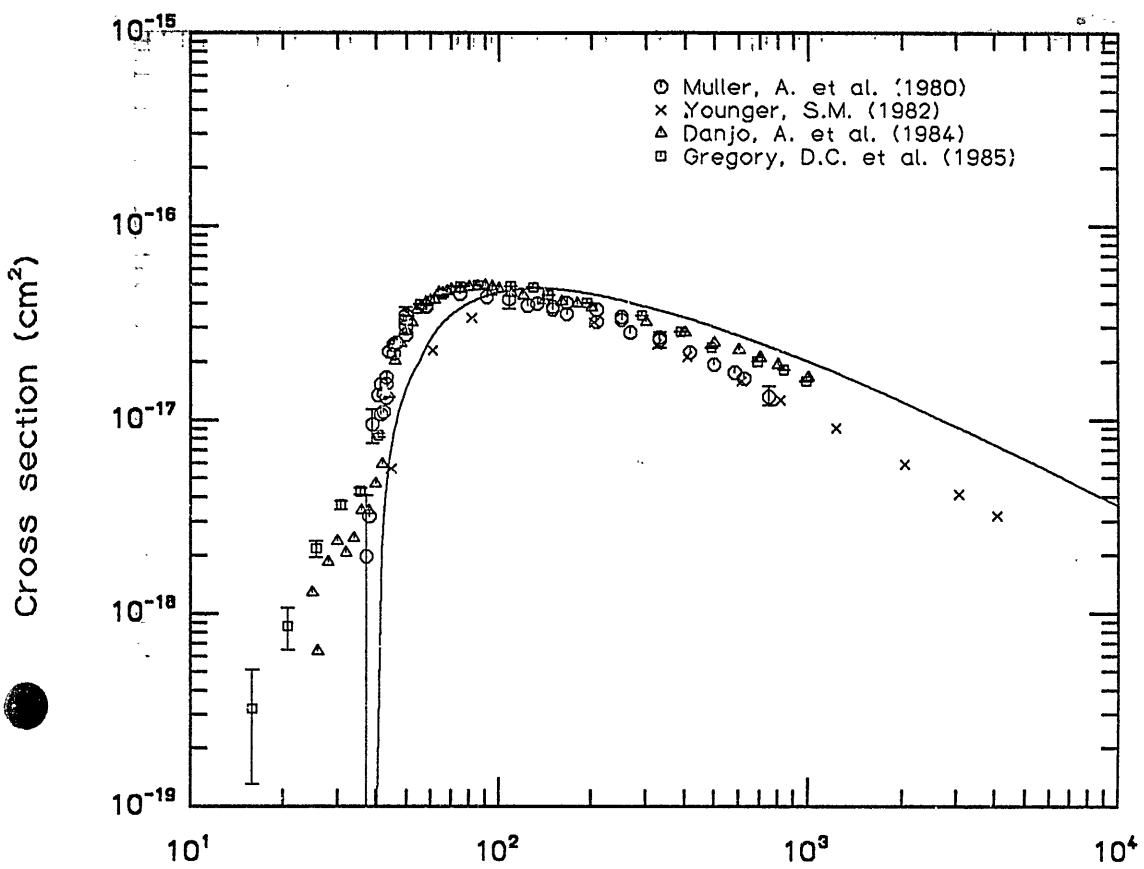
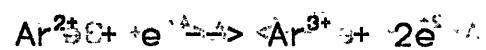


Fig. 116      Electron energy (eV)



Electron energy (eV)

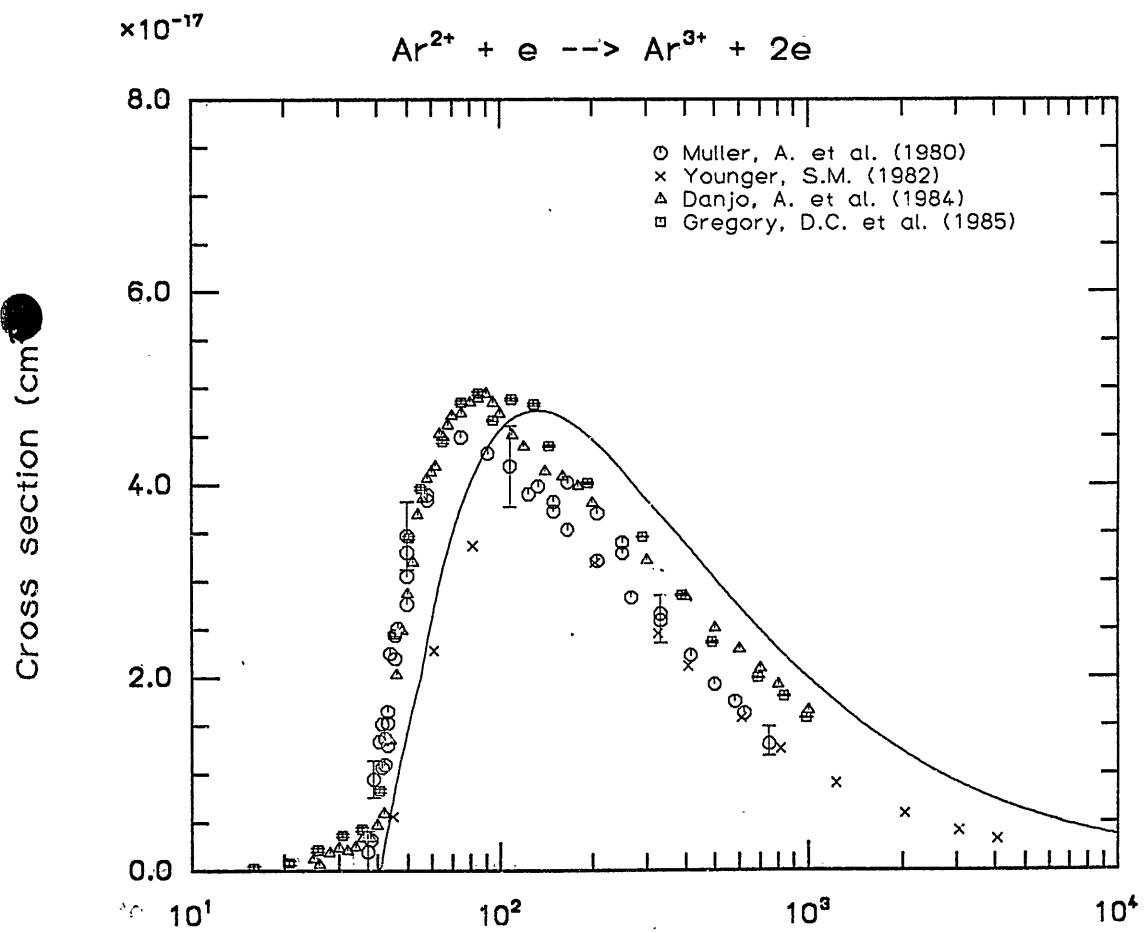


Fig. 117 : Electron energy (eV)

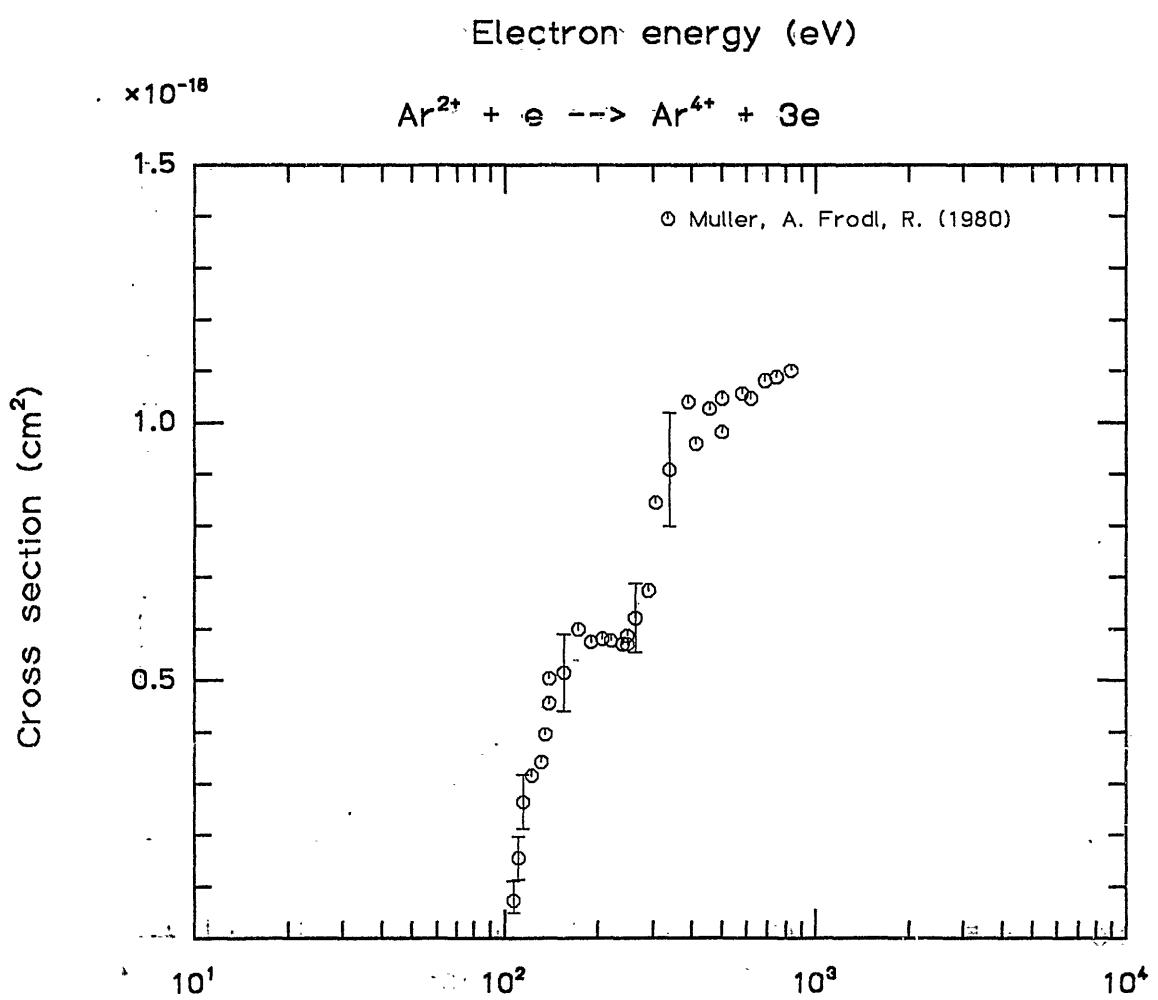
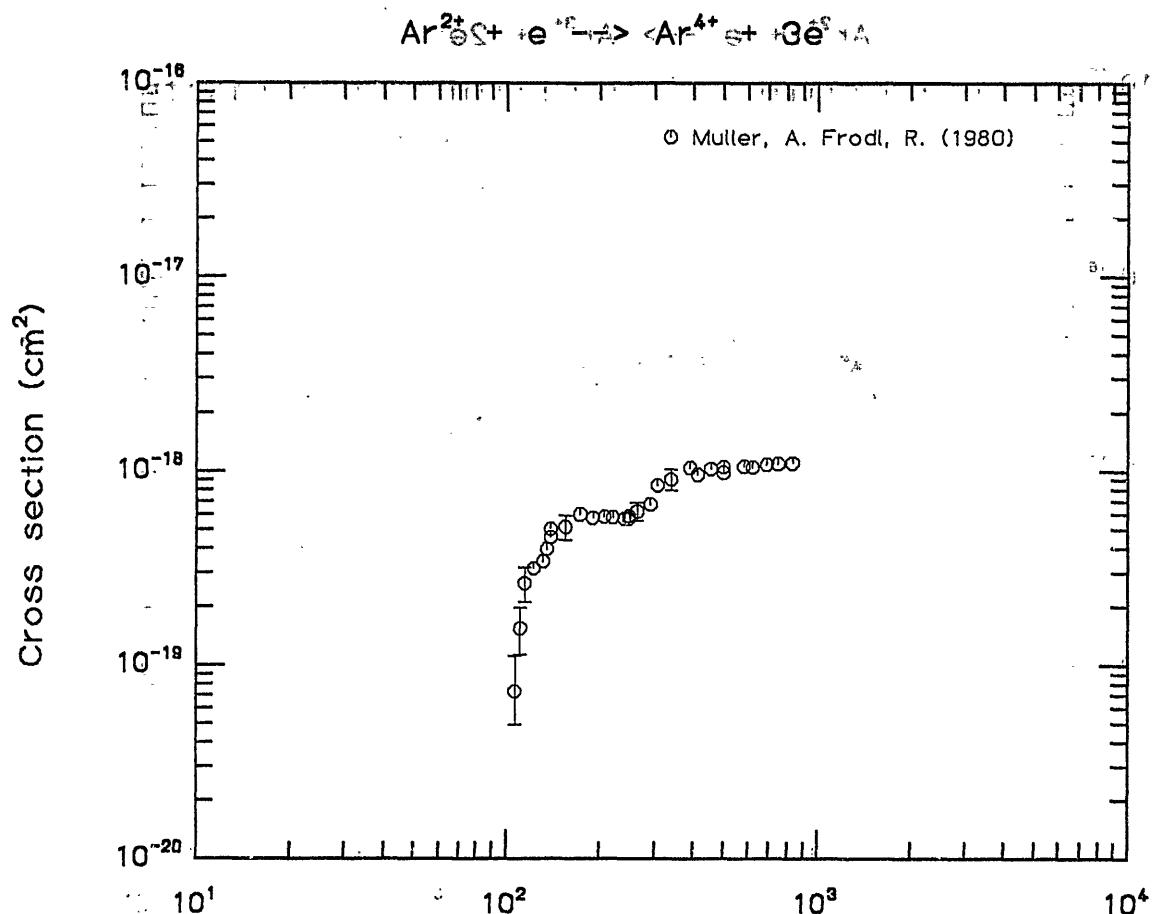


Fig. 118 Cross section vs Electron energy (eV)  $\text{Ar}^{2+} + \text{e}^- \rightarrow \text{Ar}^{4+} + 3\text{e}^-$

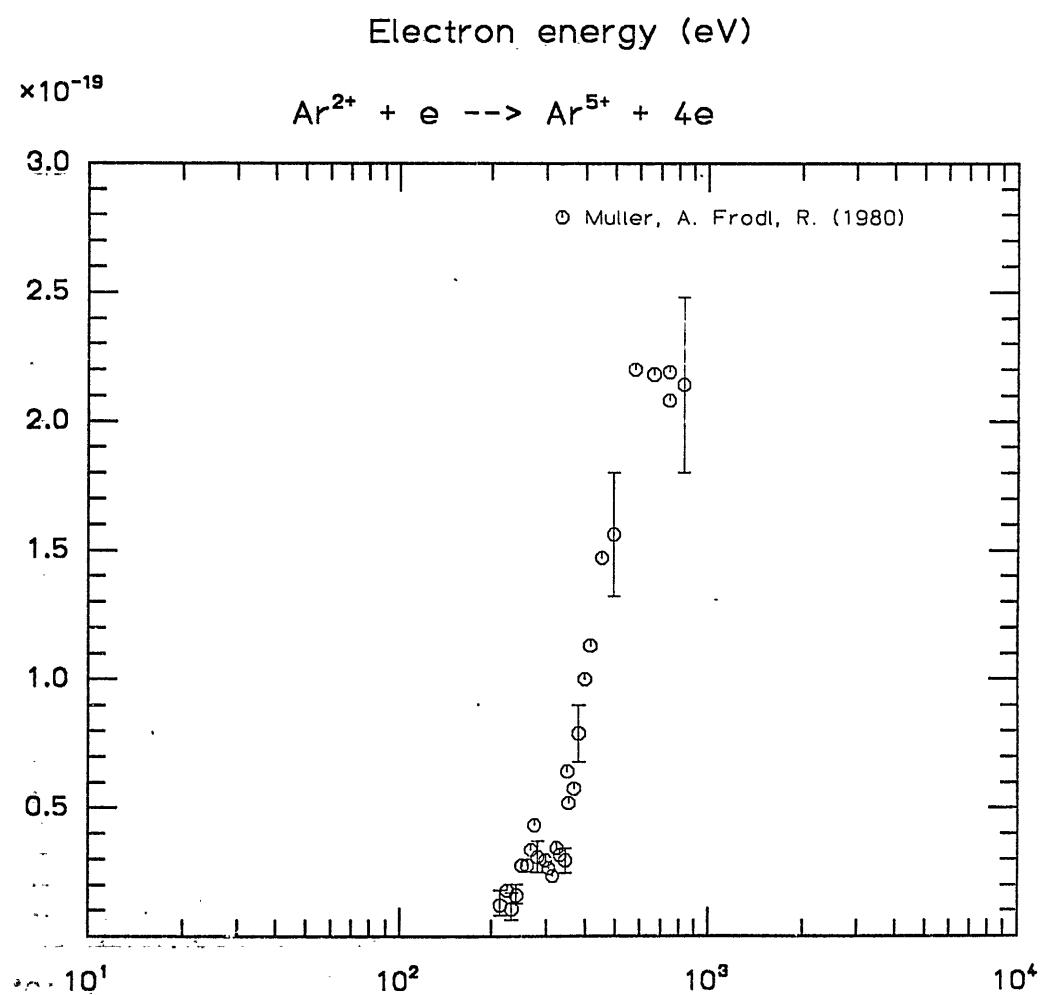
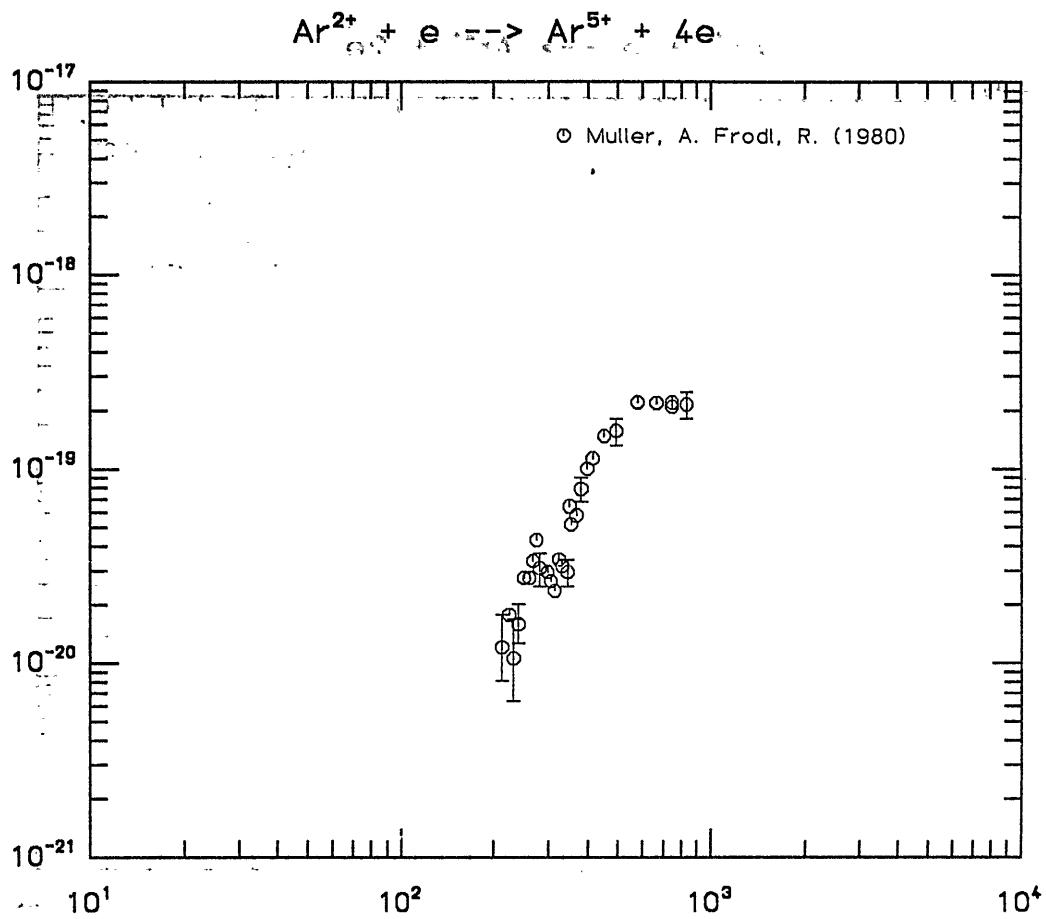


Fig. 119      Electron energy (eV)

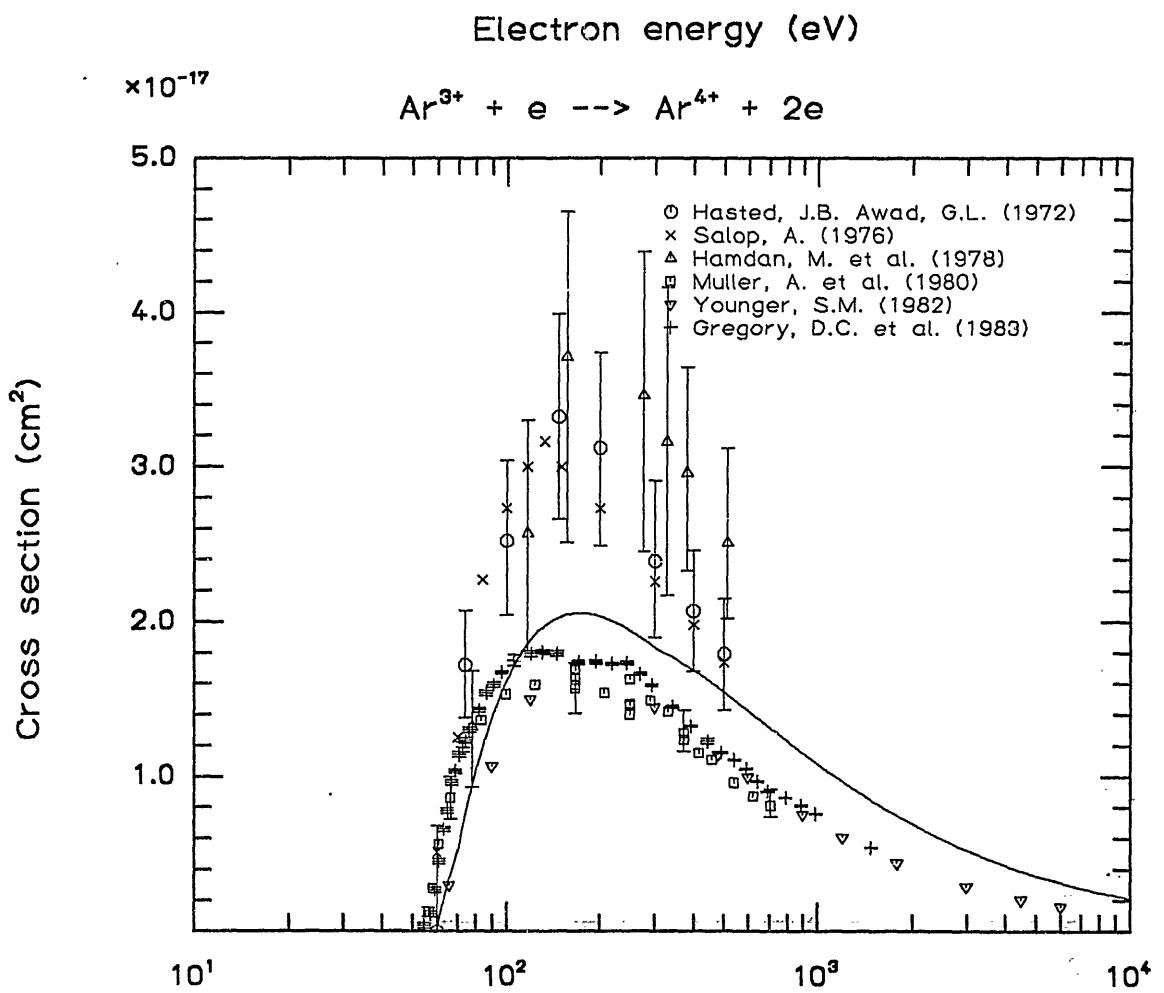
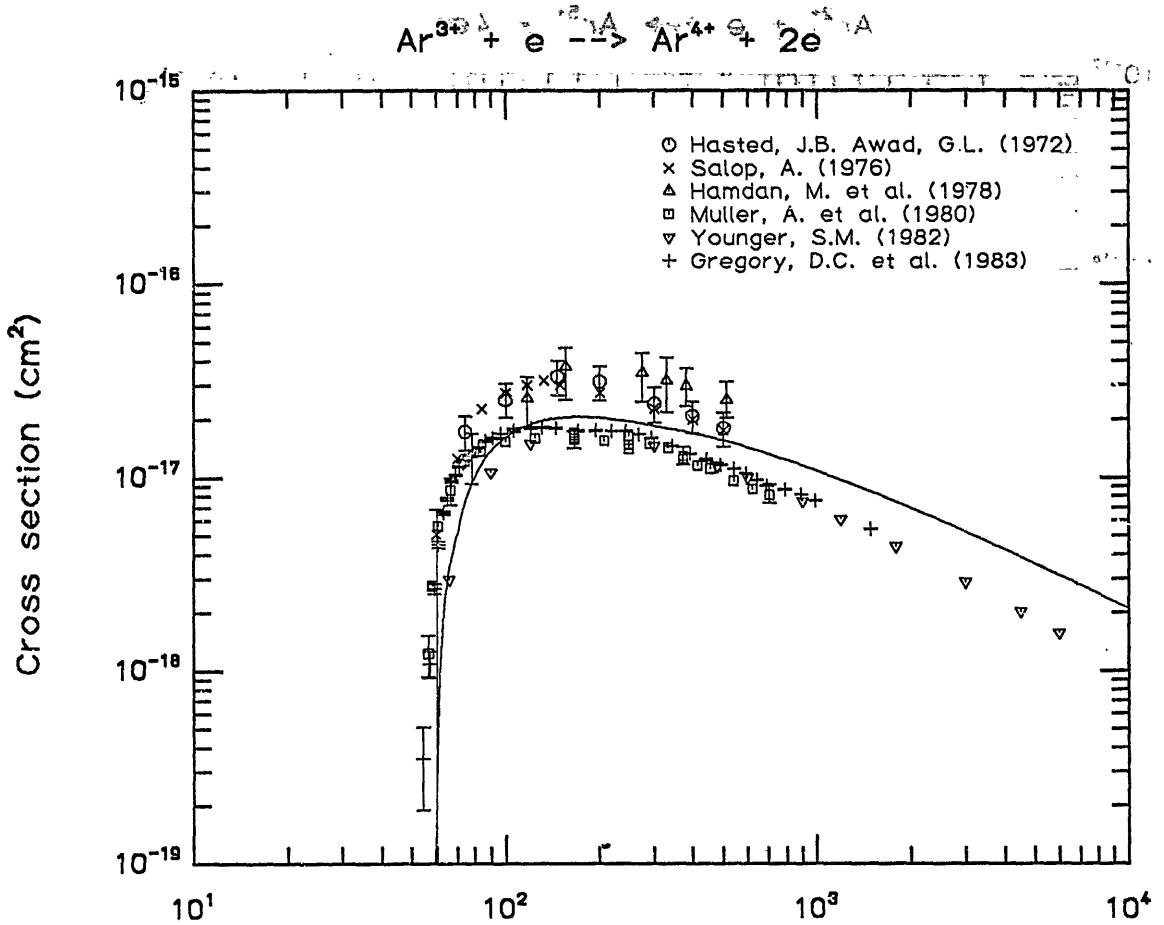


Fig. 120      (b) Electron energy (eV)      2.11      0.4

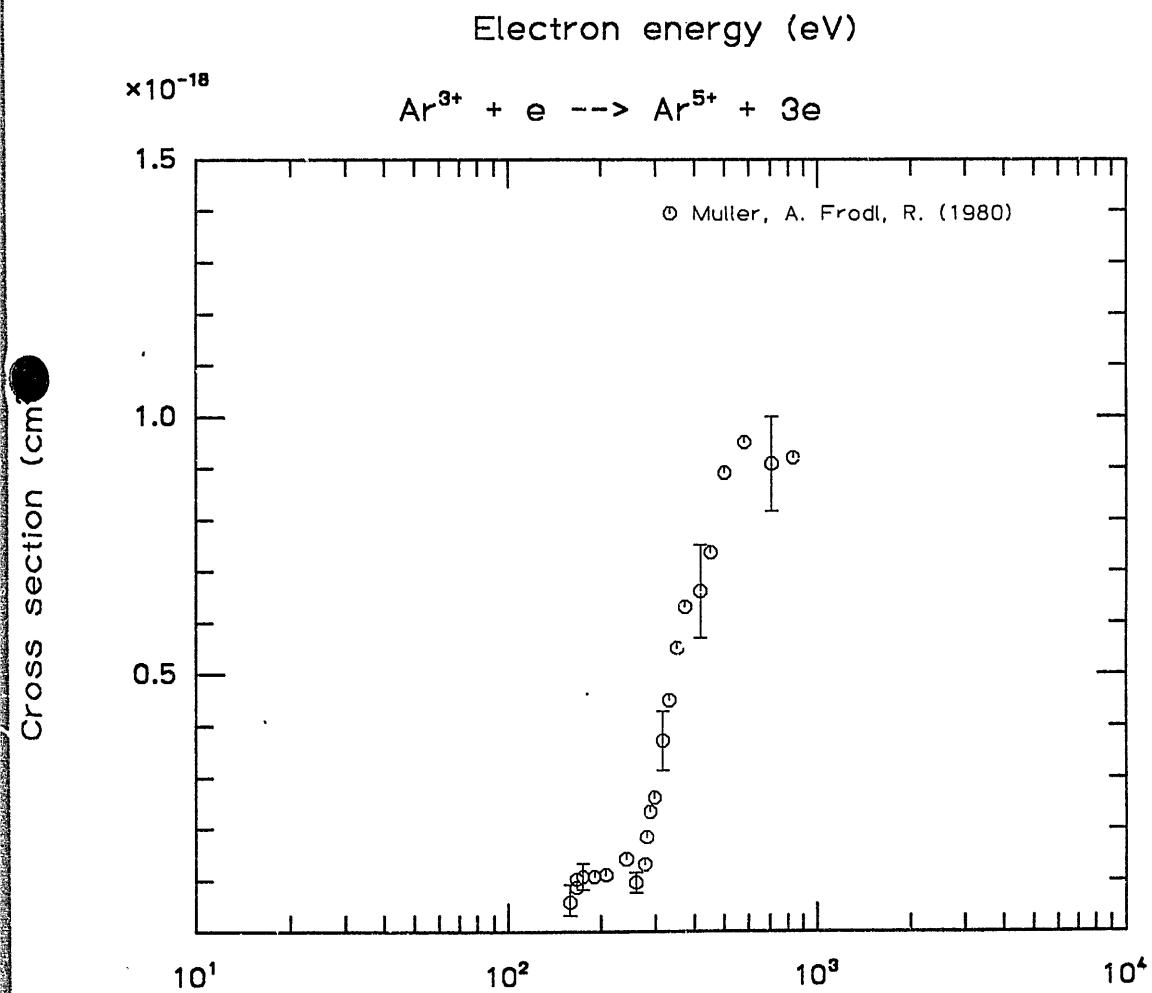
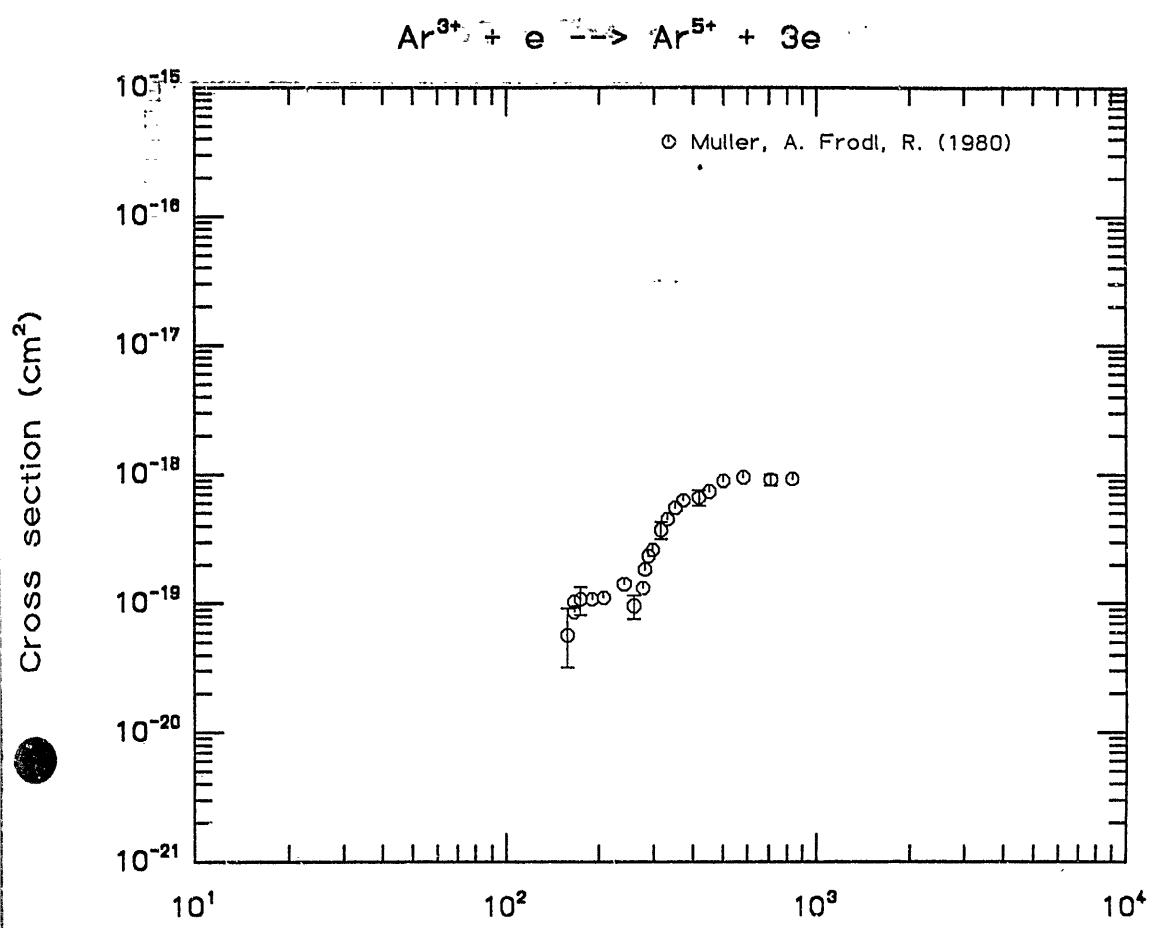


Fig. 121 . Electron energy (eV)

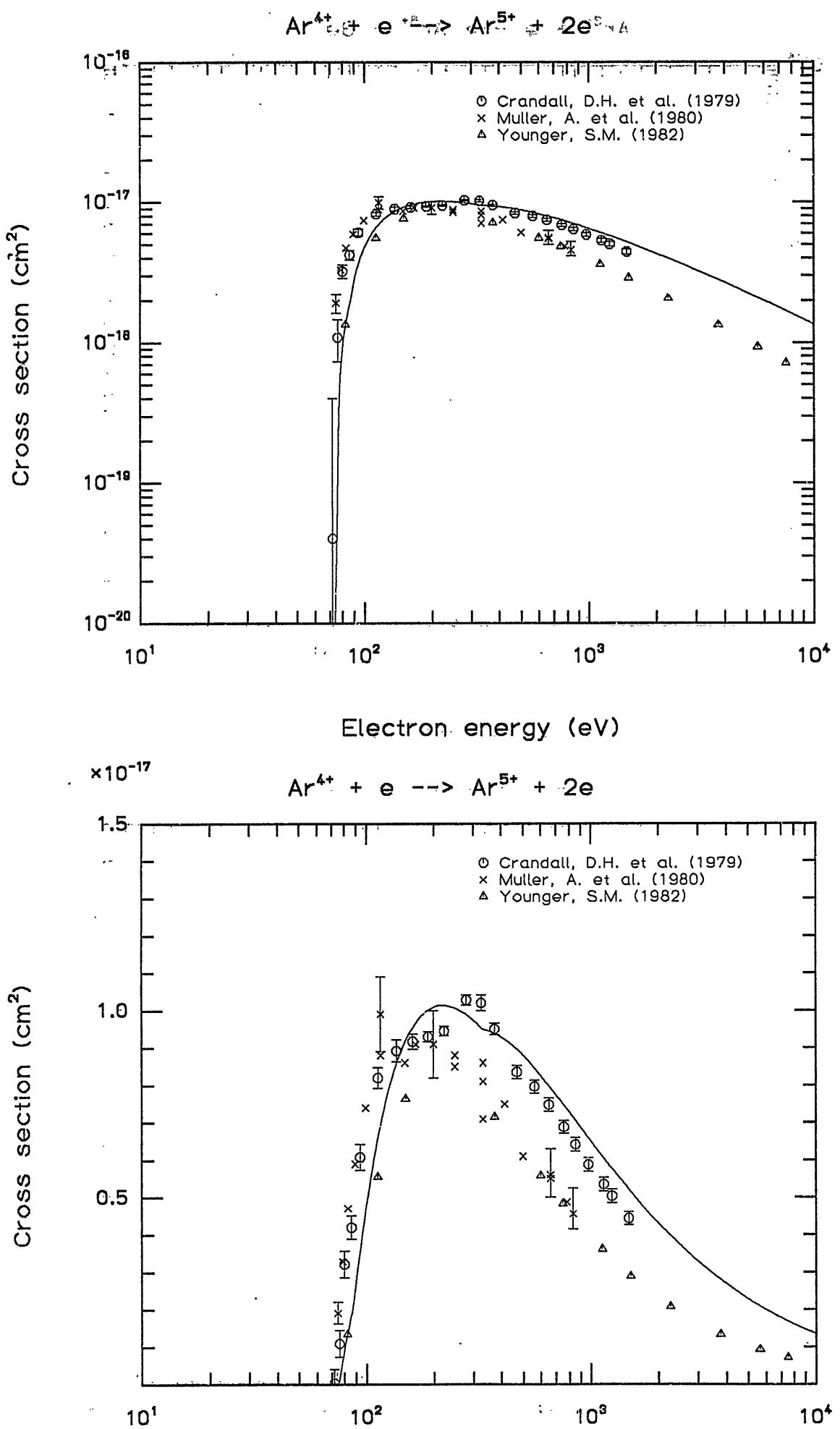
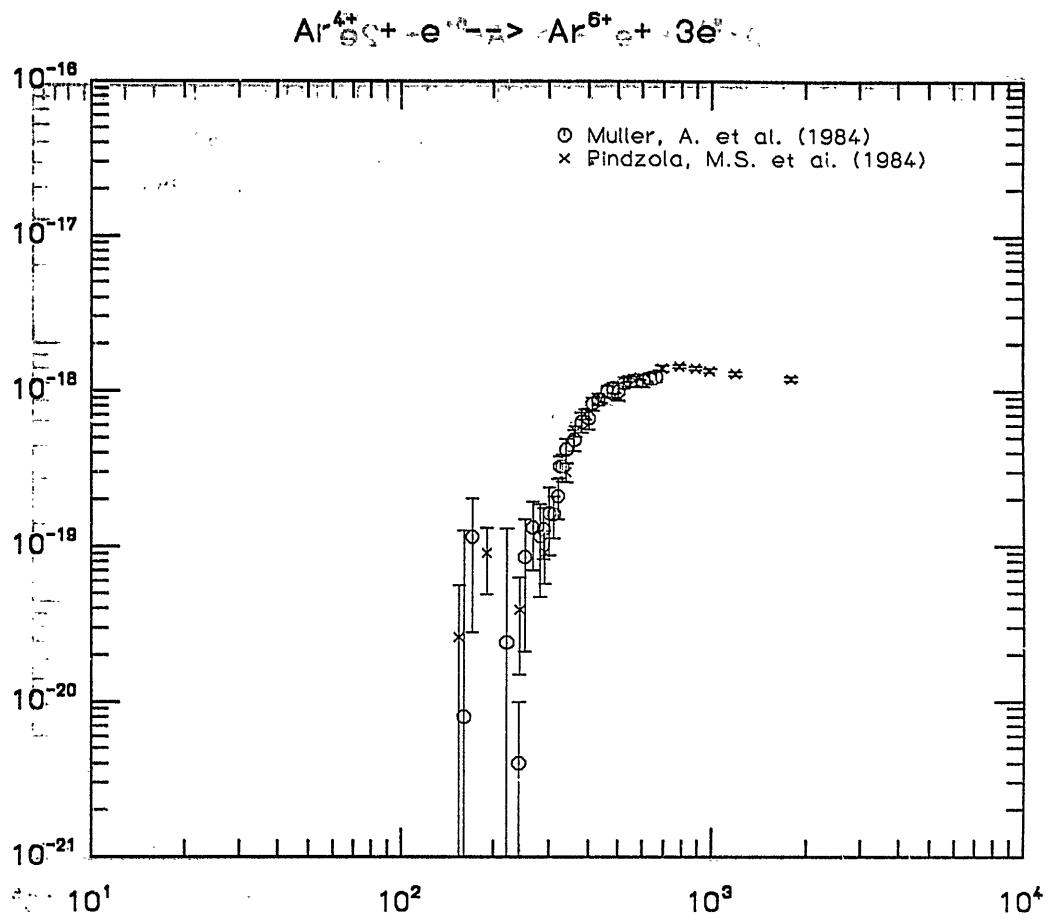


Fig. 122 Electron energy (eV)



Electron energy (eV)

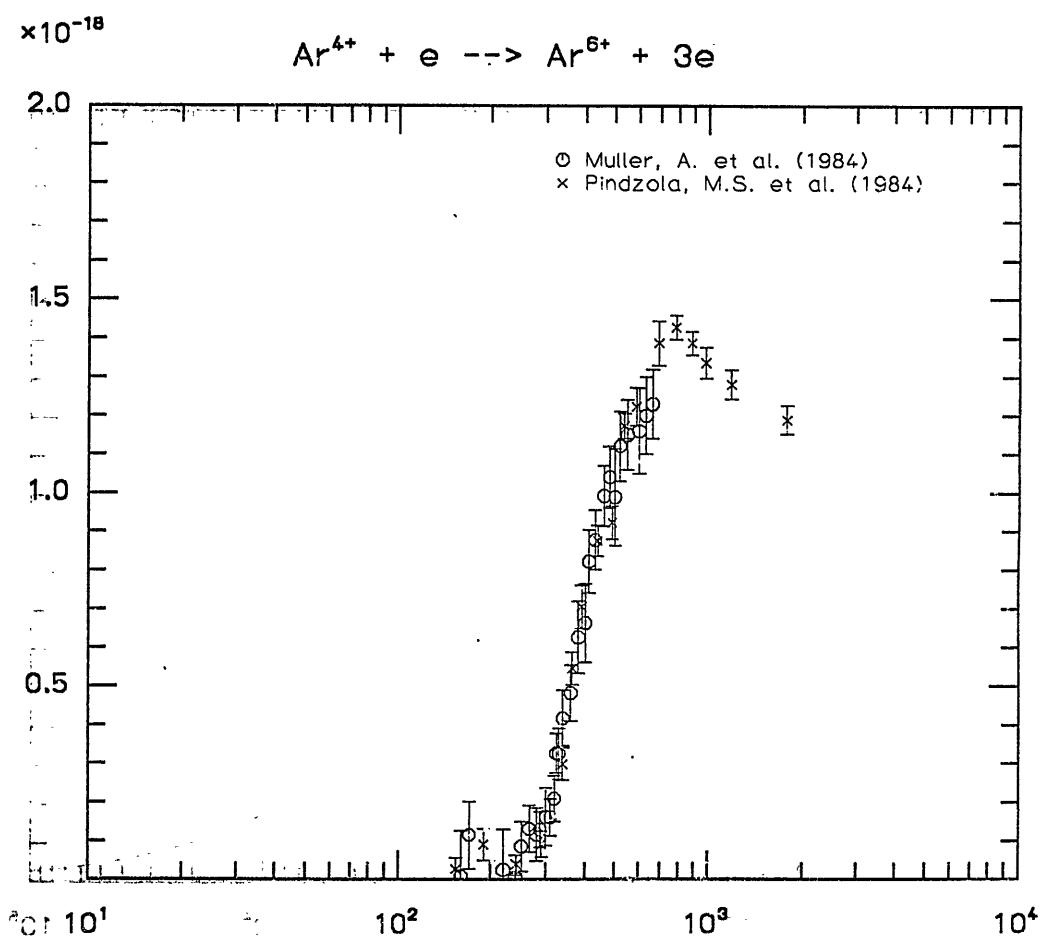


Fig. 123      Electron energy (eV)

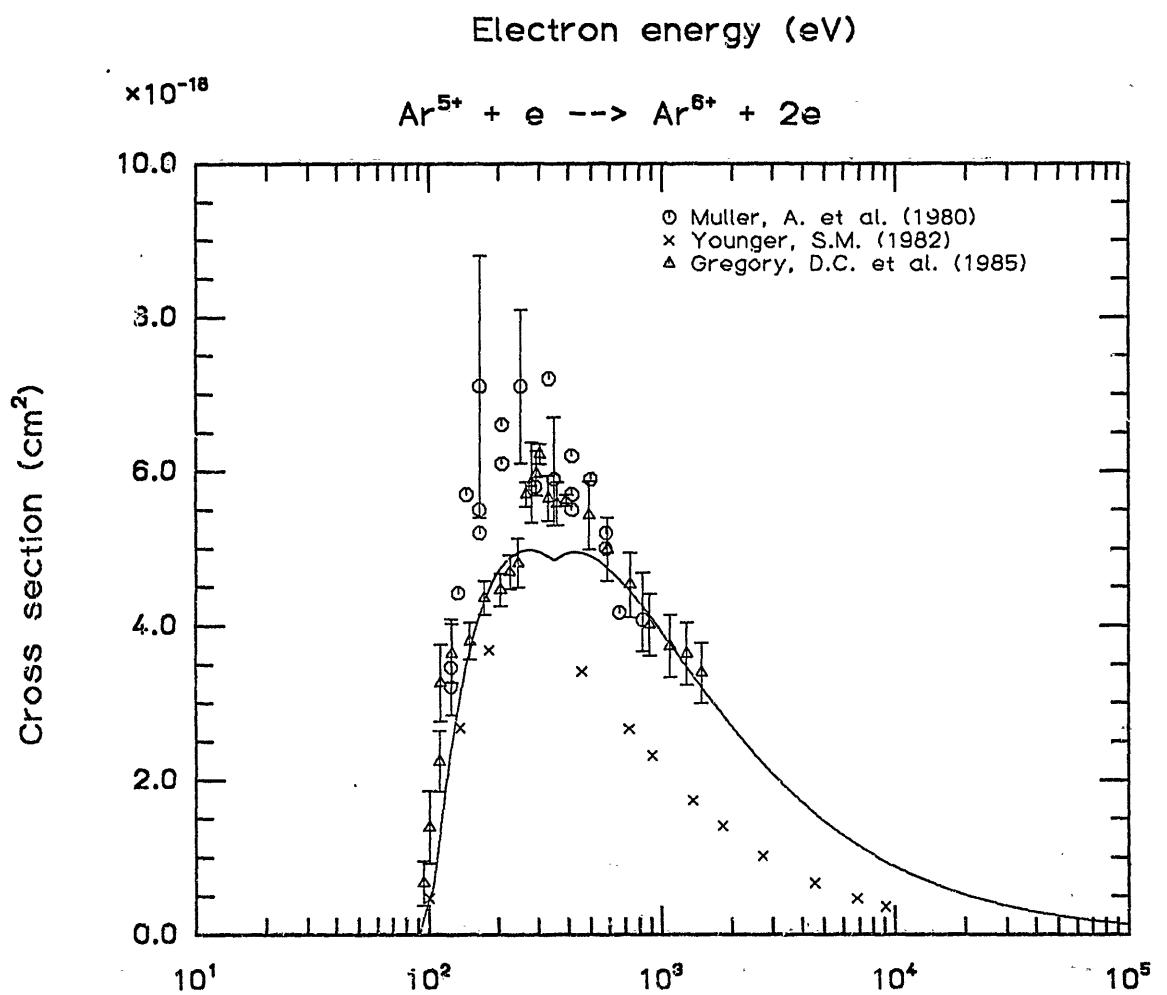
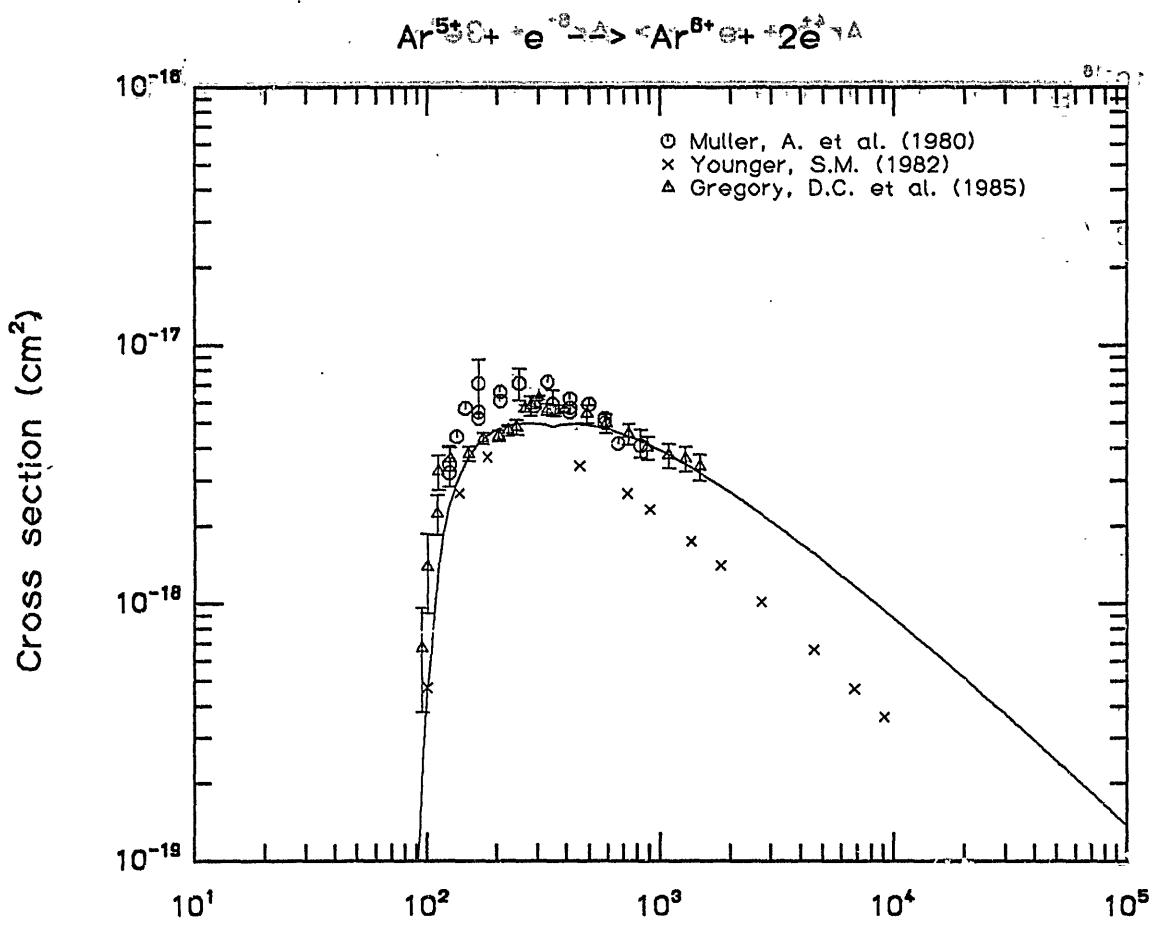
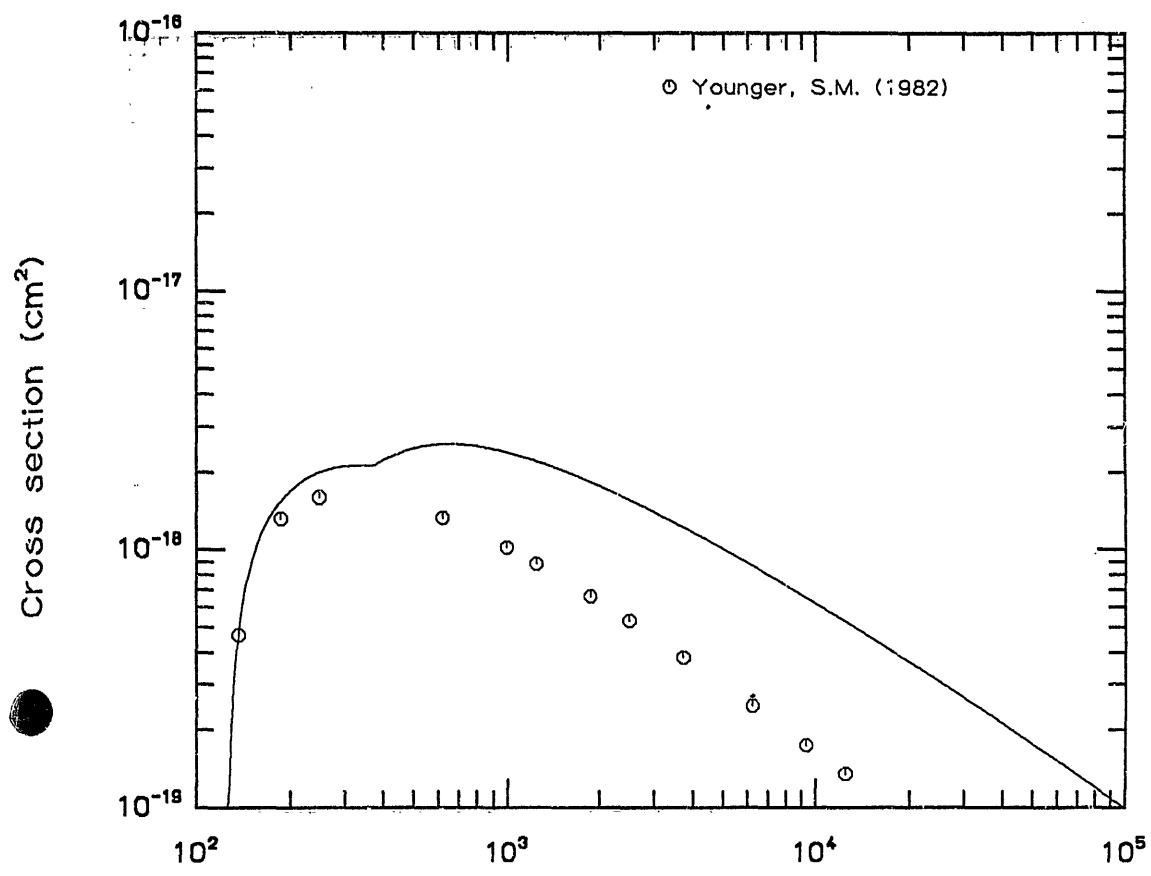
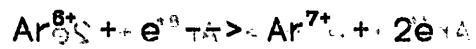


Fig. 124      Electron energy (eV)    Fig. 124



Electron energy (eV)

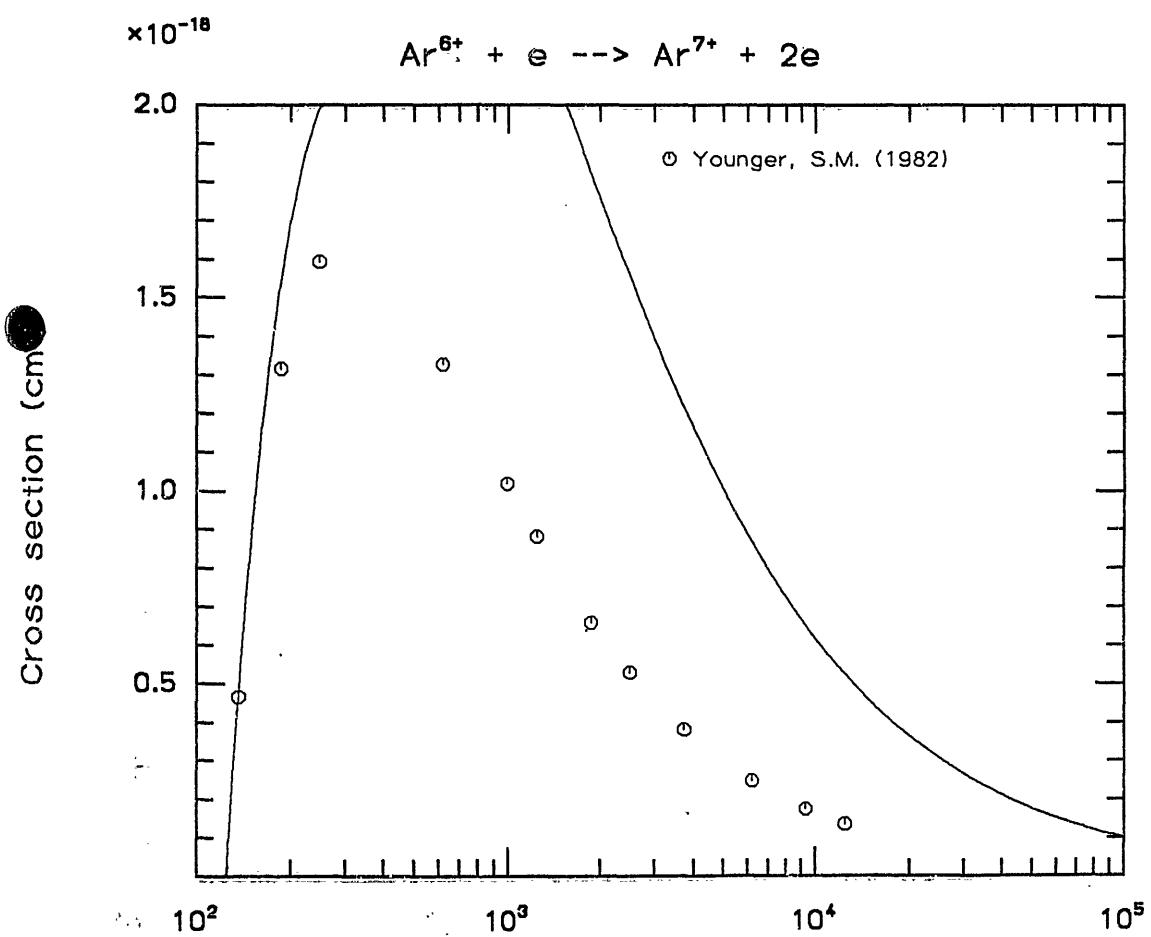


Fig. 125      Electron energy (eV)

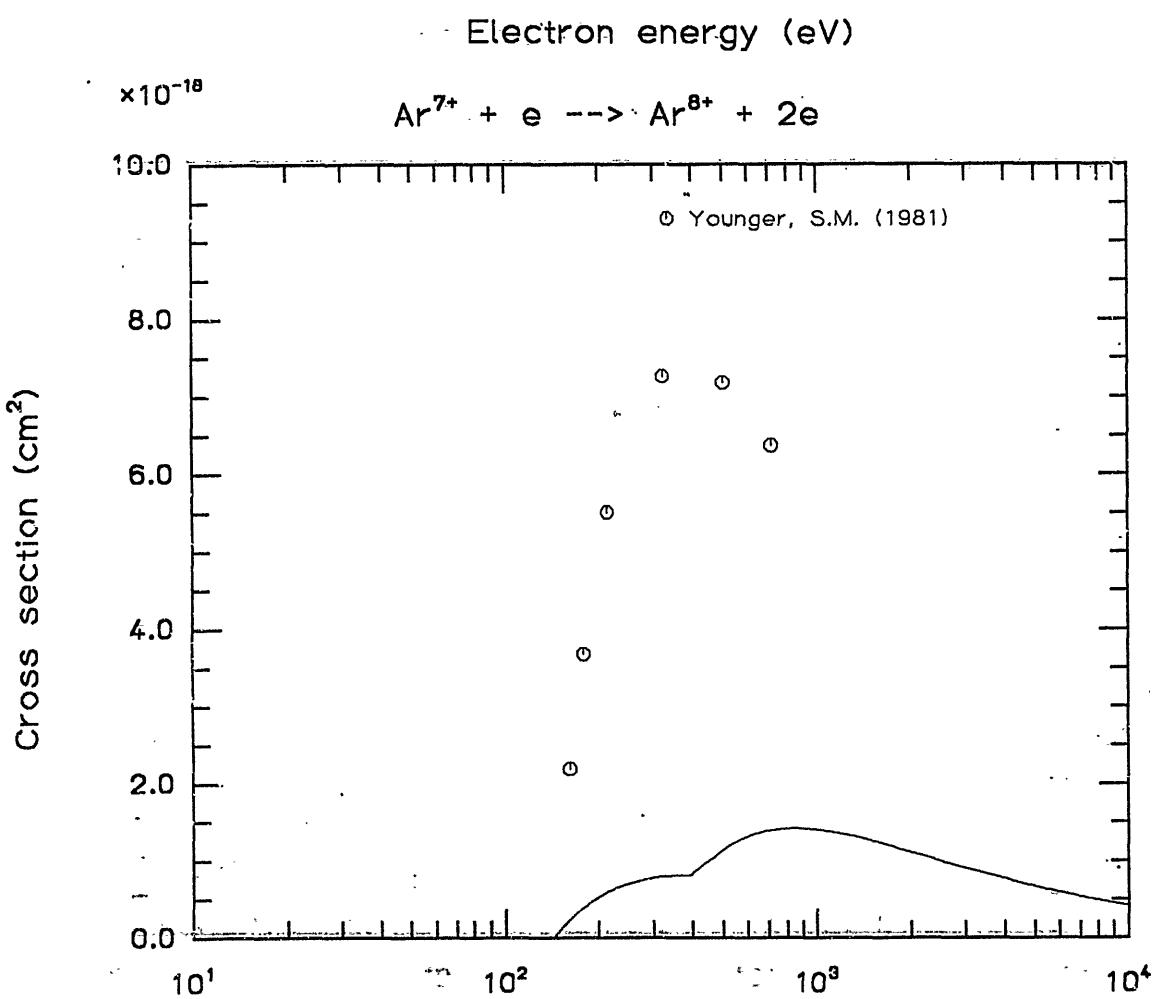
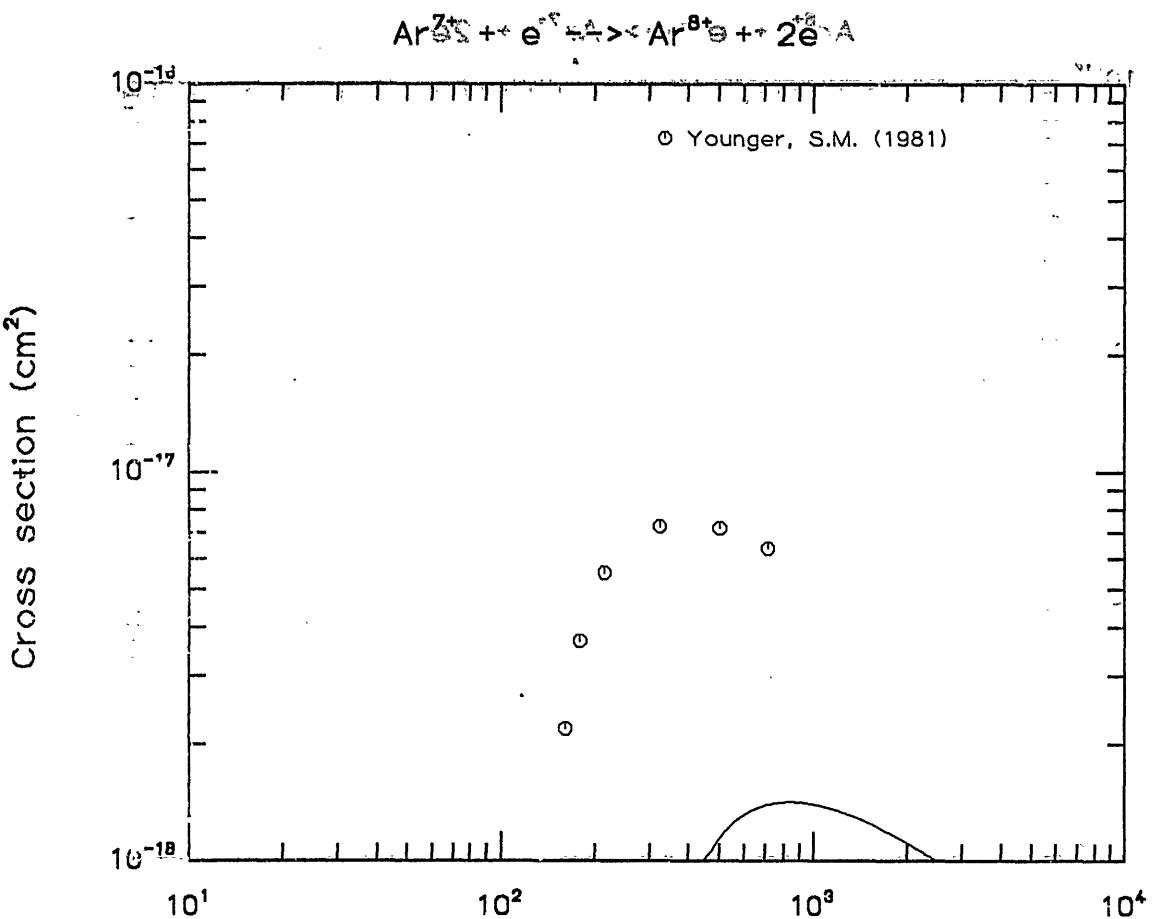
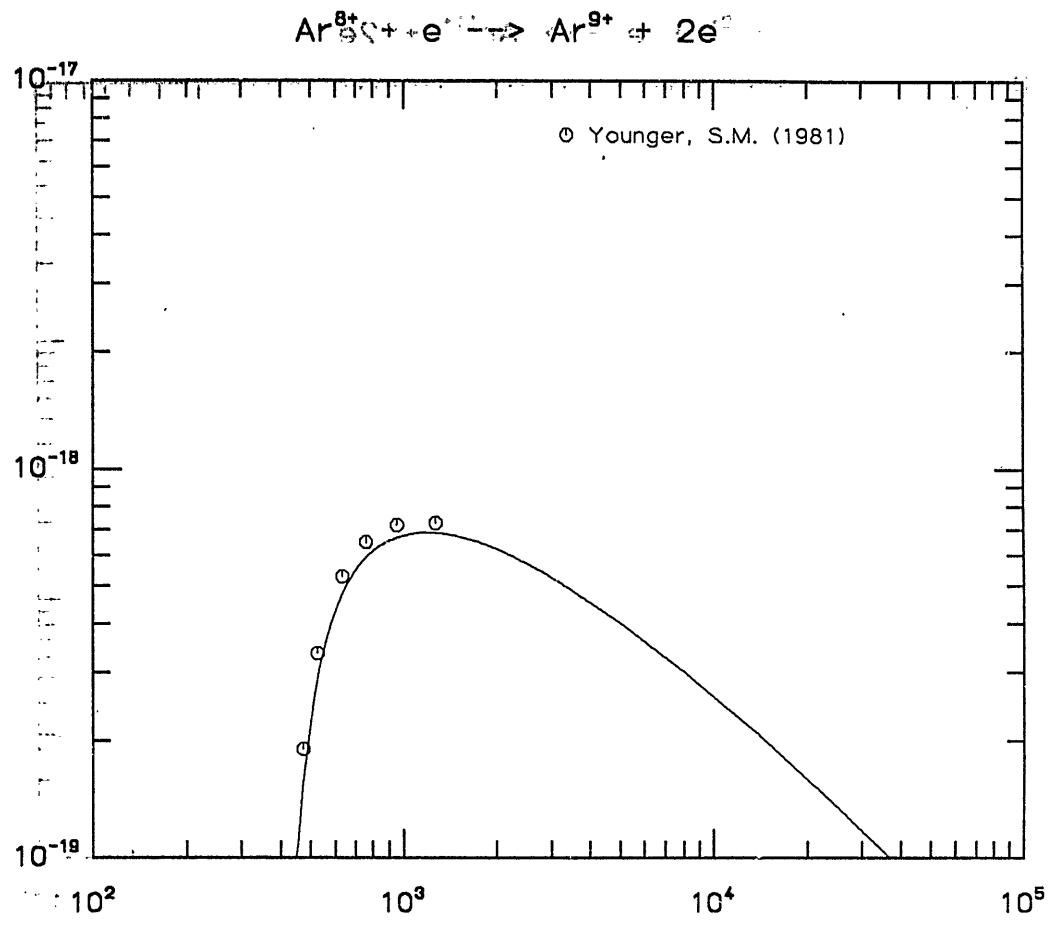


Fig. 126

Cross section ( $\text{cm}^2$ )



Electron energy (eV)

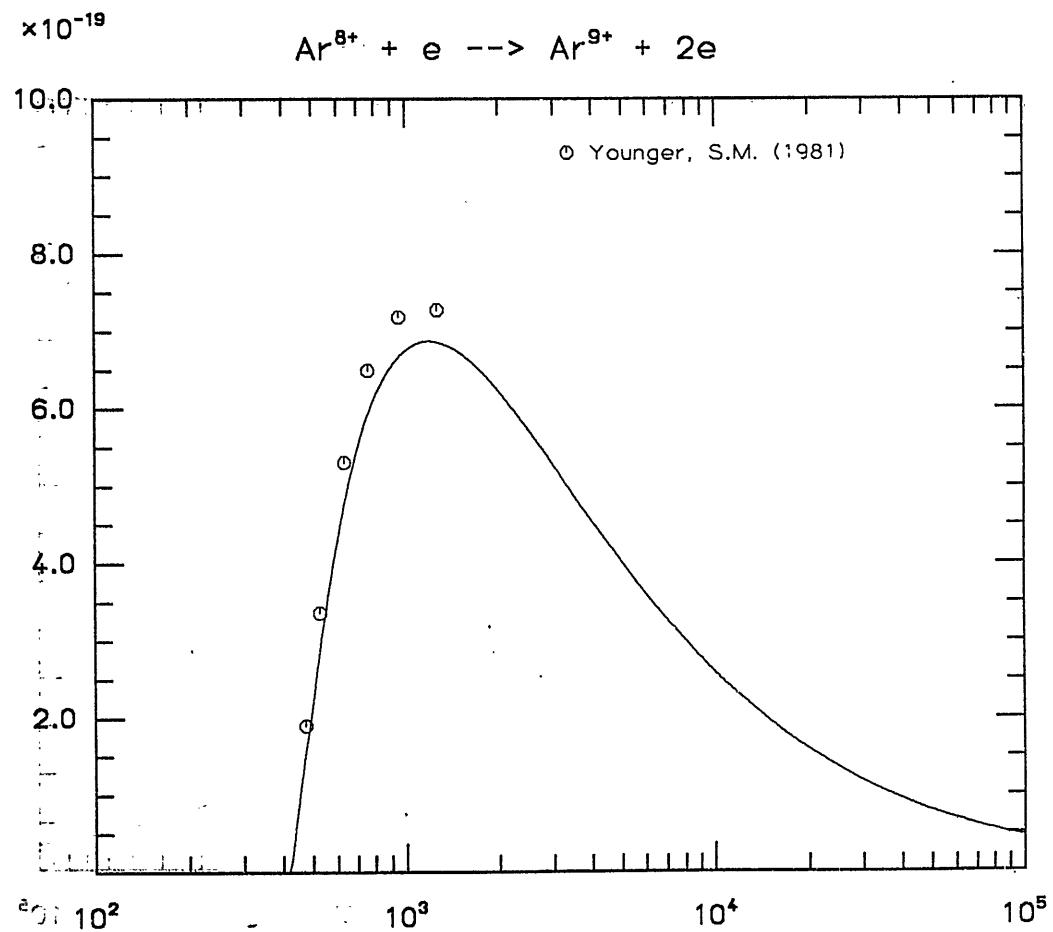


Fig. 127 Electron energy (eV)

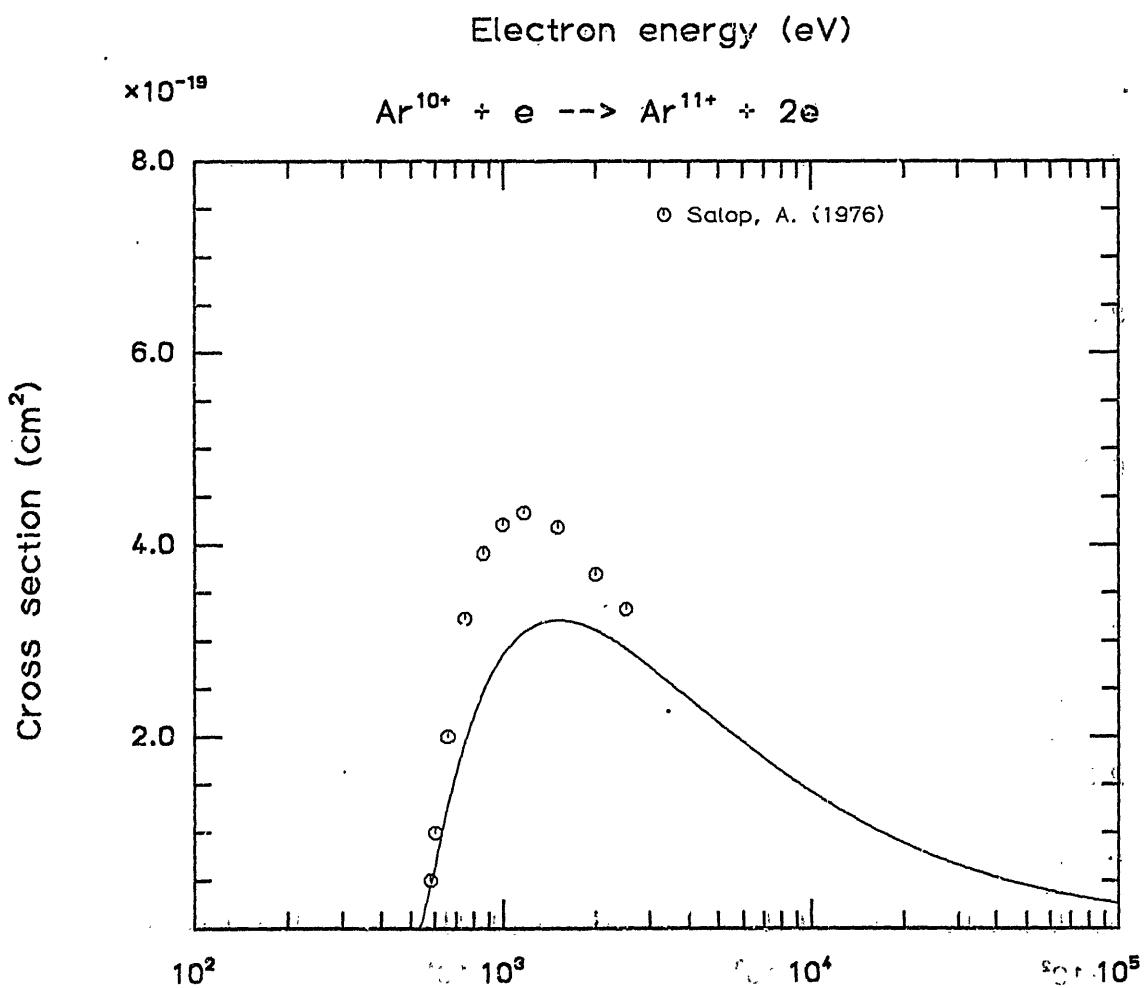
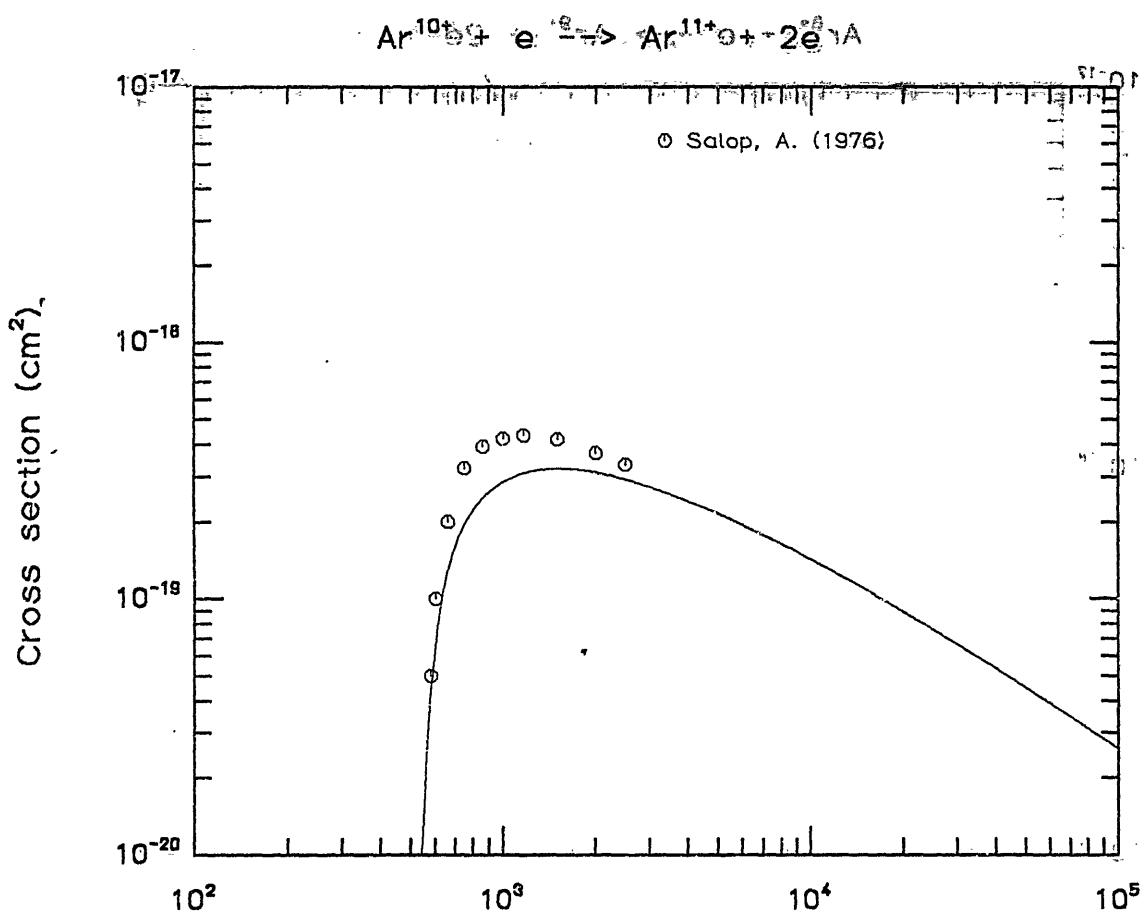
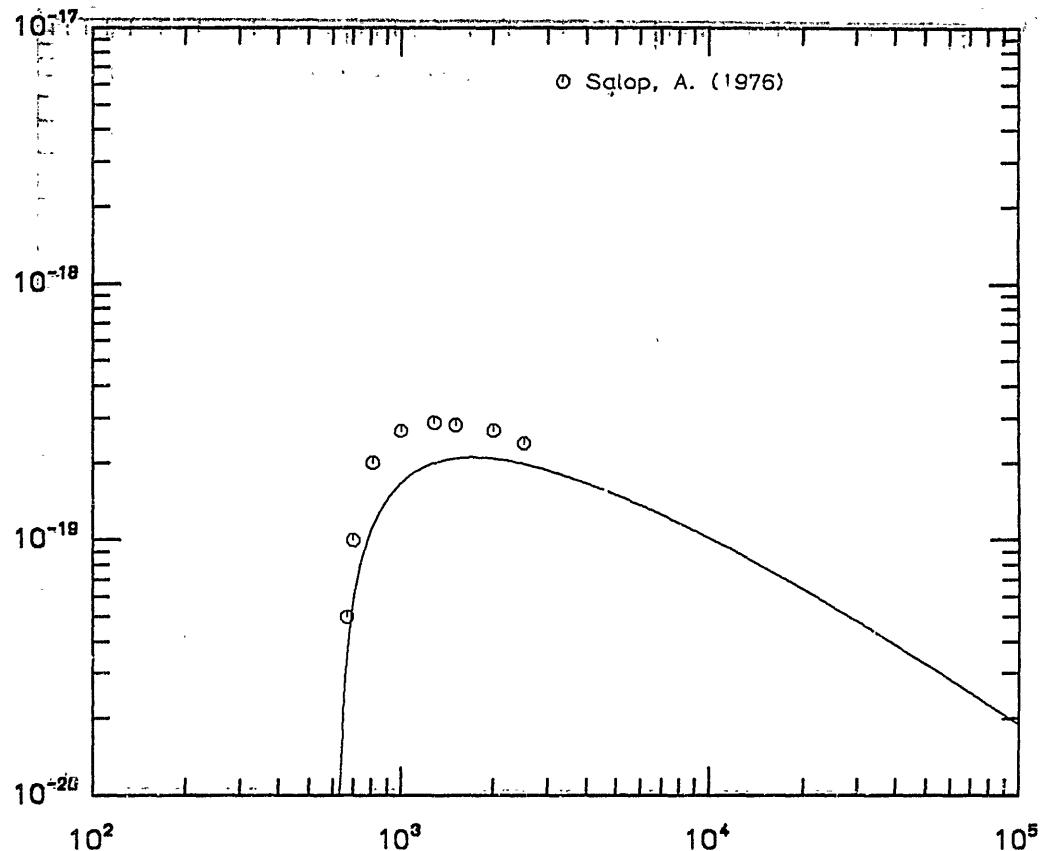


Fig. 128 (Ne) Electron energy (eV) Fig. 129



Electron energy (eV)

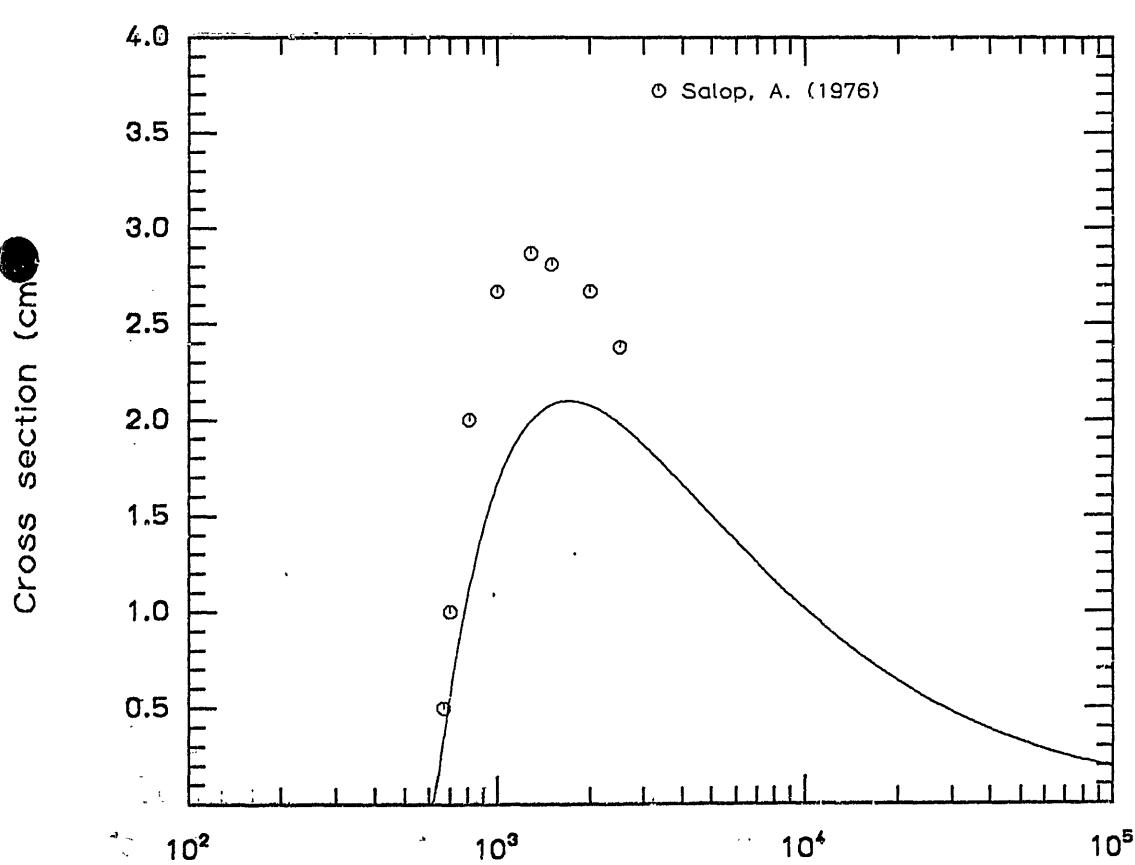
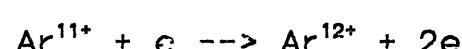


Fig. 129      Electron energy (eV)

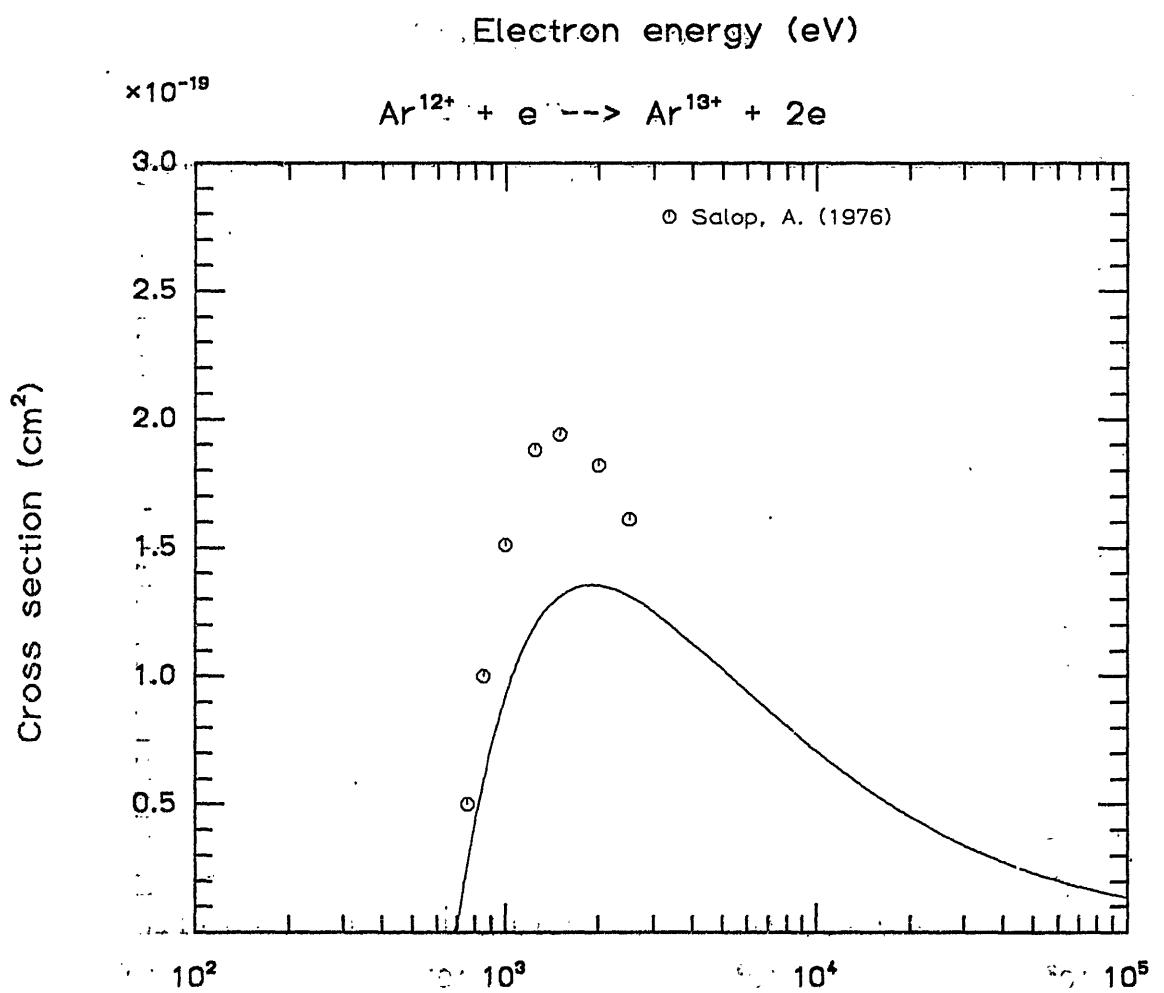
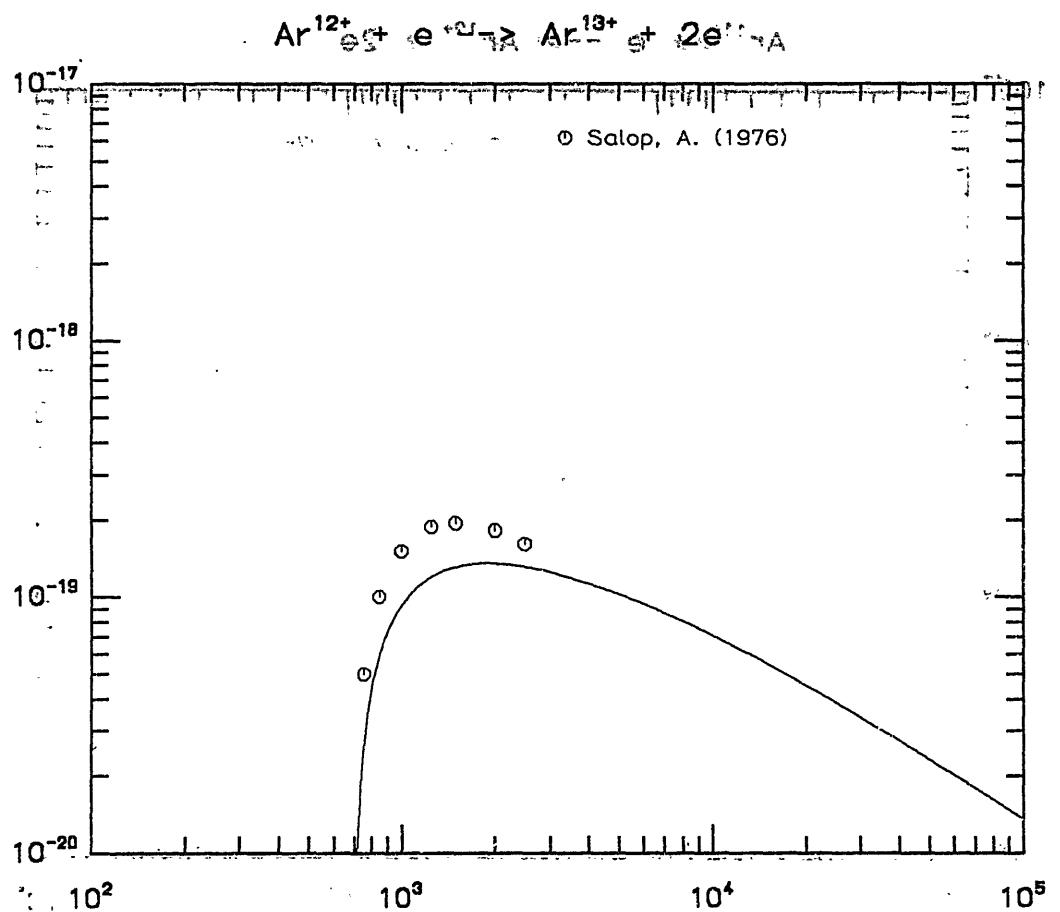


Fig. 130 (vs Electron energy(eV)) Fig. 129

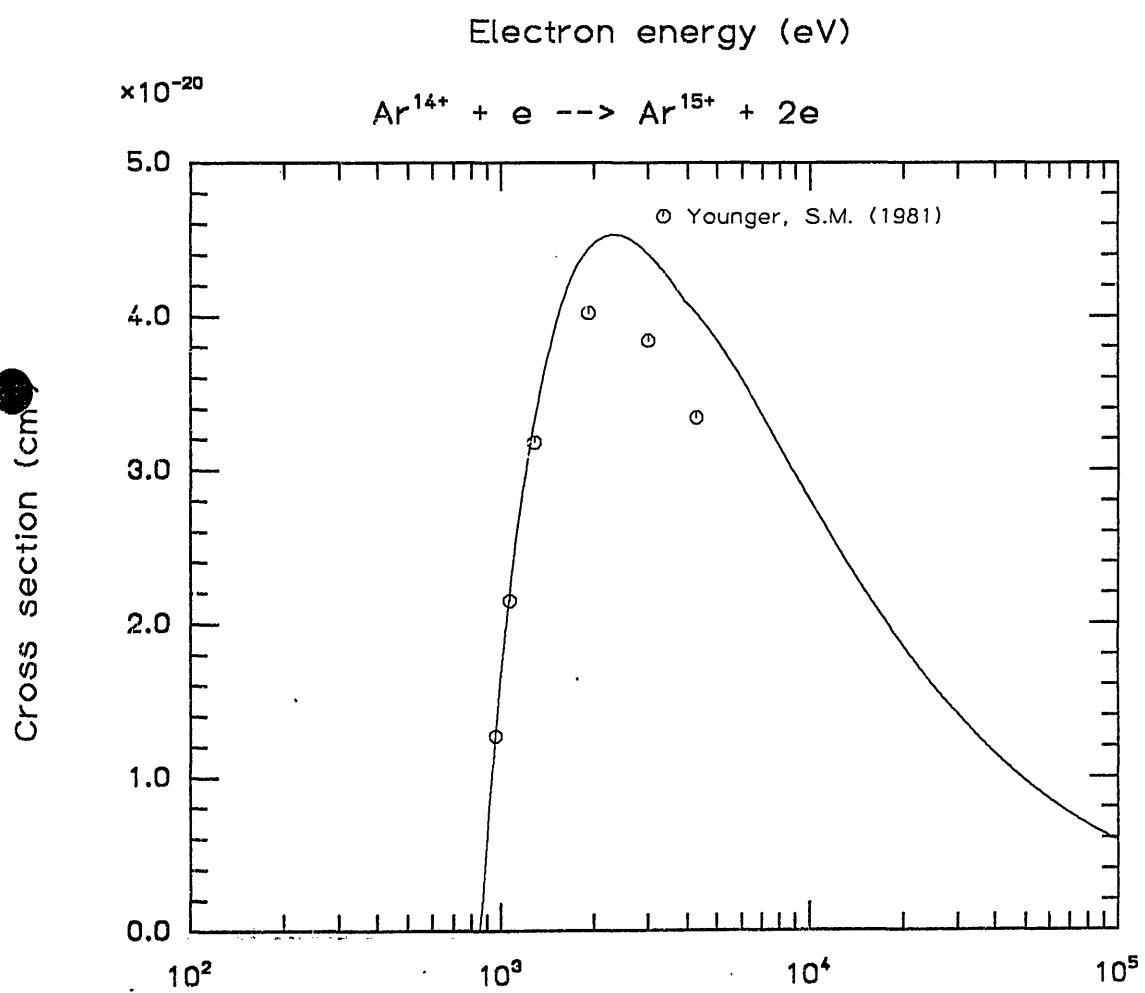
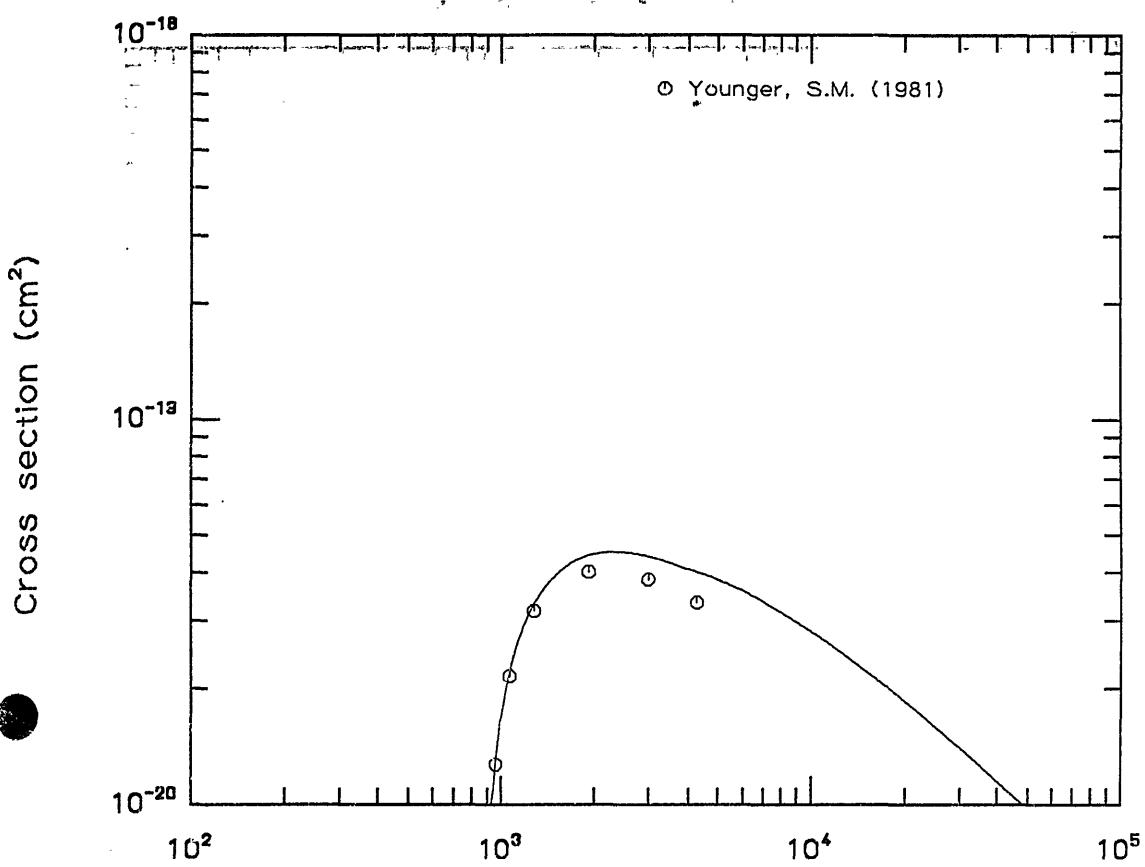
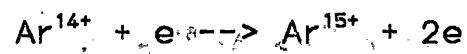


Fig. 131 Electron energy (eV)

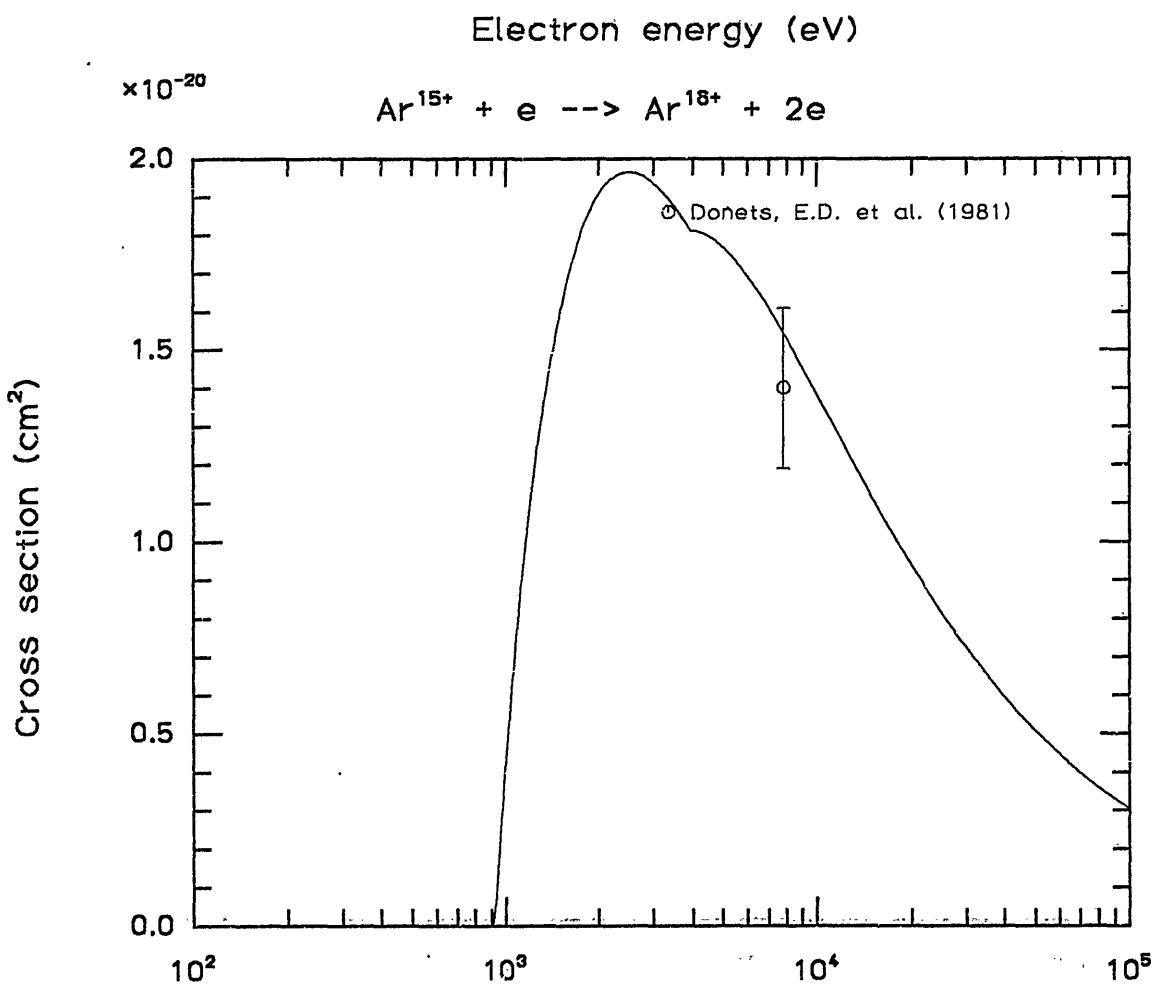
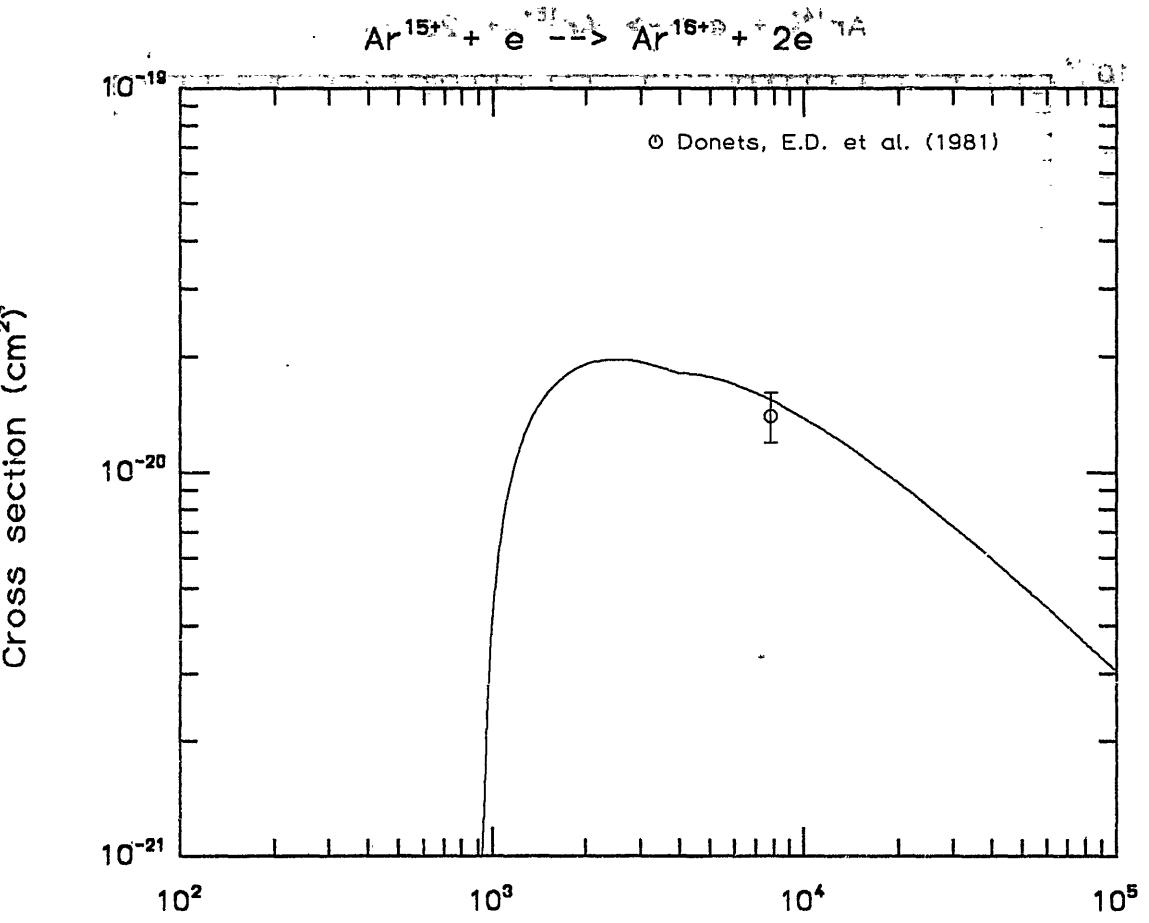


Fig. 132 / 3 Electron energy (eV) · i E · i pF

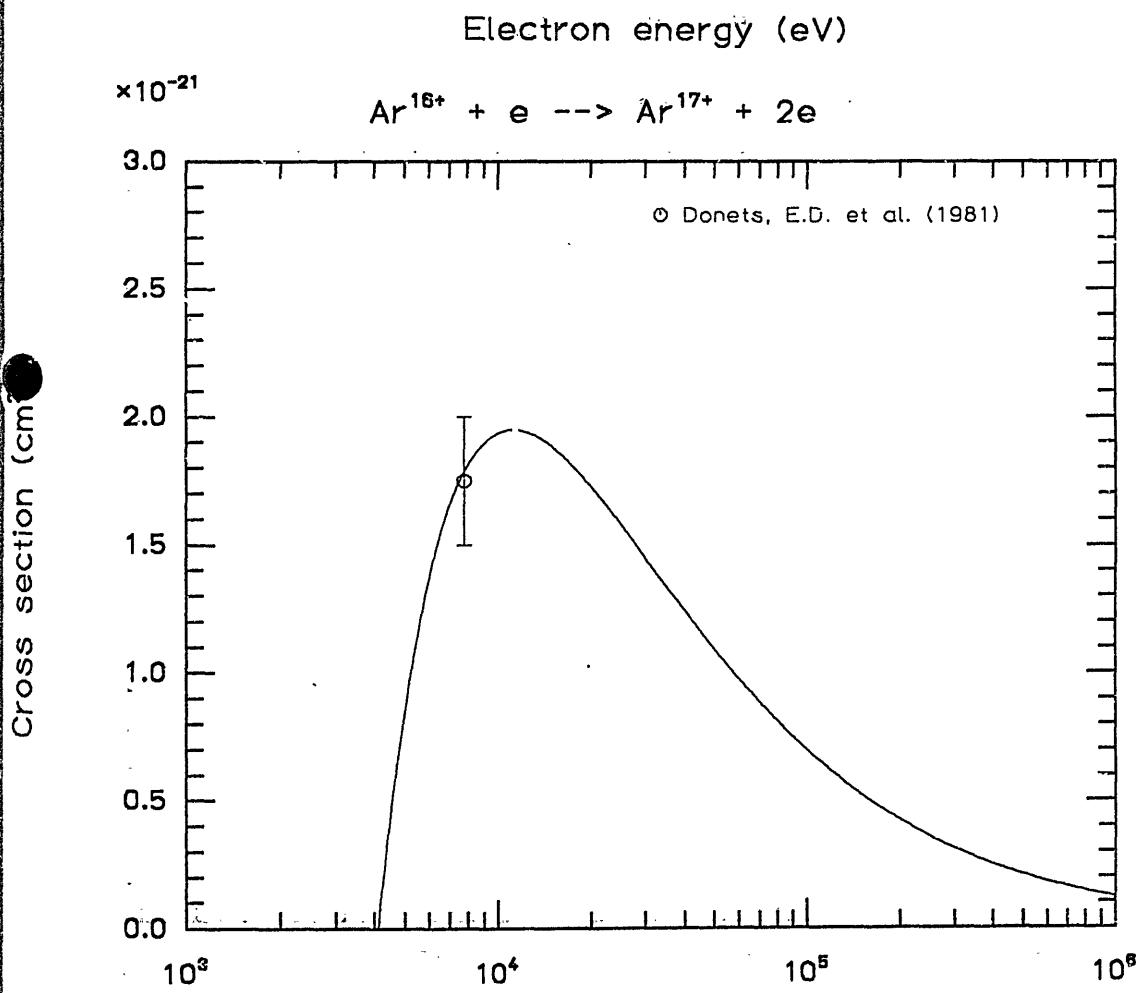
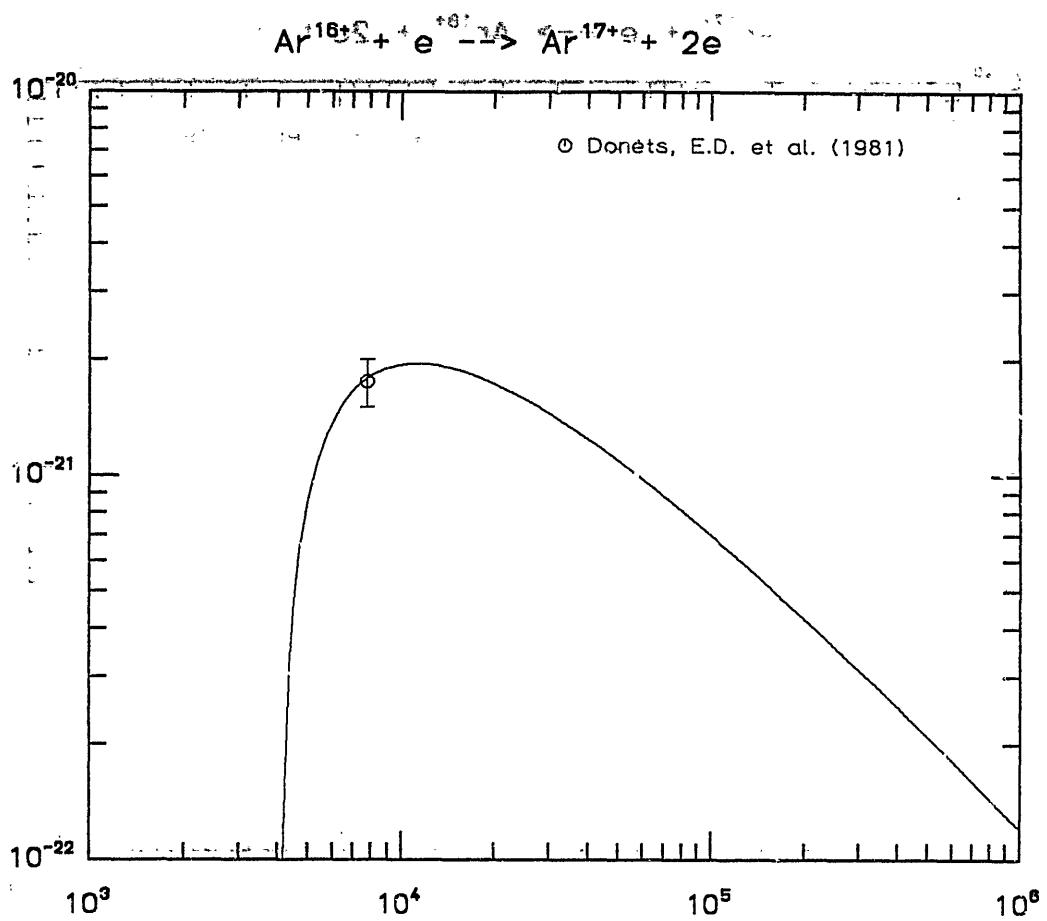


Fig. 133      Electron energy (eV)      L

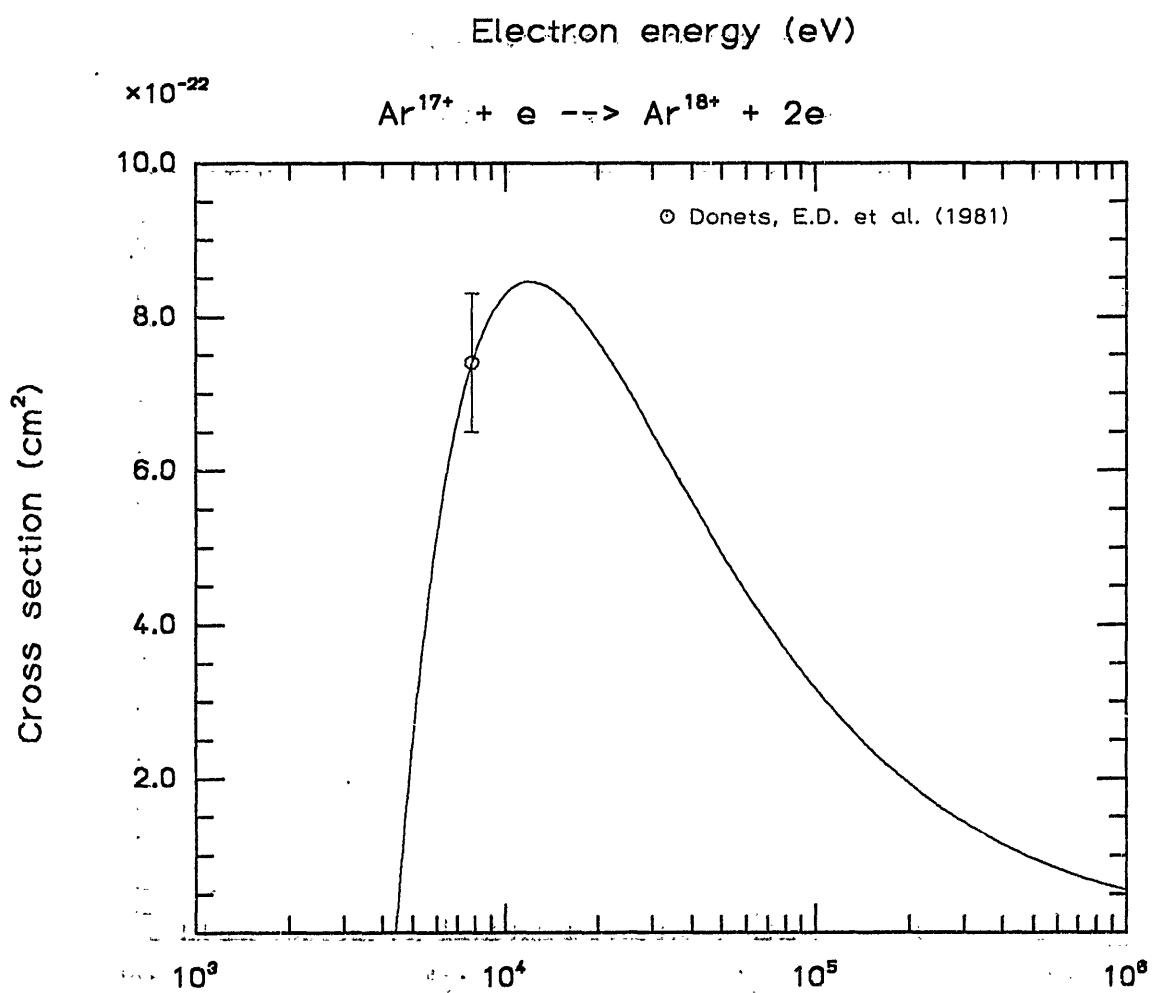
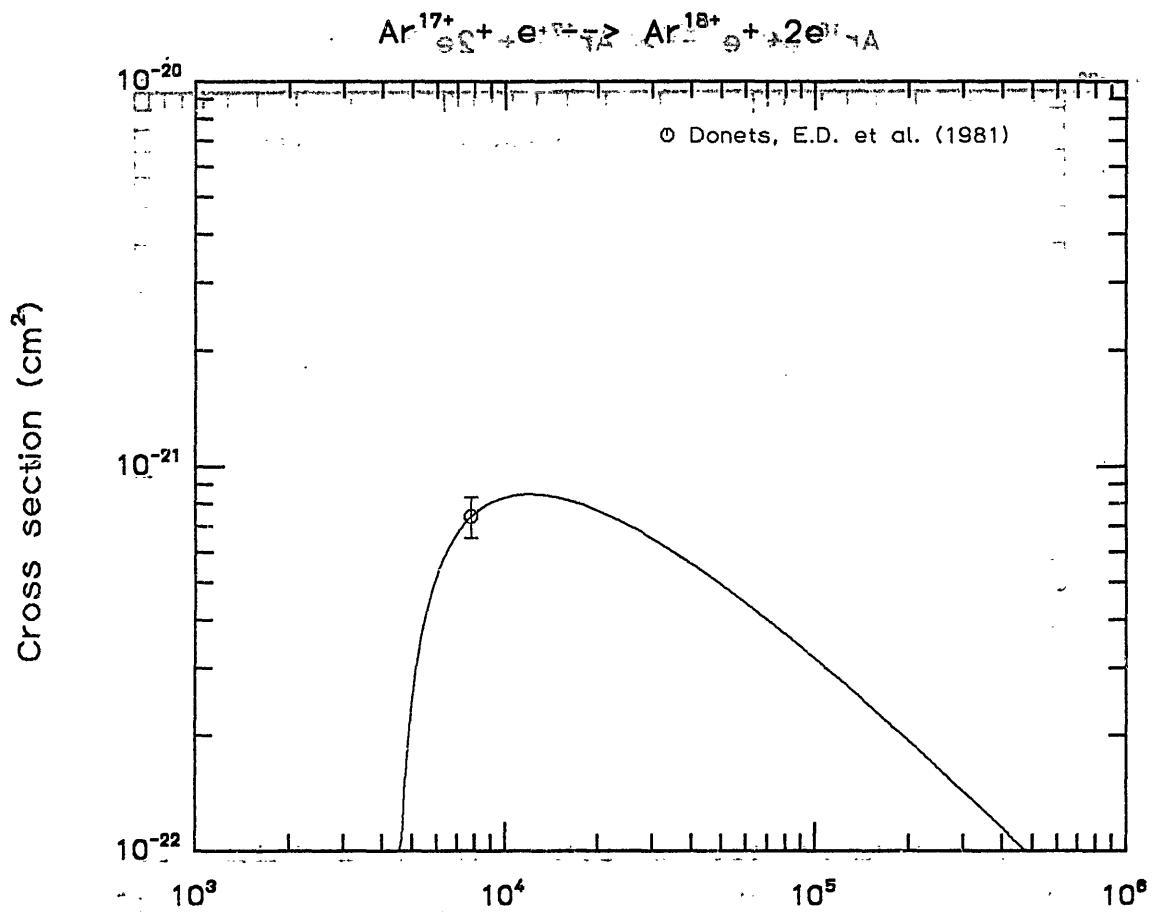
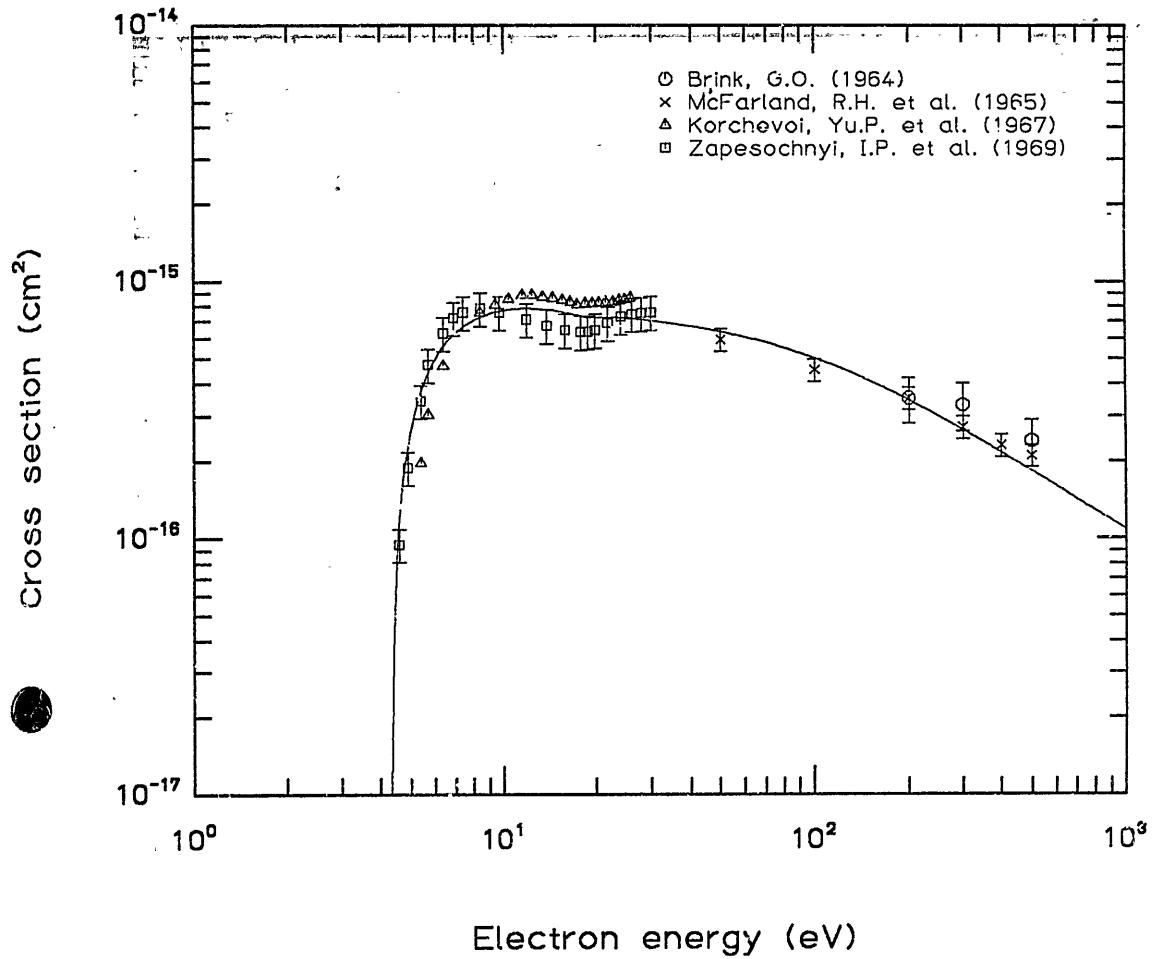
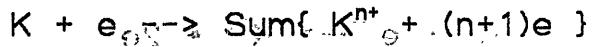


Fig. 134 (v<sub>e</sub>) Electron energy(eV) 831 83



Electron energy (eV)

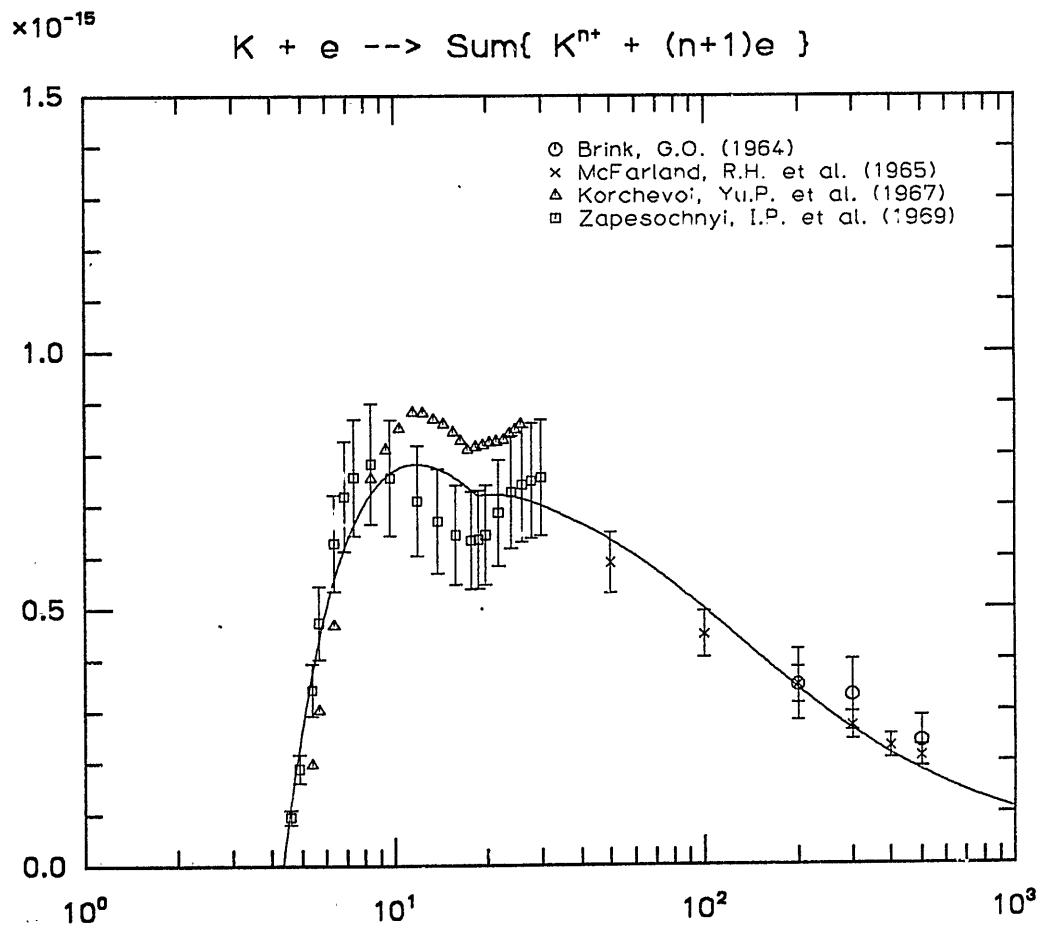


Fig. 135

Electron energy (eV)

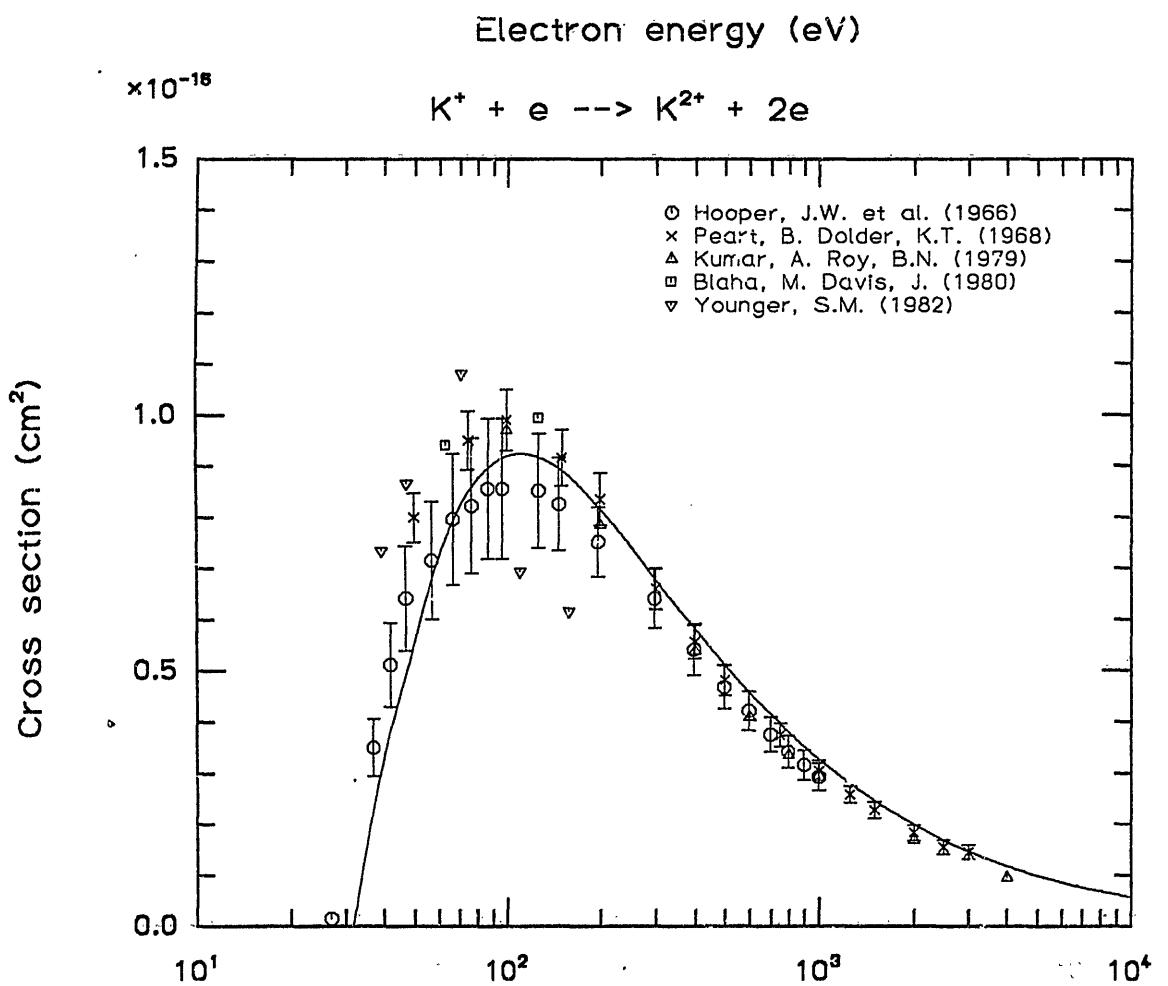
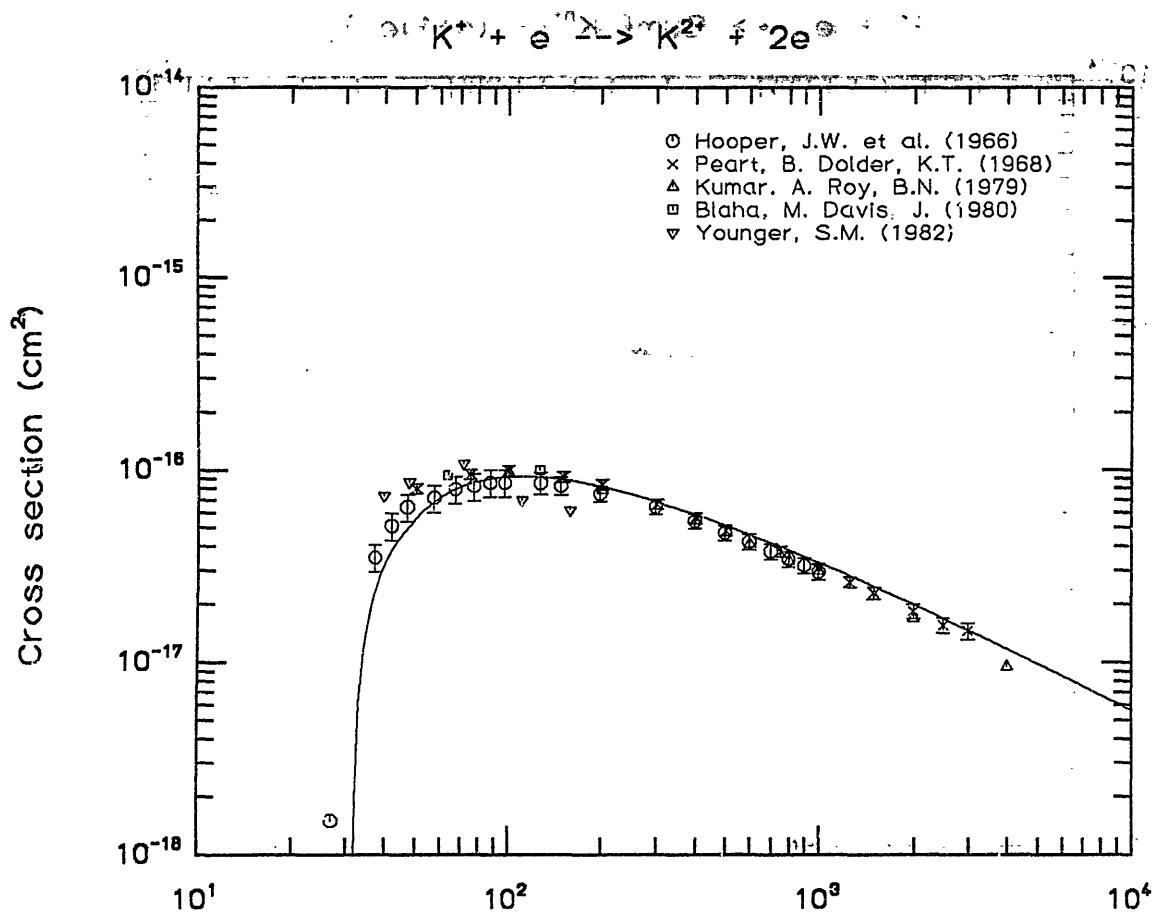
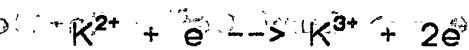
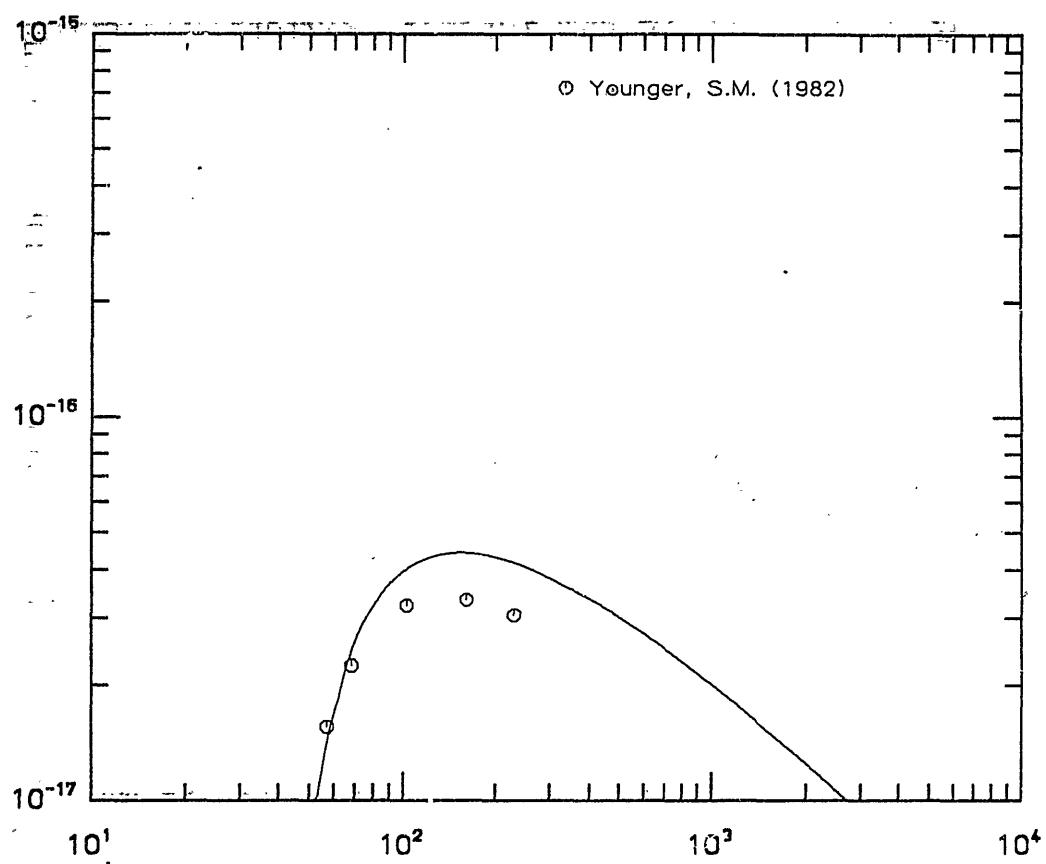


Fig. 136



Cross section ( $\text{cm}^2$ )



Cross section ( $\text{cm}^2$ )

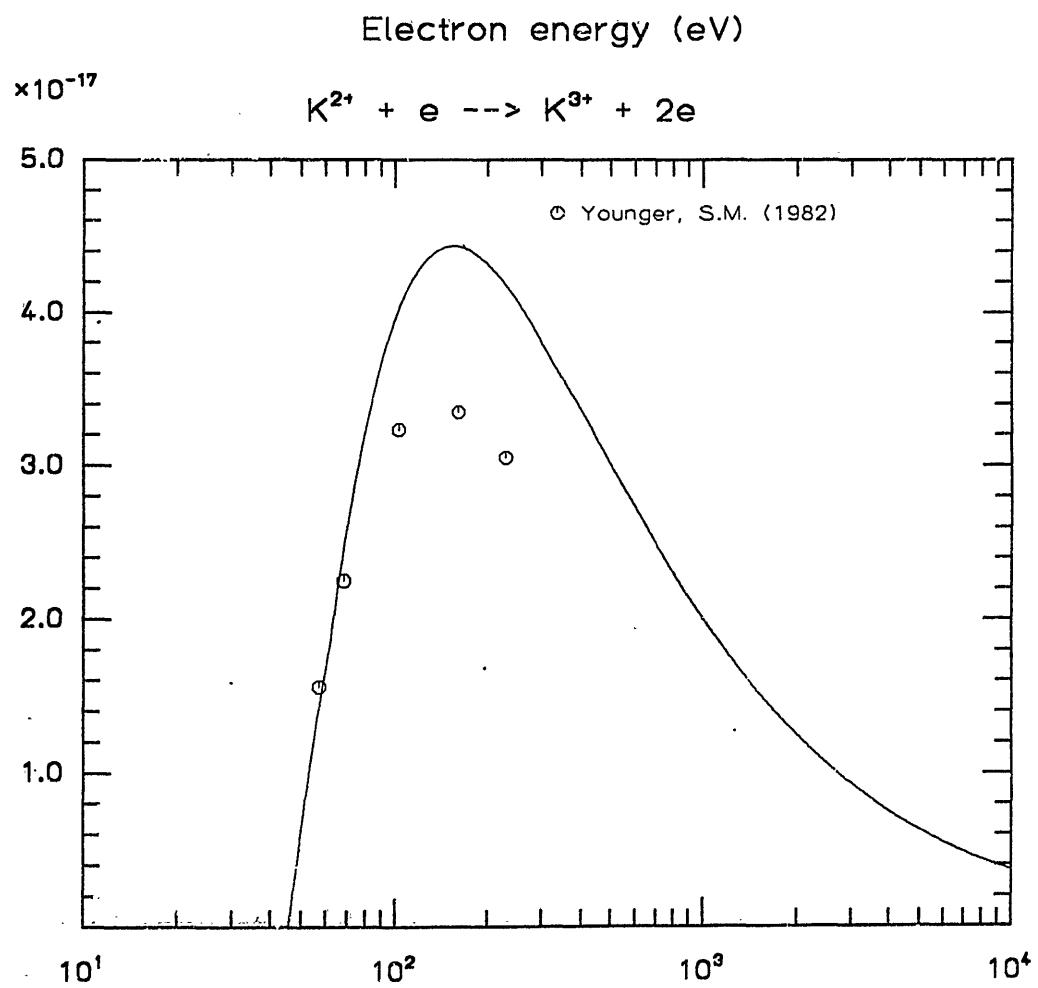


Fig. 137      Electron energy (eV)

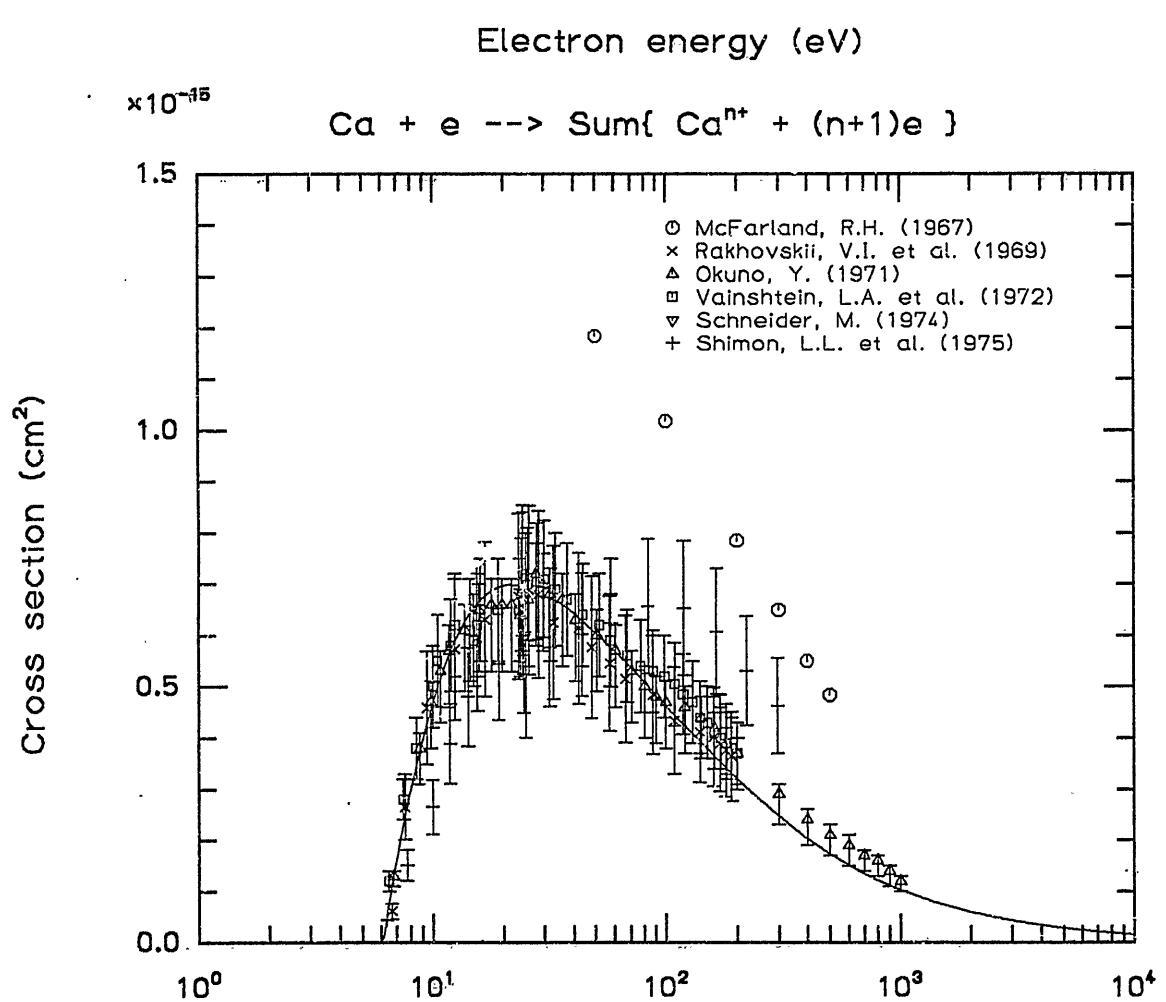
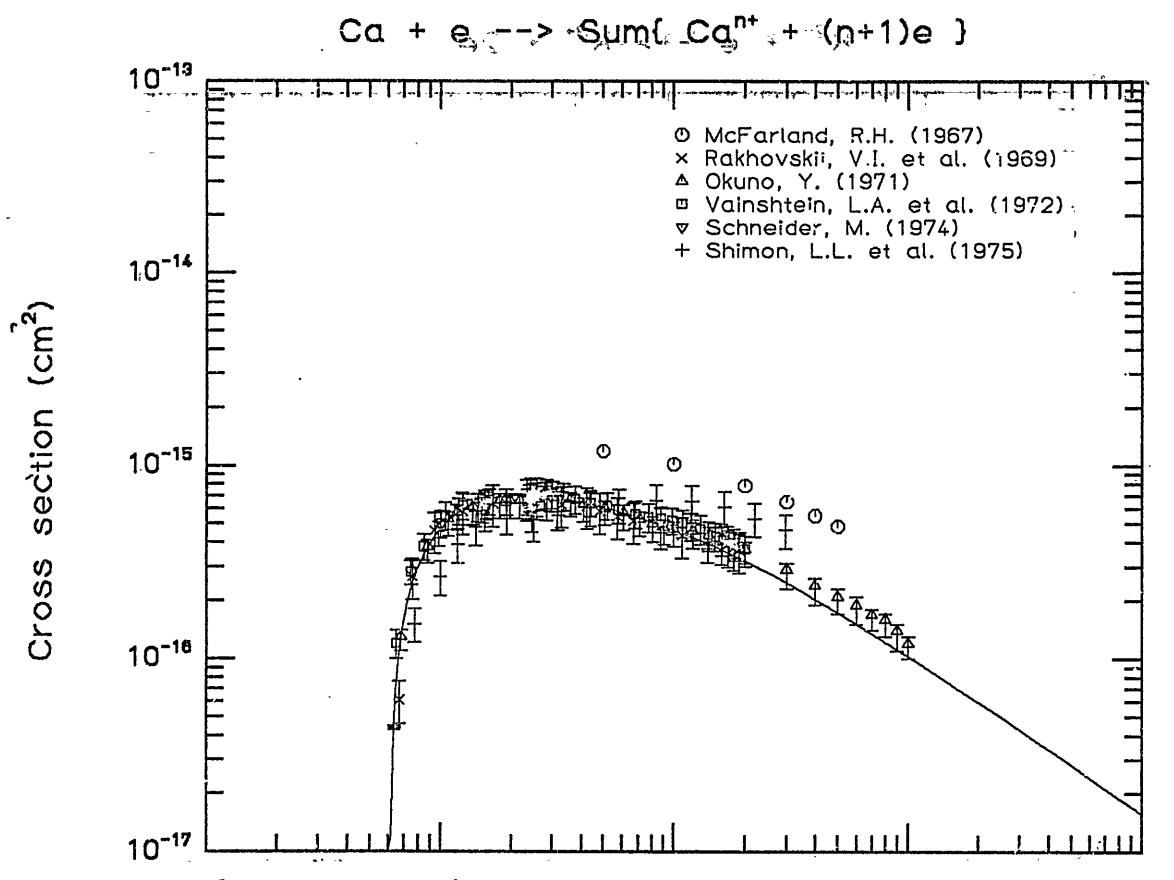
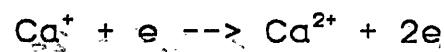
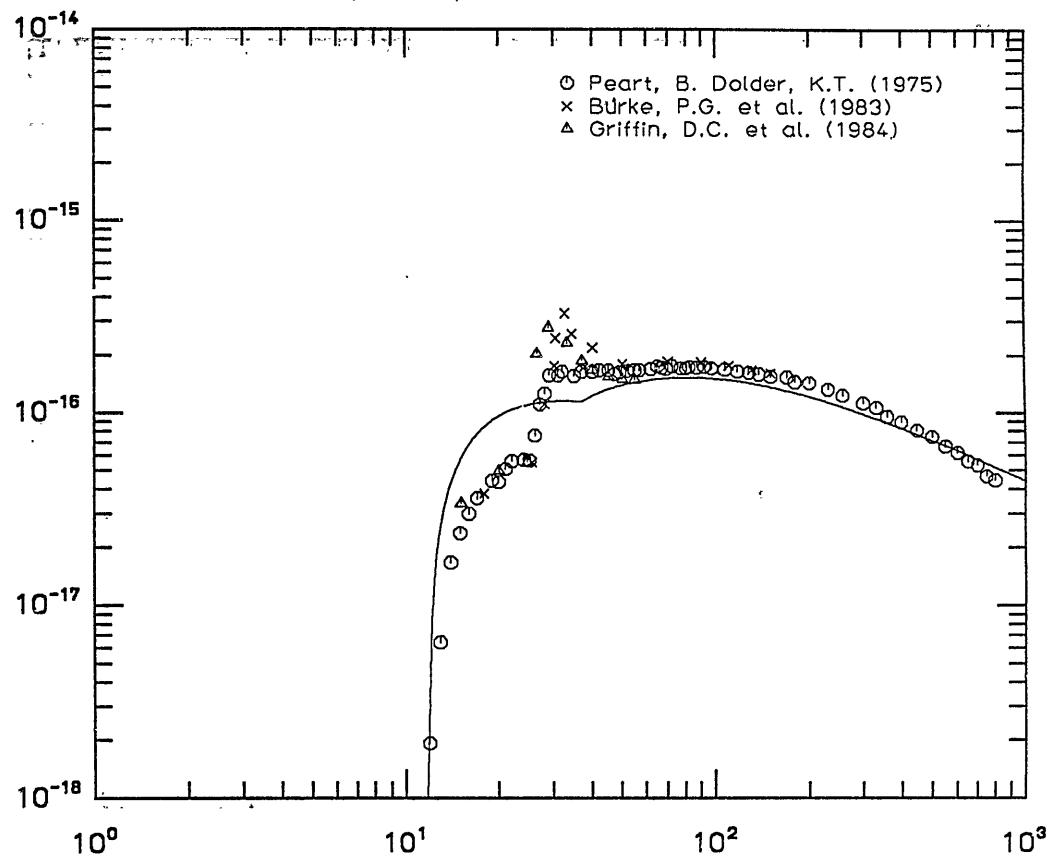


Fig. 138



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

Cross section ( $\text{cm}^2$ )

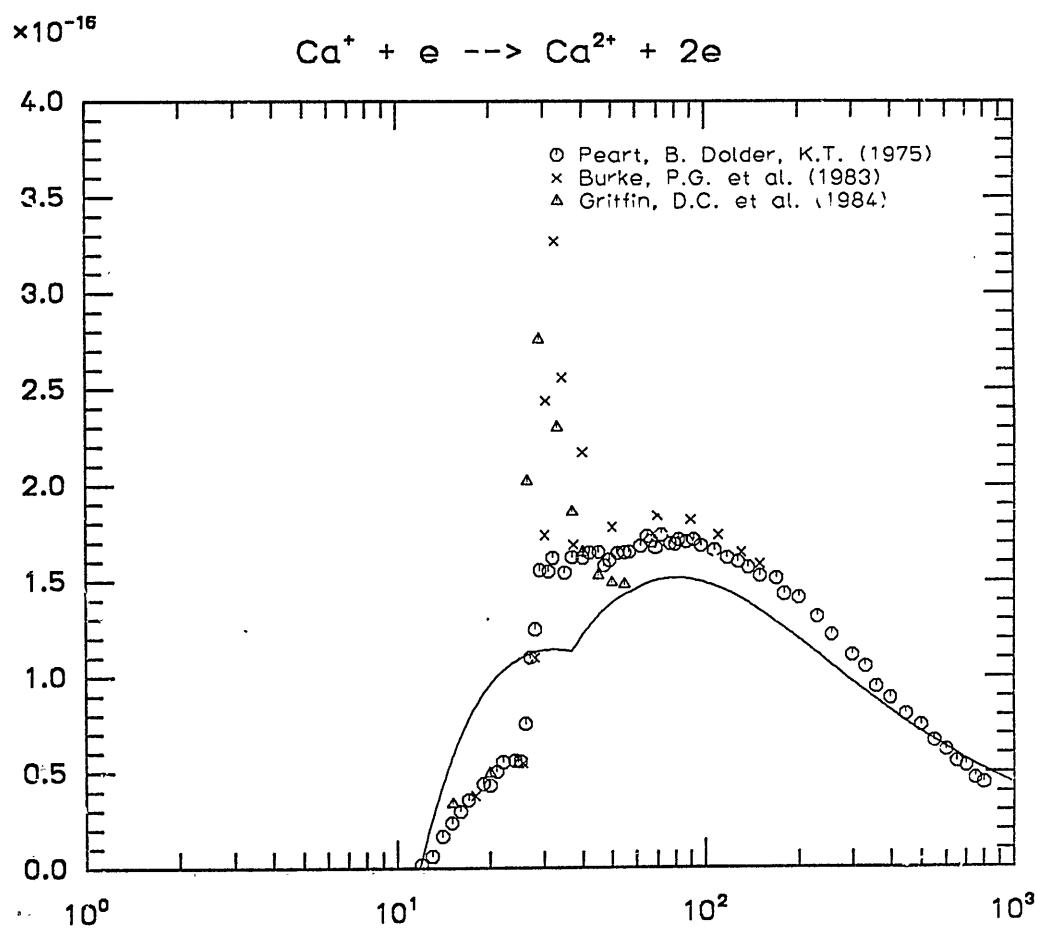


Fig. 139      Electron energy (eV)

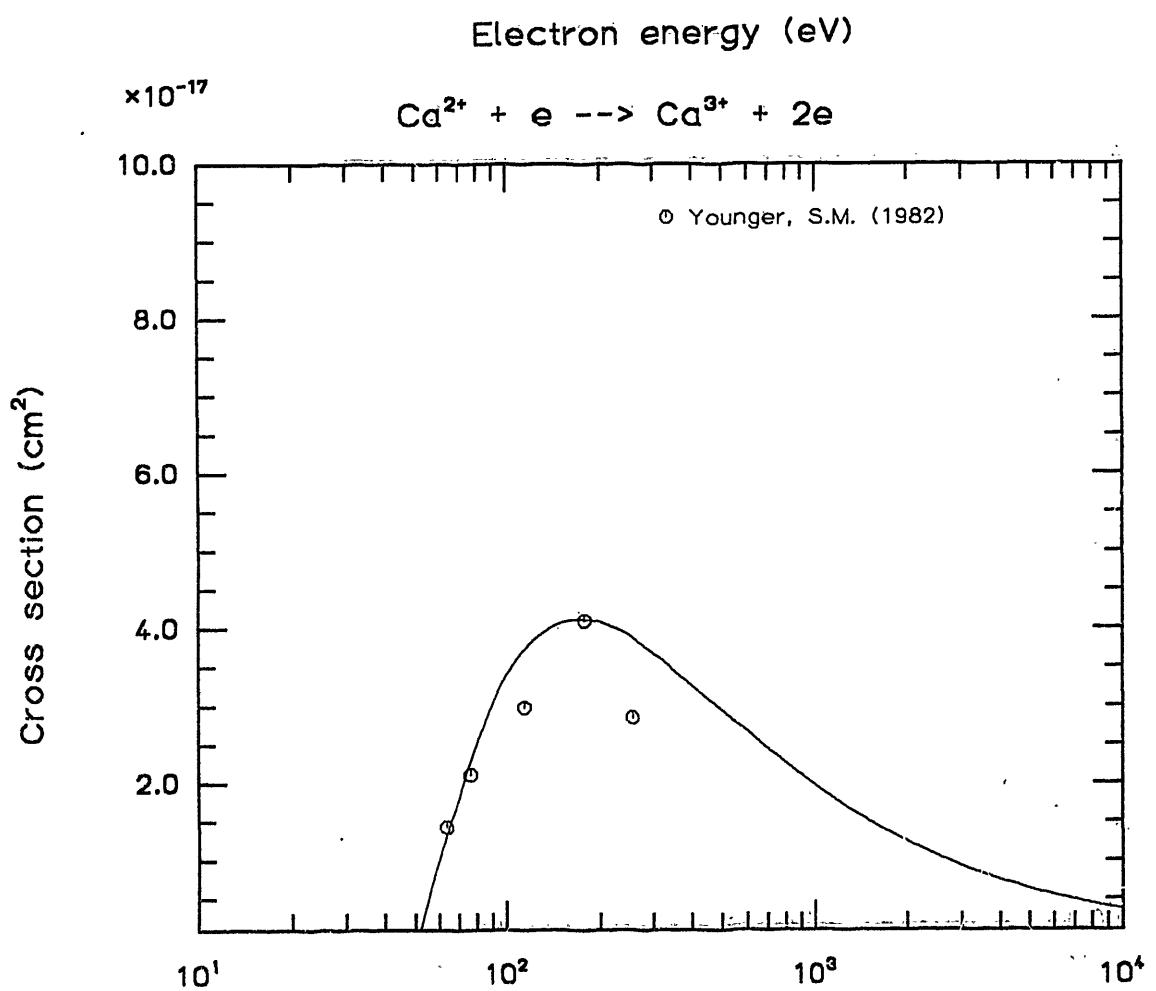
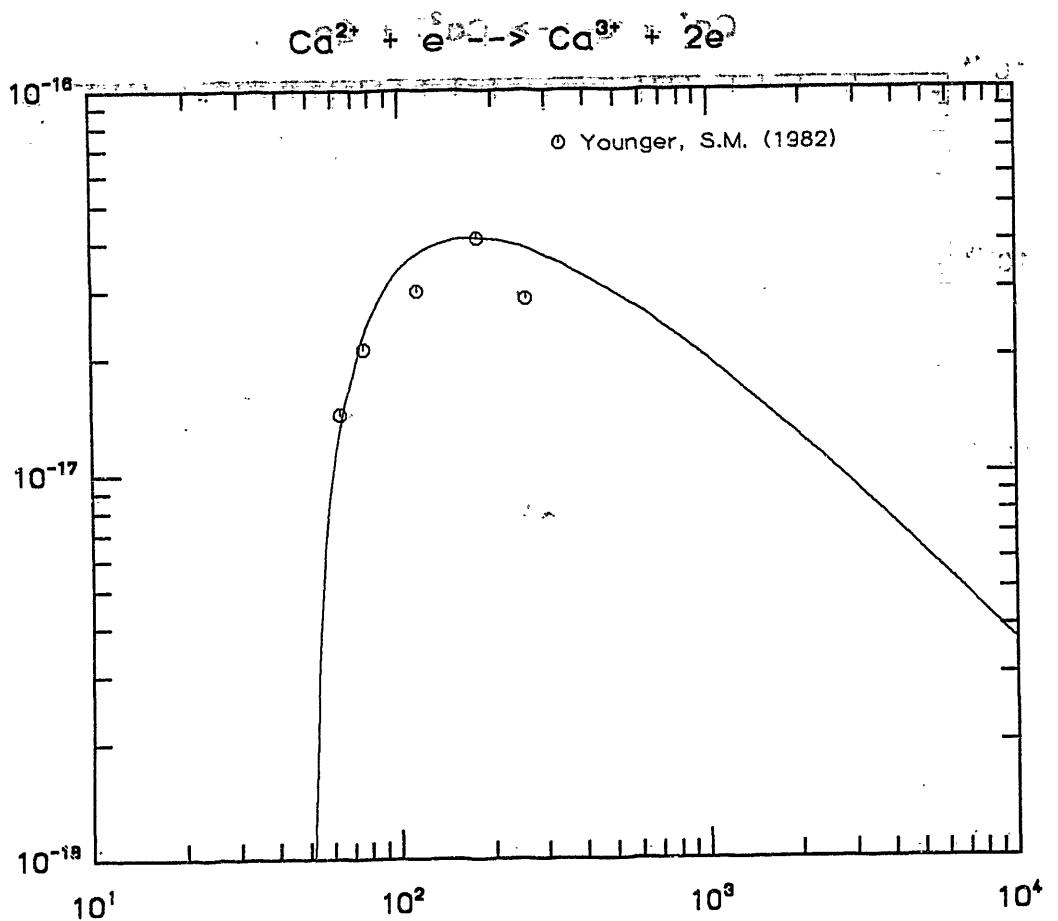


Fig. 140      Electron energy (eV)      Fig. 131

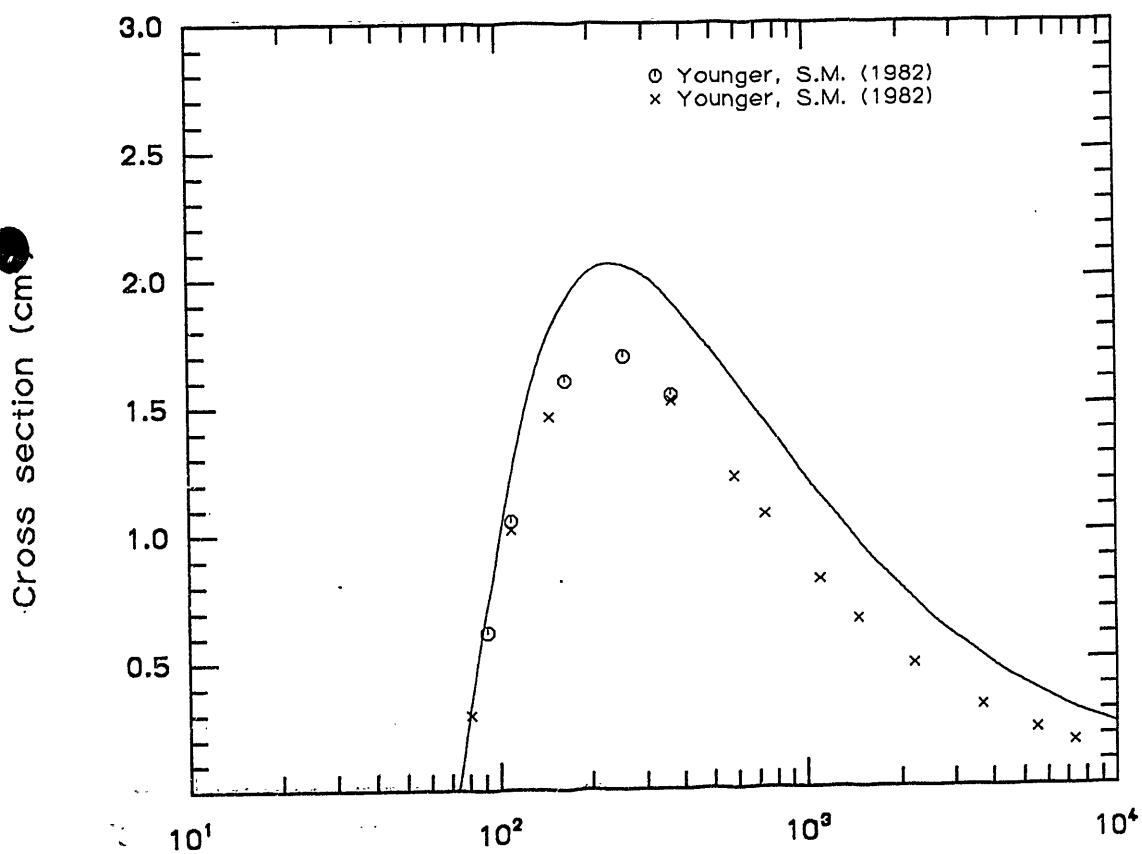
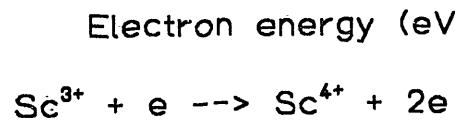
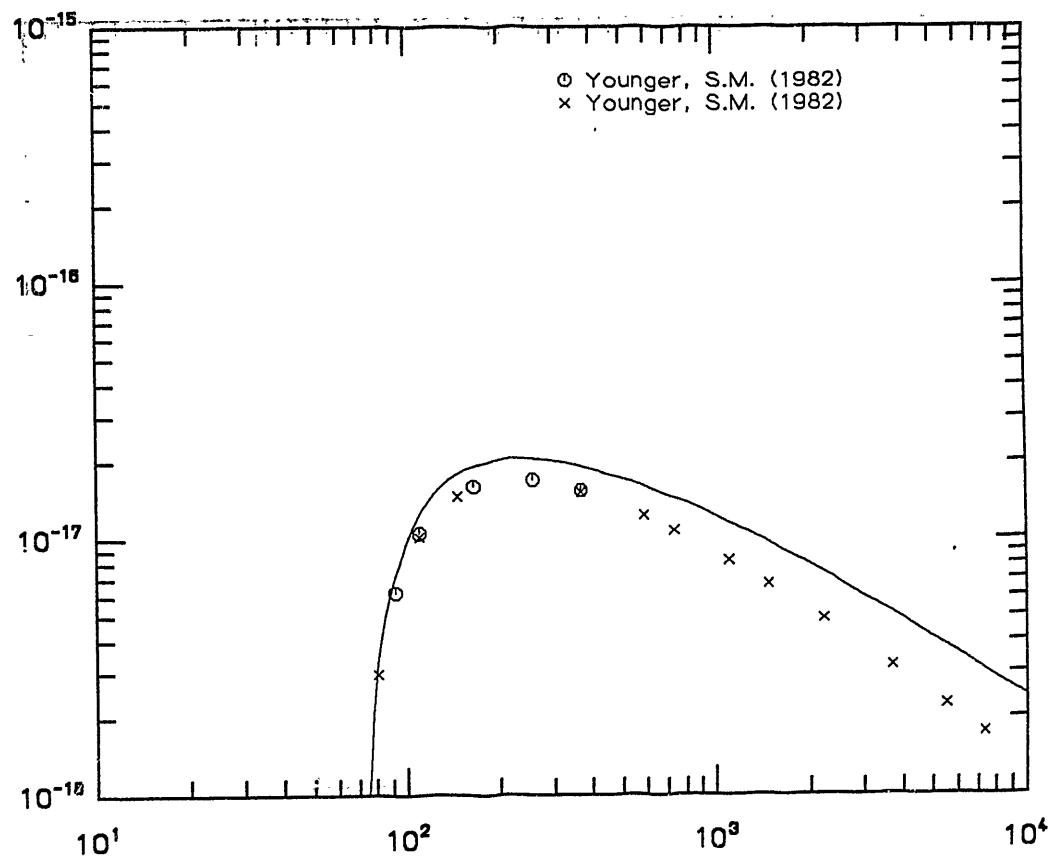
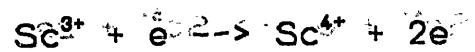


Fig. 141 : Electron energy (eV)

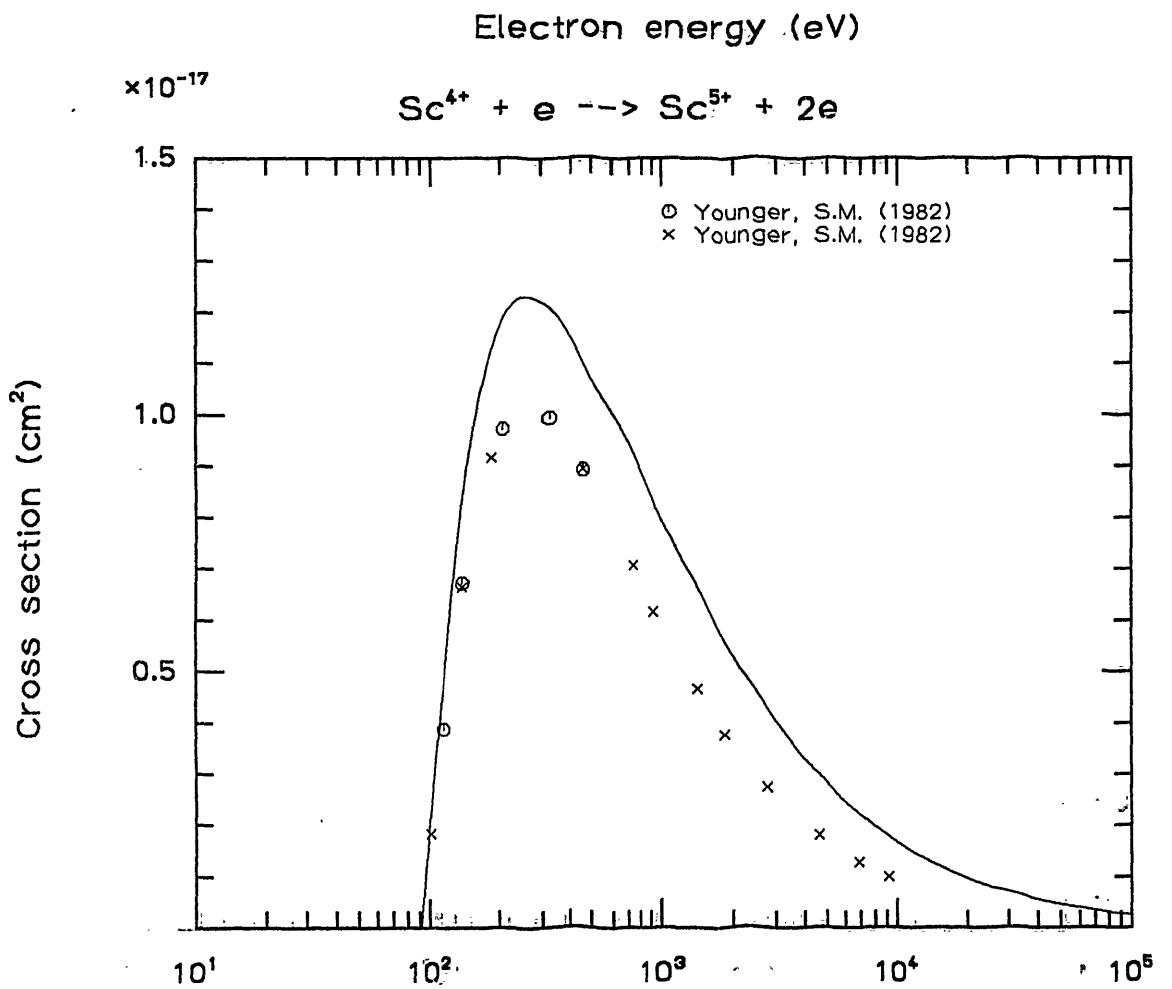
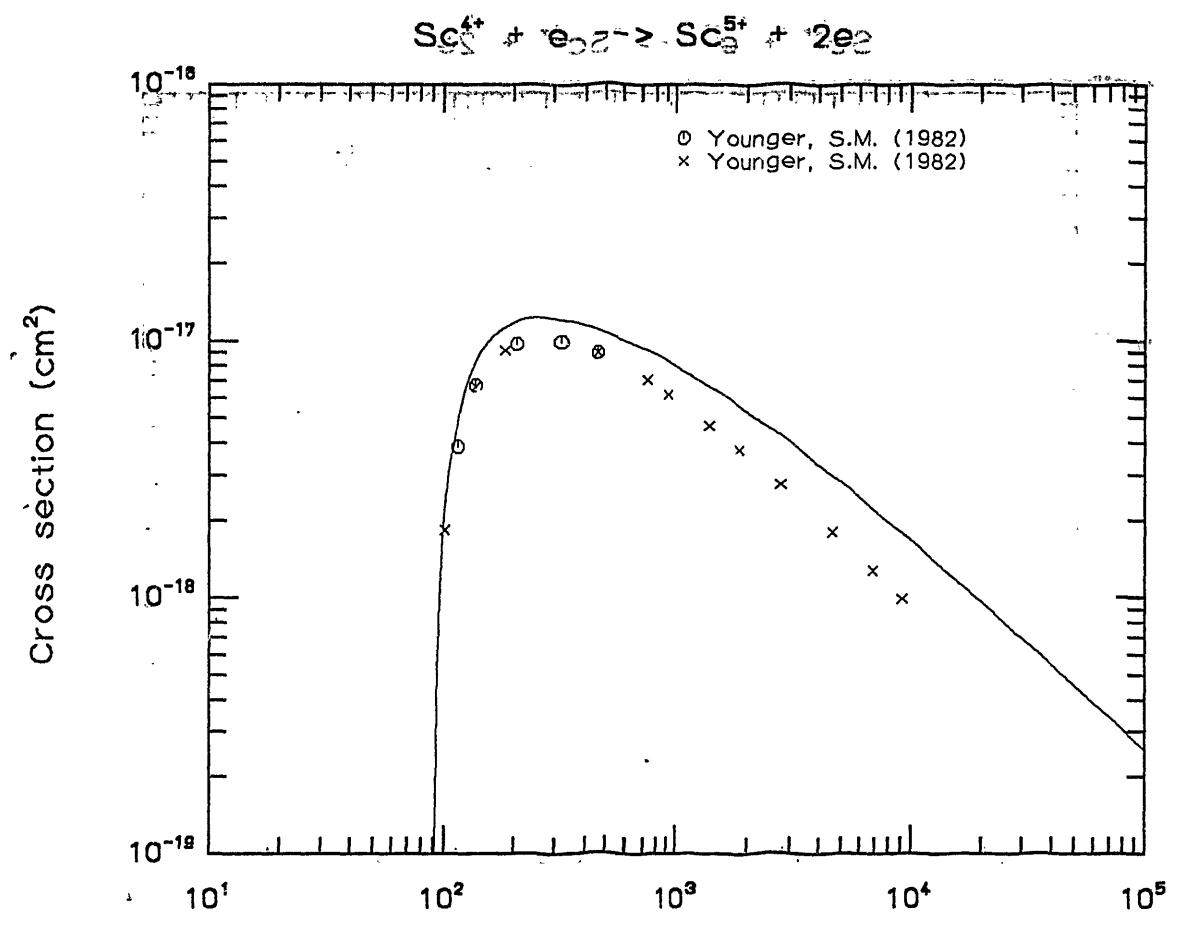
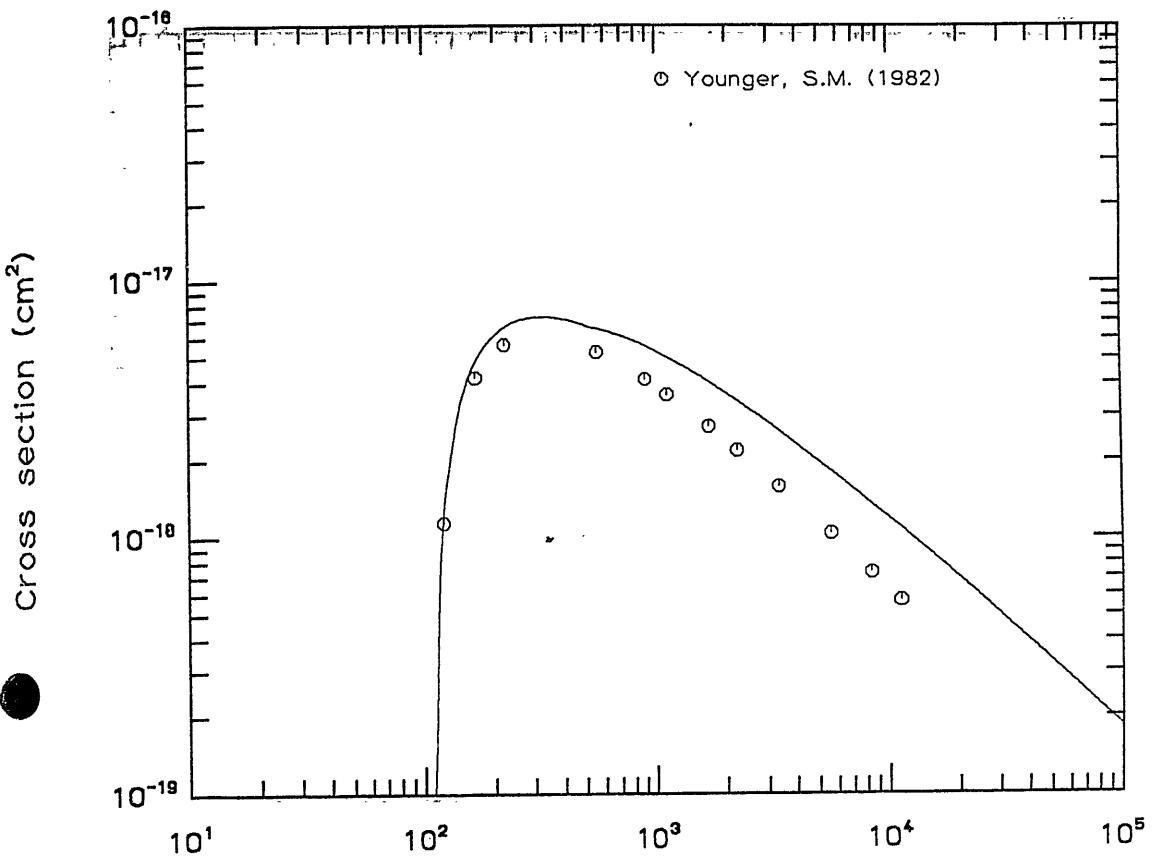
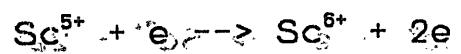


Fig. 142 (a) Electron energy (eV) (b)



Electron energy (eV)

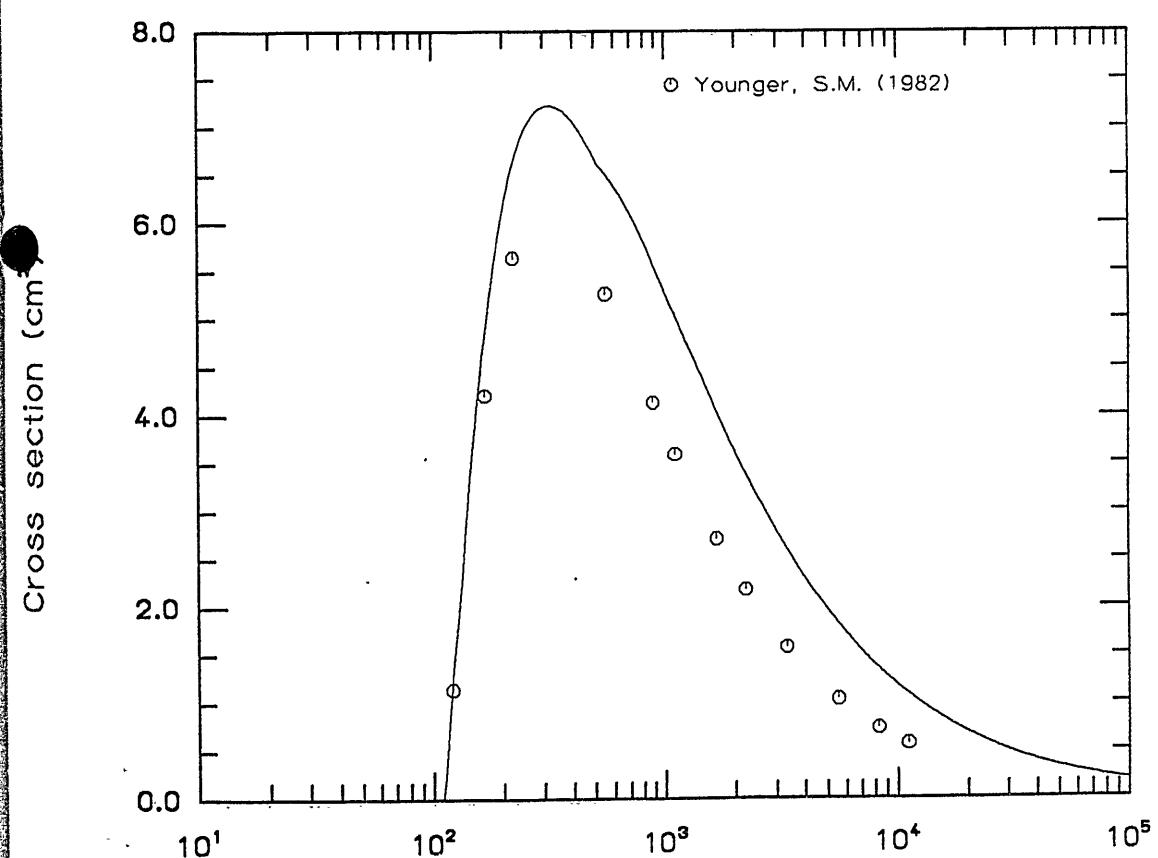
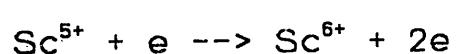


Fig. 143 Electron energy (eV)

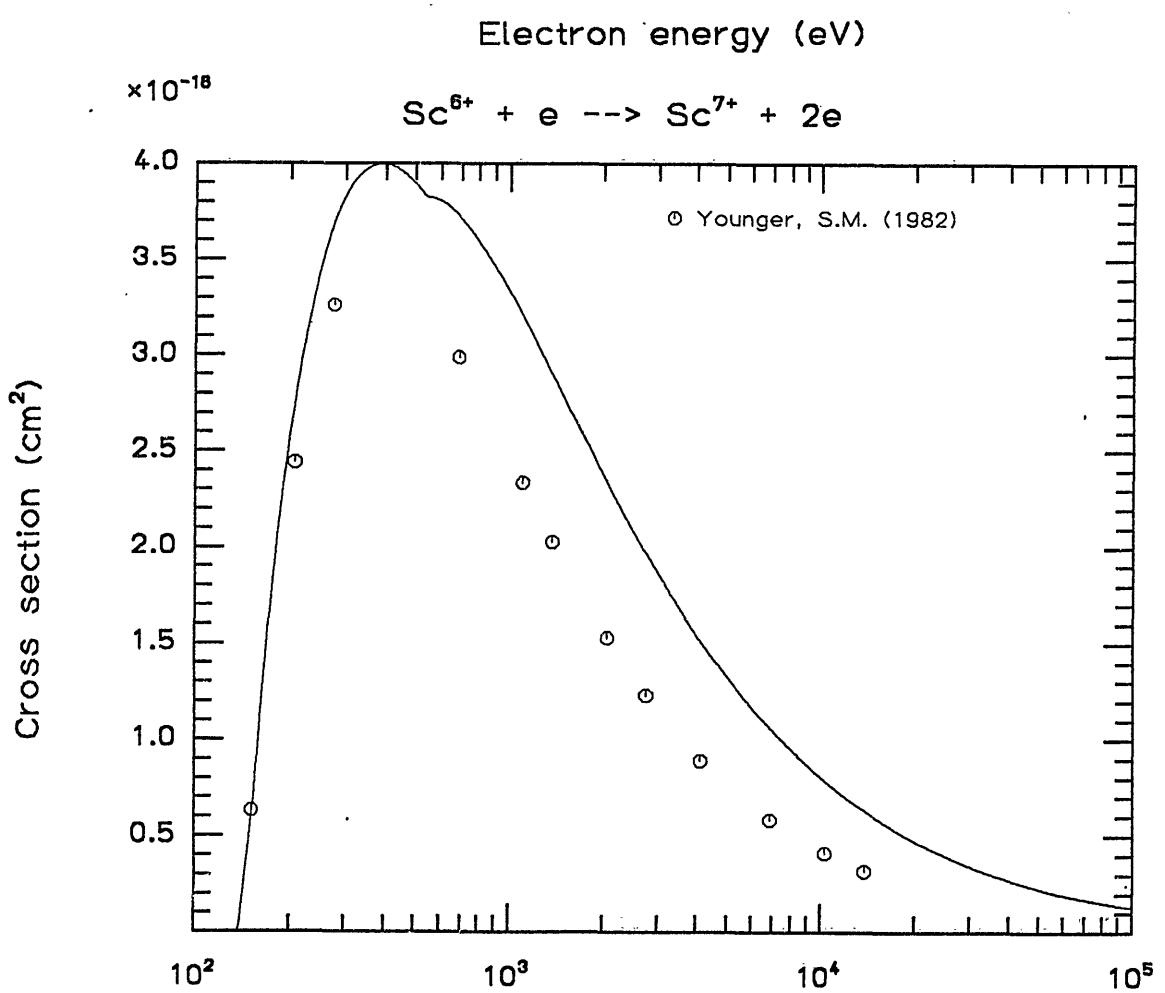
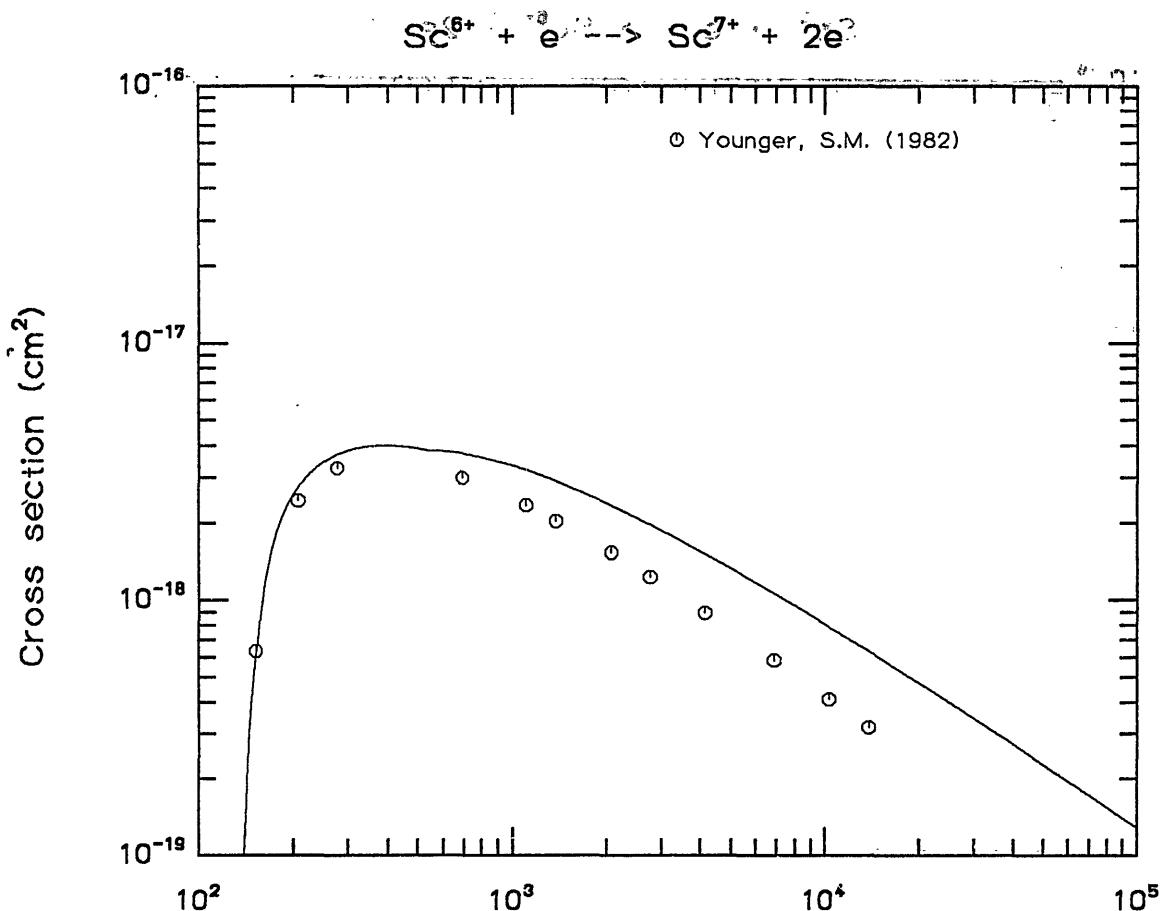
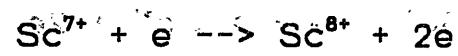
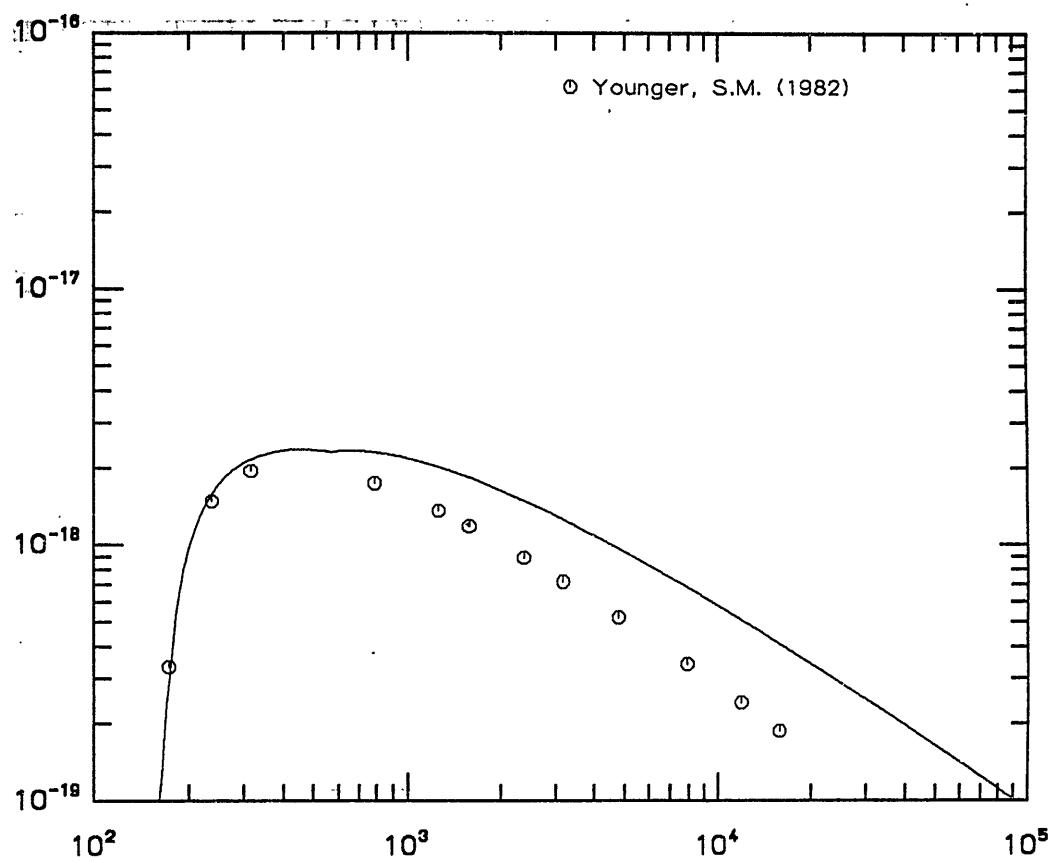


Fig. 144

Electron energy (eV)  $\text{Sc}^{6+} \rightarrow \text{Sc}^{7+}$



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

Cross section ( $\text{cm}^2$ )

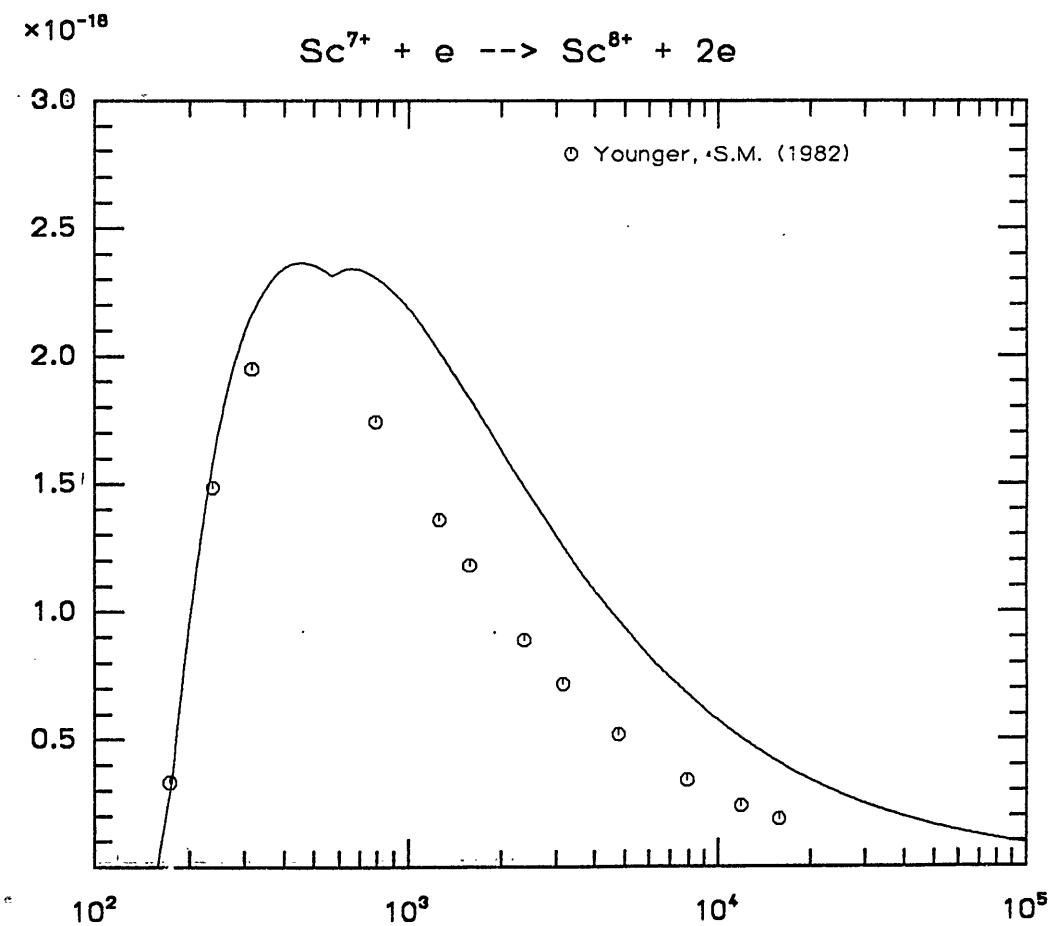


Fig. 145      • Electron energy (eV)

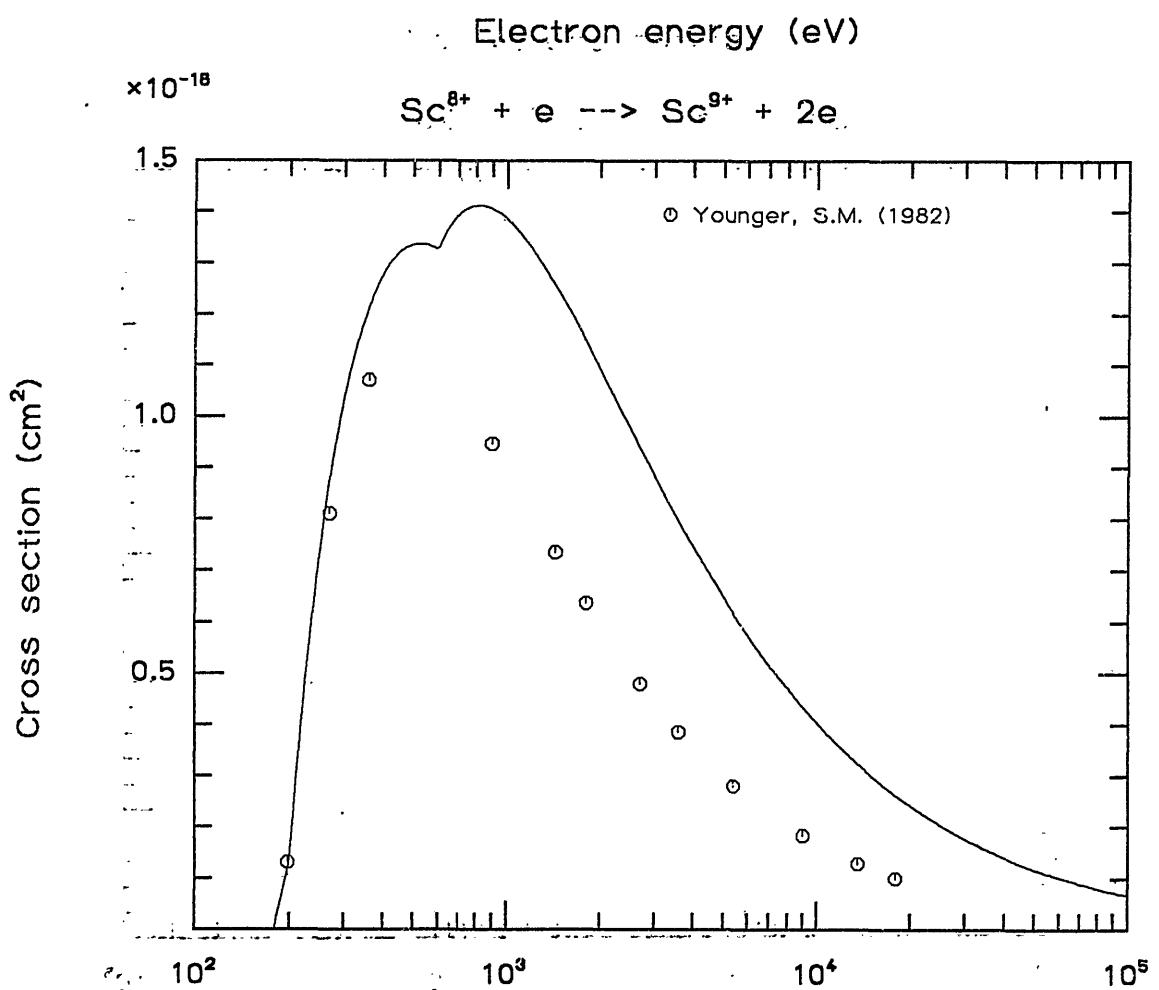
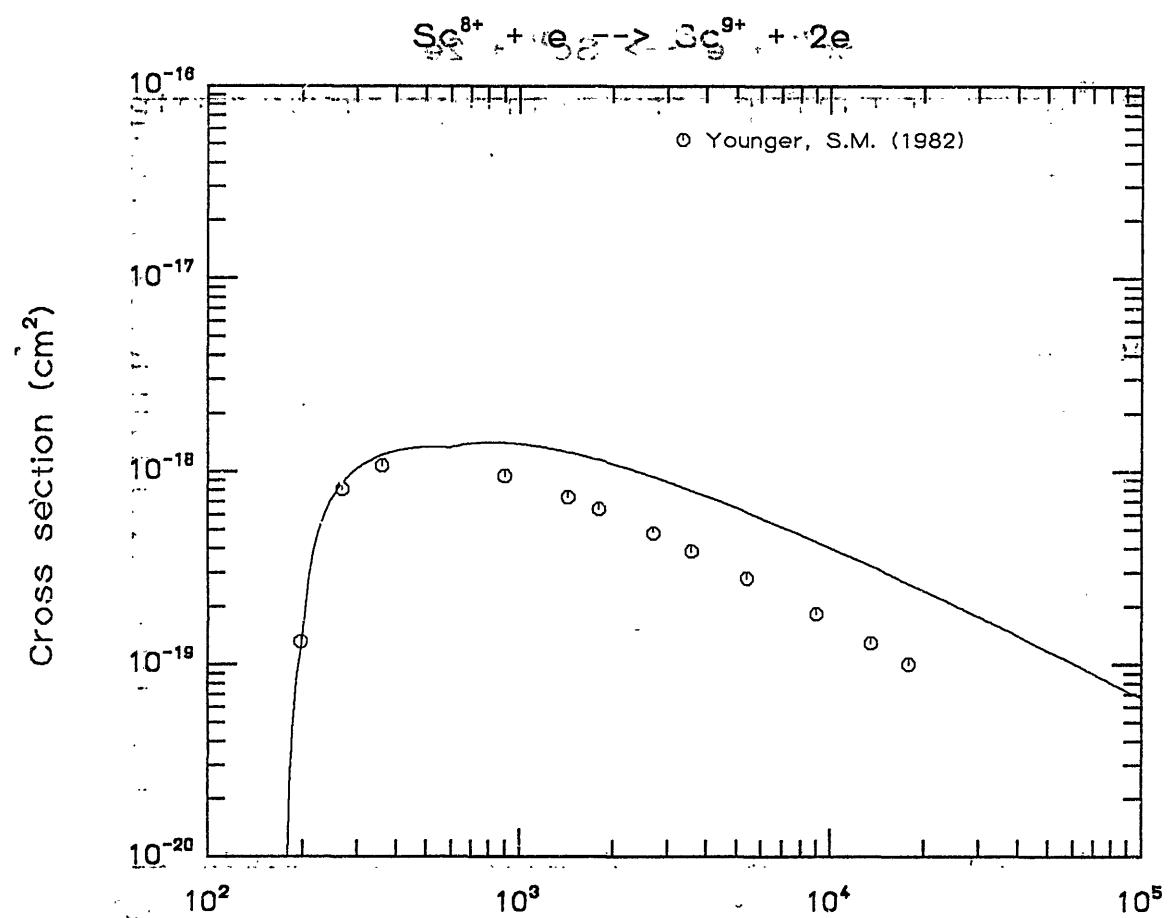


Fig. 146 (a) Electron energy (eV) (b)

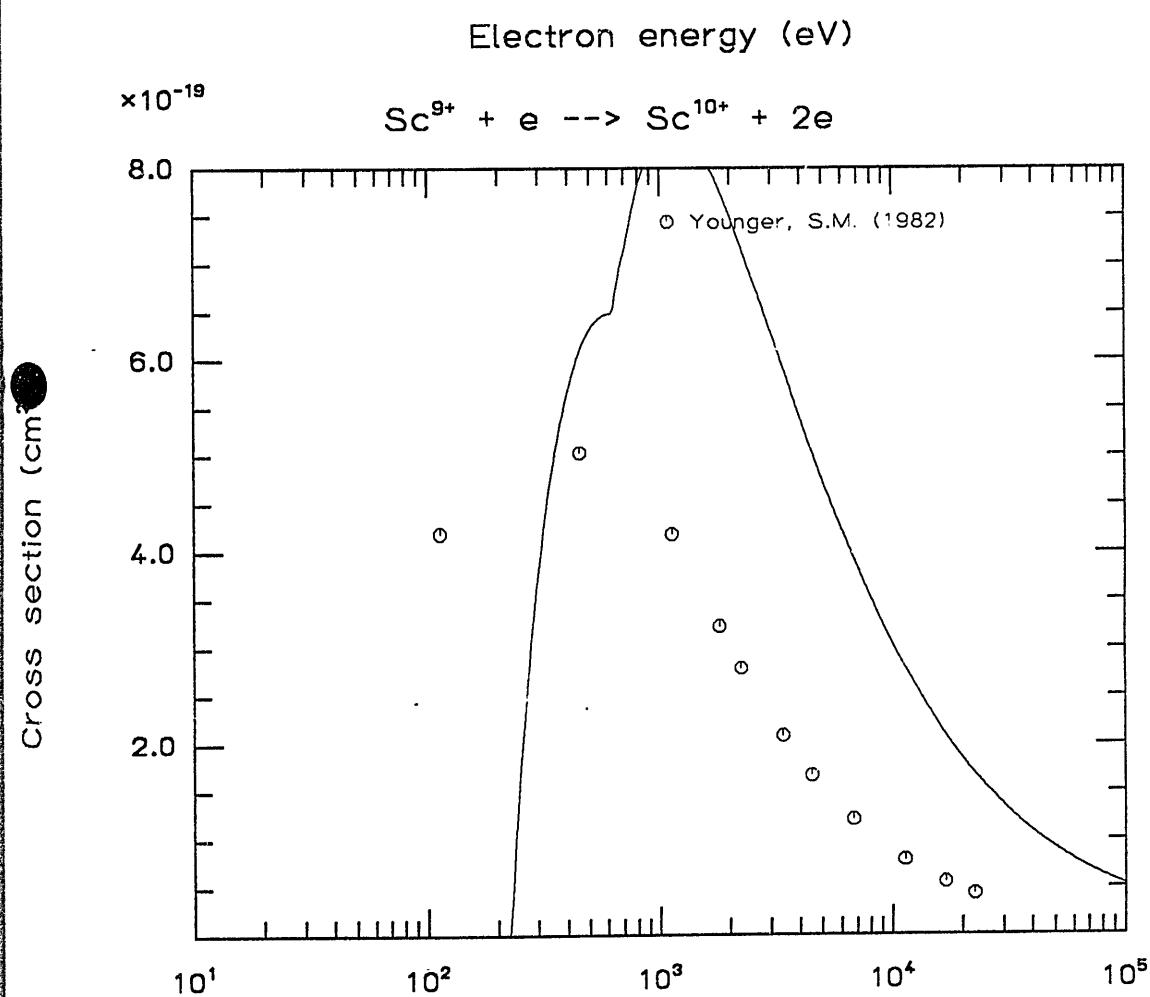
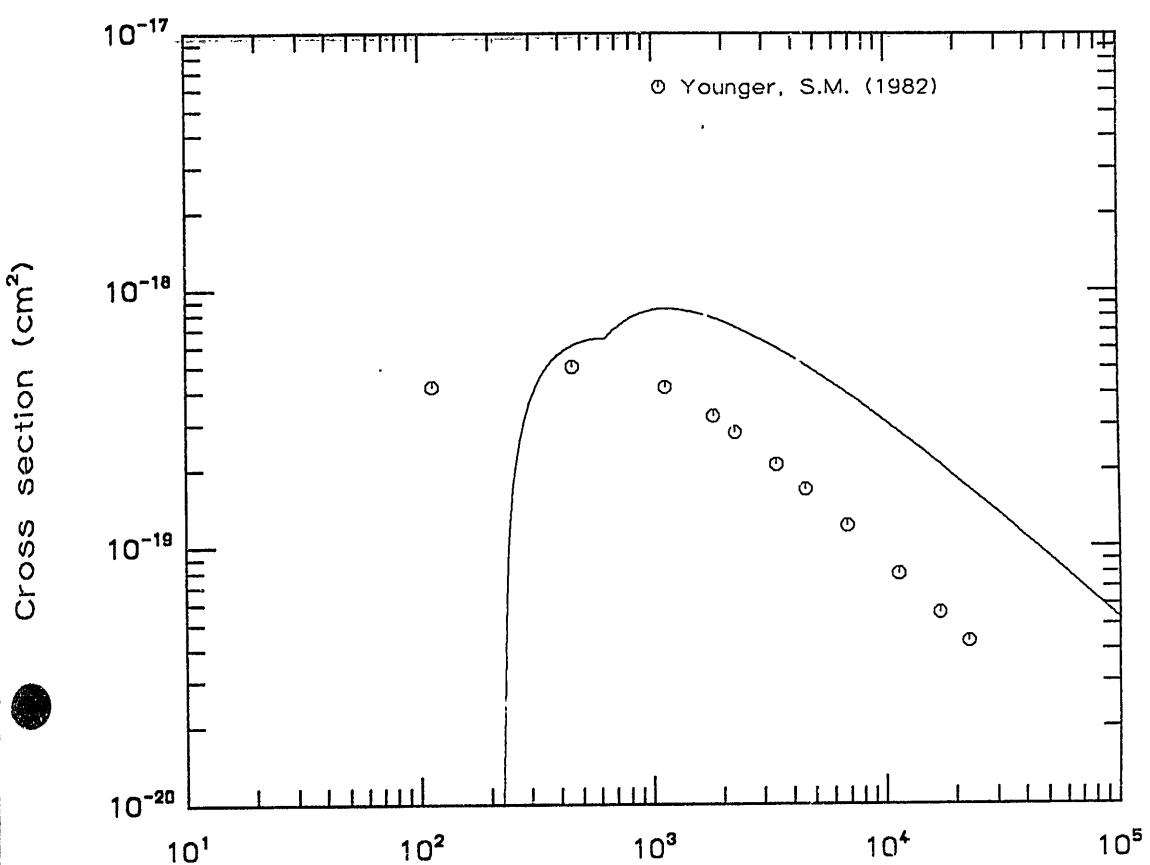
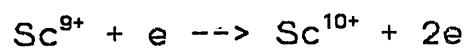


Fig. 147 . Electron energy (eV)

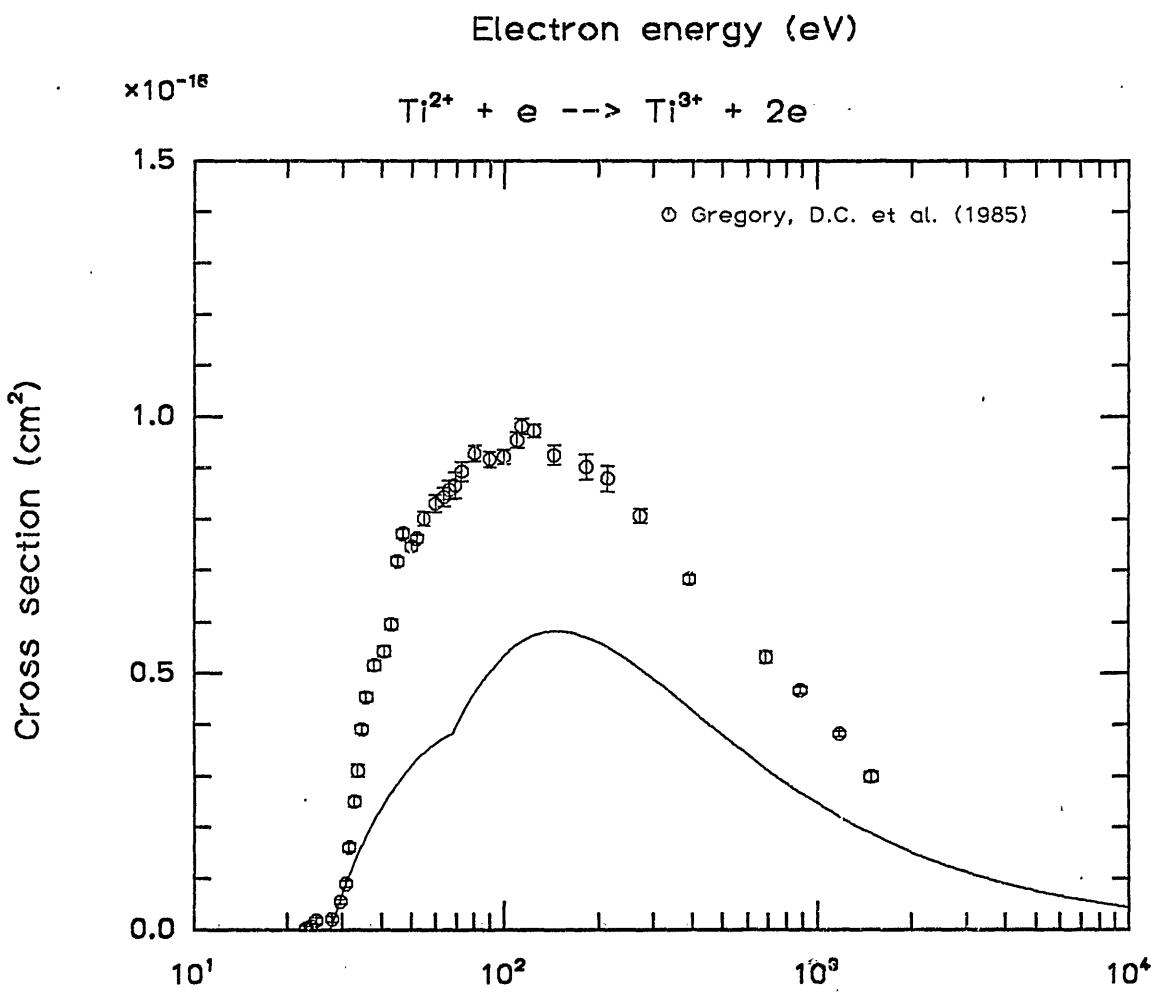
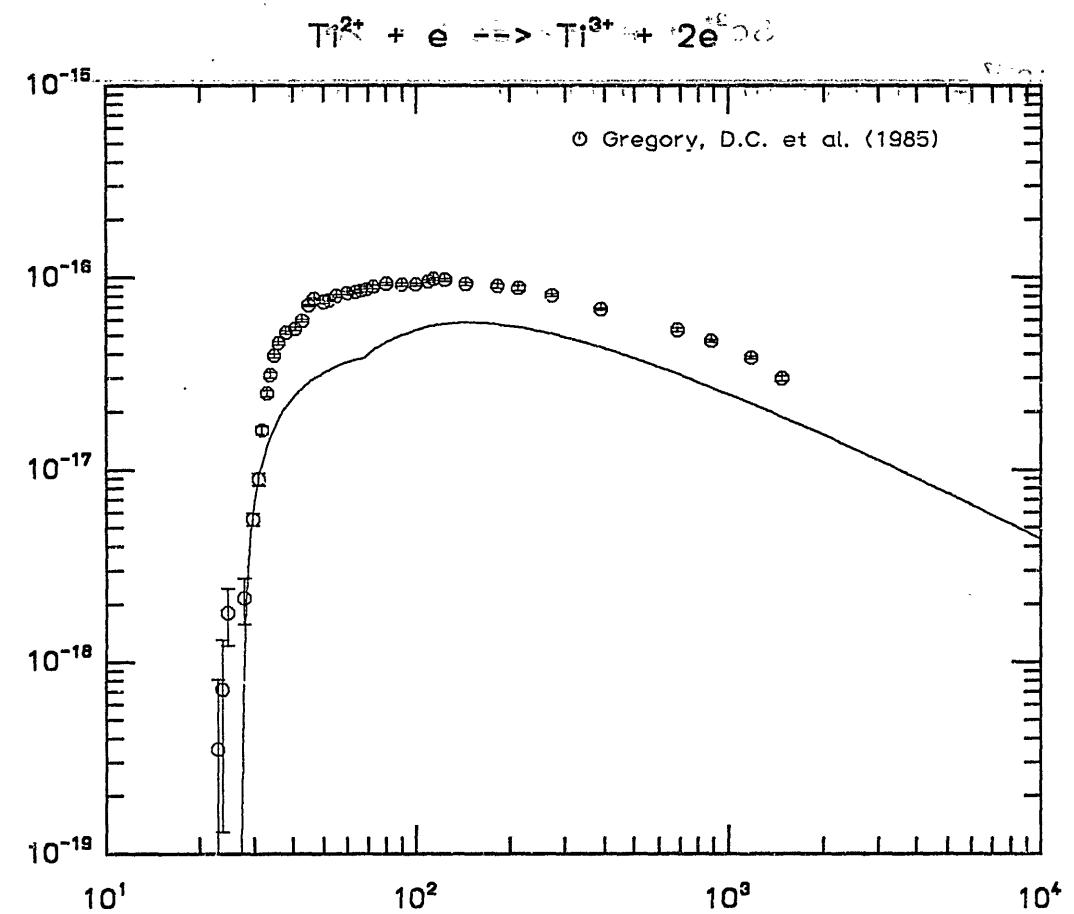


Fig. 148      Electron energy (eV)  $\sqrt{\Delta E_{\text{kin}}}$

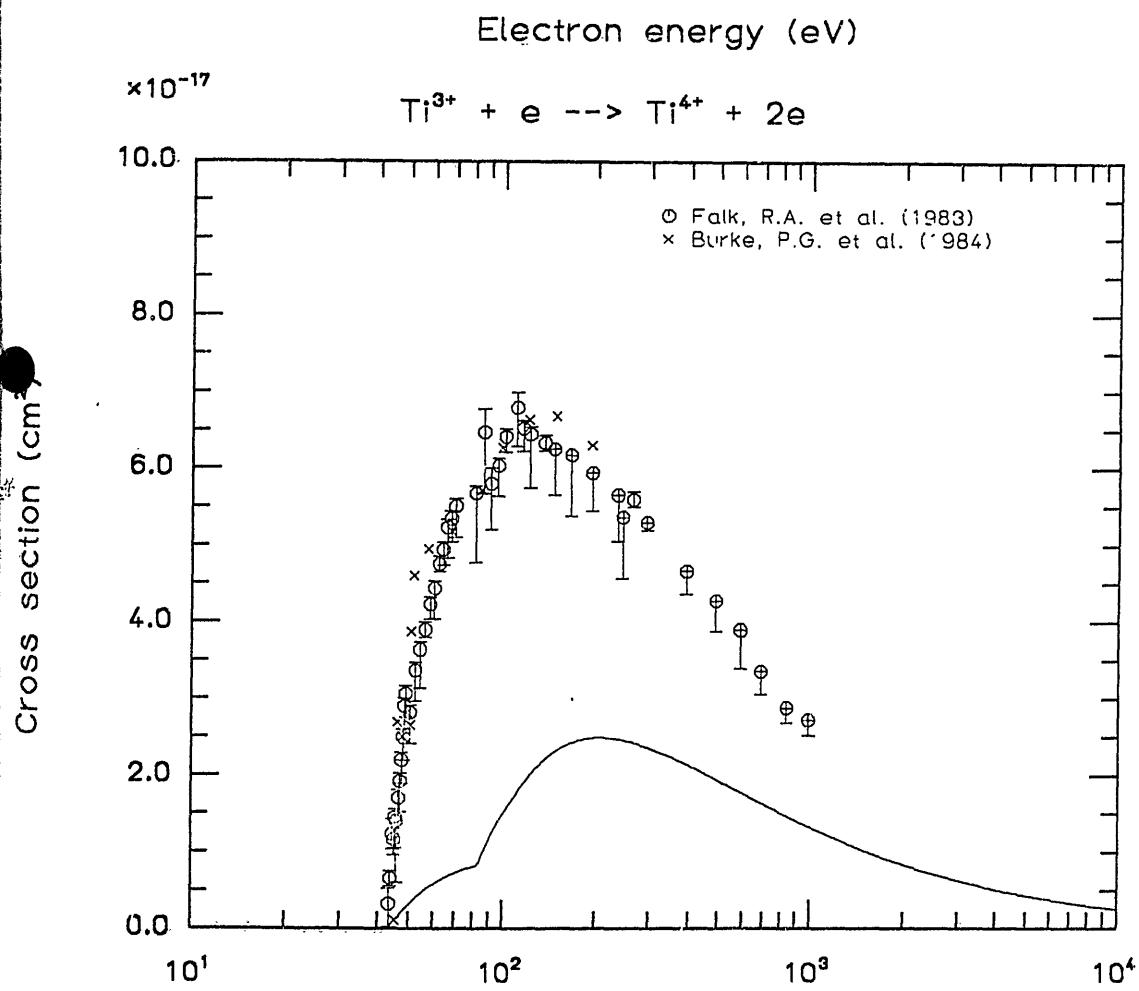
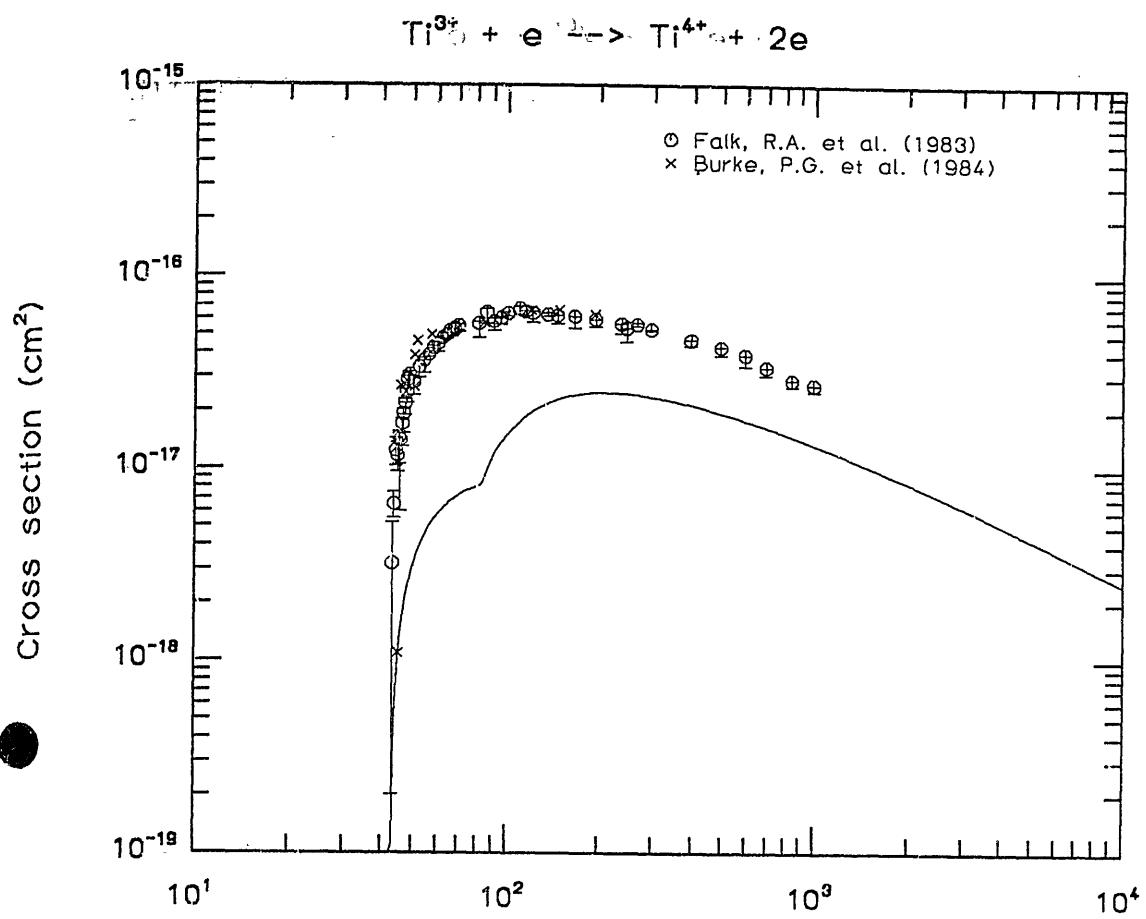


Fig. 149 · Electron energy (eV)

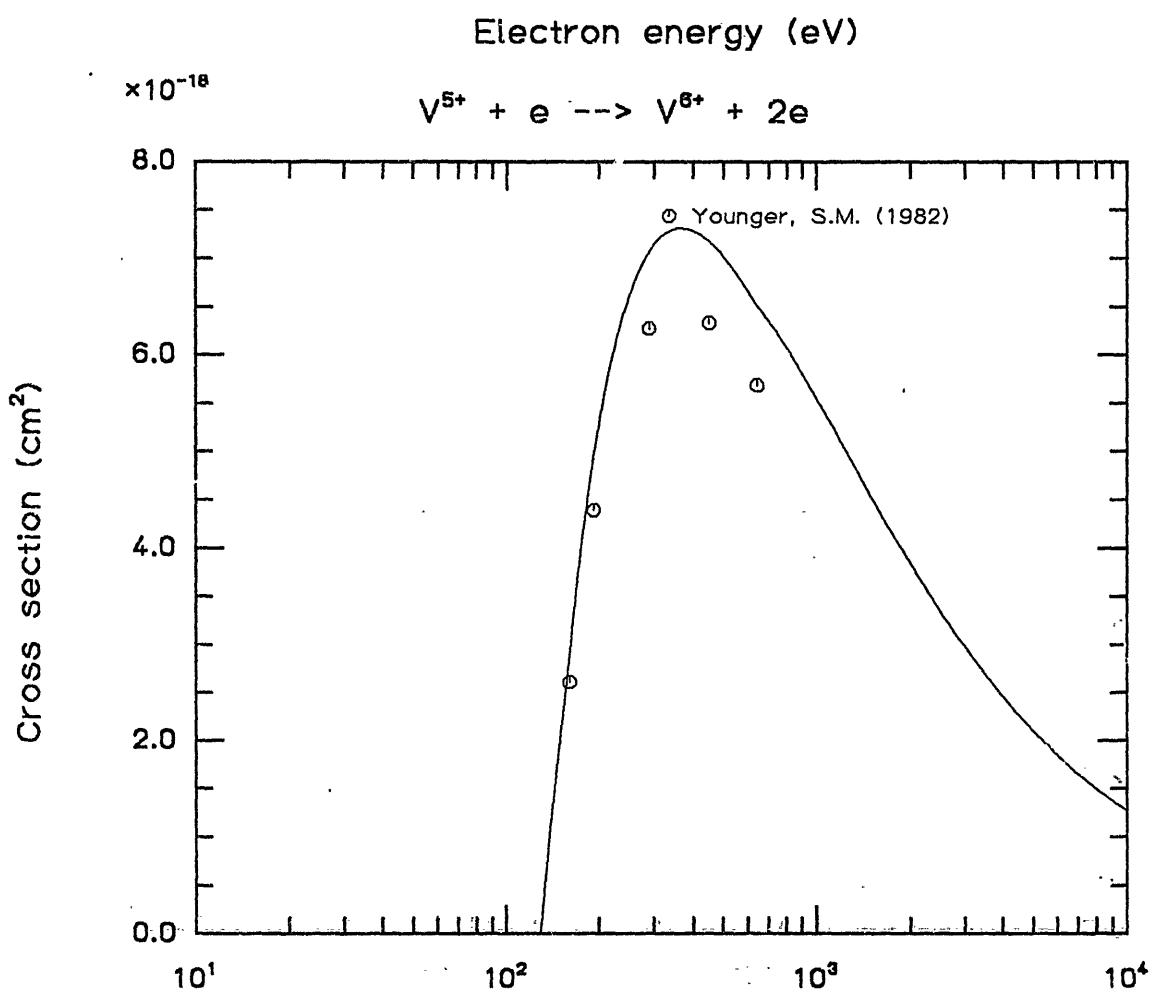
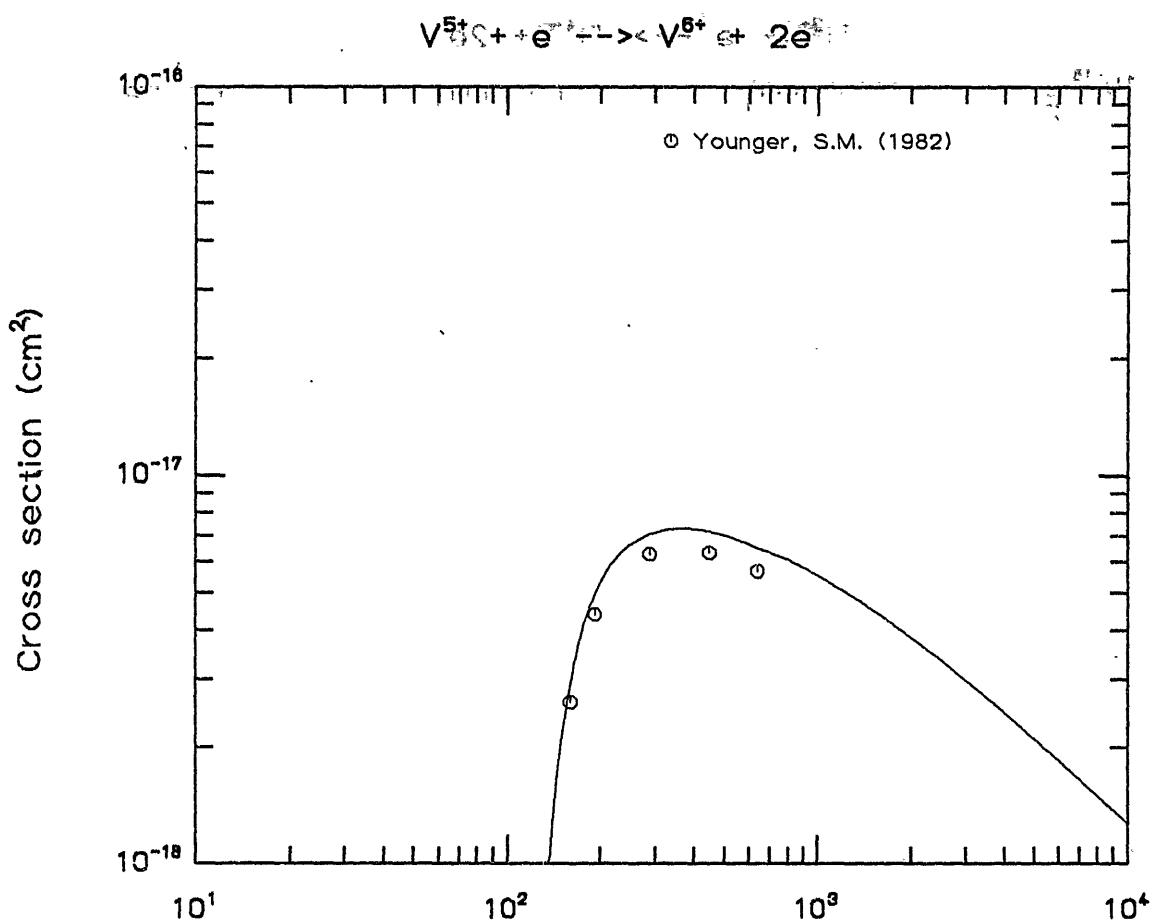
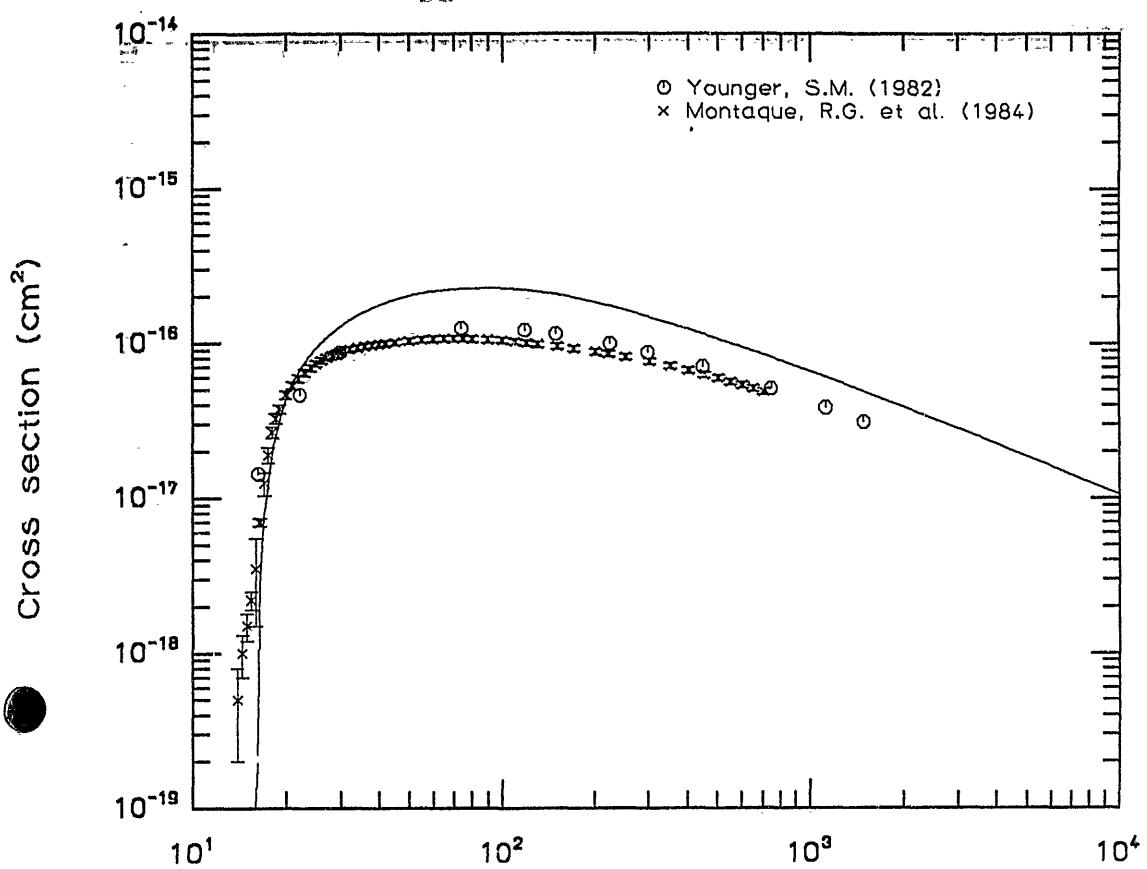
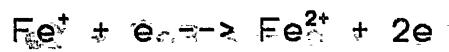


Fig. 150.  $V^{5+} + e^- \rightarrow V^{6+} + 2e^-$



Electron energy (eV)

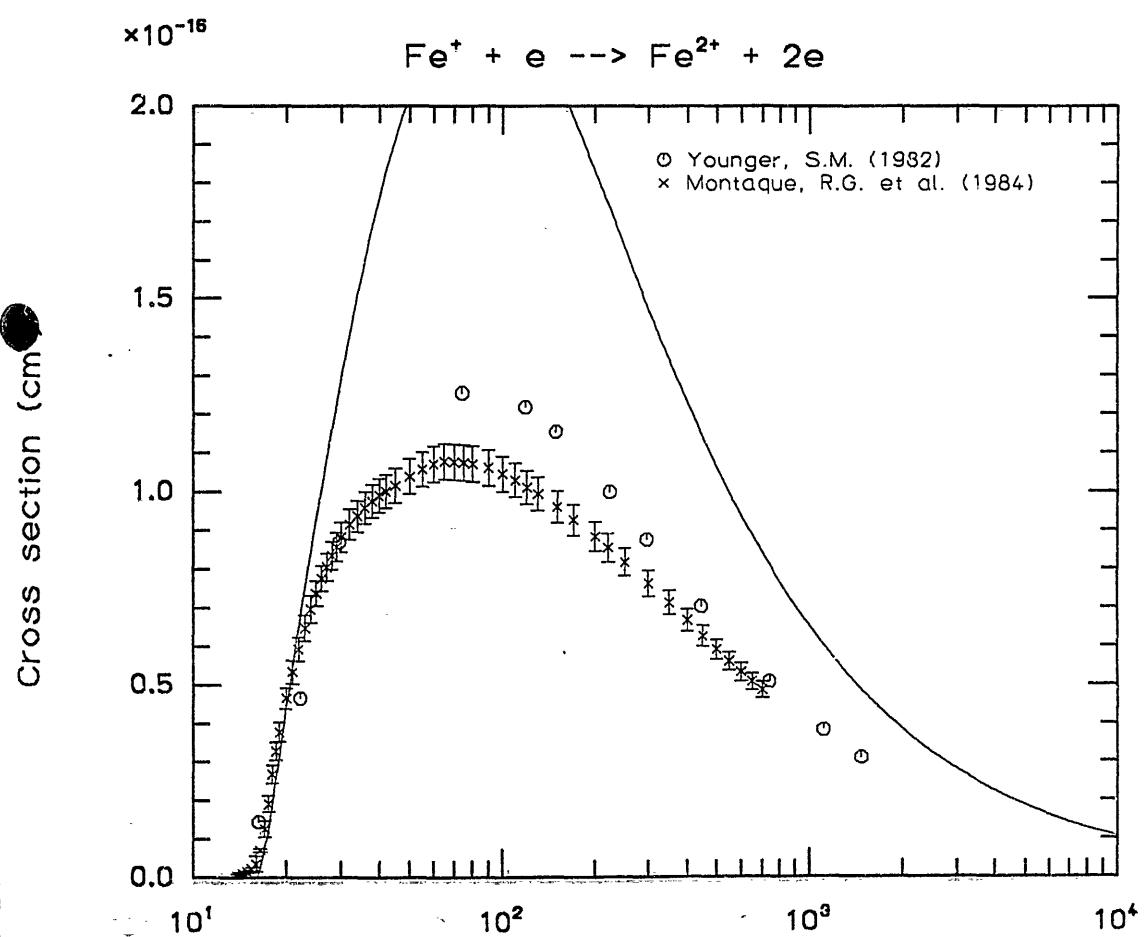


Fig. 151 (a) Electron energy (eV) (b)

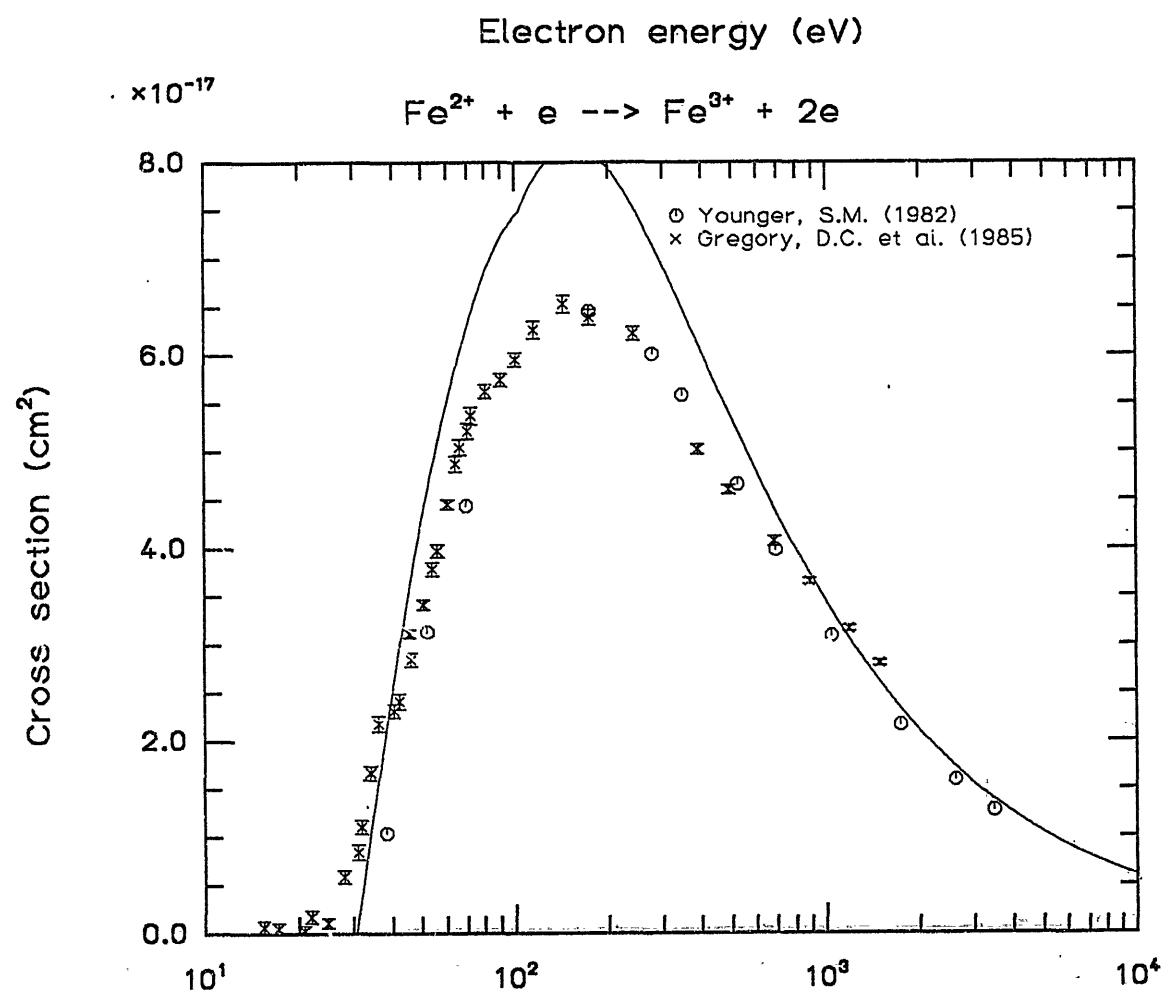
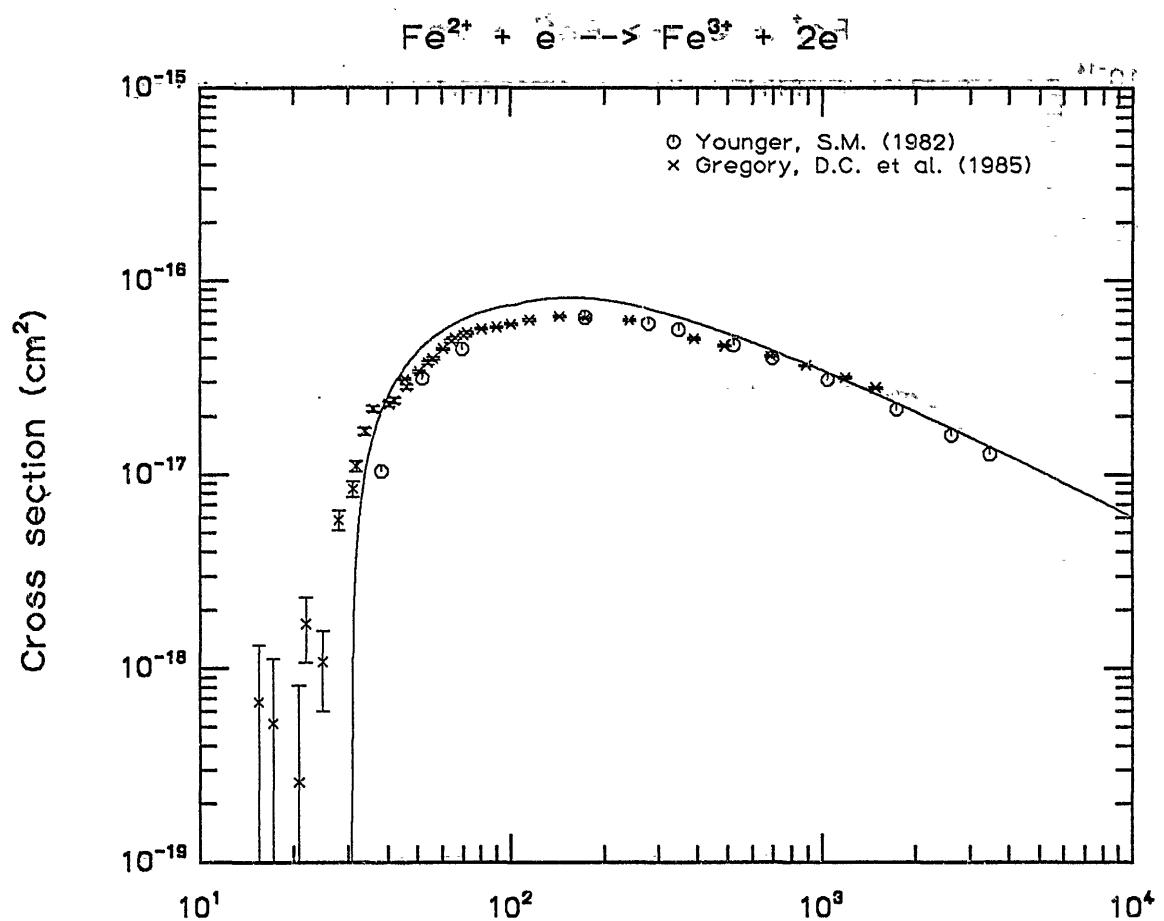
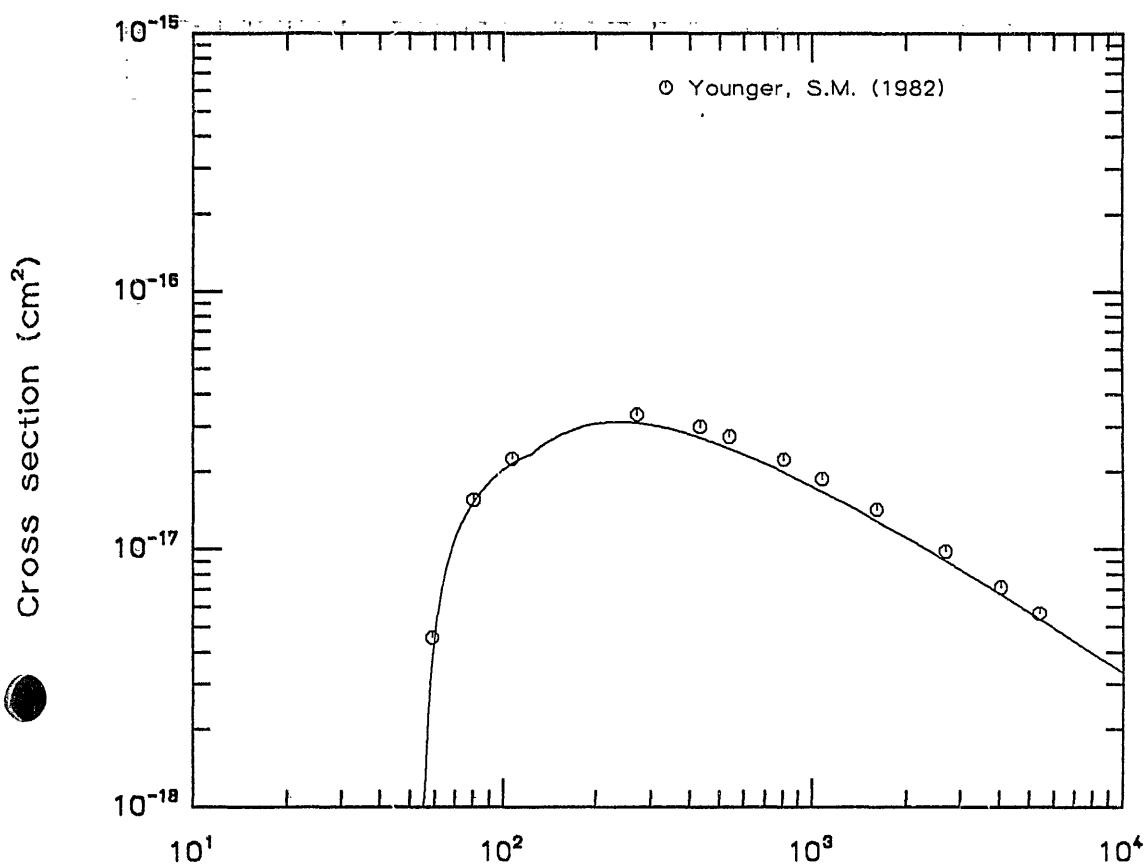
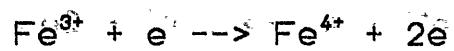


Fig. 152      Electron energy (eV)      151.pdf



Electron energy (eV)

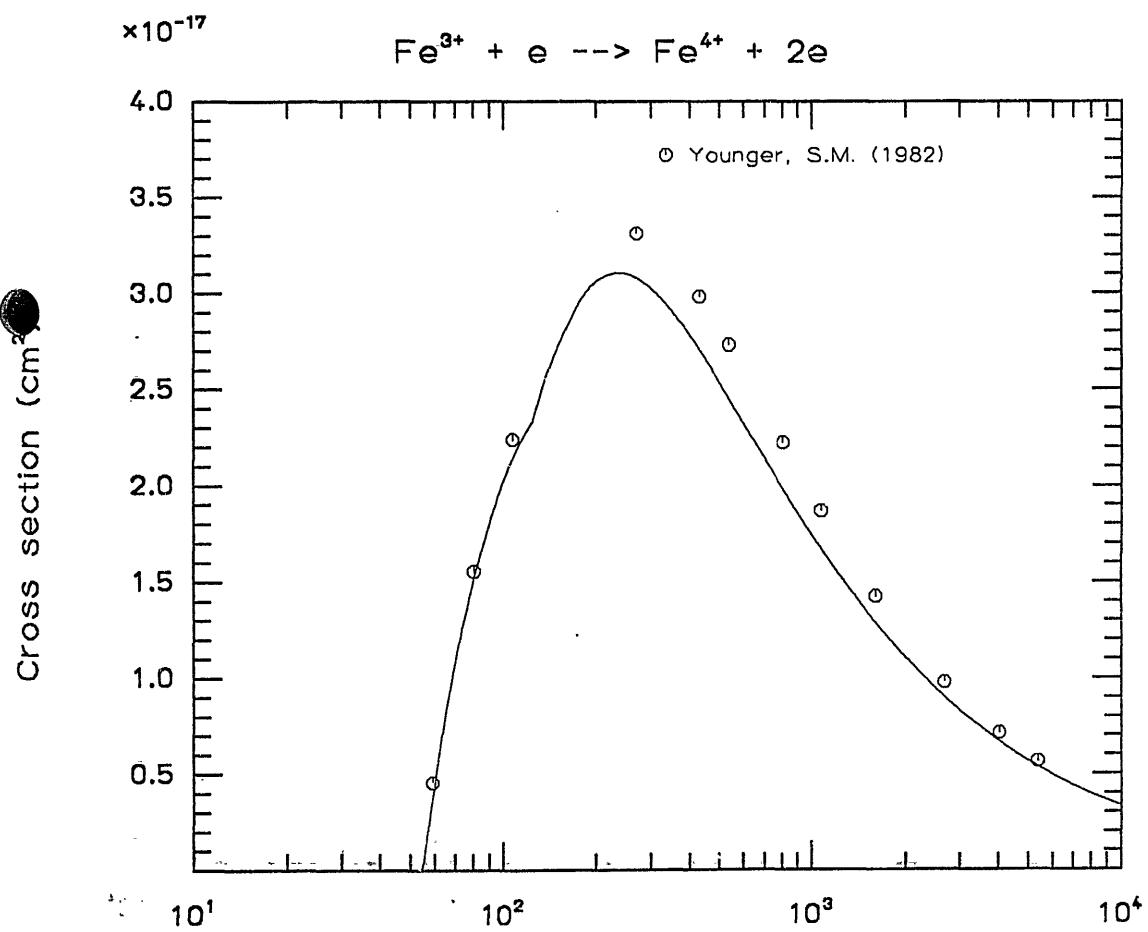


Fig. 153

Electron energy (eV)

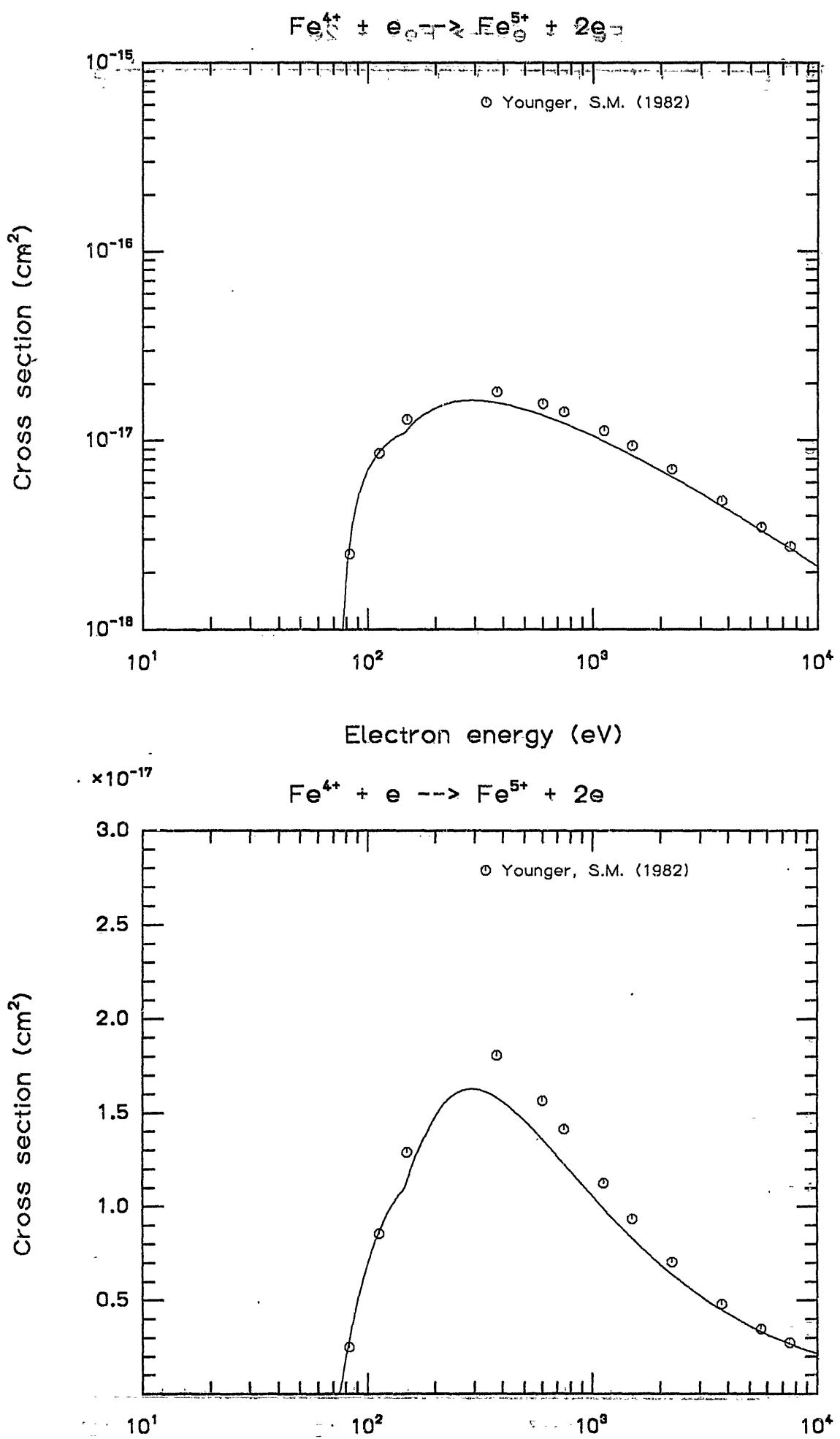
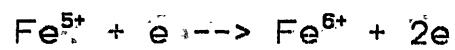
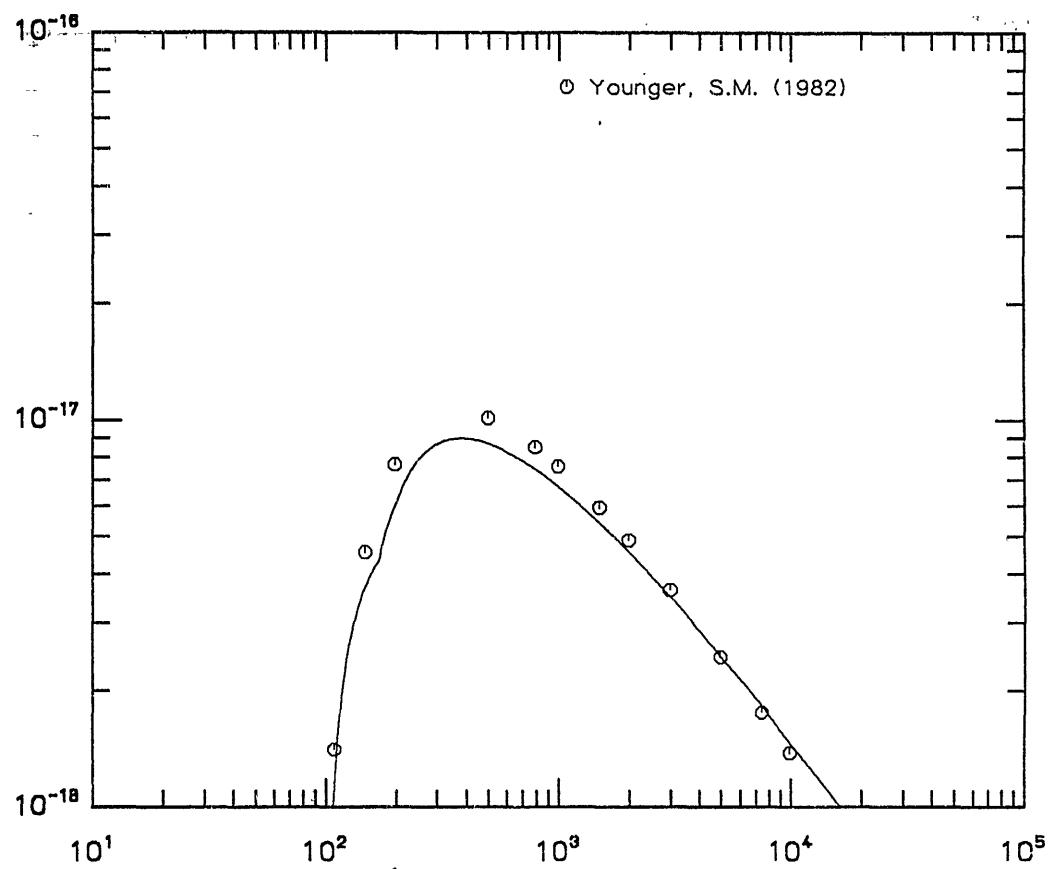


Fig. 154 (v) Electron energy (eV) ESR data



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

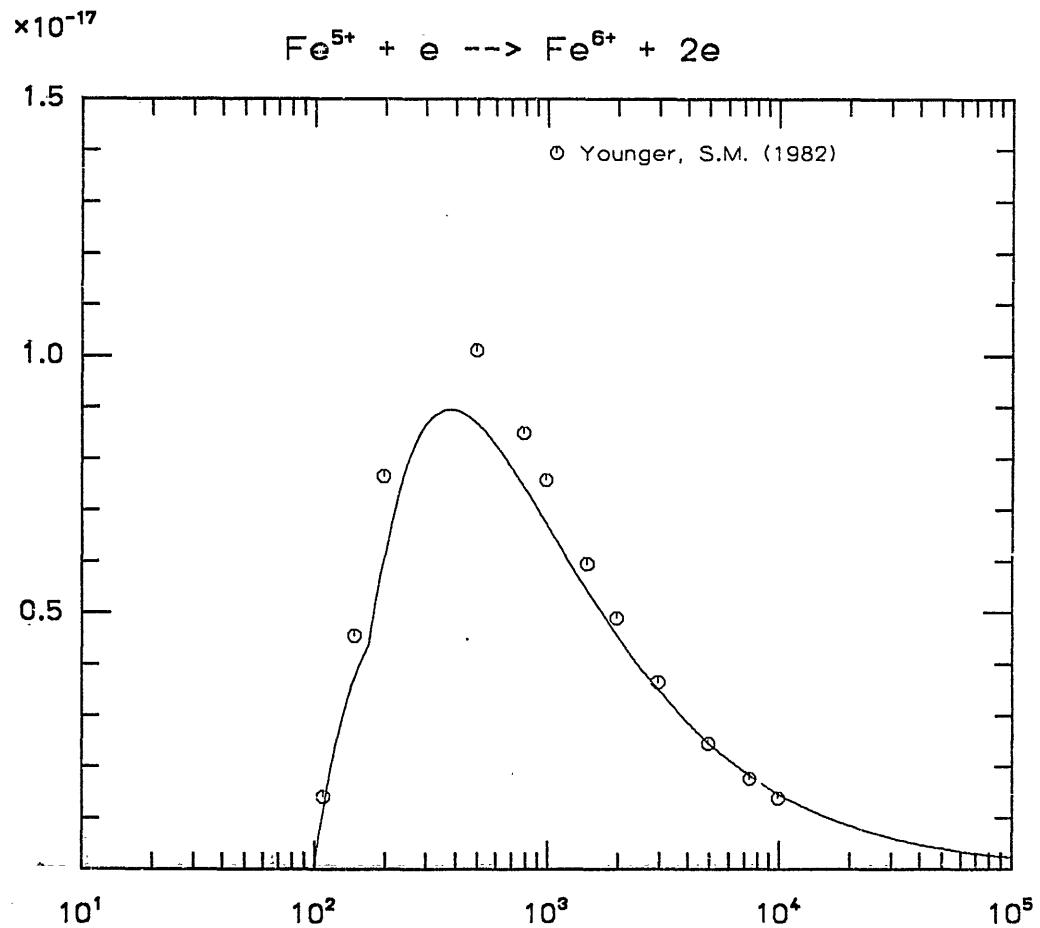


Fig. 155      Electron energy (eV)       $\text{Fe}^{5+} + e^- \rightarrow \text{Fe}^{6+} + 2e^-$

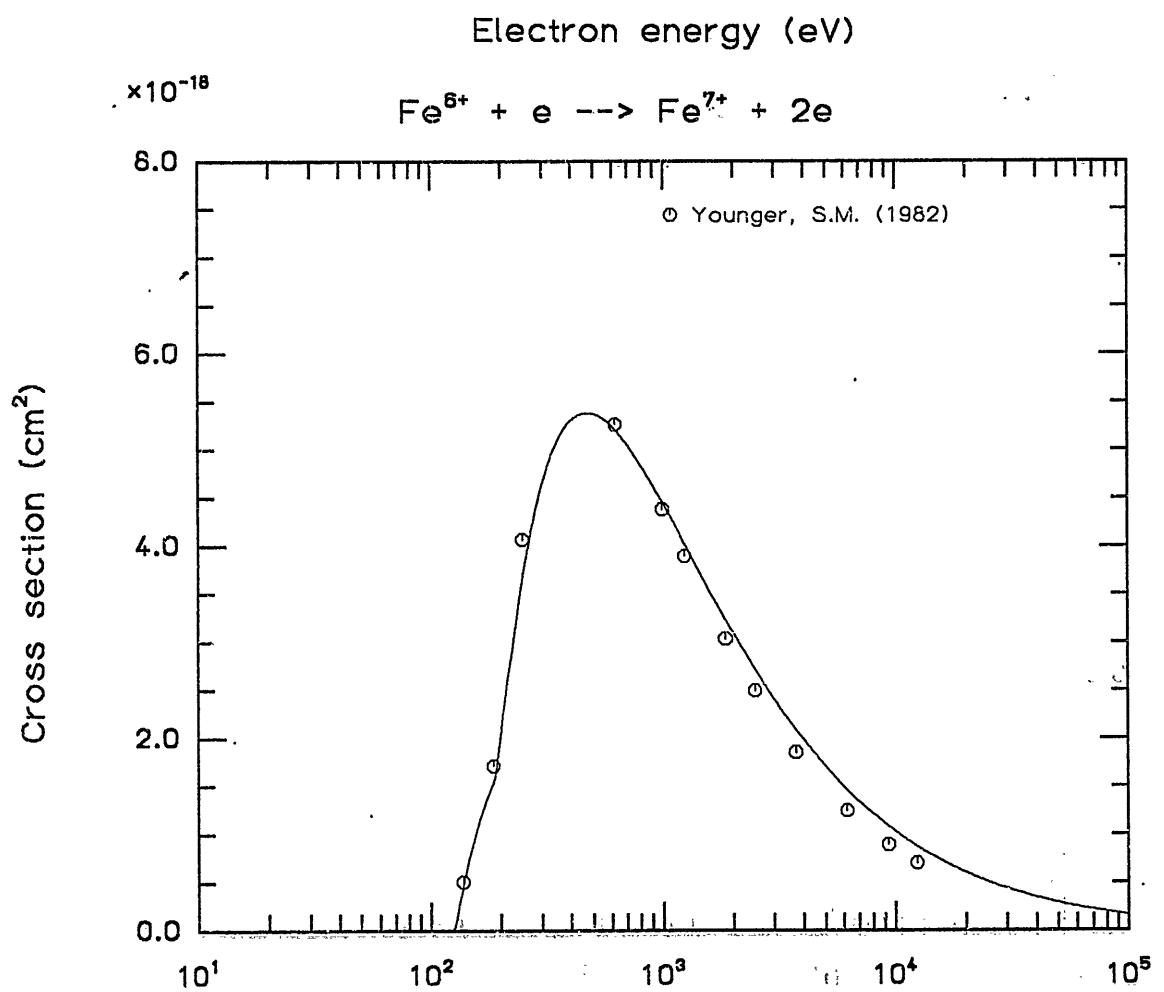
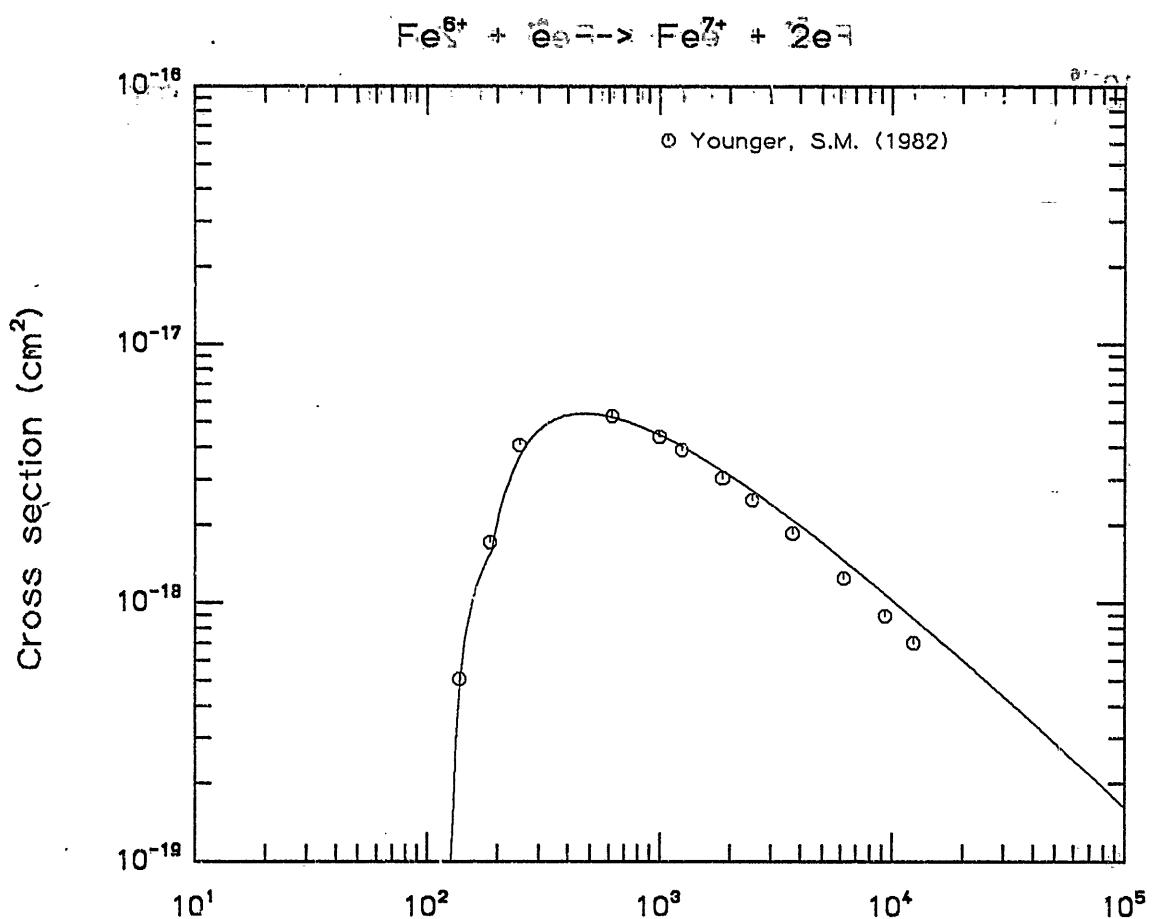
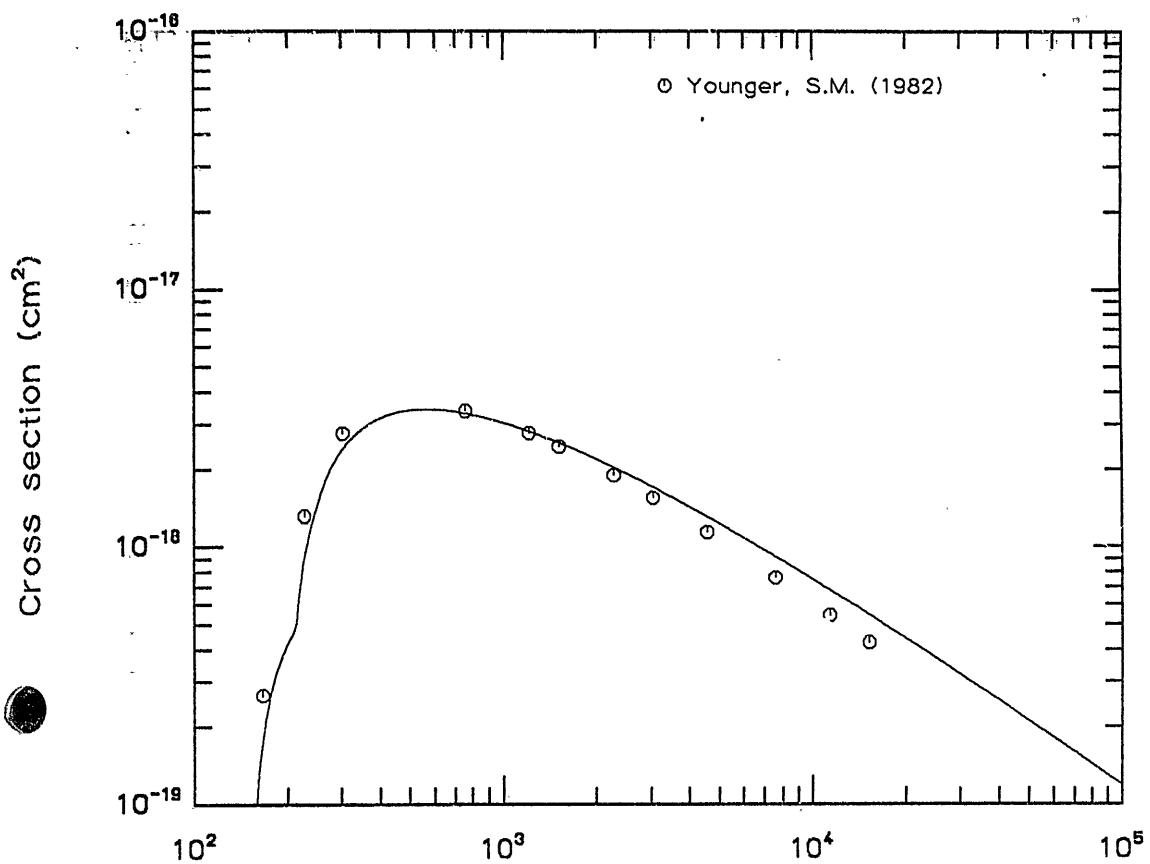
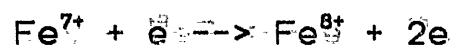


Fig. 156 Electron energy (eV) 331 pF



Electron energy (eV)

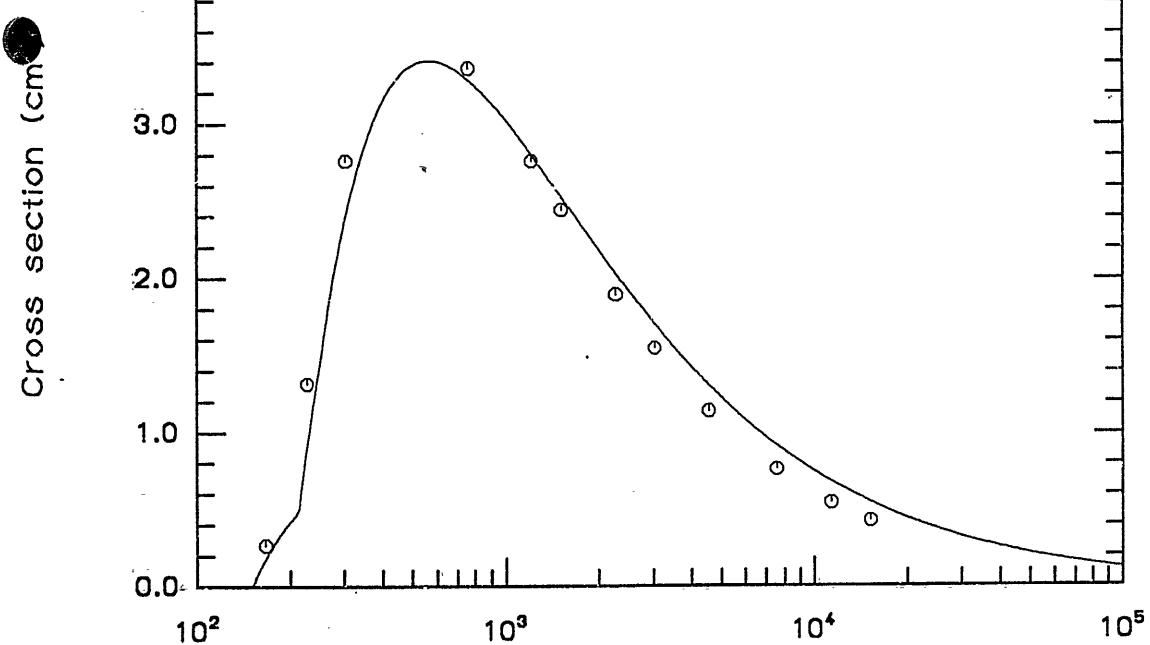
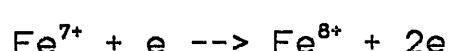
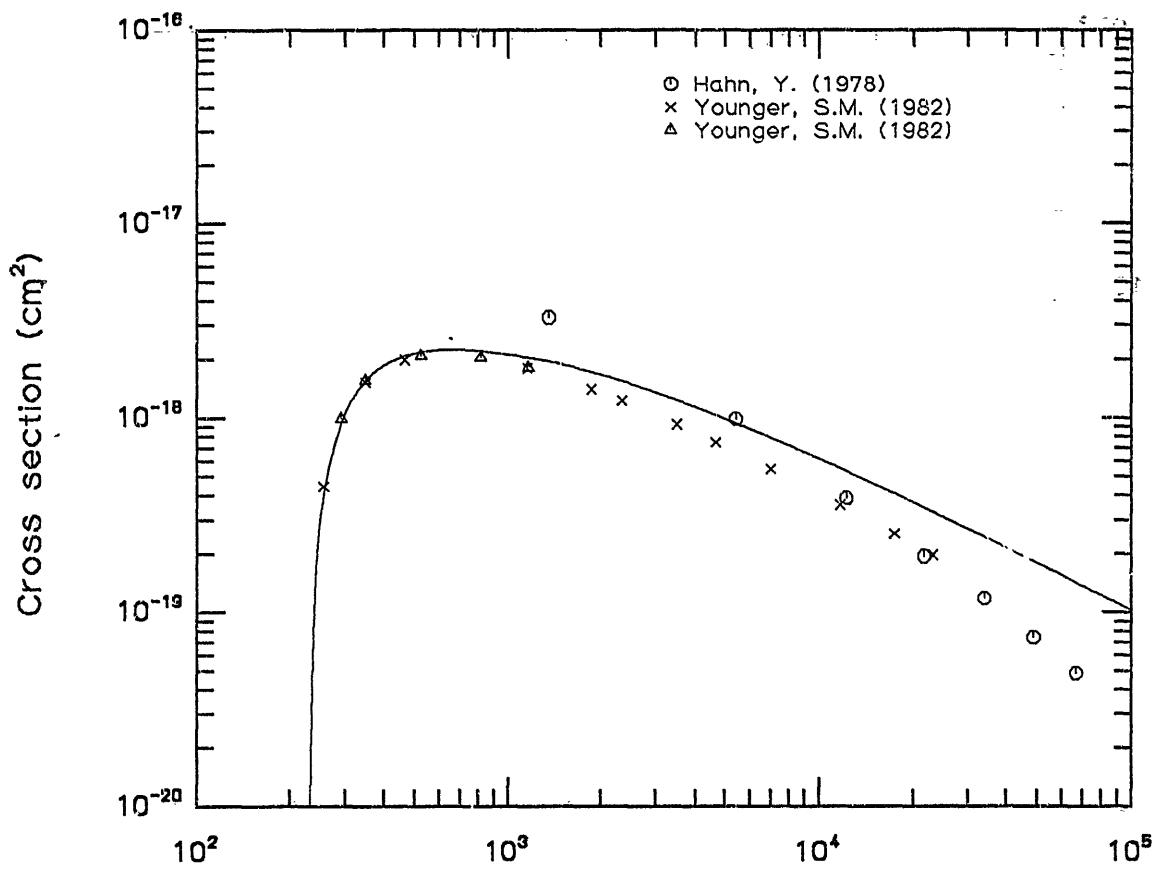
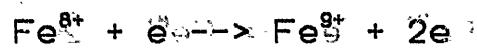


Fig. 157. Electron energy (eV)  $\times 10^{-18}$



Electron energy (eV)

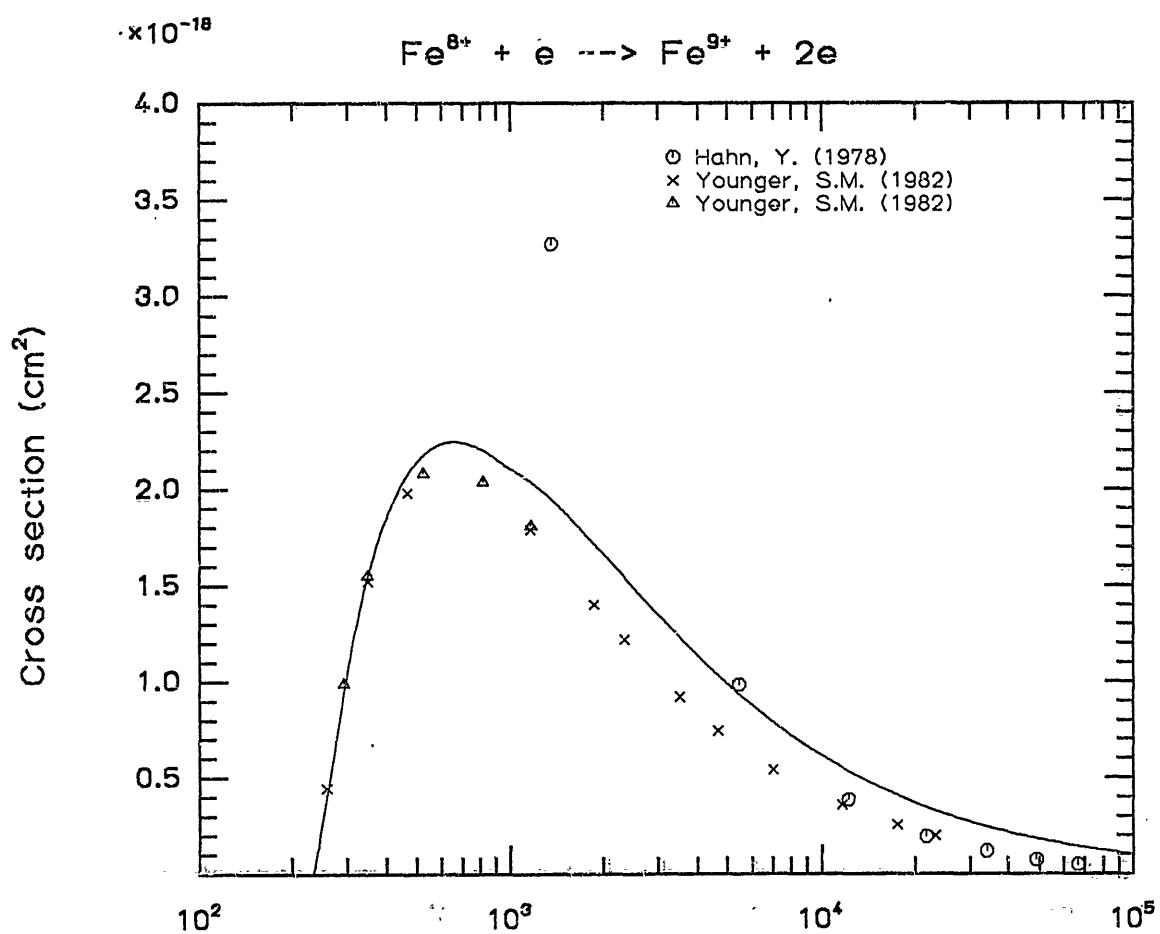
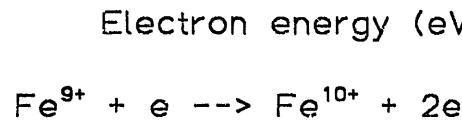
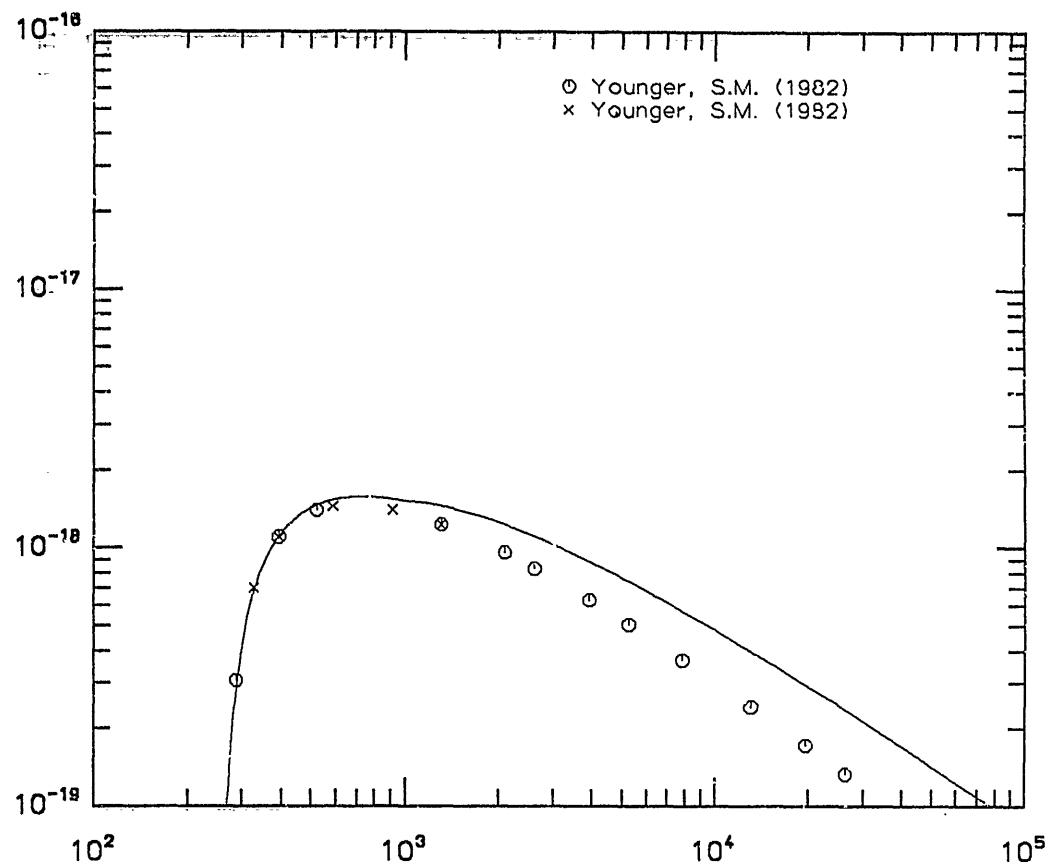


Fig. 158      Electron energy (eV)      CD 1 pF



Cross section ( $\text{cm}^2$ )



Cross section ( $\text{cm}^2$ )

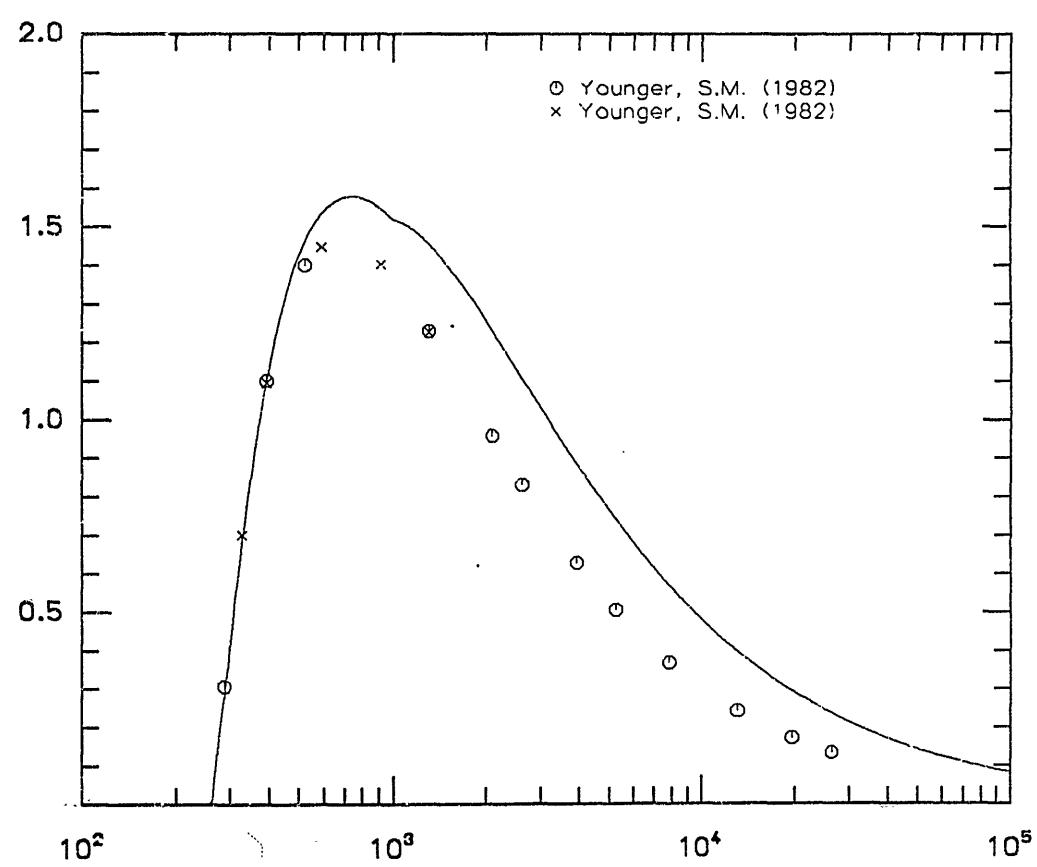


Fig. 159      Electron energy (eV)

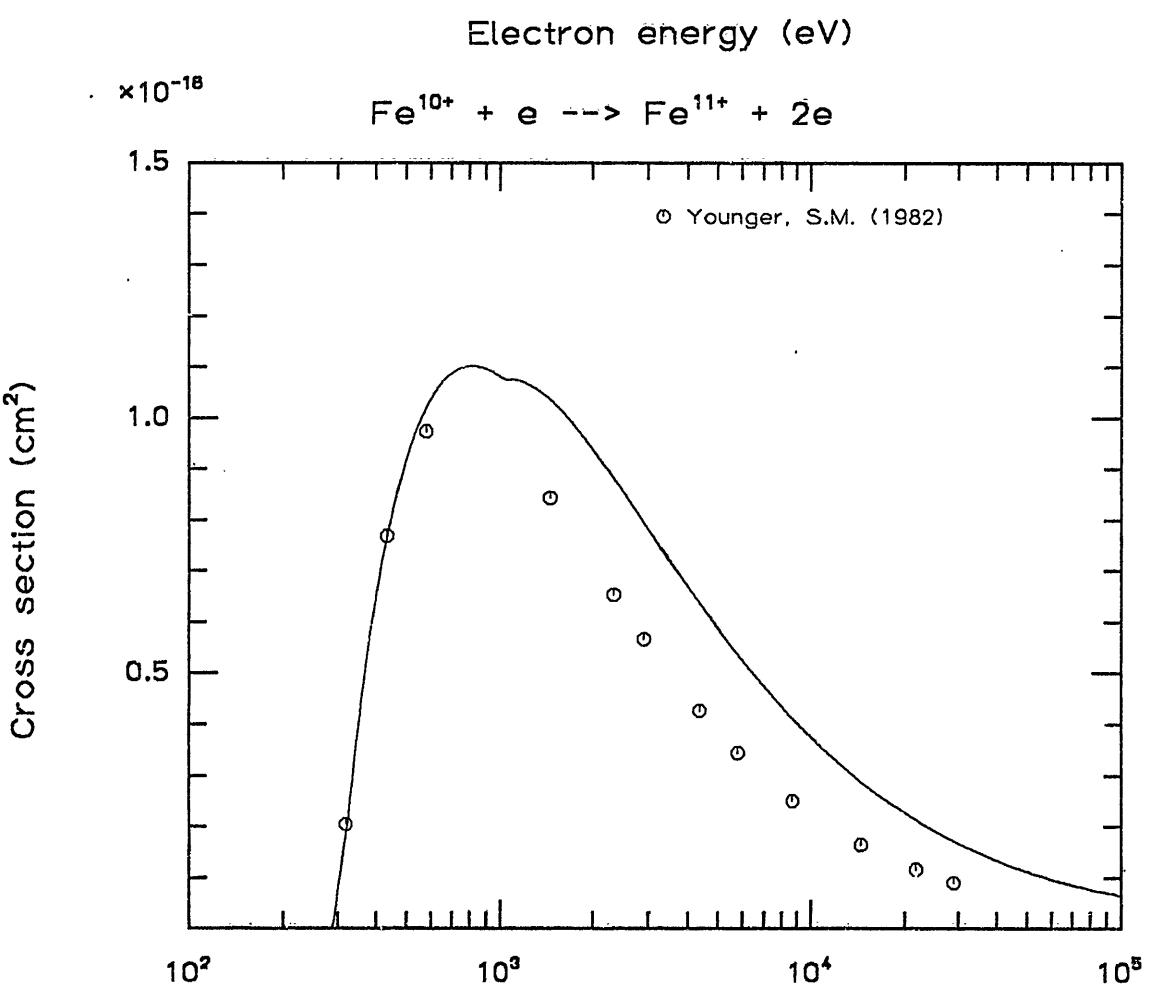
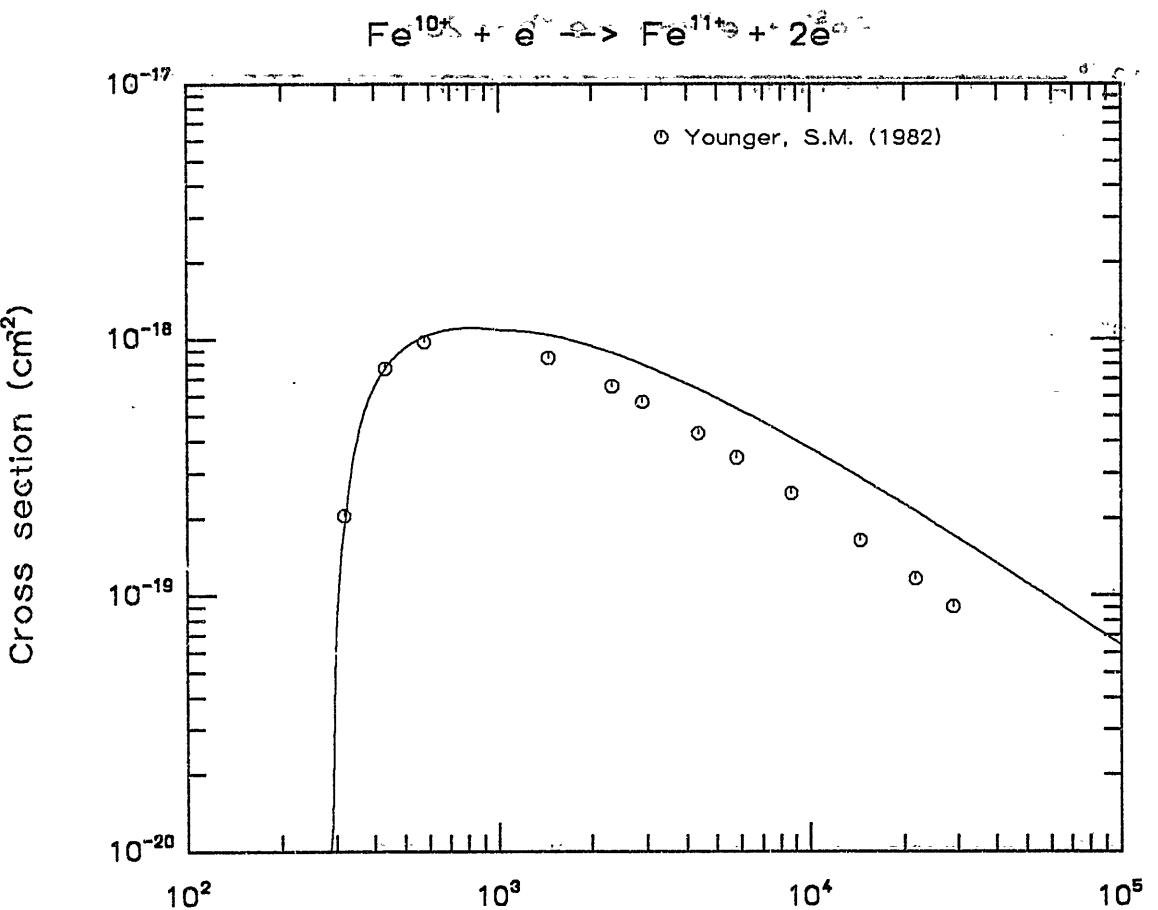
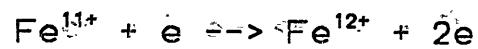
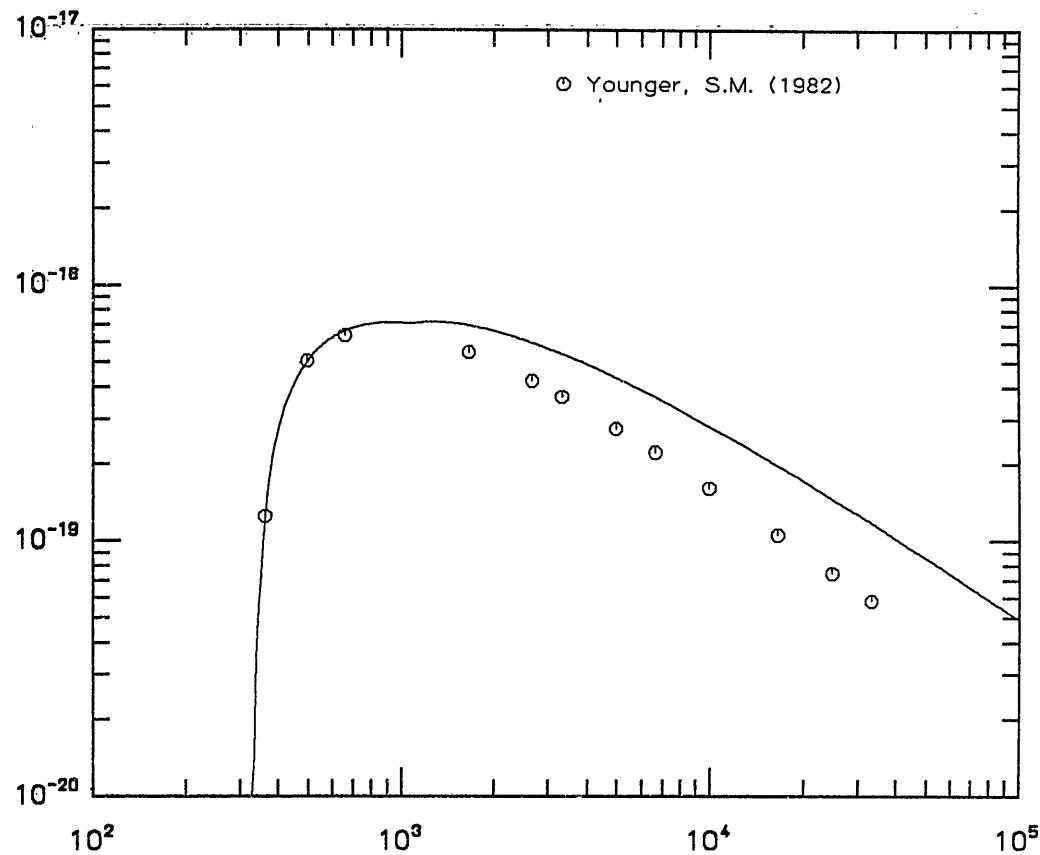


Fig. 160

Electron energy (eV)



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

Cross section ( $\text{cm}^2$ )

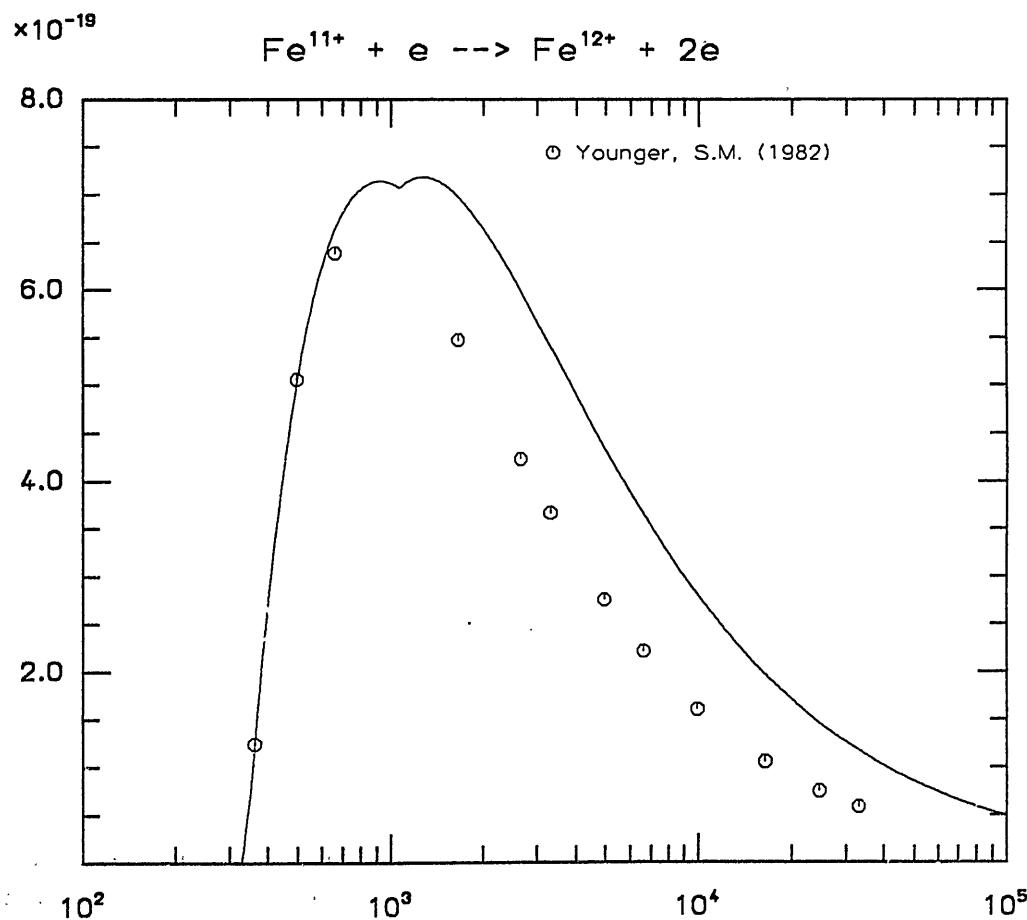


Fig. 161 Electron energy (eV)

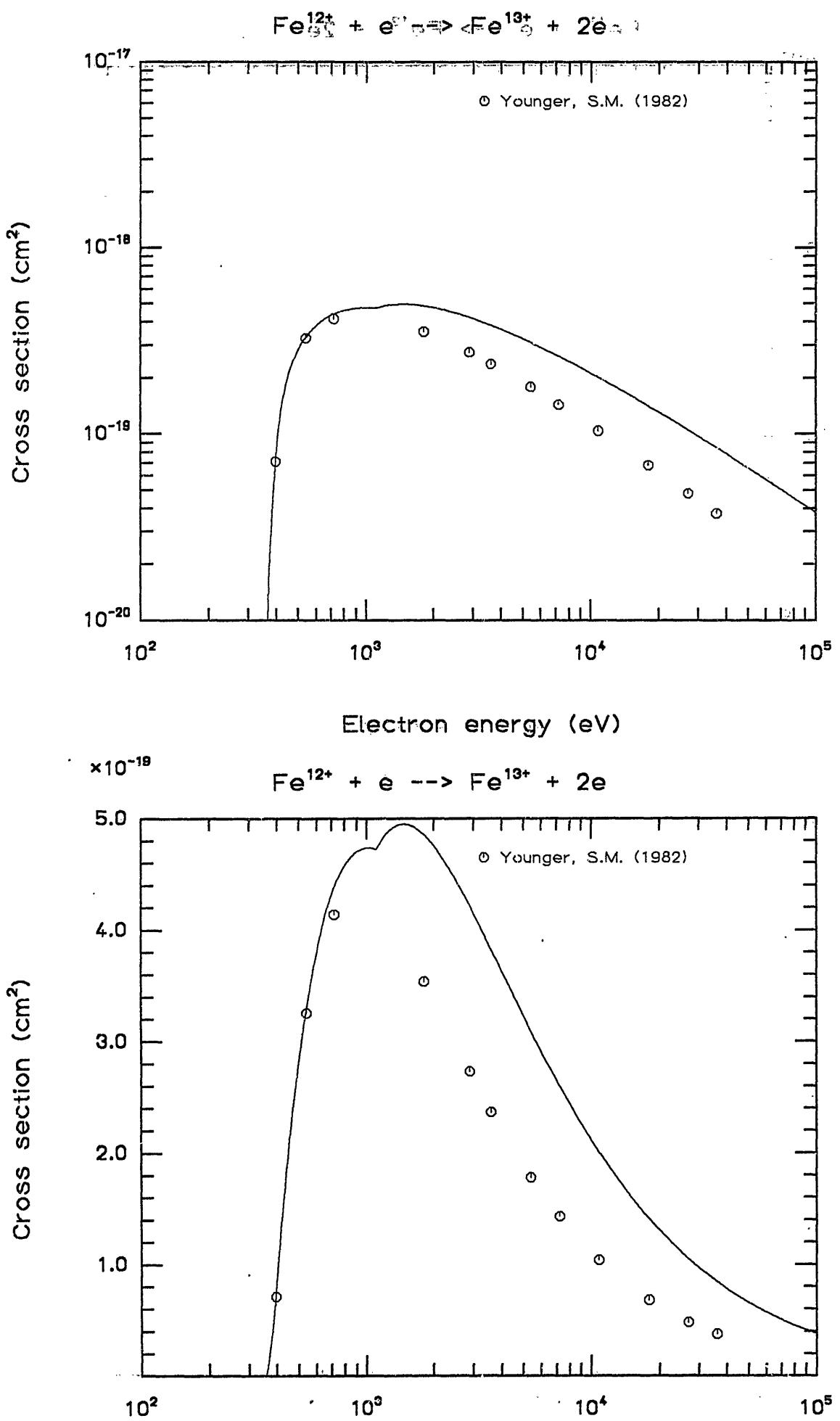
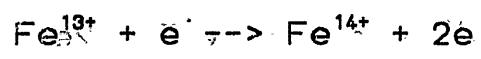
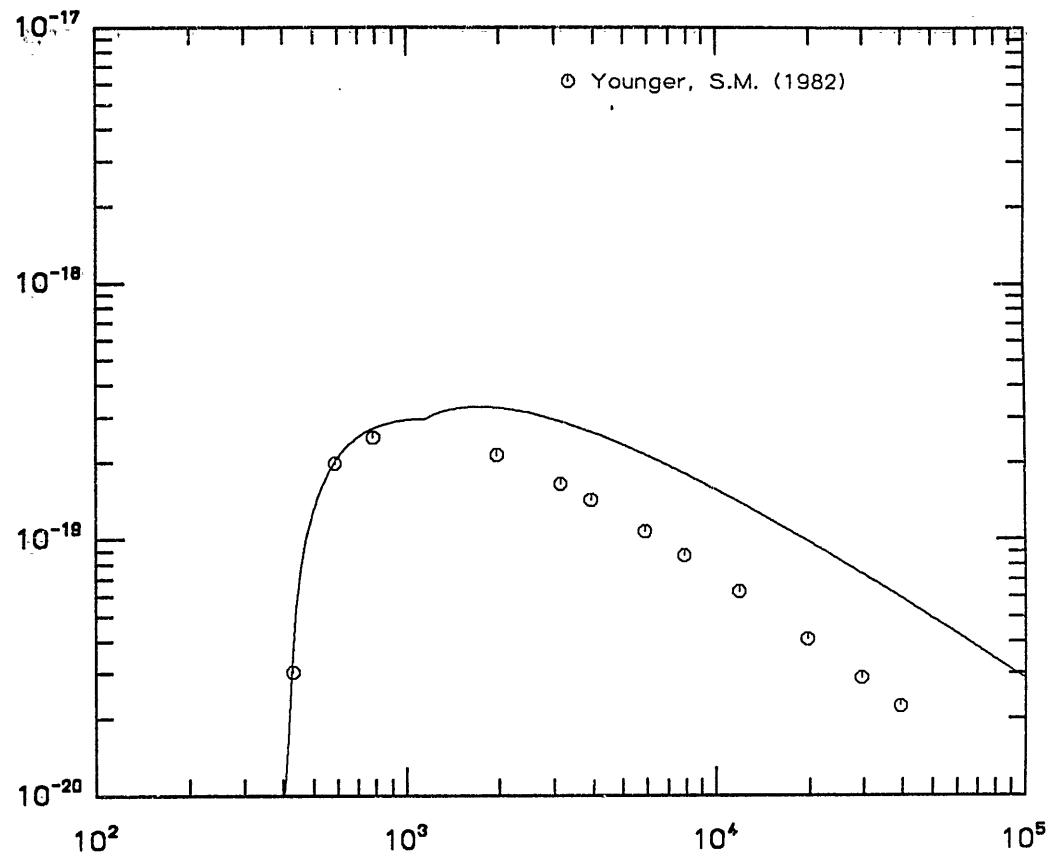


Fig. 162 Cross section vs Electron energy (eV) [31].gR



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

Cross section ( $\text{cm}^2$ )

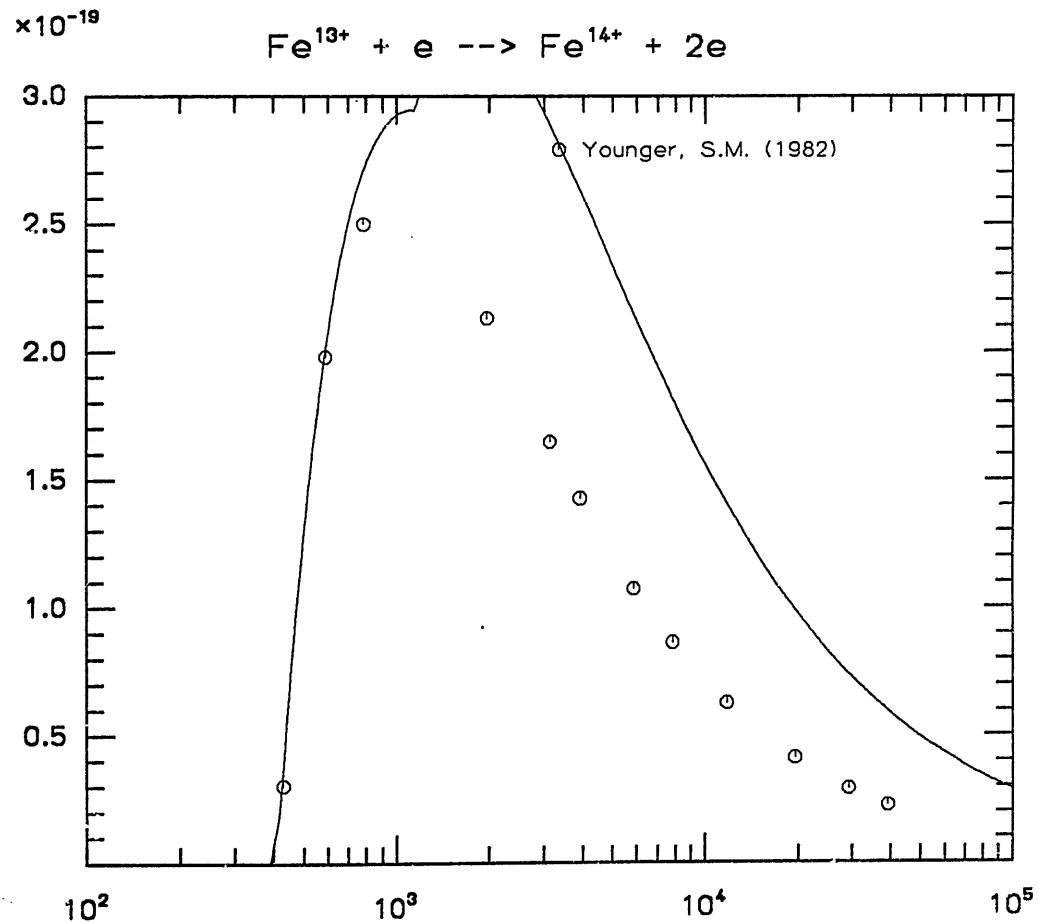


Fig. 163. Electron energy (eV)

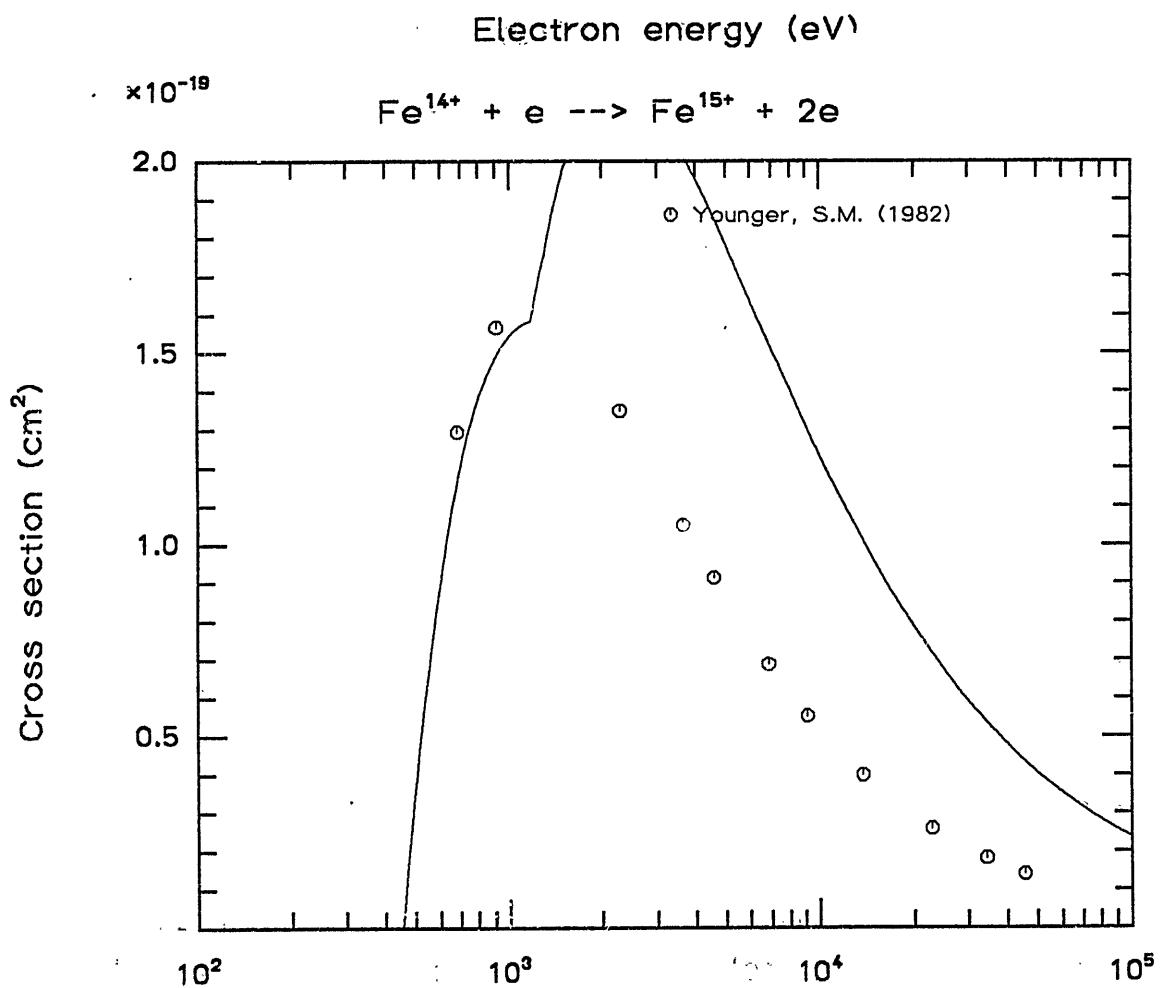
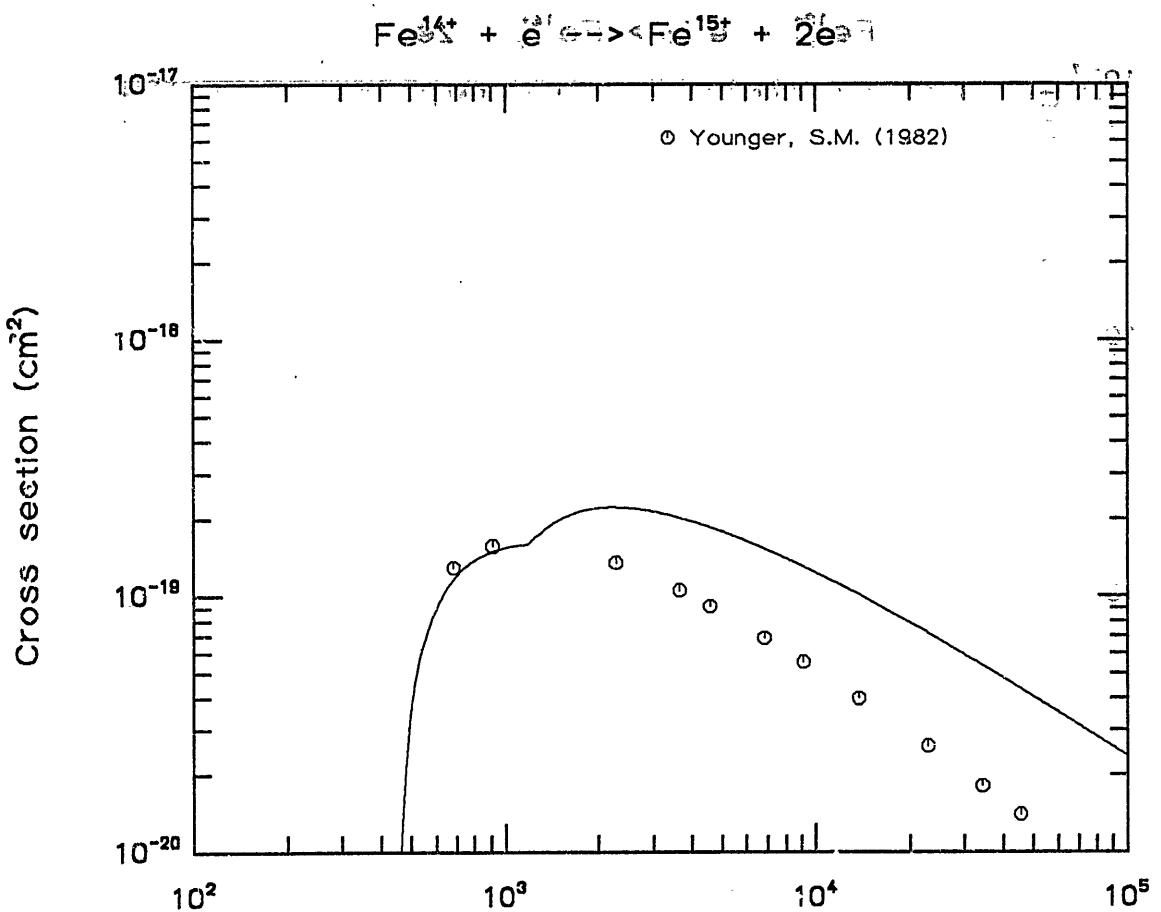
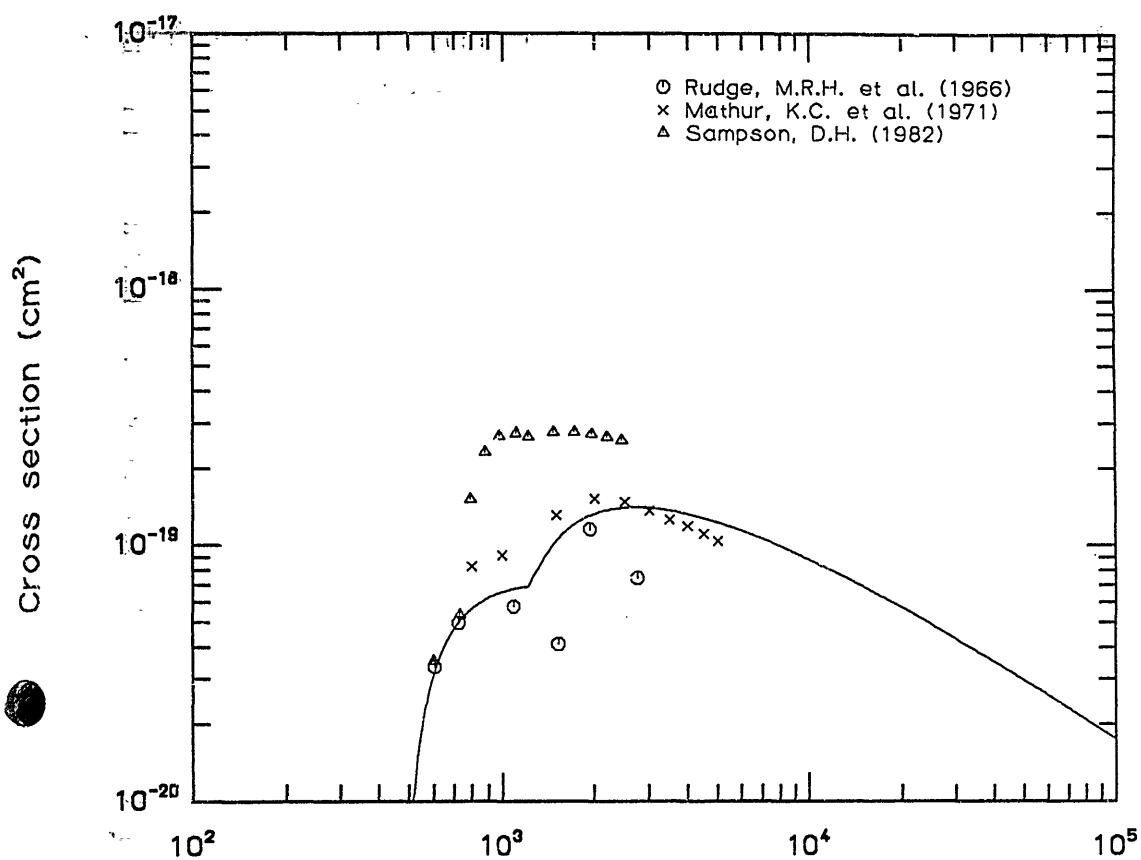
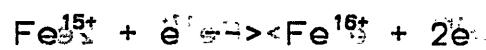


Fig. 164. Cross section vs. Electron energy (eV) for  $\text{Fe}^{14+}$



Electron energy (eV)

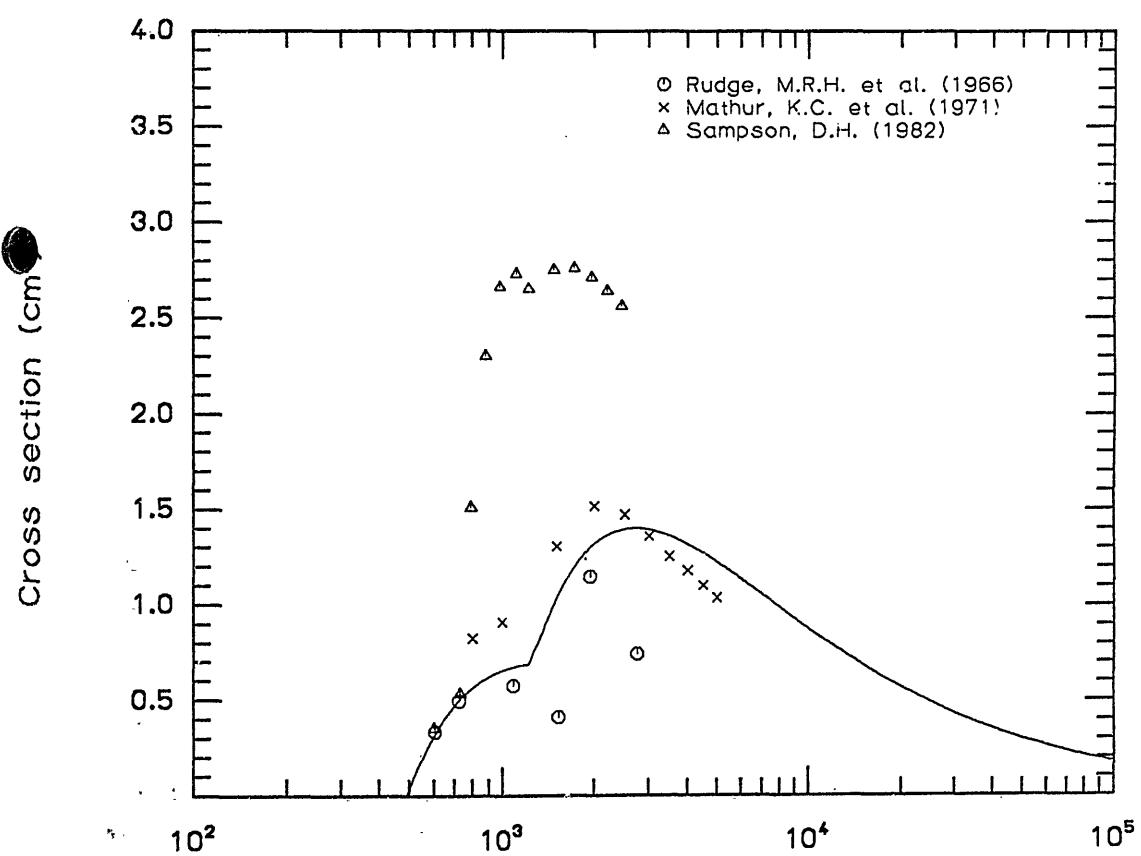
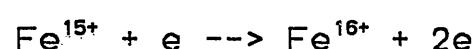


Fig. 165

Electron energy (eV)

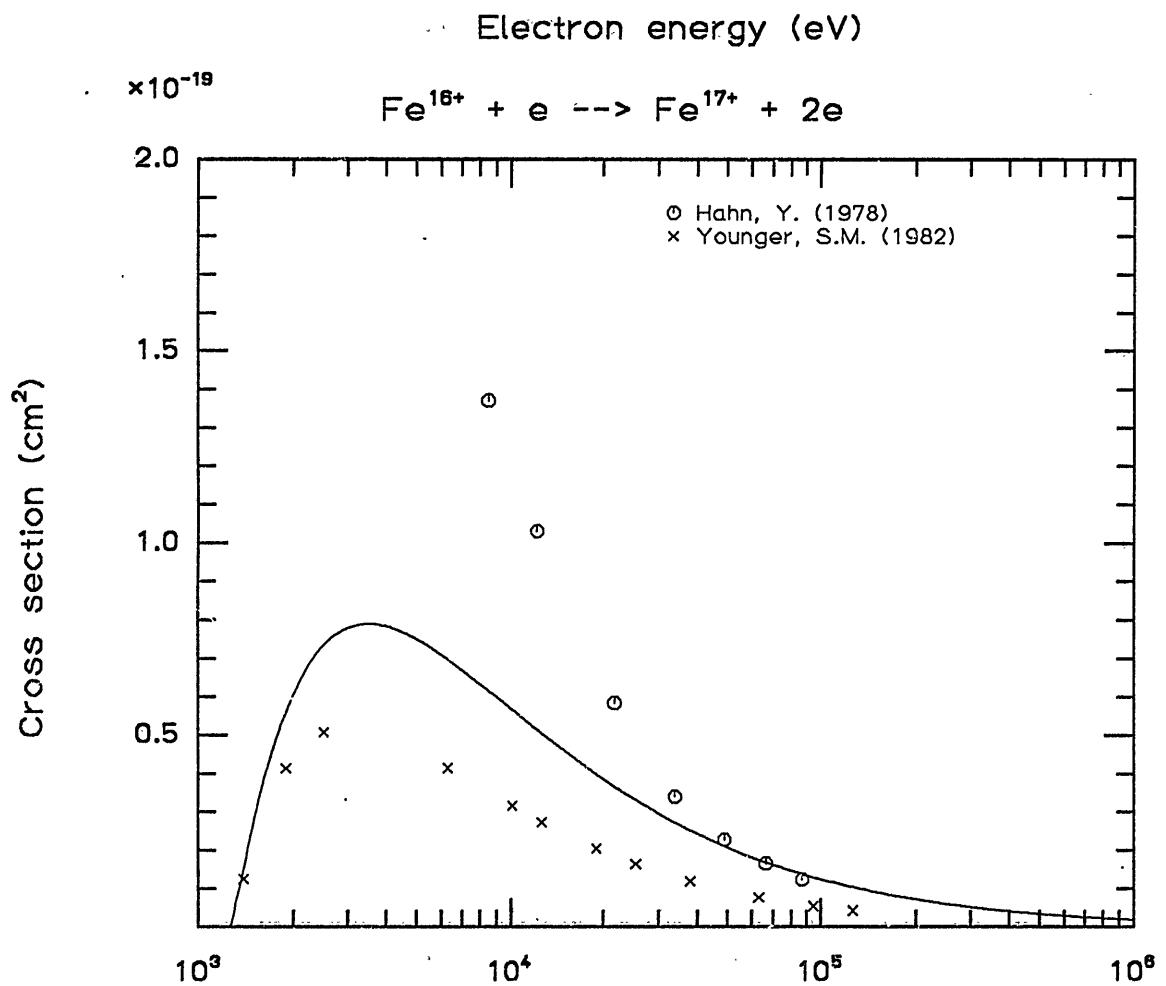
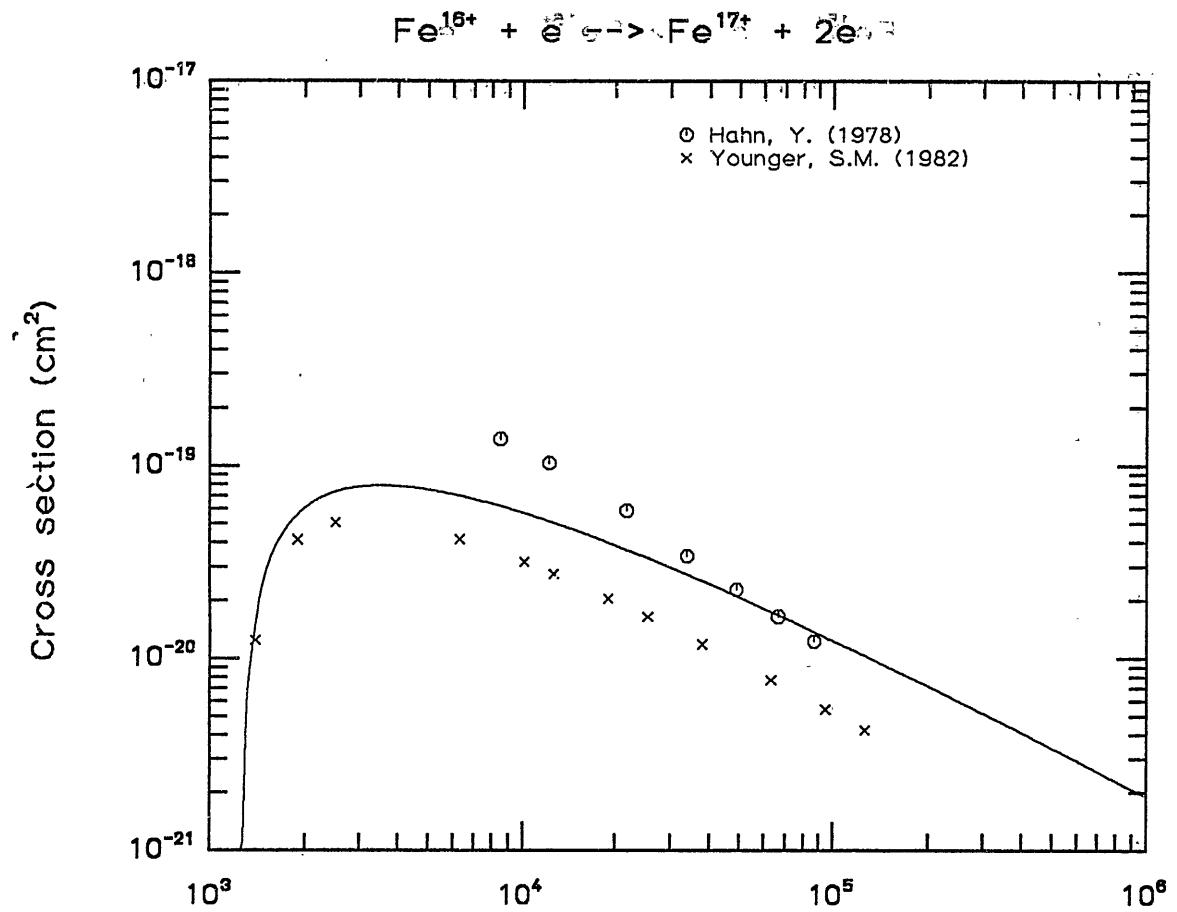
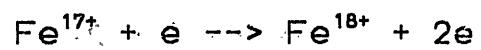
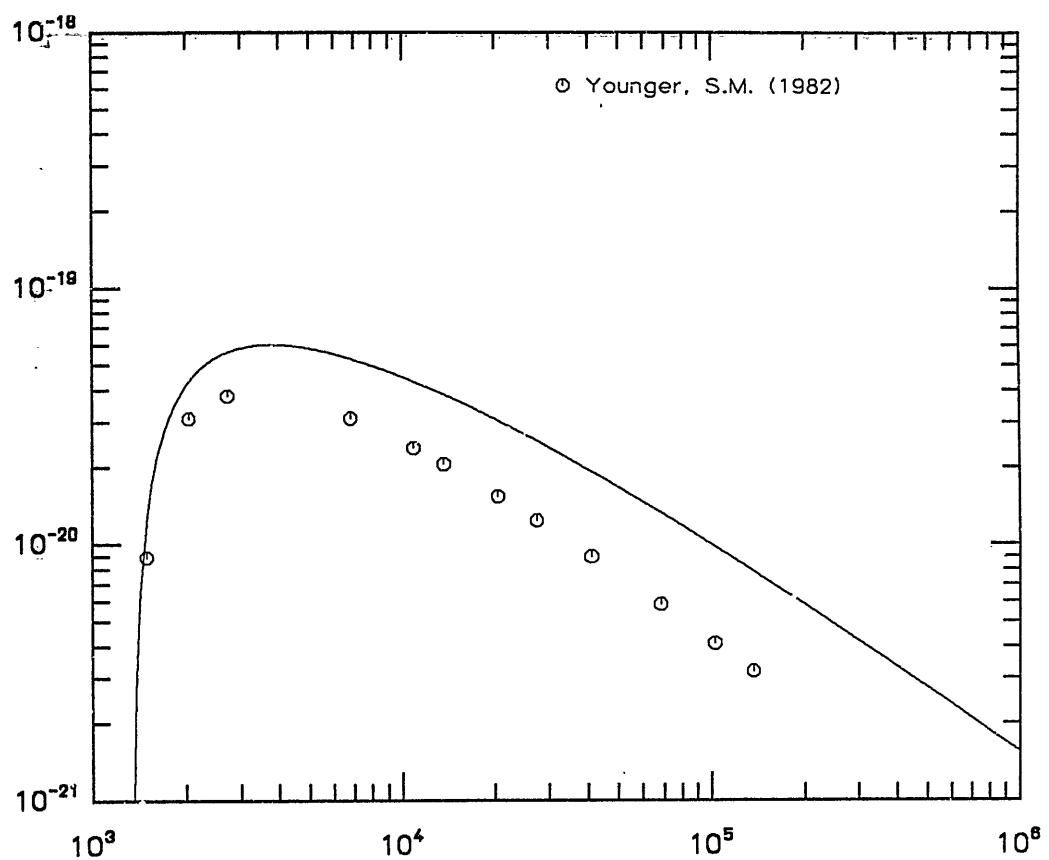


Fig. 166      Electron energy (eV)       $\text{Fe}^{16+} + e^- \rightarrow \text{Fe}^{17+} + 2e^-$



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

Cross section ( $\text{cm}^2$ )

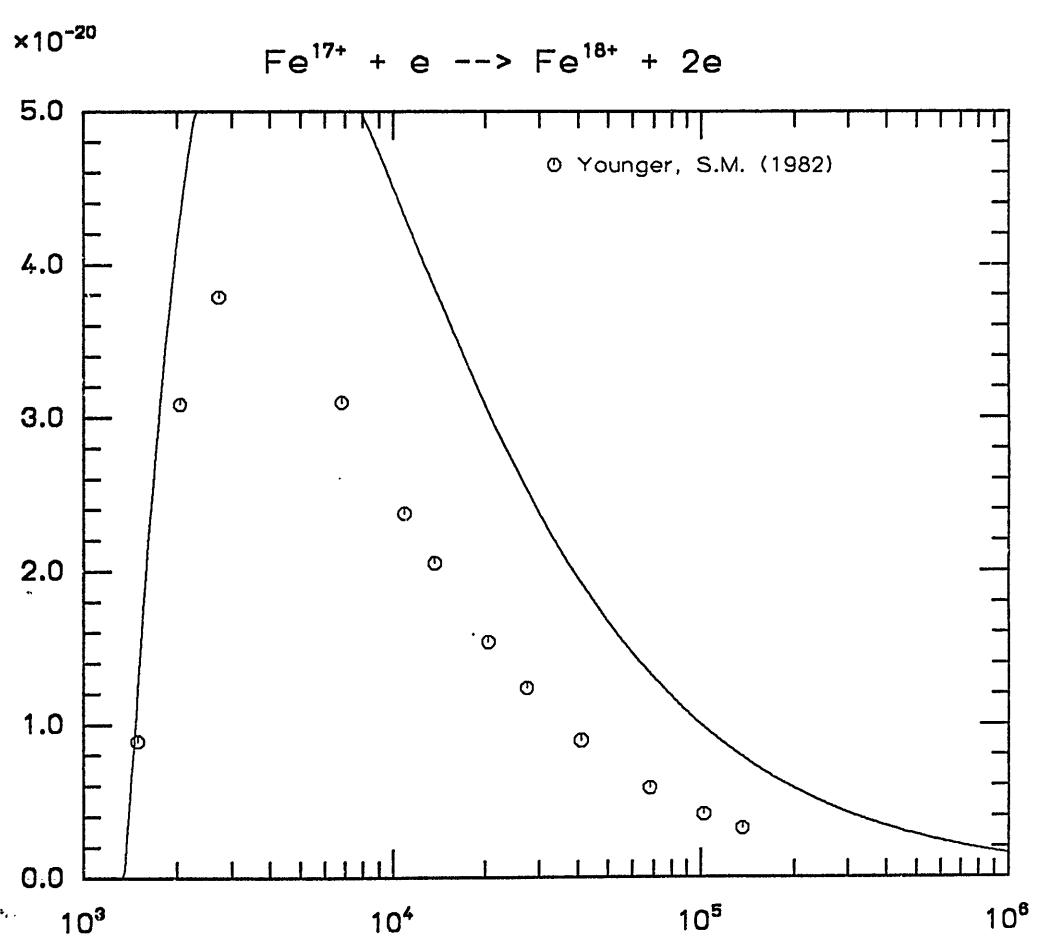


Fig. 167      Electron energy (eV)

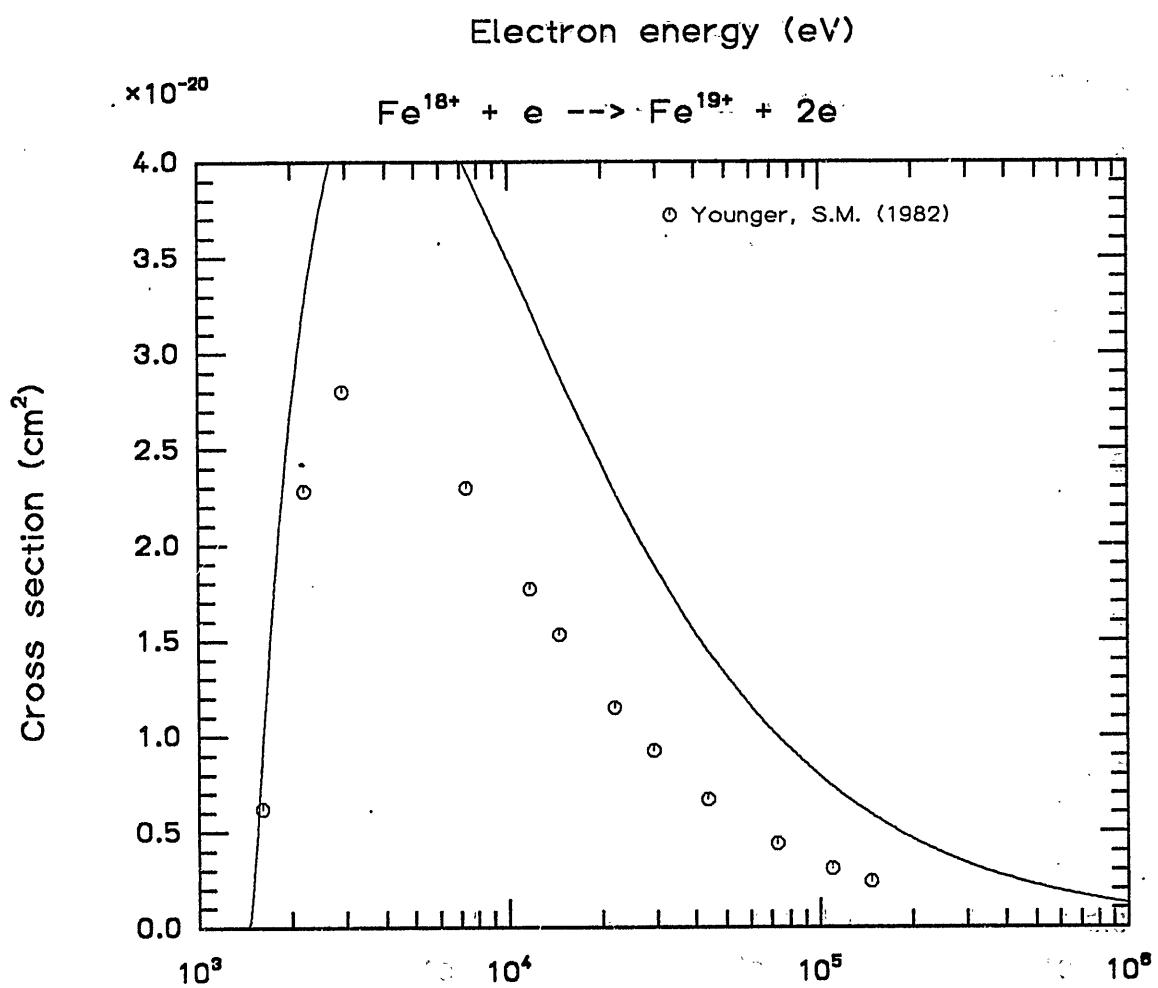
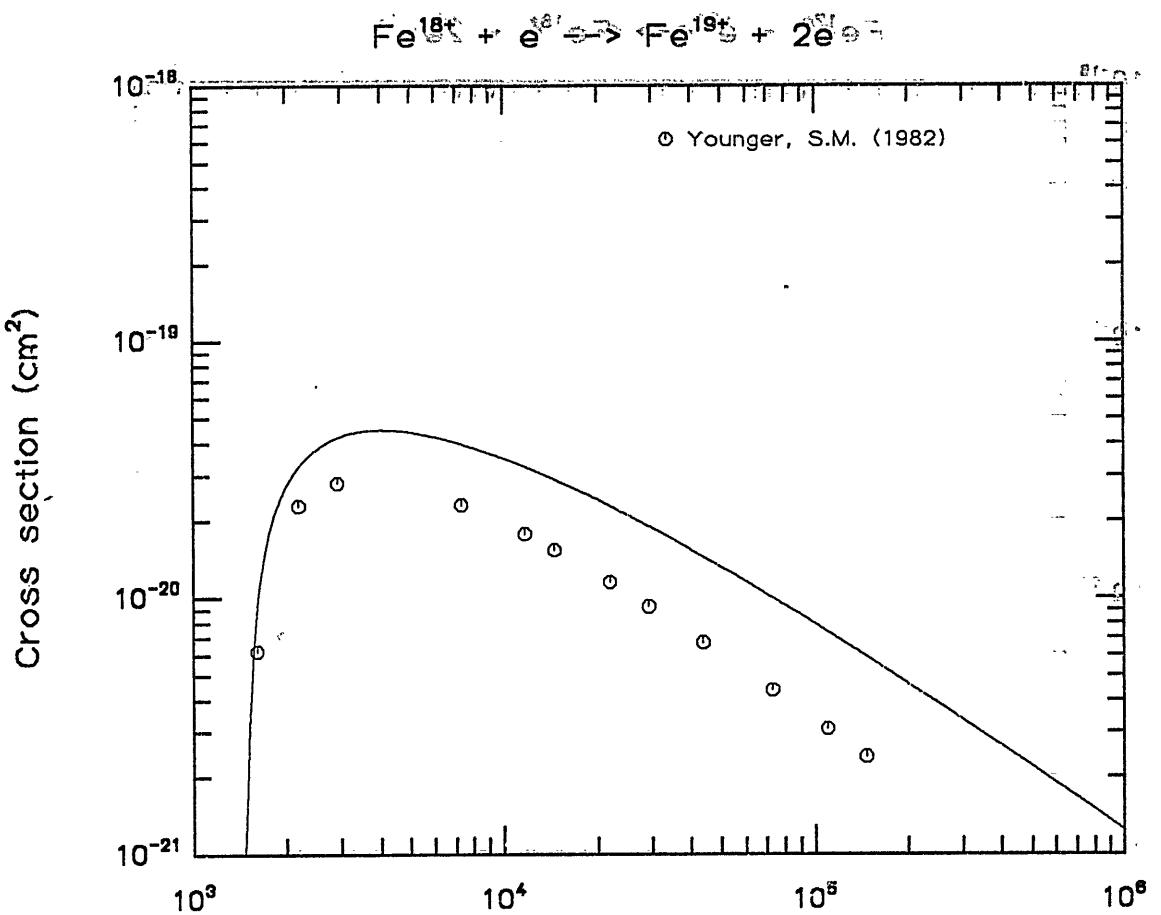
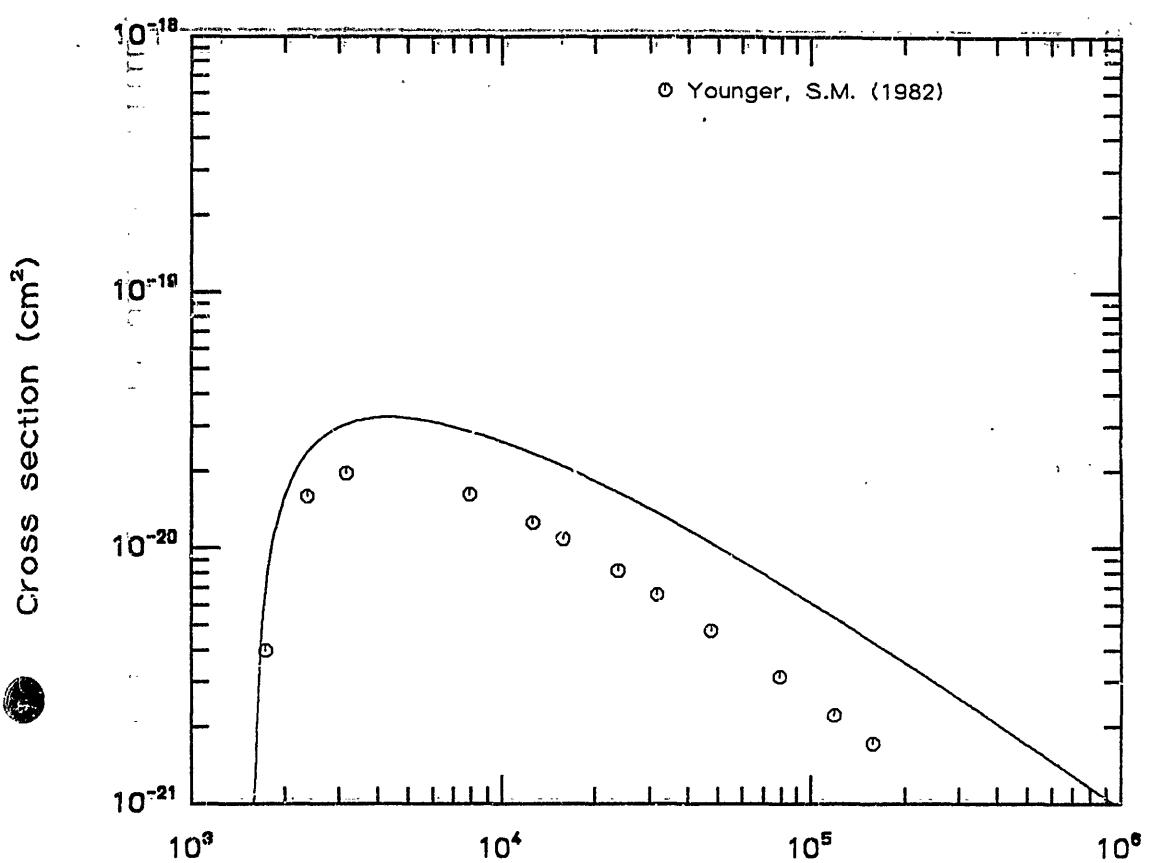
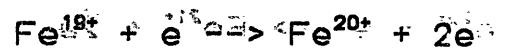


Fig. 168 (top) Electron energy (eV) for  $\text{Fe}^{18+} + e^- \rightarrow \text{Fe}^{19+} + 2e^-$



Electron energy (eV)

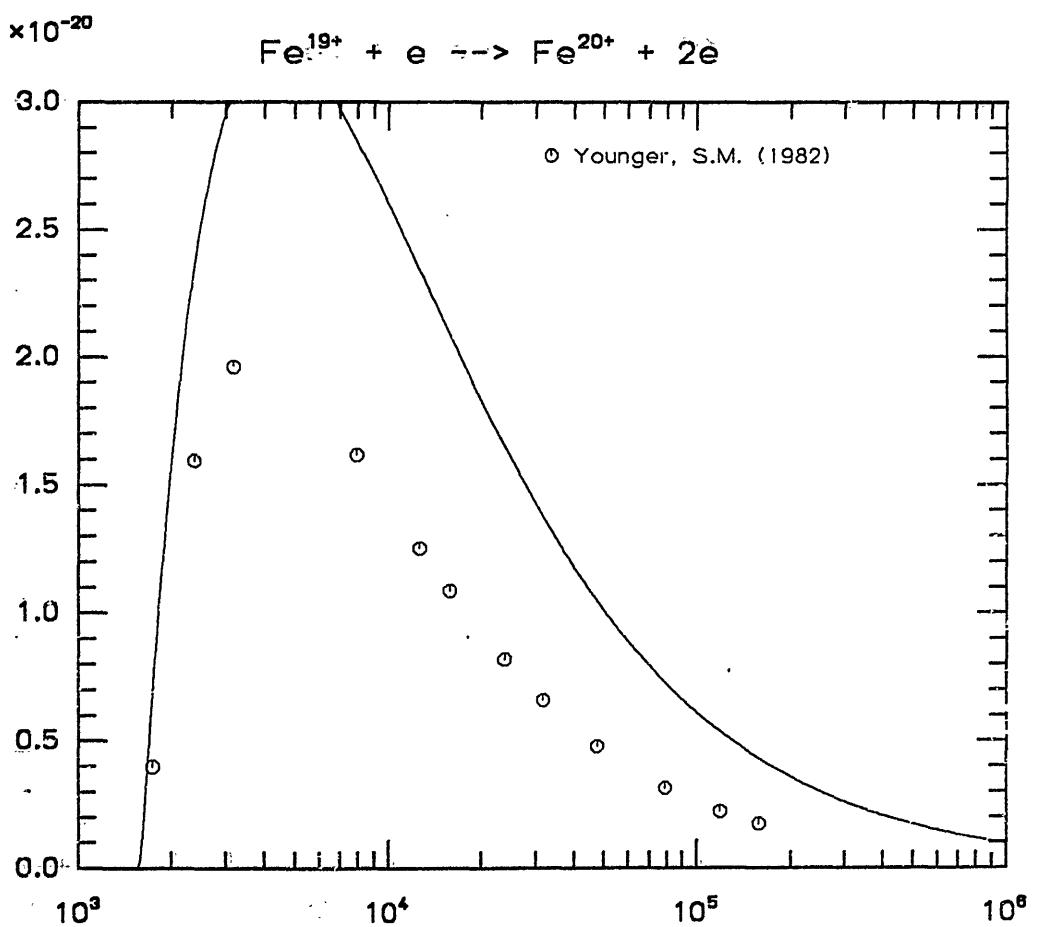


Fig. 169 ○ Electron energy (eV)

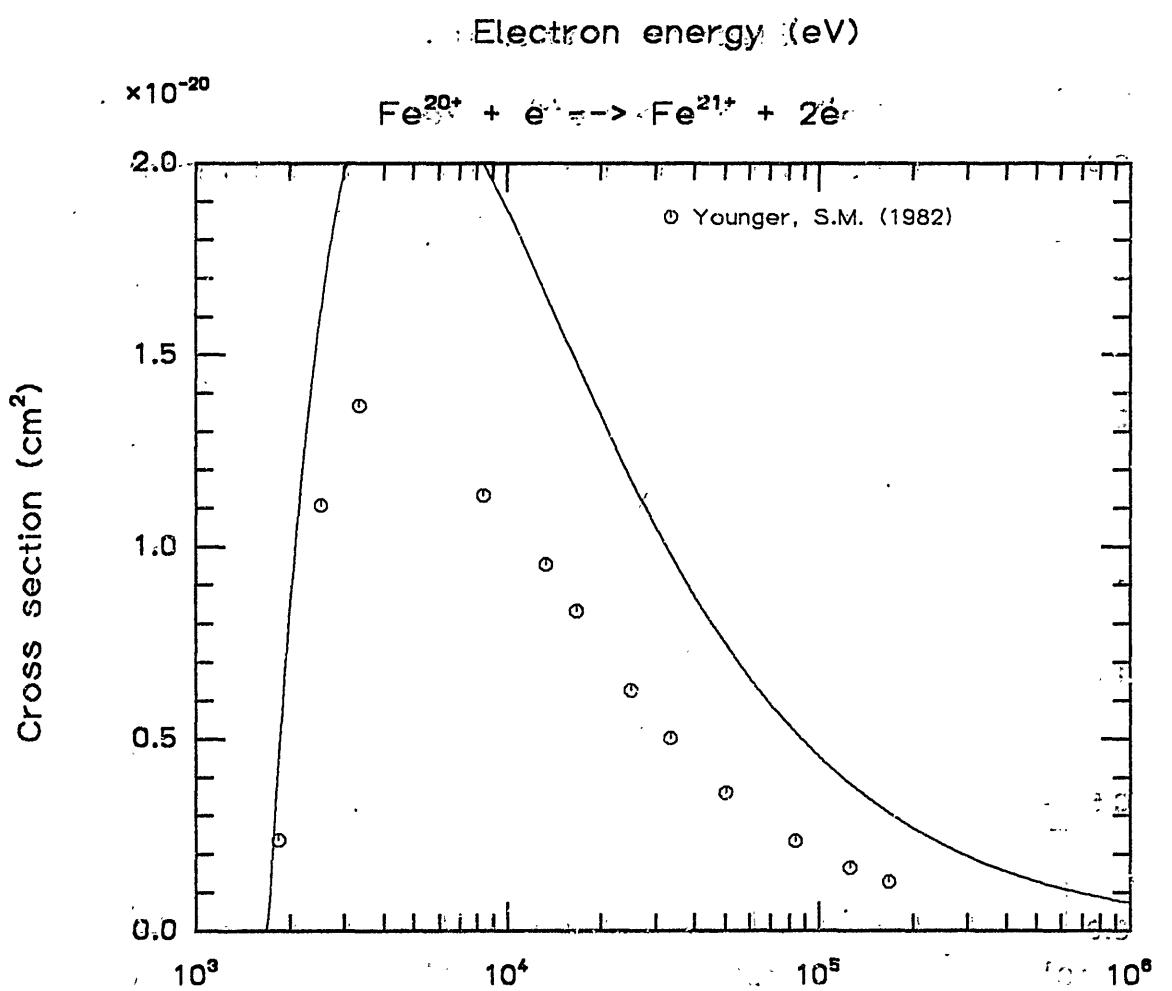
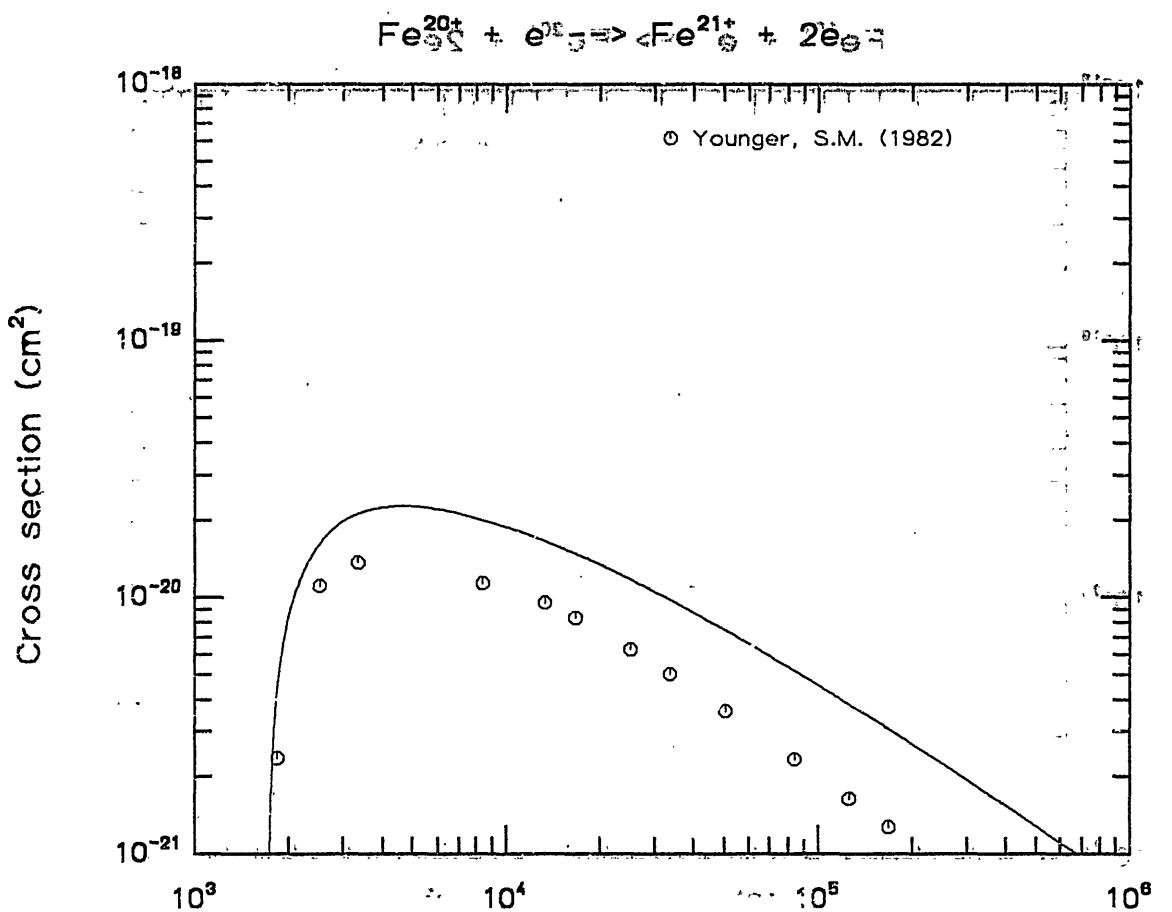
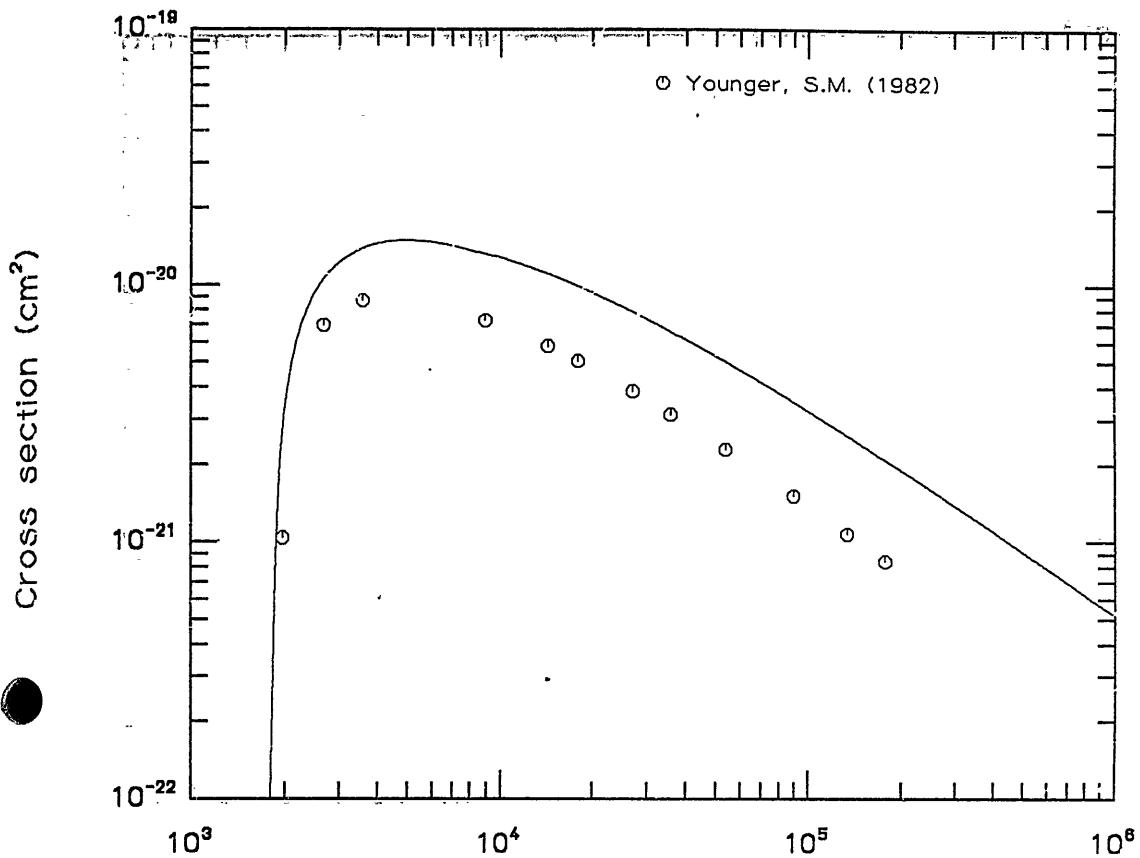
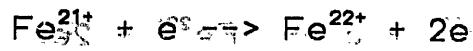


Fig. 170 (a) Electron energy (eV) 0 to 10<sup>6</sup>



Electron energy (eV)

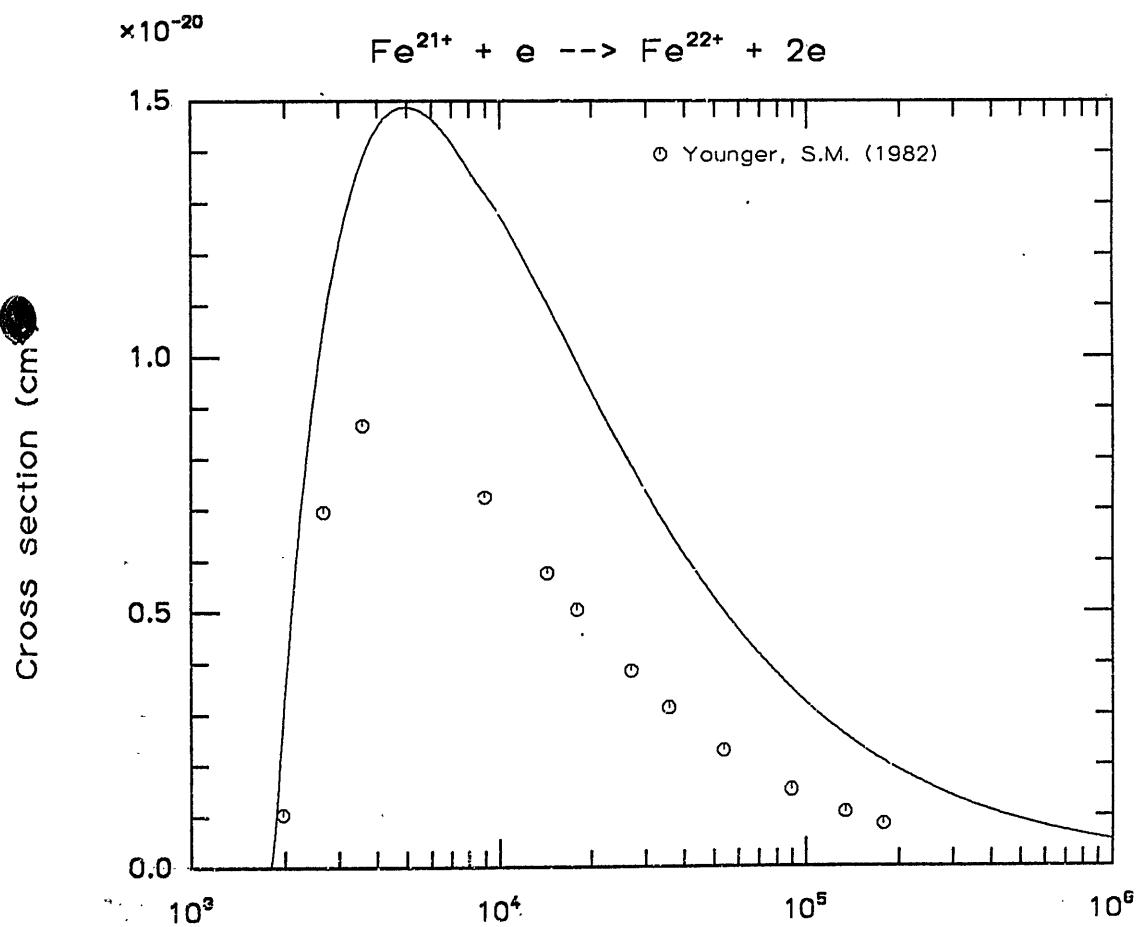


Fig. 171 → Electron energy (eV)

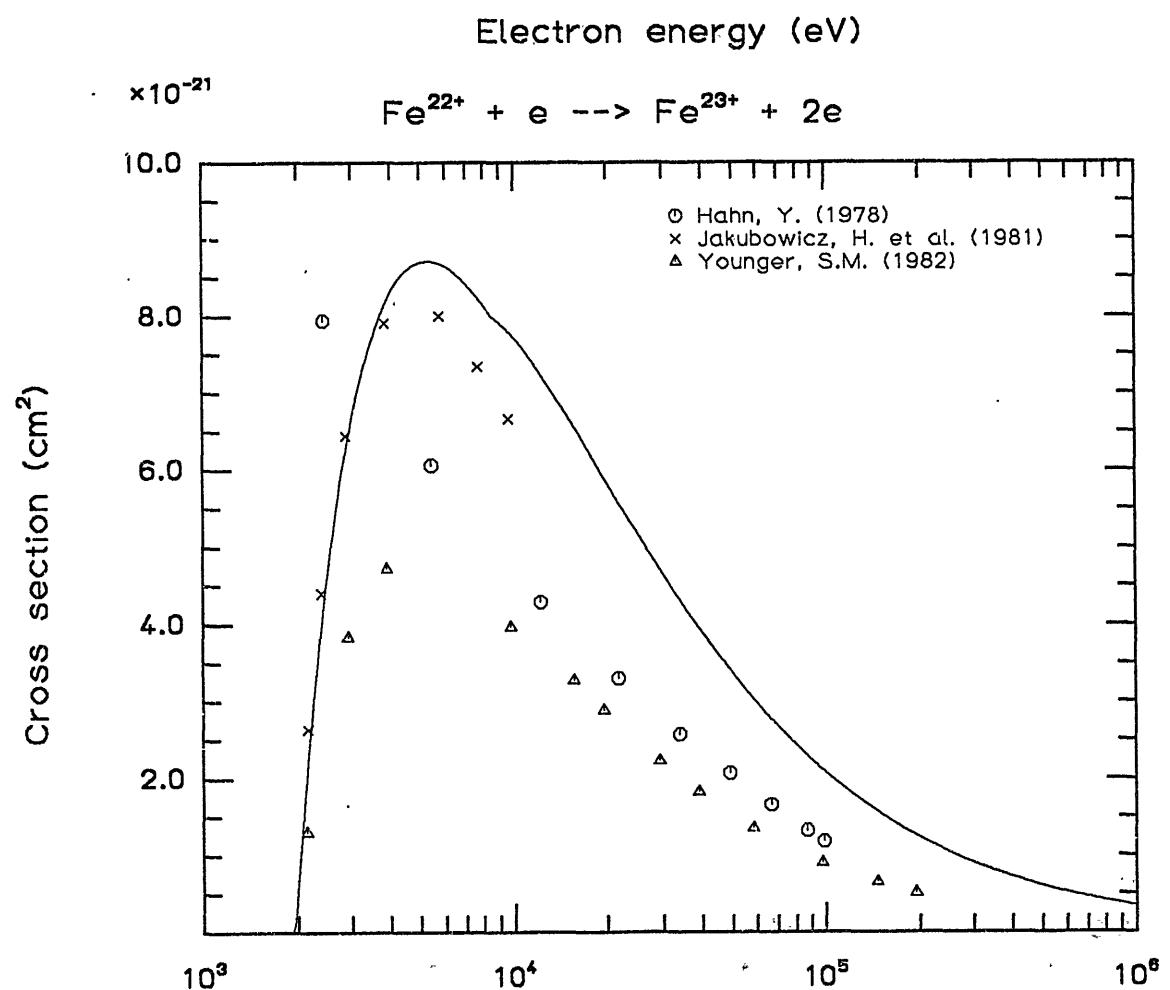
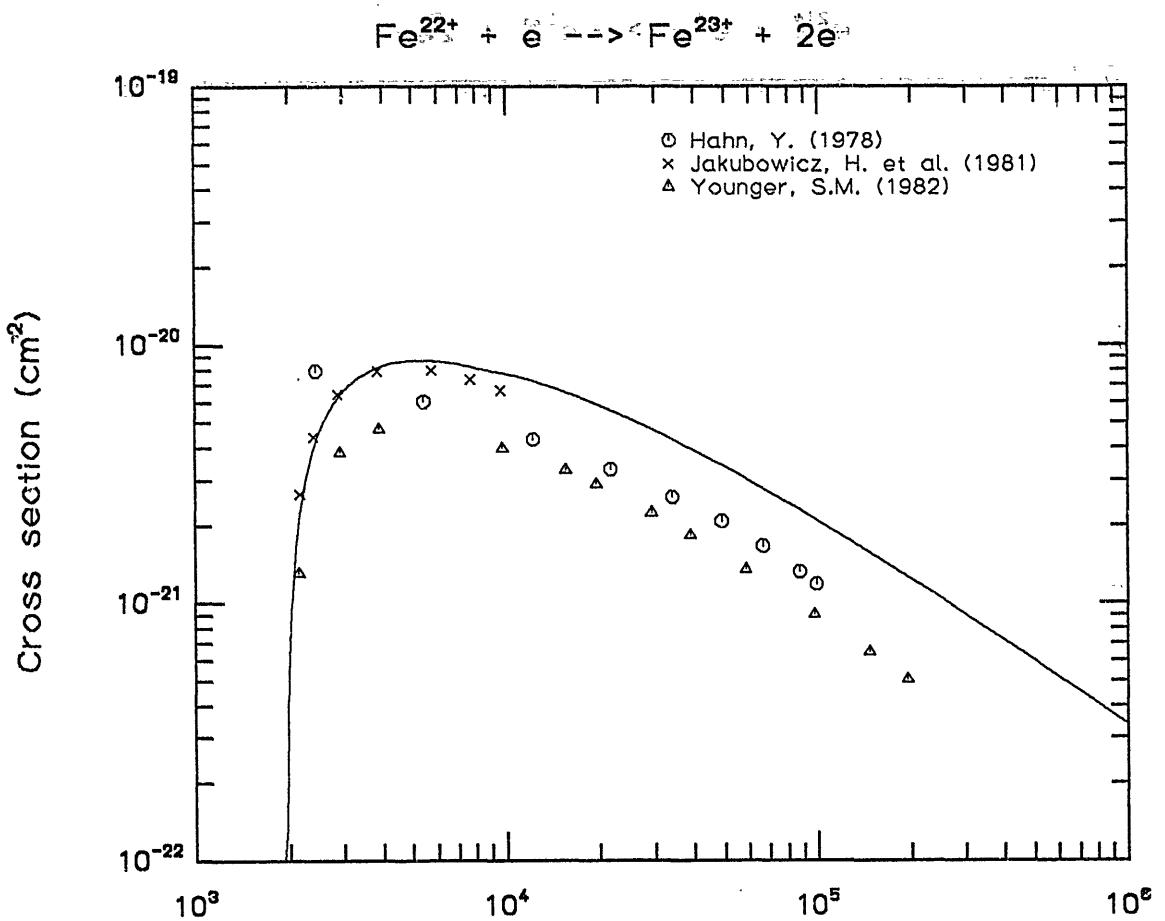
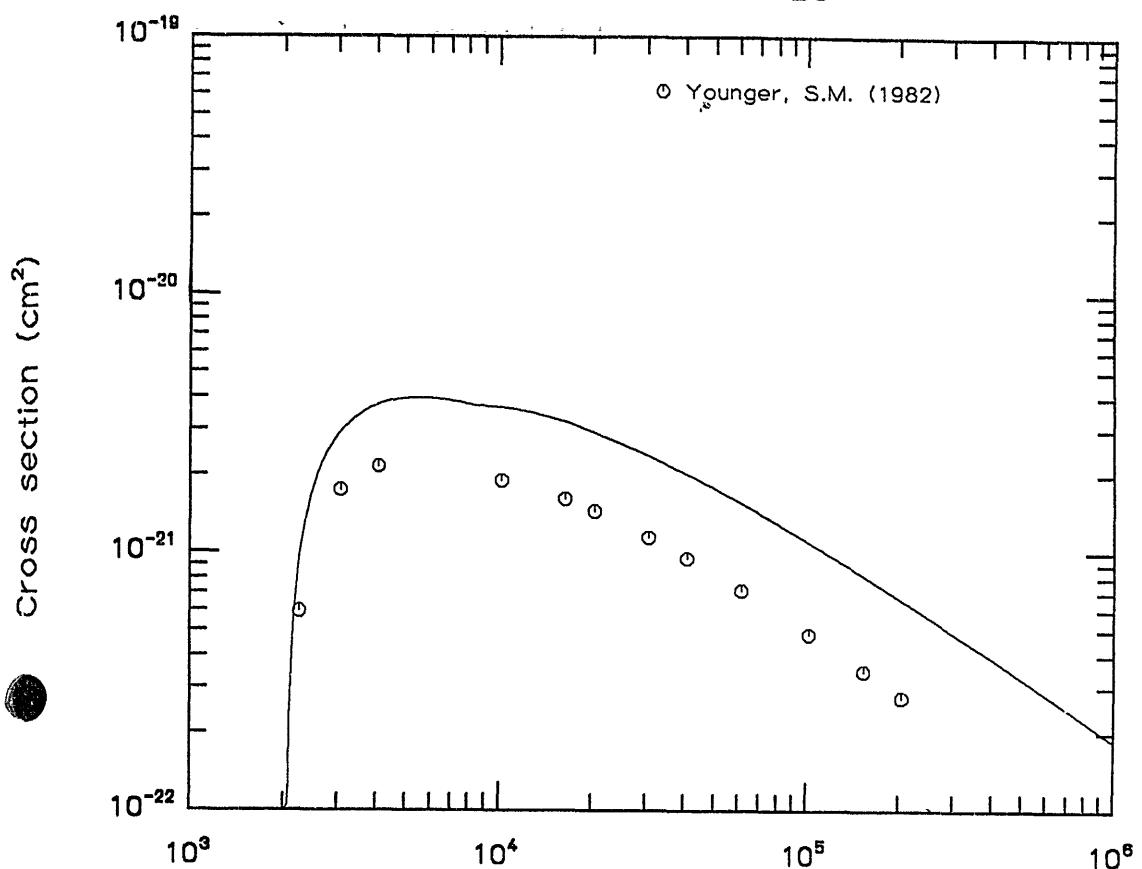
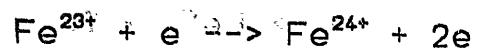
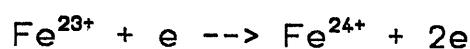


Fig. 172      Electron energy (eV)       $\text{cm}^2$       pA



Electron energy (eV)

$\times 10^{-21}$



Cross section (cm<sup>2</sup>)

3.0

2.5

2.0

1.5

1.0

0.5

0.0

○ Younger, S.M. (1982)

10<sup>3</sup>

10<sup>4</sup>

10<sup>5</sup>

10<sup>6</sup>

Fig. 173

Electron energy (eV)

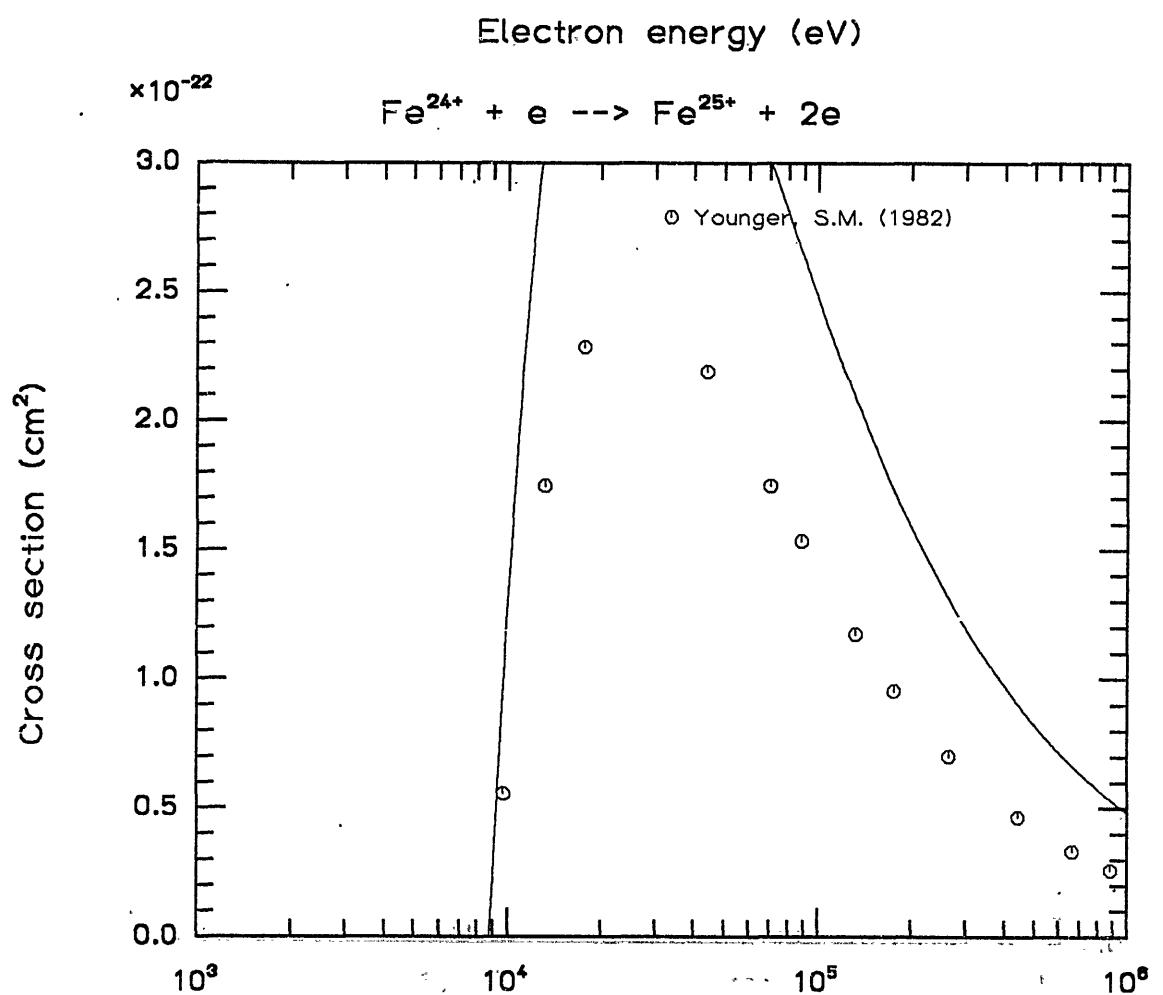
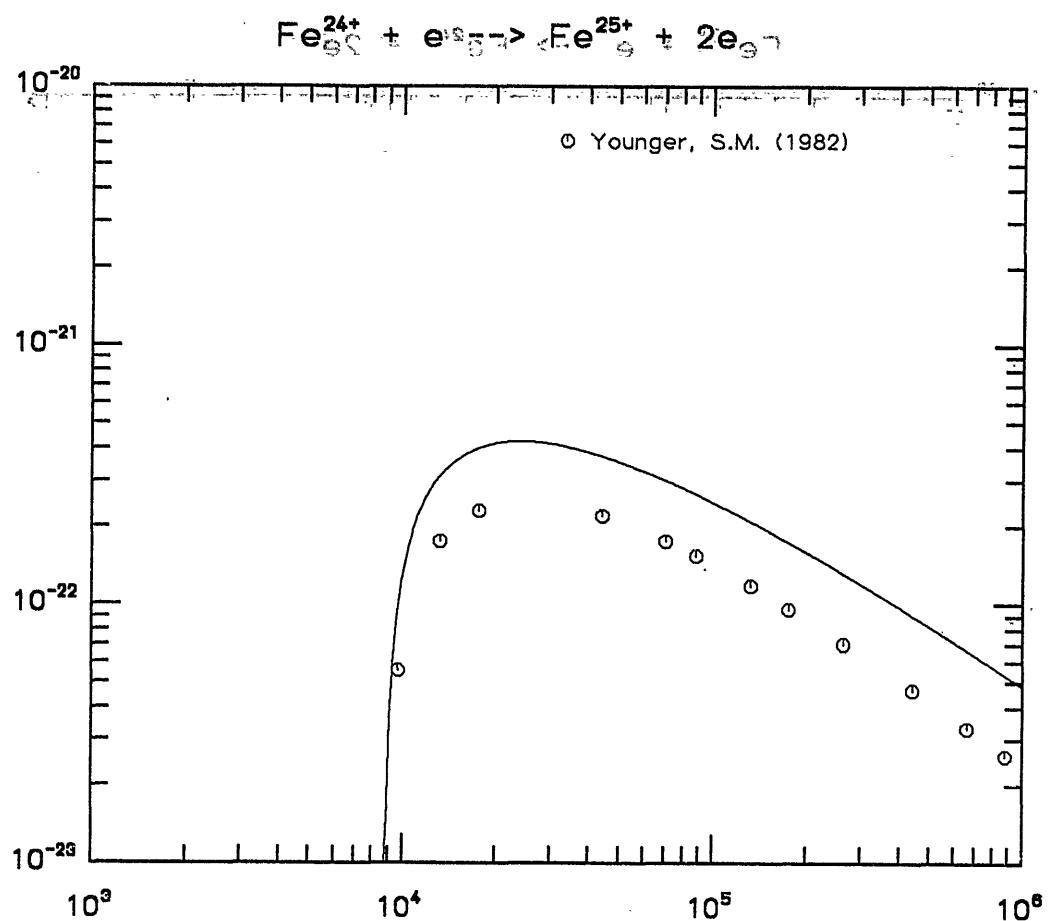
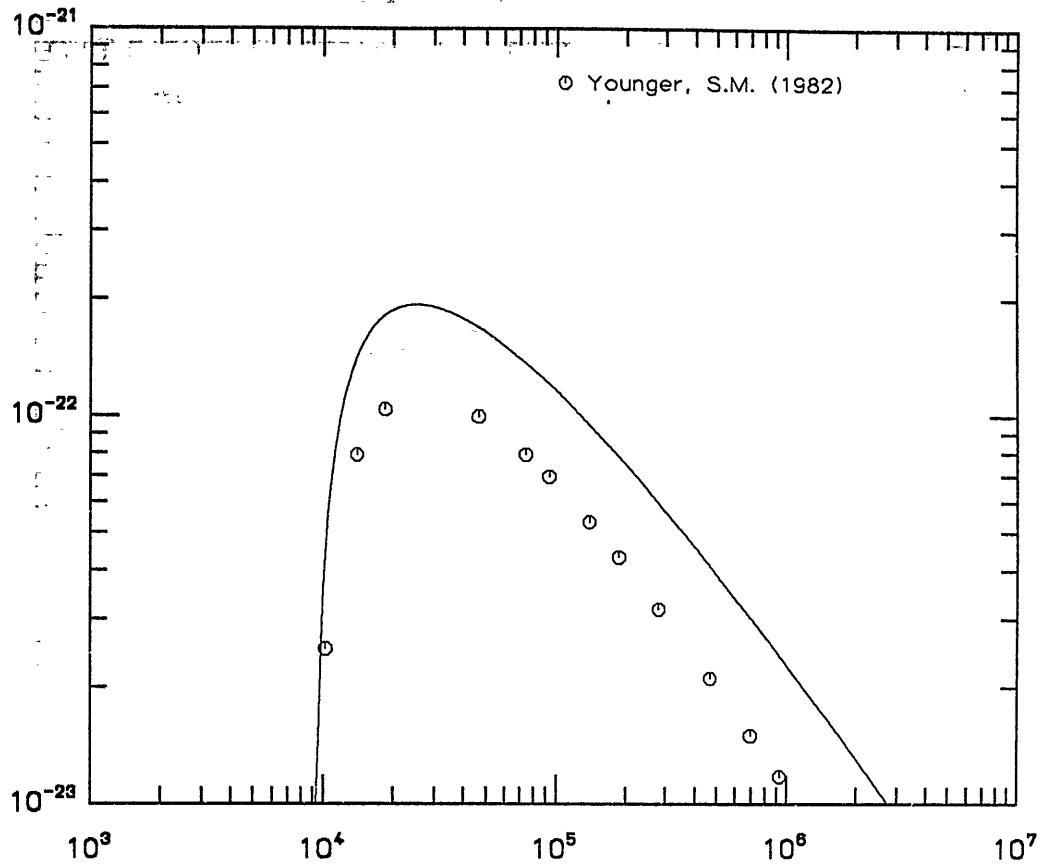
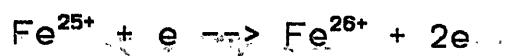


Fig. 174 Electron energy (eV)  $\times 10^{-22}$



Electron energy (eV)

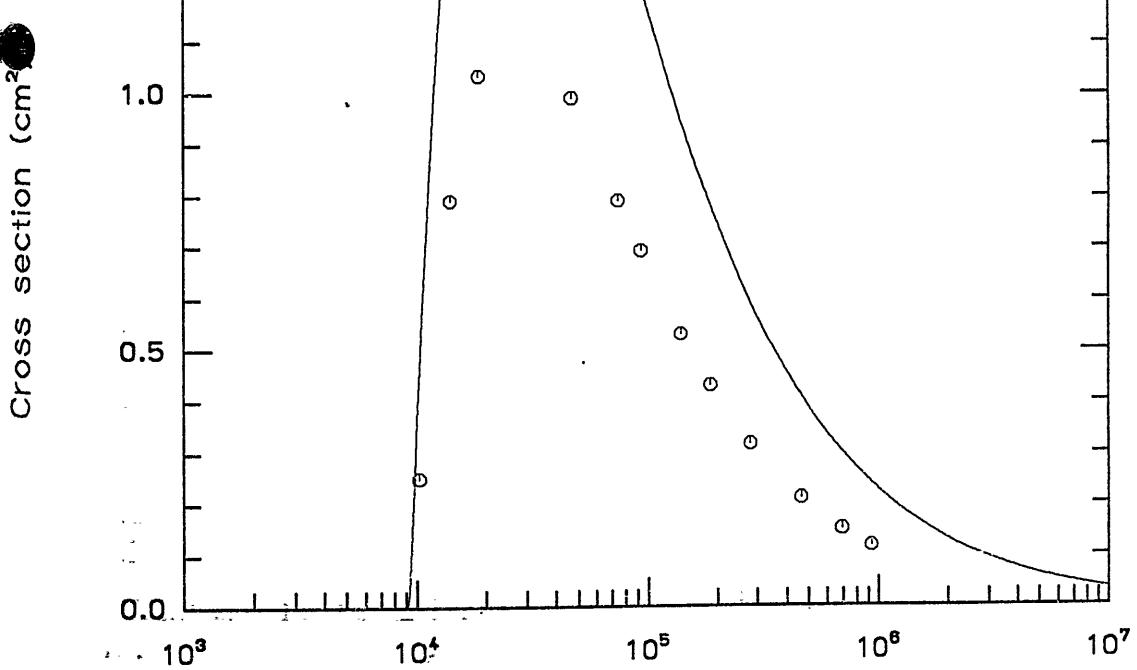
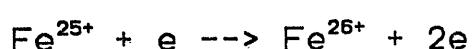


Fig. 175

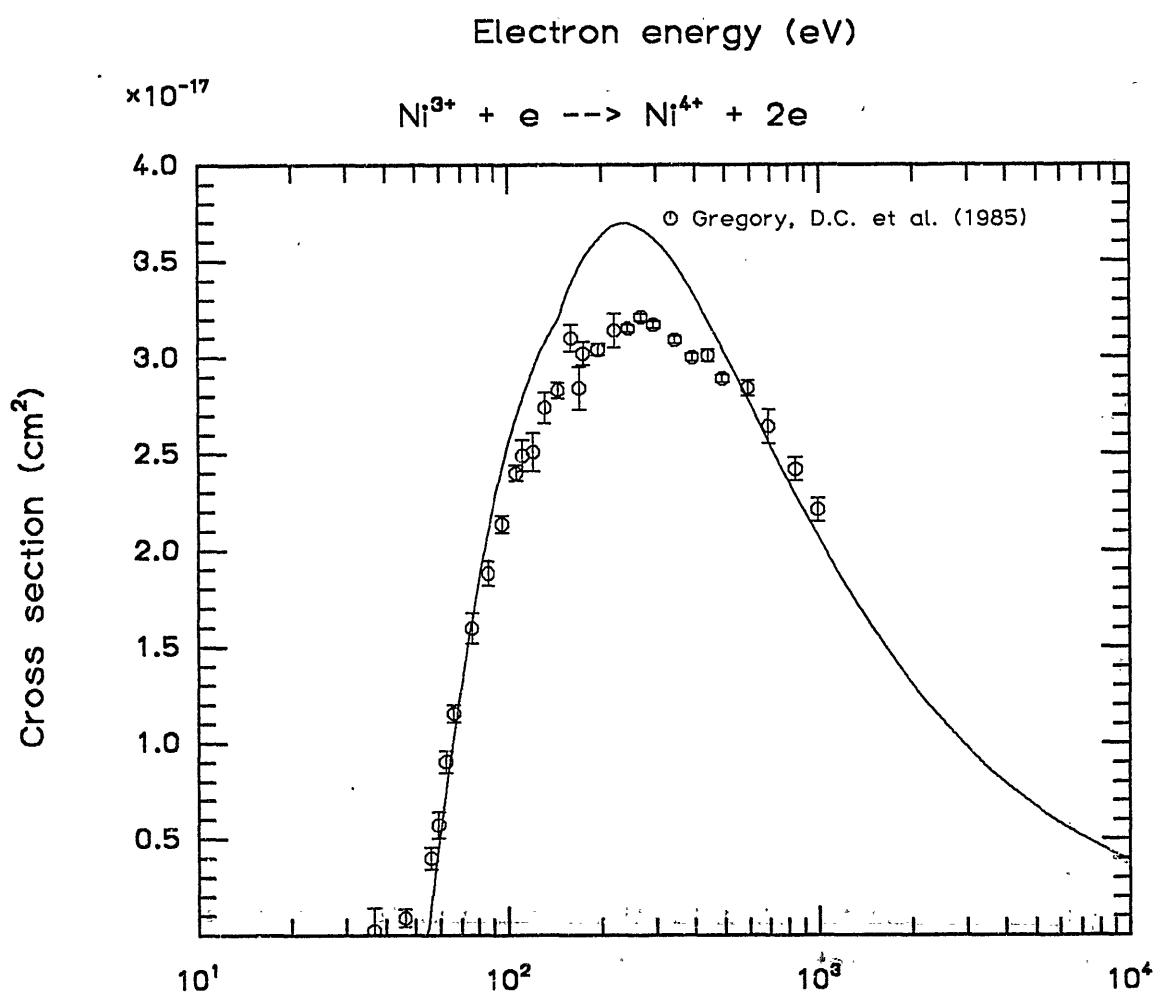
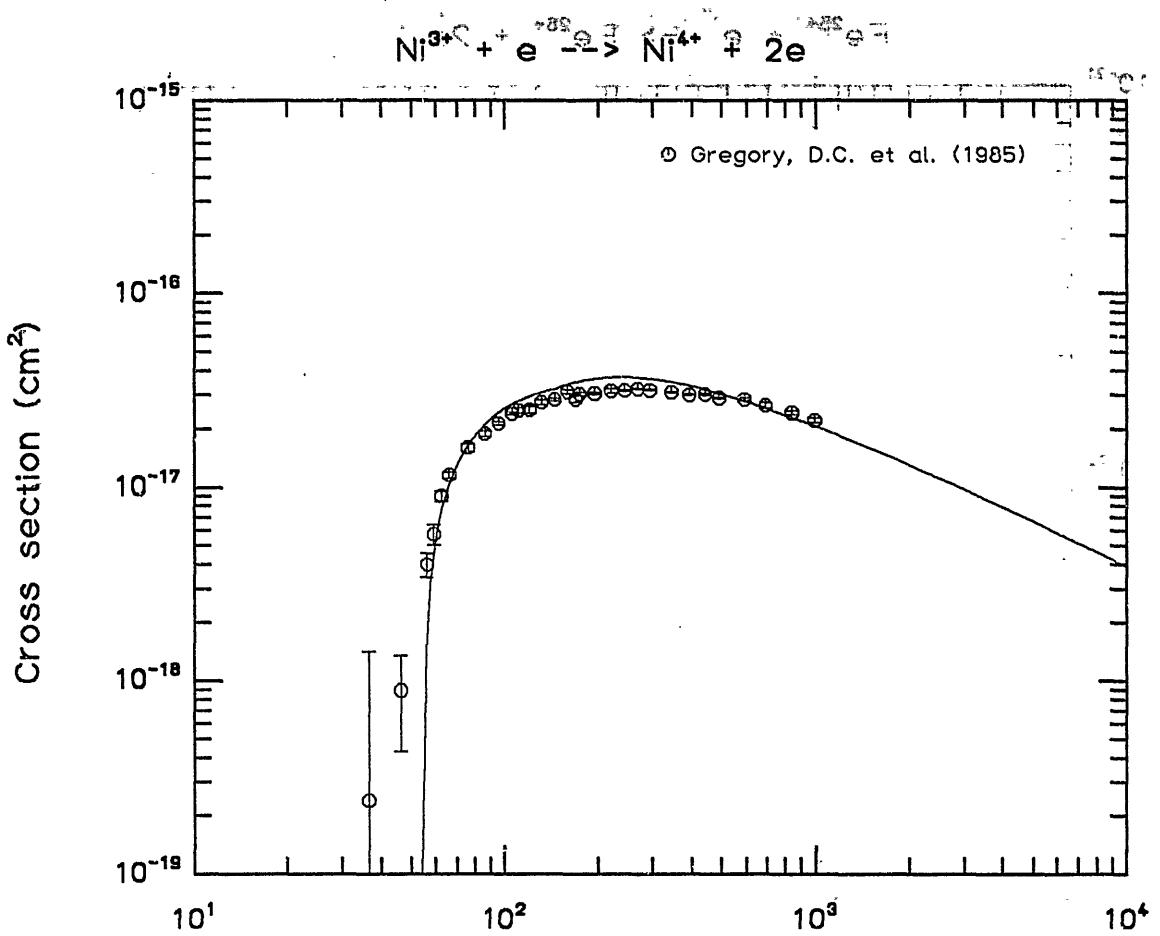
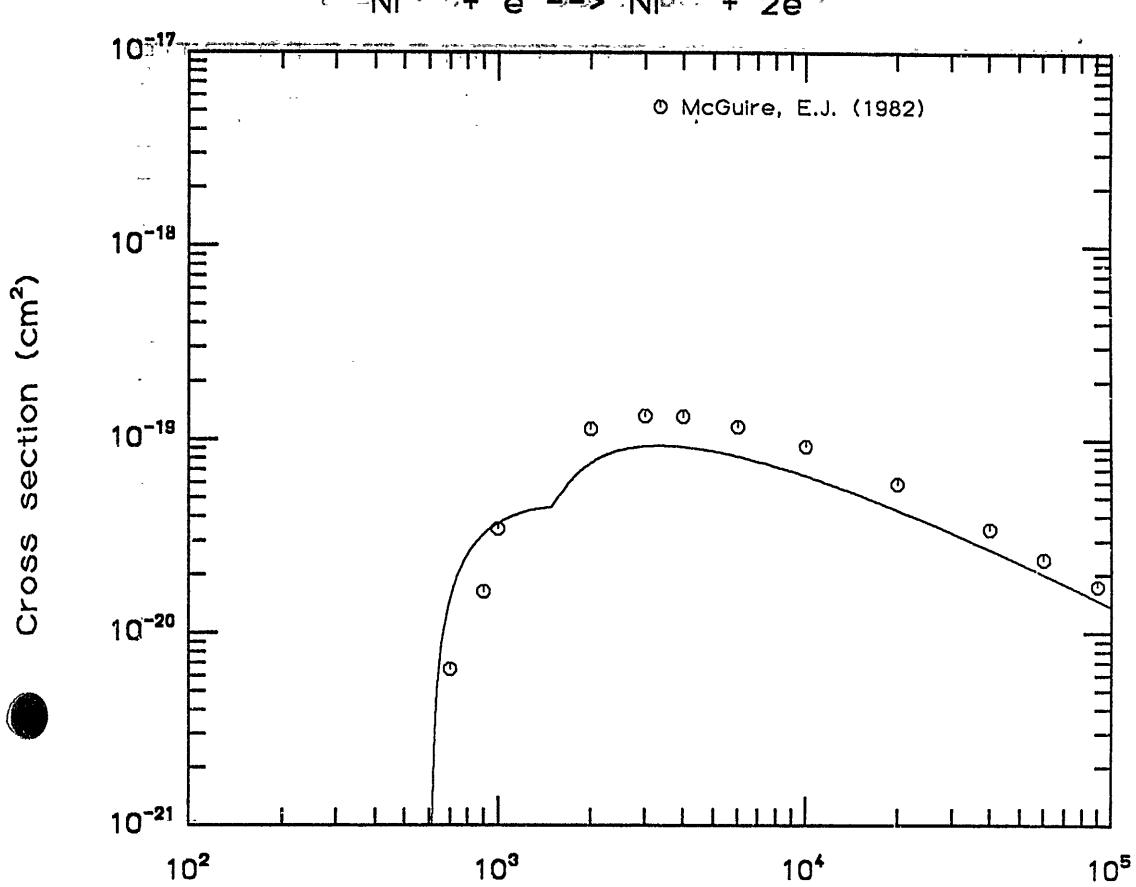
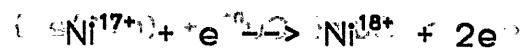


Fig. 176



Electron energy (eV)

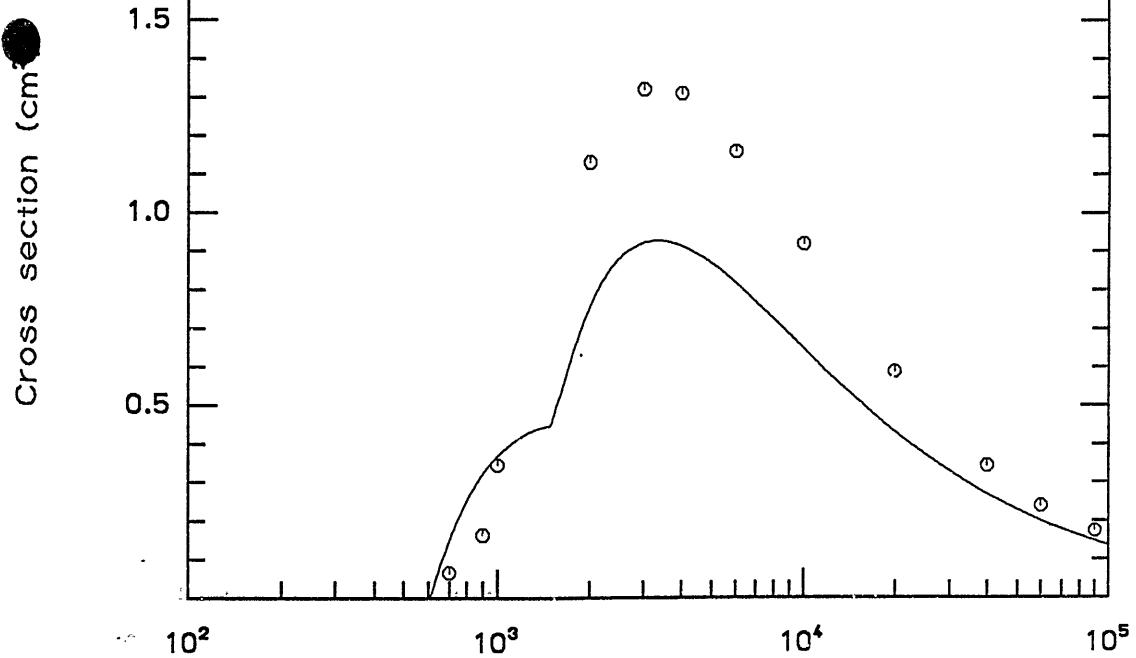
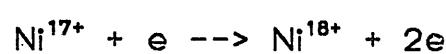


Fig. 177      Electron energy (eV)

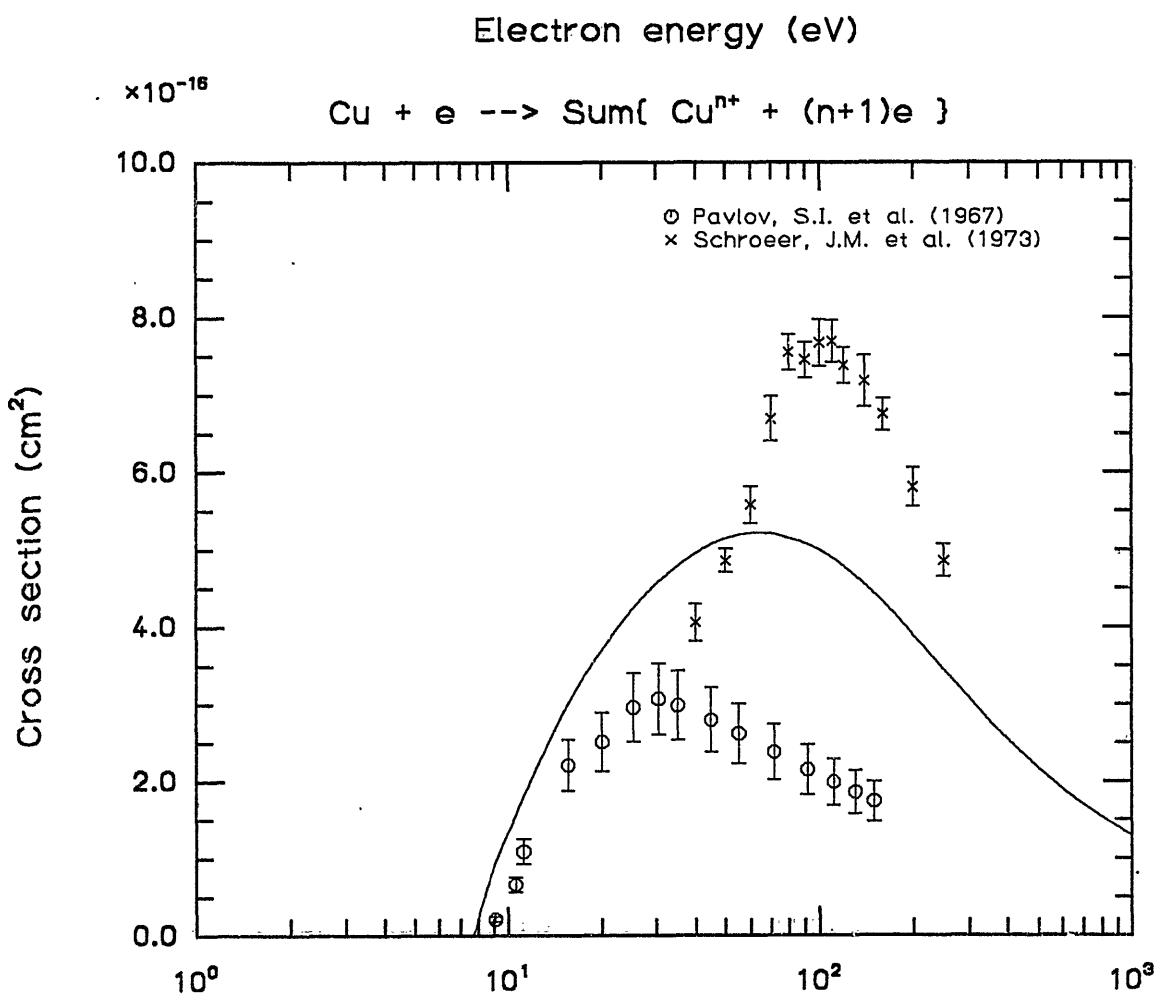
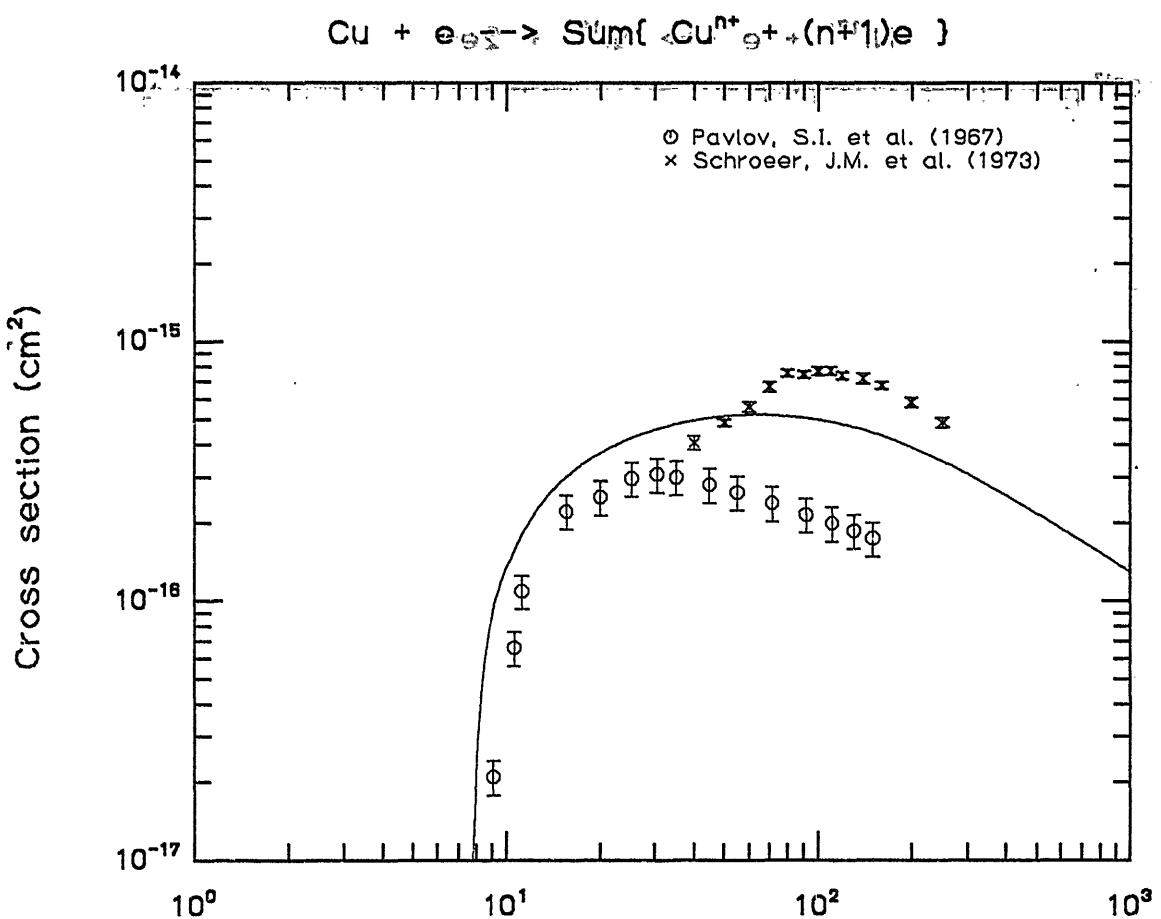


Fig. 178      Electron energy (eV)       $\text{Cu} + e \rightarrow \text{Sum}\{ \text{Cu}^{n+} + (n+1)e \}$

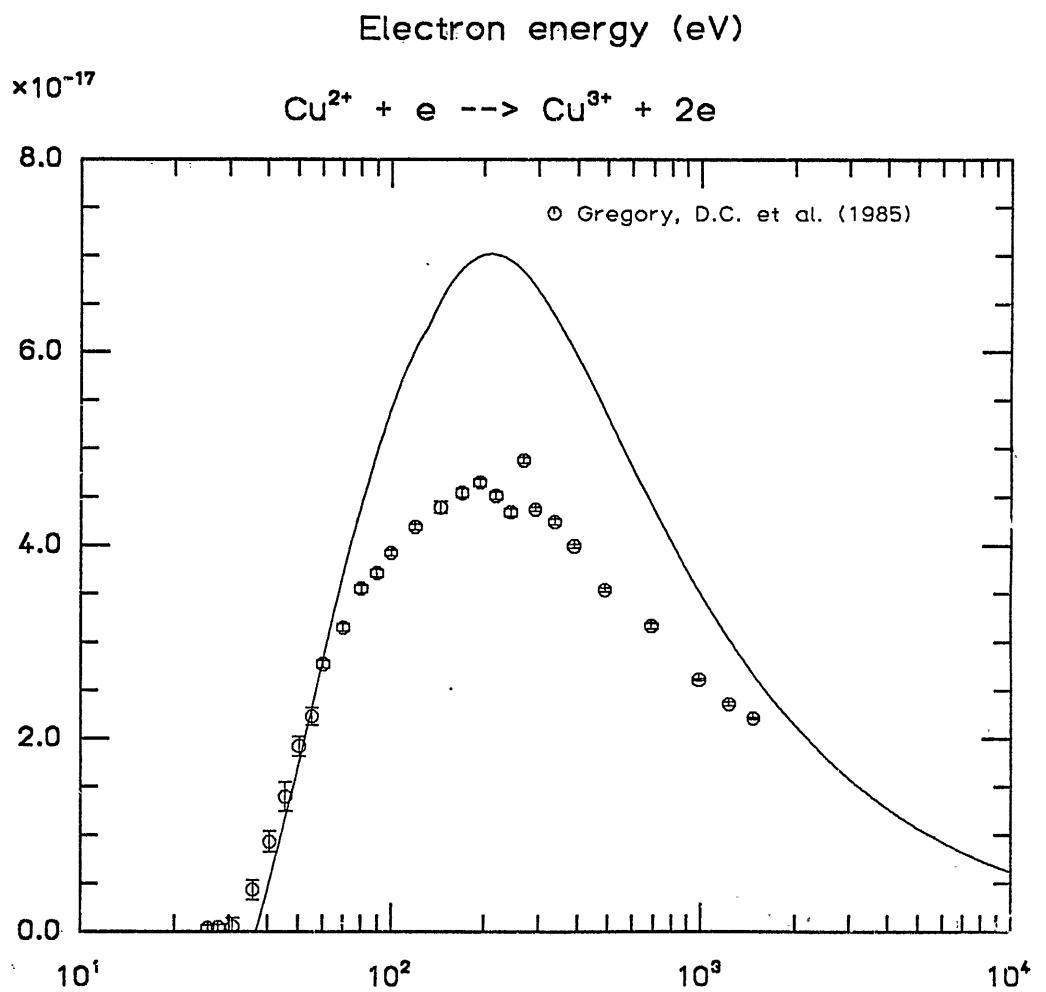
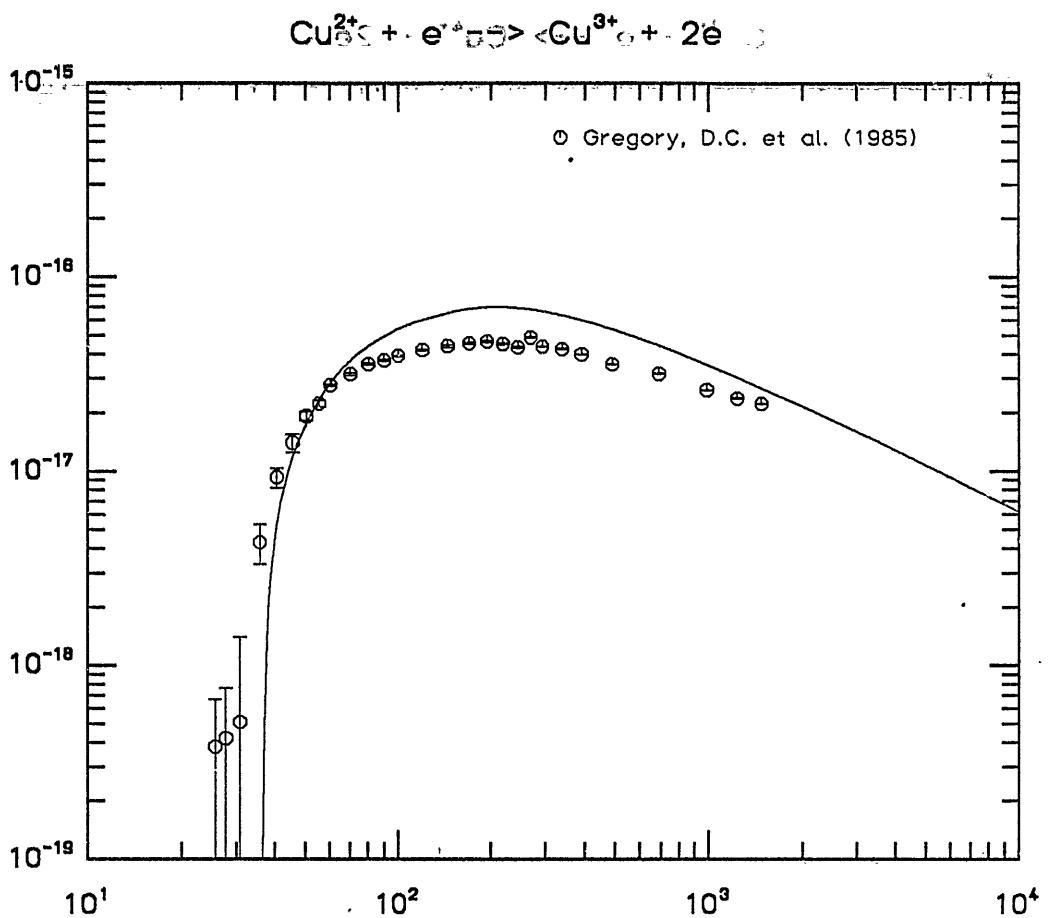


Fig. 179 Electron energy (eV)

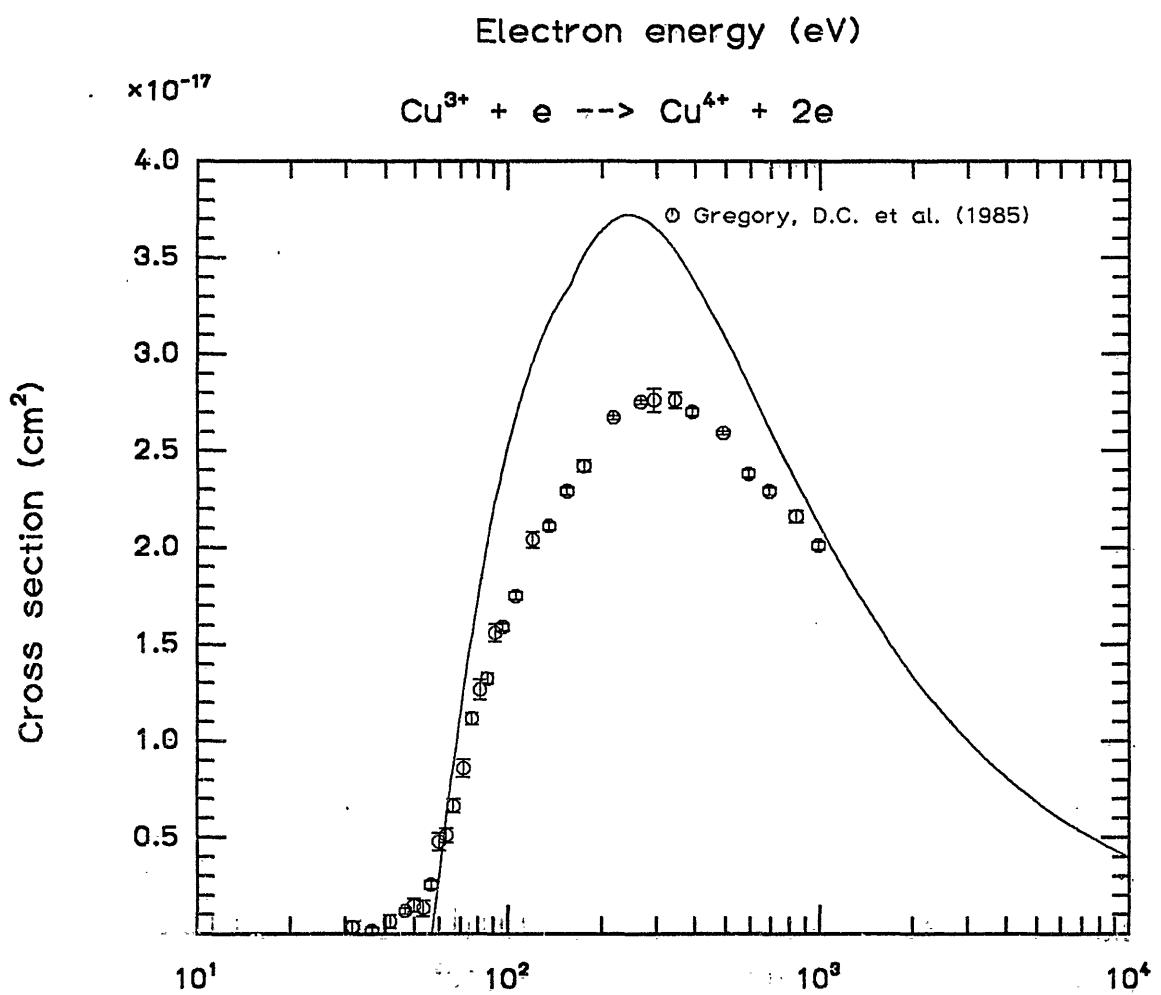
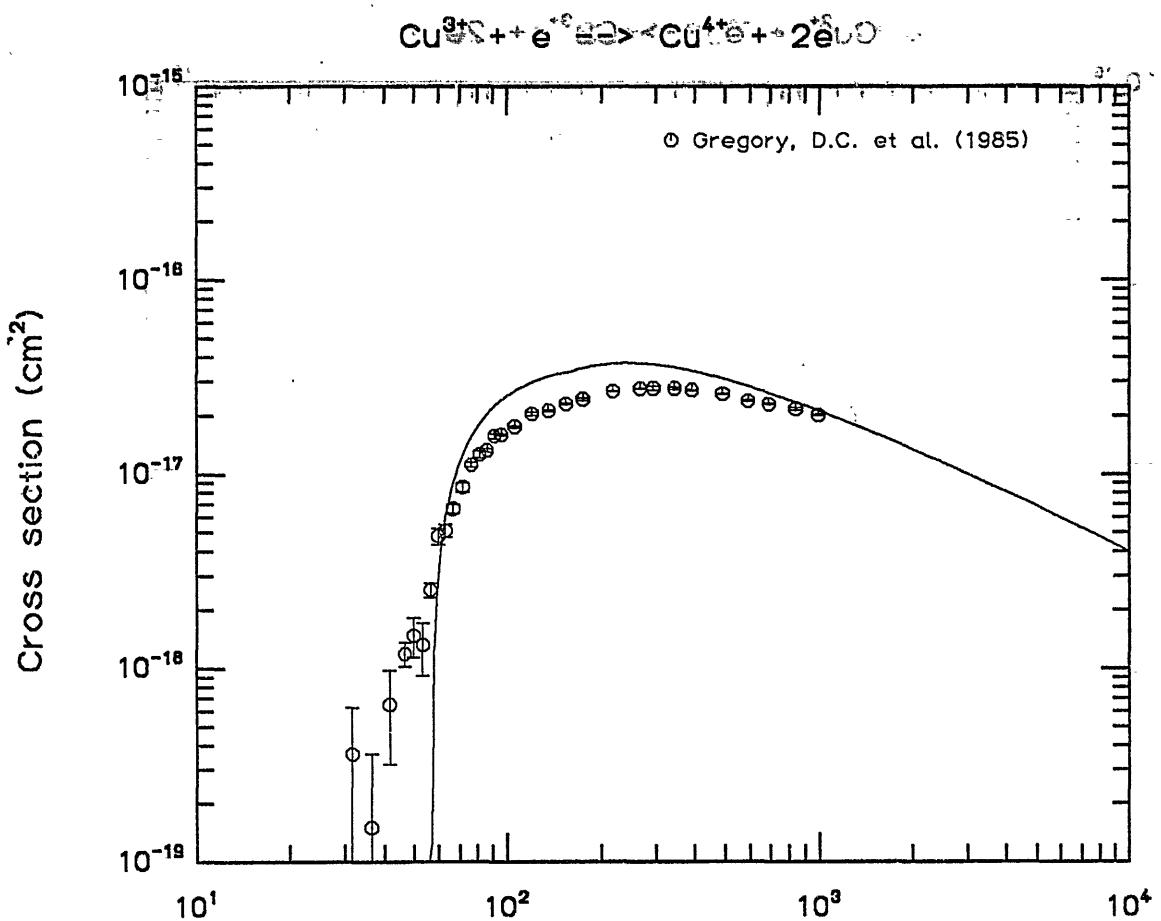
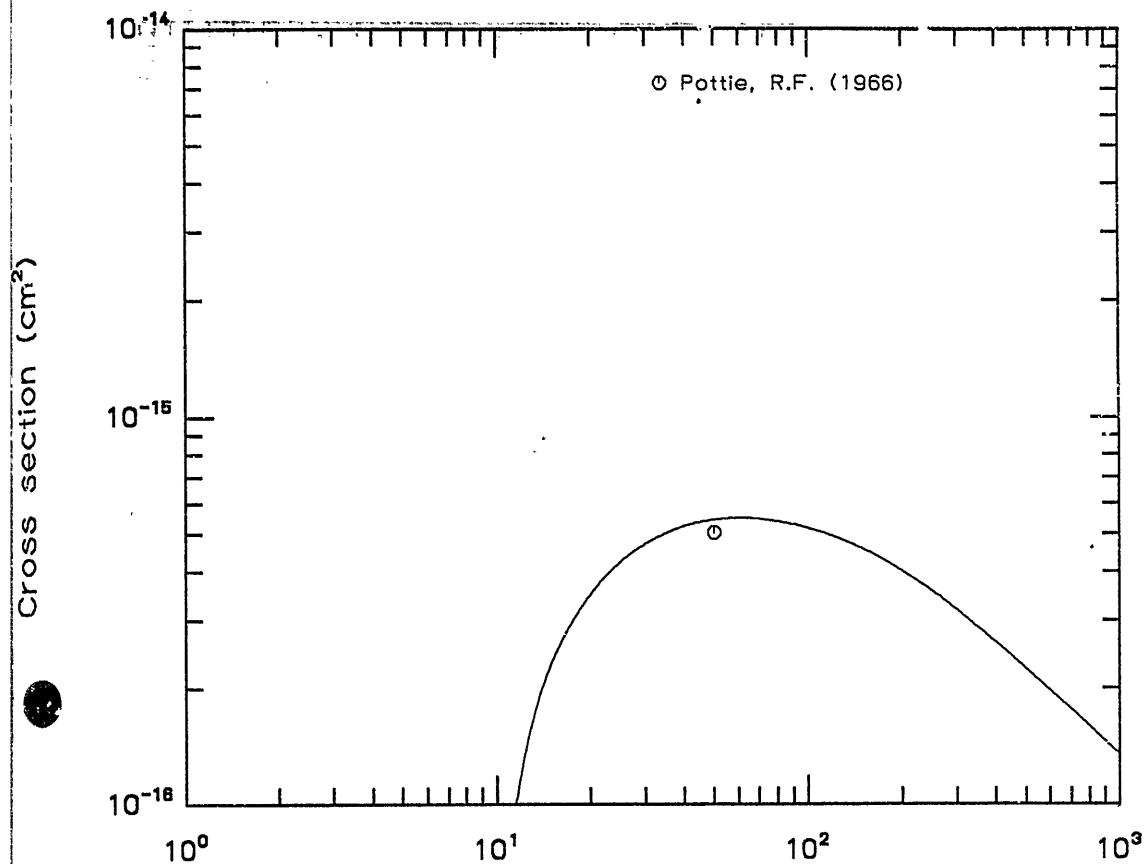
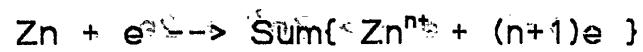


Fig. 180 Cross section vs Electron energy (eV) (Ref. 18)



Electron energy (eV)

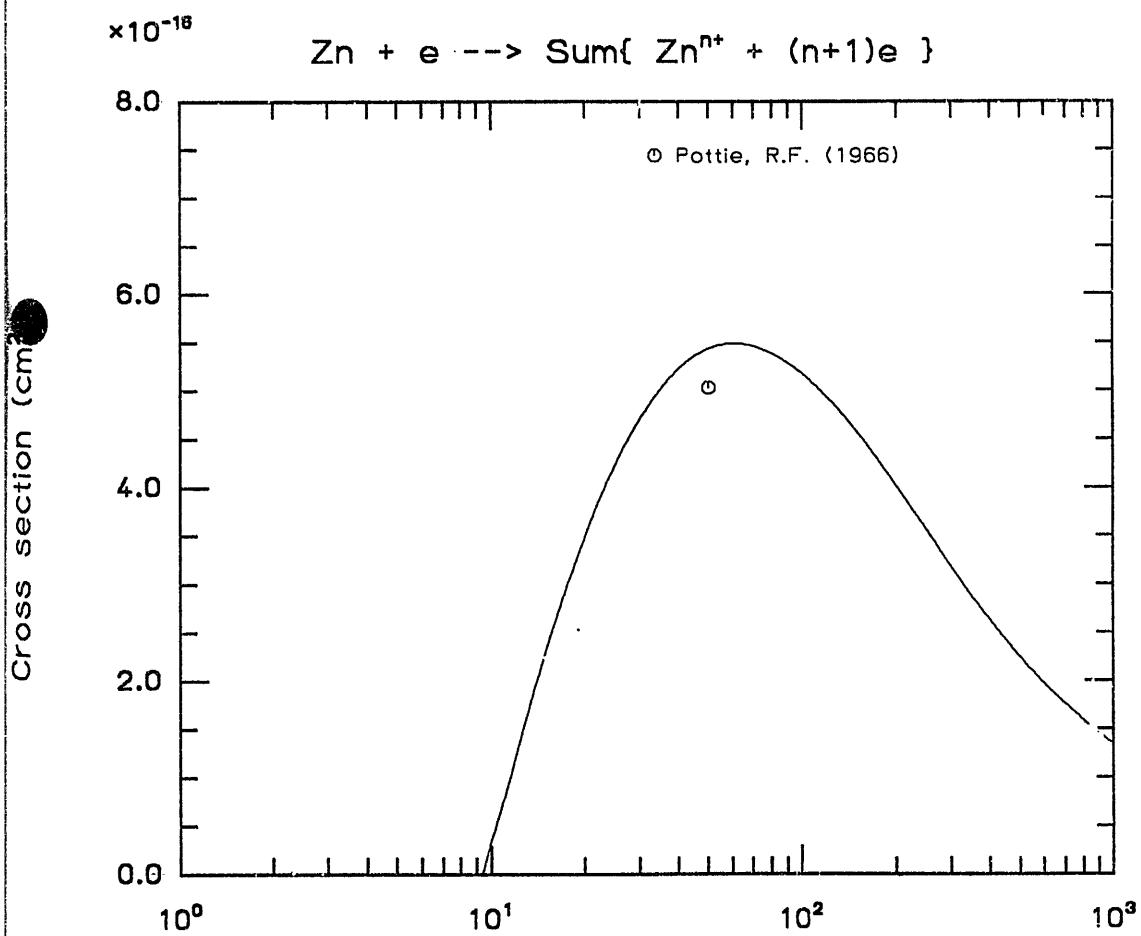


Fig. 181

Electron energy (eV)

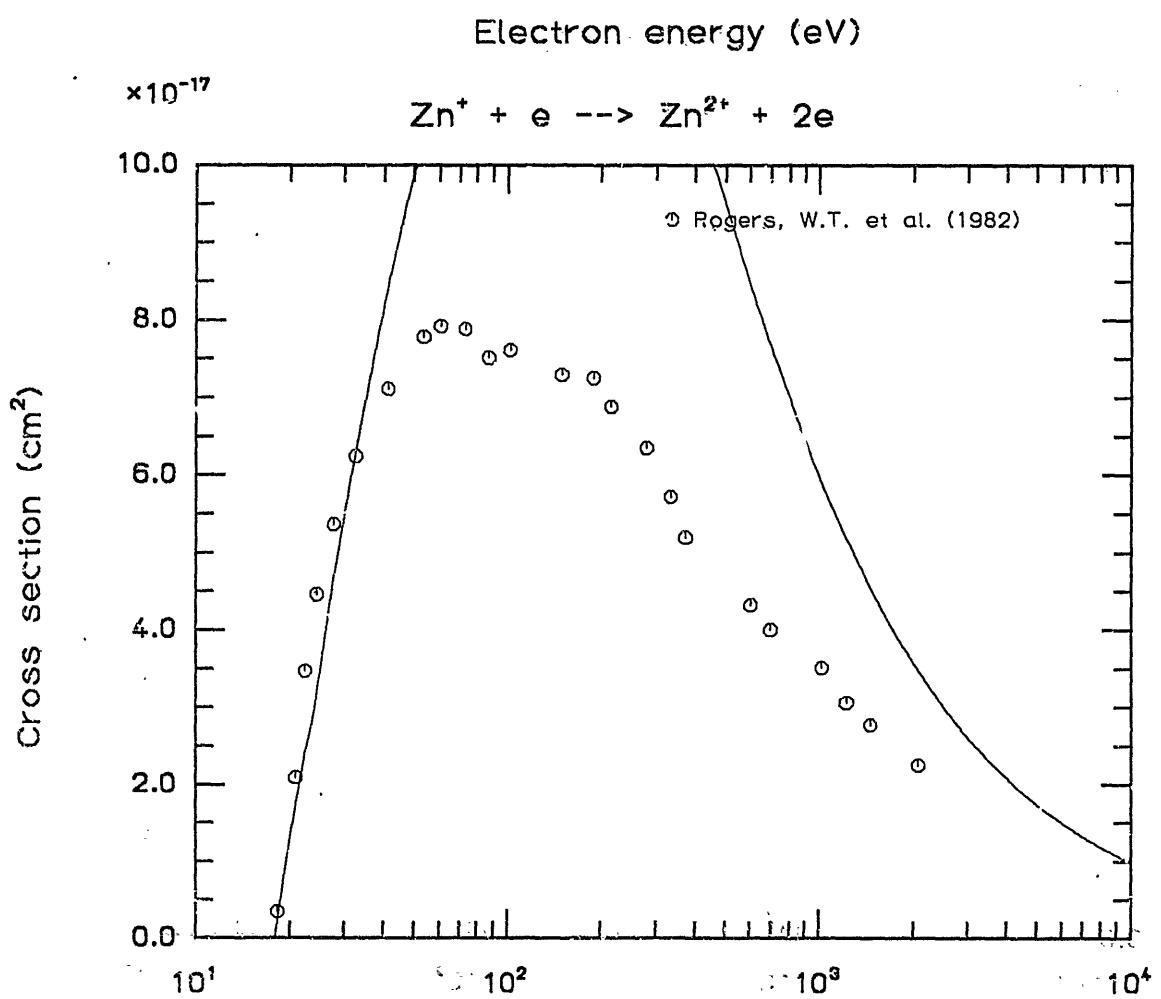
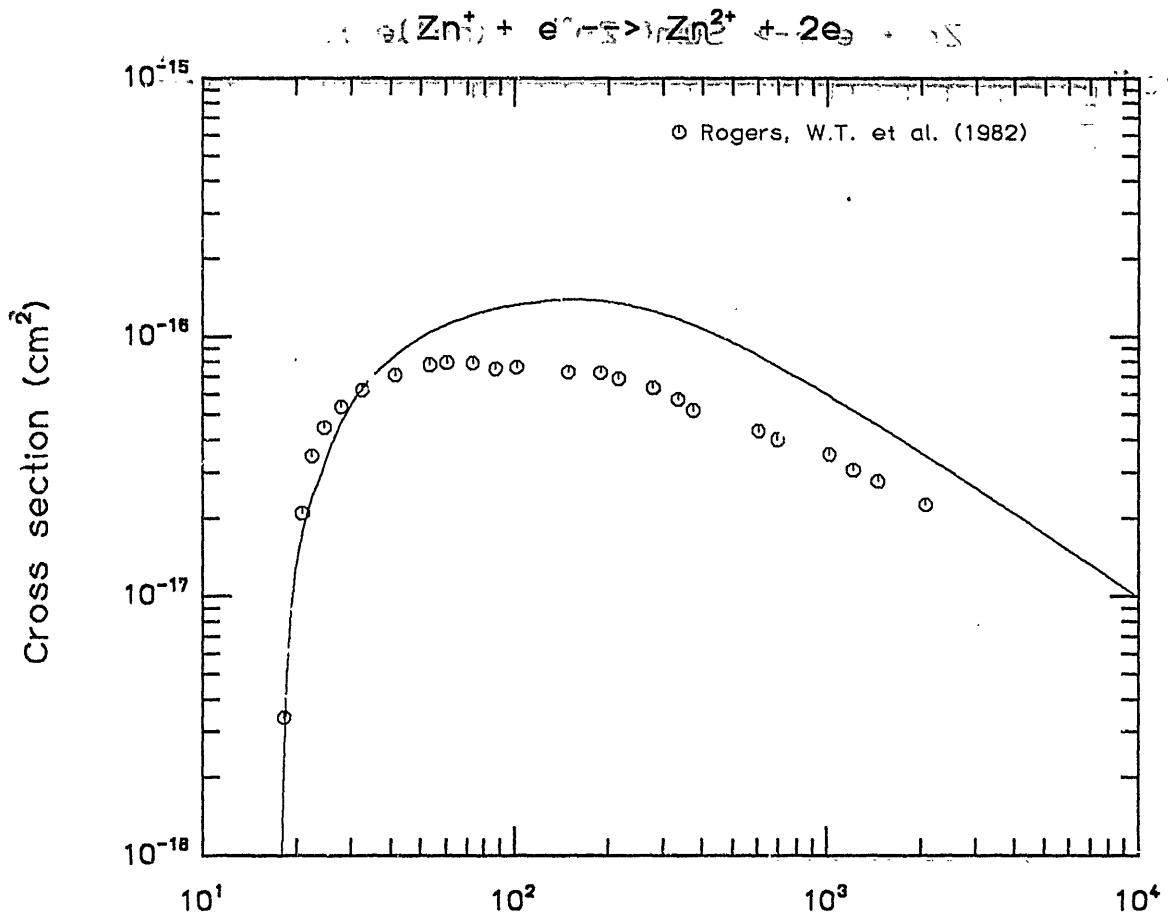


Fig. 182 (vs Electron energy (eV)) 181.pdf

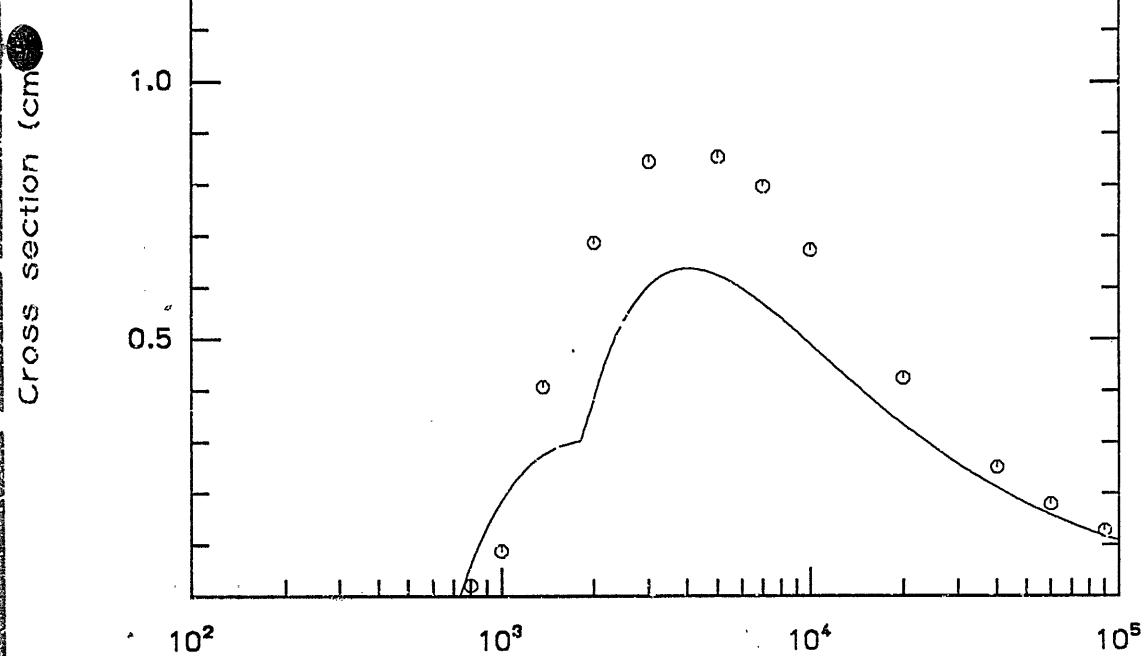
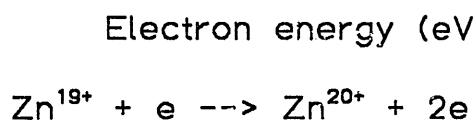
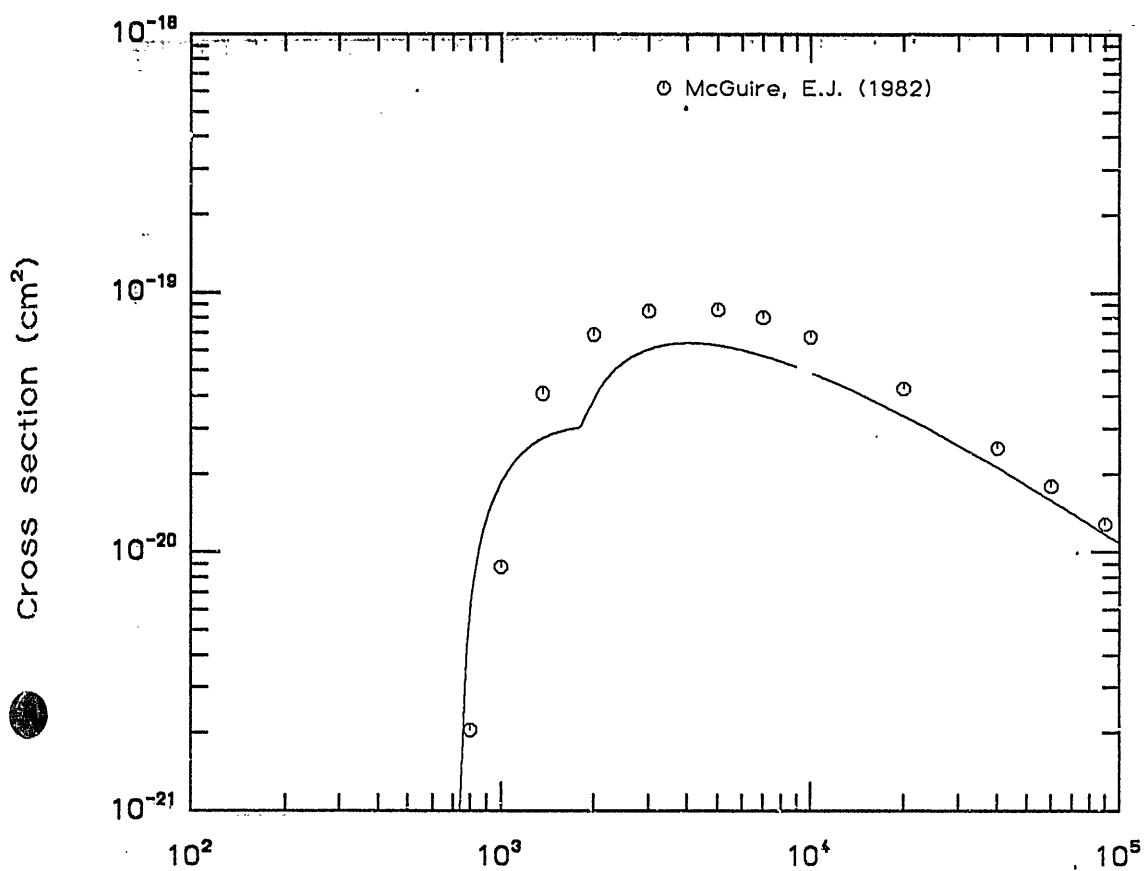
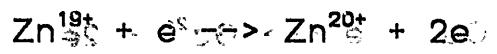


Fig. 183      Electron energy (eV)

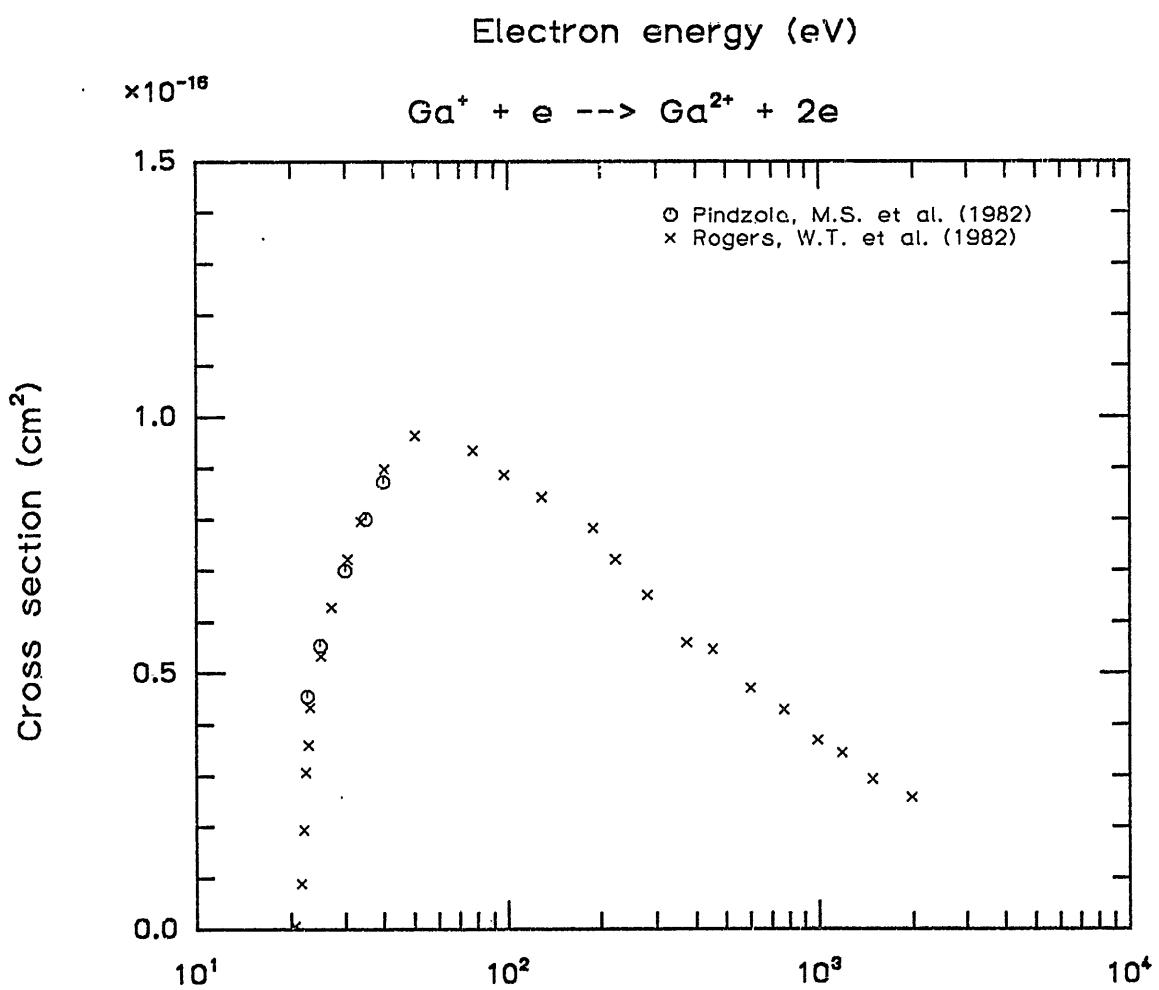
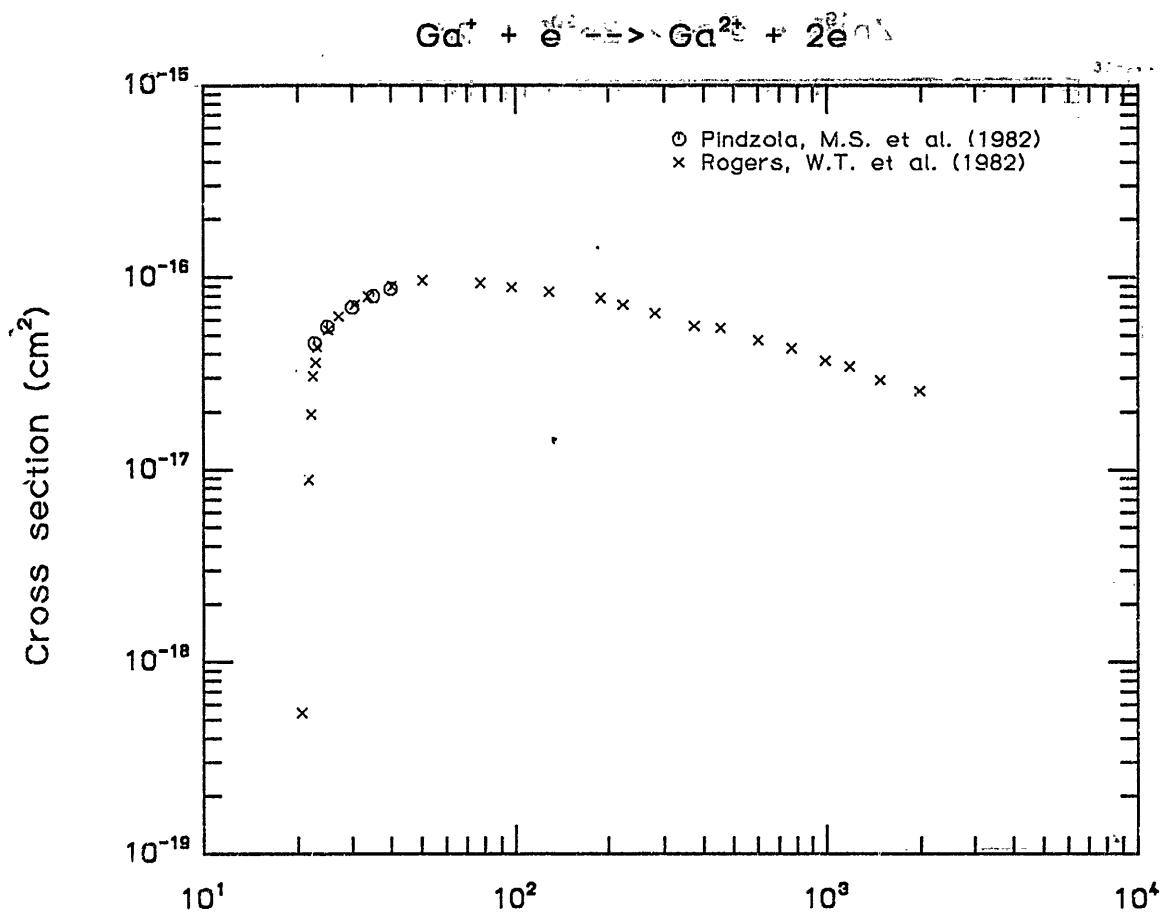


Fig. 184 Electrom energy (eV) 251 p.4

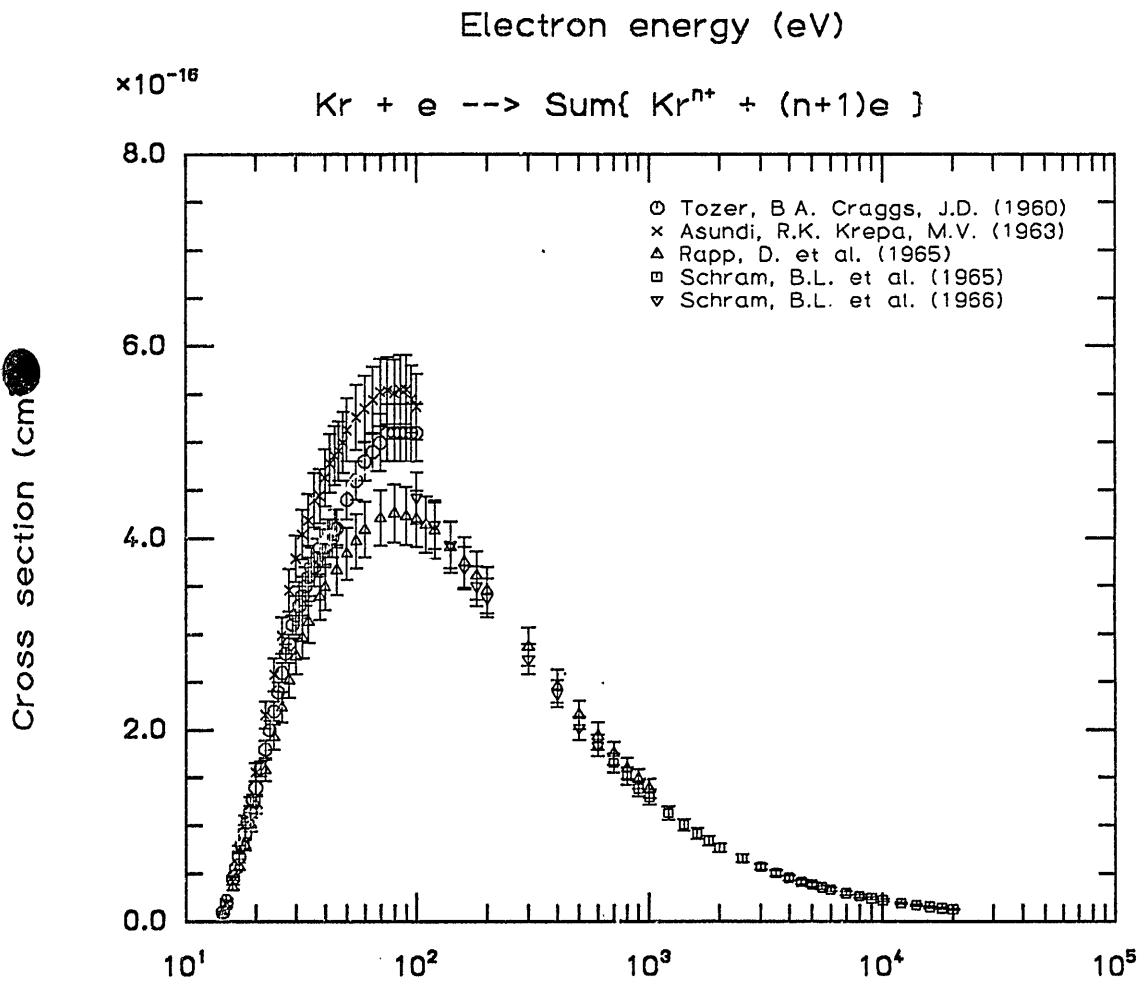
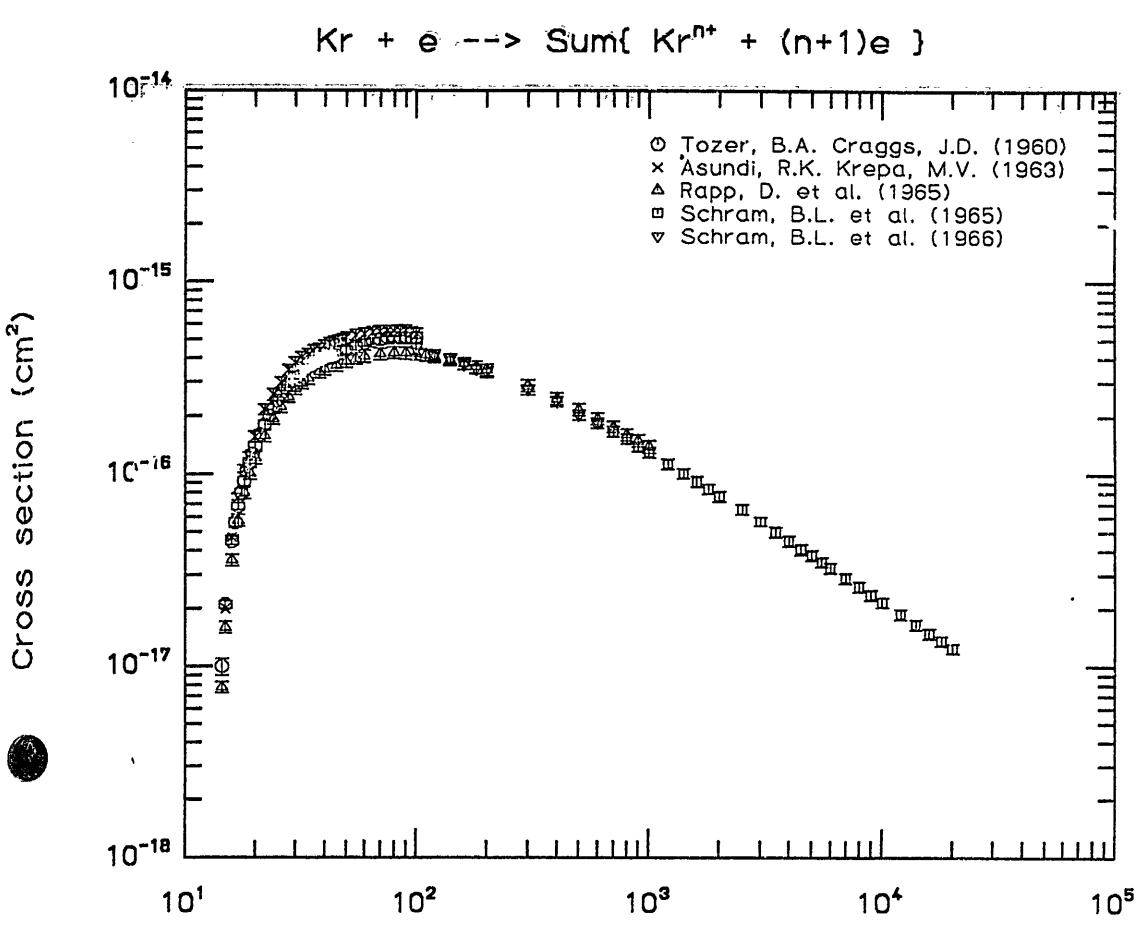


Fig. 185      Electron energy (eV)

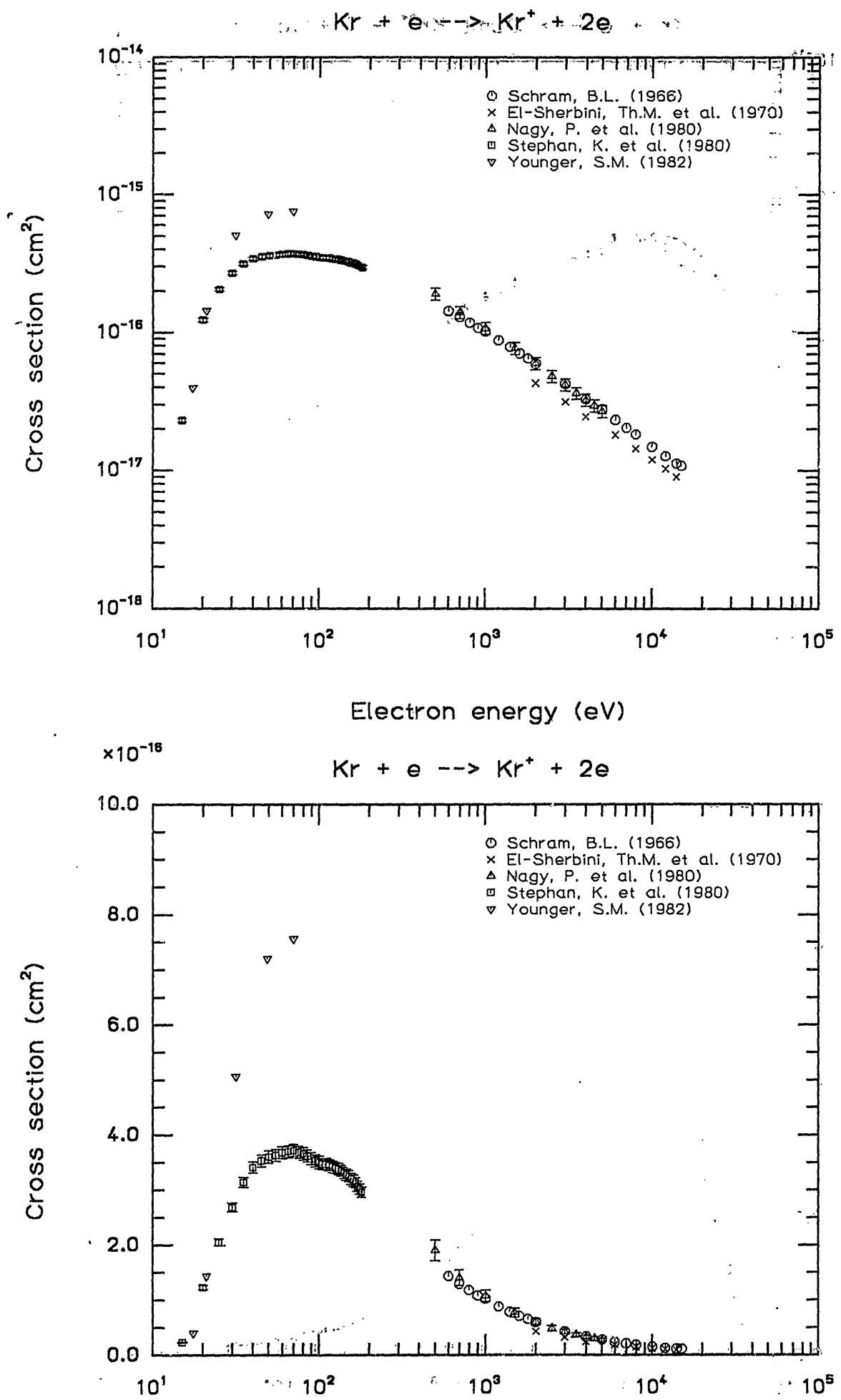
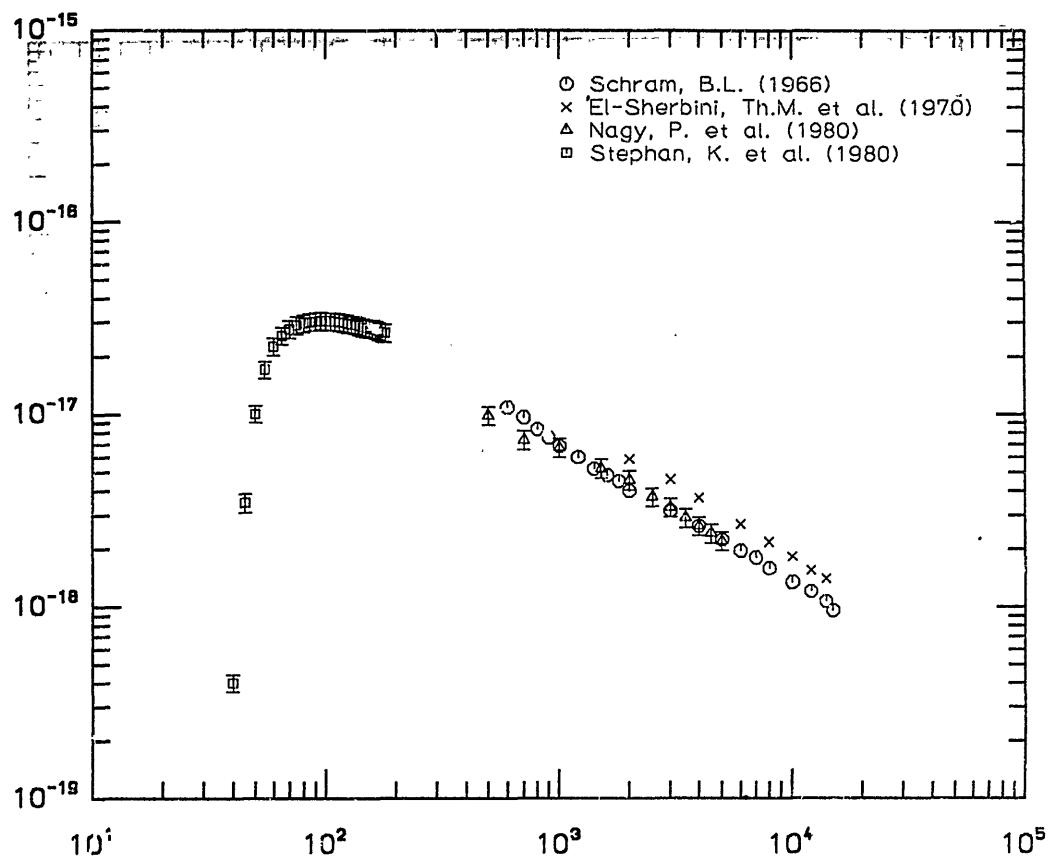


Fig. 186 (v) Electron energy (eV) 281 281



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

Cross section ( $\text{cm}^2$ )

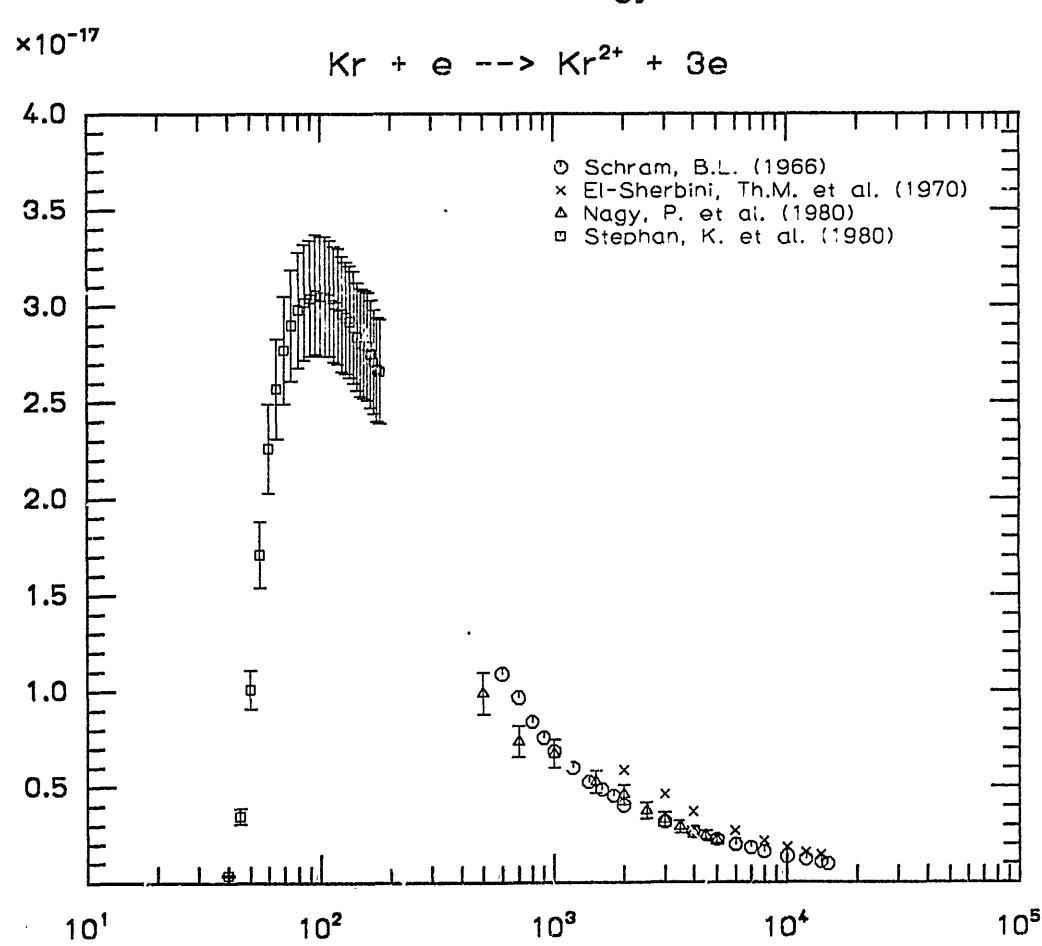


Fig. 187

Electron energy (eV)

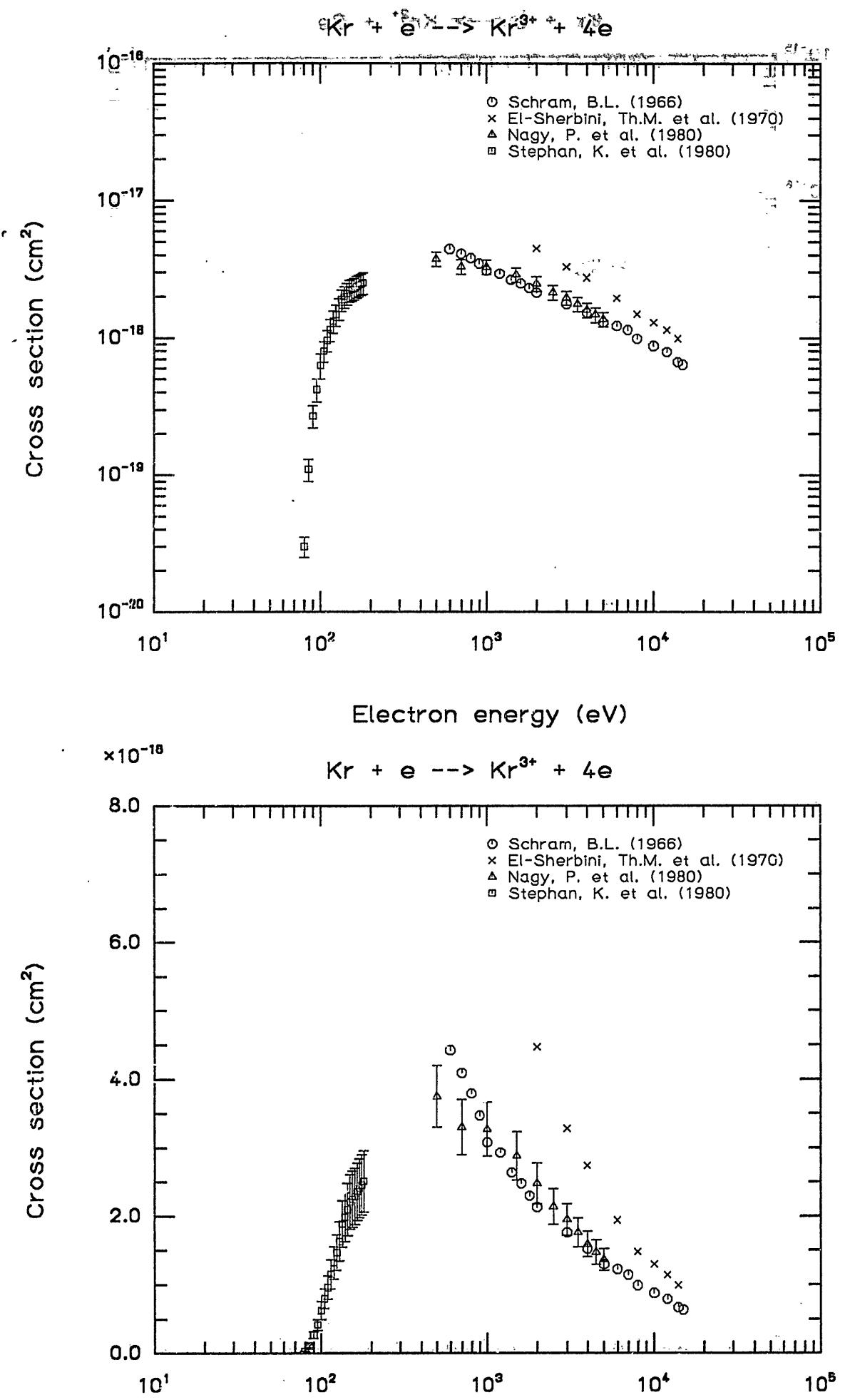
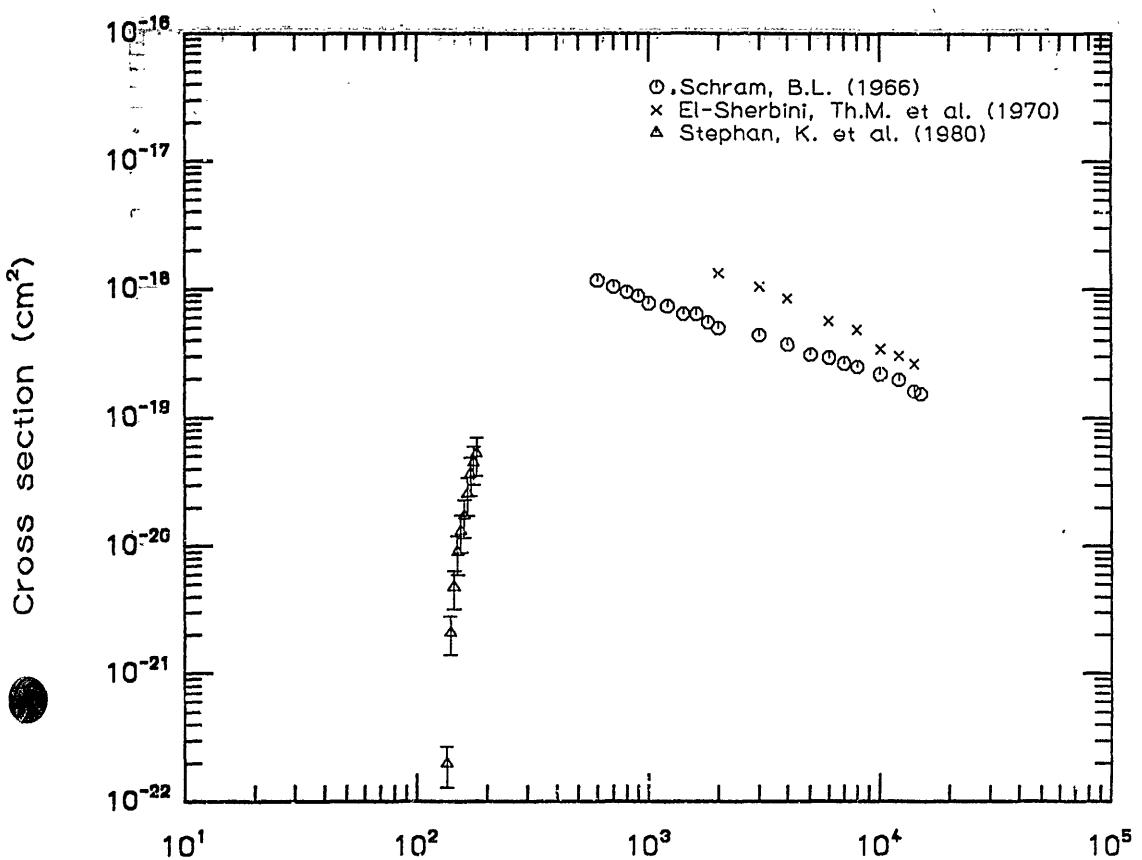
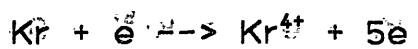


Fig. 188 Electron energy (eV) 18.1.01



Electron energy (eV)

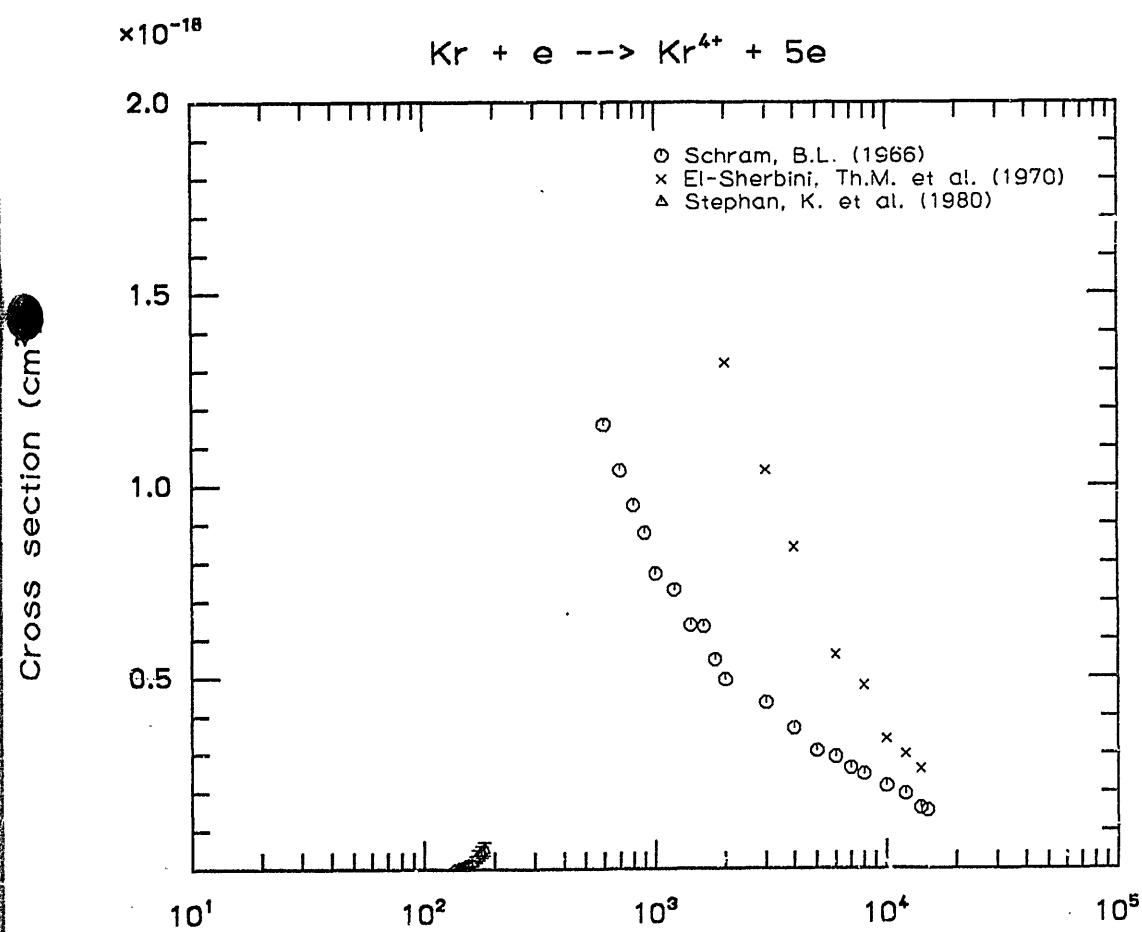


Fig. 189

Electron energy (eV)

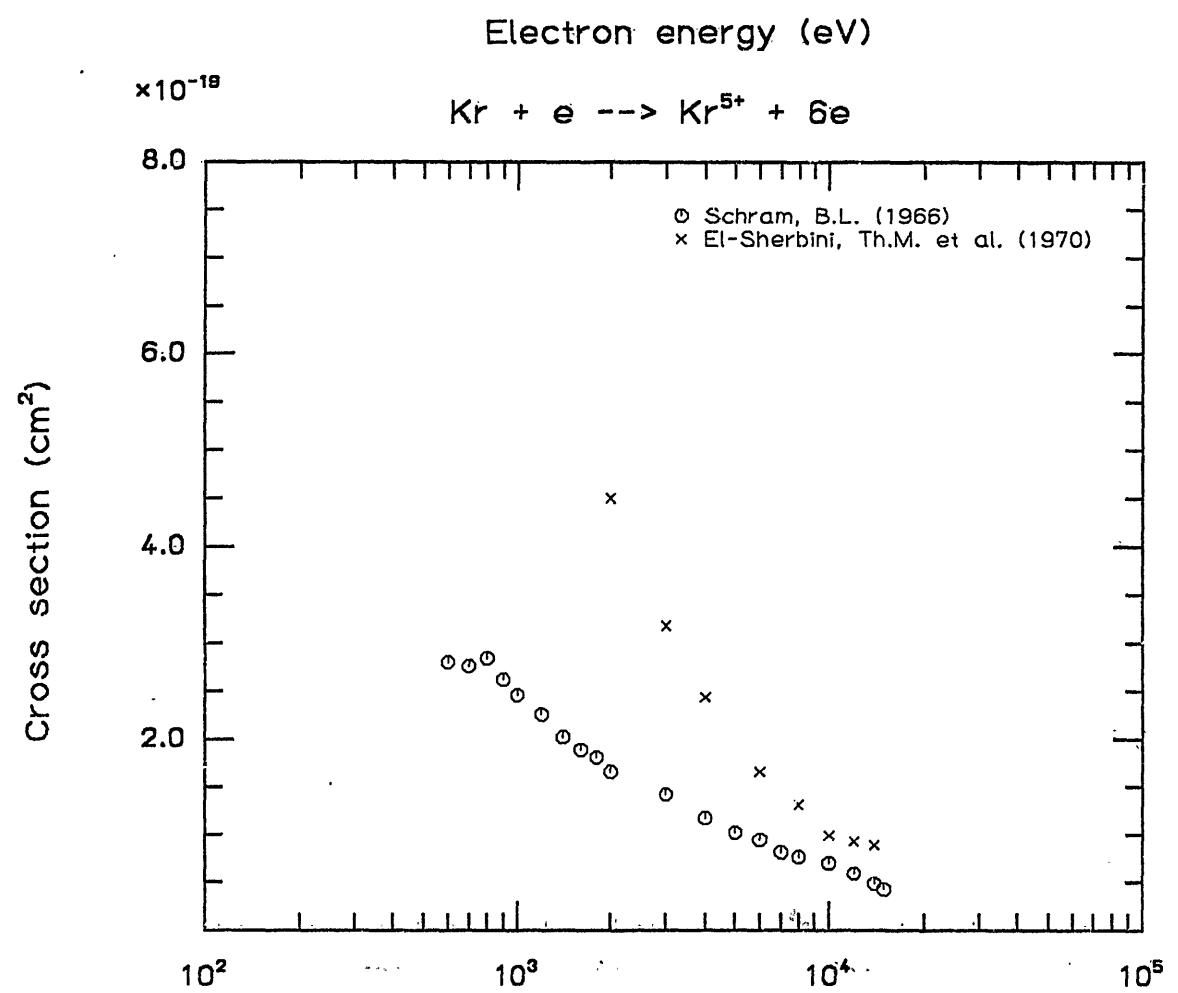
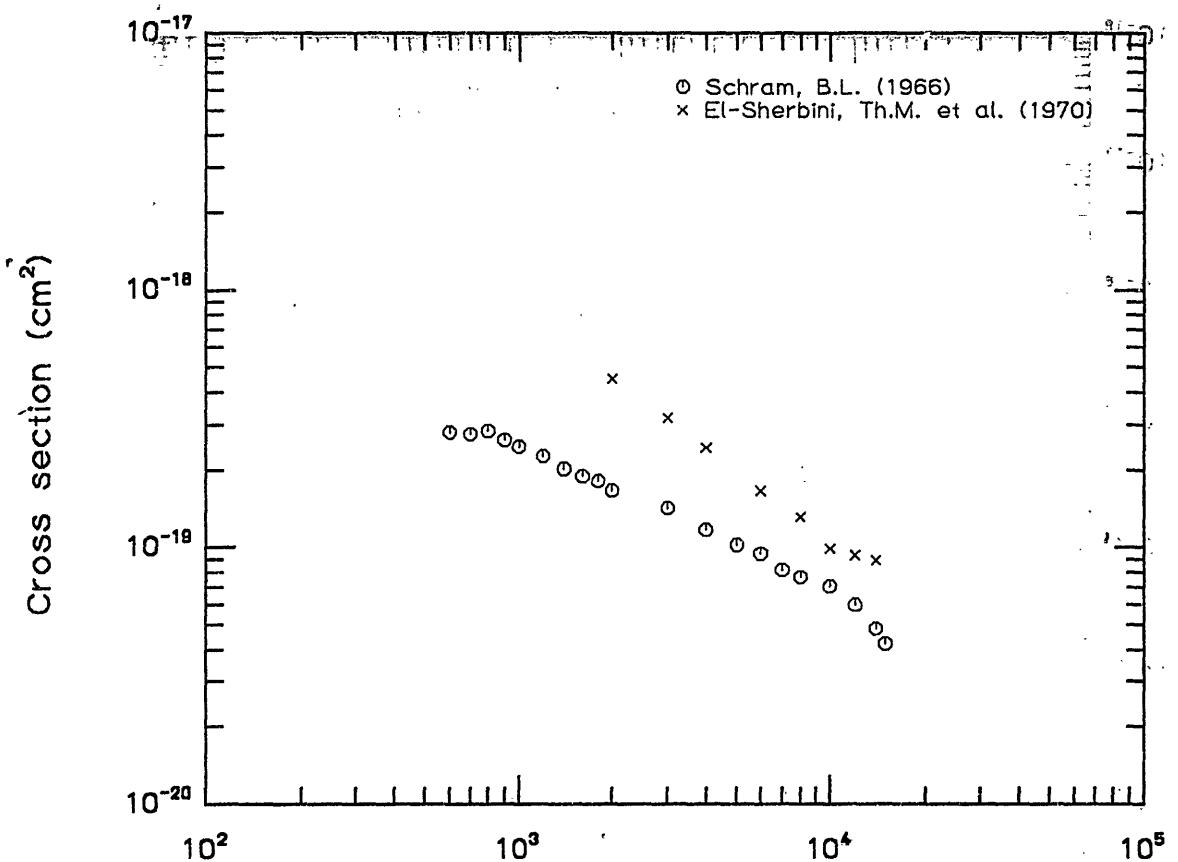
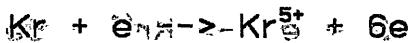
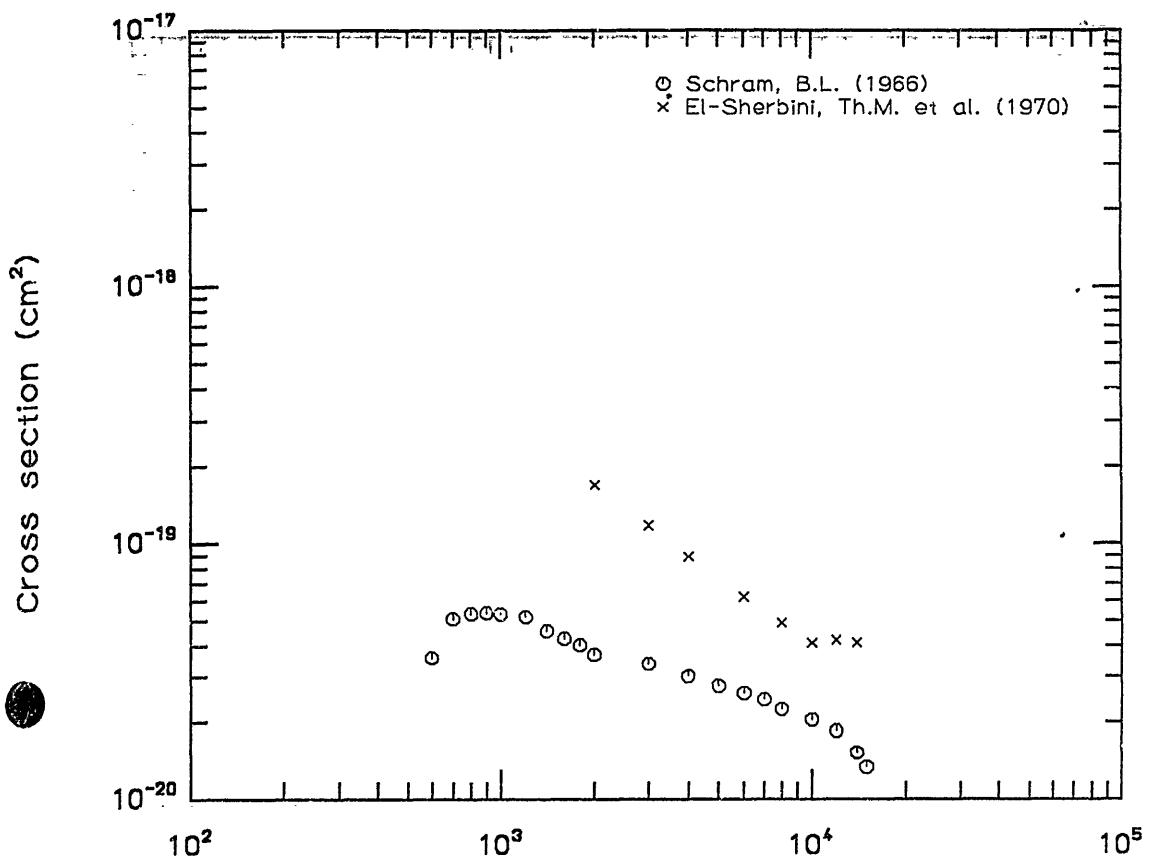


Fig. 190 Electron energy (eV) 281 p.F



Electron energy (eV)

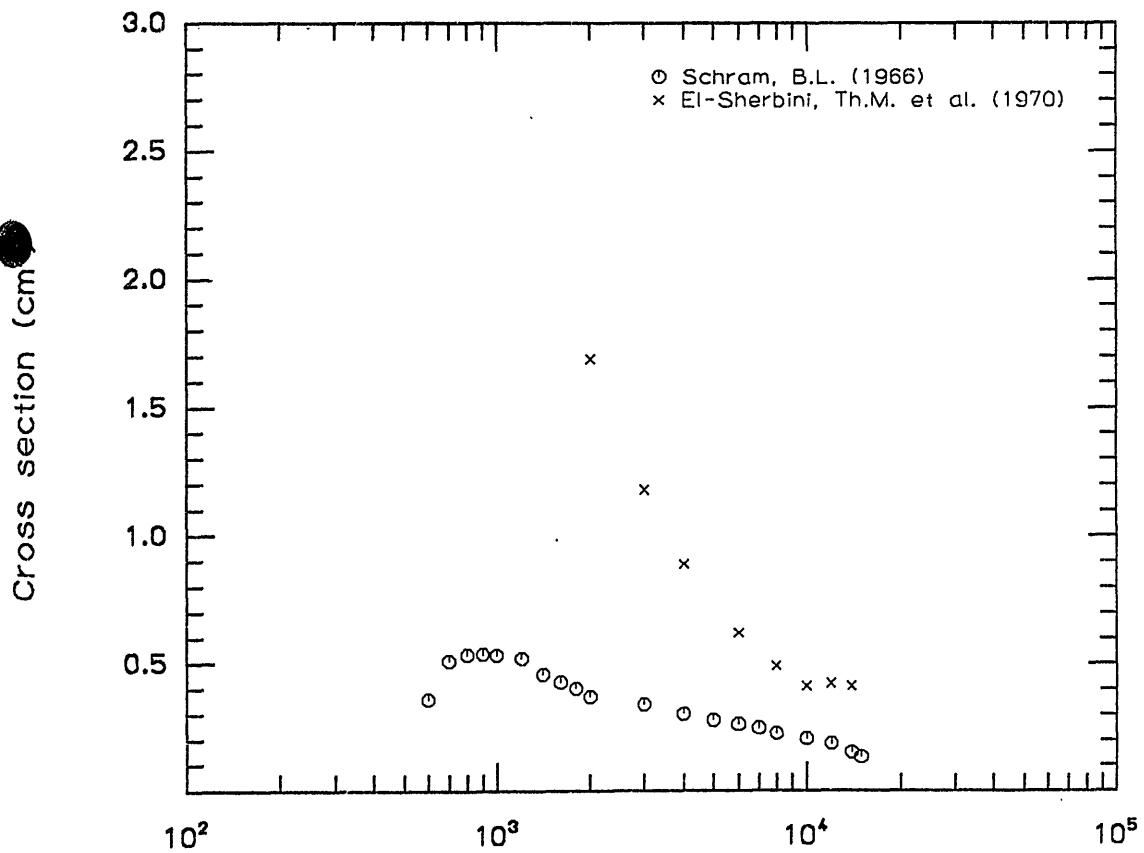
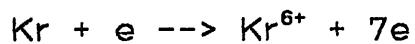


Fig. 191      Electron energy (eV)

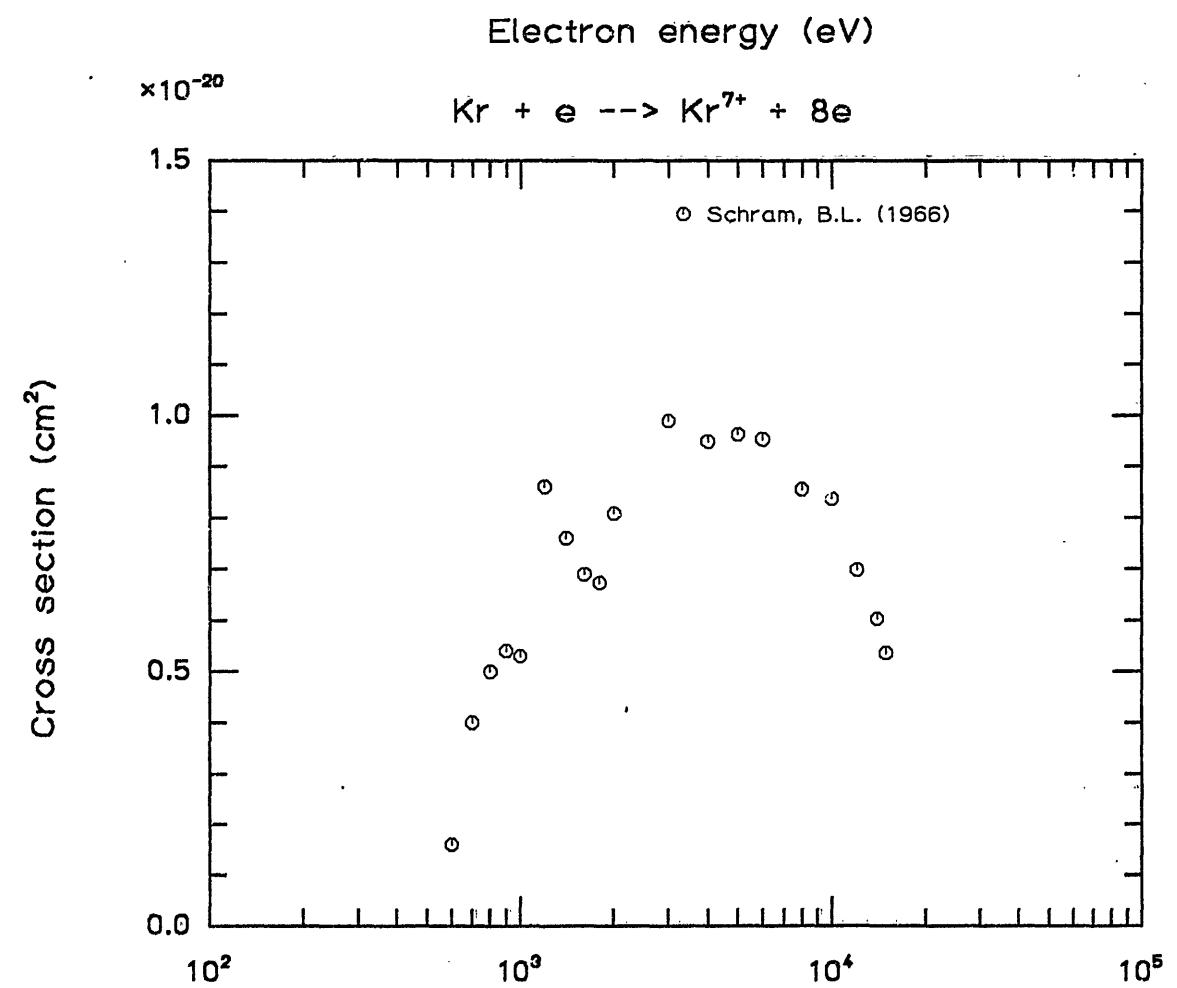
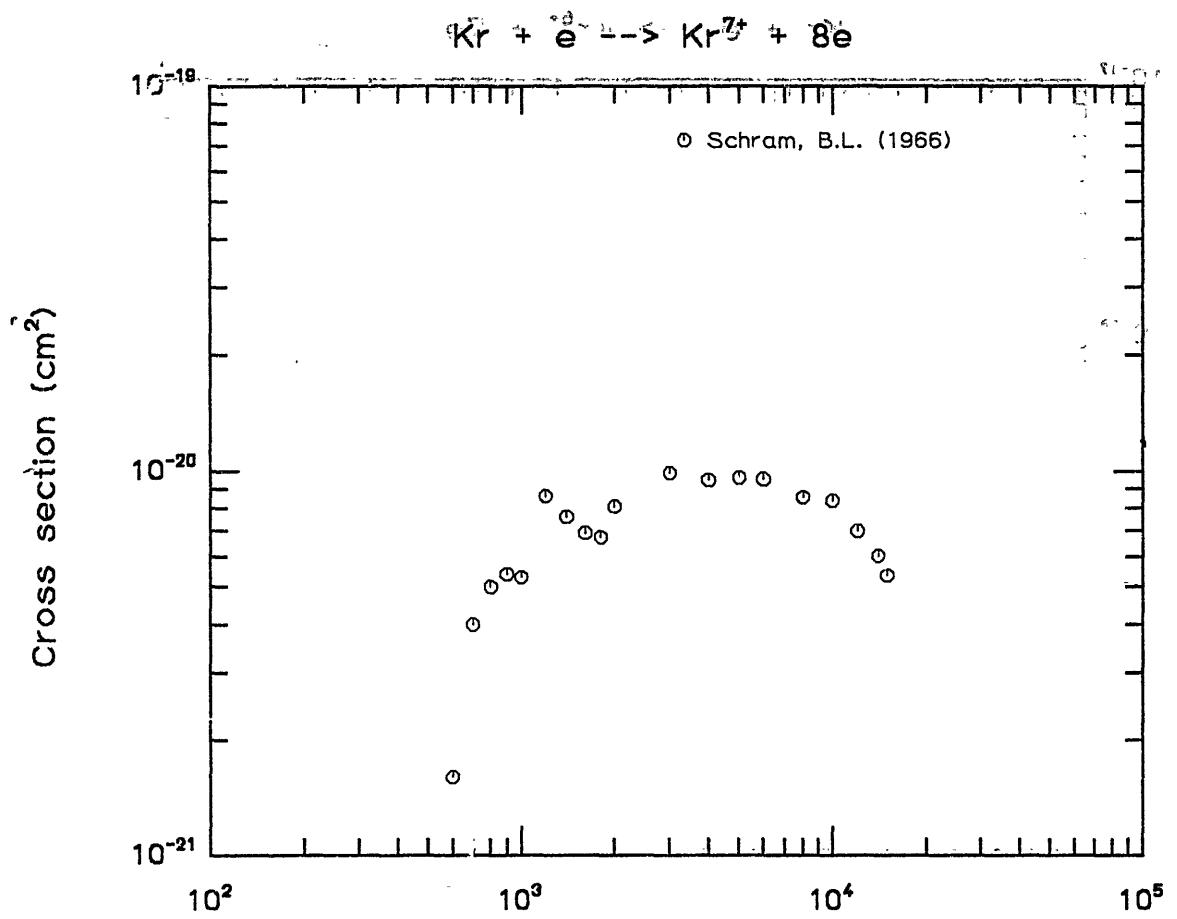
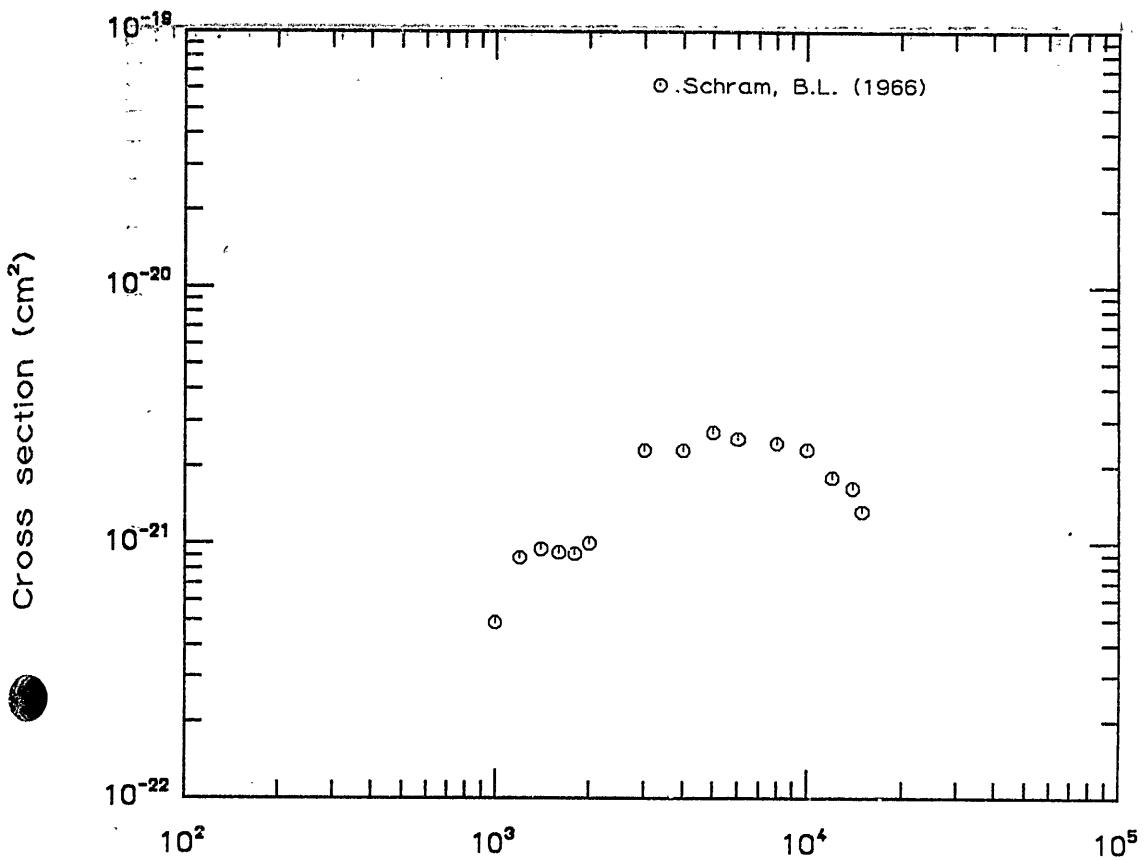
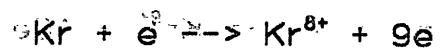


Fig. 192      → Electron energy (eV)      Ref. 193



Electron energy (eV)

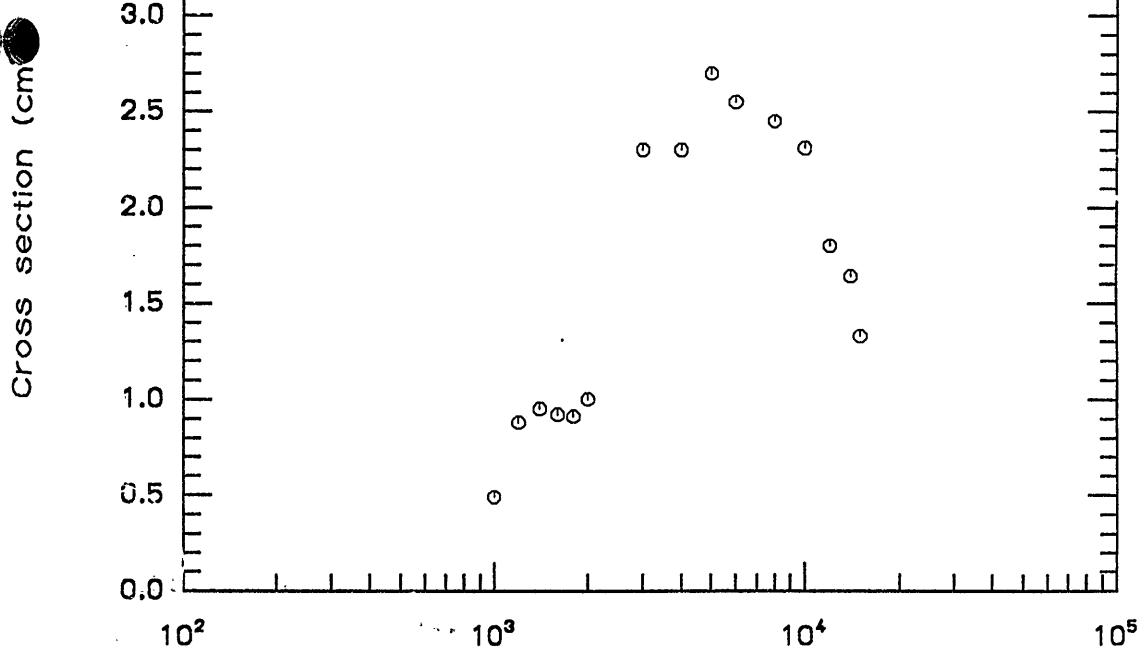
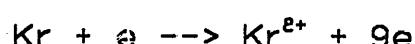


Fig. 193      Electron energy (eV)

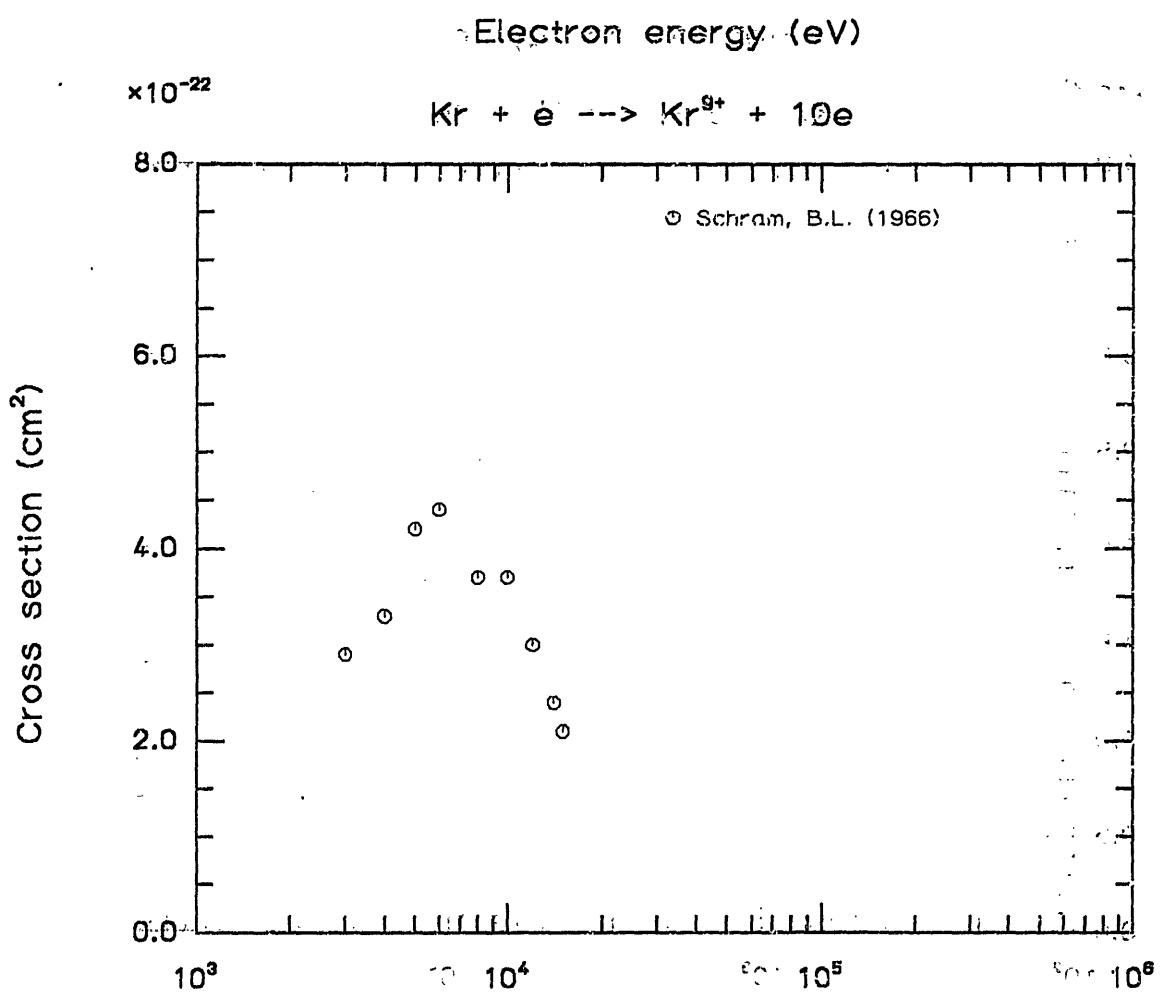
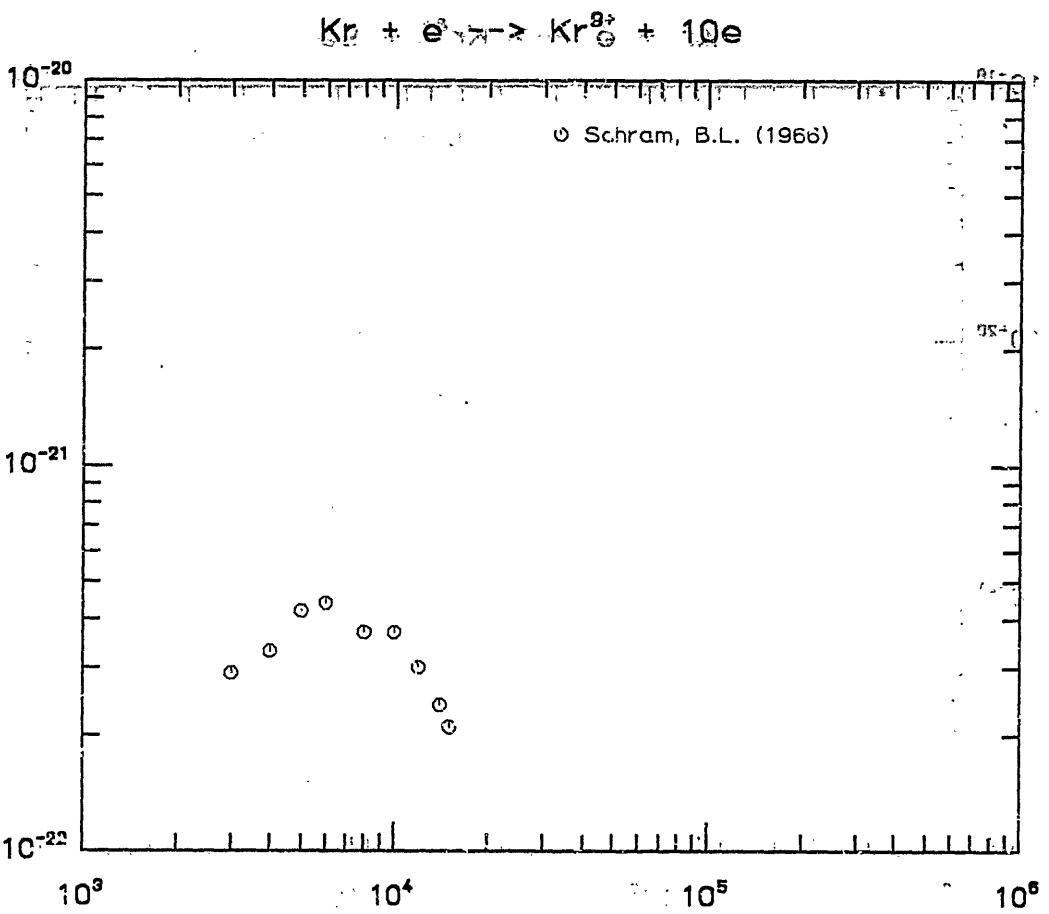
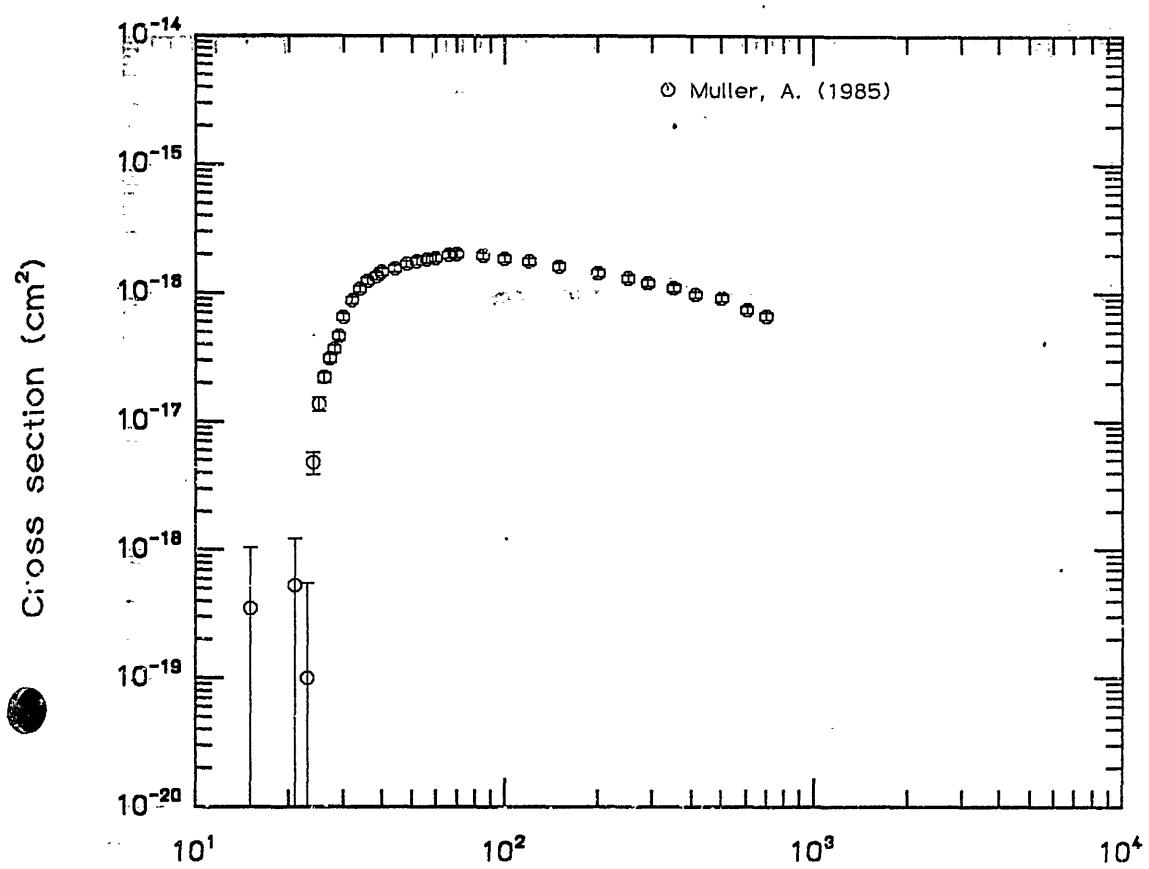
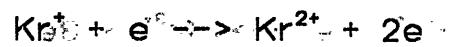


Fig. 194 (a) Electron energy (eV) 801 p14



Electron energy (eV)

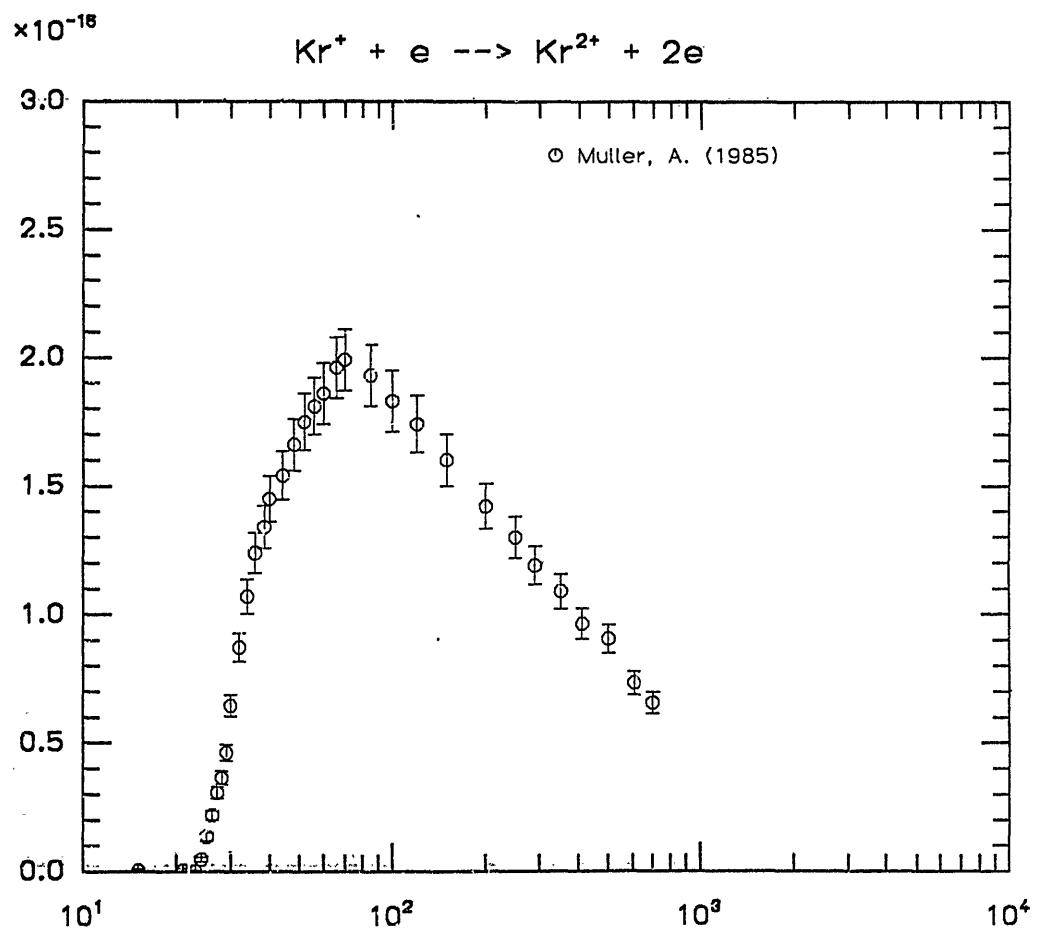


Fig. 195      Électron energy (eV)

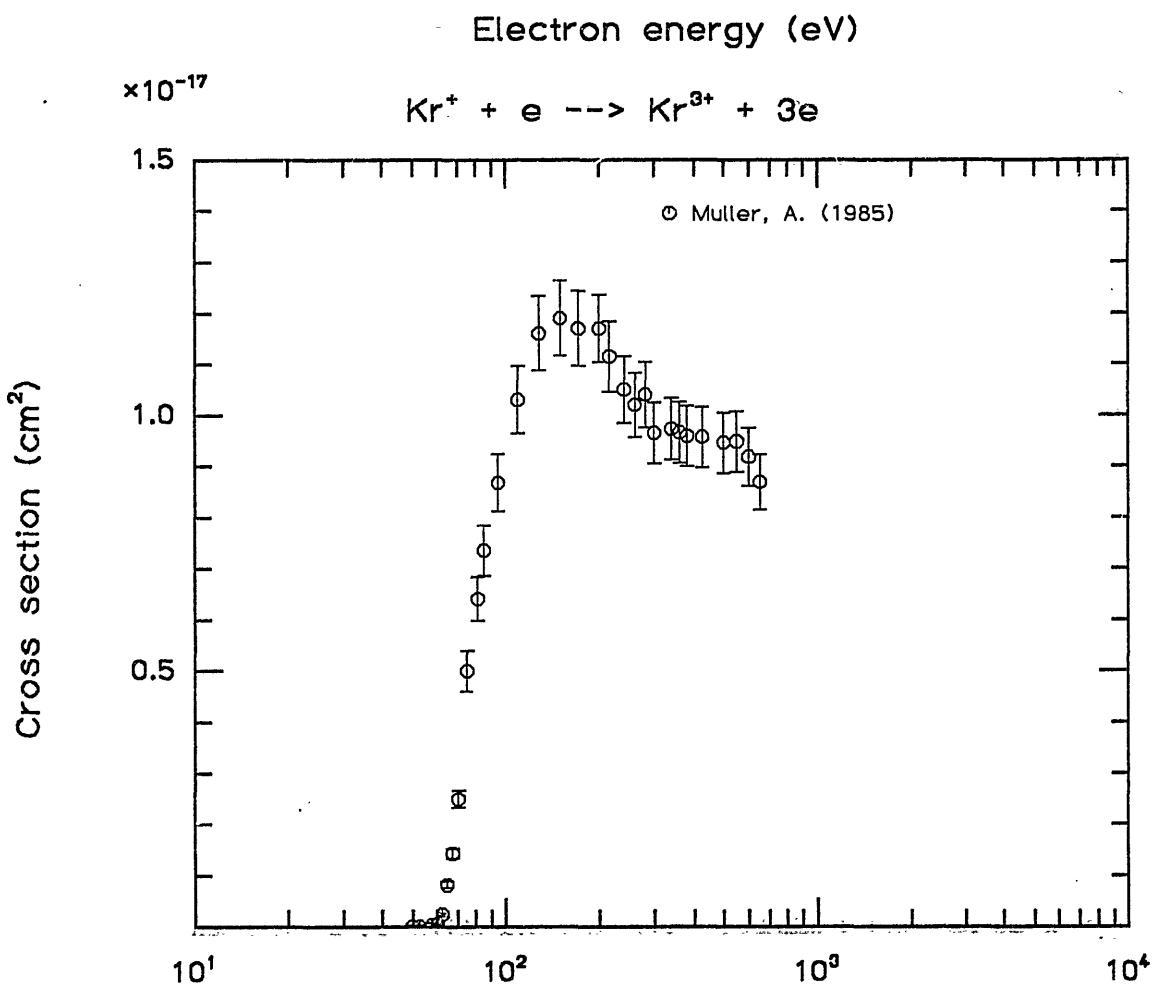
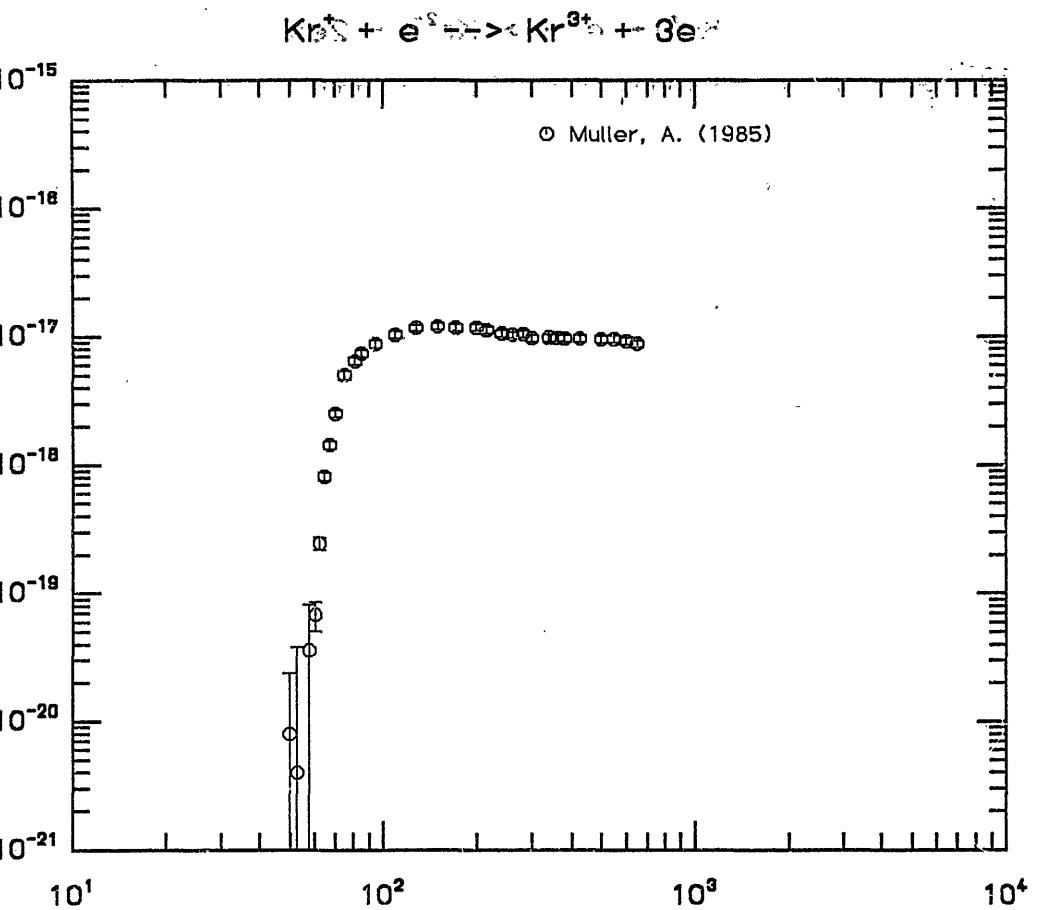


Fig. 196

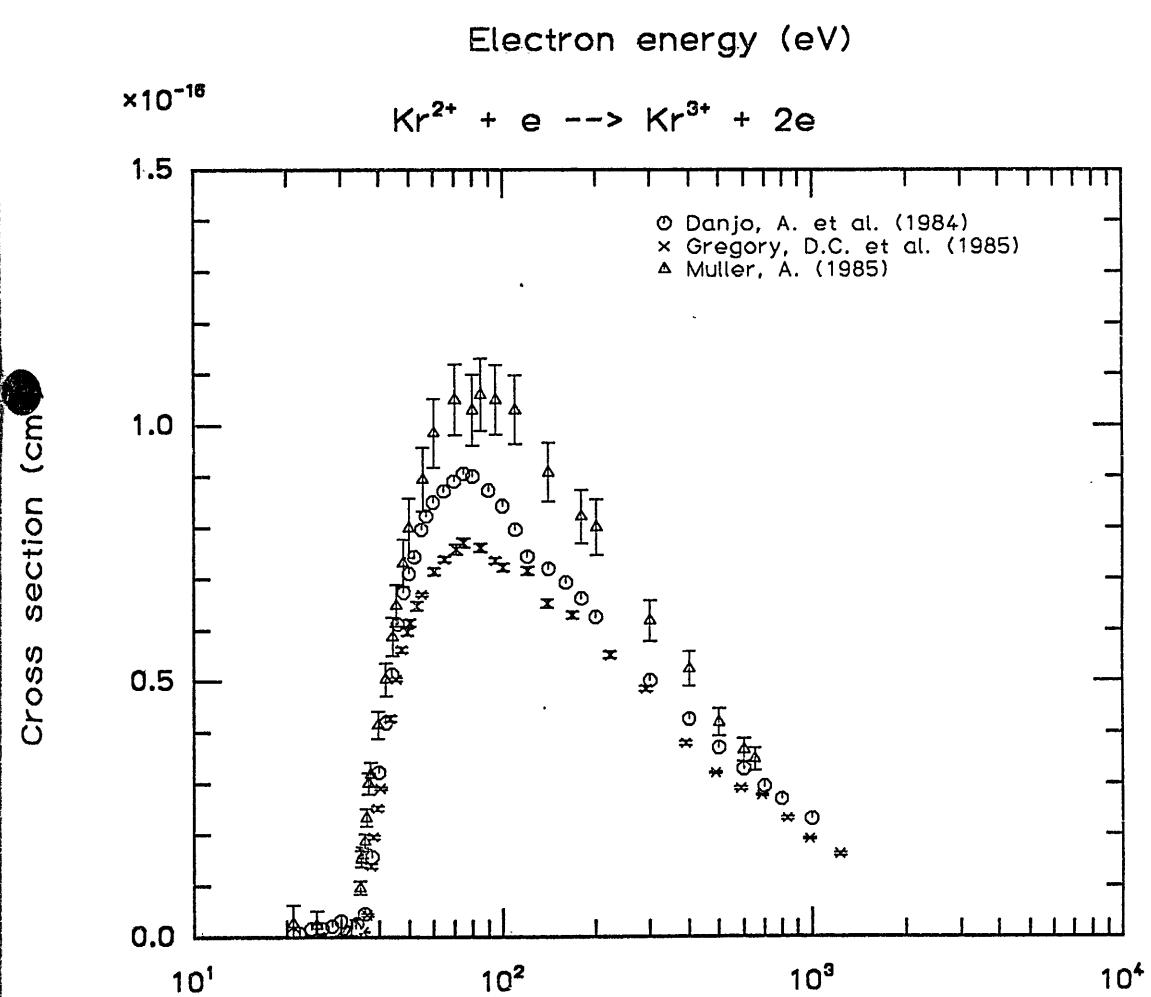
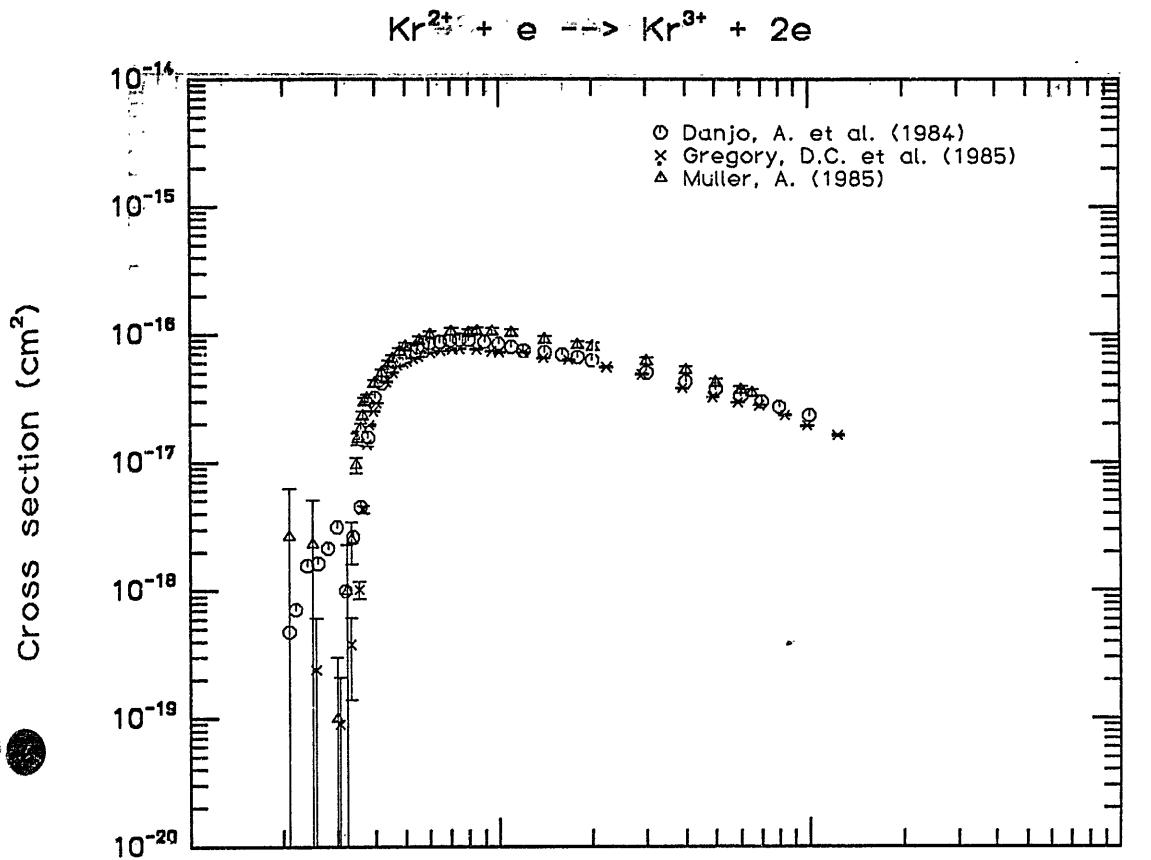


Fig. 197      Electron energy (eV)

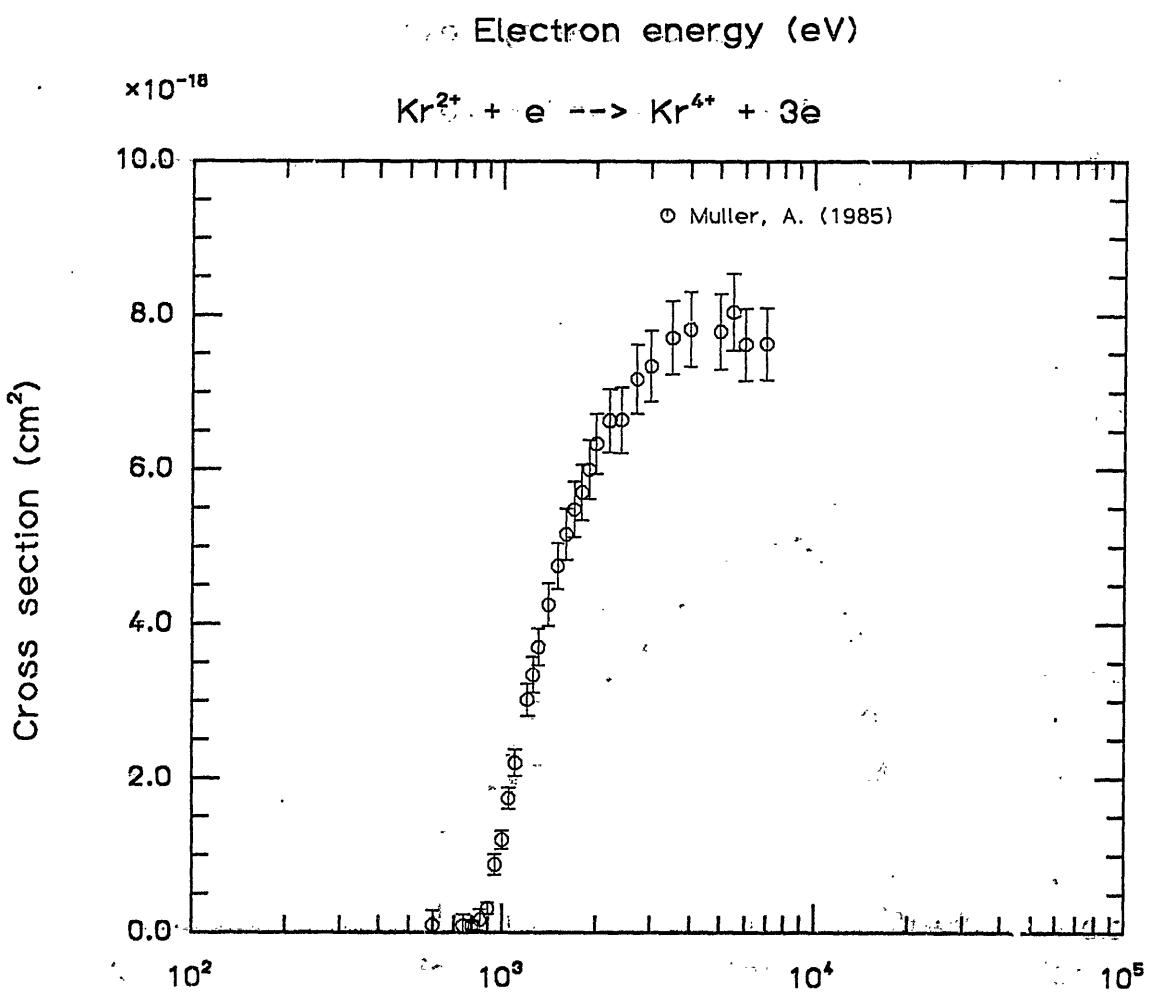
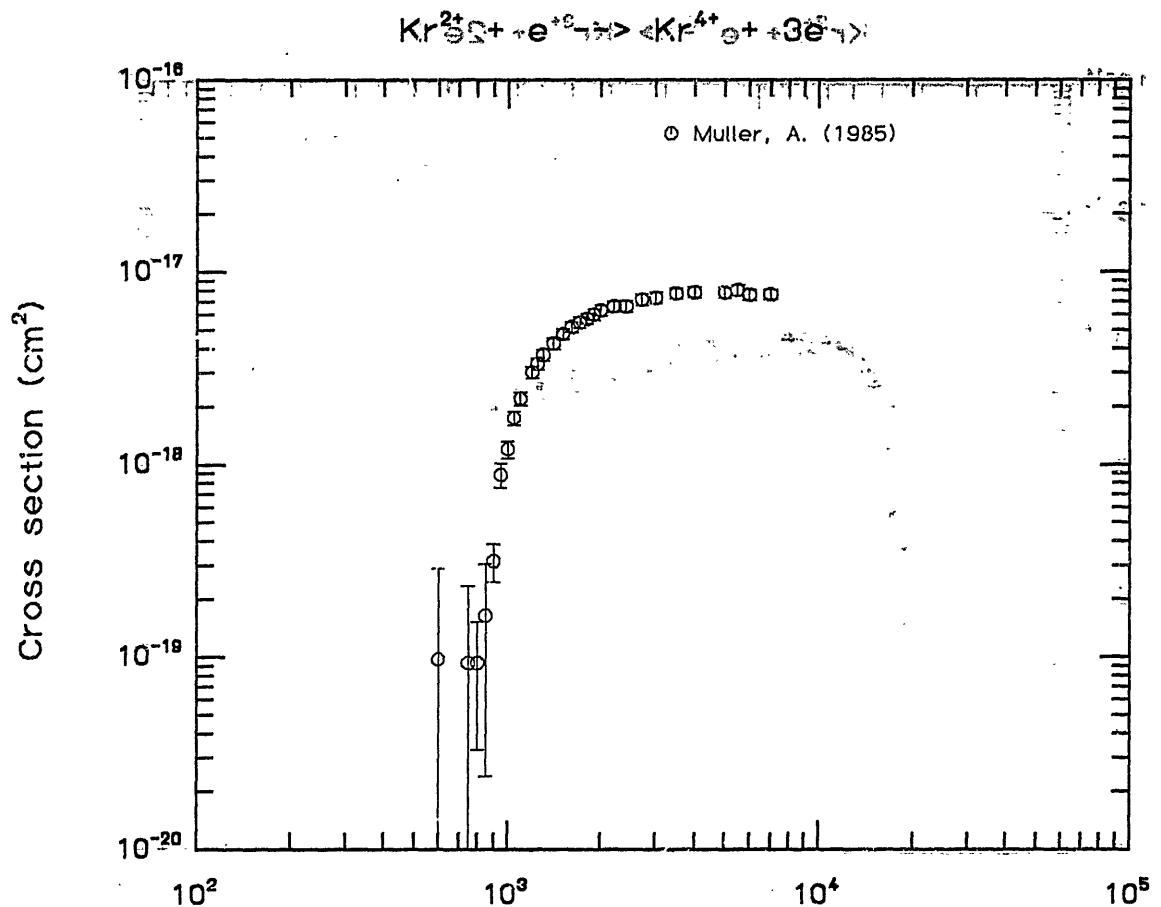
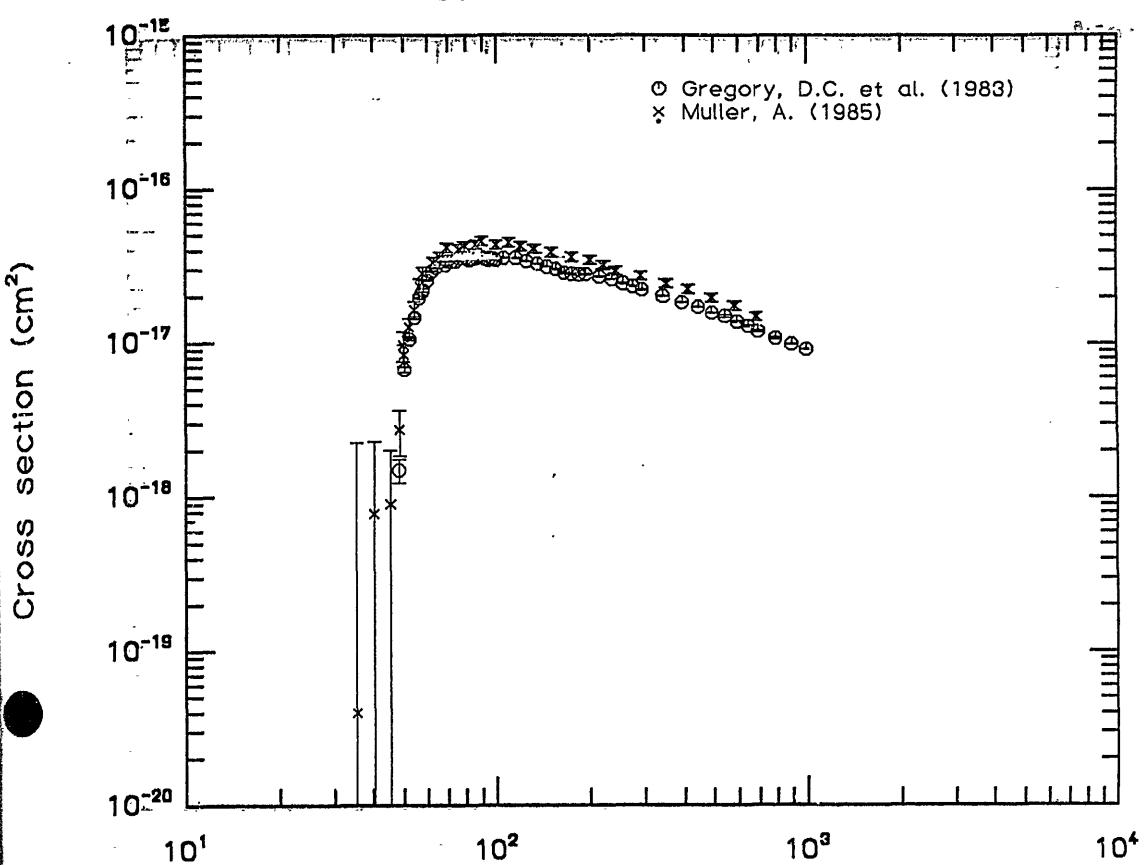
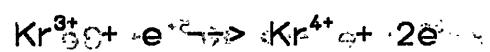


Fig. 198       $\text{Kr}^{2+} + \text{e}^{-} \rightarrow \text{Kr}^{4+} + 3\text{e}^{-}$



Electron energy (eV)

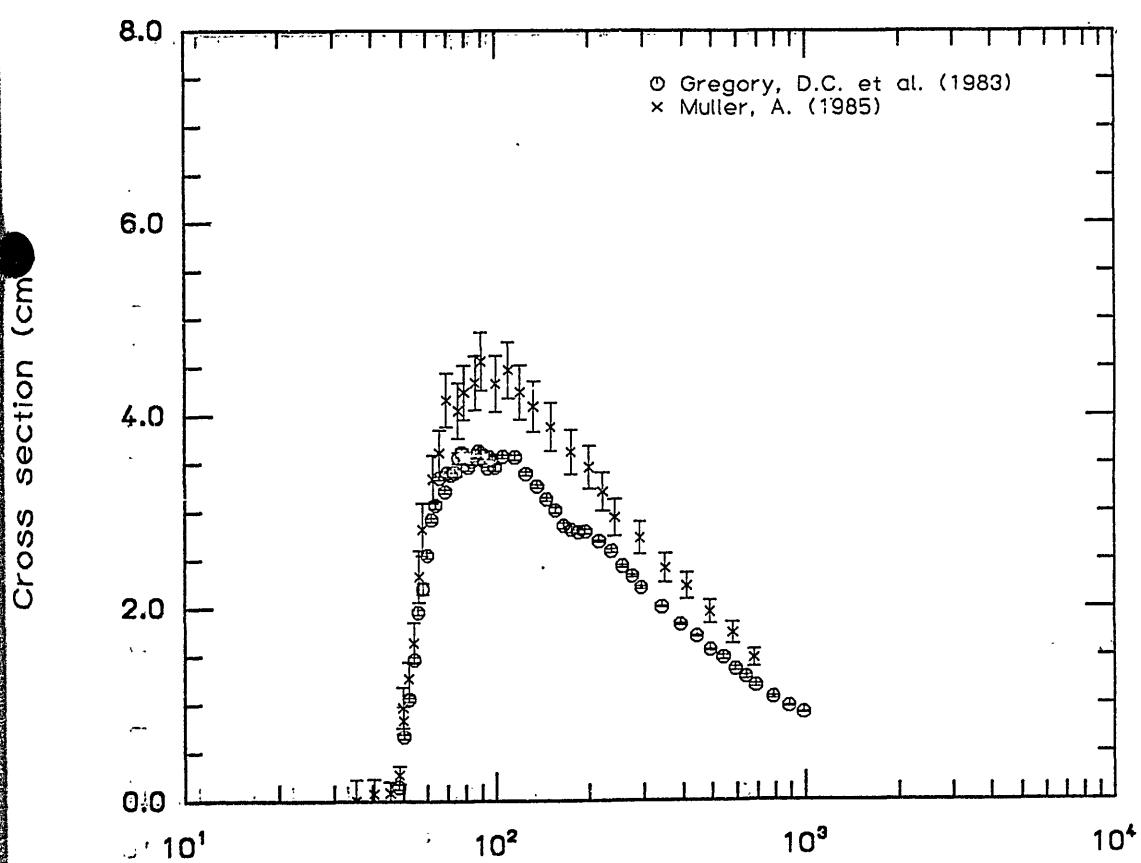
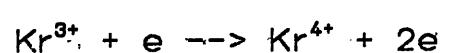


Fig. 199      Electron energy (eV)

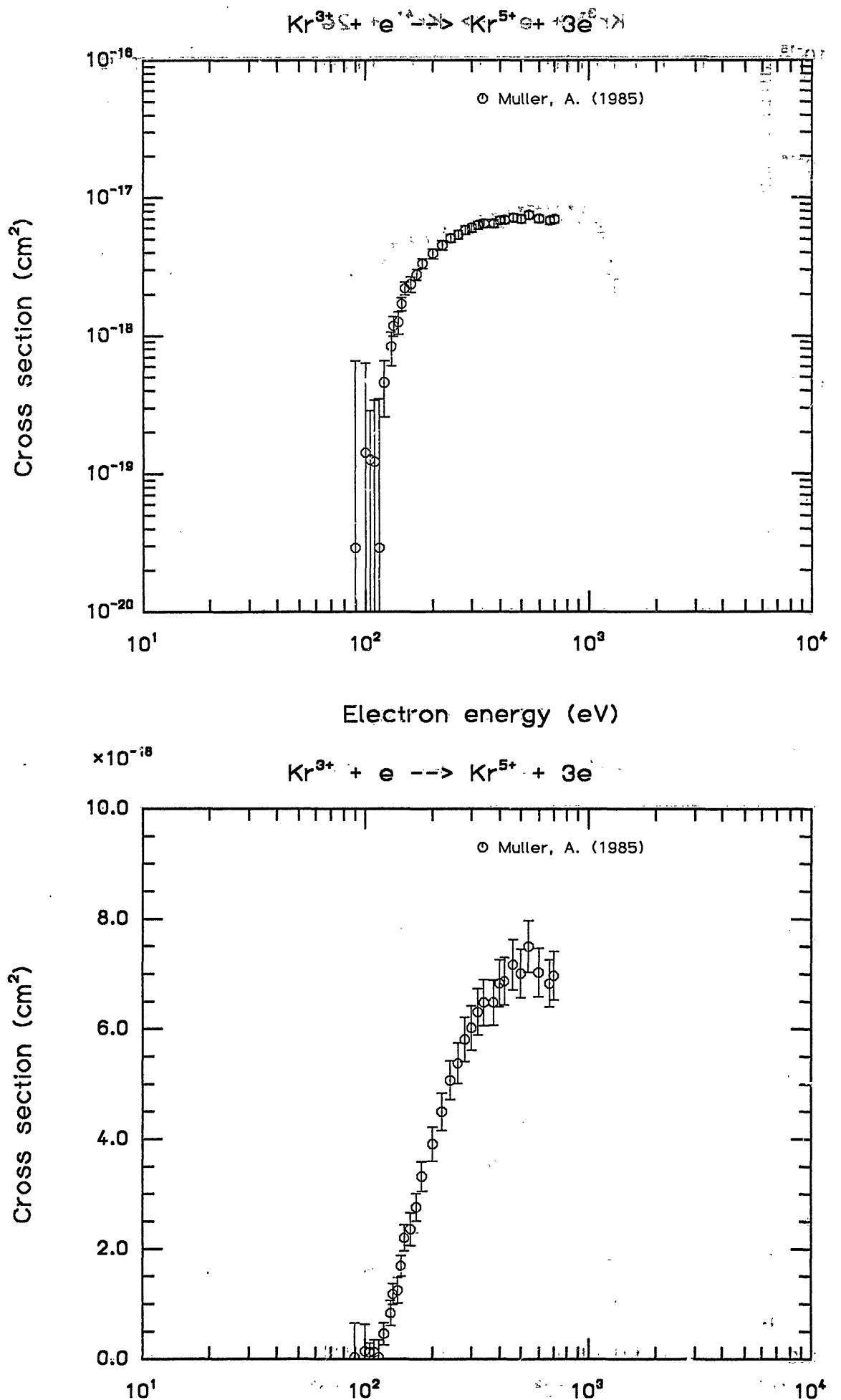
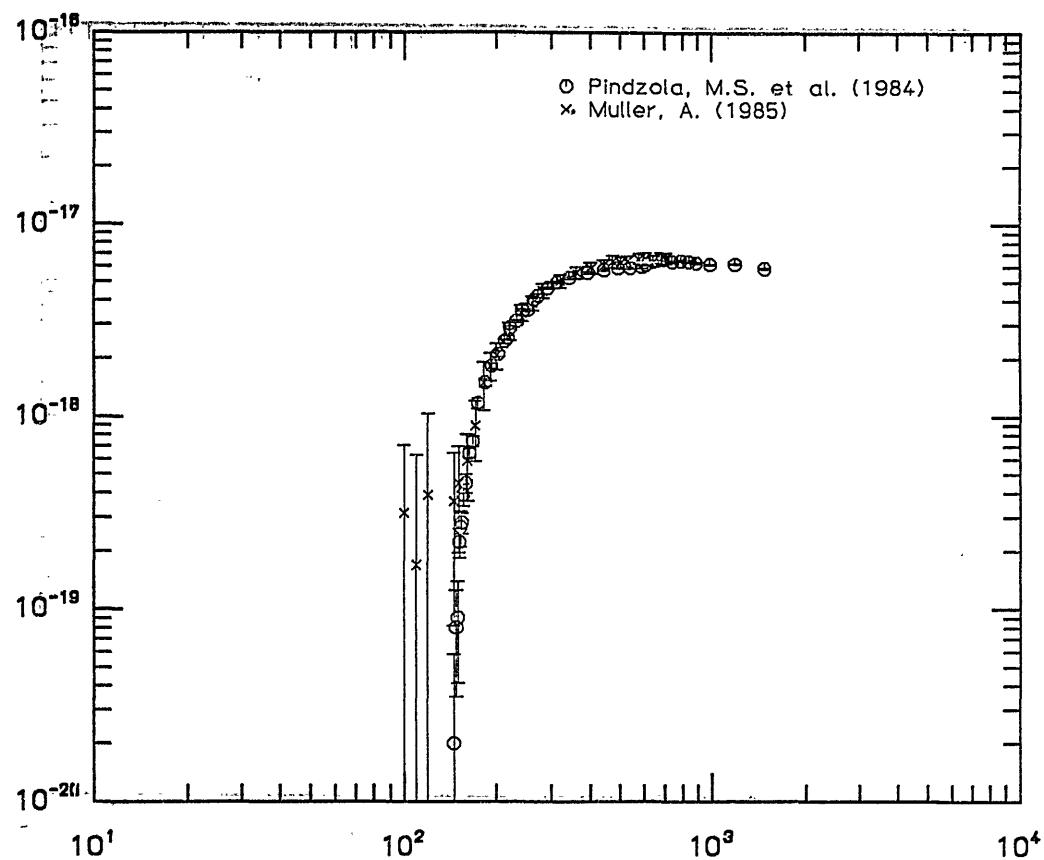


Fig. 200 Electron energy (eV)  $\times 10^{-18}$   $\text{cm}^2$



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

Cross section ( $\text{cm}^2$ )

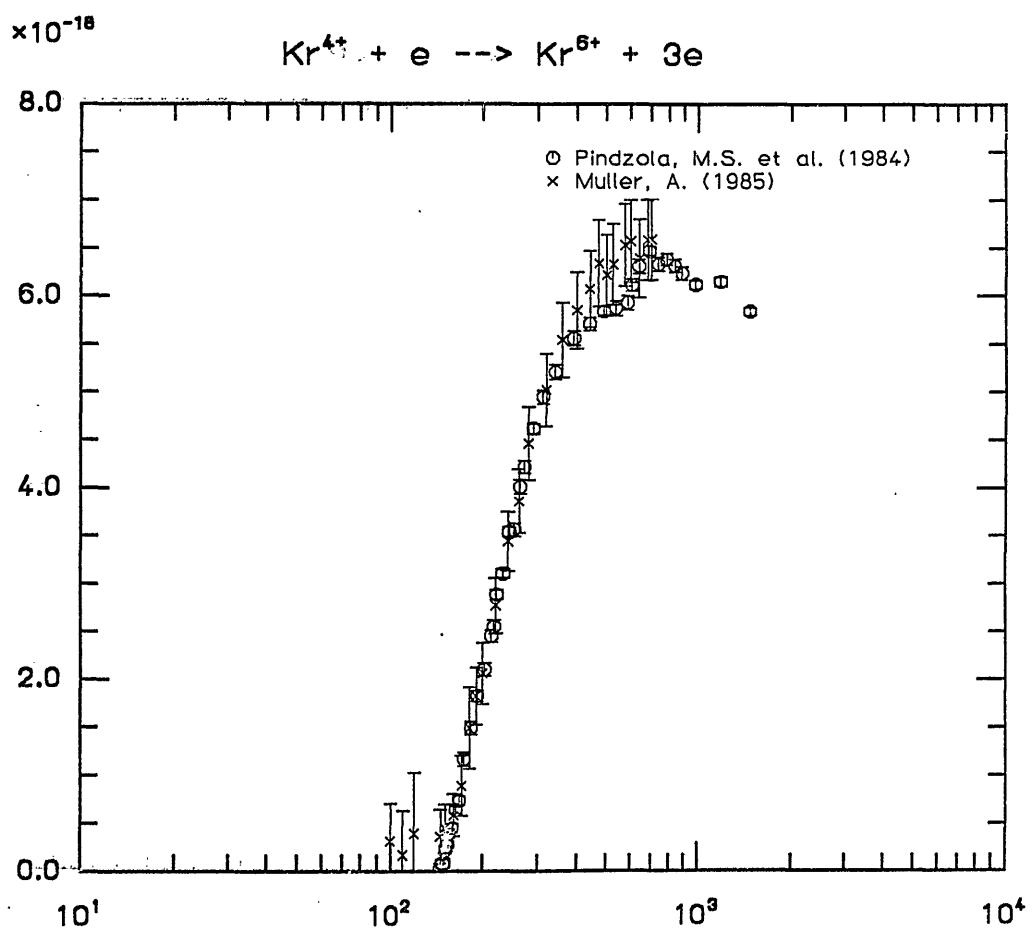


Fig. 201 Electron energy (eV)

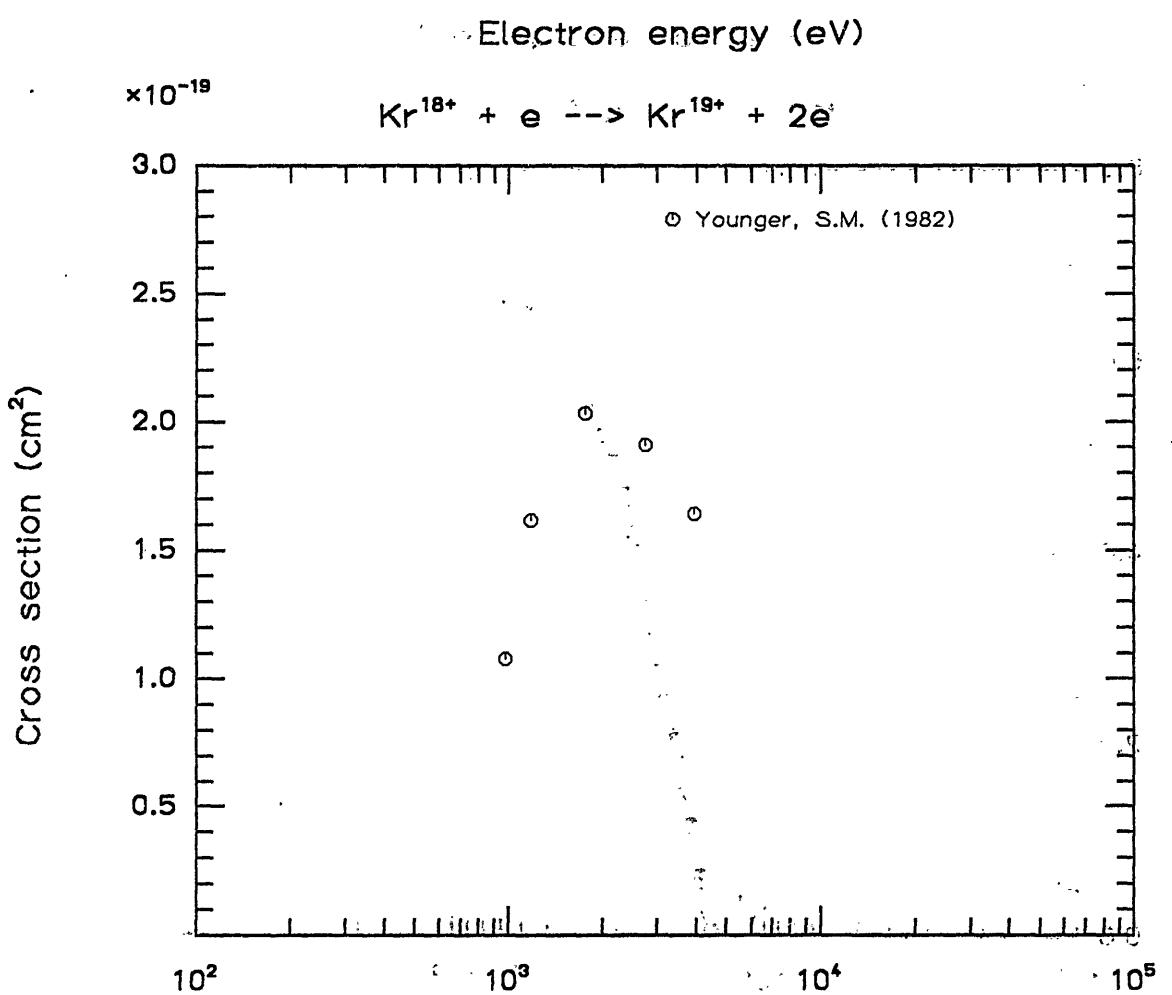
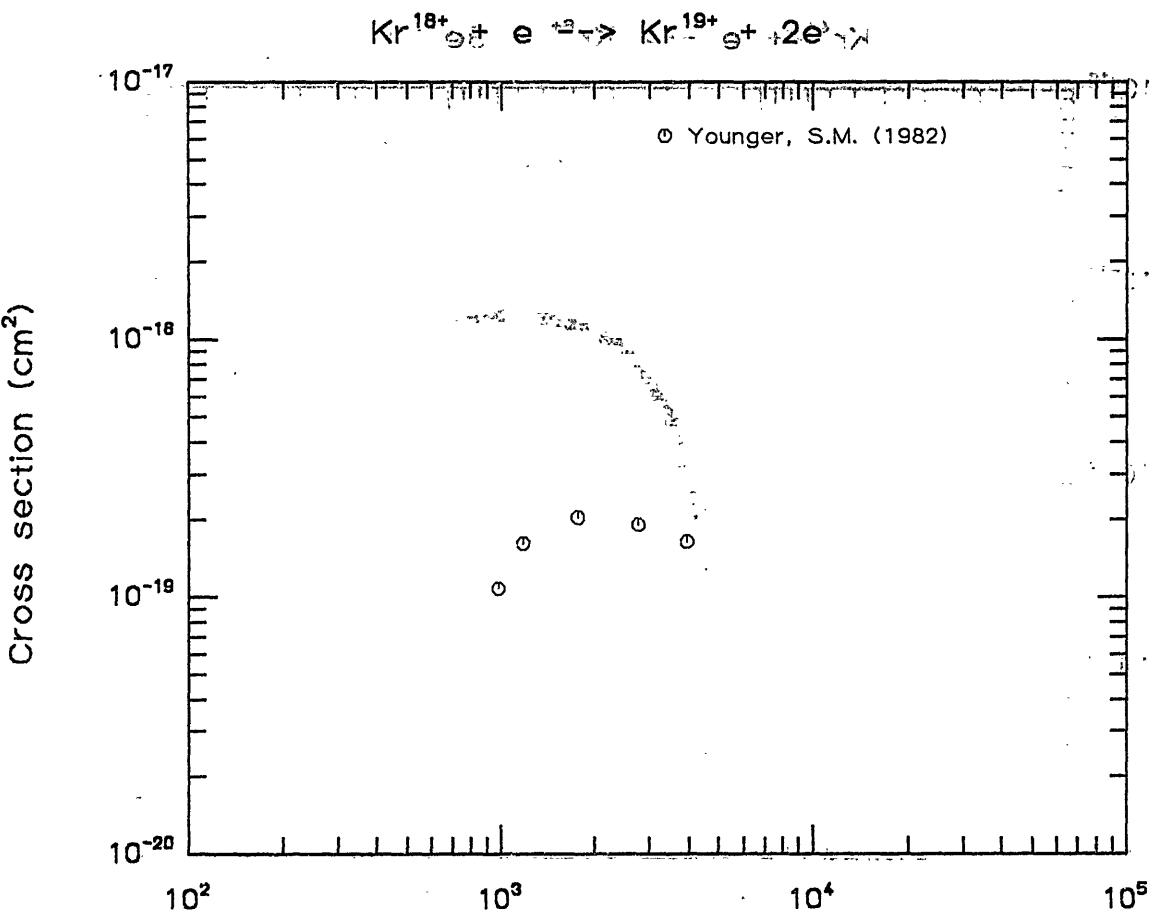


Fig. 202 (a) Electron energy (eV) Fig. 202 (b)

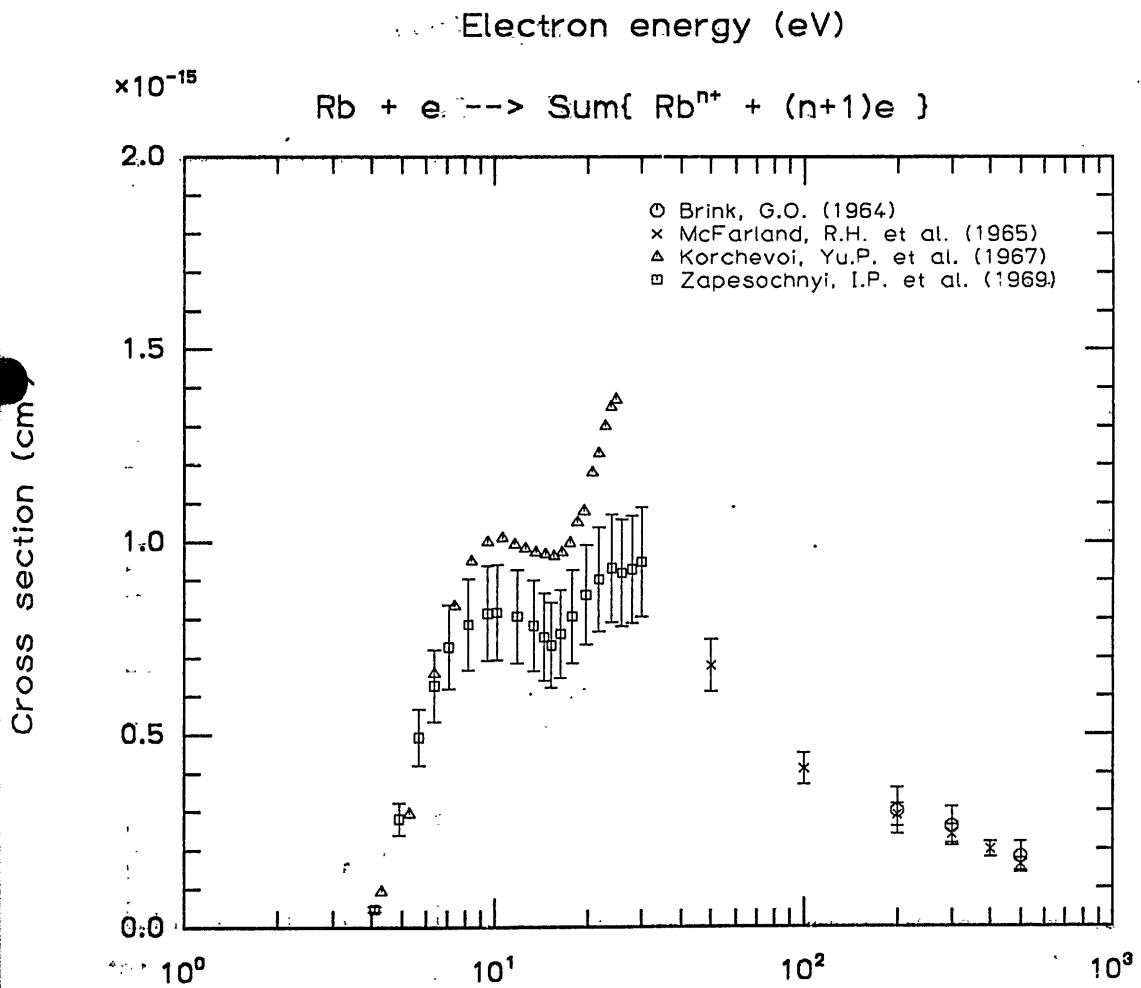
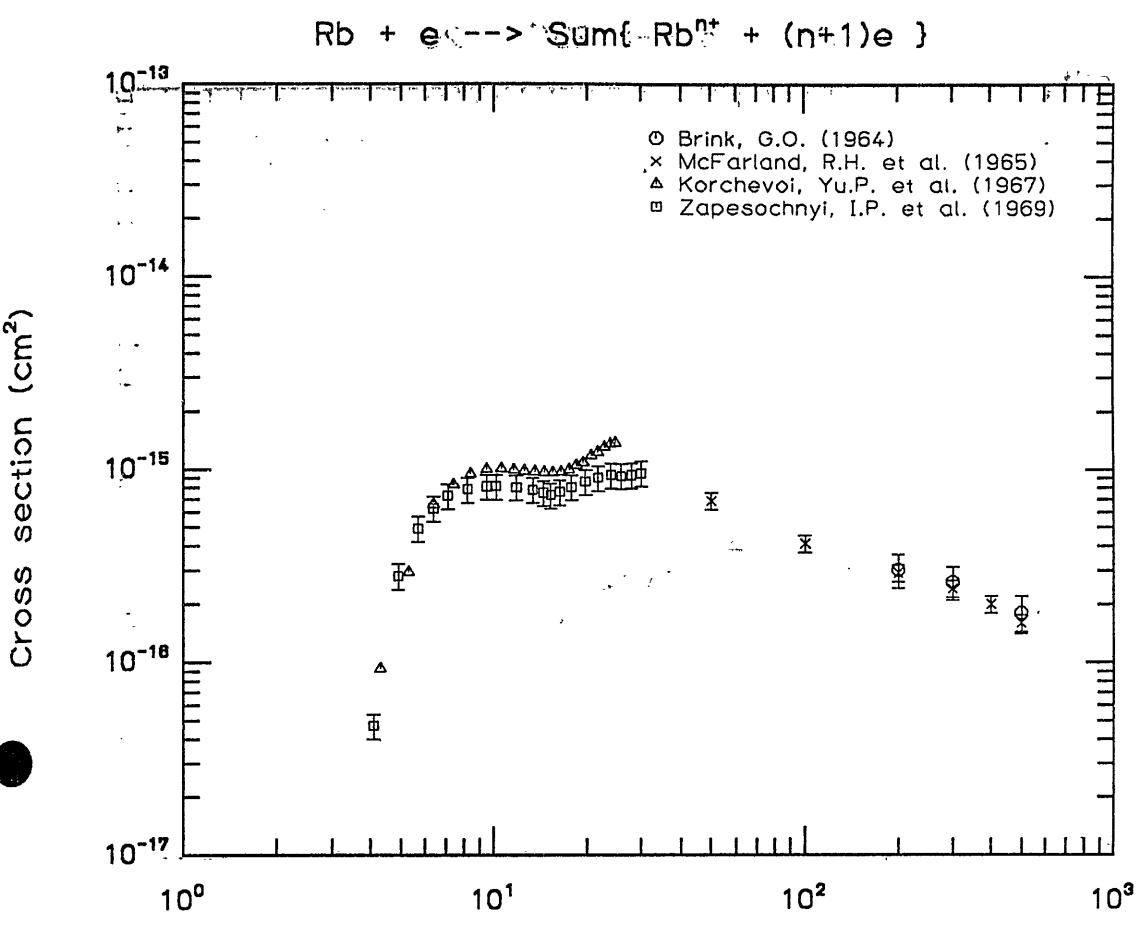


Fig. 203

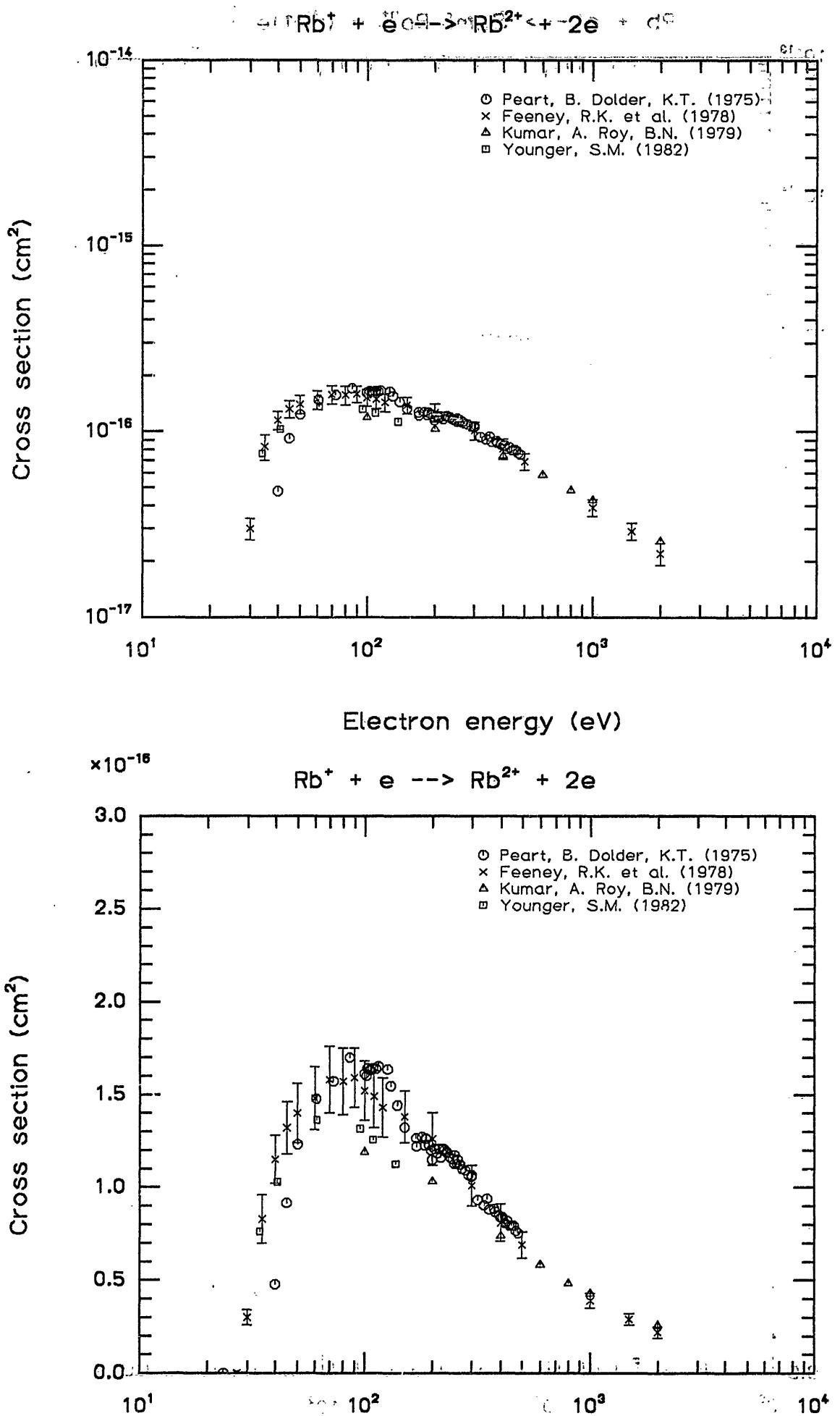
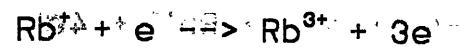
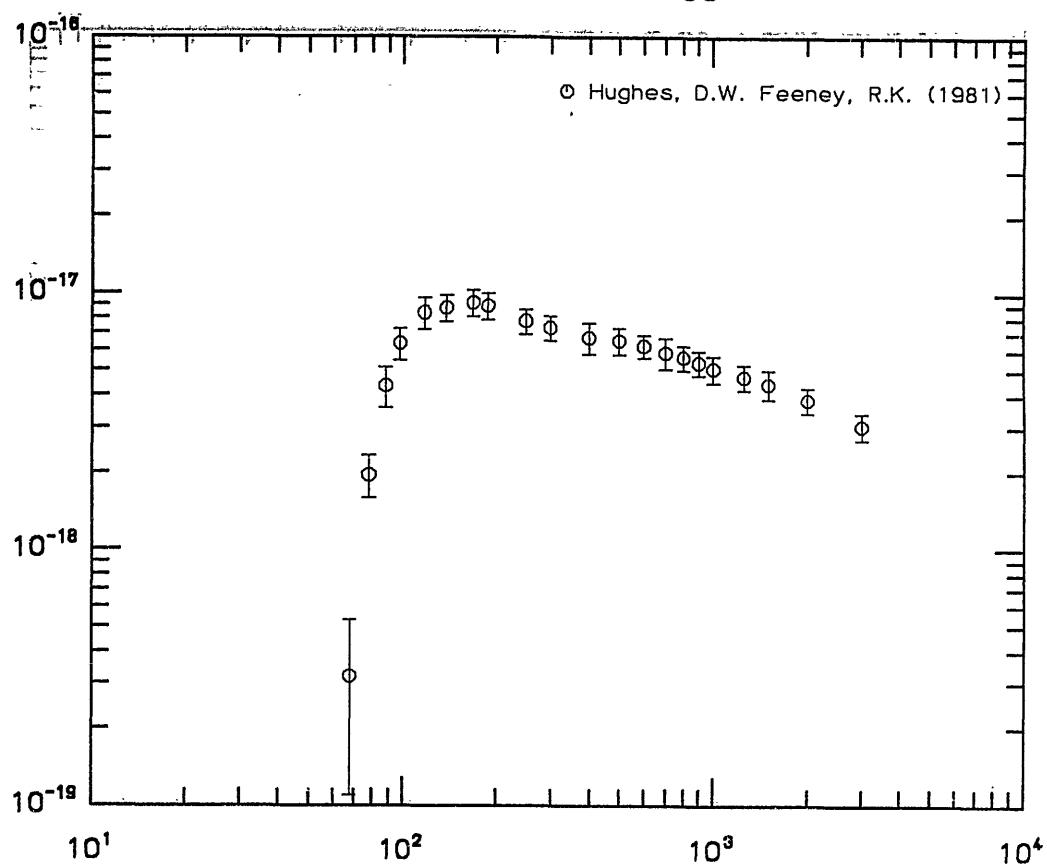


Fig. 204. The Electron energy (eV) COS .prf



Cross Section ( $\text{cm}^2$ )



Electron energy (eV)

Cross section ( $\text{cm}^2$ )

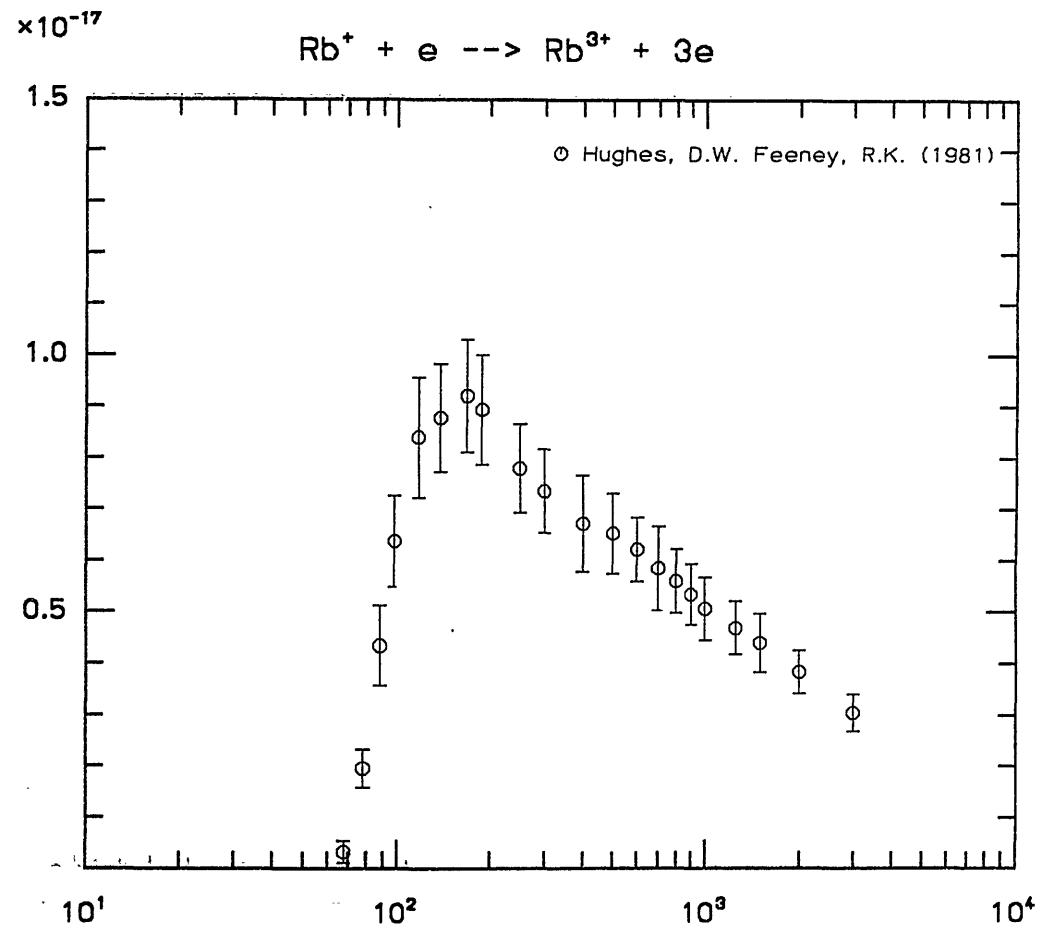


Fig. 205      Electron energy (eV)

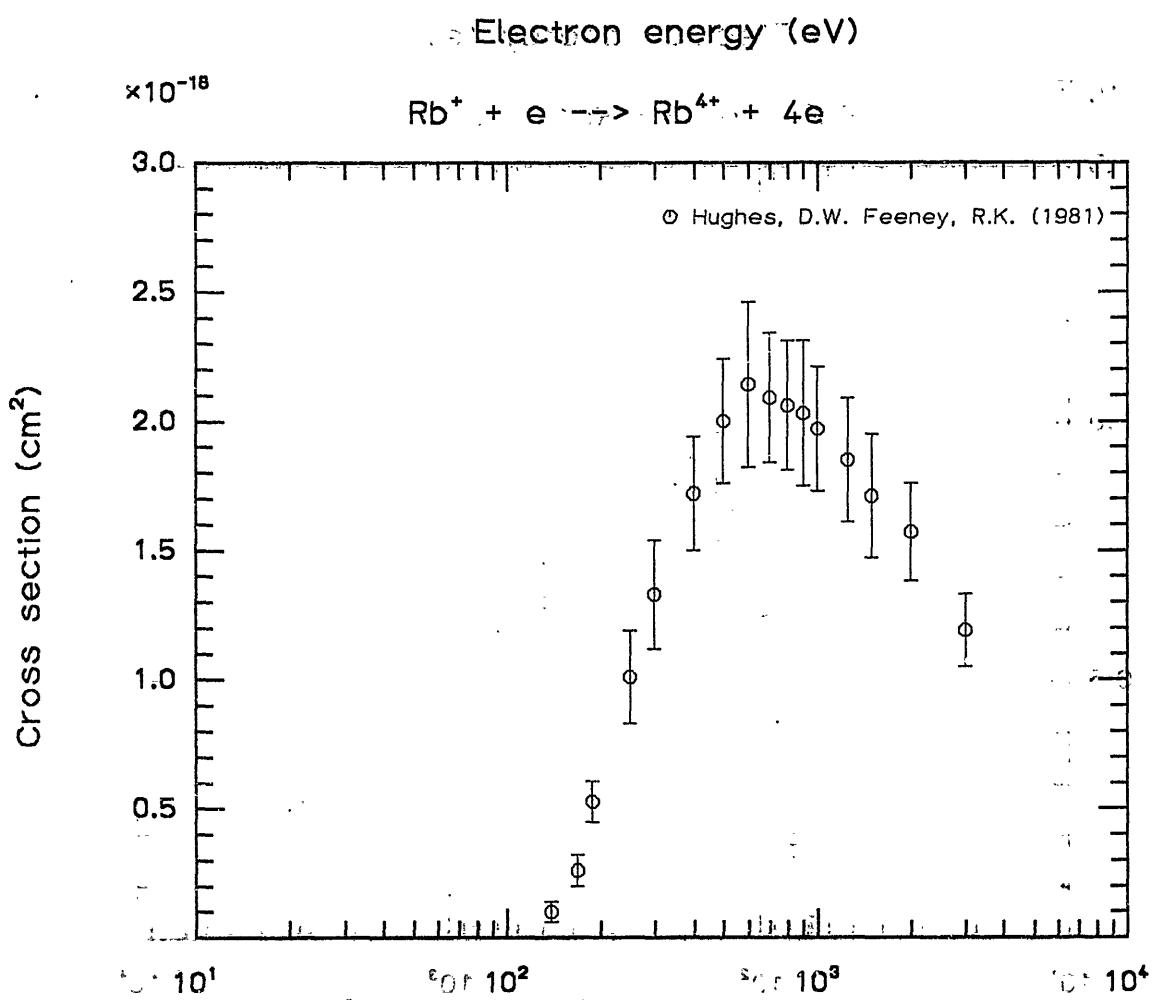
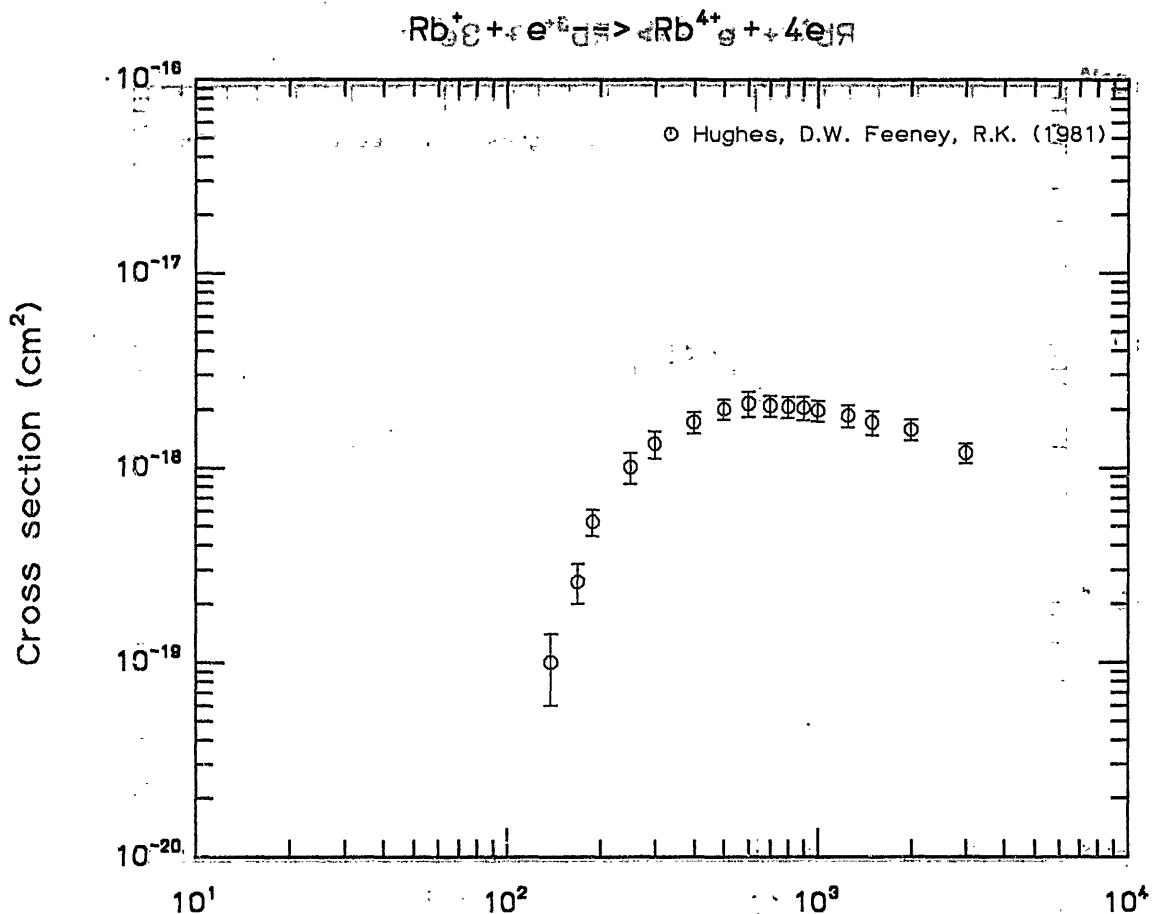


Fig. 206 (vs Electron energy (eV)) Fig. 205

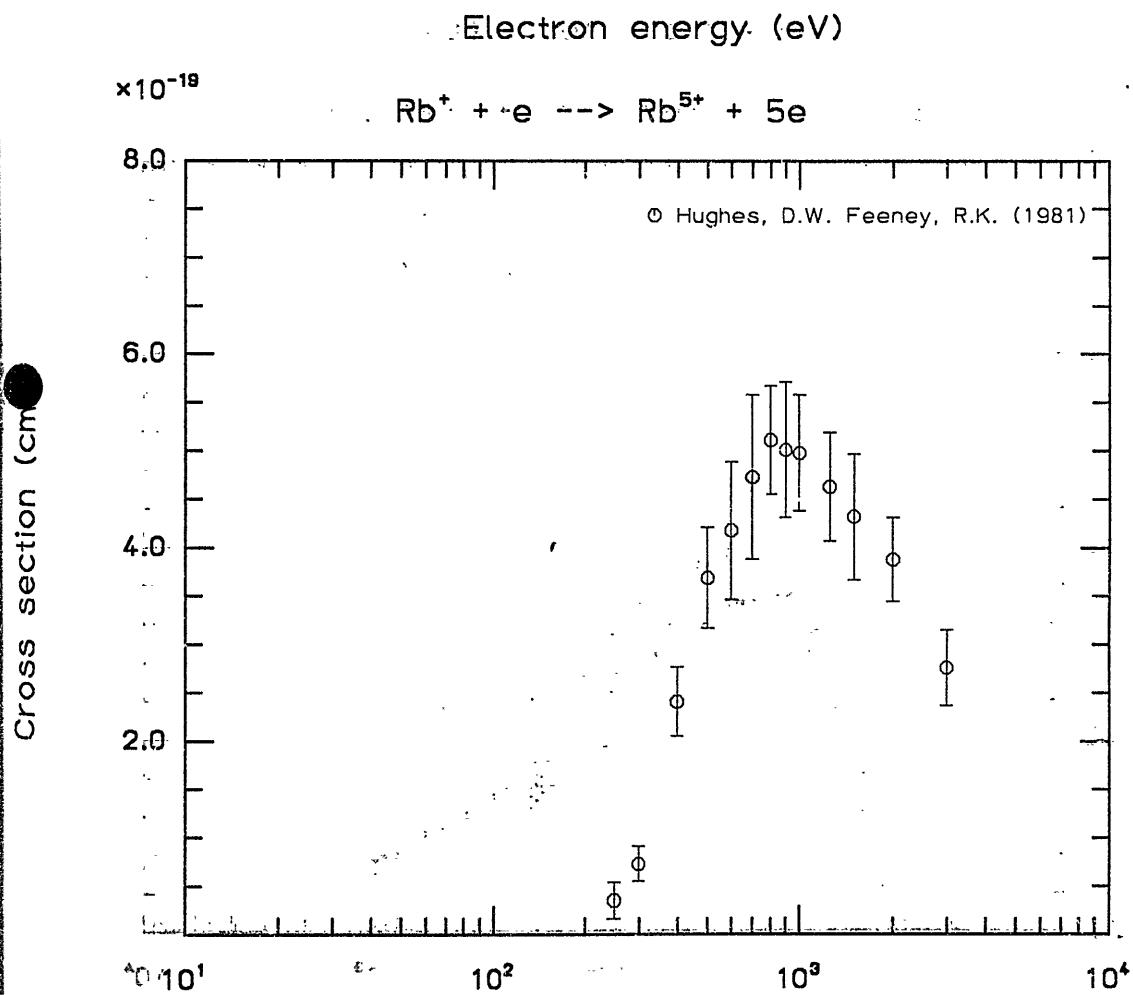
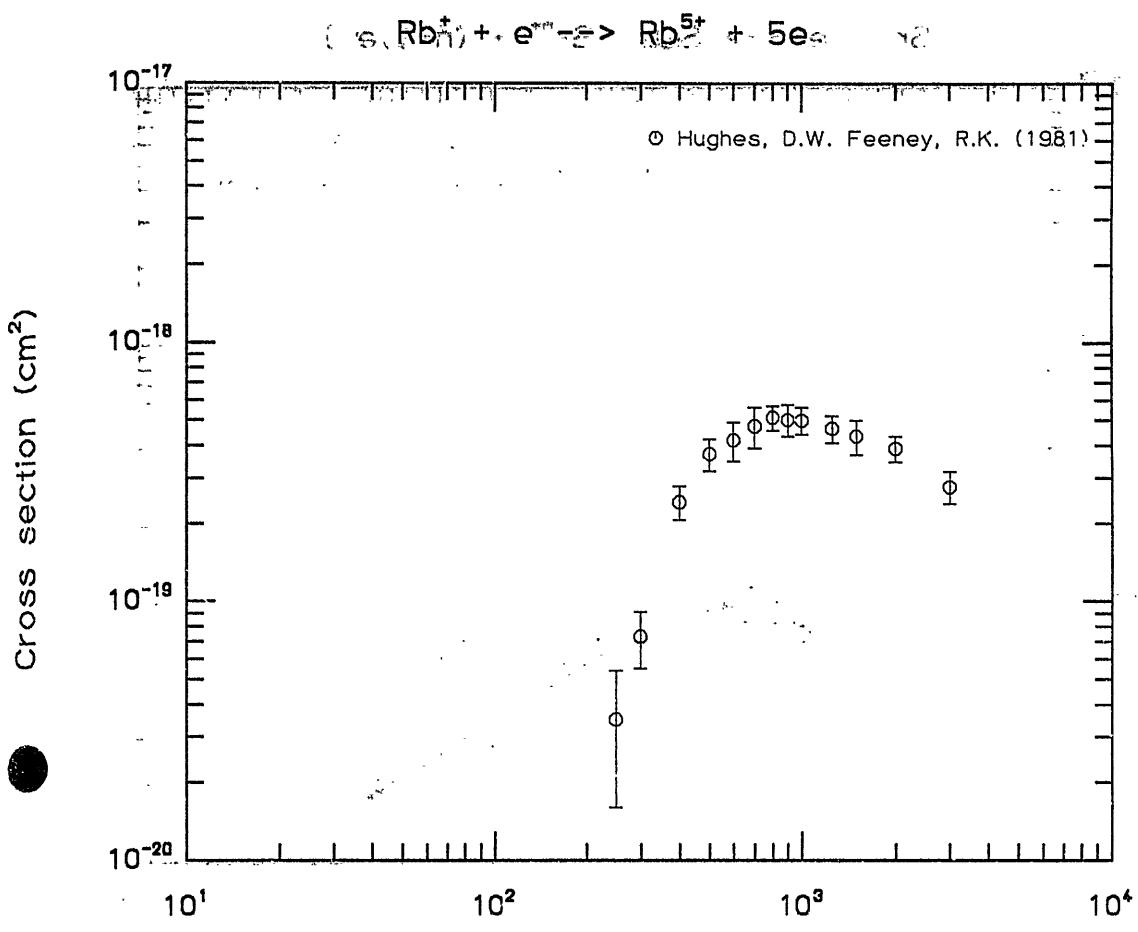


Fig. 207 • (Electron energy (eV)) SUT 2.7

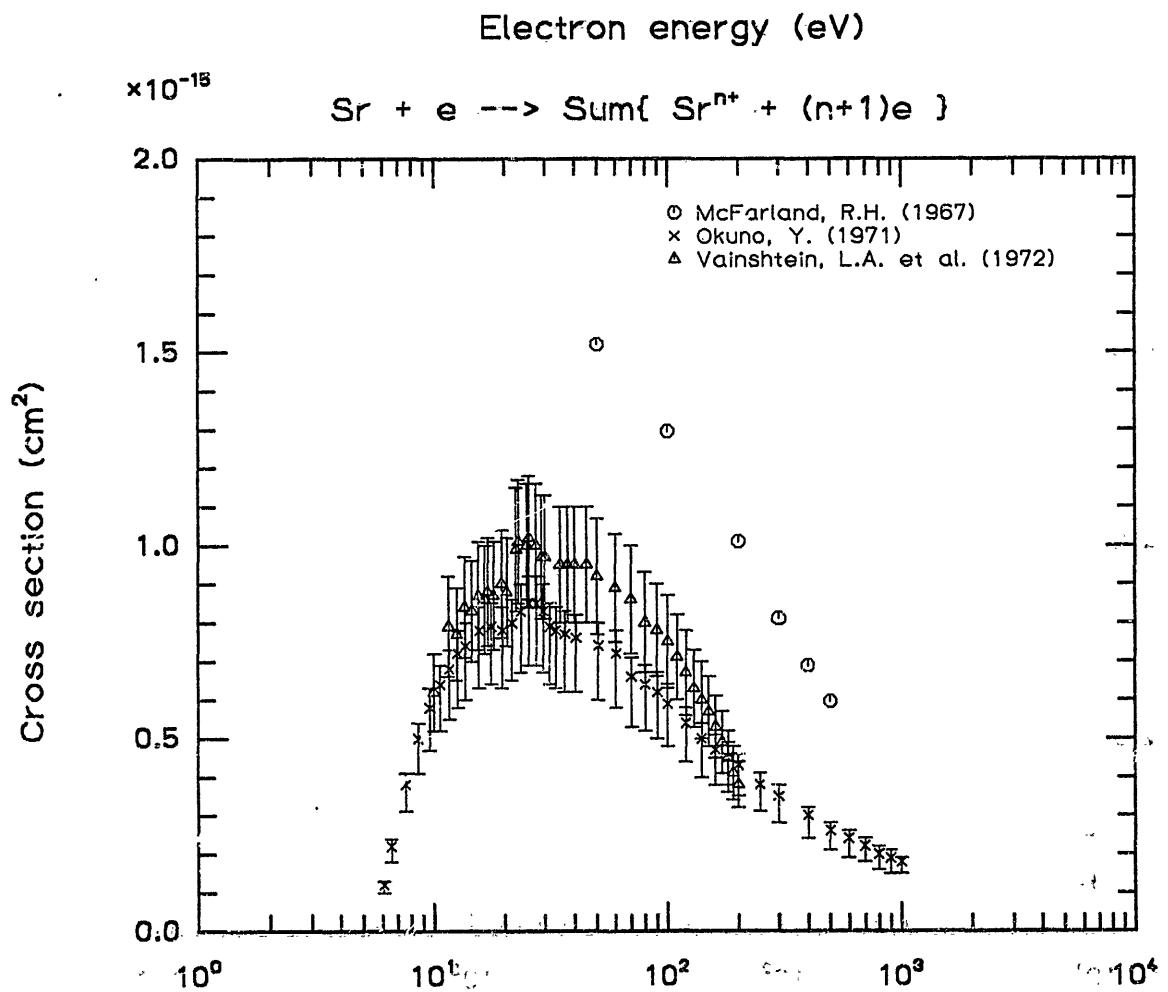
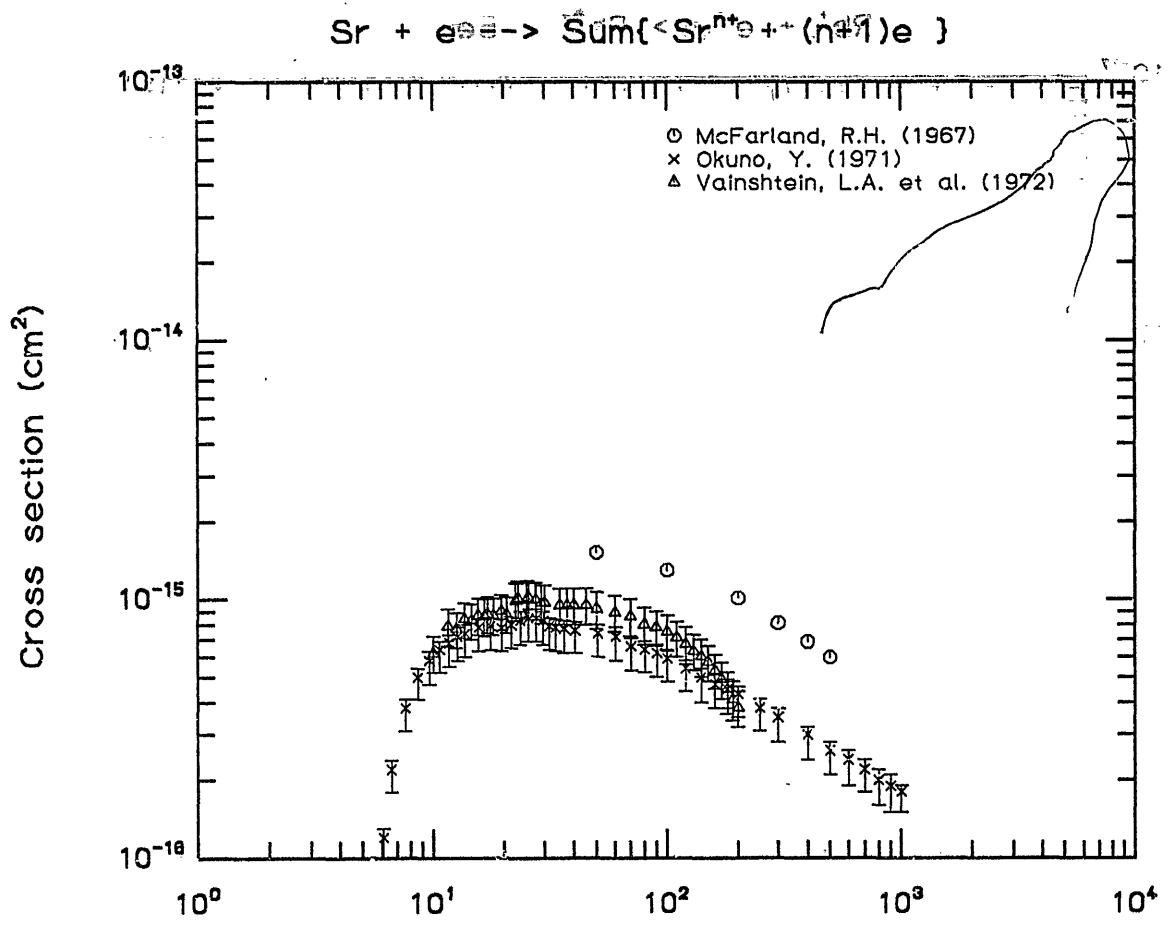
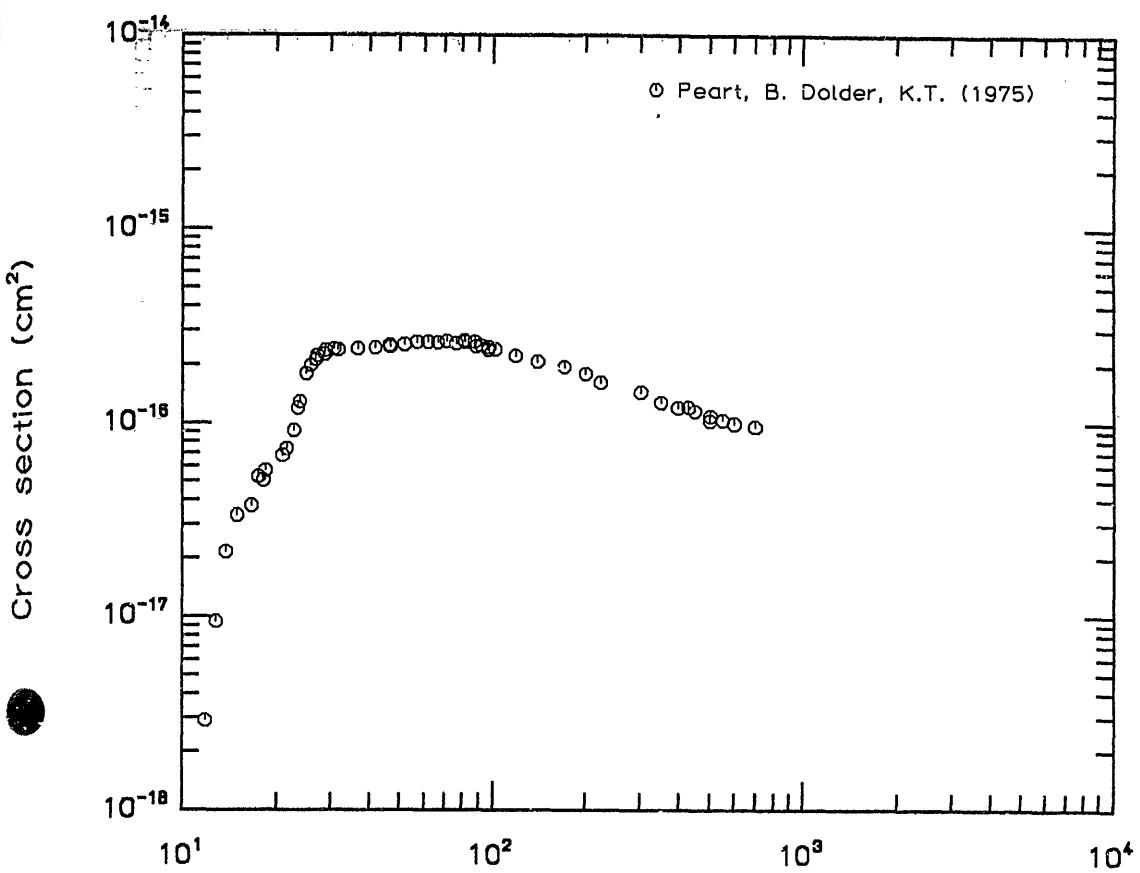
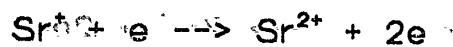


Fig. 208. vs Electron energy (eV) Fig. 209. id.



Electron energy (eV)

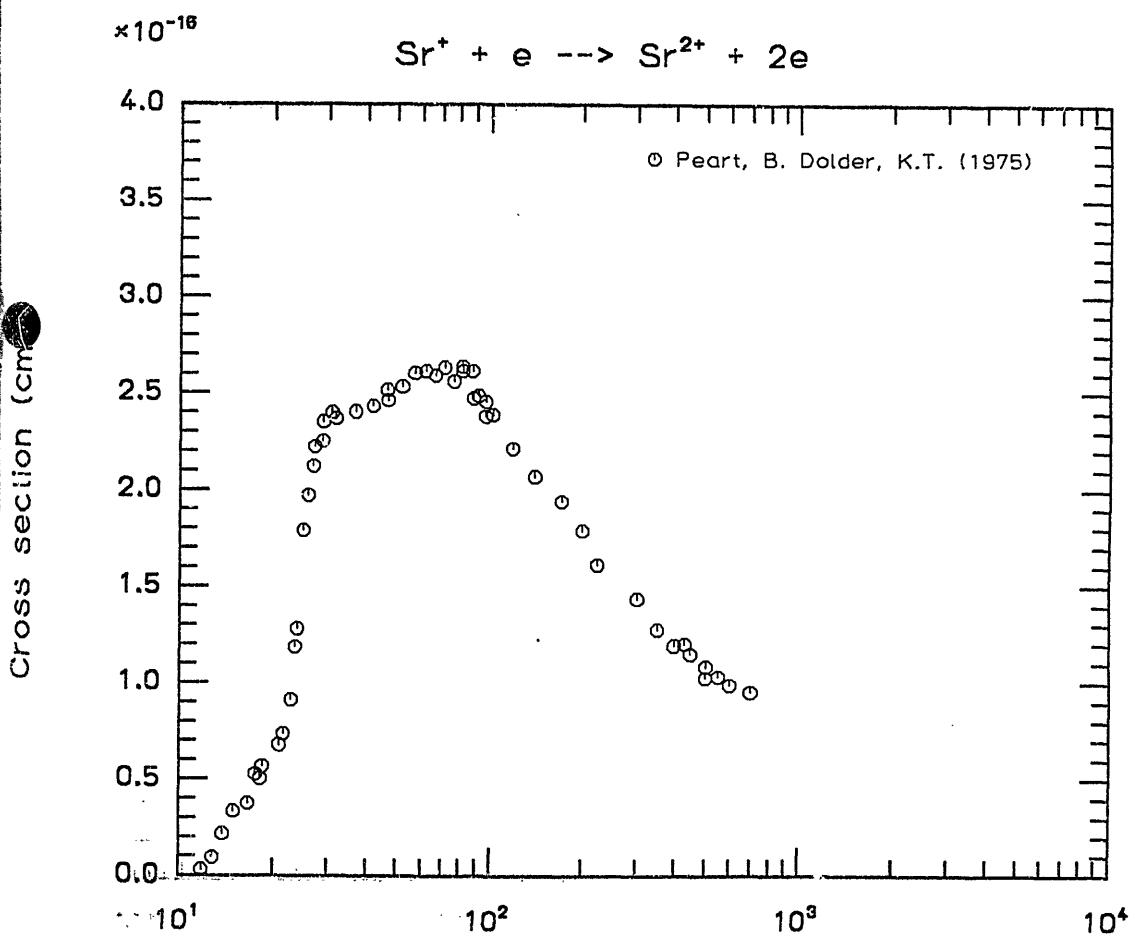


Fig. 209      Electron energy (eV)

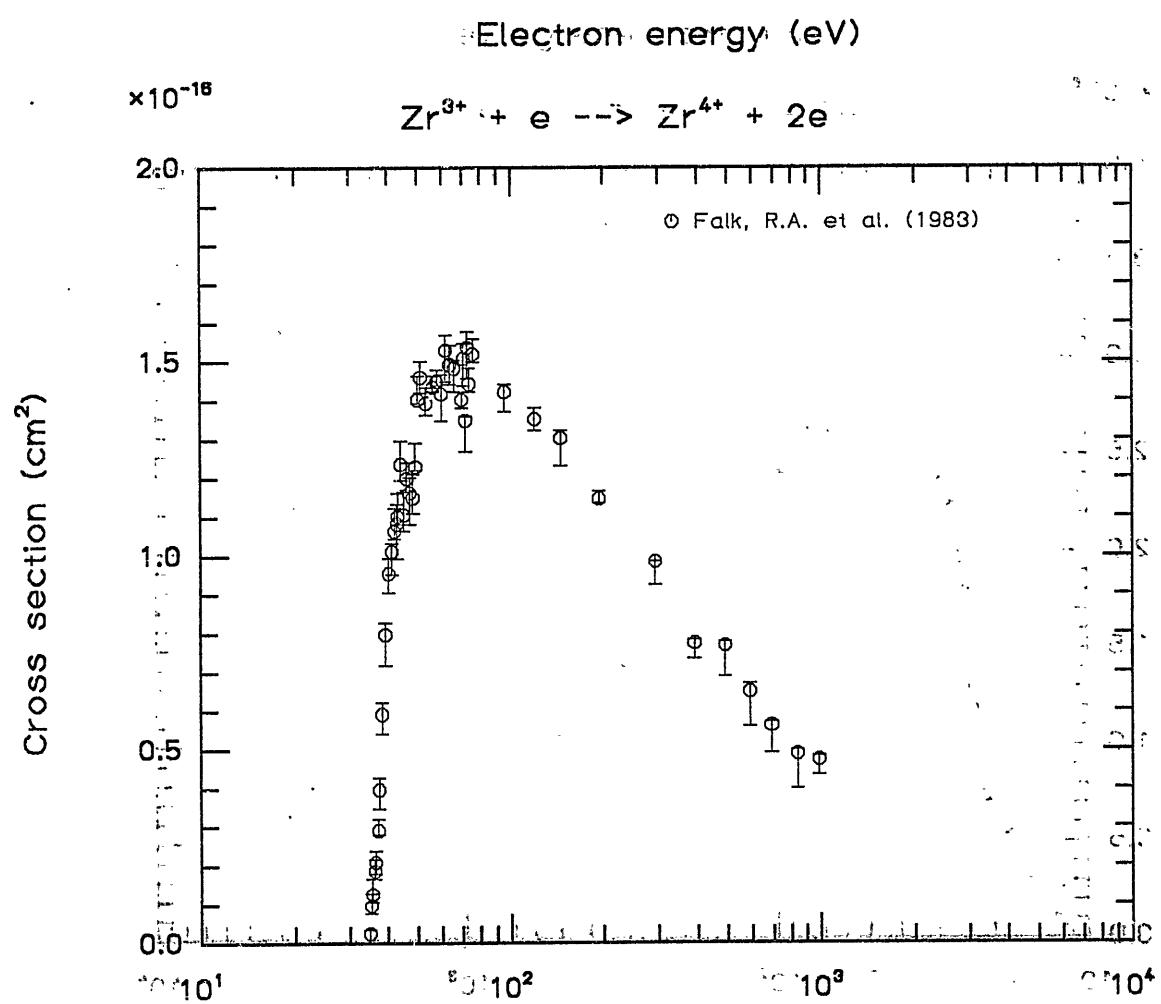
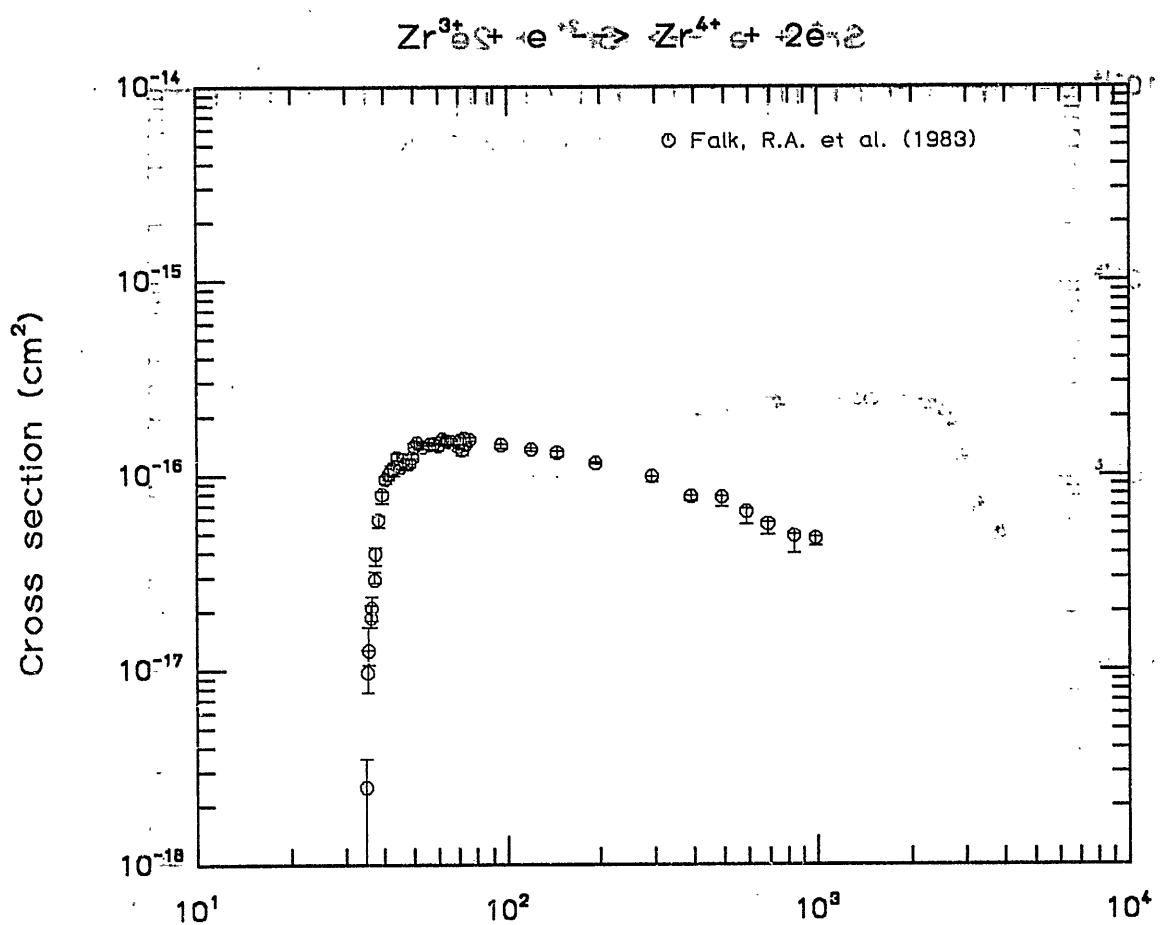
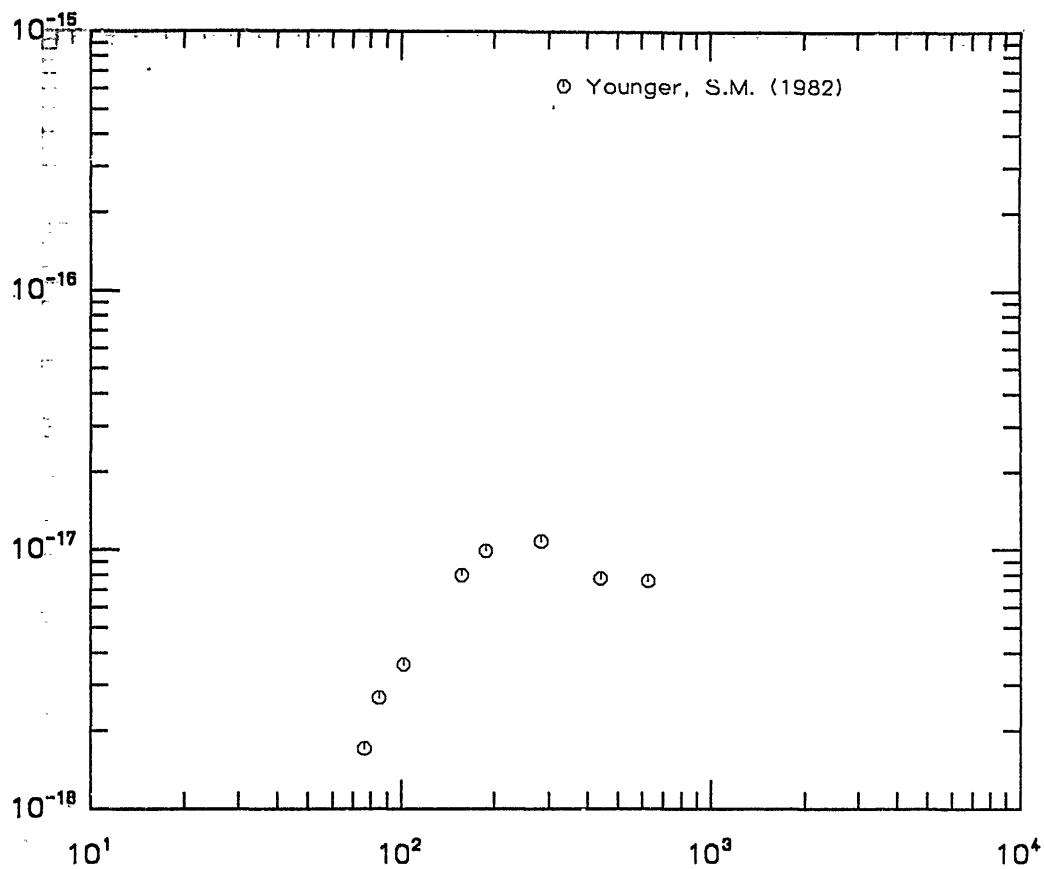
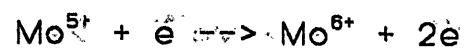


Fig. 210 (v Electron energy (eV)) Fig. 209



Electron energy (eV)

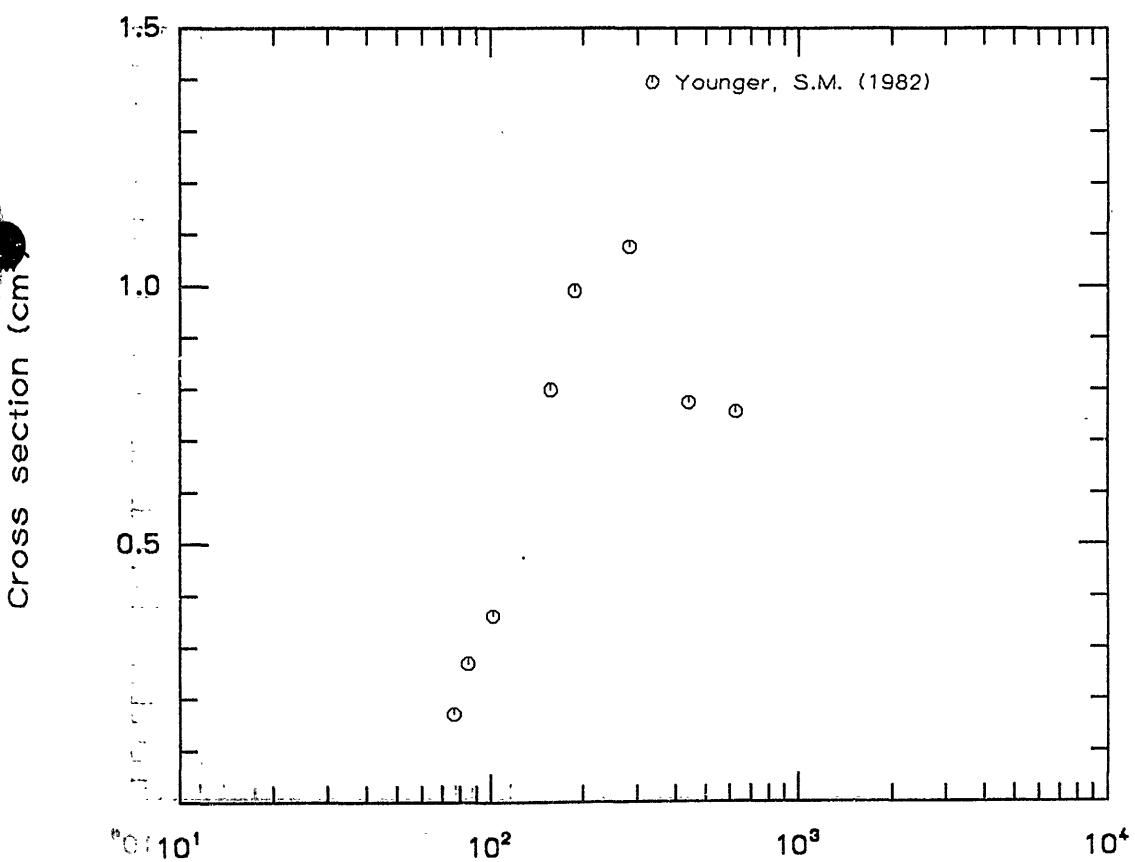
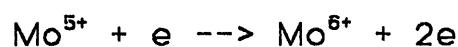


Fig. 211  $\text{Mo}^{5+} + e^- \rightarrow \text{Mo}^{6+} + 2e^-$

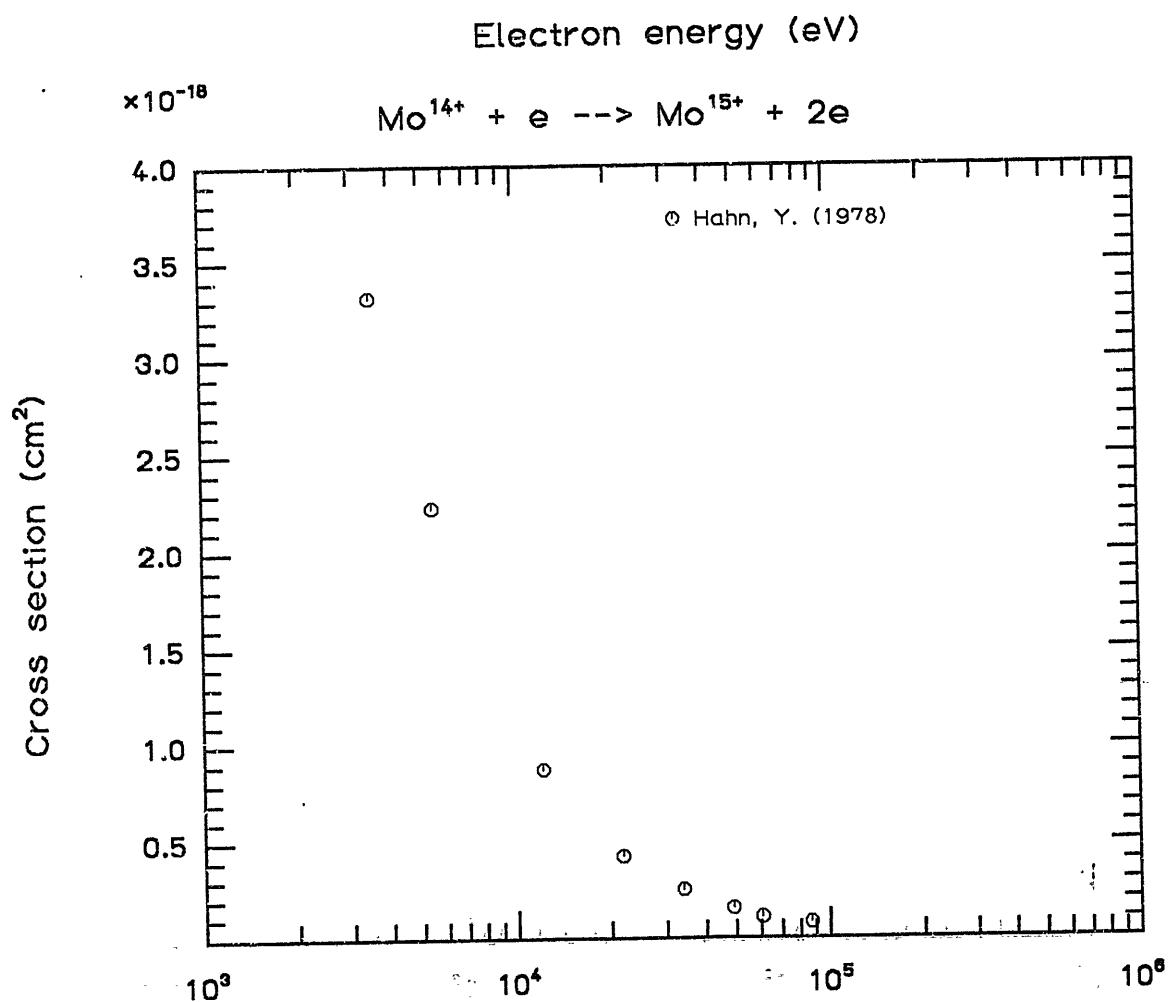
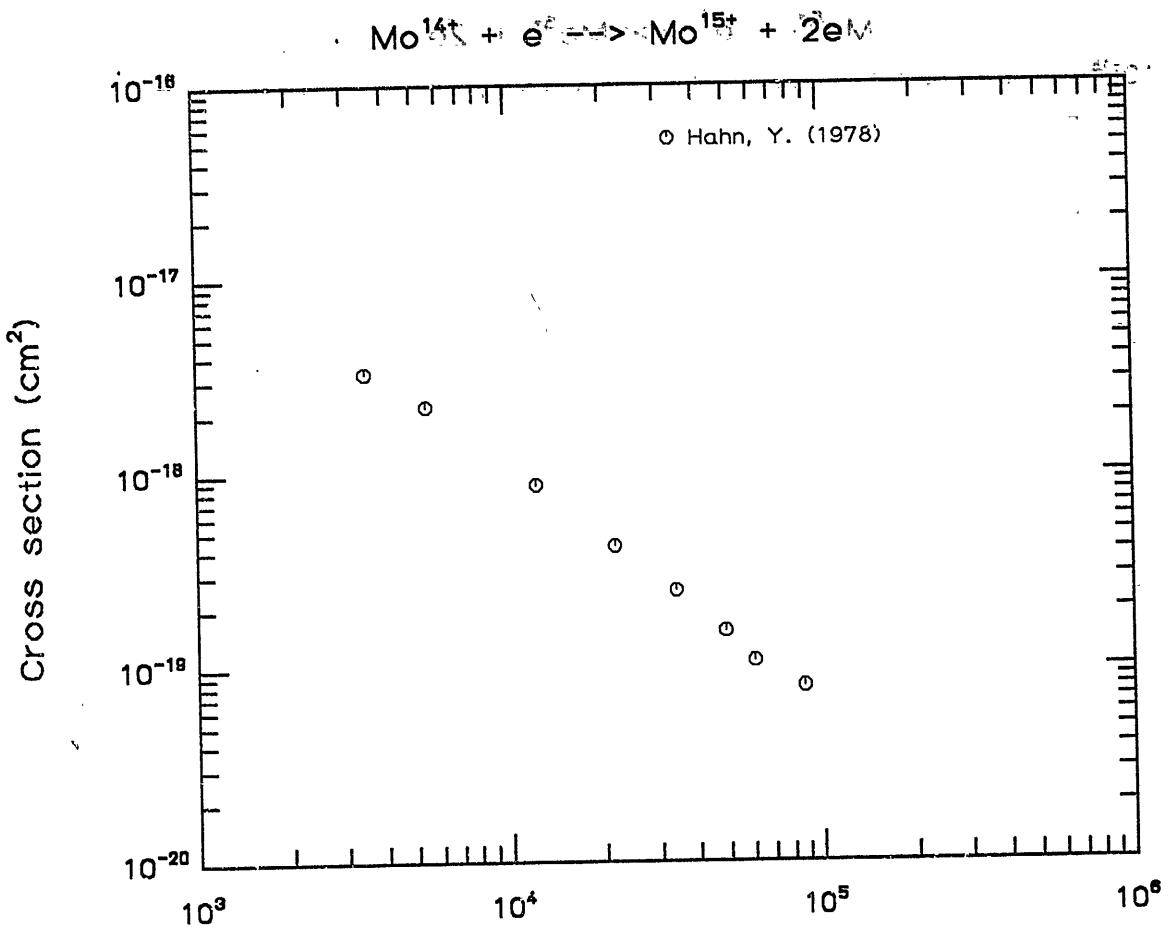
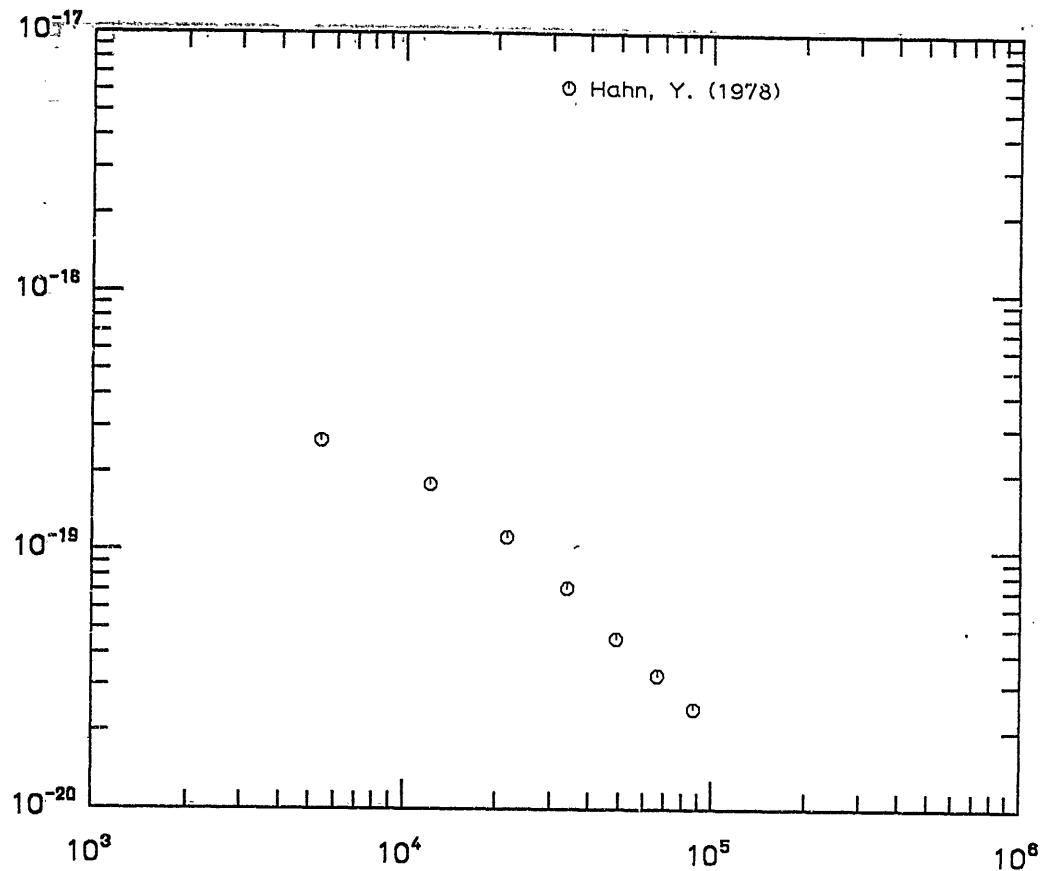


Fig. 212 (a) Electron energy (eV)

Fig. 212 (b)



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

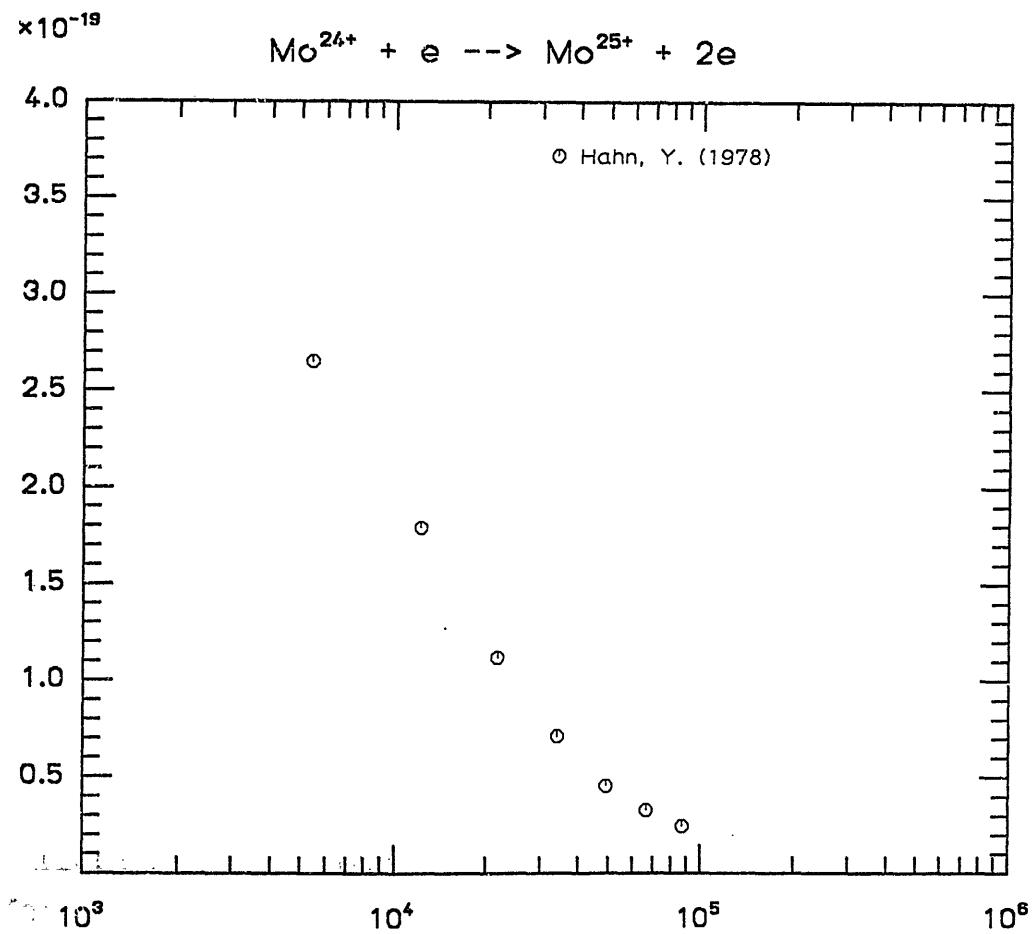


Fig. 213 Electron energy (eV)

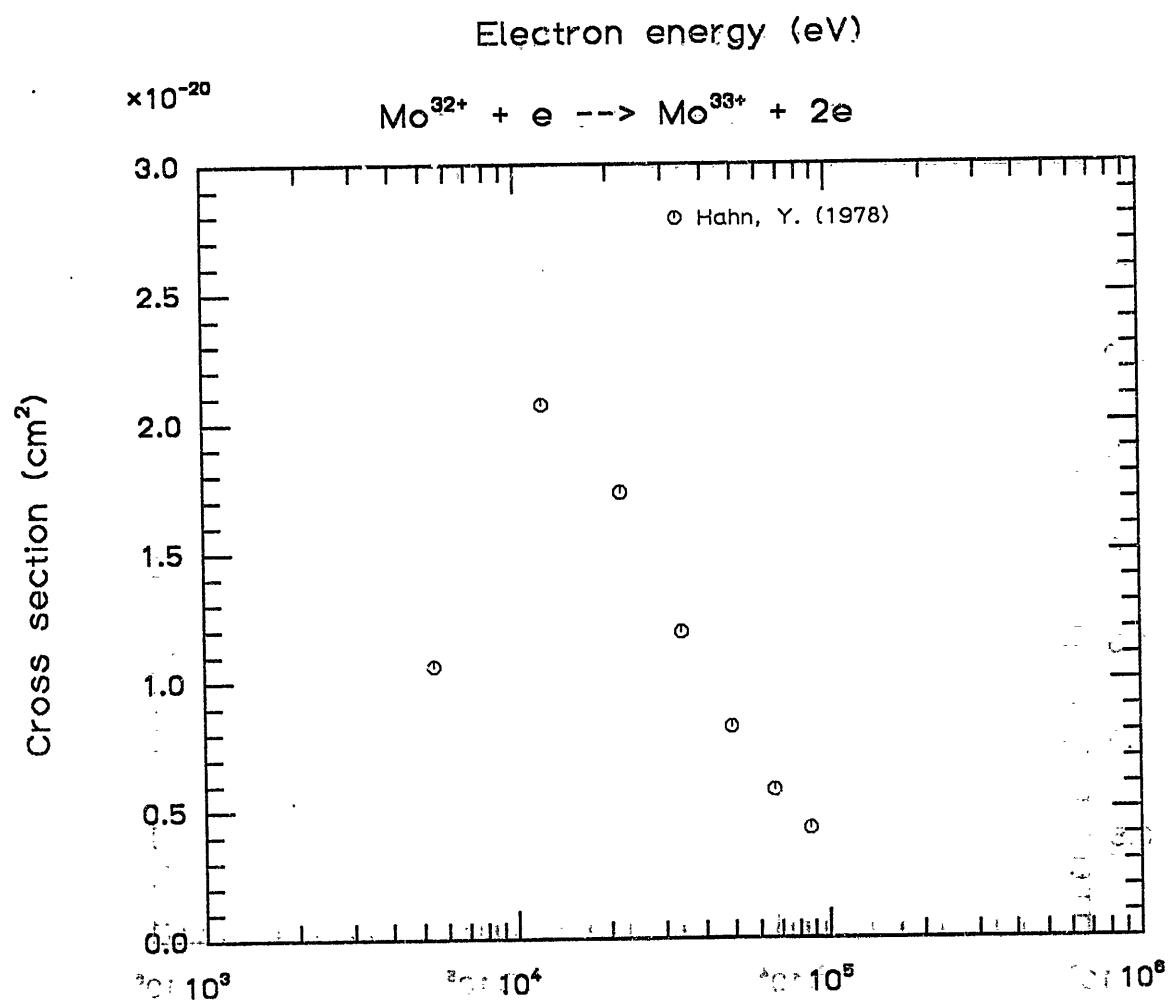
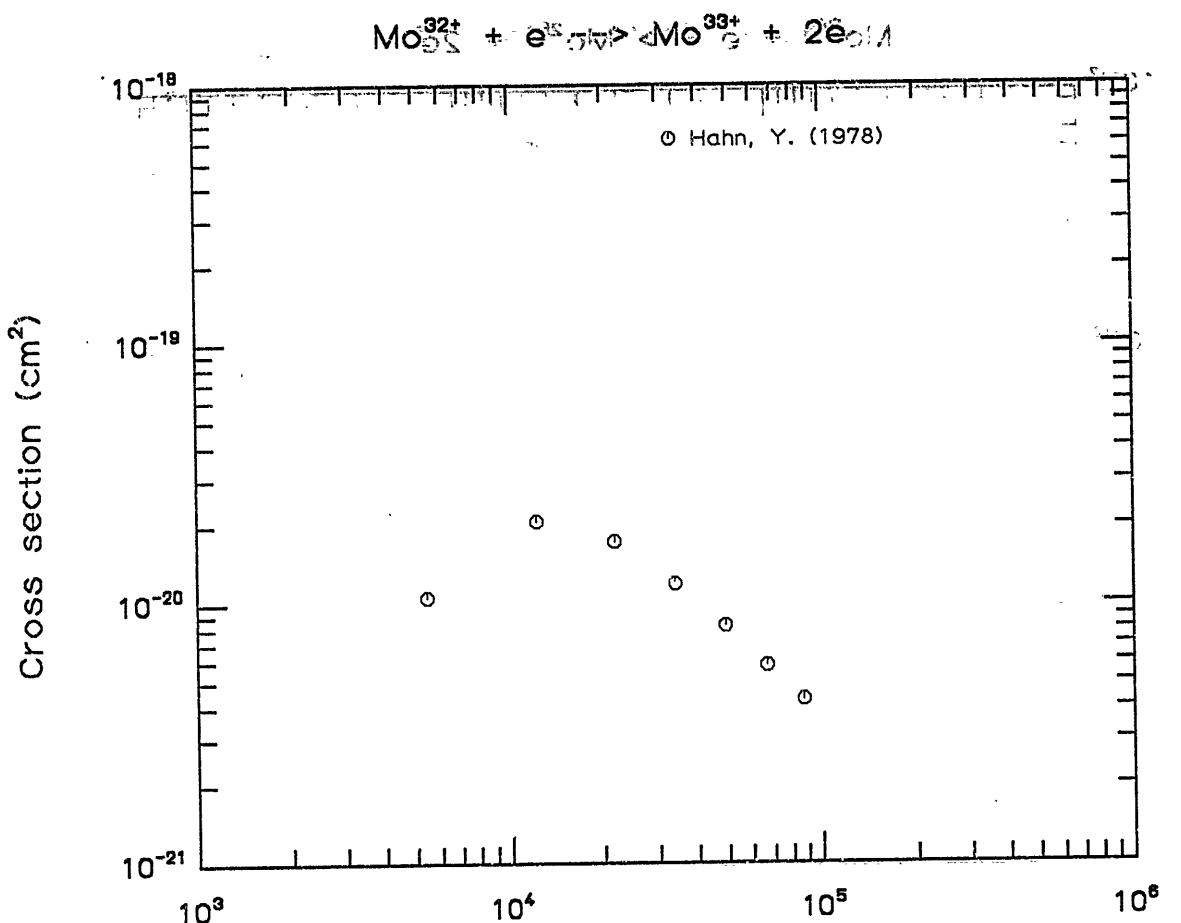
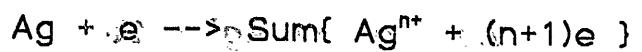
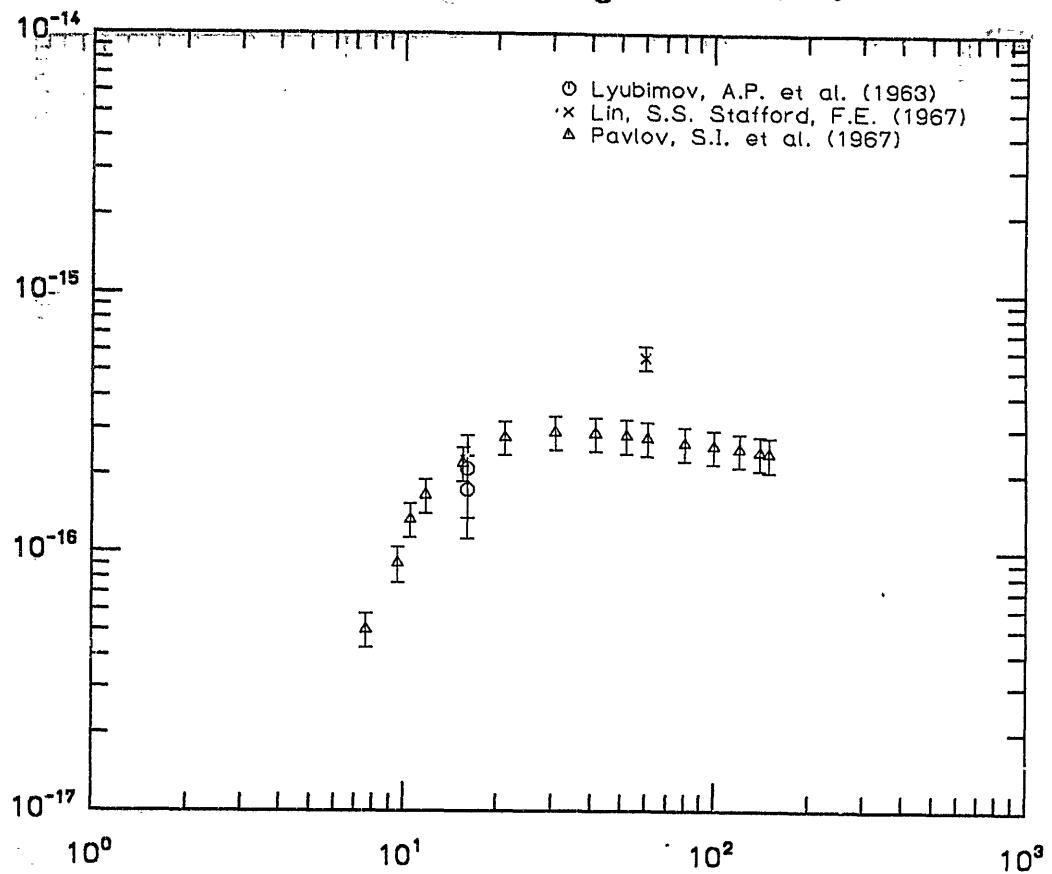


Fig. 22.14 (Mo<sup>32+</sup> + e → Mo<sup>33+</sup> + 2e) Fig. 22.14



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

Cross section ( $\text{cm}^2$ )

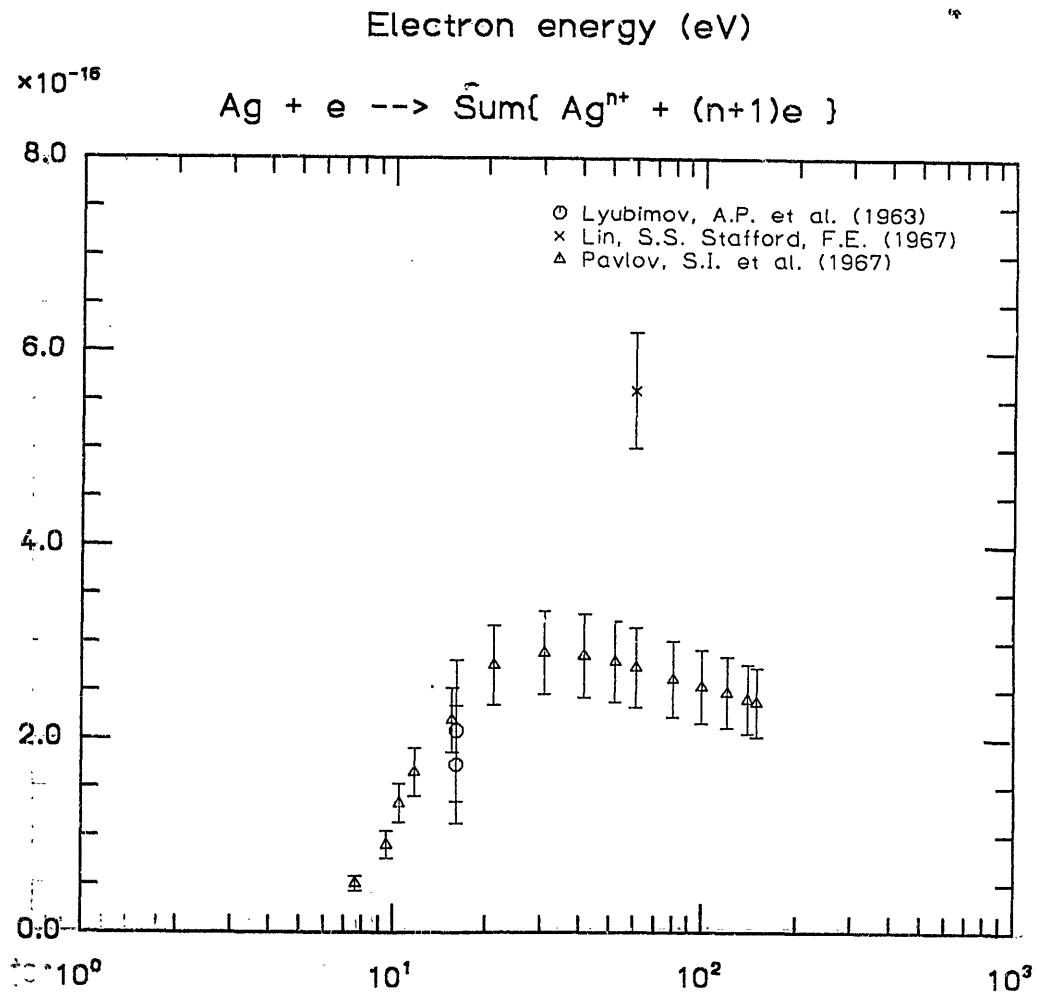


Fig. 215 Electron energy (eV)  $\text{cm}^2$

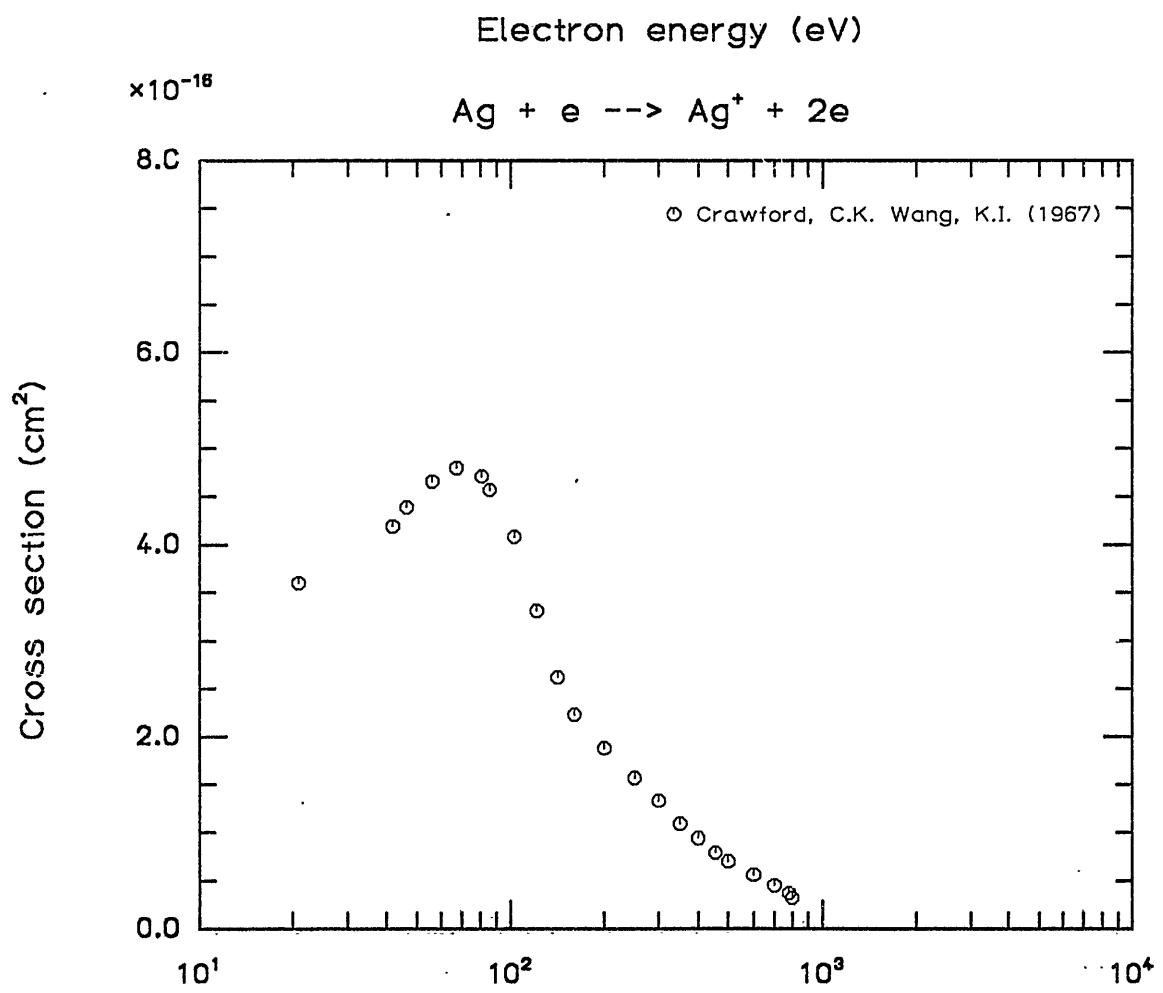
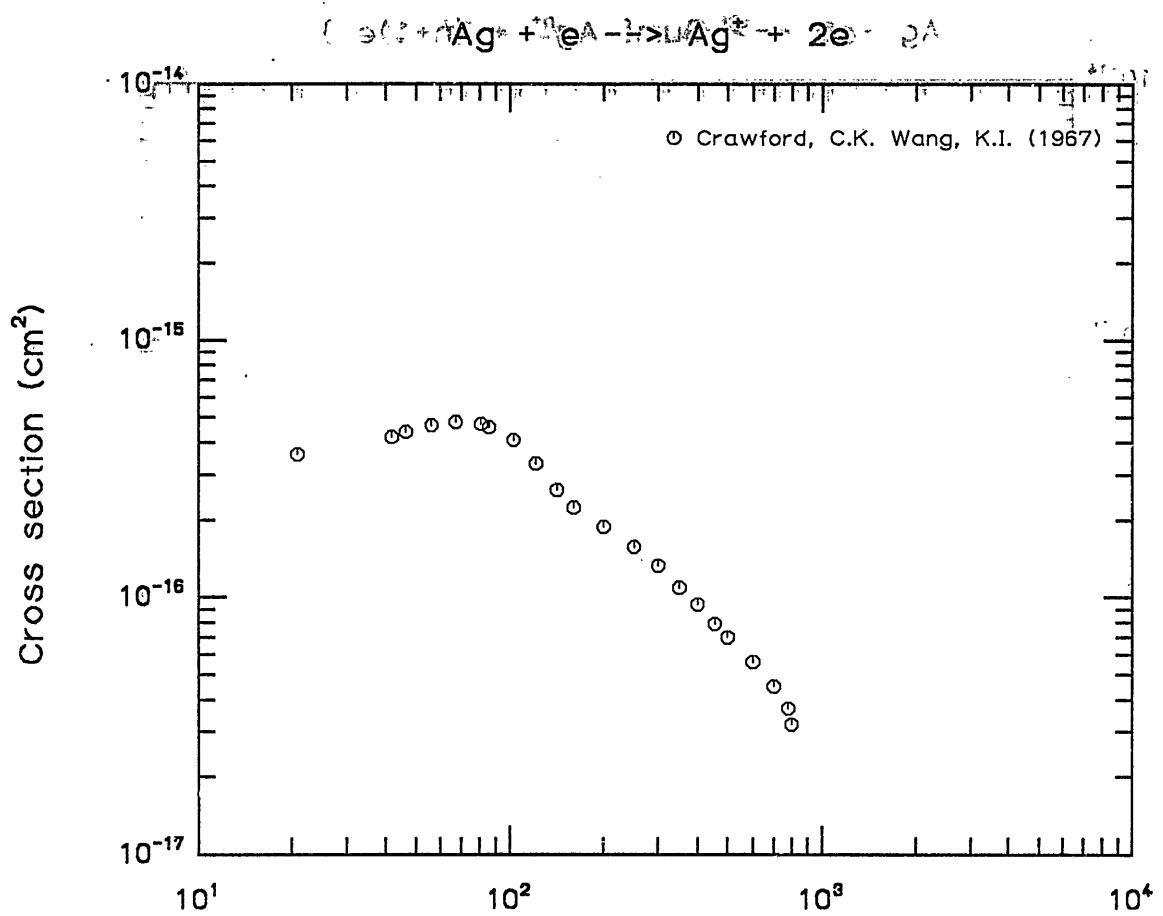


Fig. 216      Electron energy (eV)      215.0

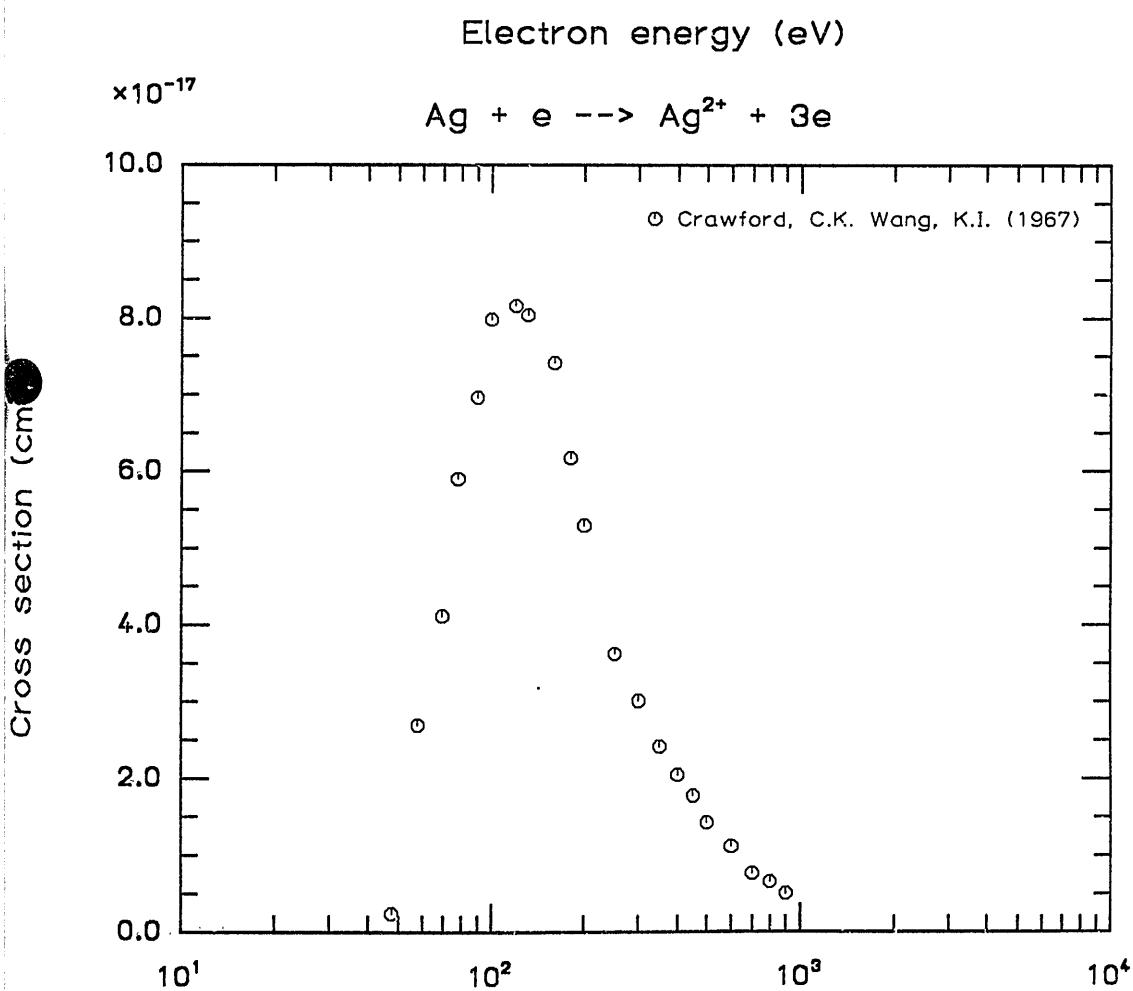
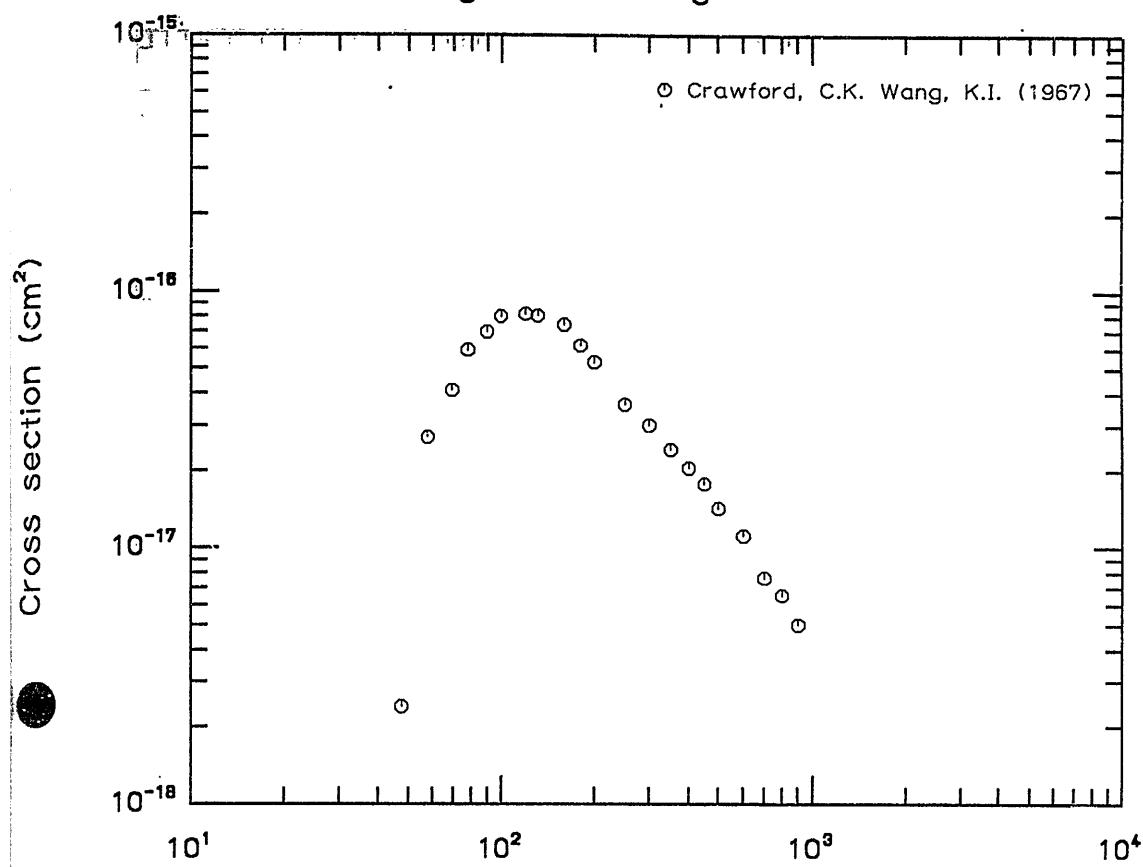
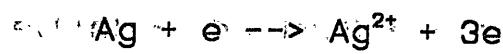


Fig. 217 Electron energy (eV)

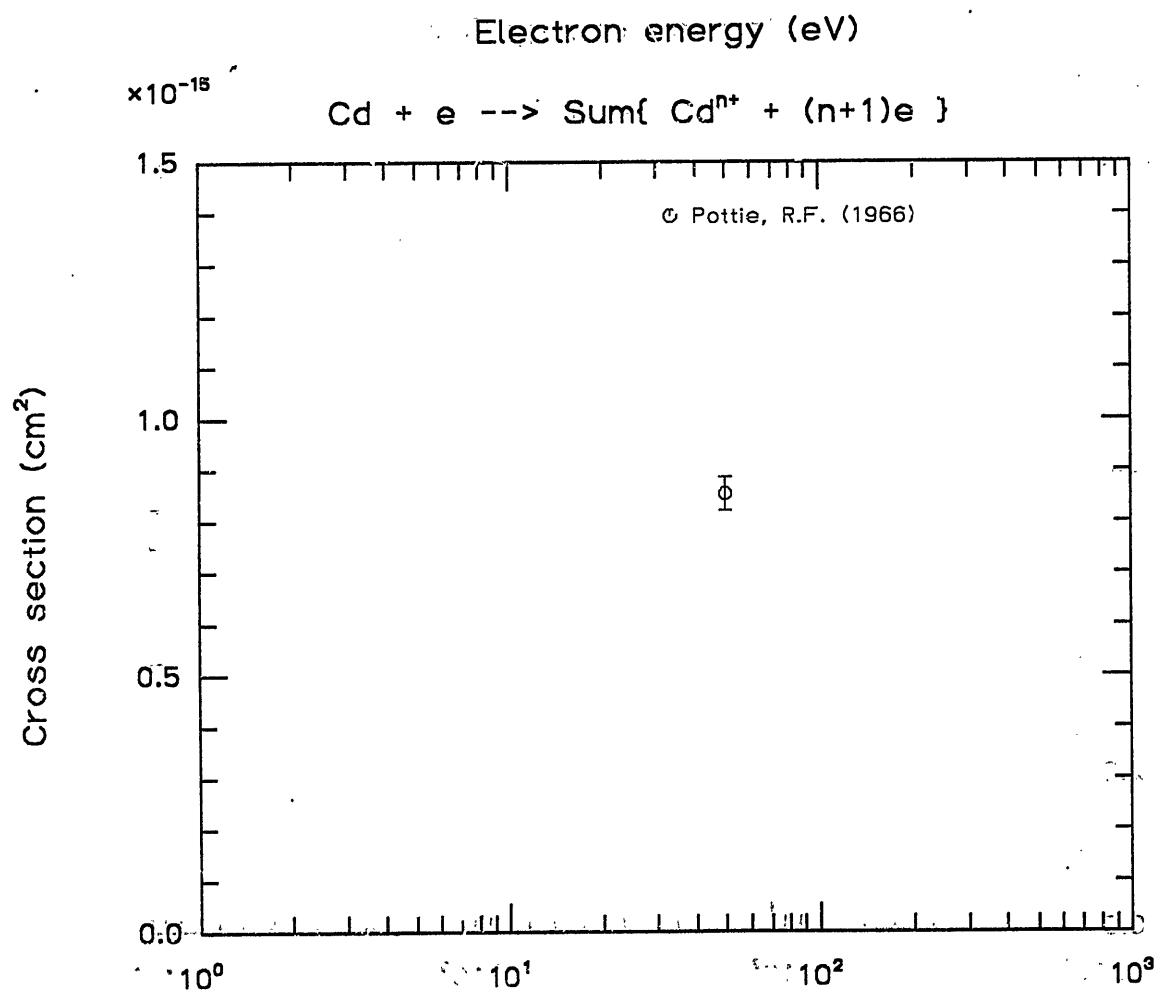
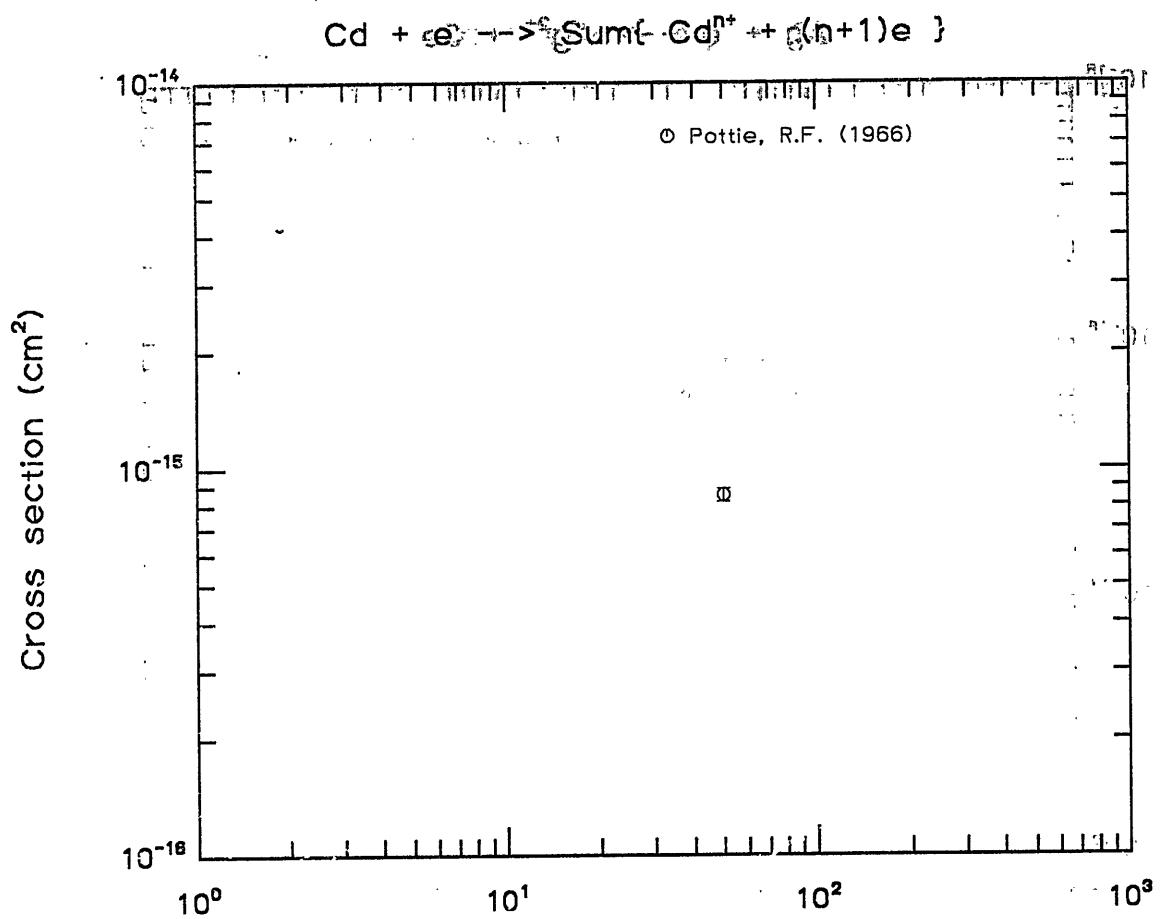
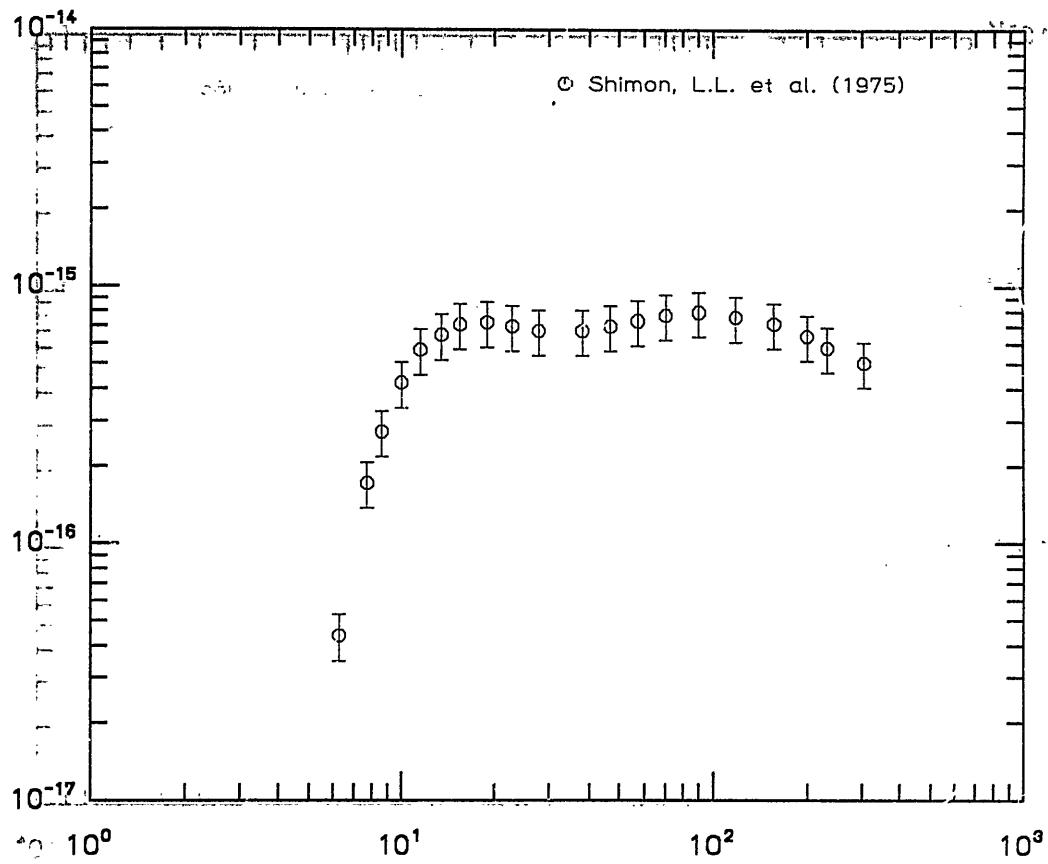
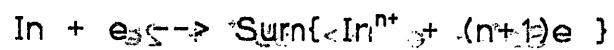
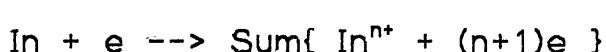


Fig. 218 (Ne) Electron energy (eV) Fig. 219 (e)



Electron energy (eV)



Cross section ( $\text{cm}^2$ )

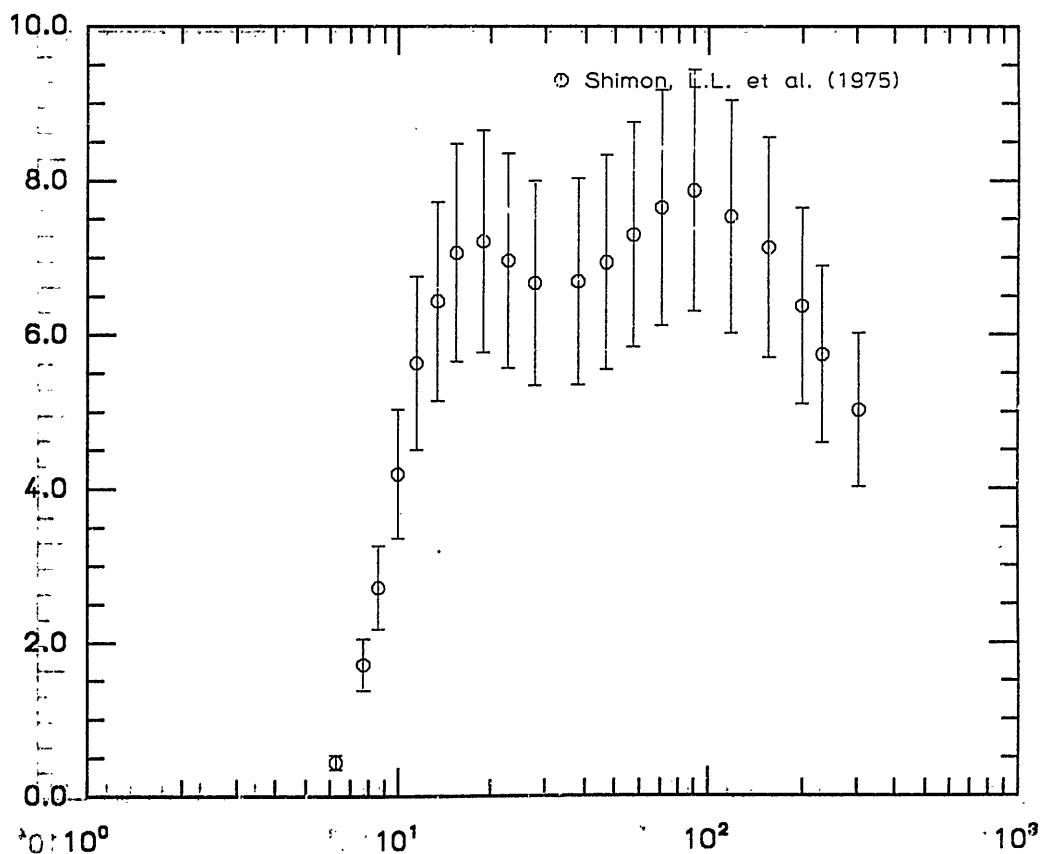


Fig. 219 (a) Electron energy (eV) (b)  $\text{cm}^2$

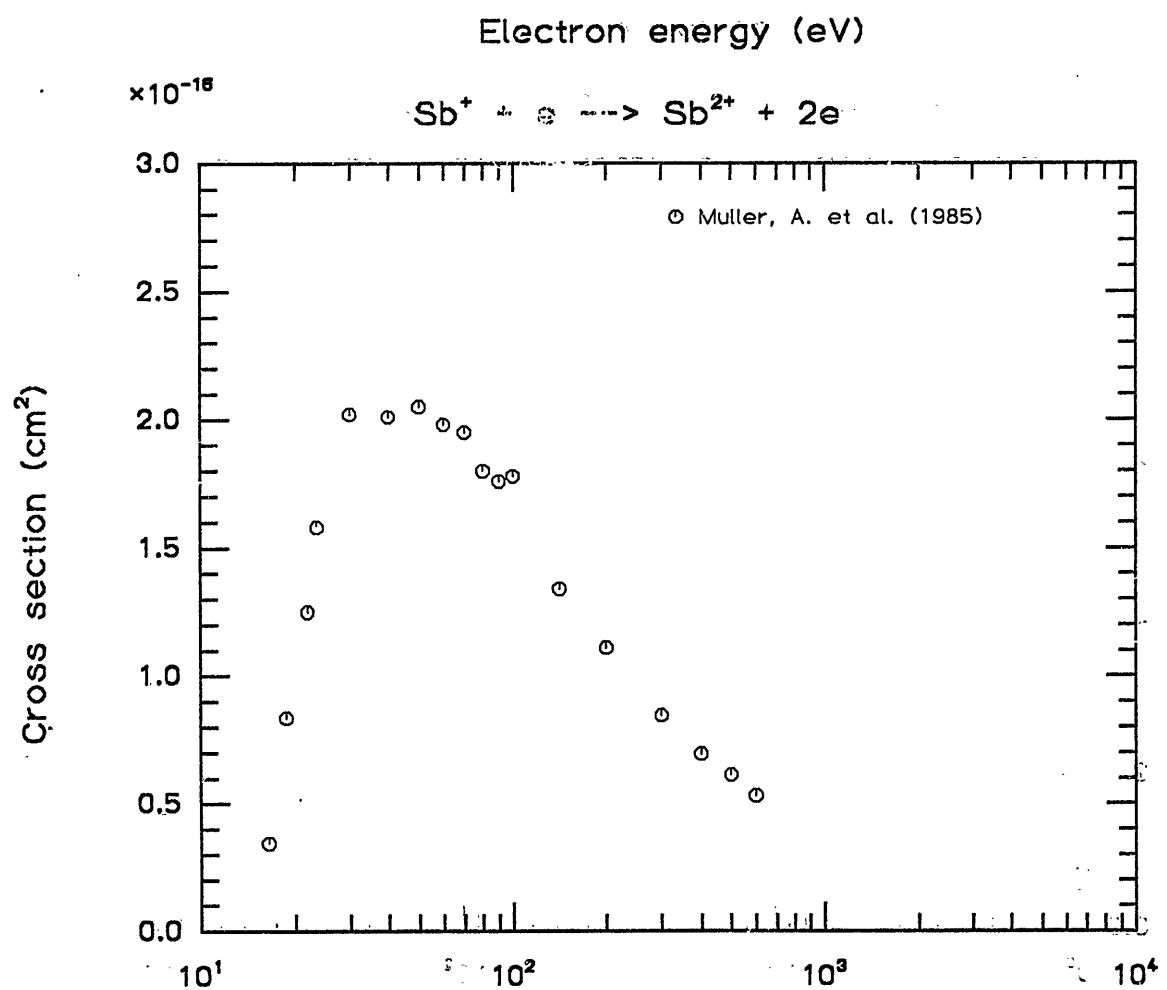
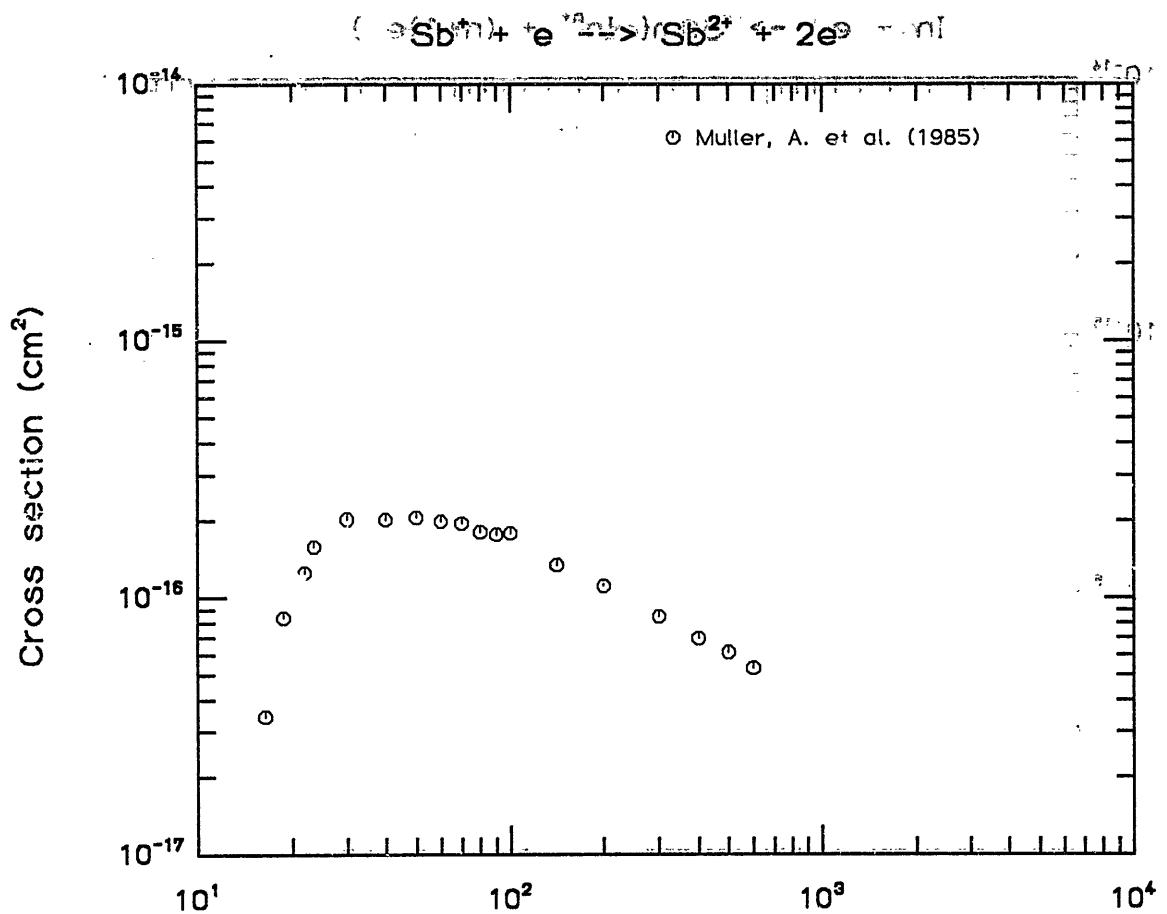
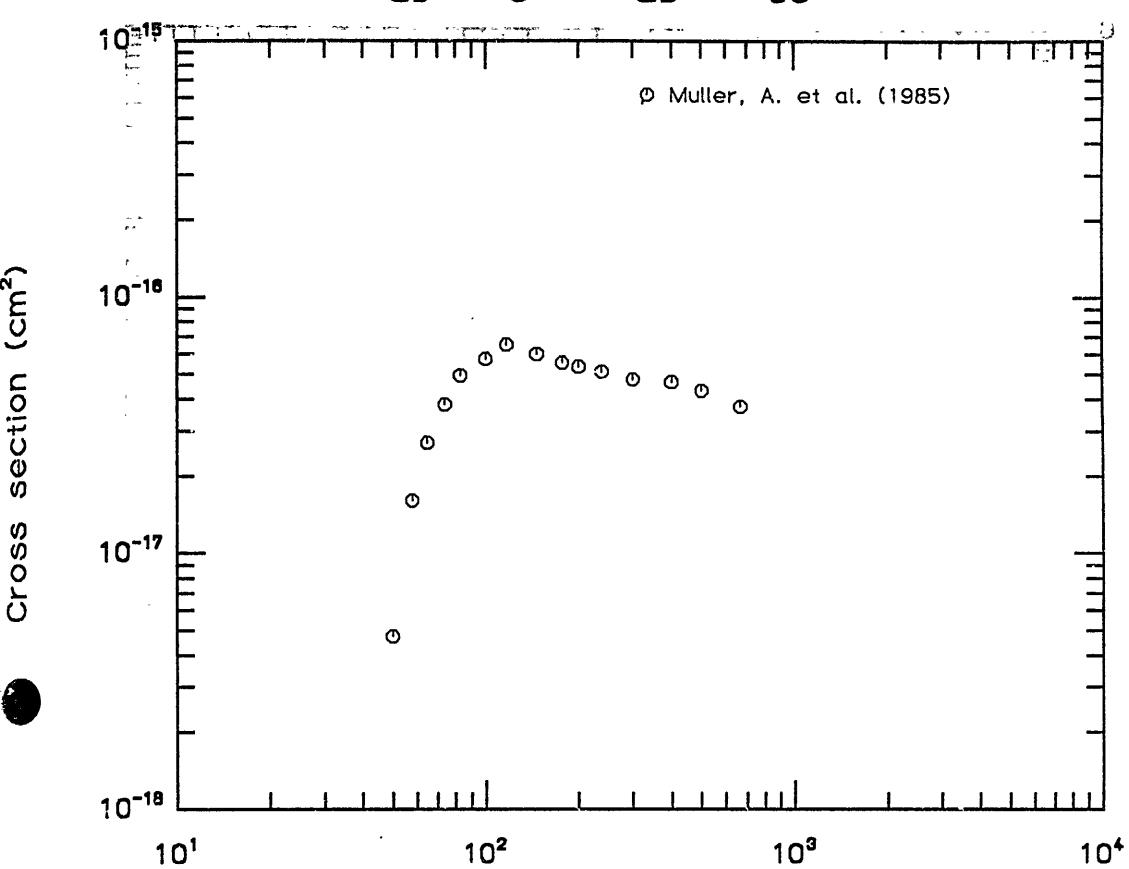
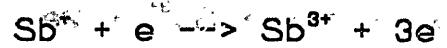


Fig. 220 (Top) Electron energy (eV) of Sb<sup>m</sup>



Electron energy (eV)

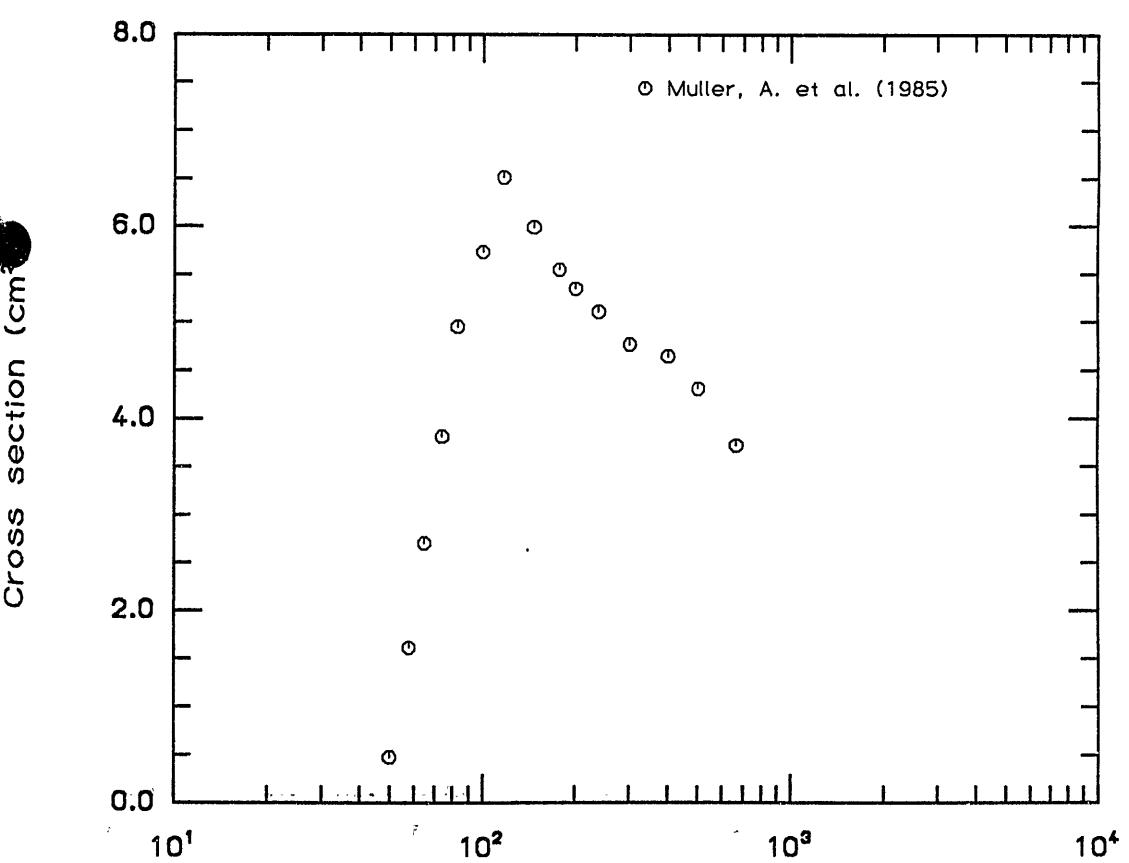
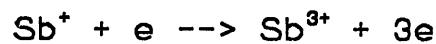


Fig. 221

Electron energy (eV)

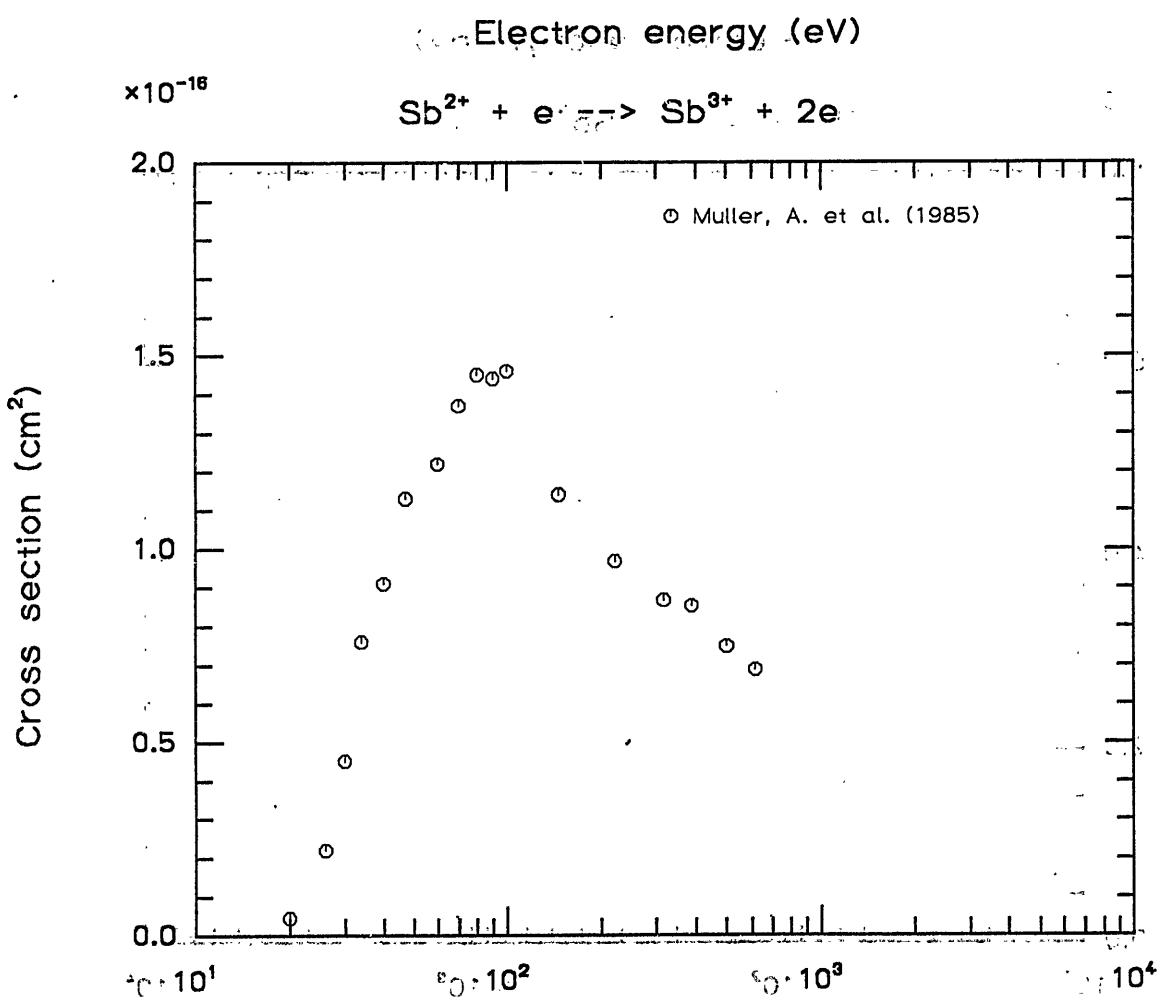
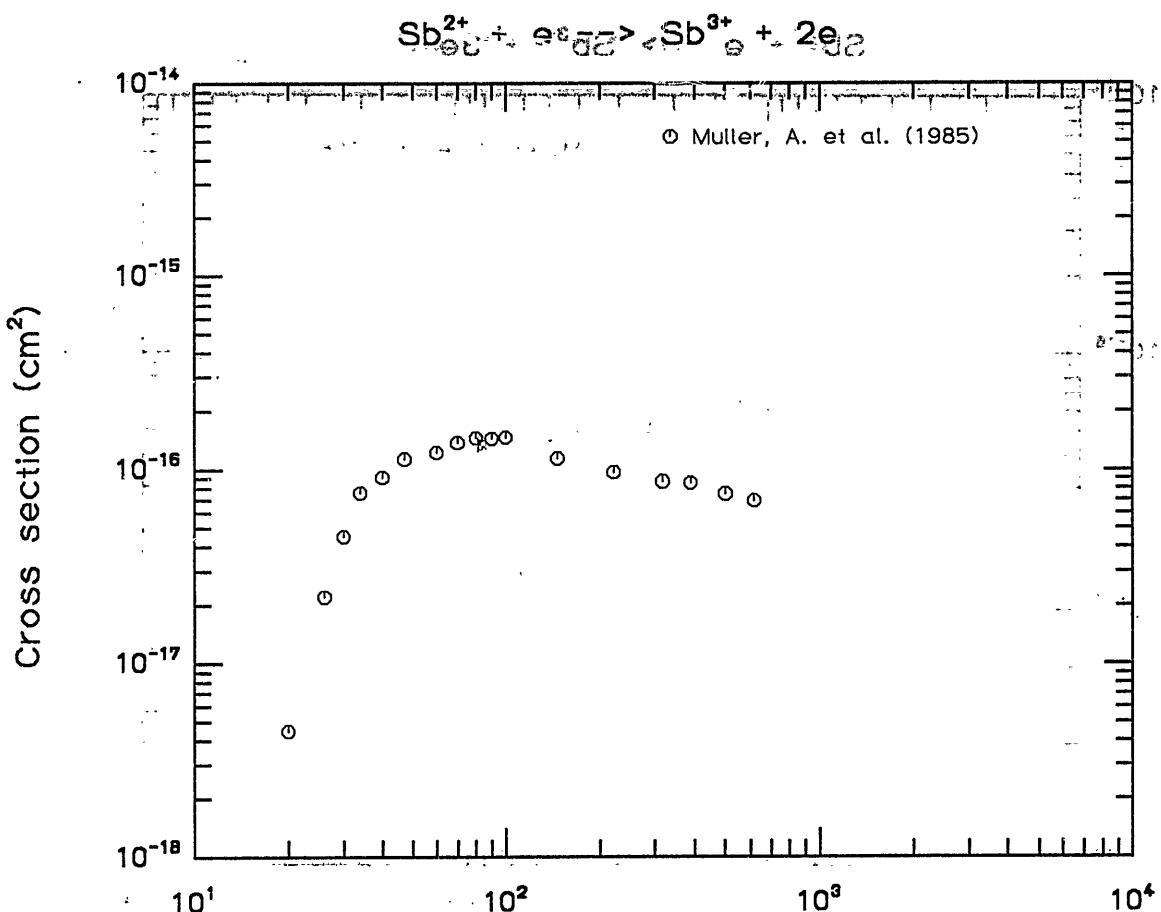


Fig. 222 (v) Electron energy (eV) 122 pB

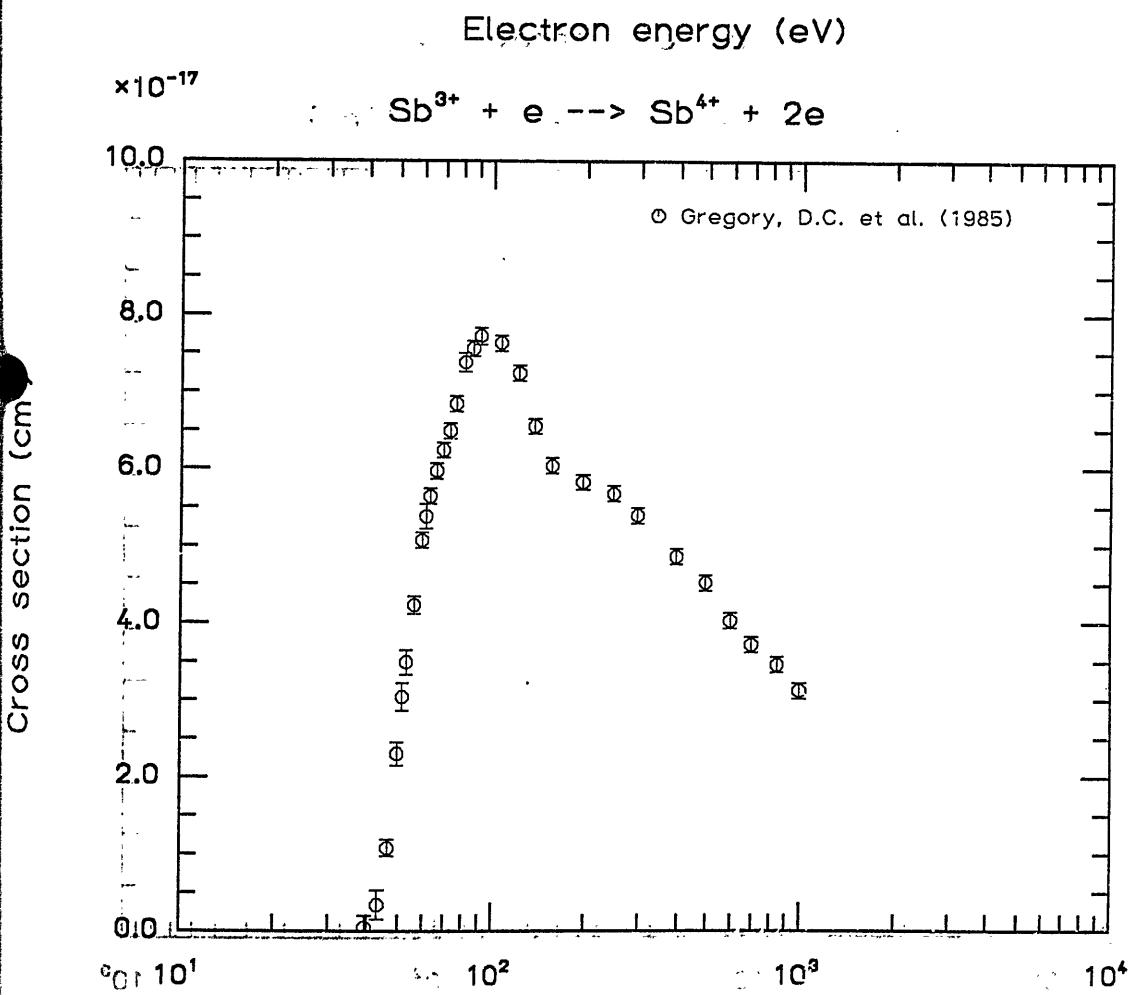
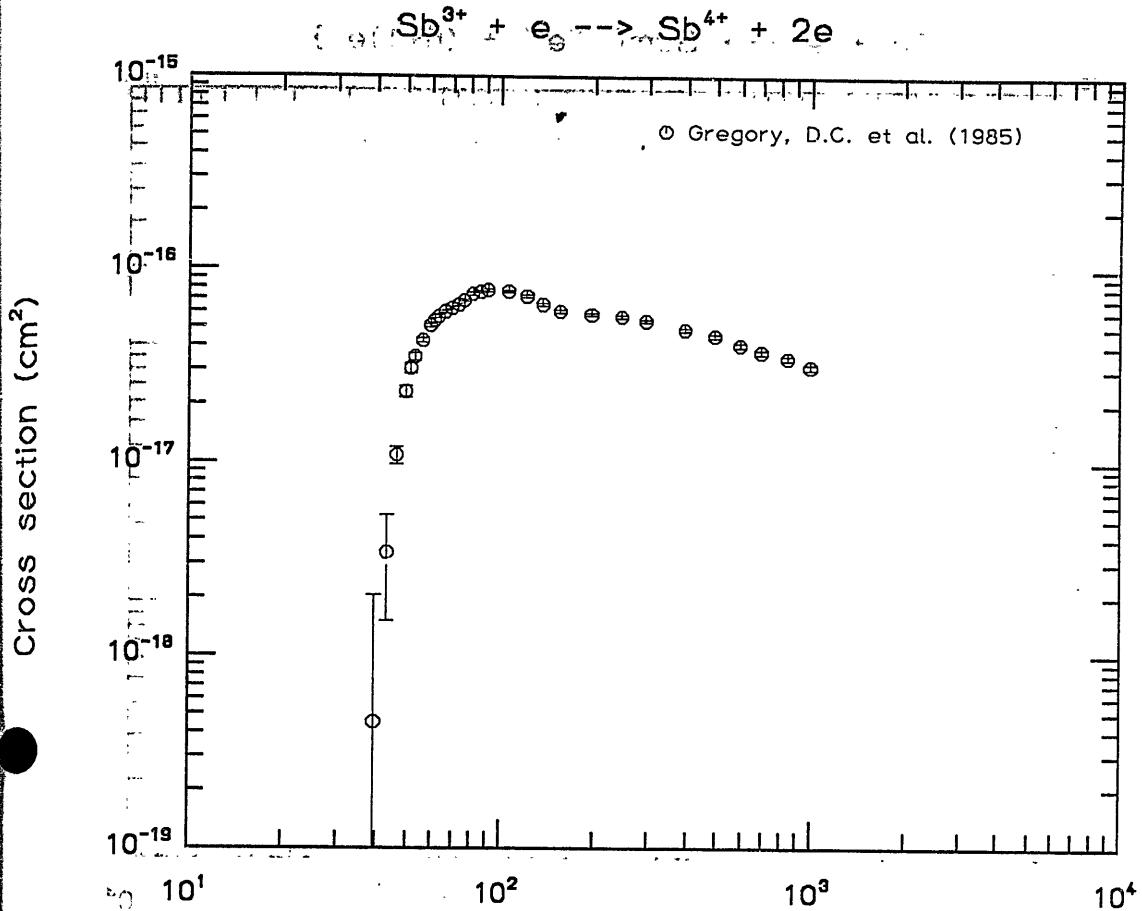


Fig. 223 (a) Electron energy (eV) ACS pH

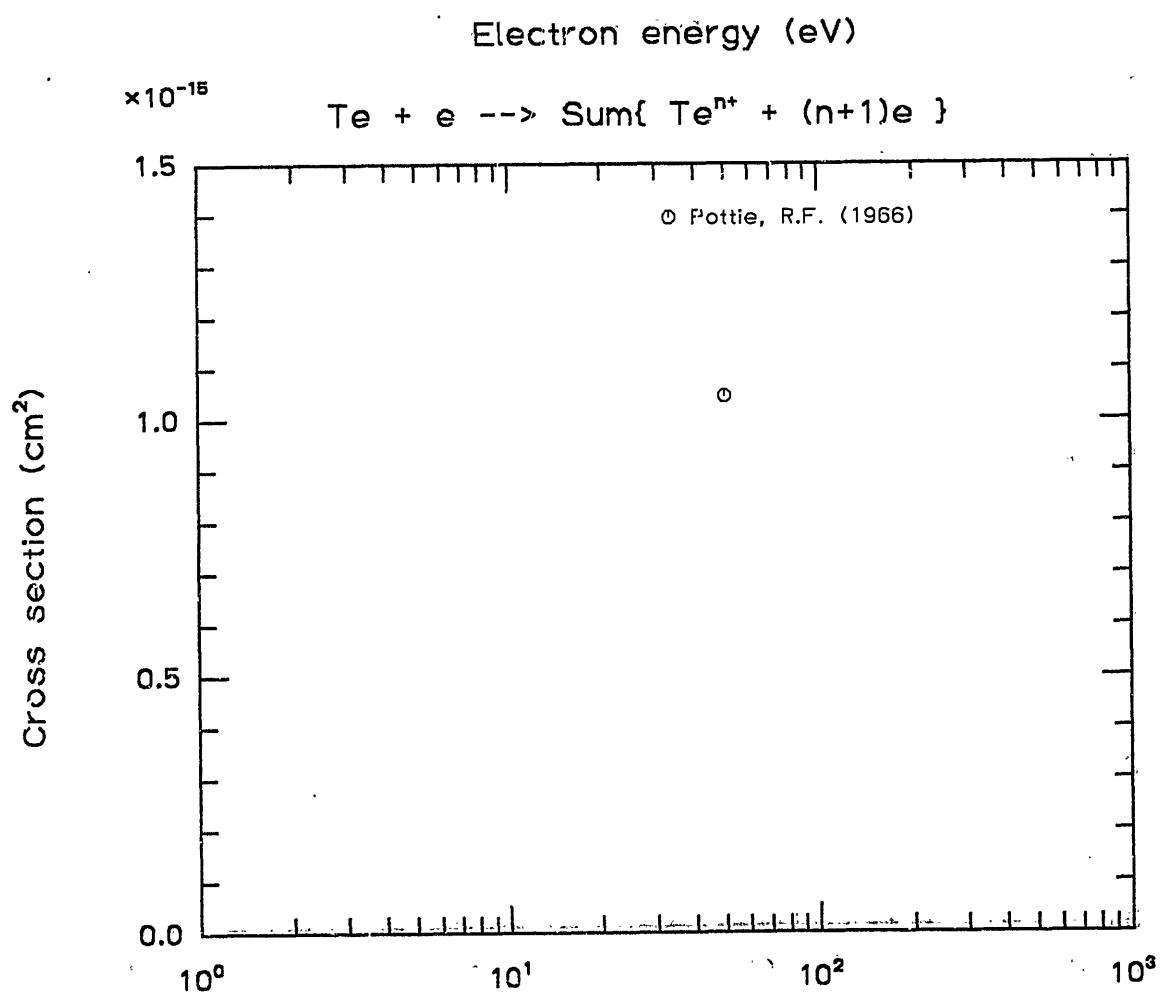
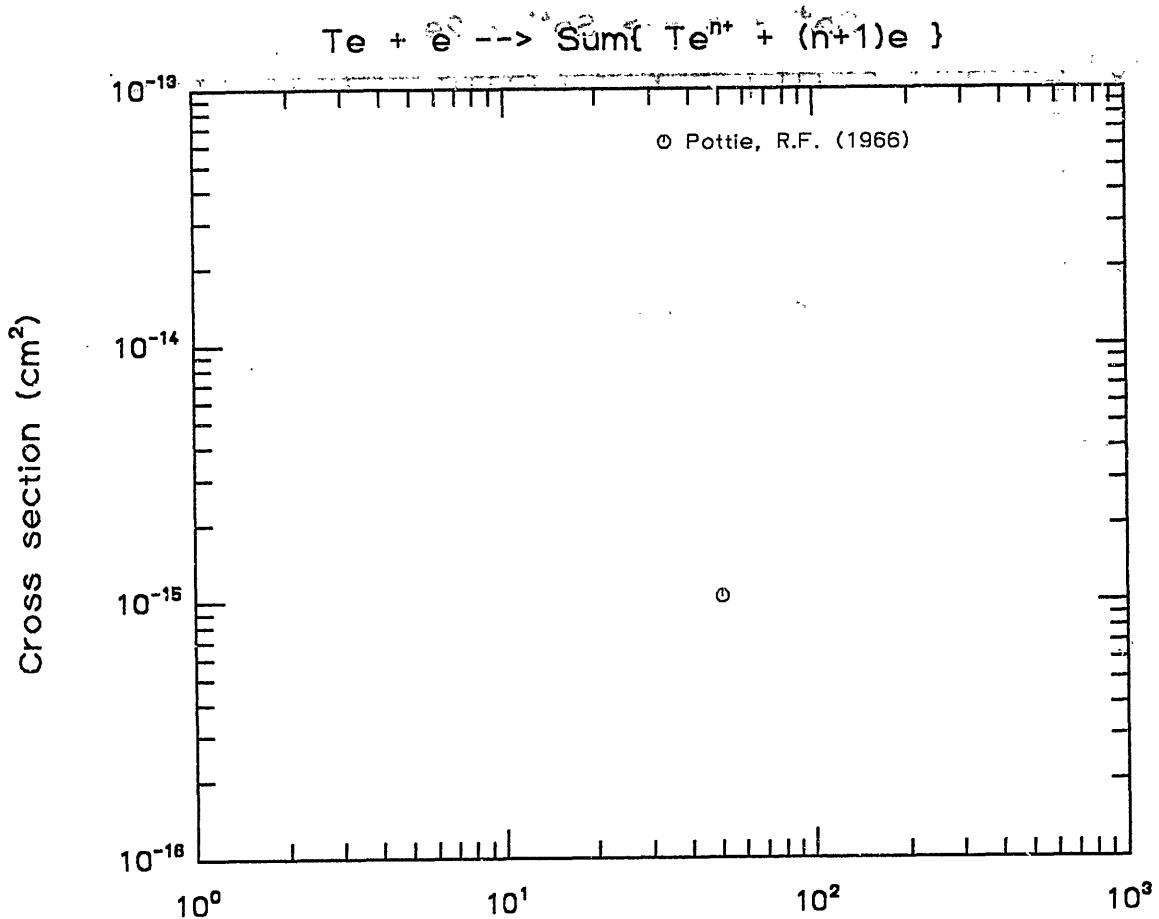


Fig. 224      Electron energy (eV)    ESS . pH

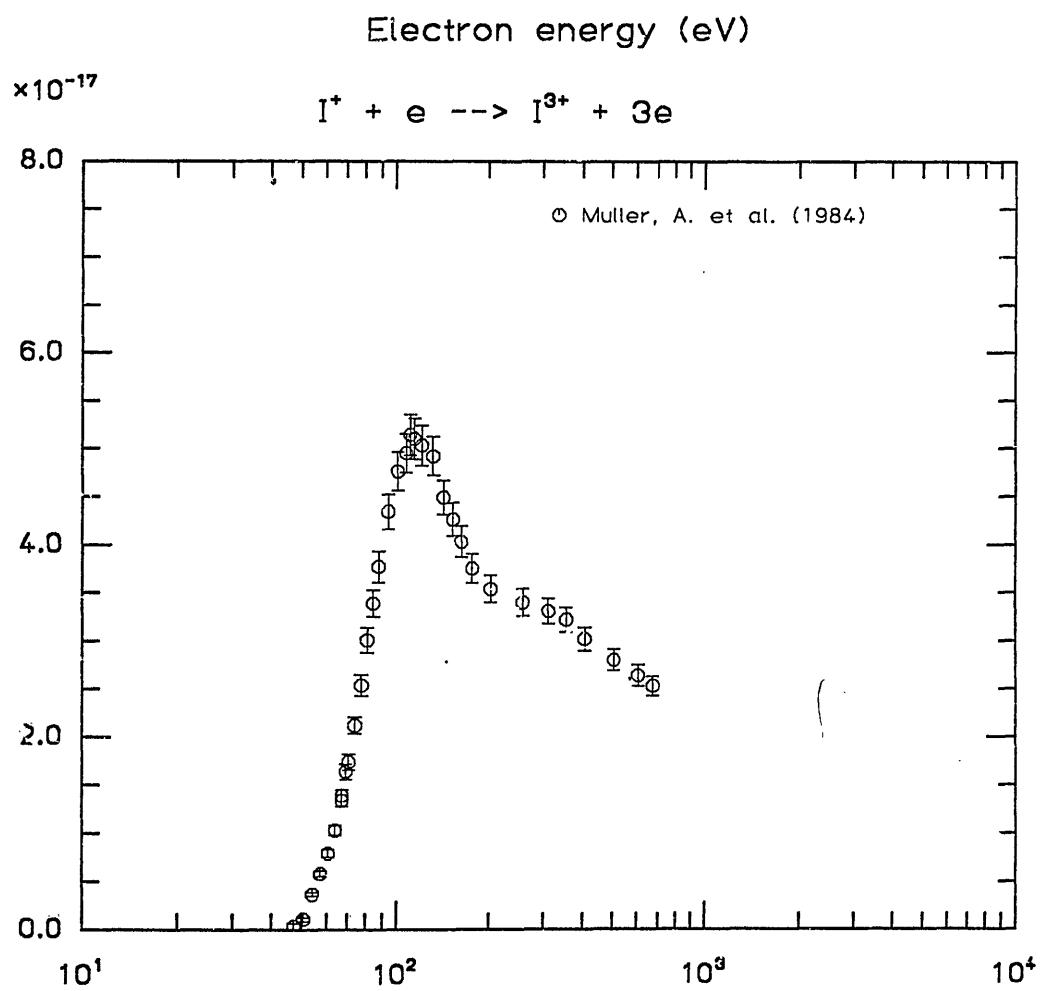
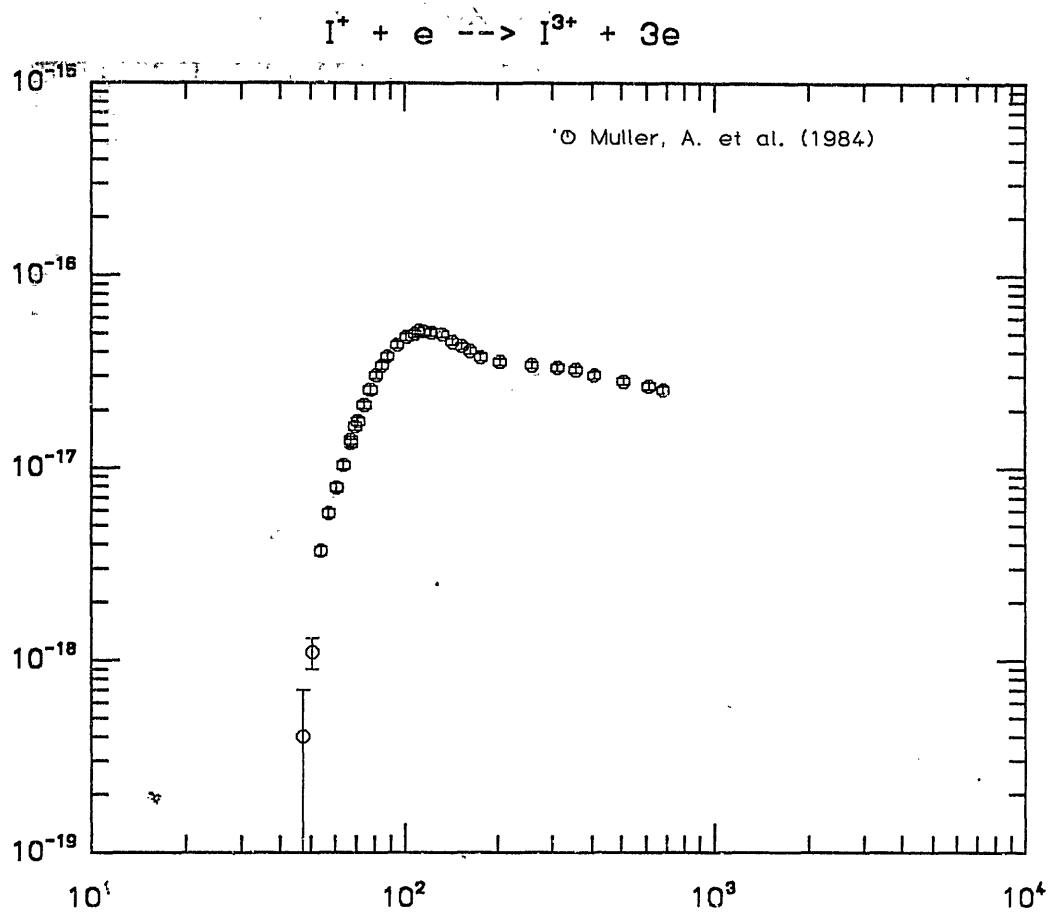


Fig. 225

Electron energy (eV)

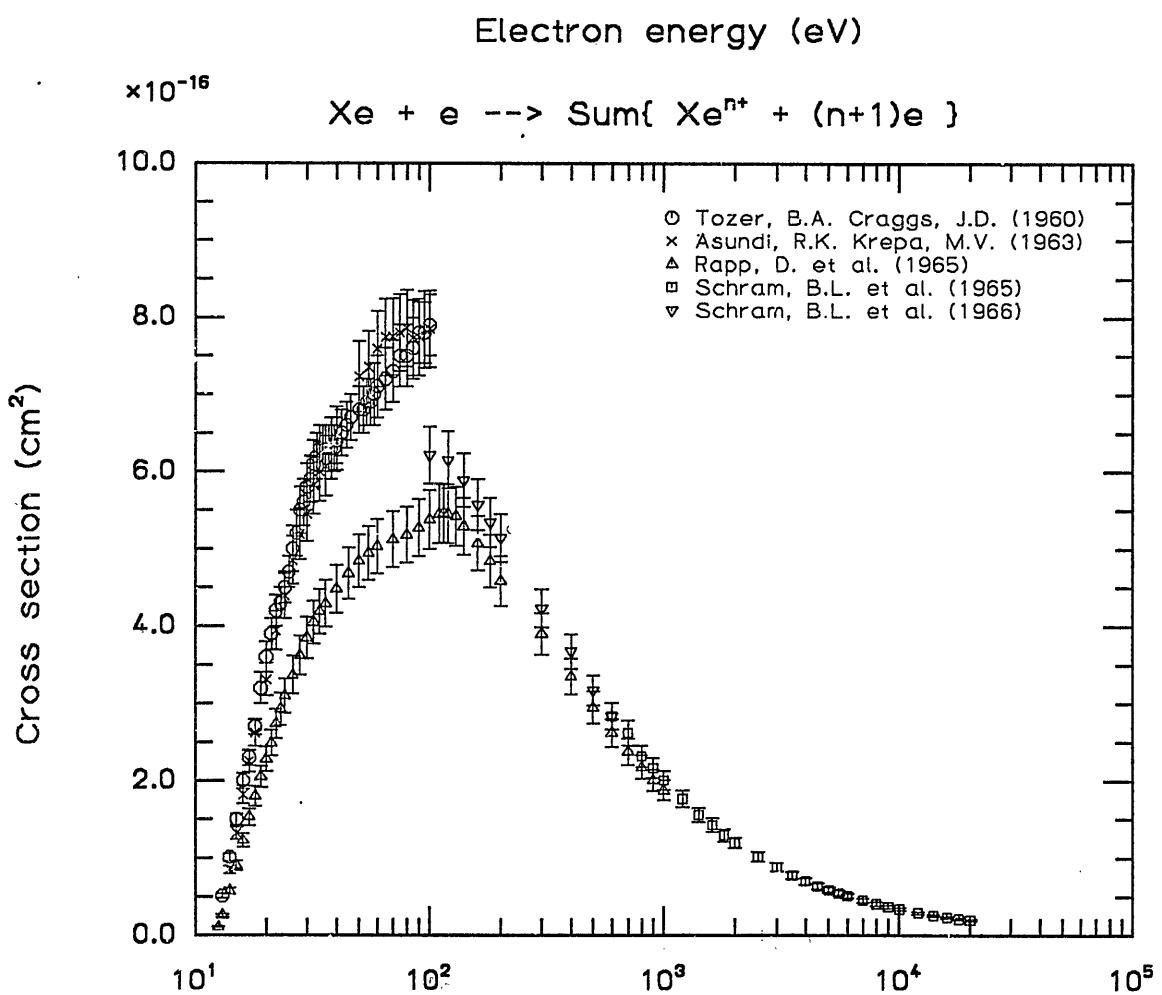
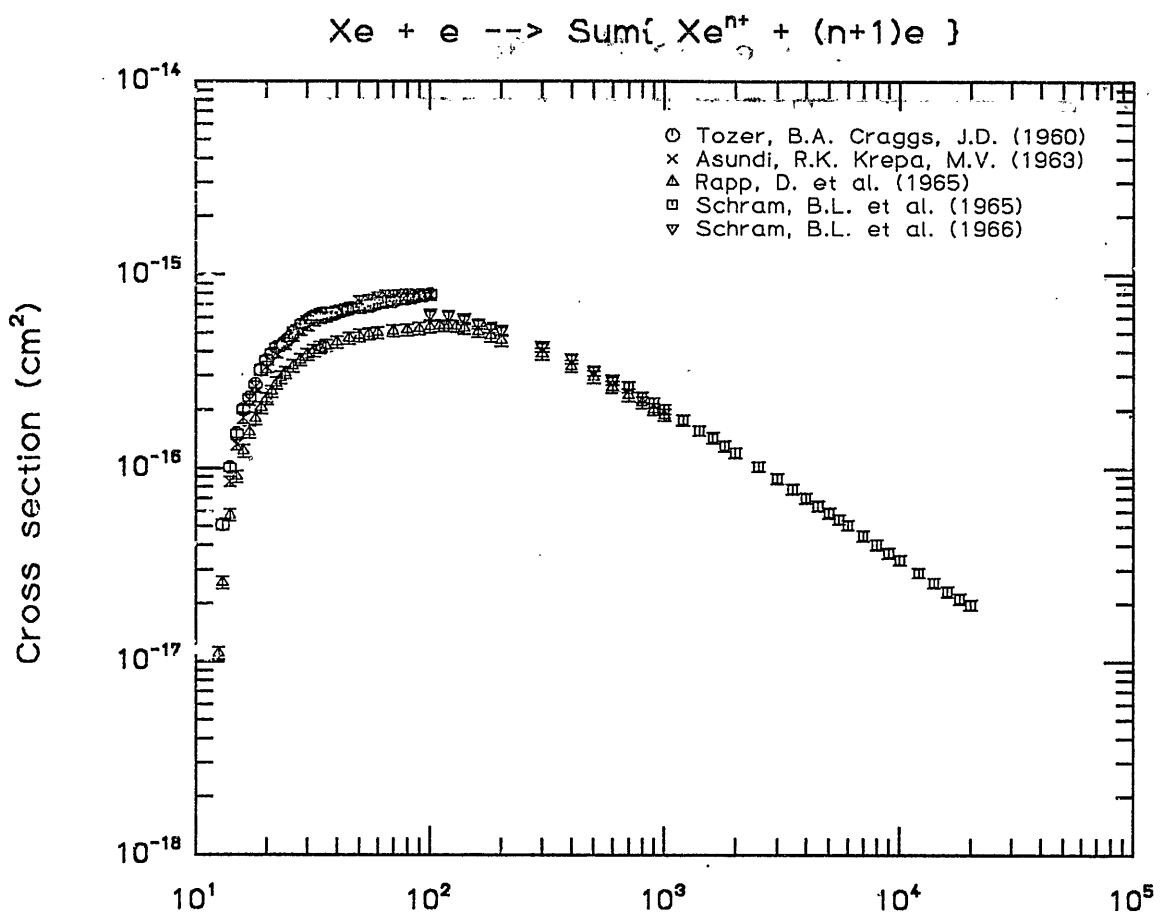
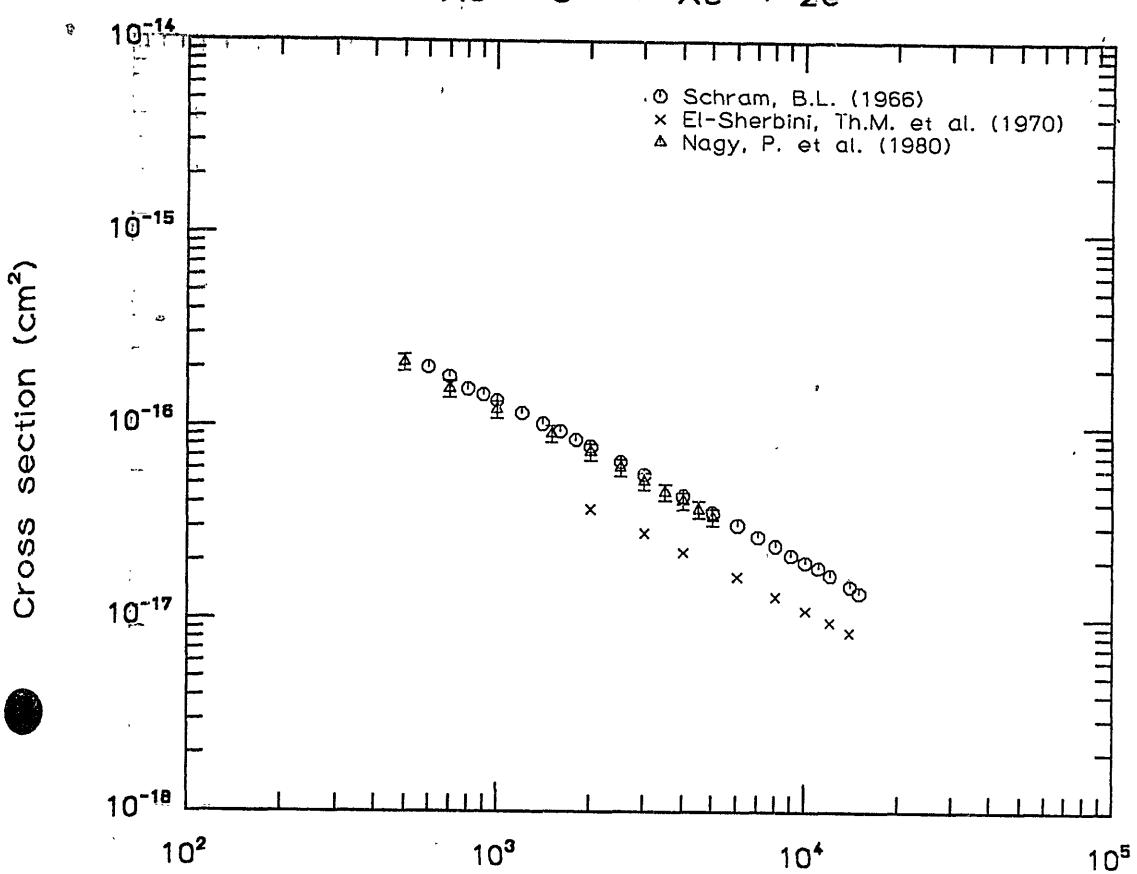
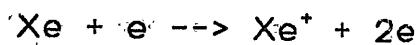


Fig. 226



Electron energy (eV)

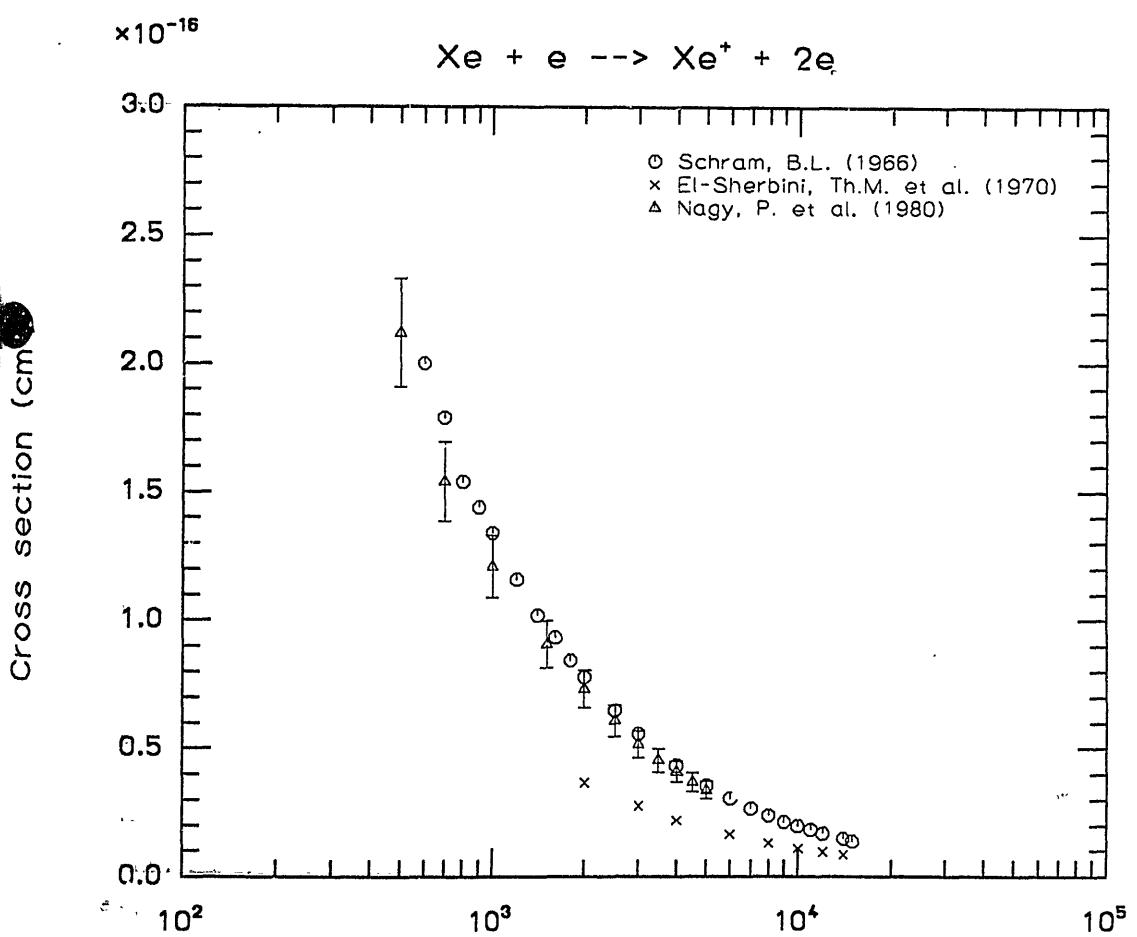


Fig. 227      Electron energy (eV)

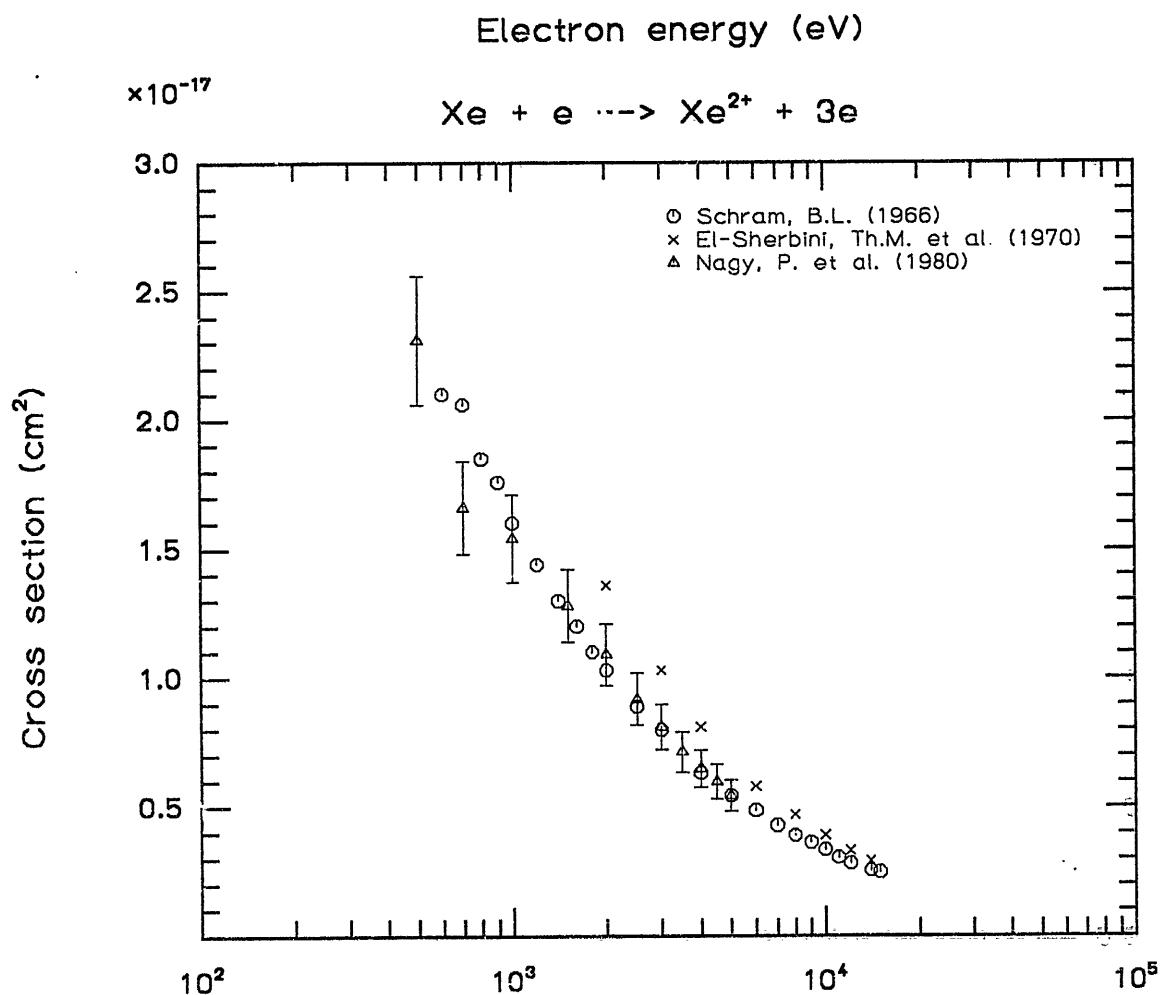
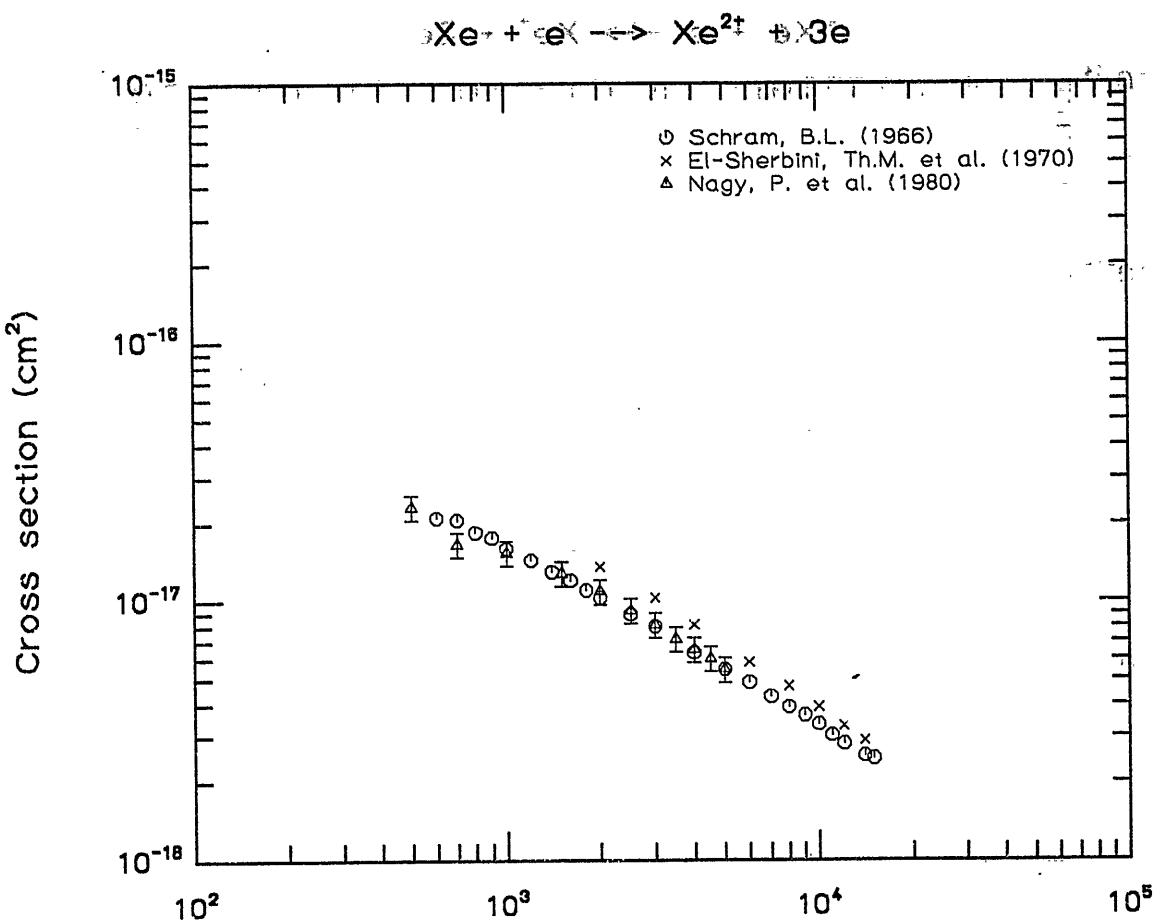


Fig. 228      Electron energy (eV)      10<sup>2</sup>      10<sup>3</sup>      10<sup>4</sup>

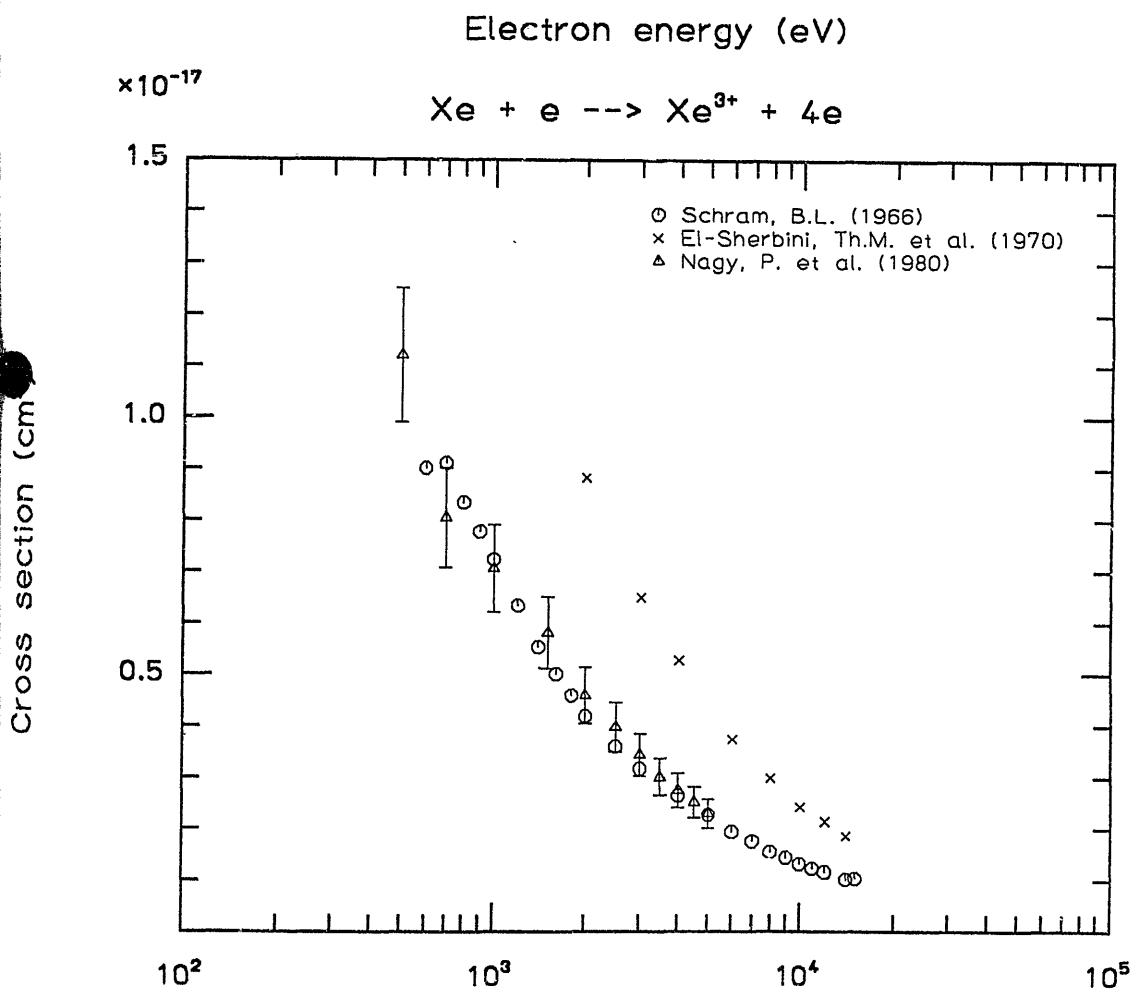
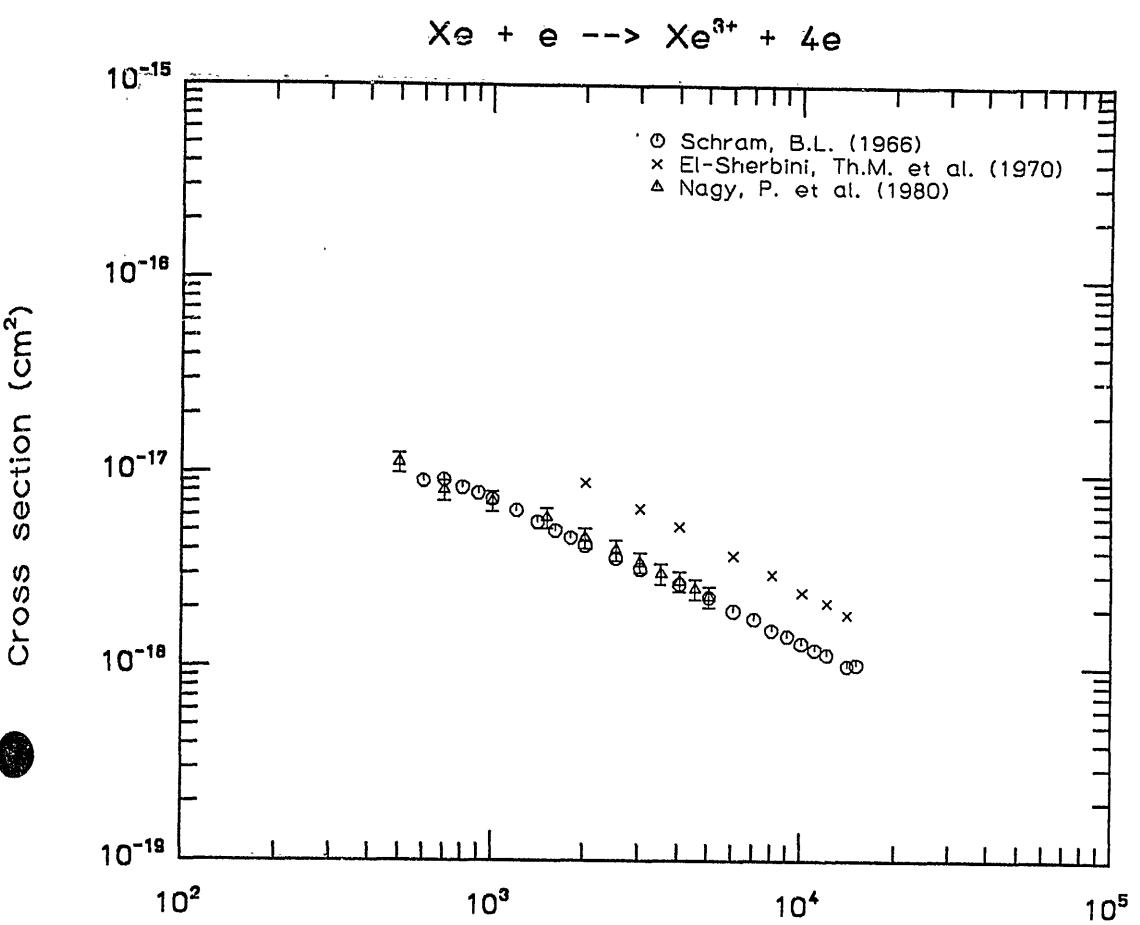


Fig. 229

Electron energy (eV)

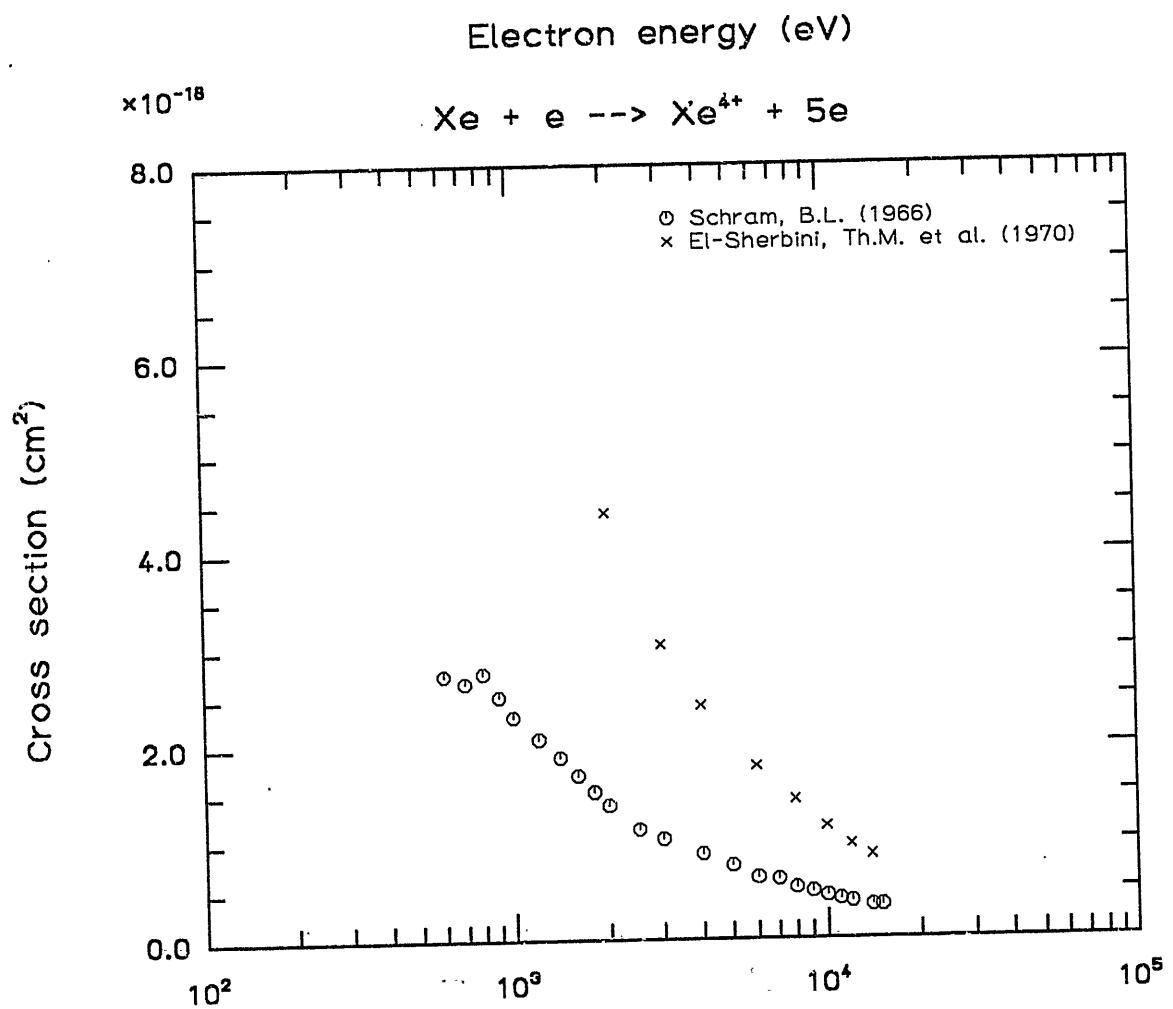
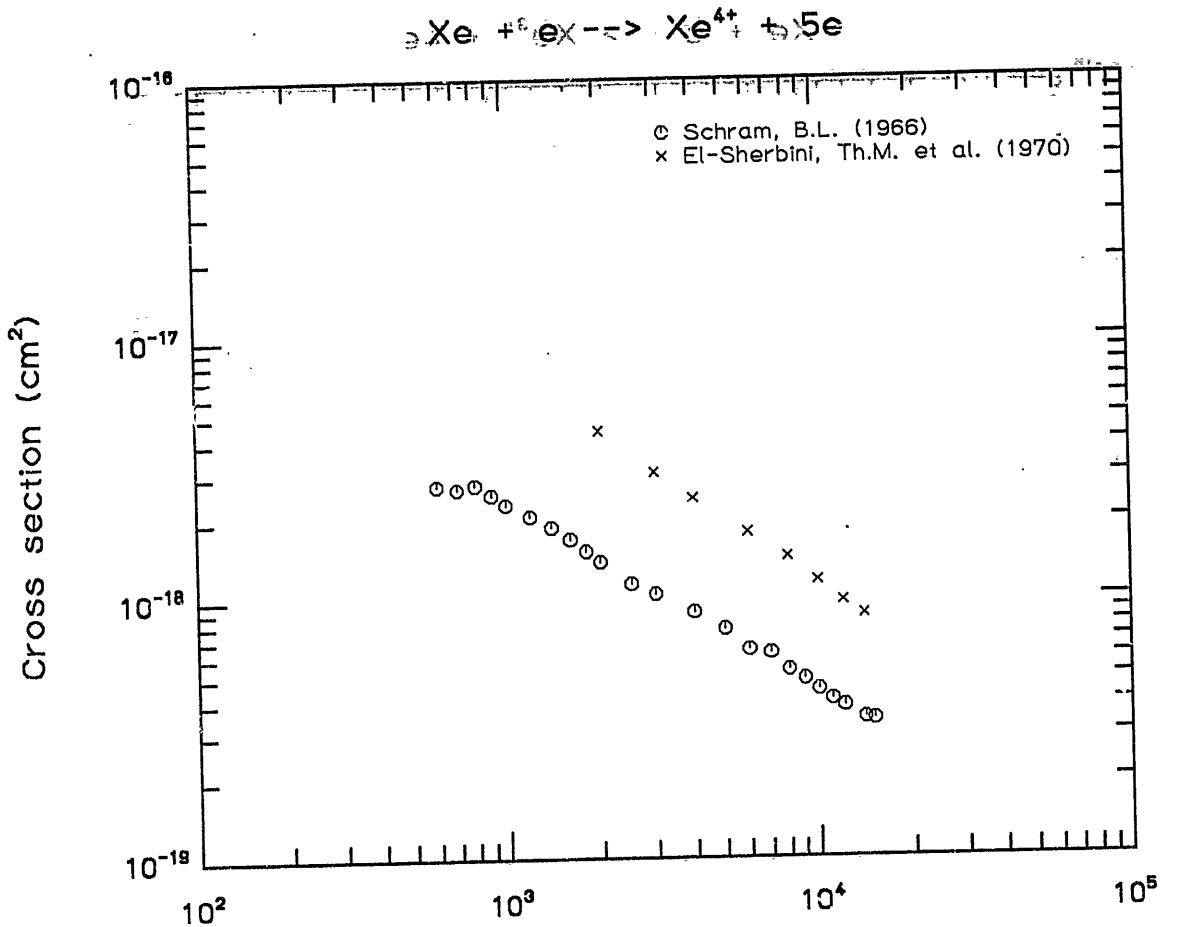


Fig. 230 (a) Electron energy (eV) 0.55-10<sup>5</sup>

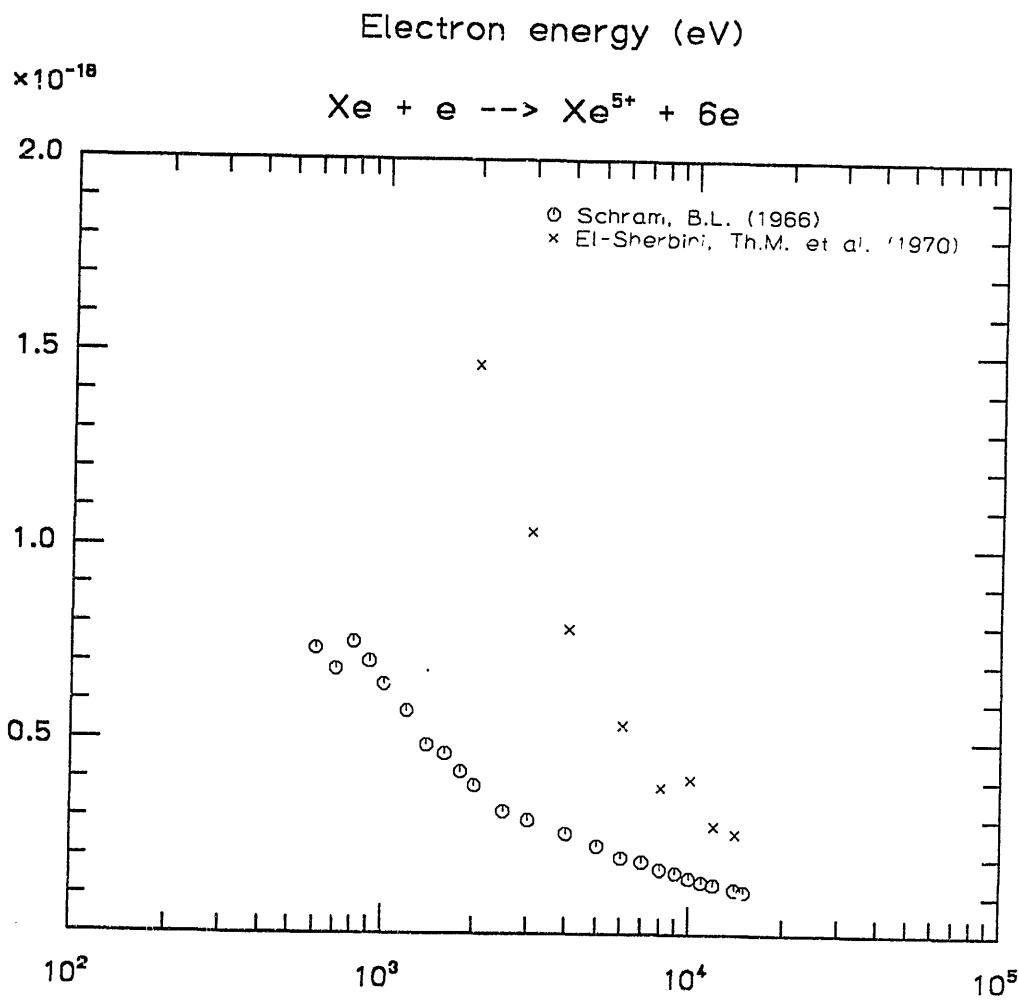
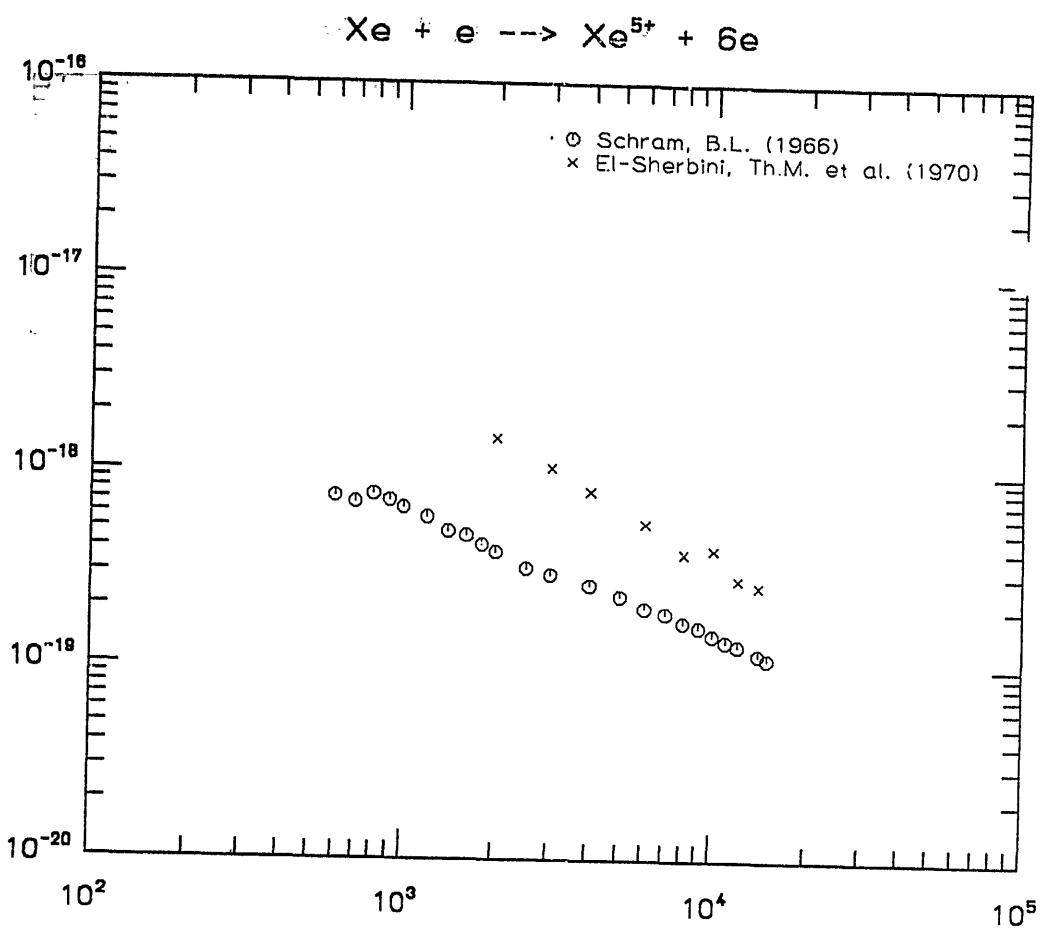


Fig. 231      Electron energy (eV)

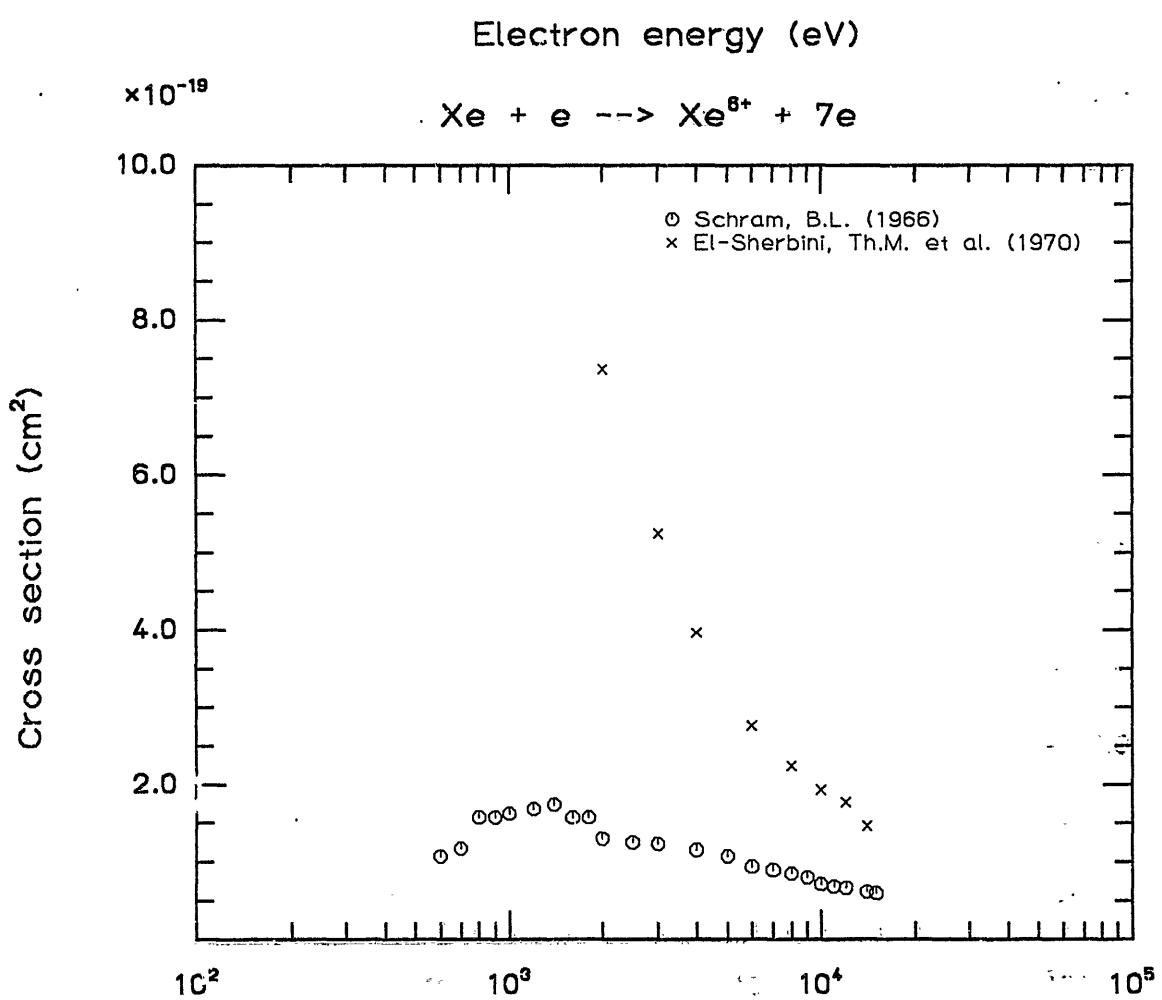
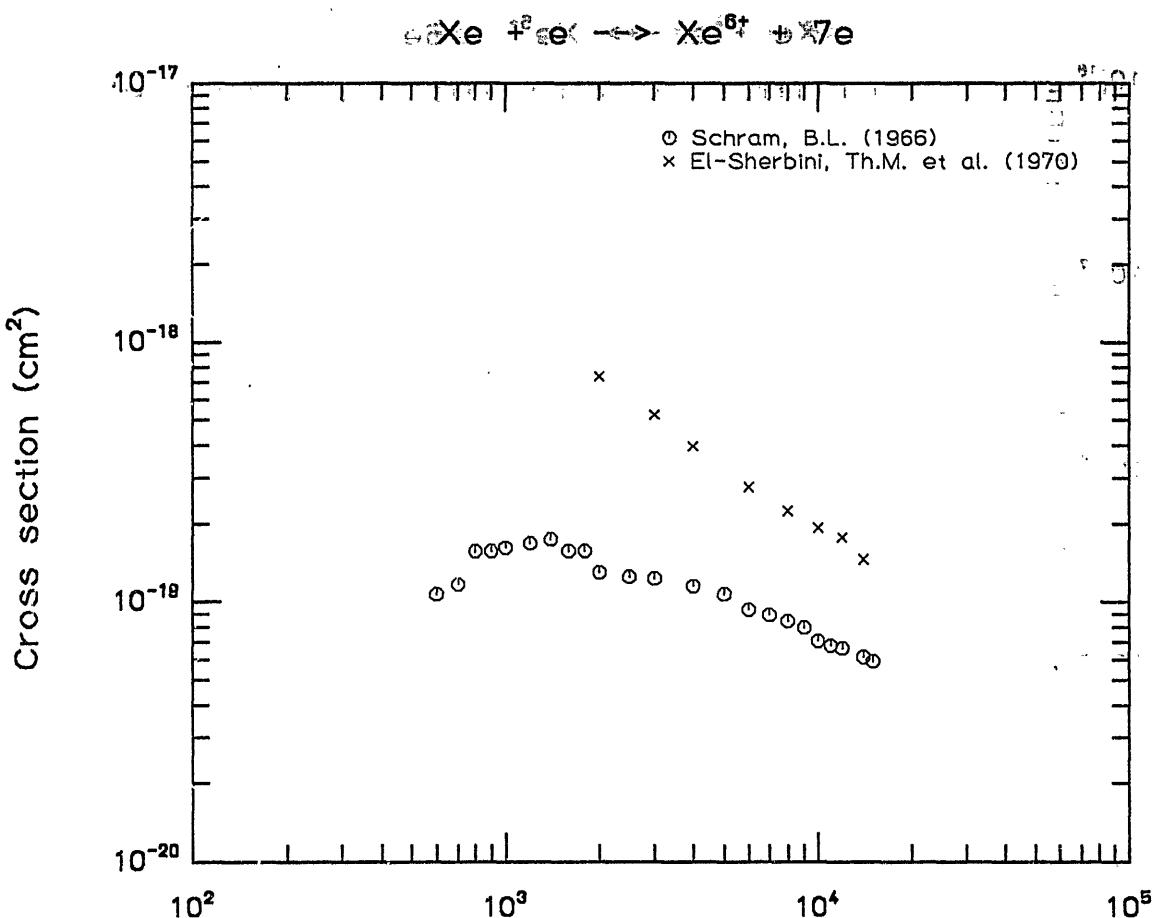
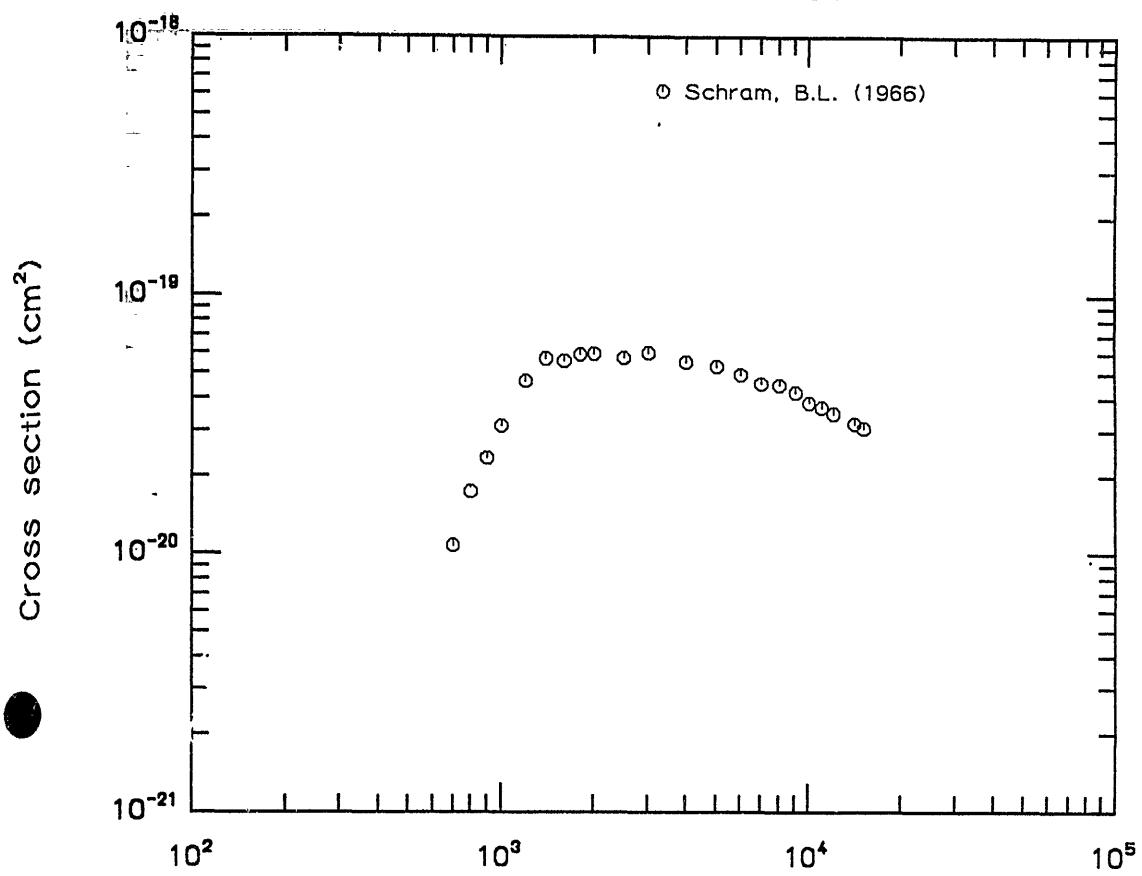
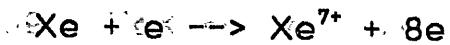


Fig. 232 (a) Electron energy (eV) Fig. 232 (b) Electron energy (eV)



Electron energy (eV)

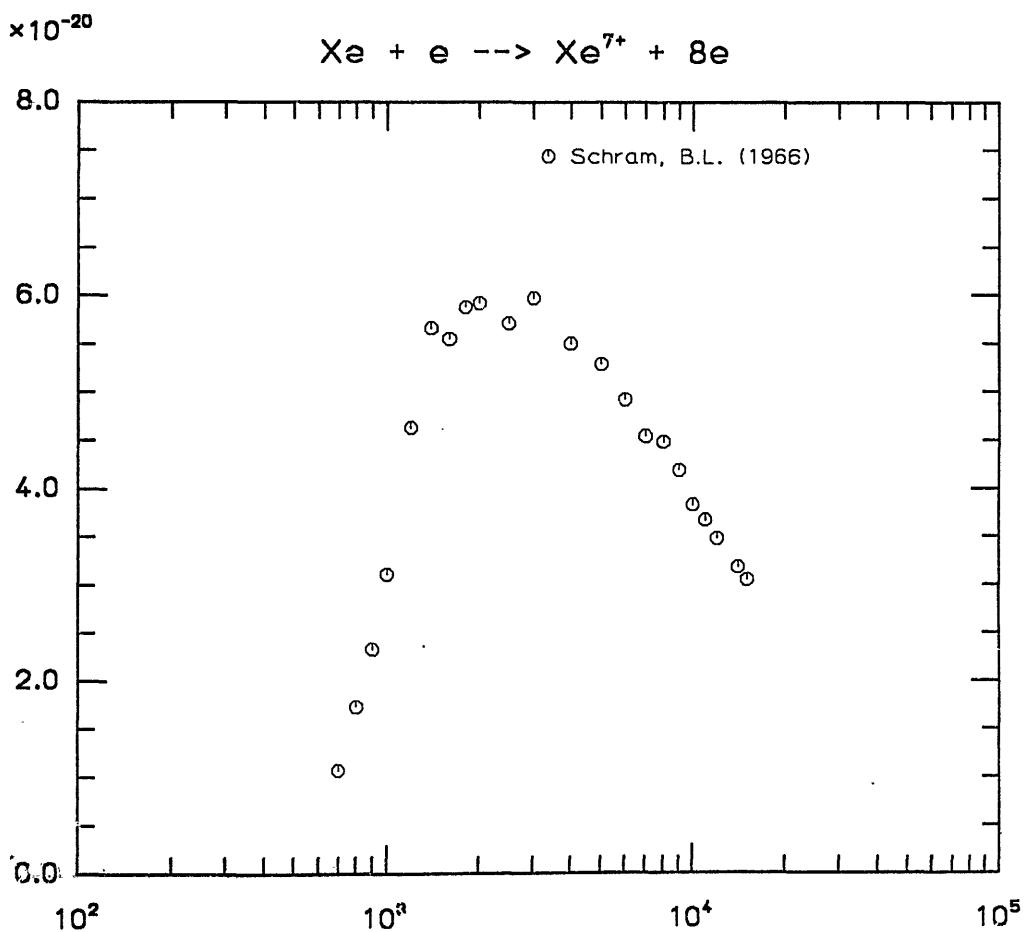


Fig. 233 . Electron energy (eV)

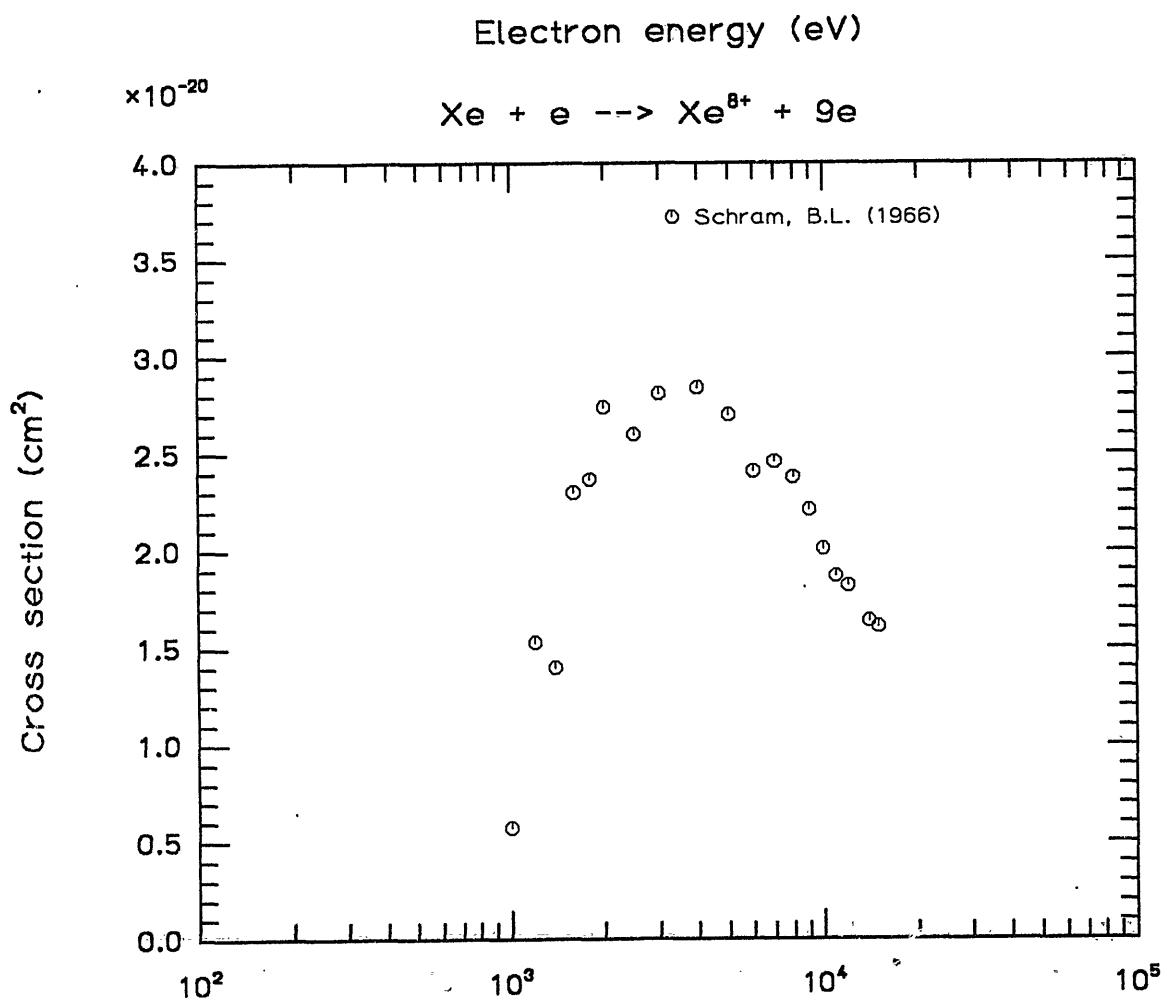
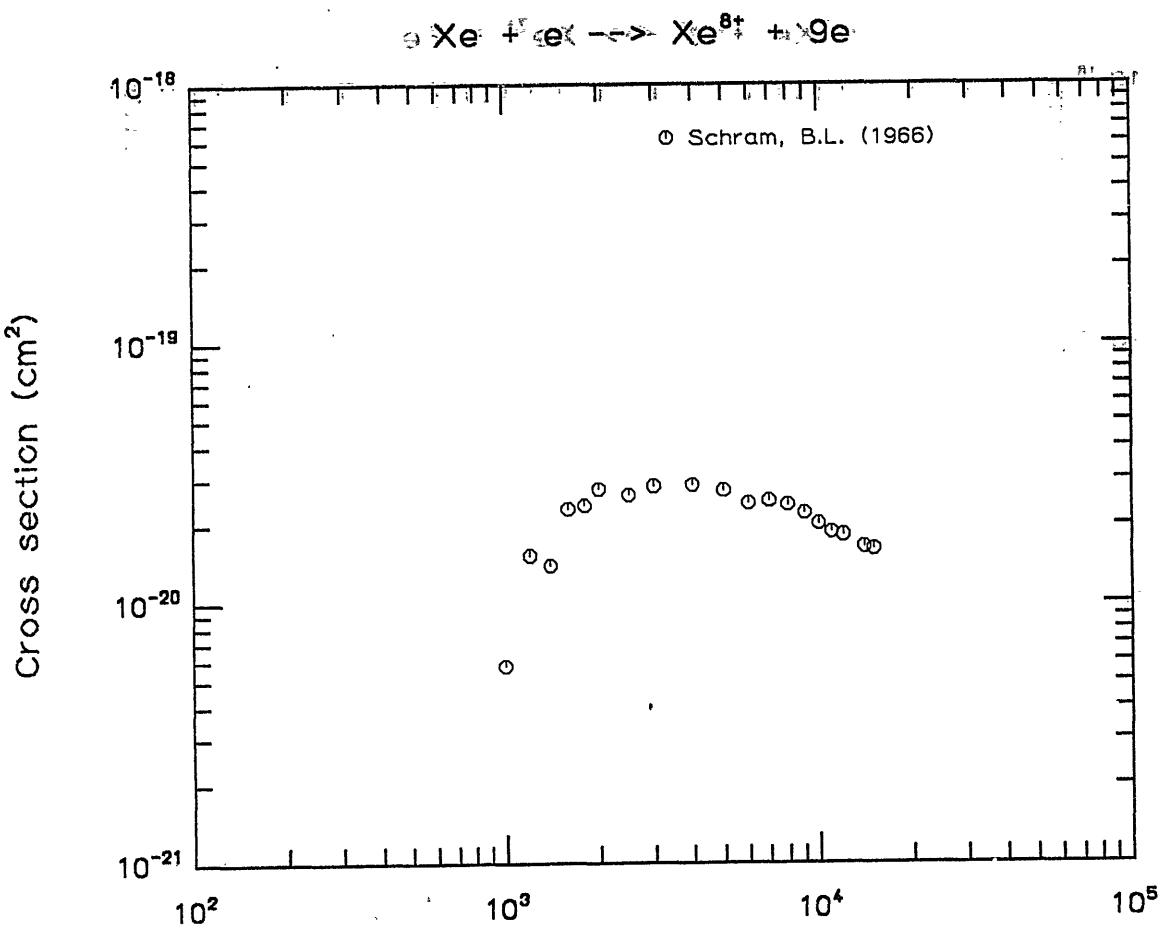
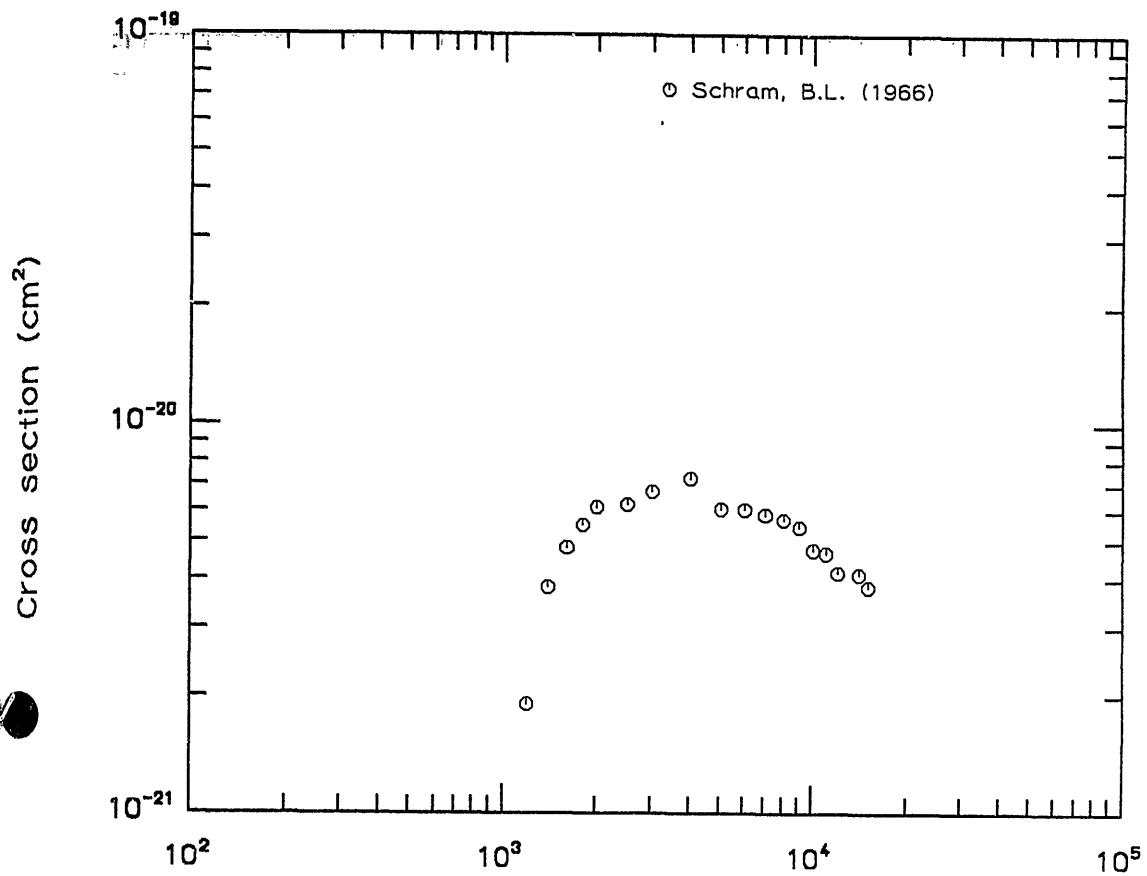
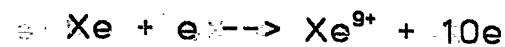


Fig. 234. Cross section vs. Electron energy (eV) for  $\text{Xe} + \text{e} \rightarrow \text{Xe}^{8+} + 9\text{e}$



Electron energy (eV)

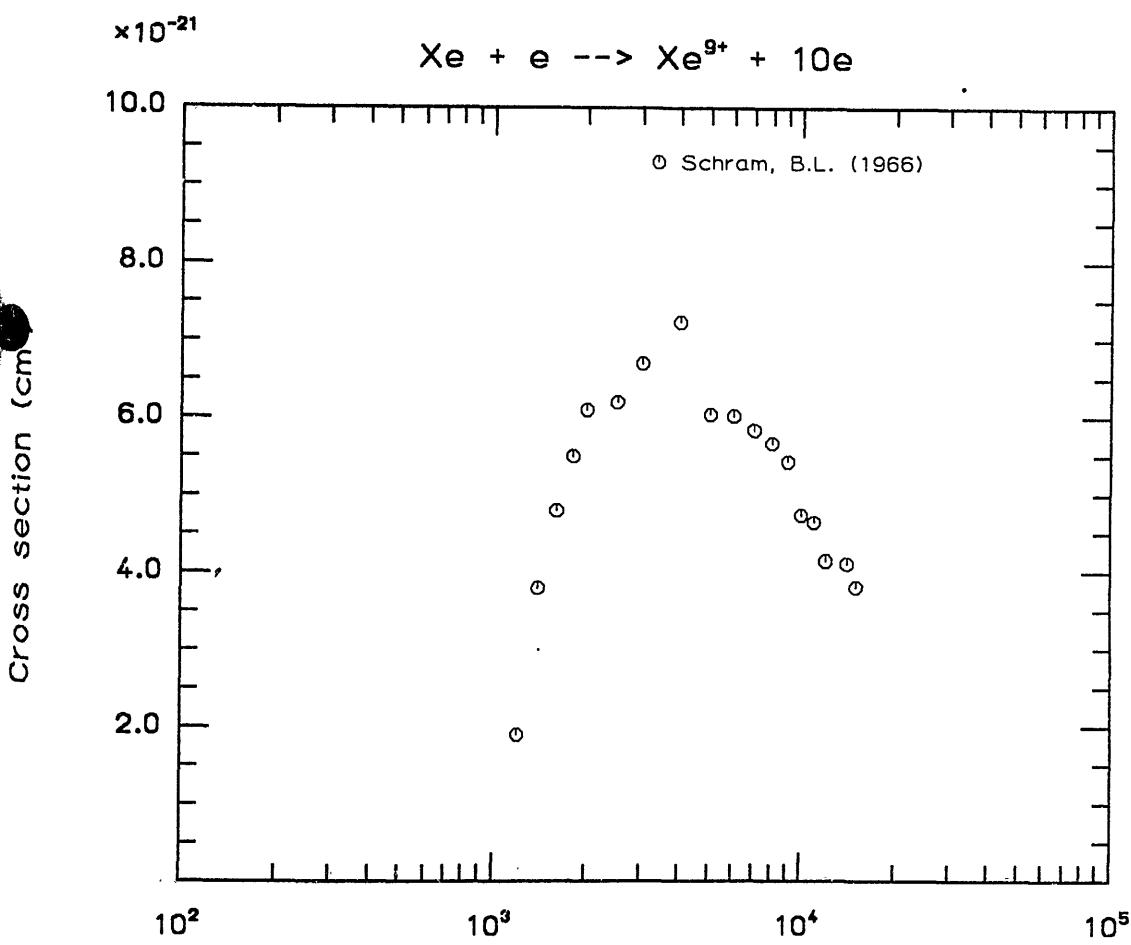


Fig. 235      Electron energy (eV)

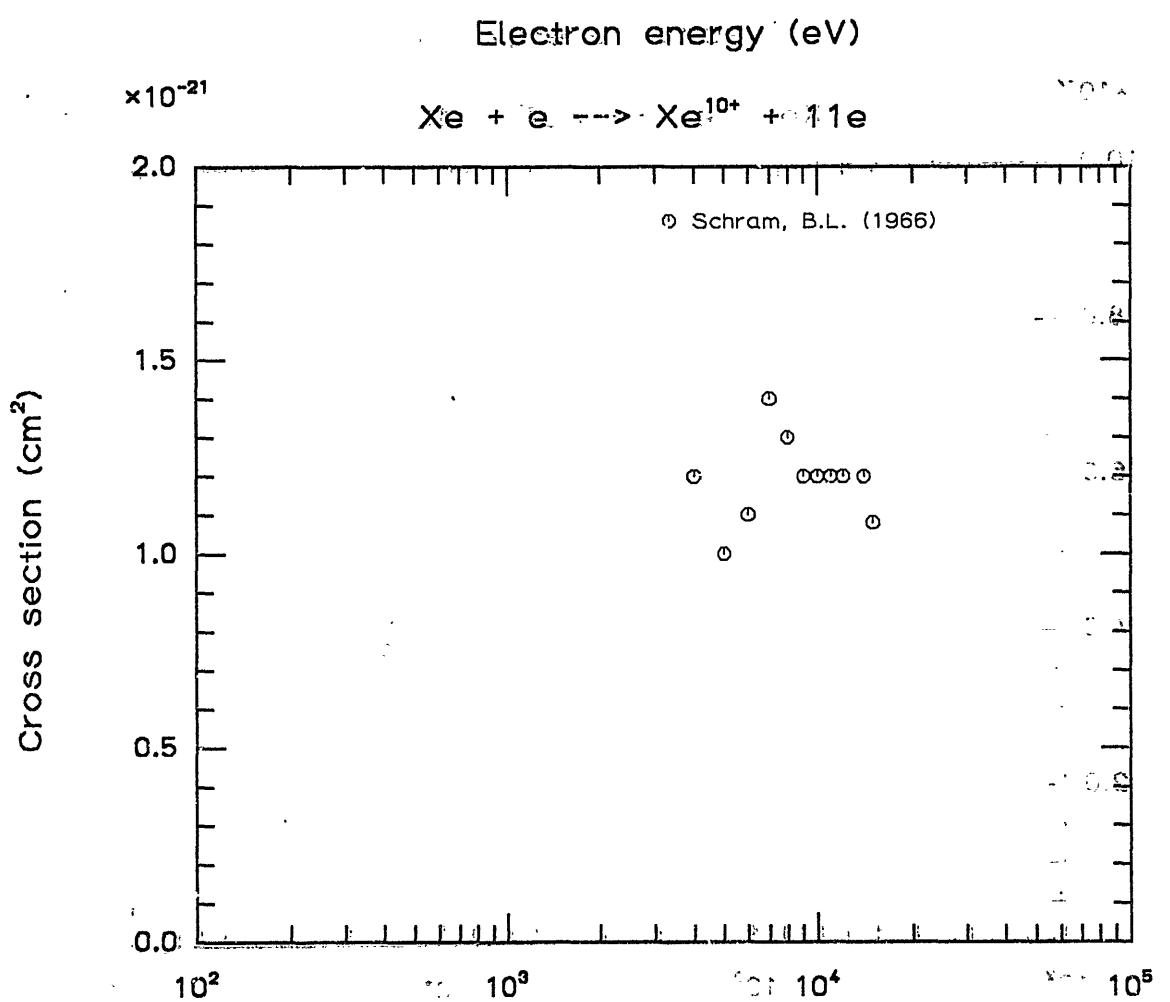
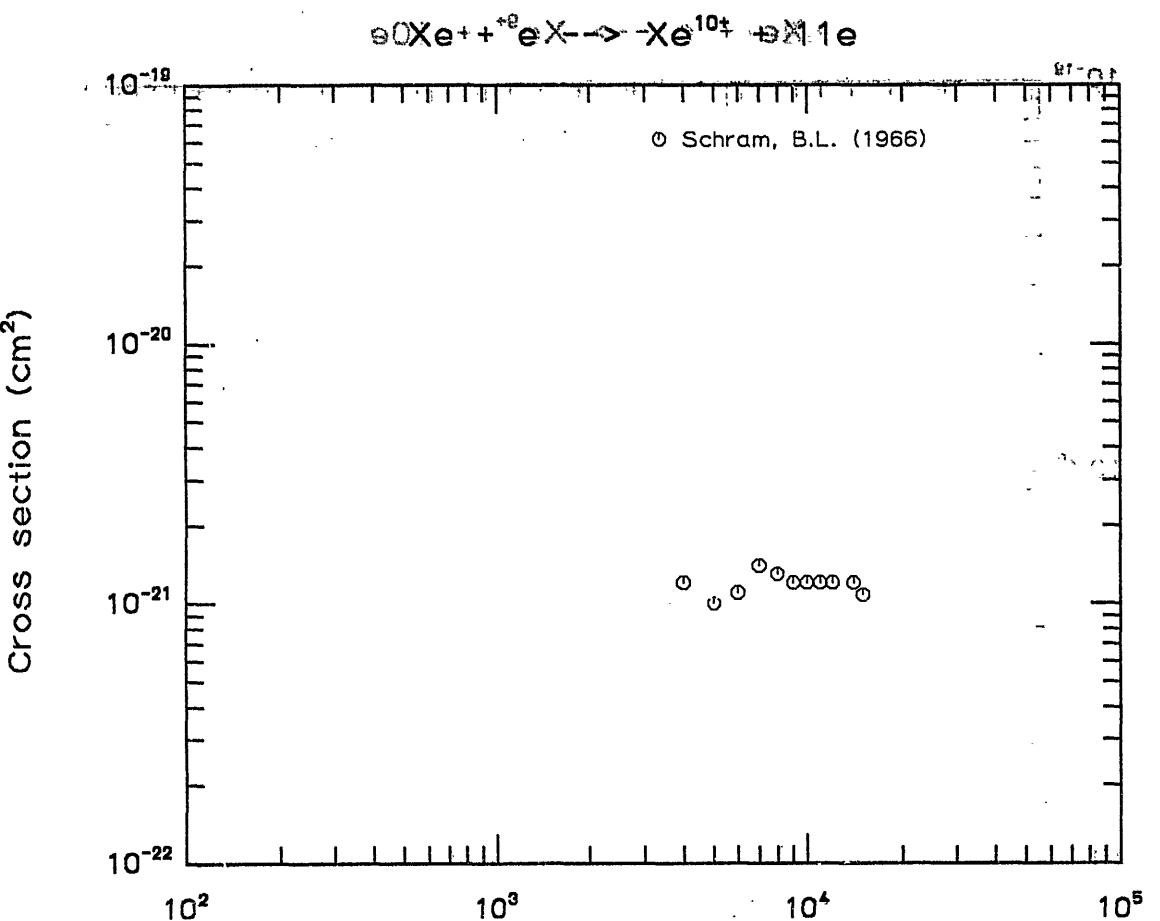
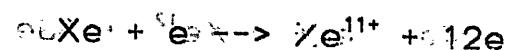
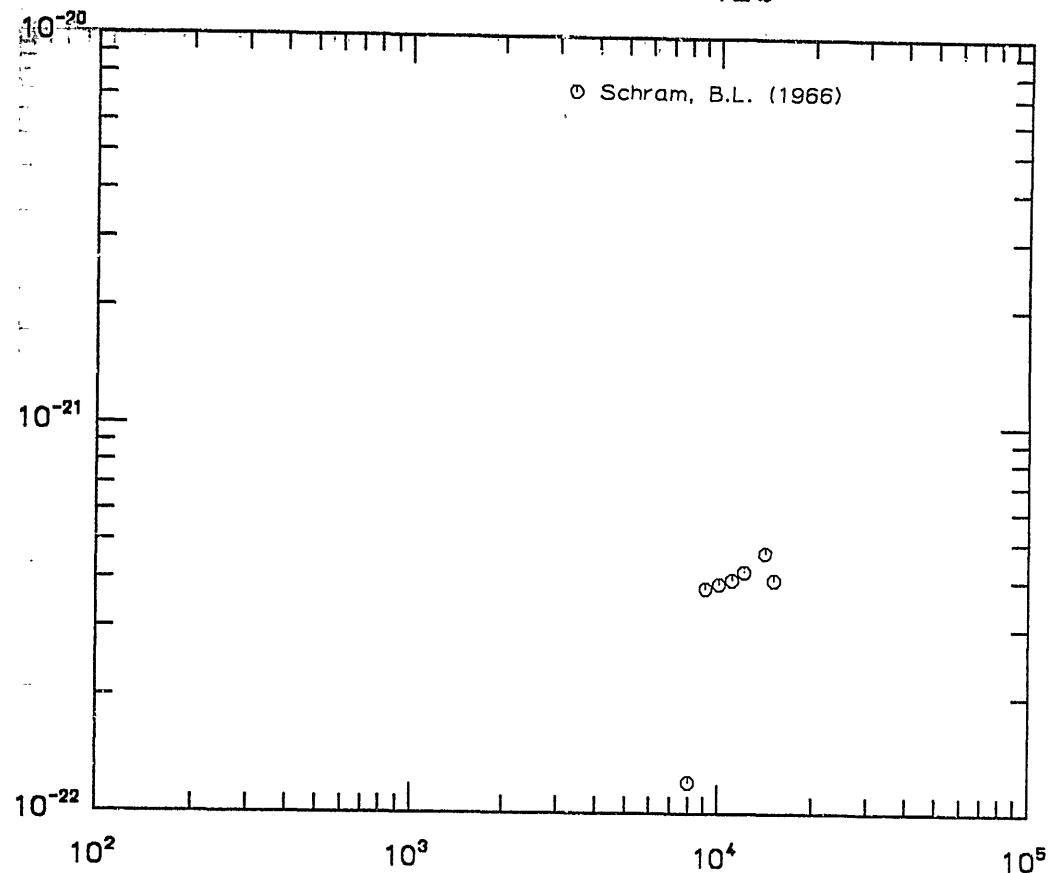


Fig. 236 (Top) Electron energy (eV) Fig. 236 (Bottom)



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

Cross section ( $\text{cm}^2$ )

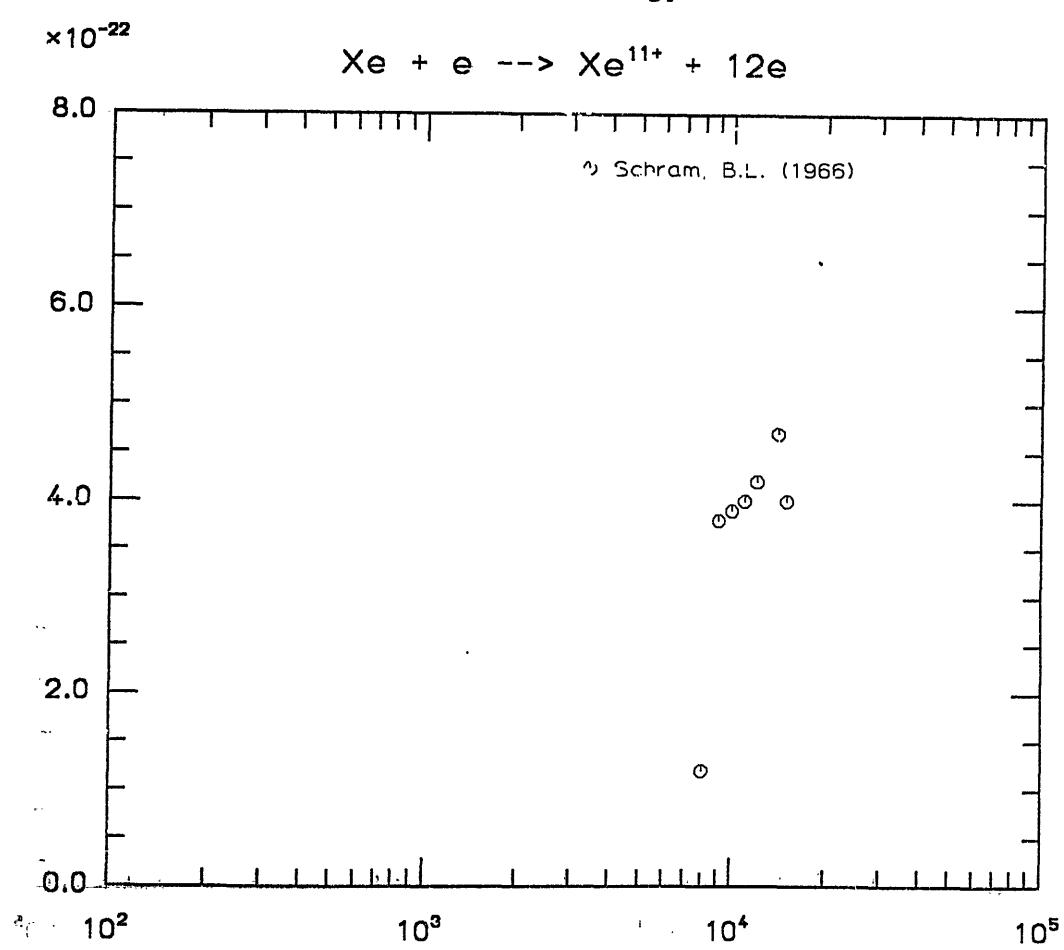


Fig. 237 (v) Electron energy (eV) 88 N D. A.

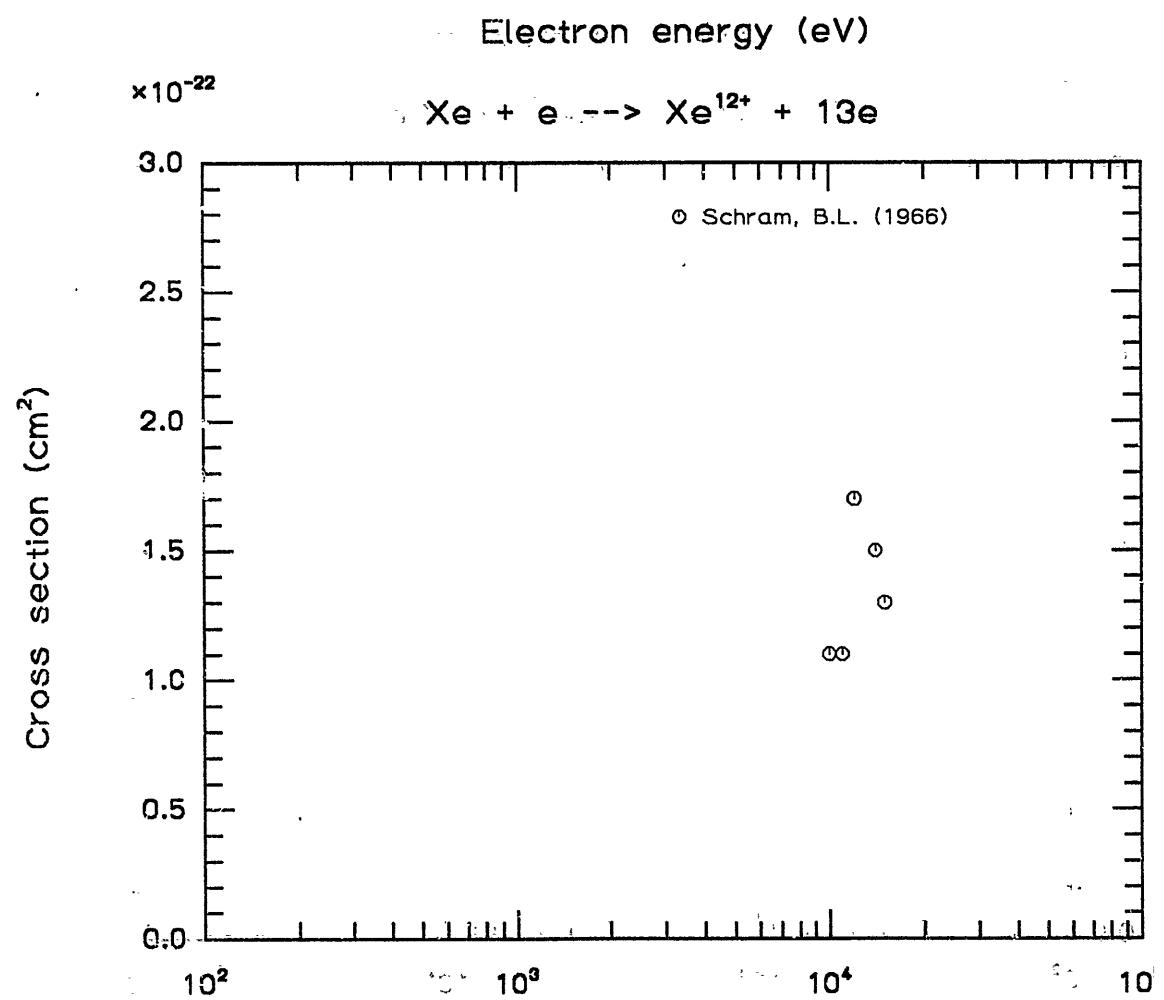
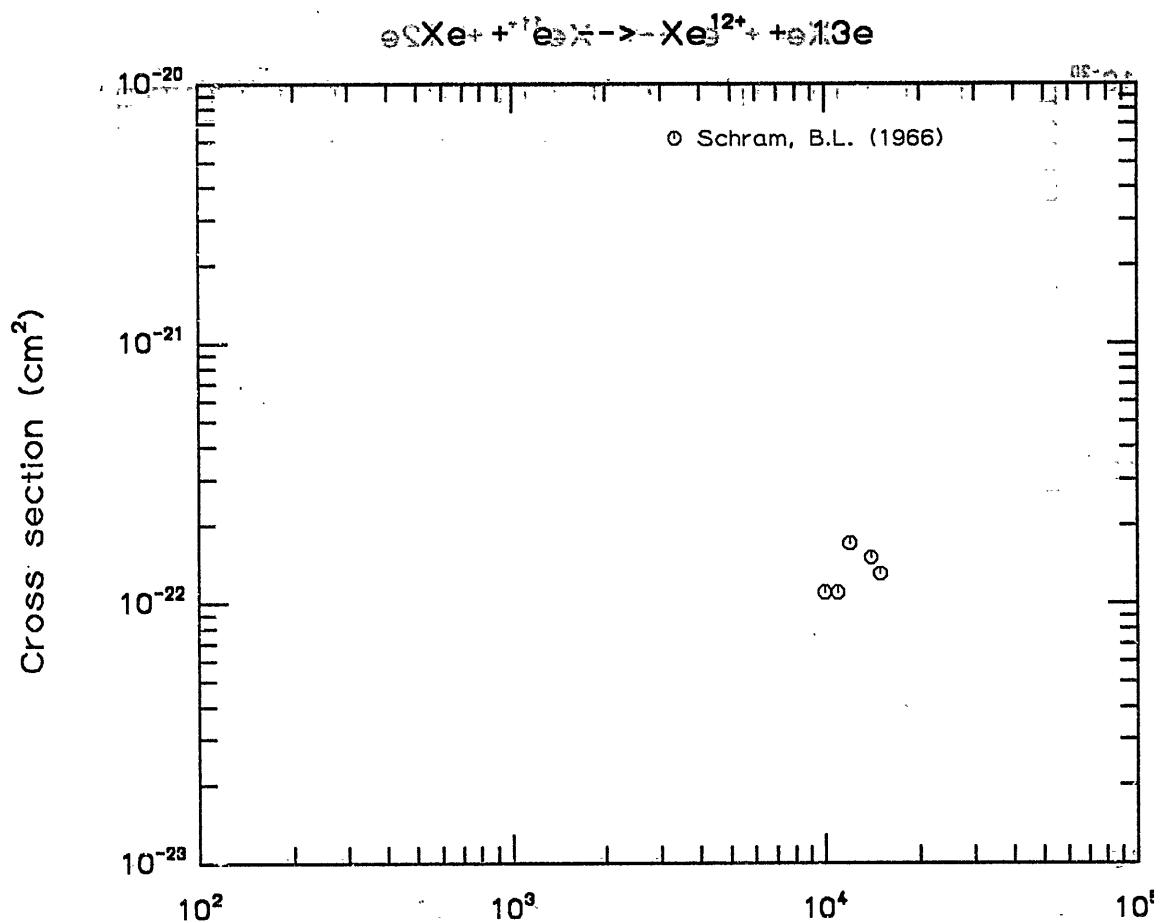
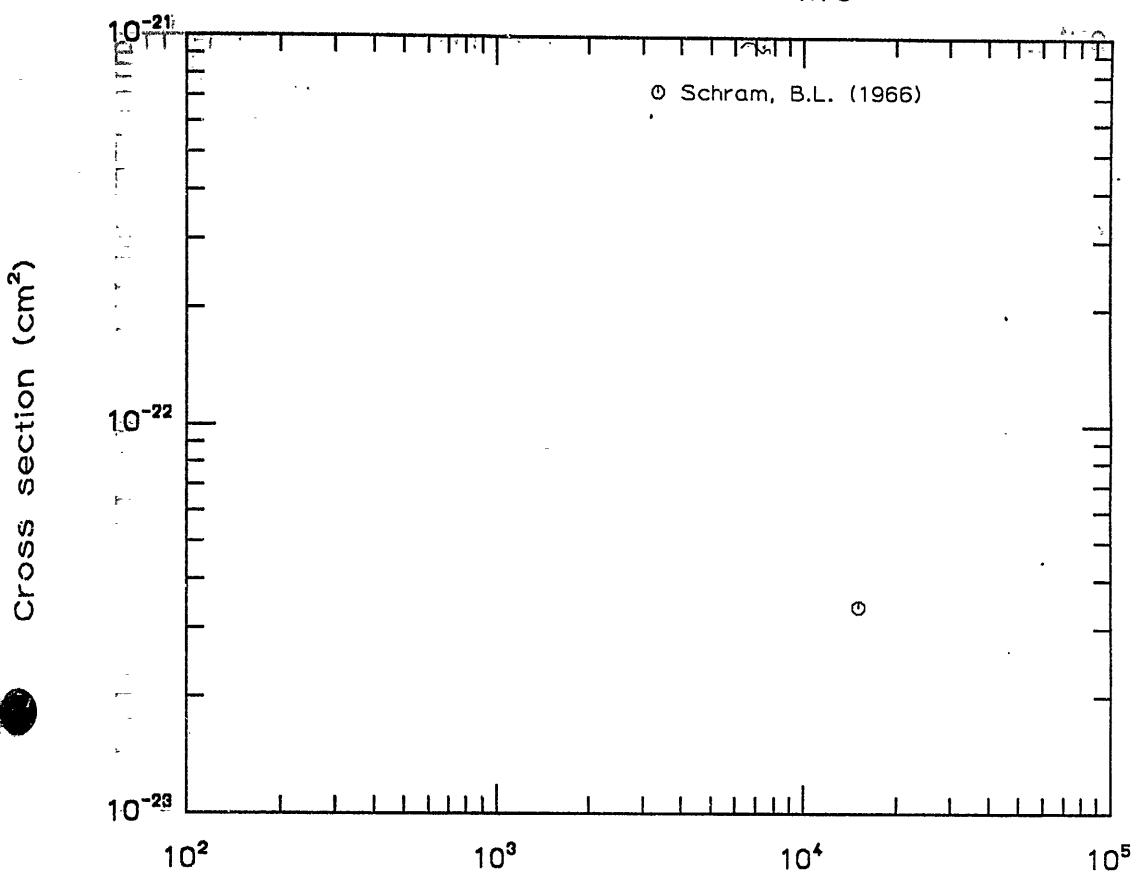
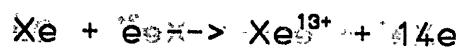


Fig. 238. Cross section vs. Electron energy (eV)  $Xe + e^- \rightarrow Xe^{12+} + 13e^-$



Electron energy (eV)

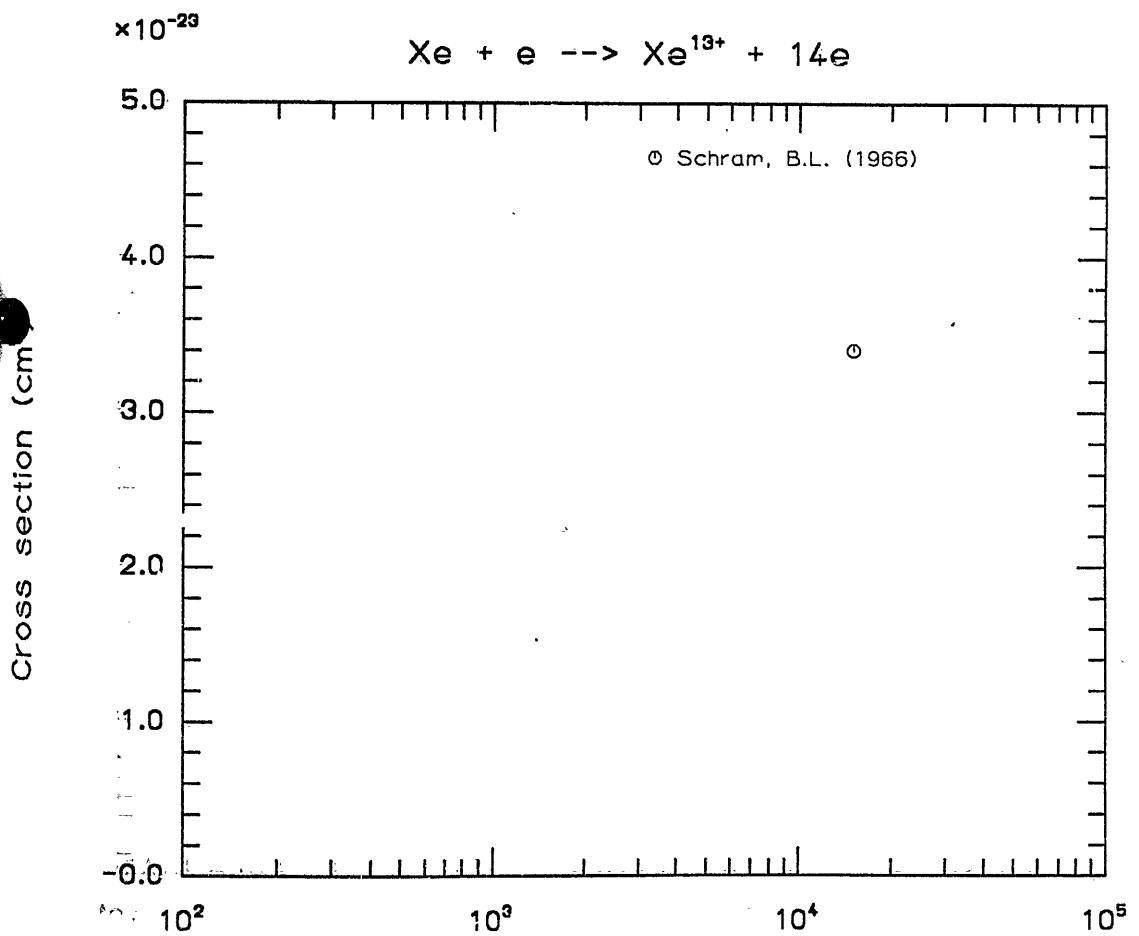
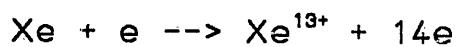


Fig. 239      Electron energy (eV)       $\text{Xe} + \text{e}^- \rightarrow \text{Xe}^{13+} + 14\text{e}$

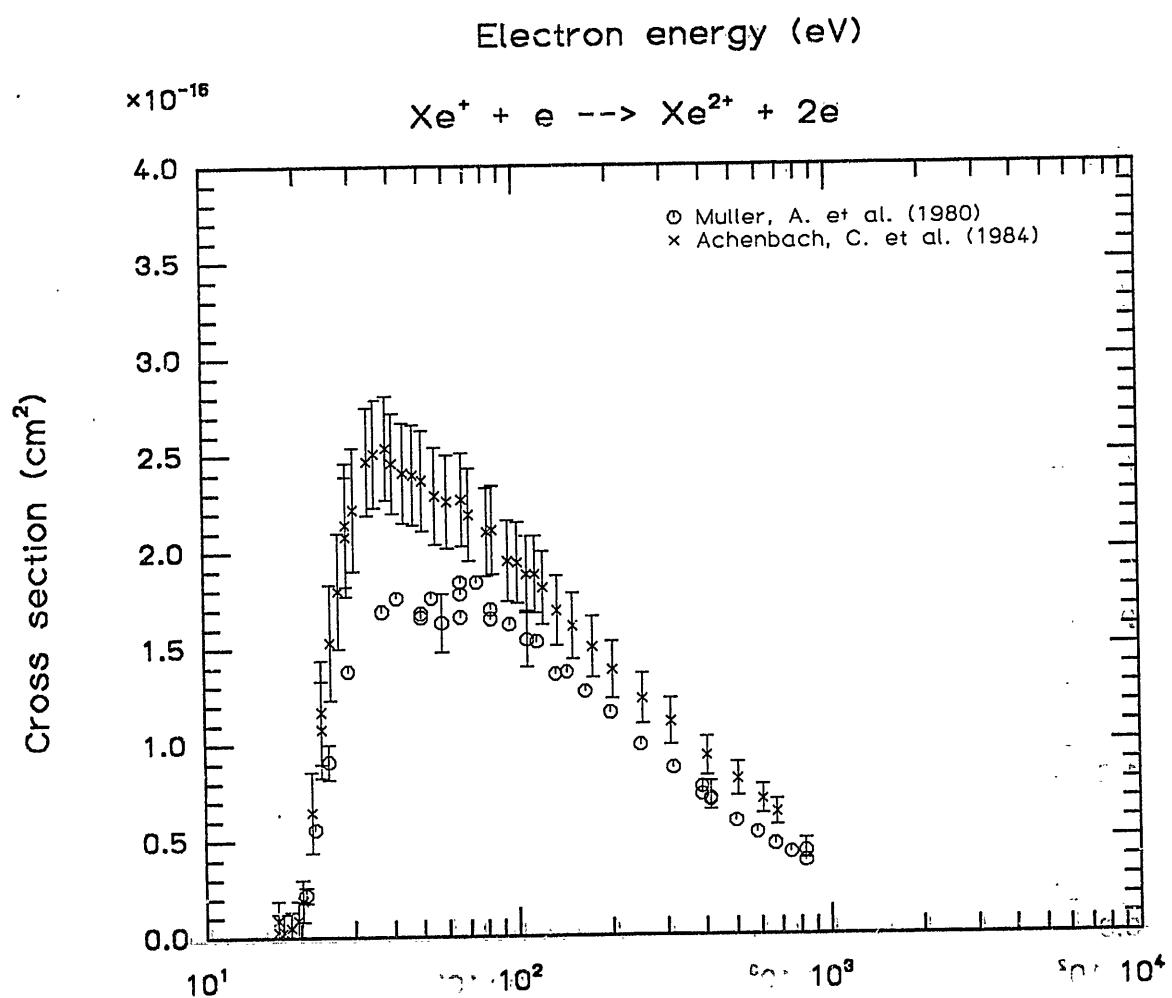
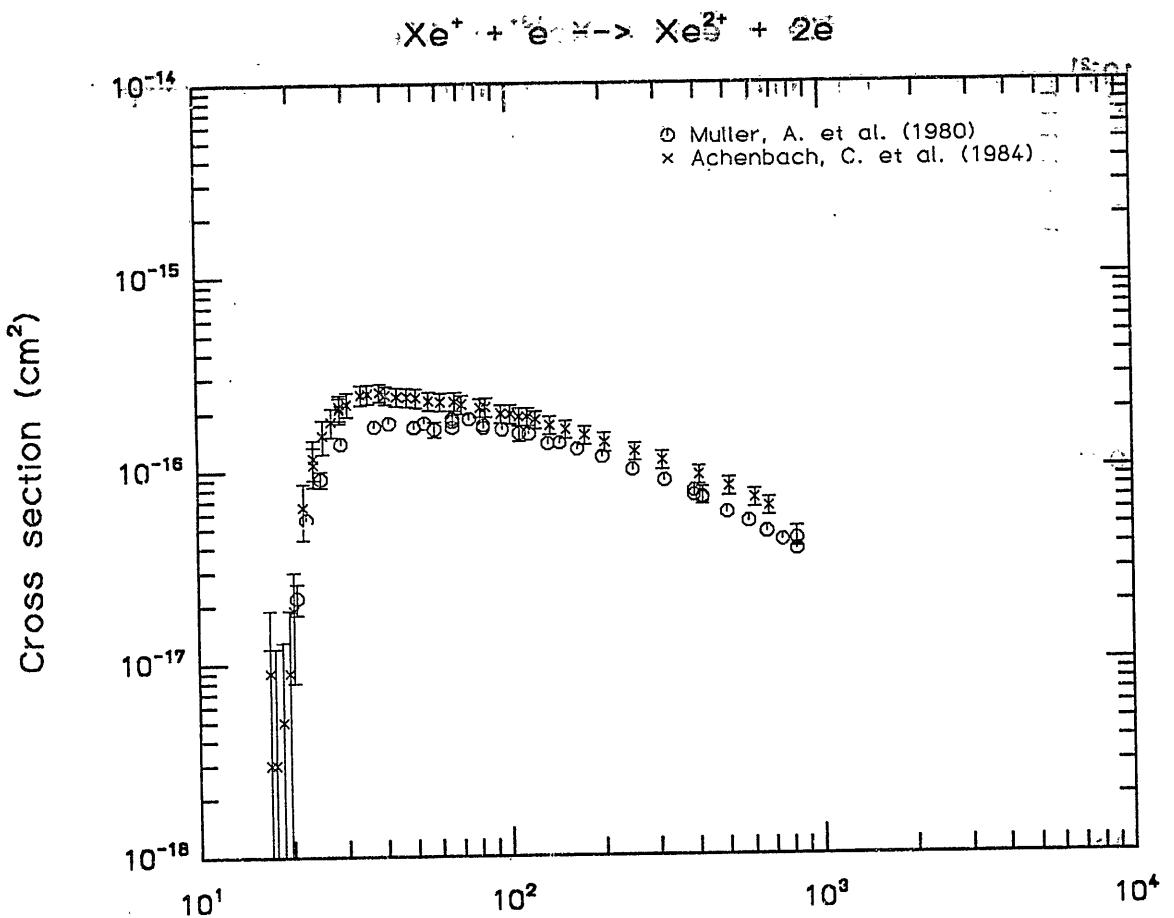
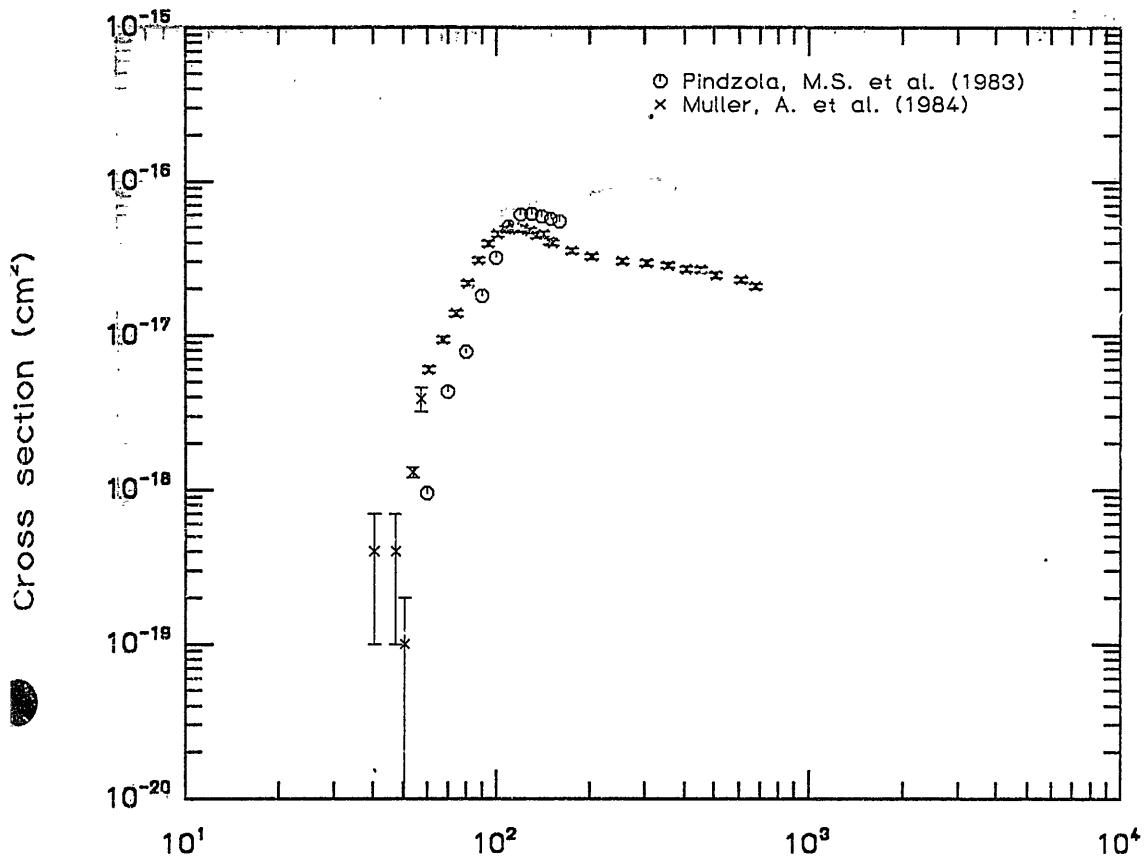
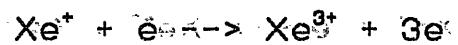


Fig. 240 (top) Electron energy (eV) 832 fig



Electron energy (eV)

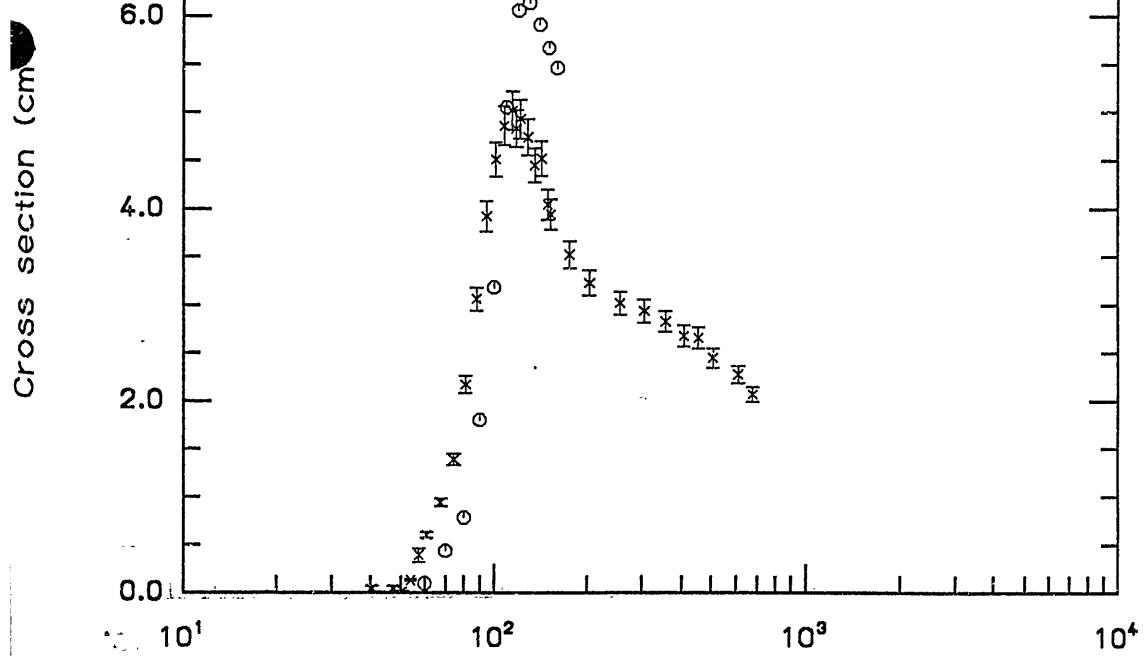
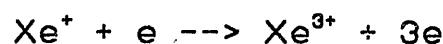


Fig. 241      Electron energy (eV)

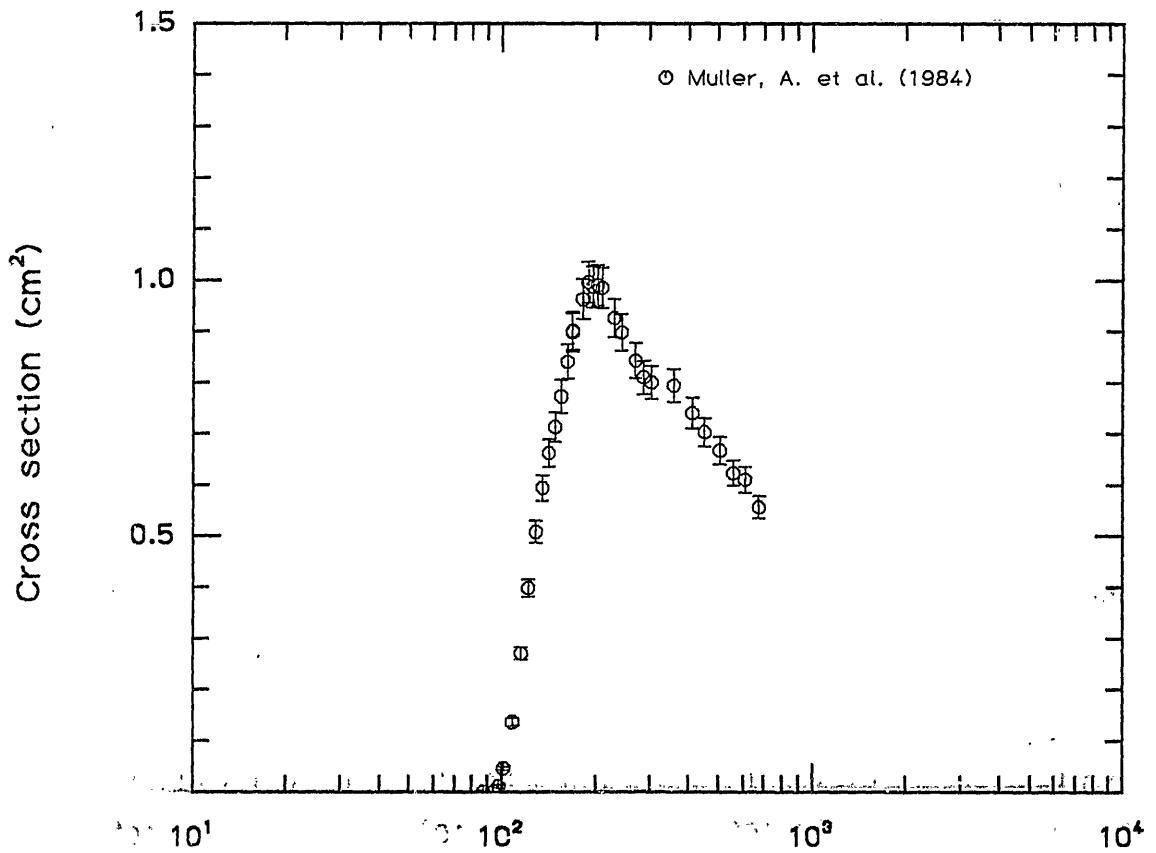
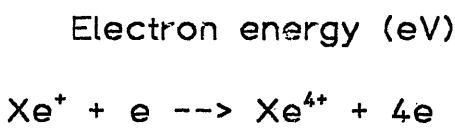
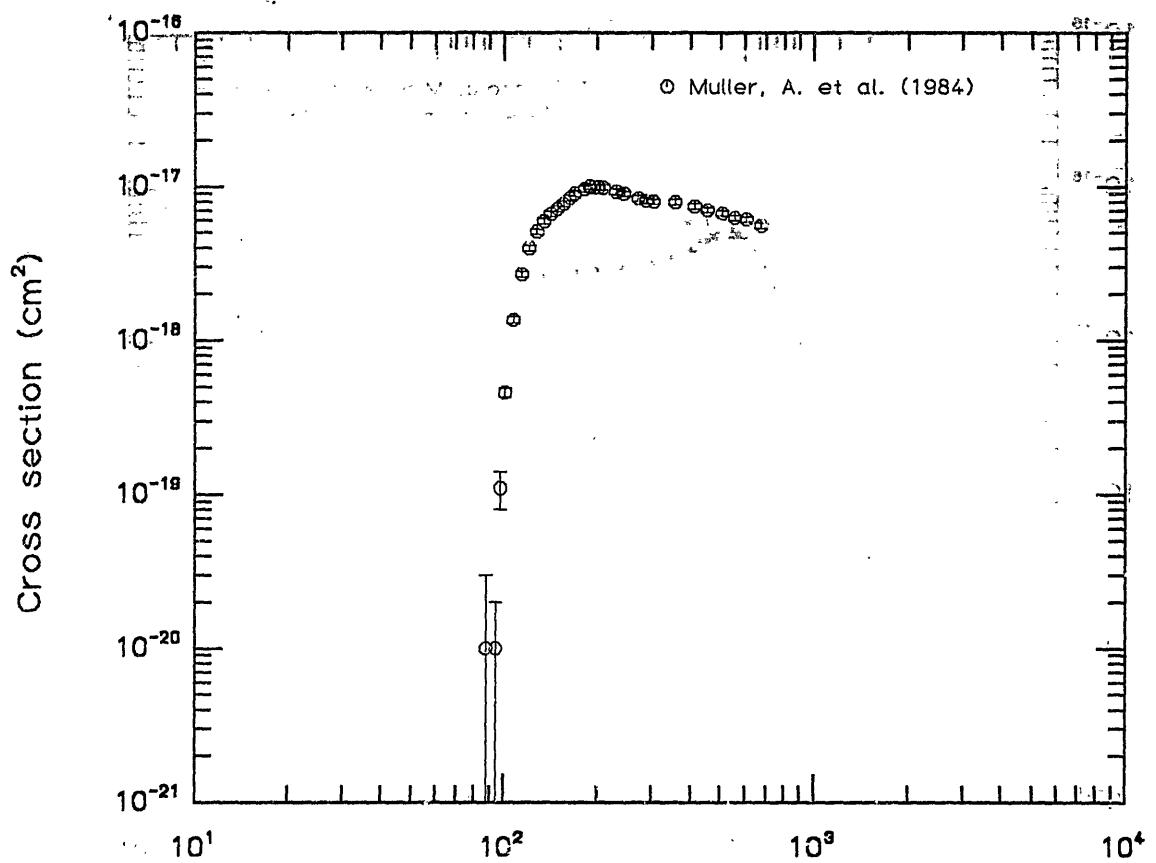
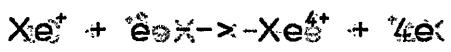


Fig. 242 (vs Electron energy (eV)) 142 pi

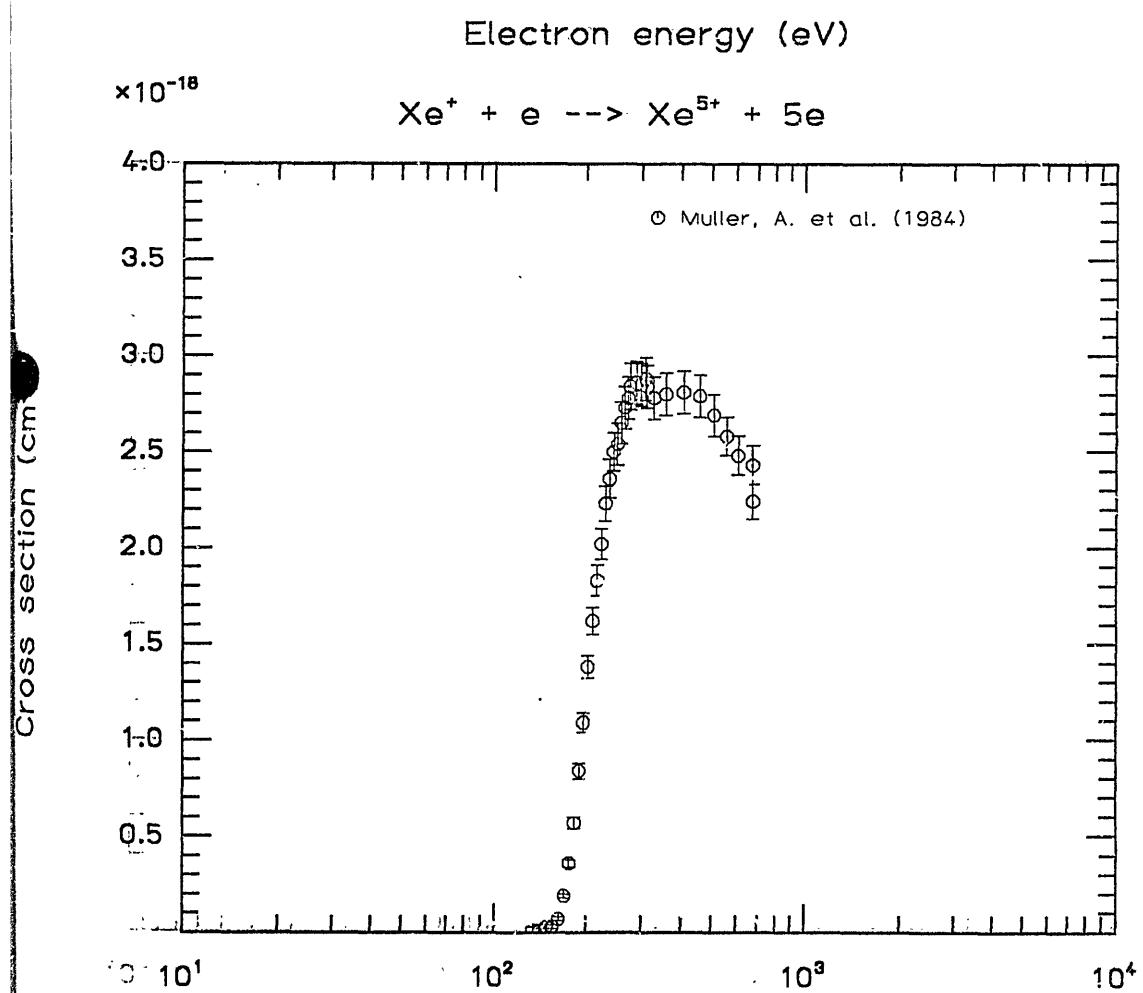
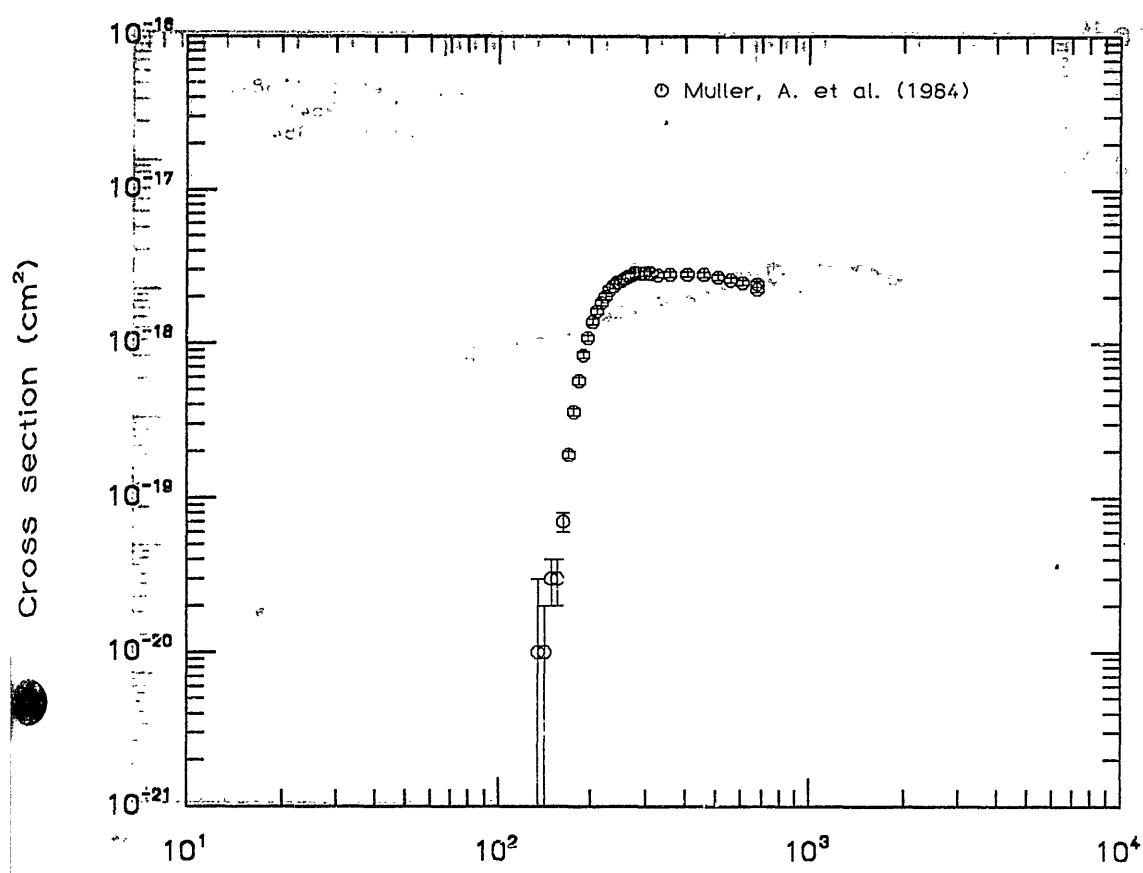
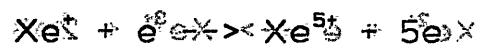


Fig. 243

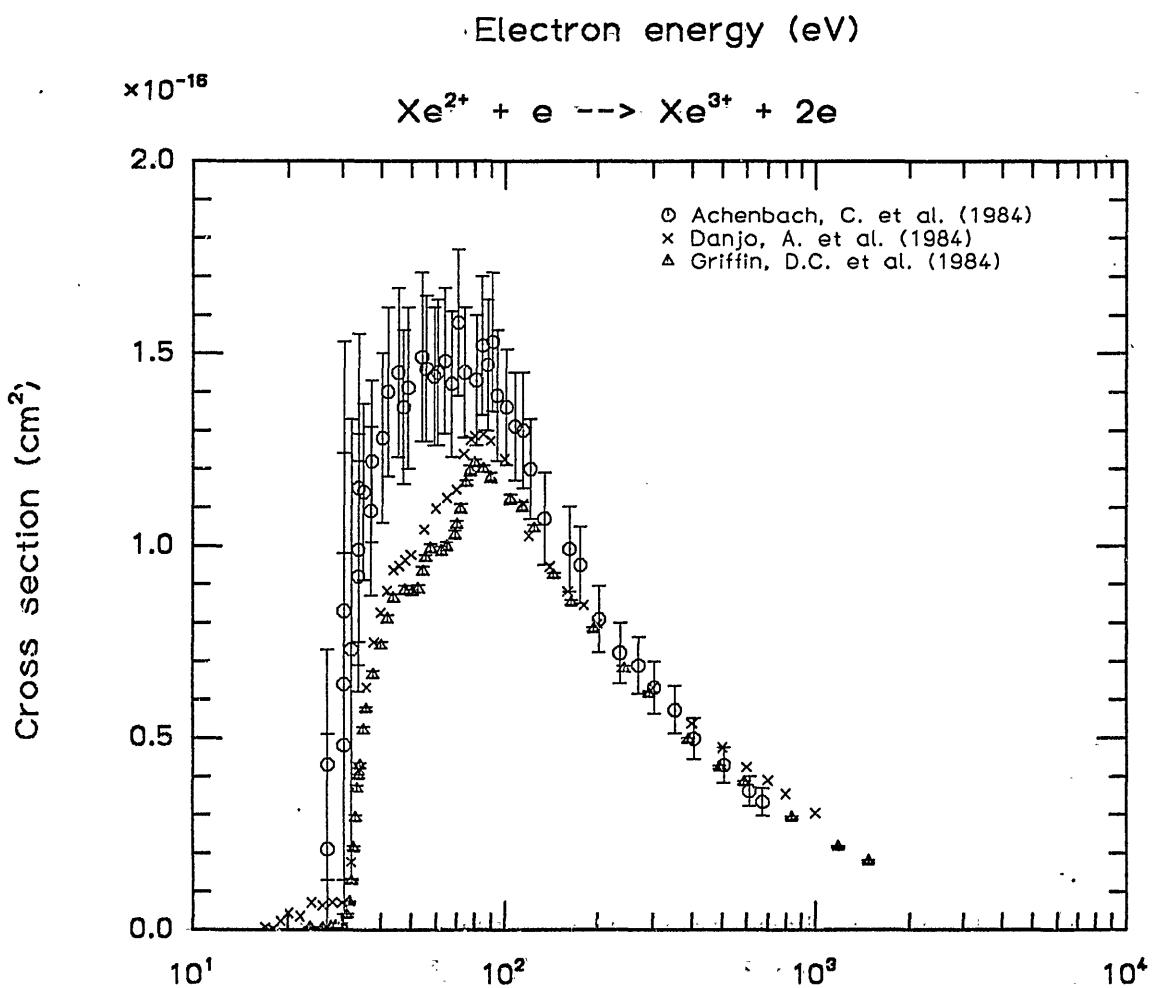
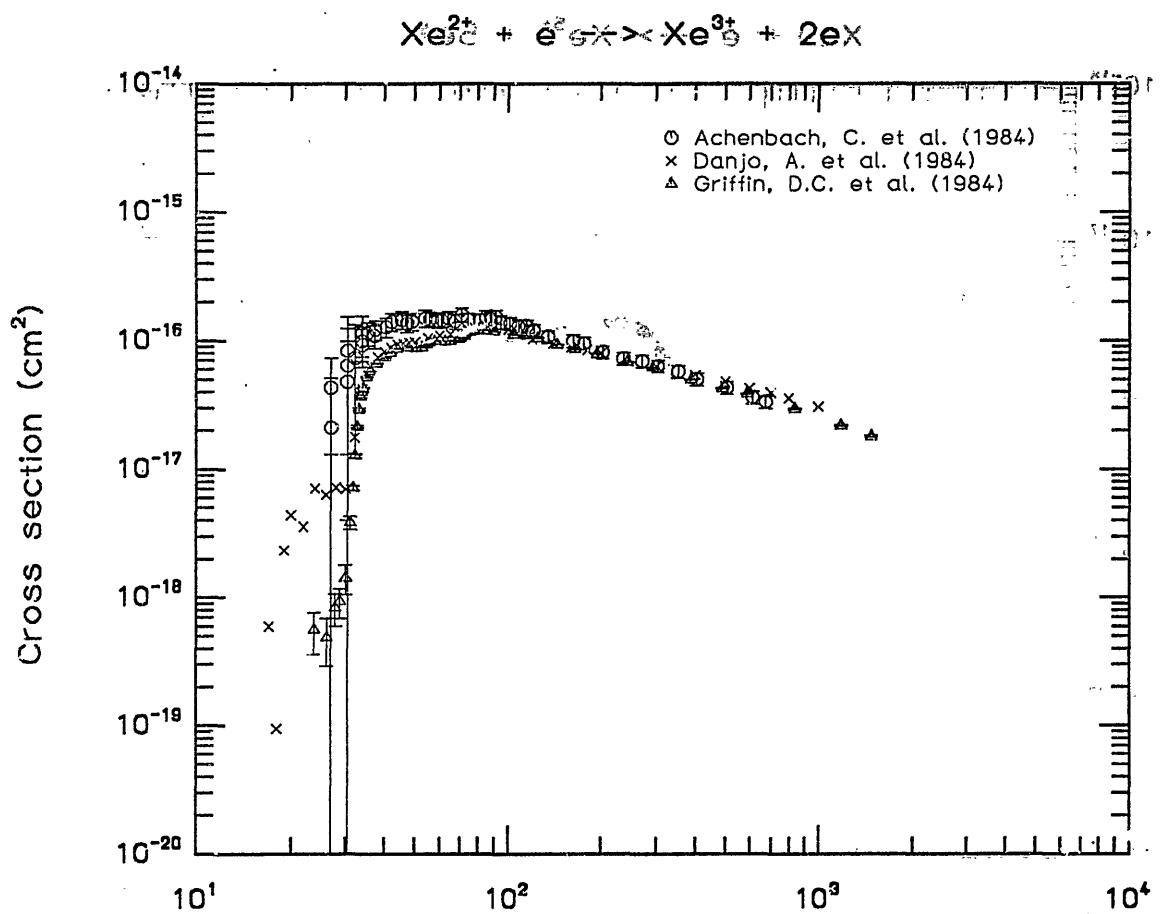
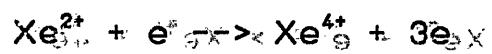
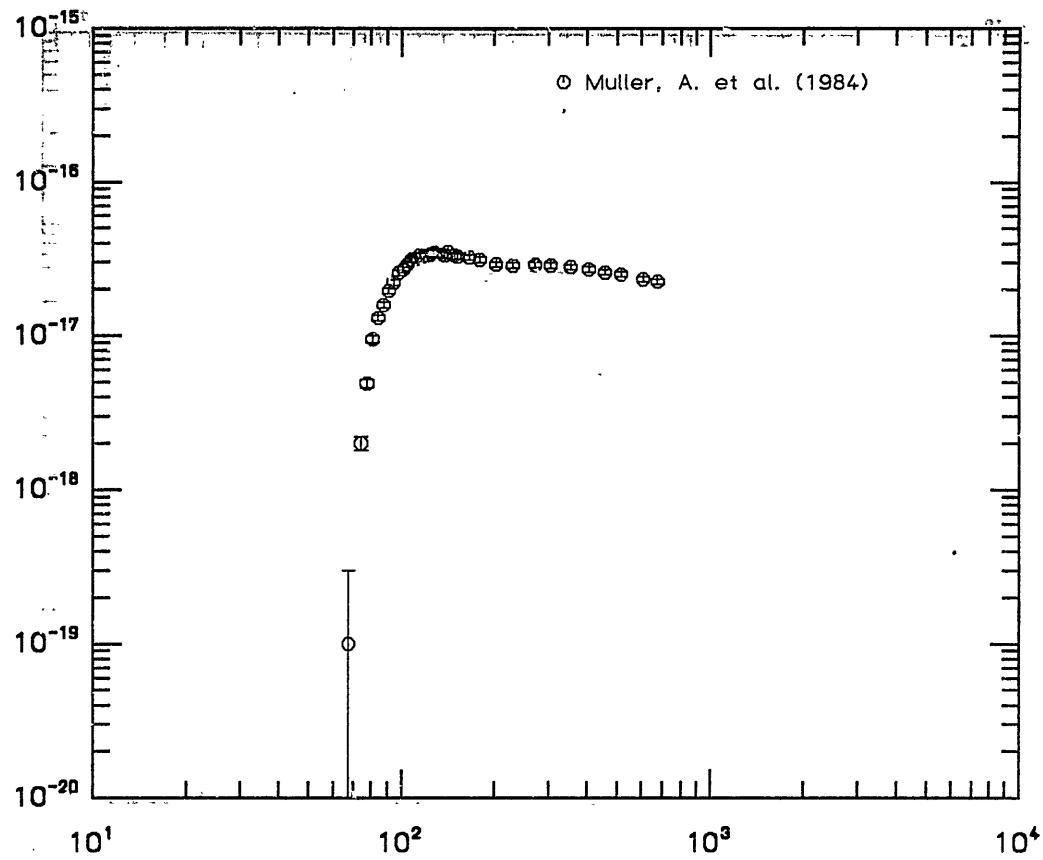


Fig. 244. Cross section vs Electron energy (eV)  $\times 10^{-16}$  cm<sup>2</sup> (Fig. 243)



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

Cross section ( $\text{cm}^2$ )

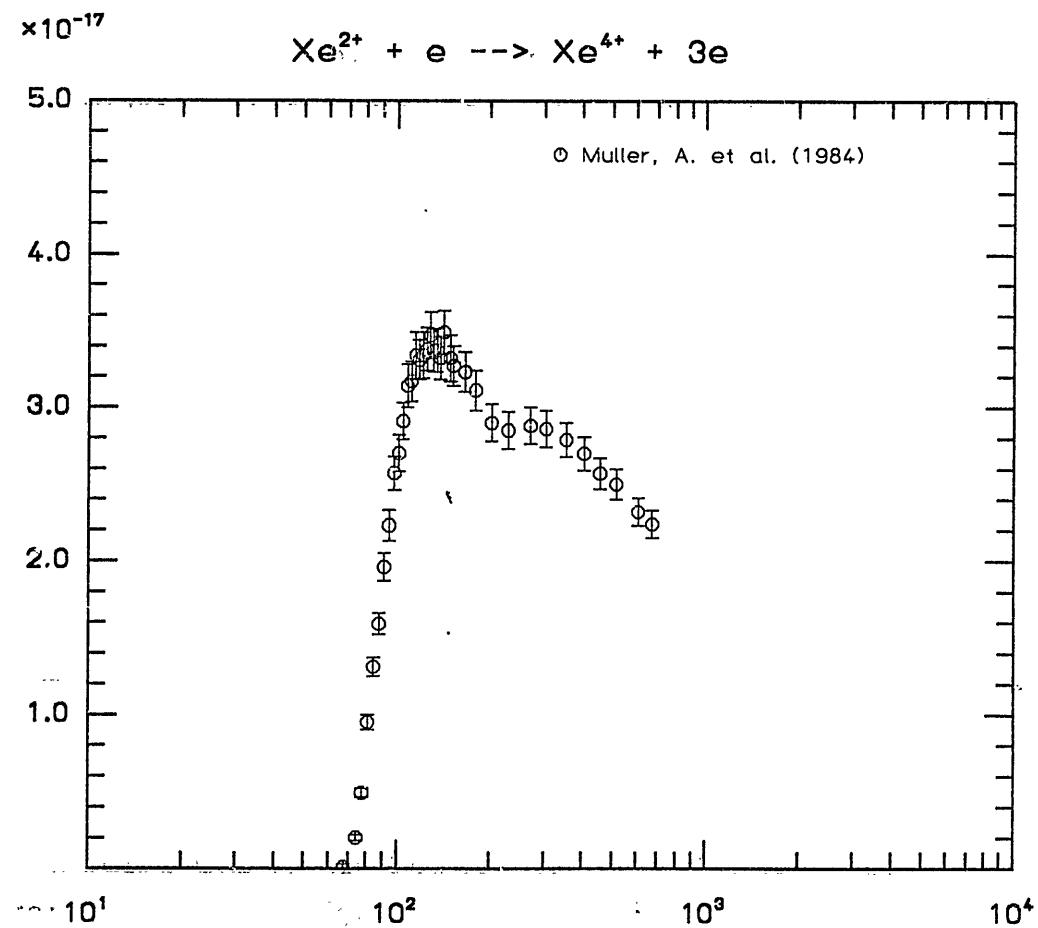


Fig. 245 (a) Electron energy (eV) (b)  $\text{A.S.}$

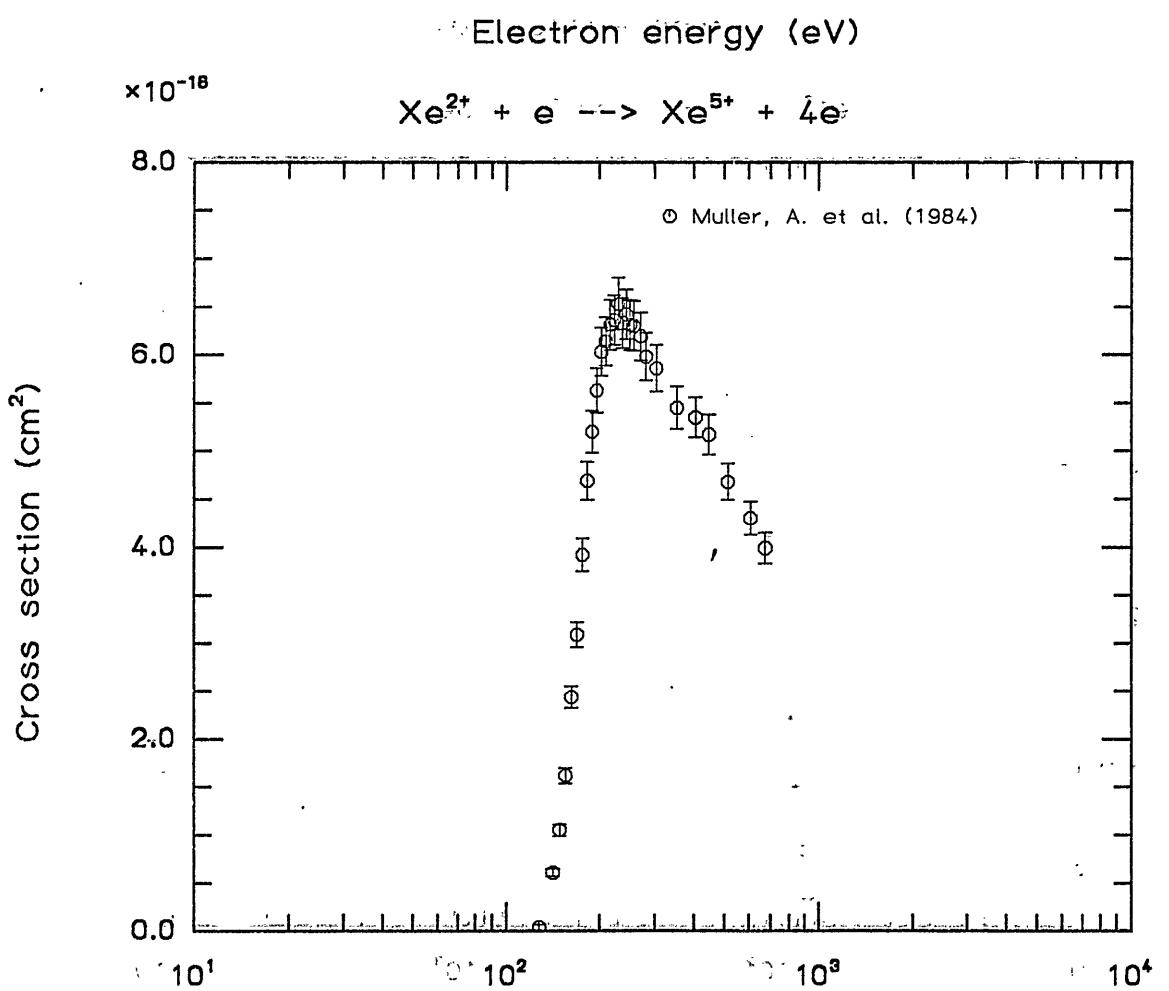
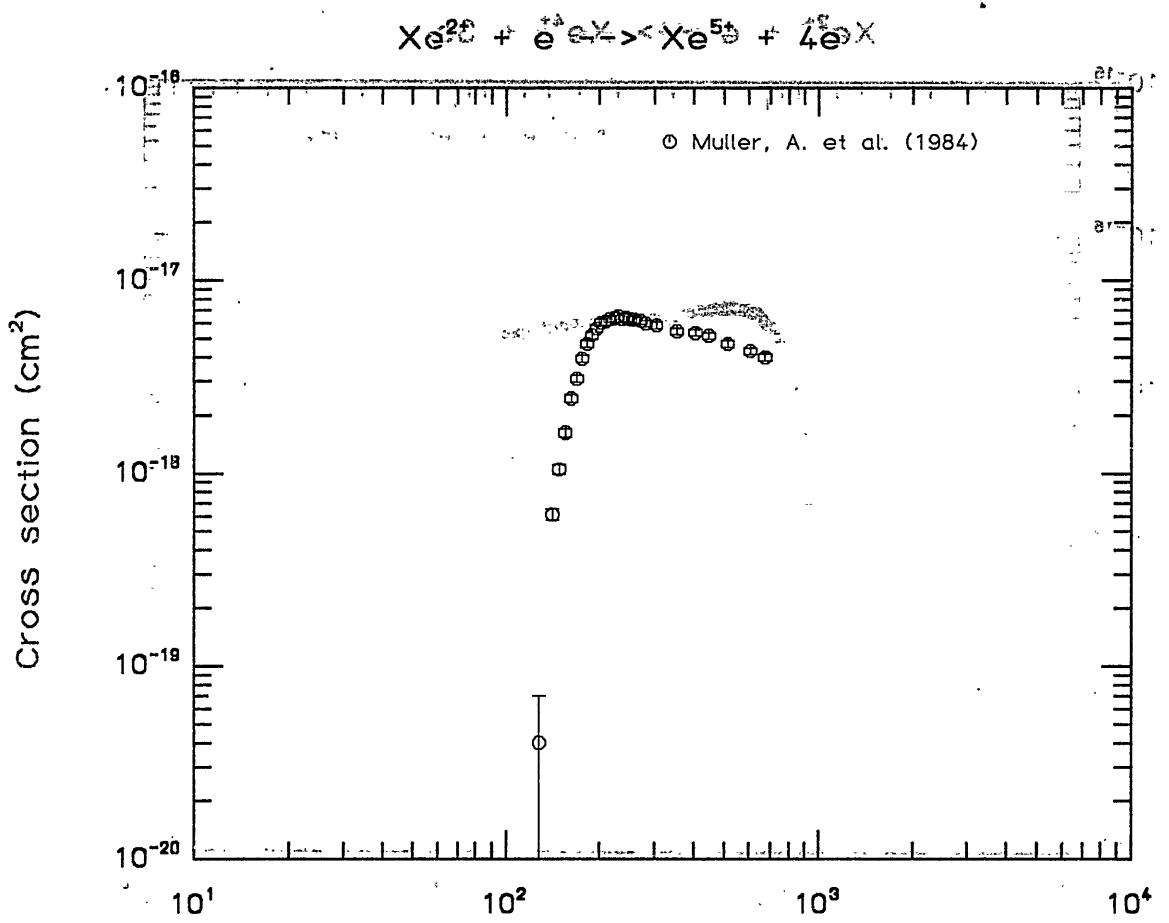
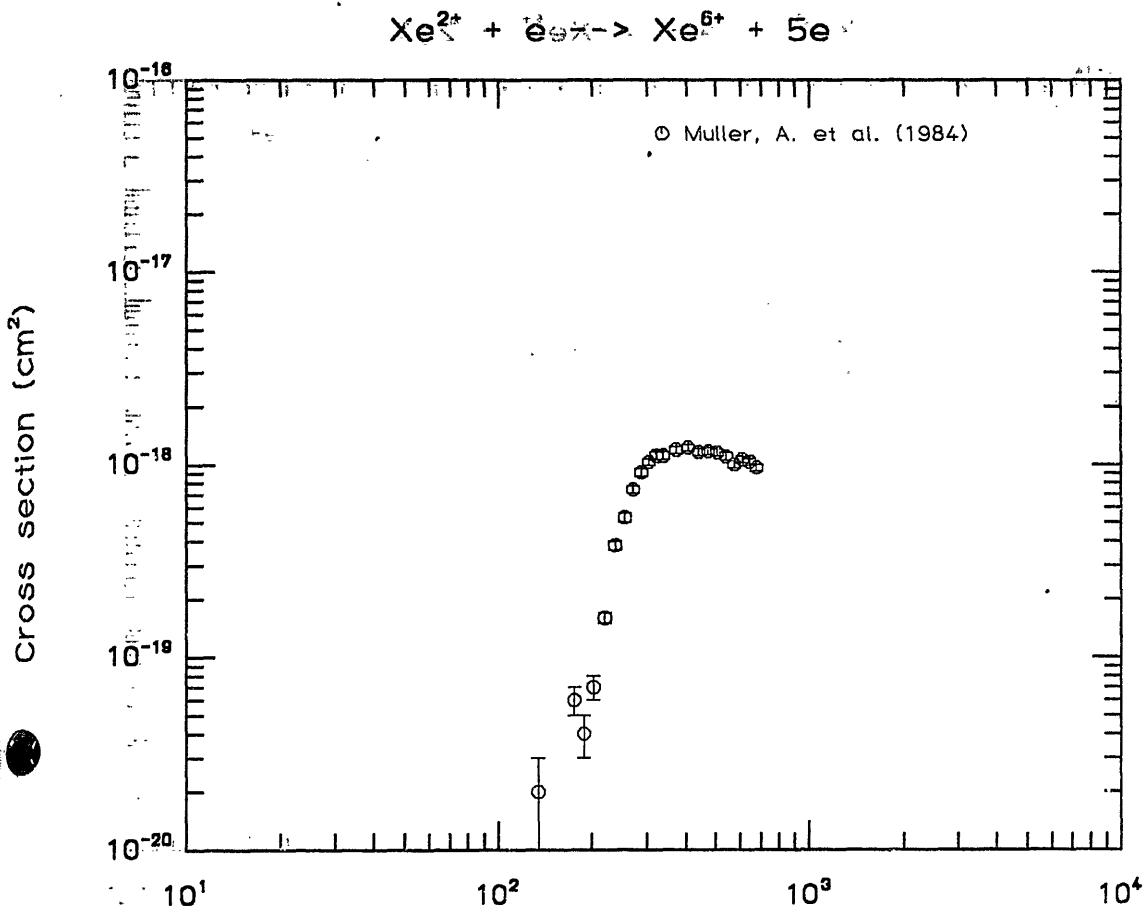
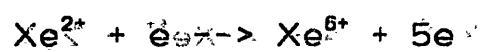


Fig. 246 (V) Electron energy (eV) 246.pif



Electron energy (eV)

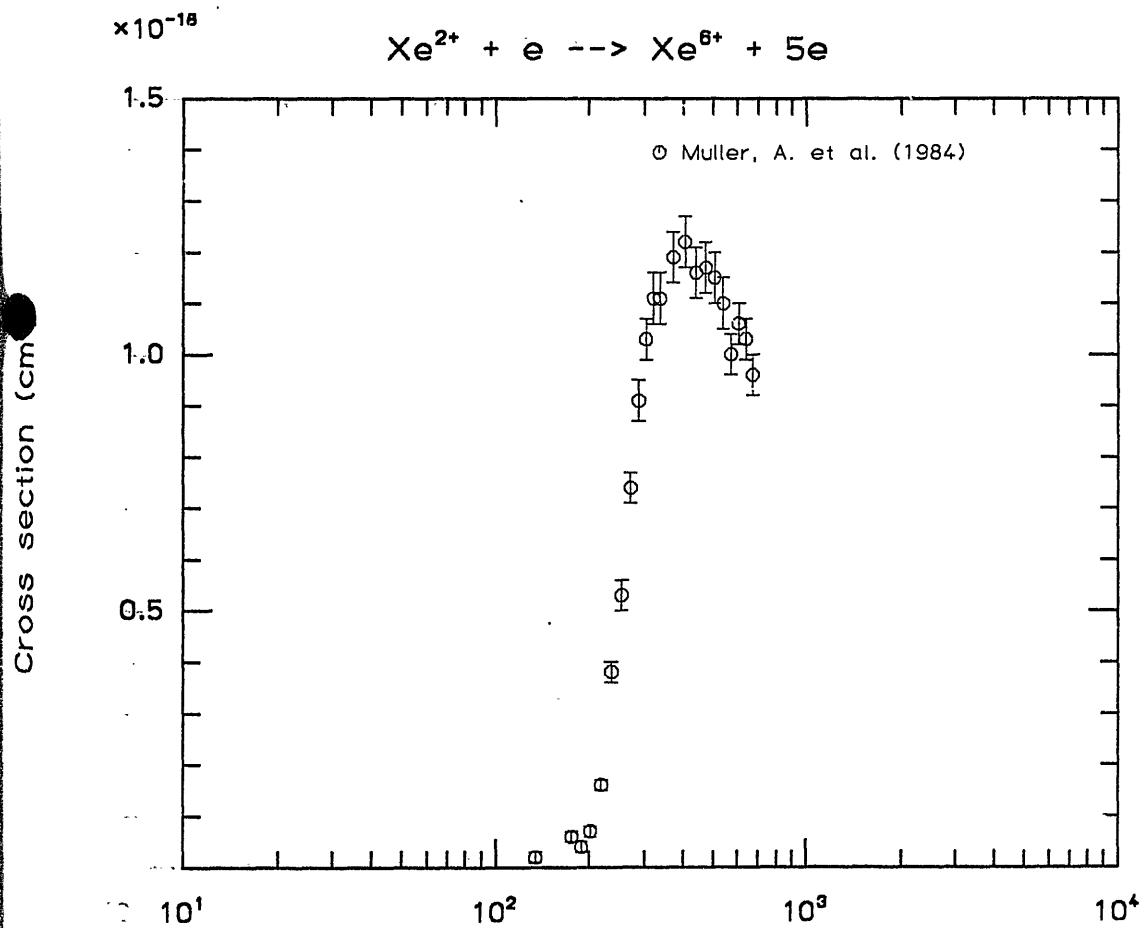


Fig. 247 Electron energy (eV)

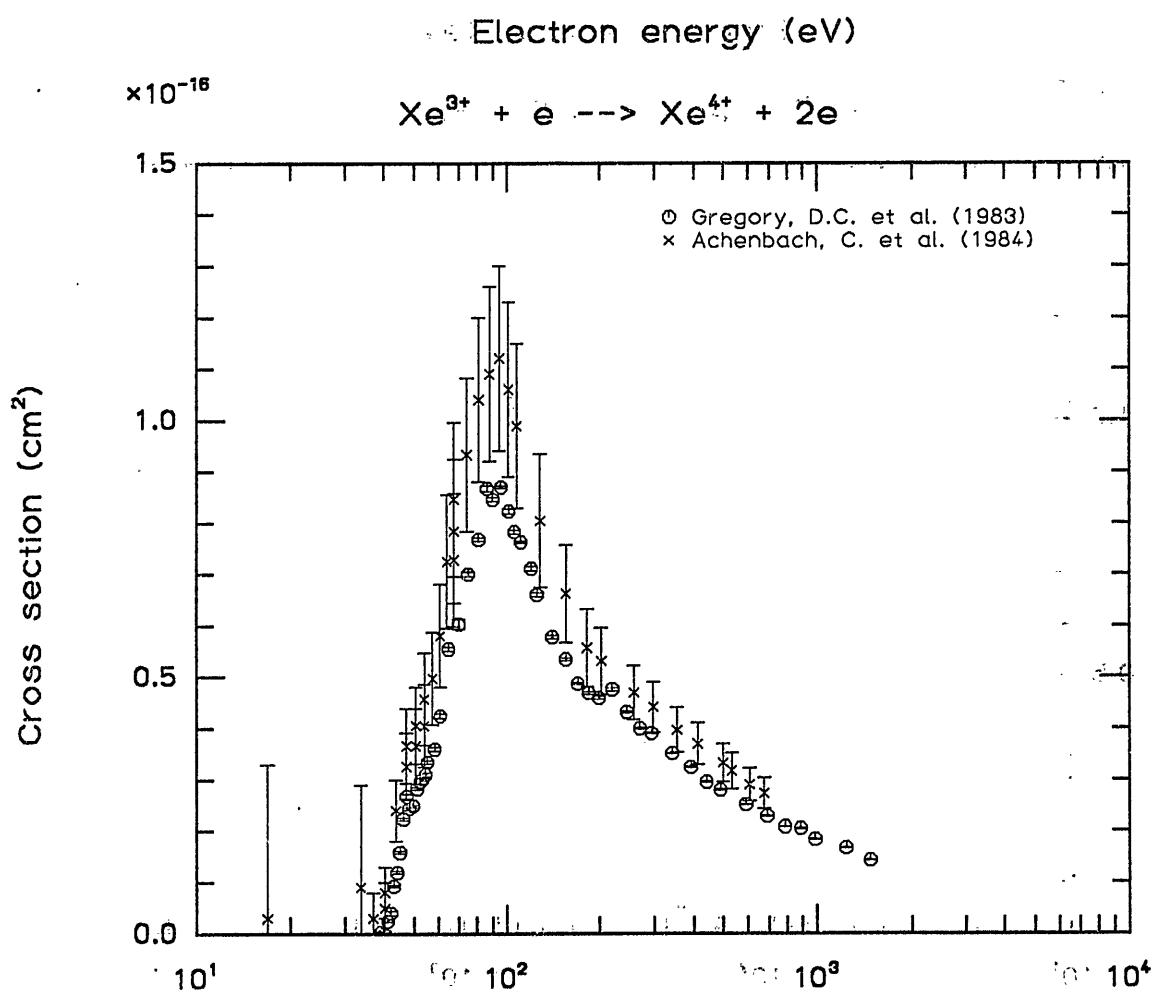
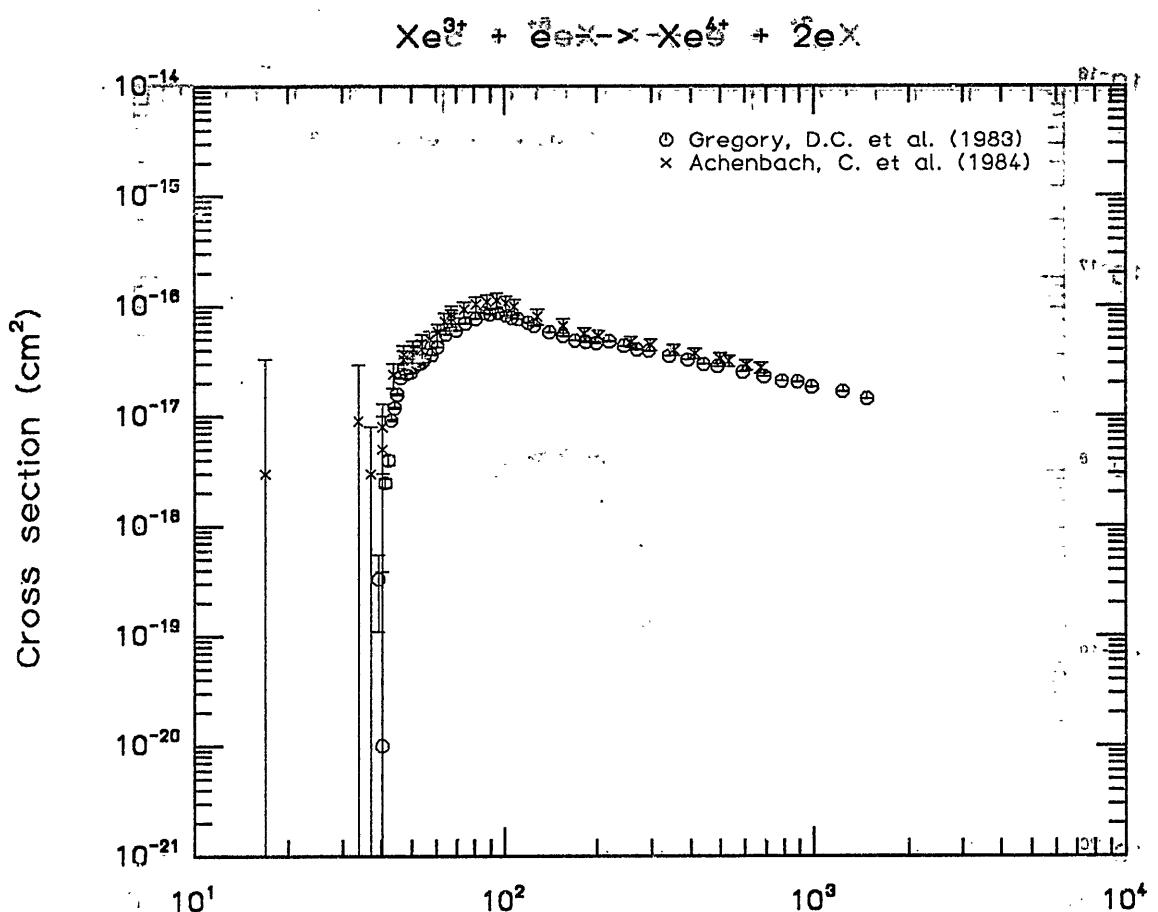
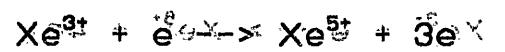
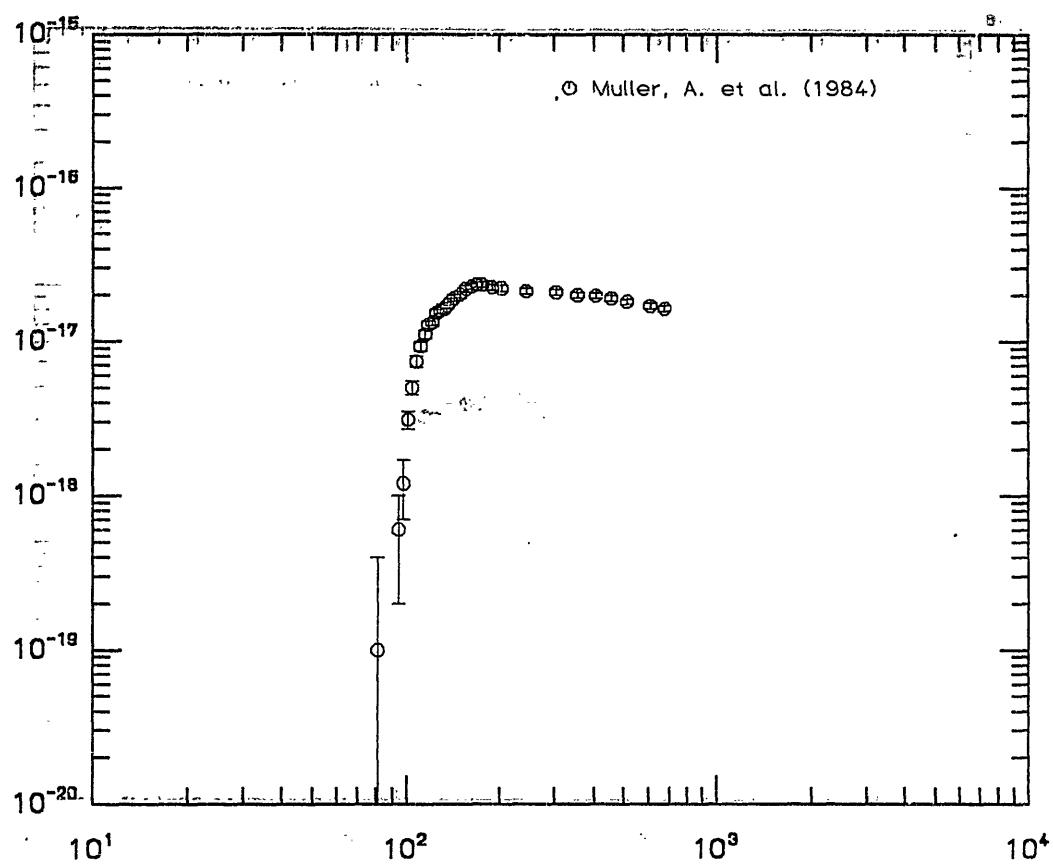


Fig. 248 (Ne) Electron energy (eV) Fig. 248 (Ne)



Cross section ( $\text{cm}^2$ )



Cross section ( $\text{cm}^2$ )

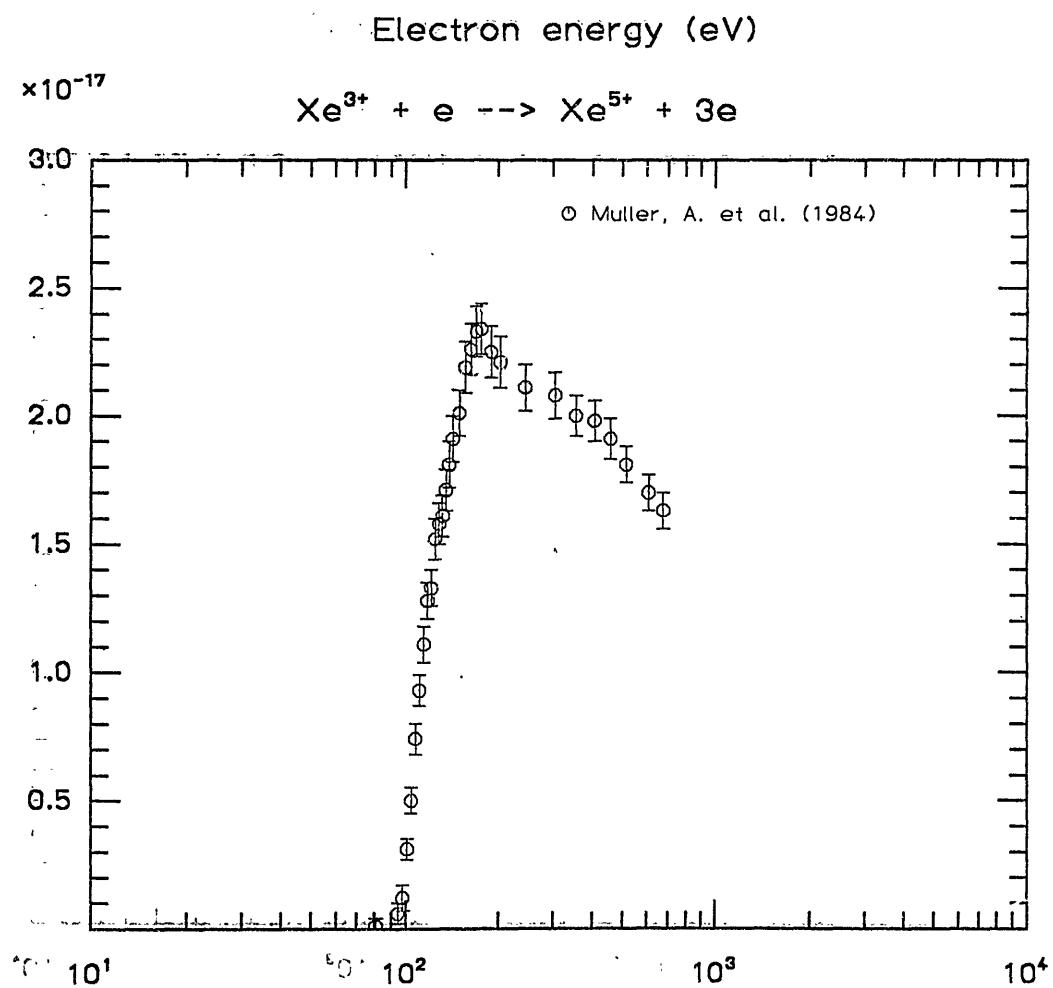


Fig. 249 Cross section vs. Electron energy (eV)

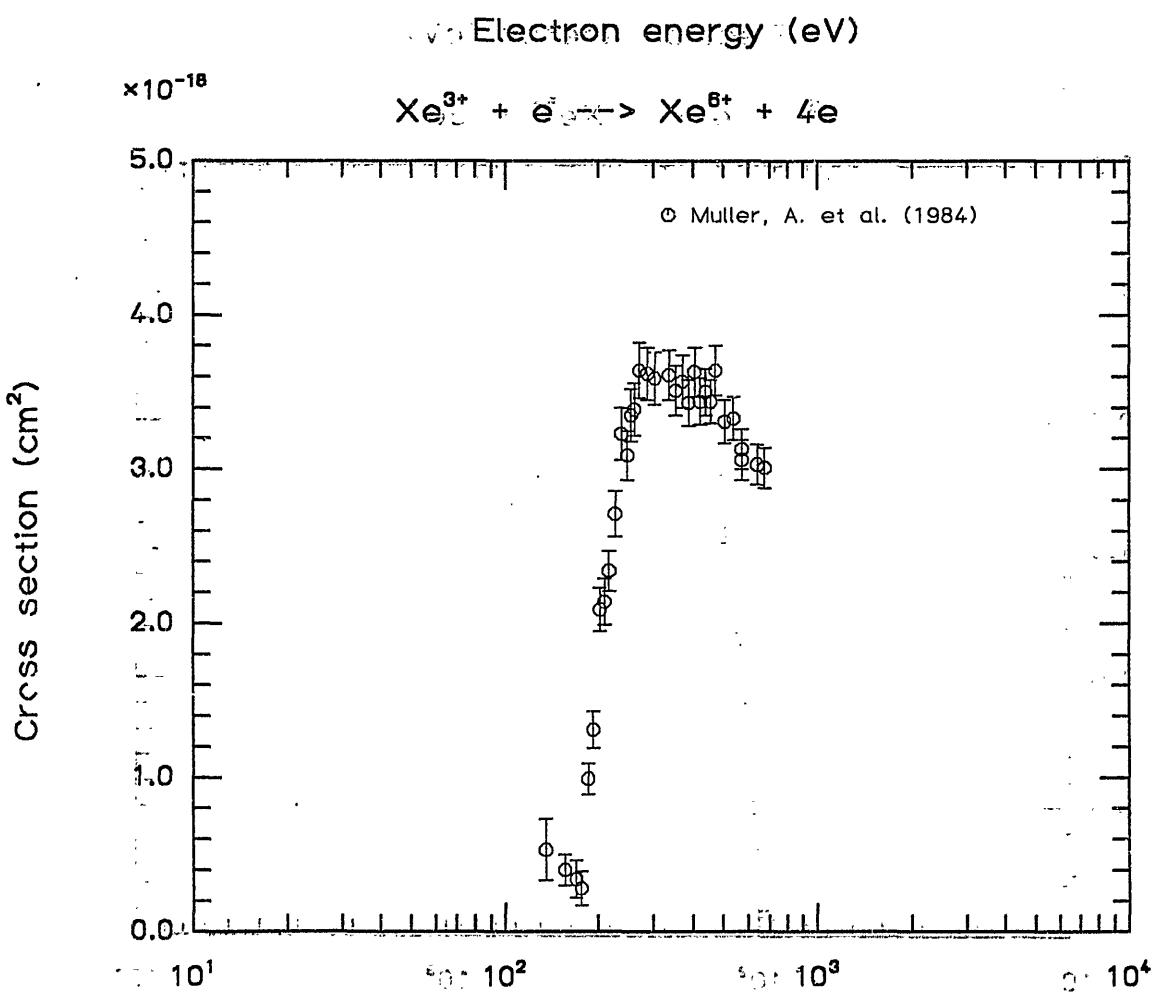
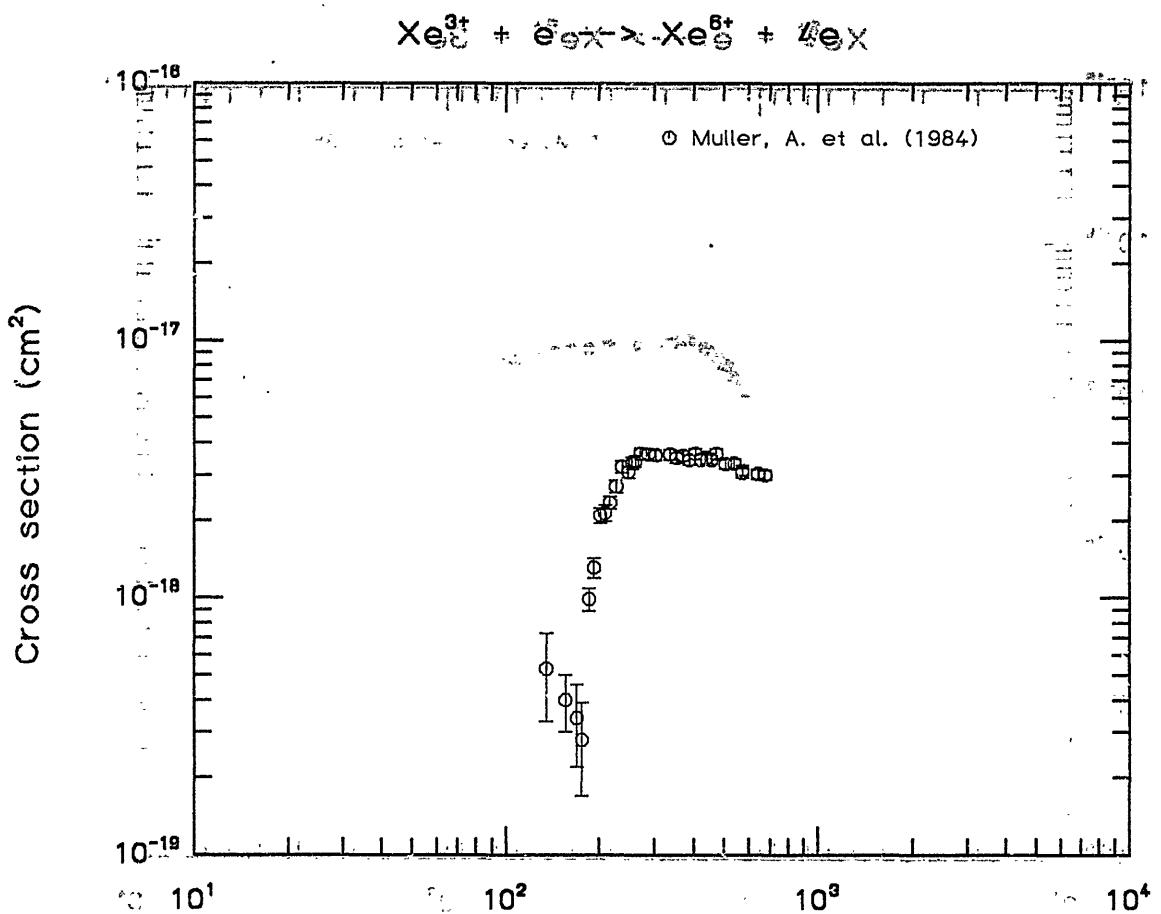


Fig. 250 (ve) Electron energy (eV) 042.pif

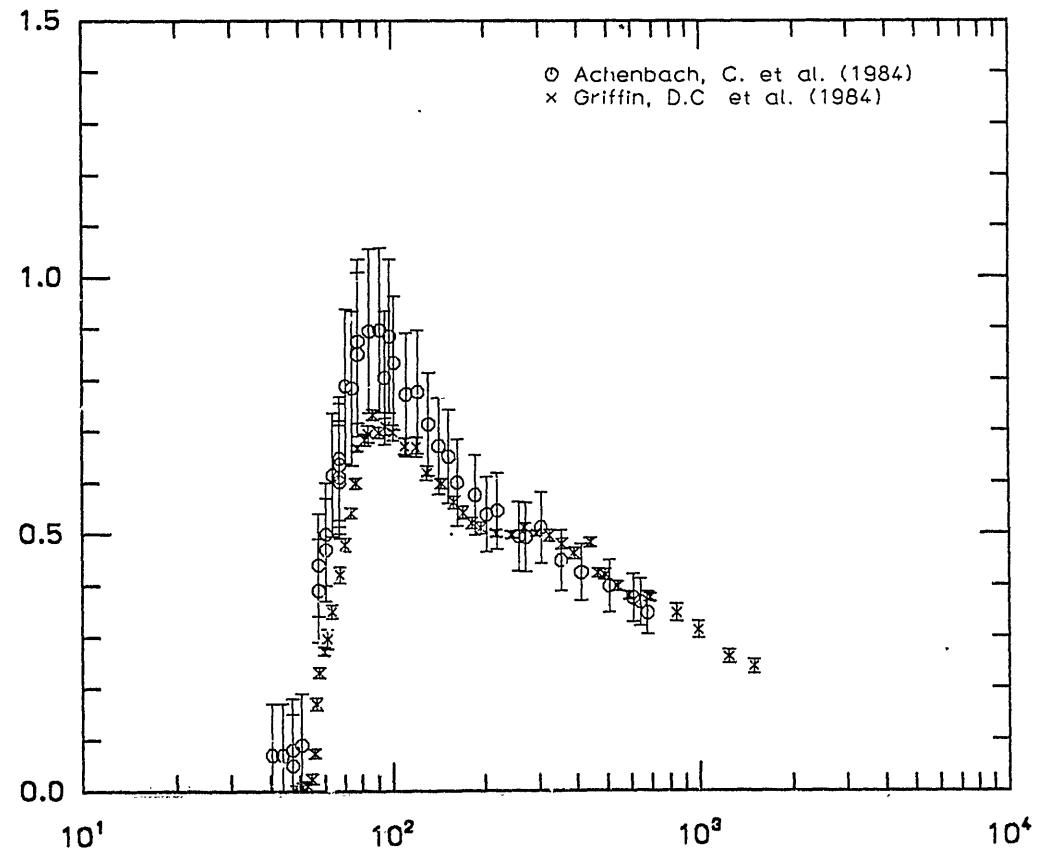
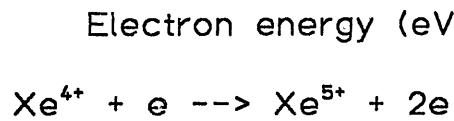
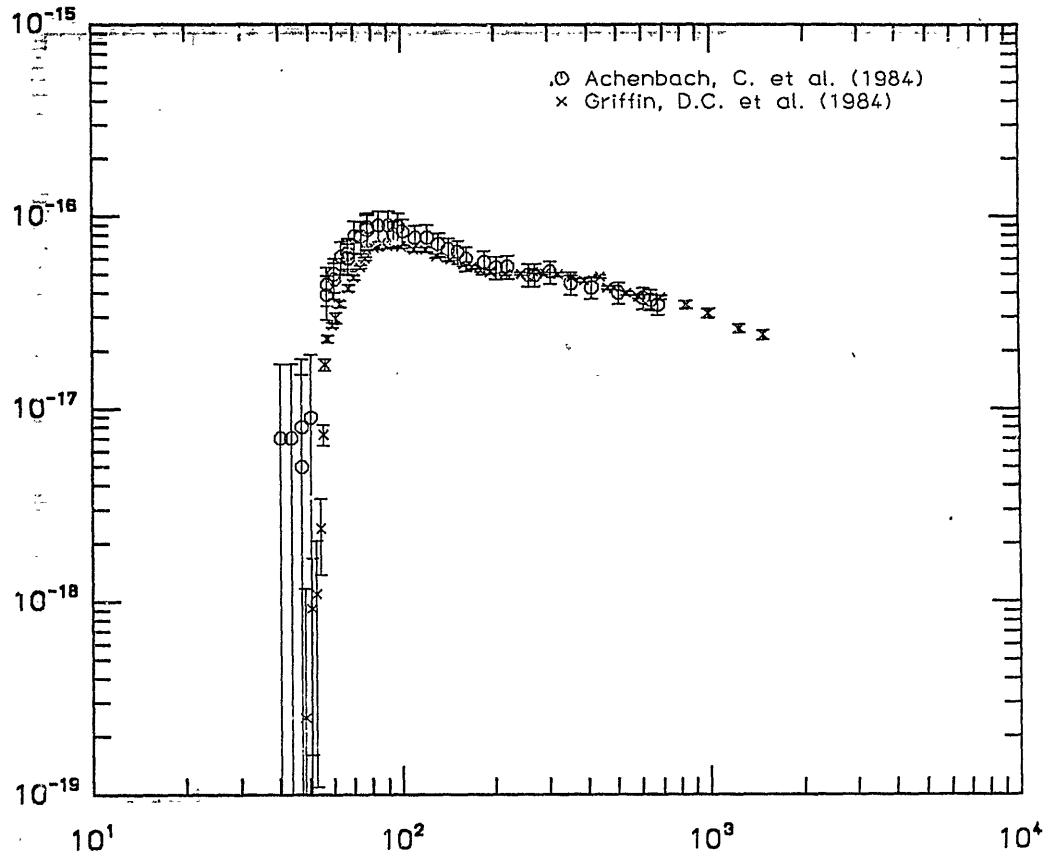
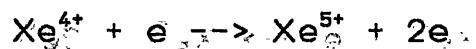


Fig. 251

Electron energy (eV)

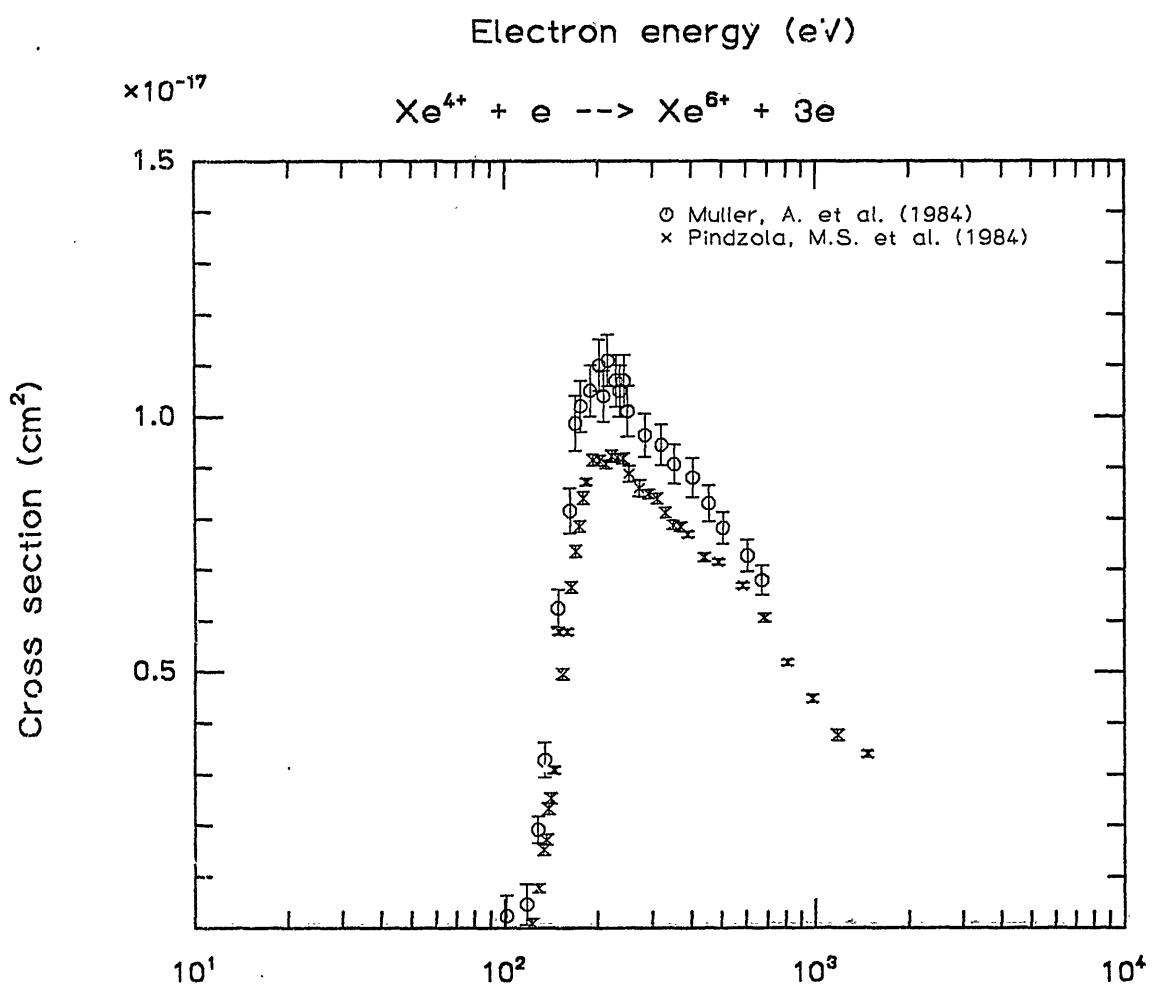
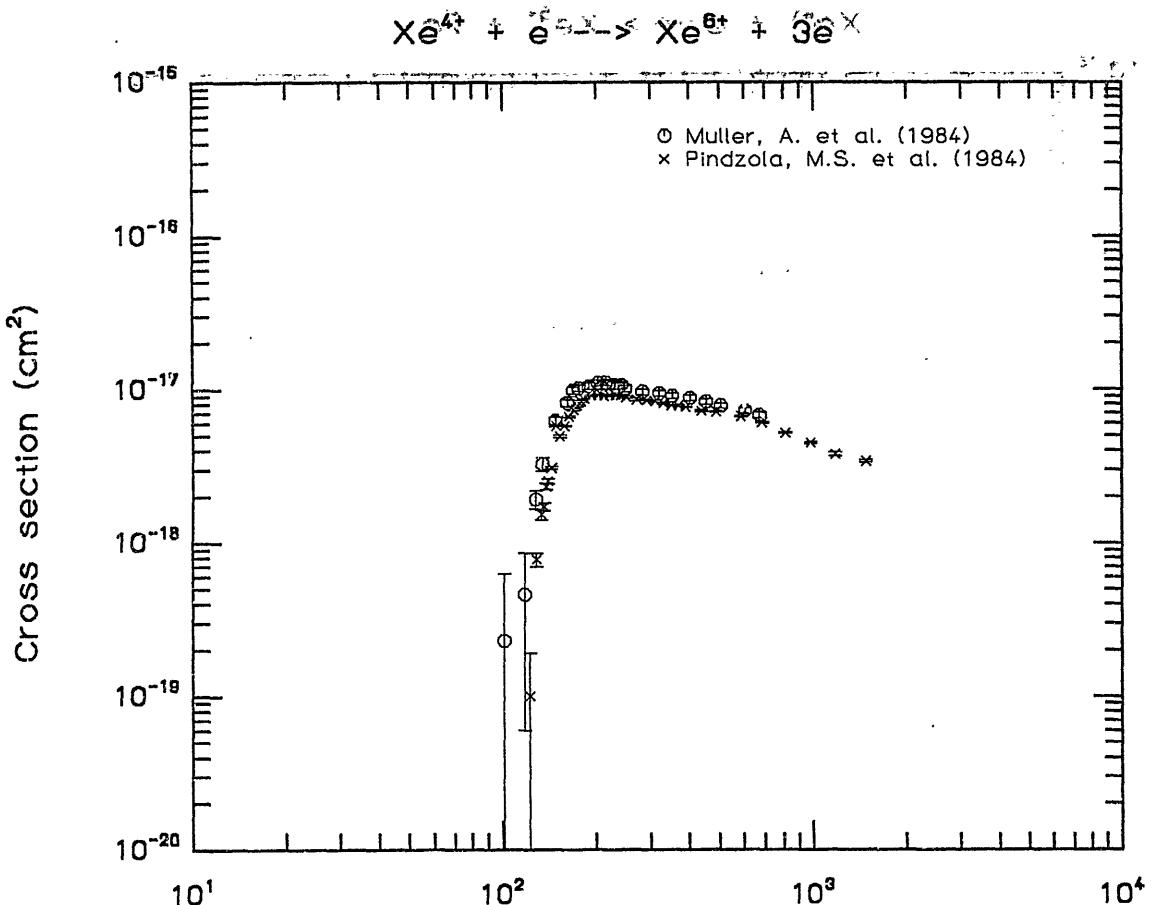
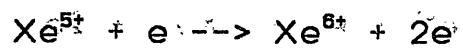
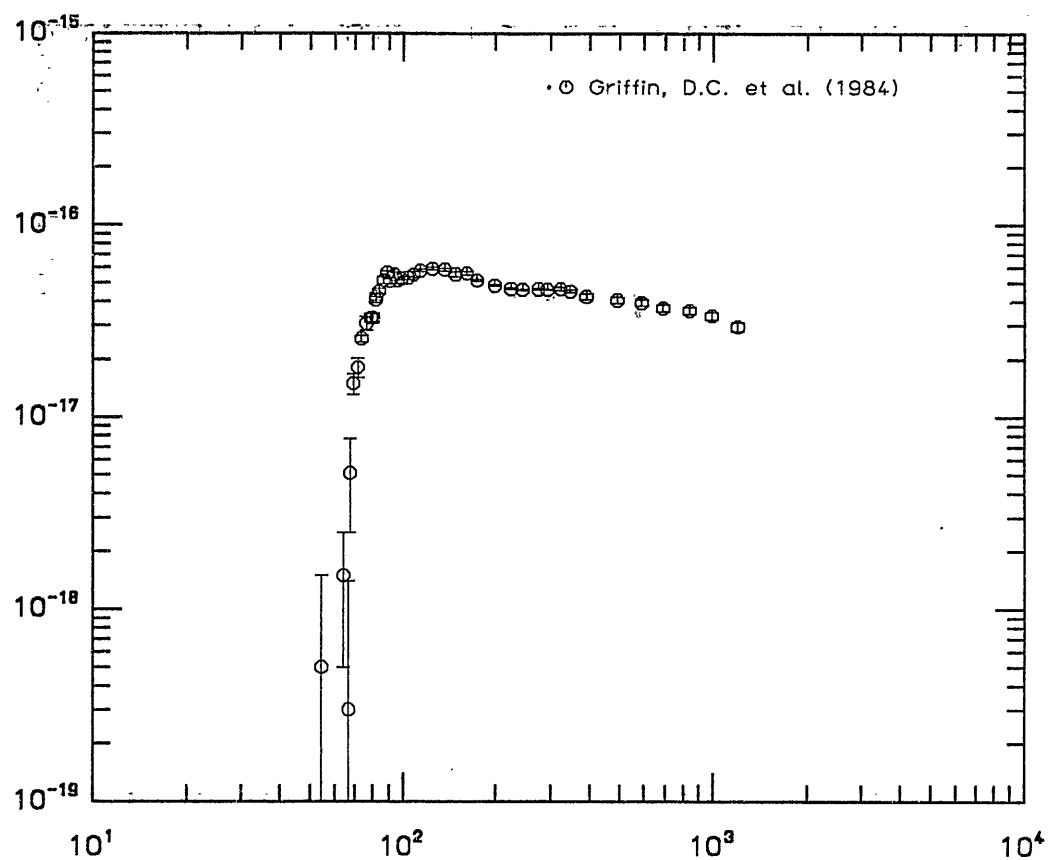


Fig. 252      Electron energy (eV)



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

Cross section ( $\text{cm}^2$ )

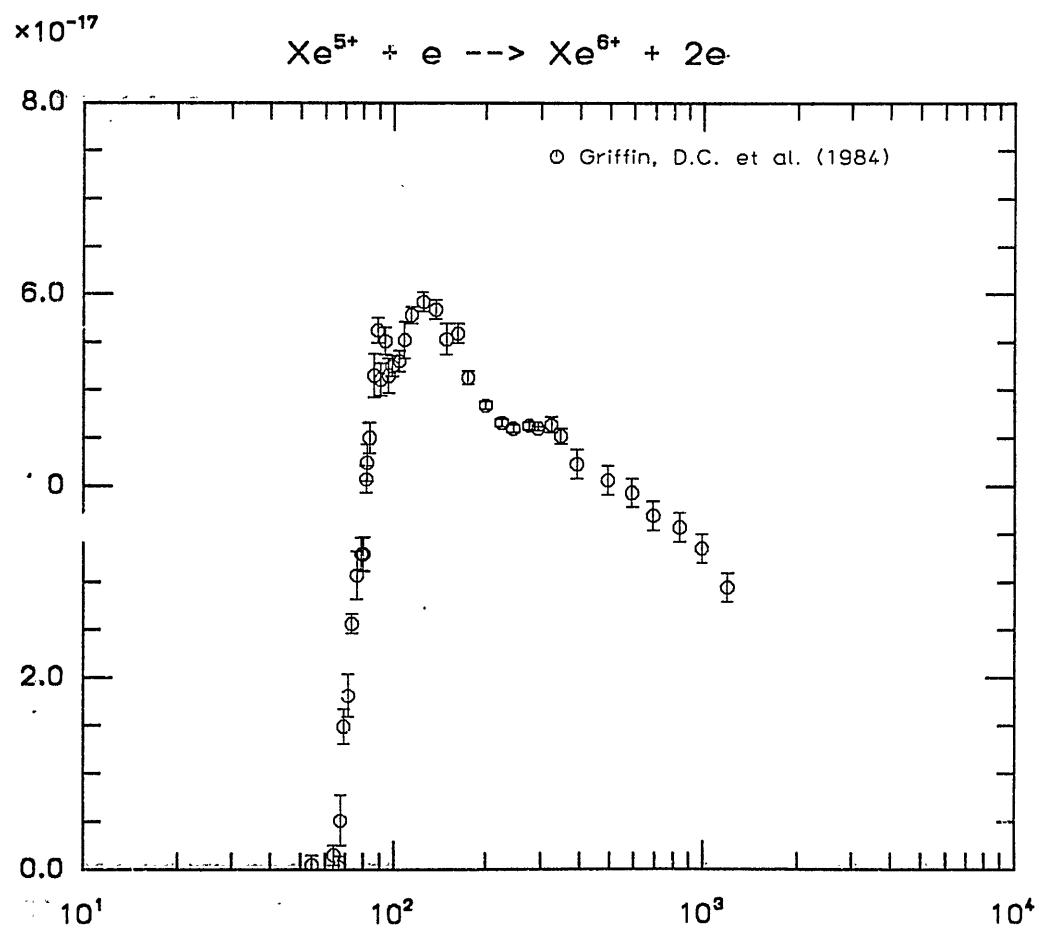


Fig. 253

Electron energy (eV)

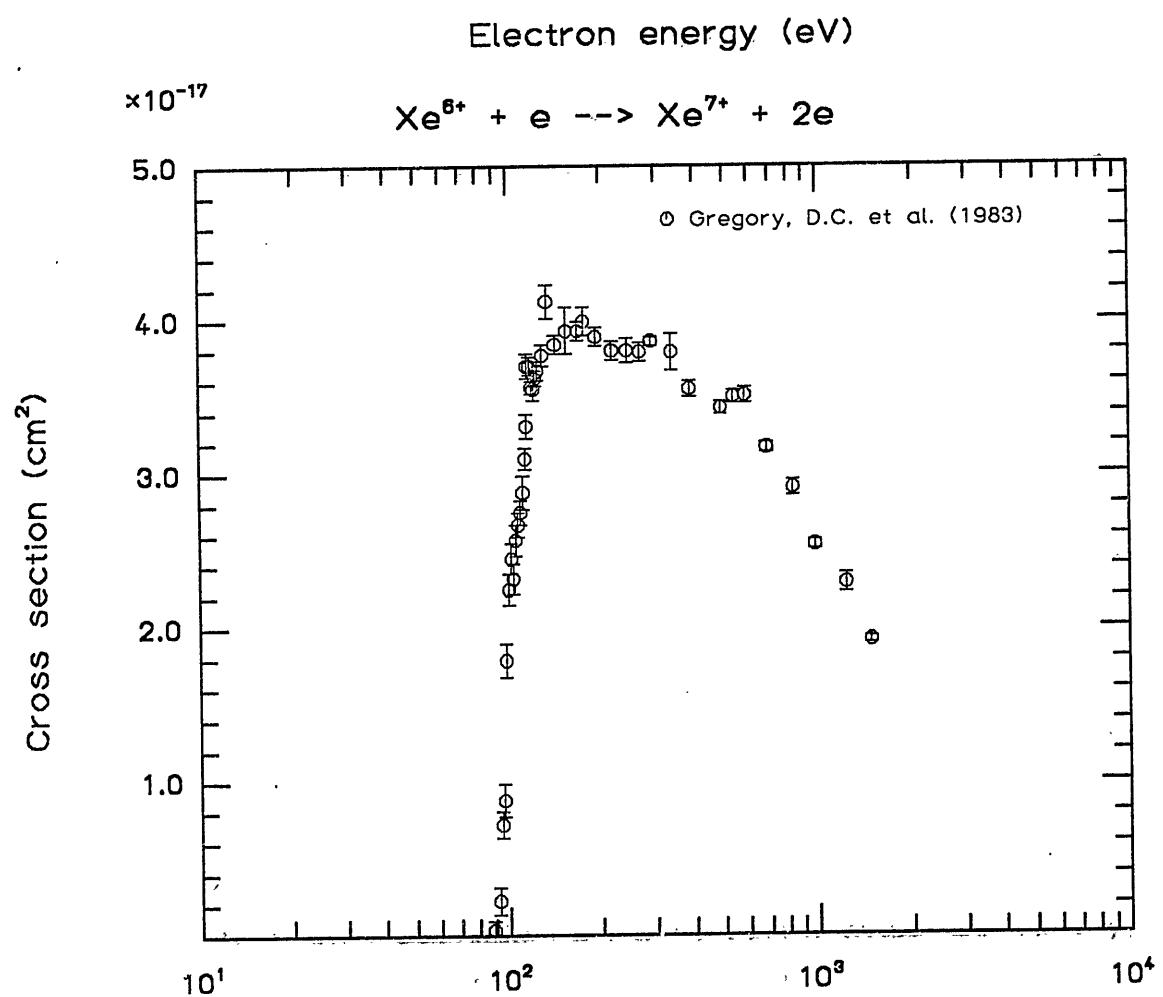
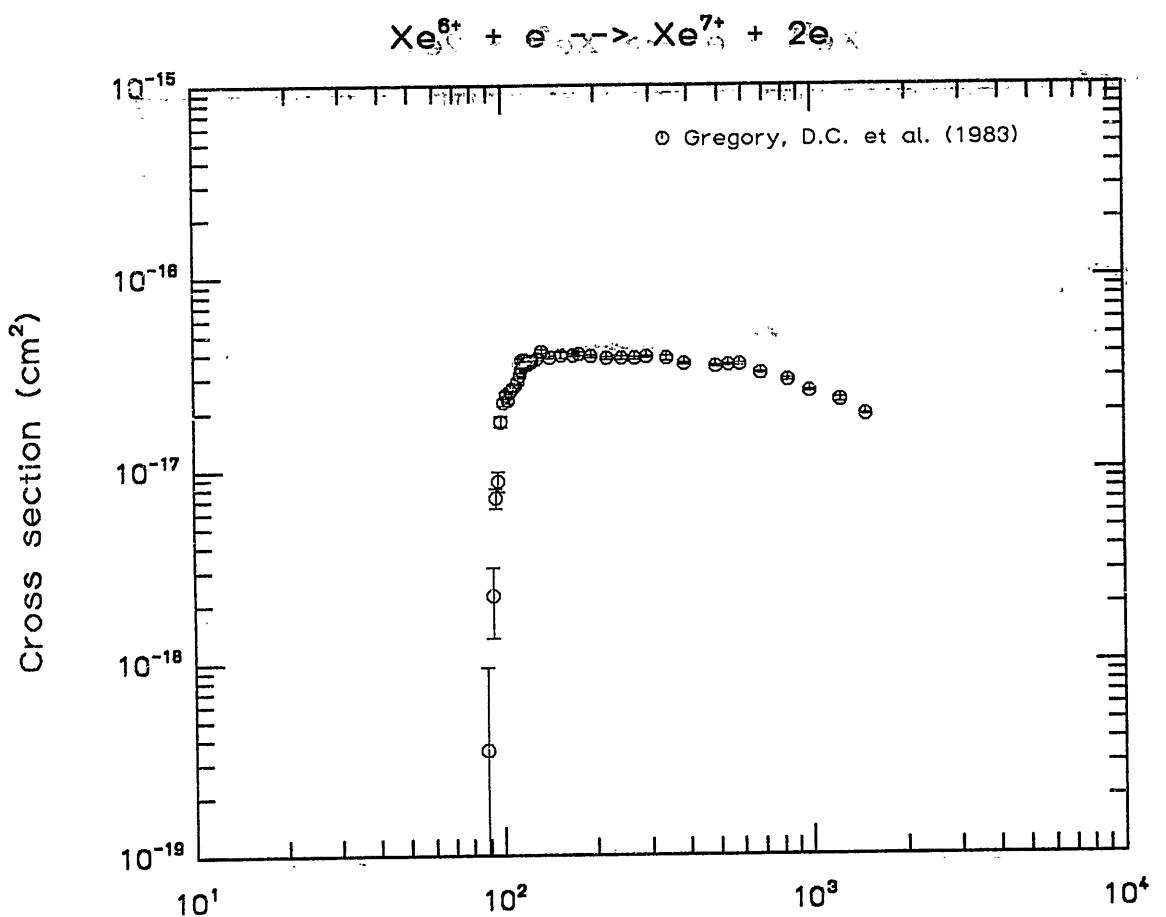


Fig. 254      Electron energy (eV)       $\Sigma \bar{\sigma} \text{ cm}^2$

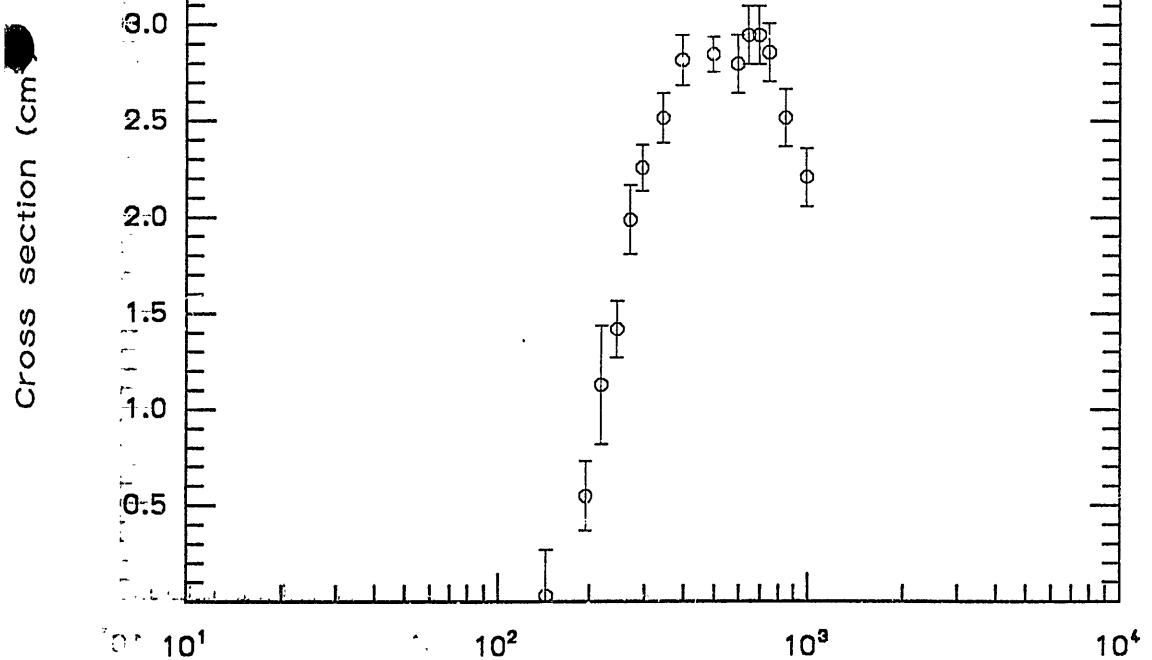
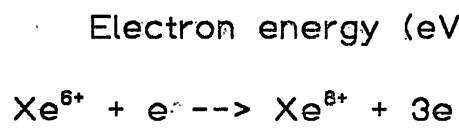
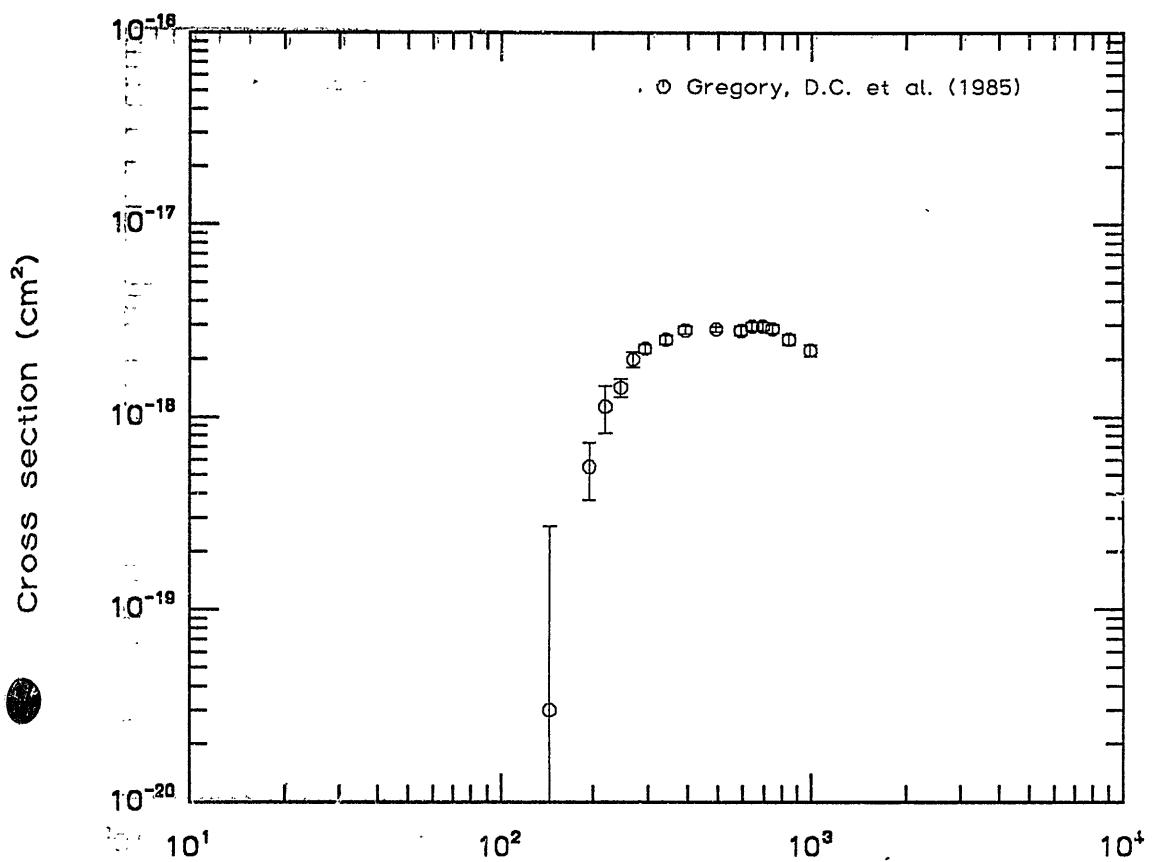
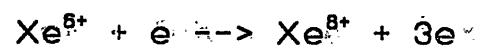


Fig. 255 Electron energy (eV)

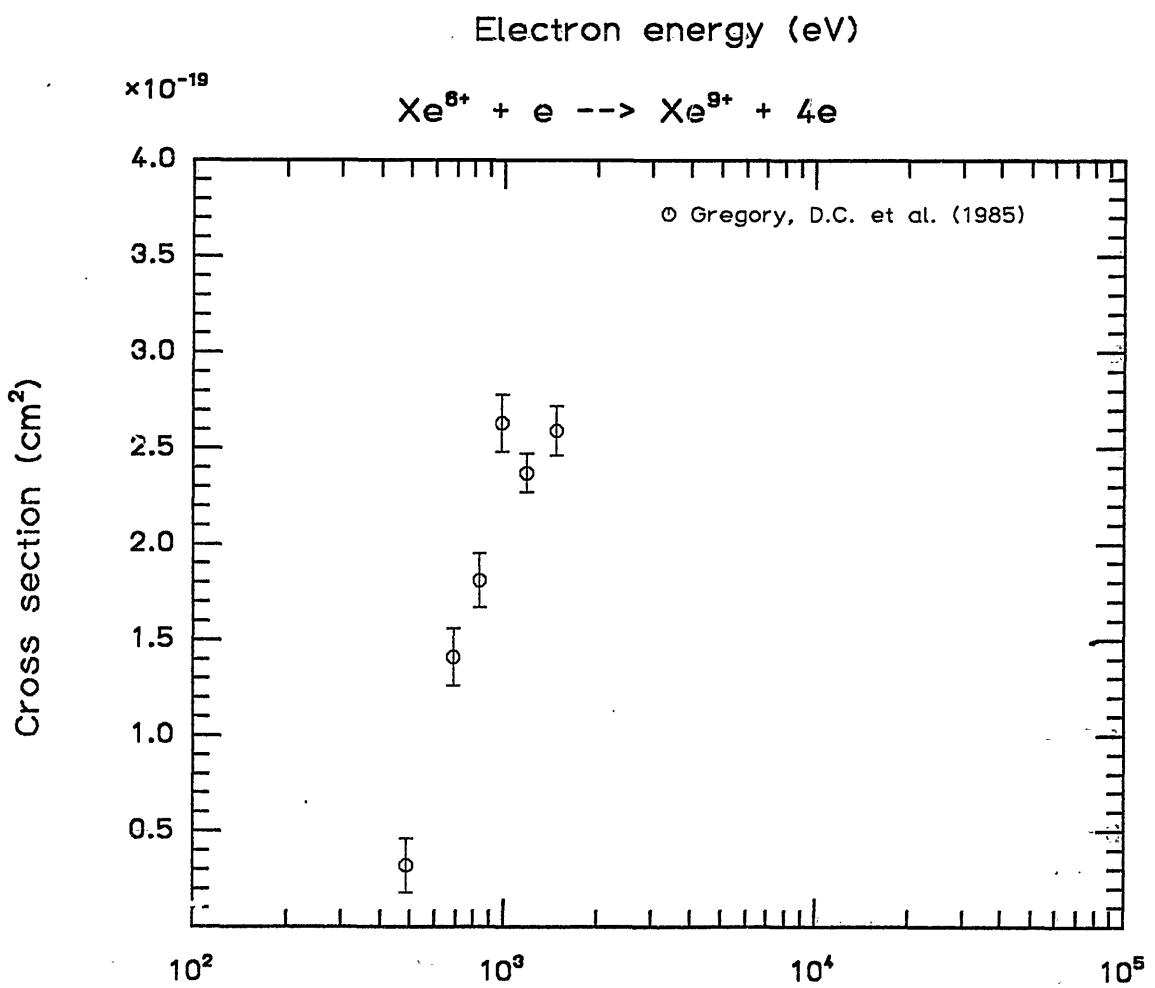
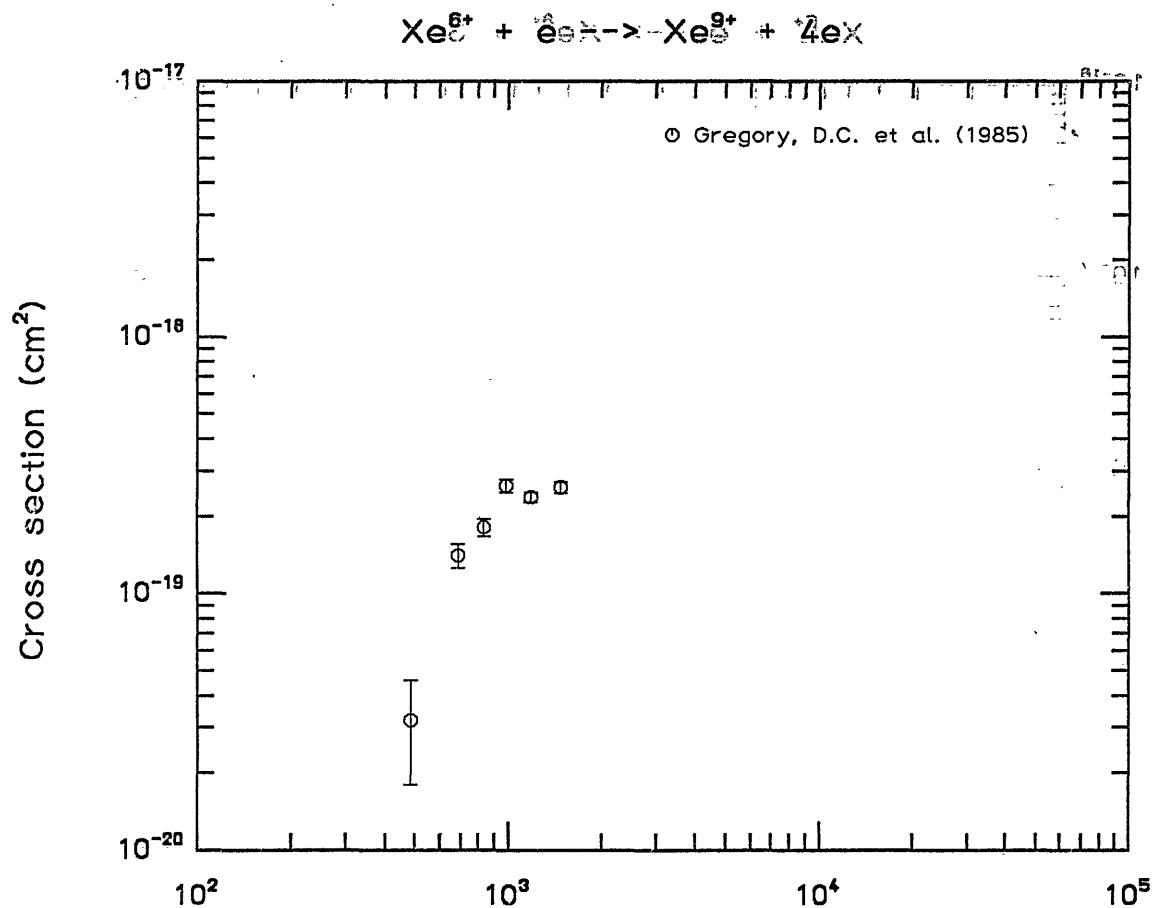


Fig. 256      Electron energy (eV)      225 pB

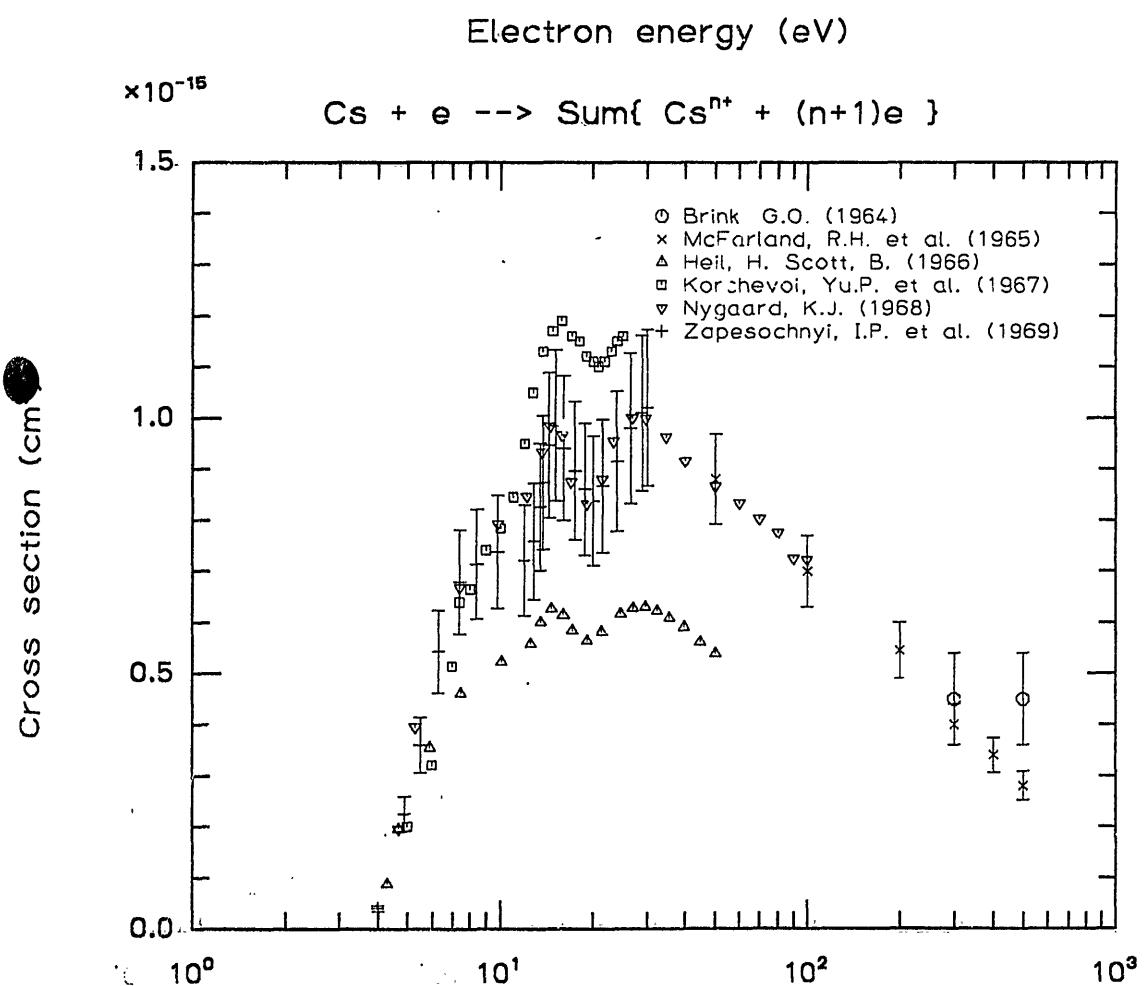
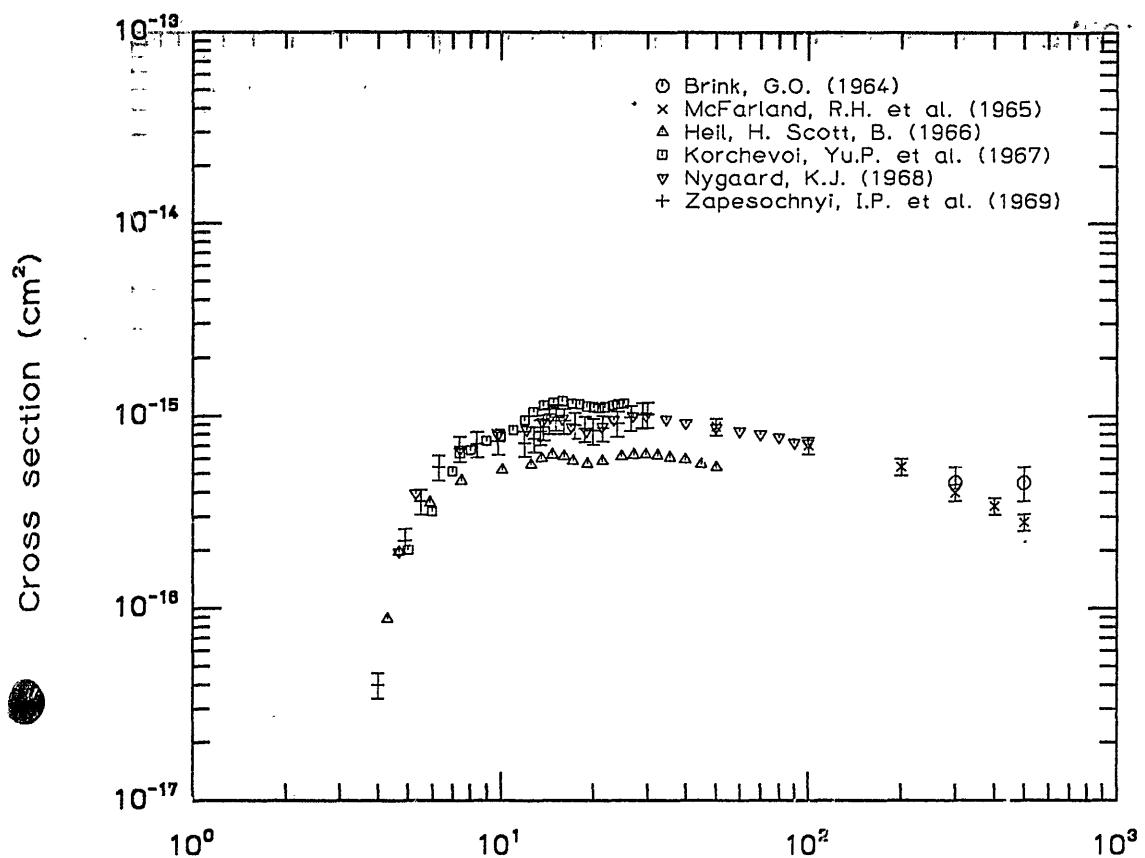
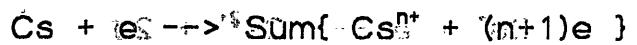


Fig. 257. The Electron energy (eV) 885 pp. 1

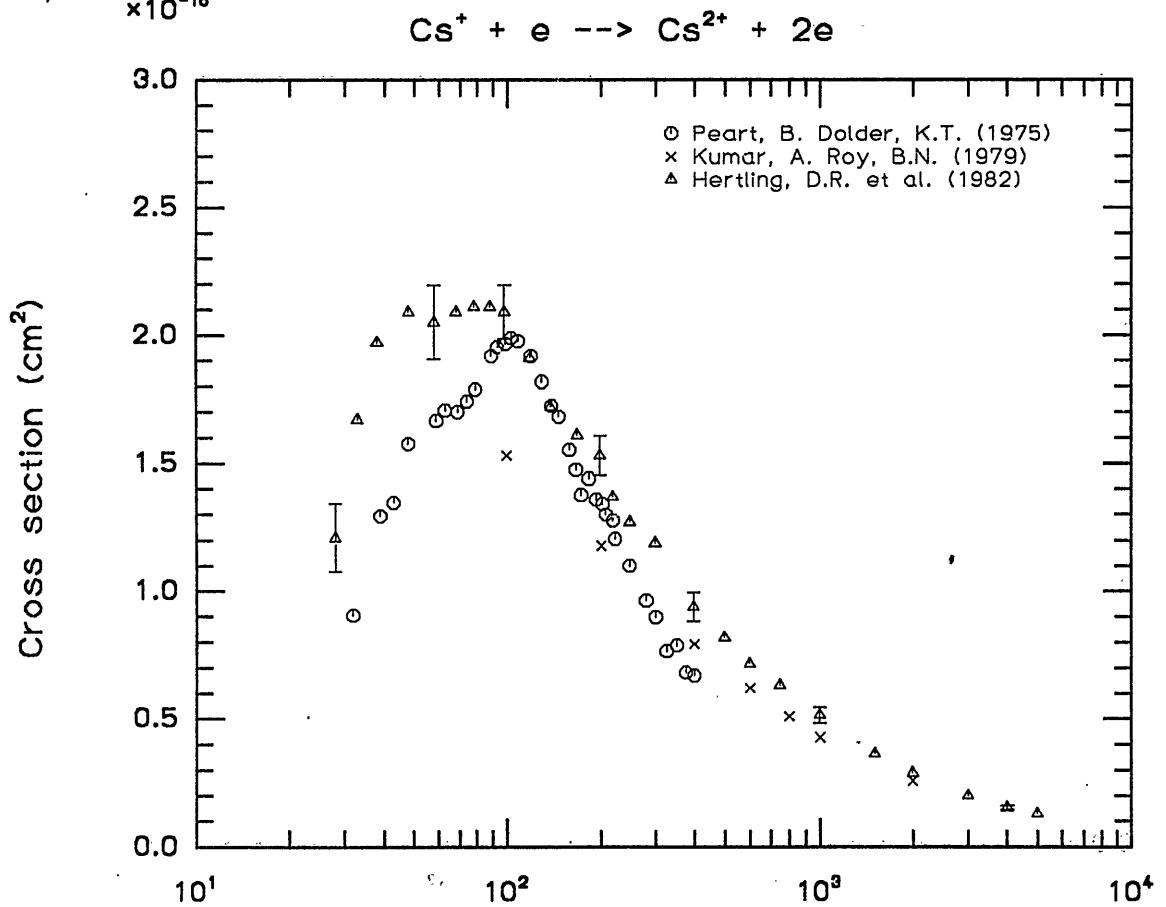
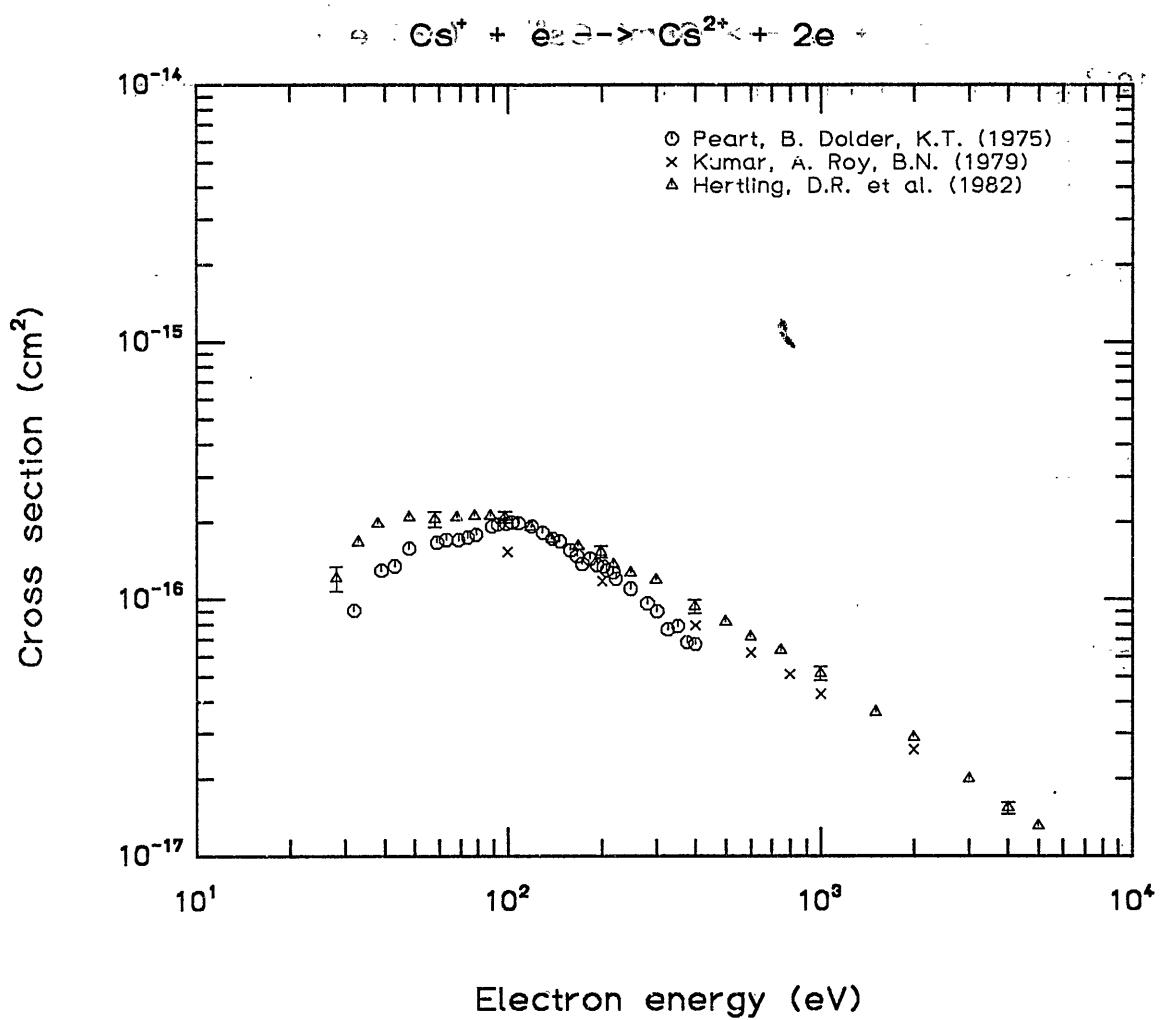


Fig. 258 Electron energy (eV)  $\text{Cs}^+ + \text{e}^- \rightarrow \text{Cs}^{2+} + 2\text{e}^-$

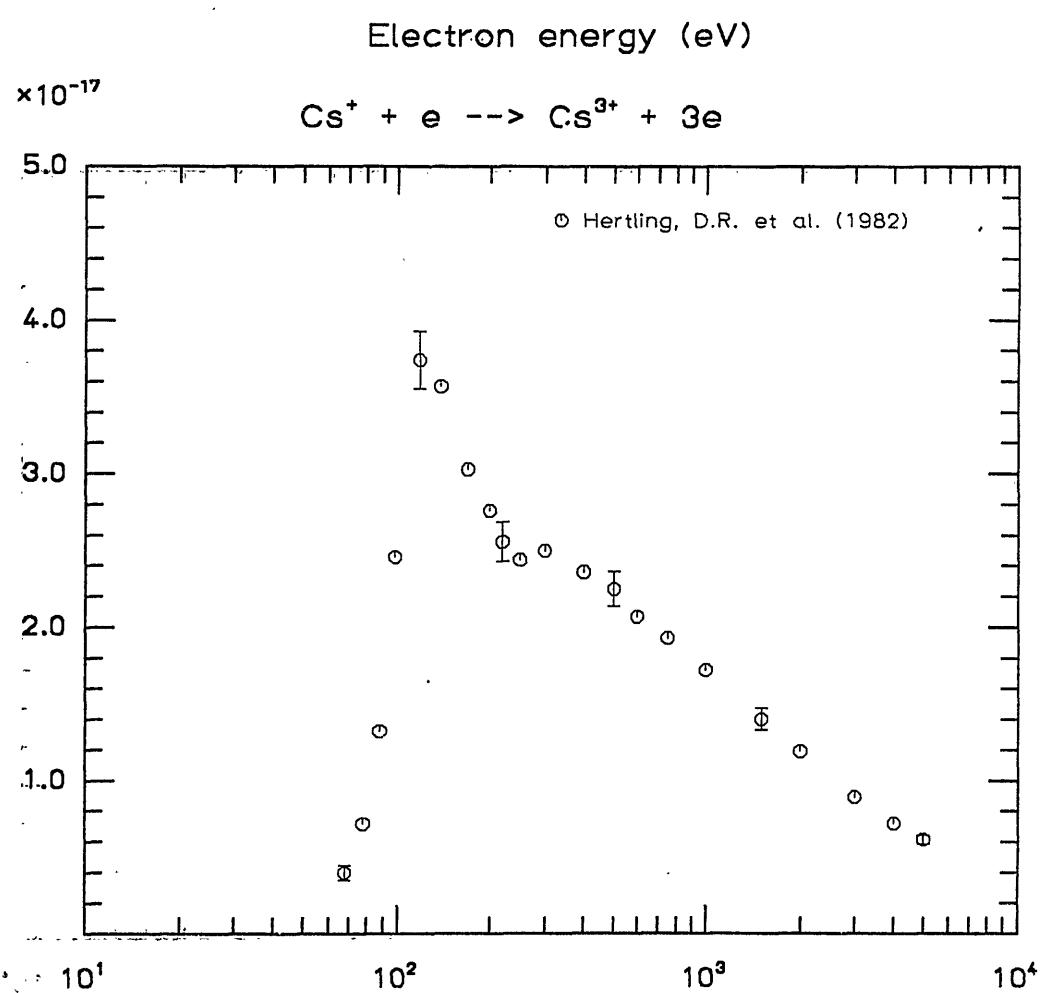
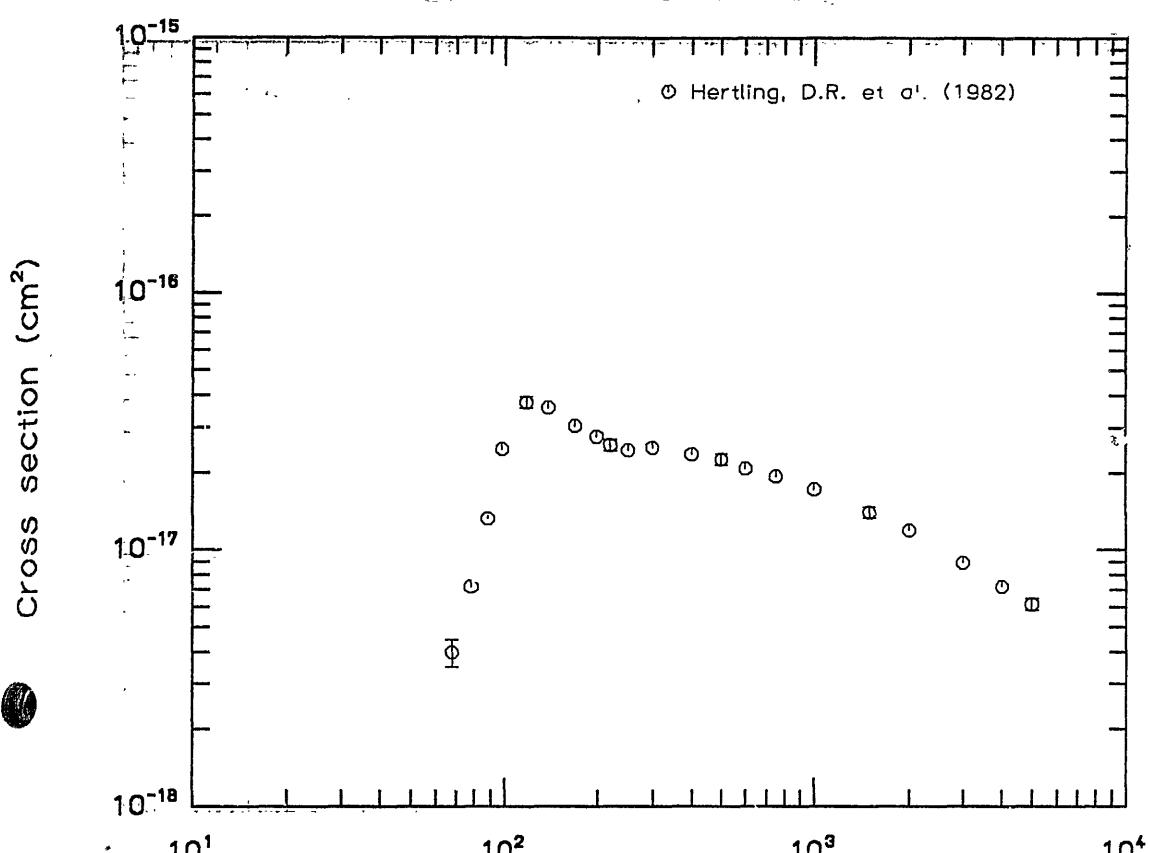
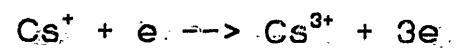


Fig. 259

Electron energy (eV)

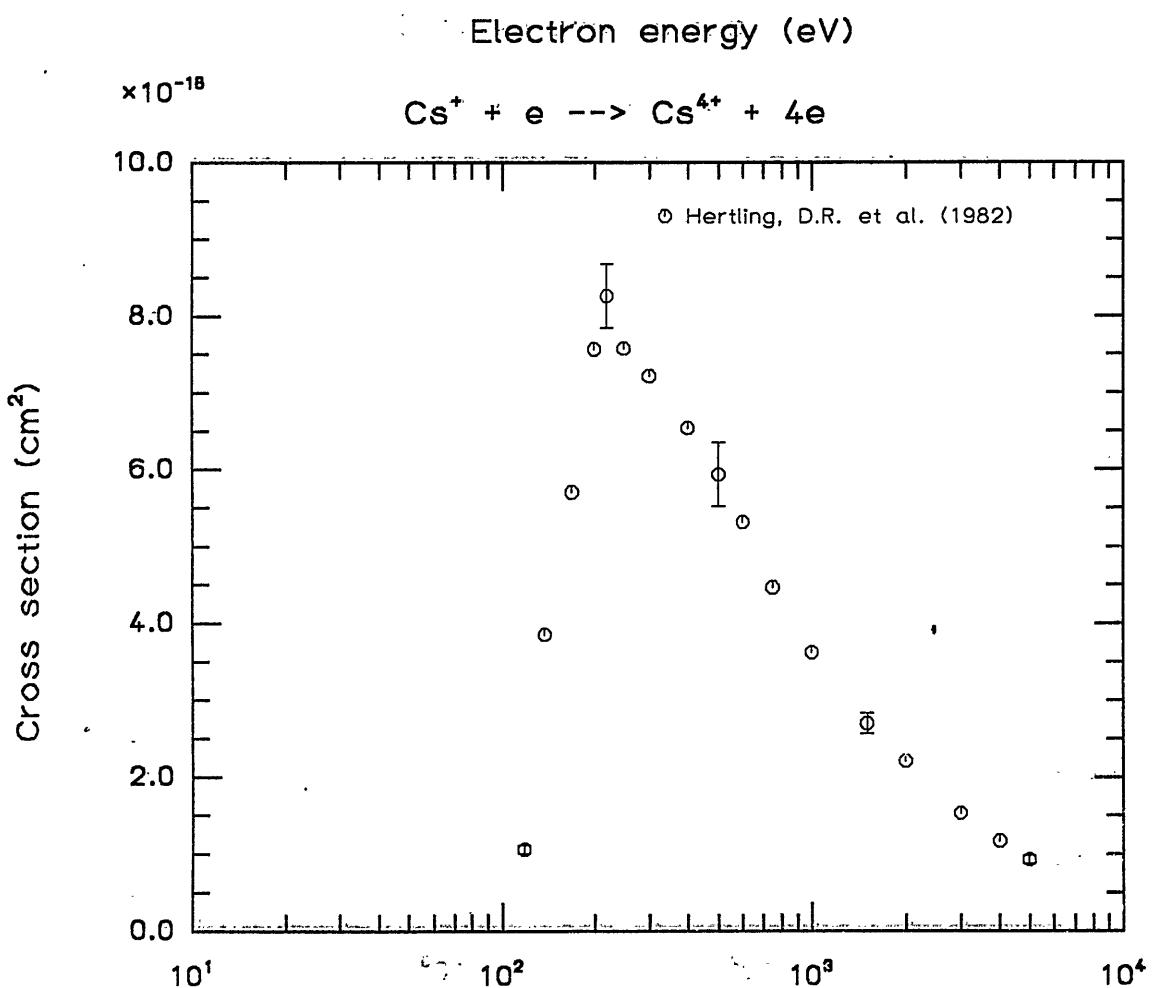
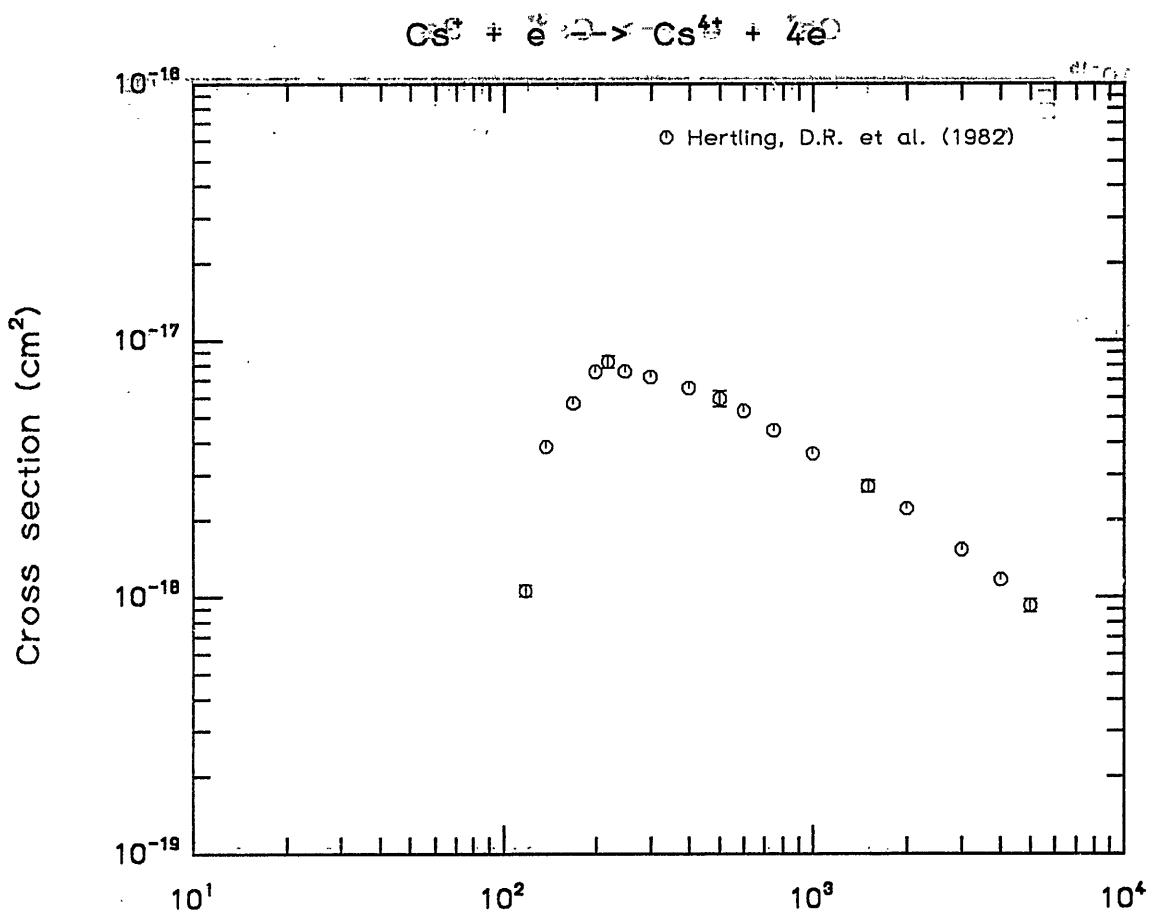
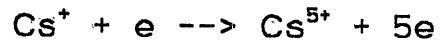
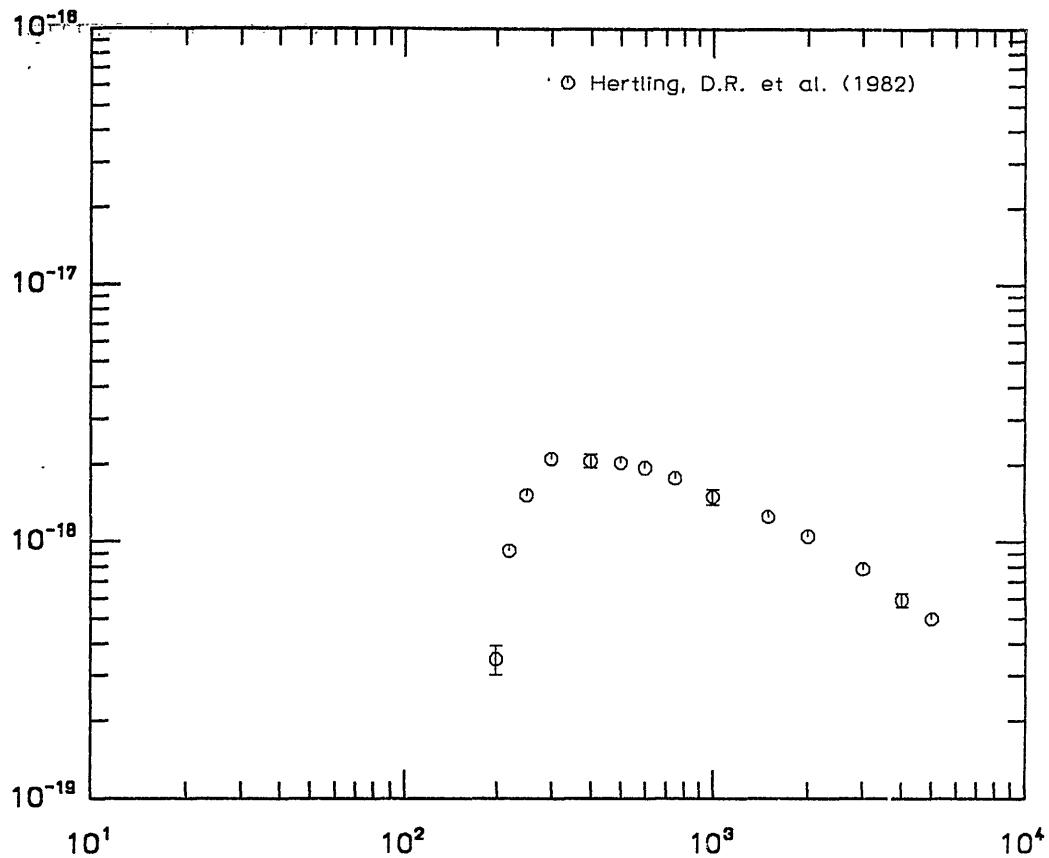


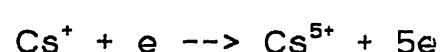
Fig. 260 Electron energy (eV)  $\text{Cs}^+ + e^- \rightarrow \text{Cs}^{4+} + 4e^-$



• Cross section ( $\text{cm}^2$ )



Electron energy (eV)



• Cross section ( $\text{cm}^2$ )

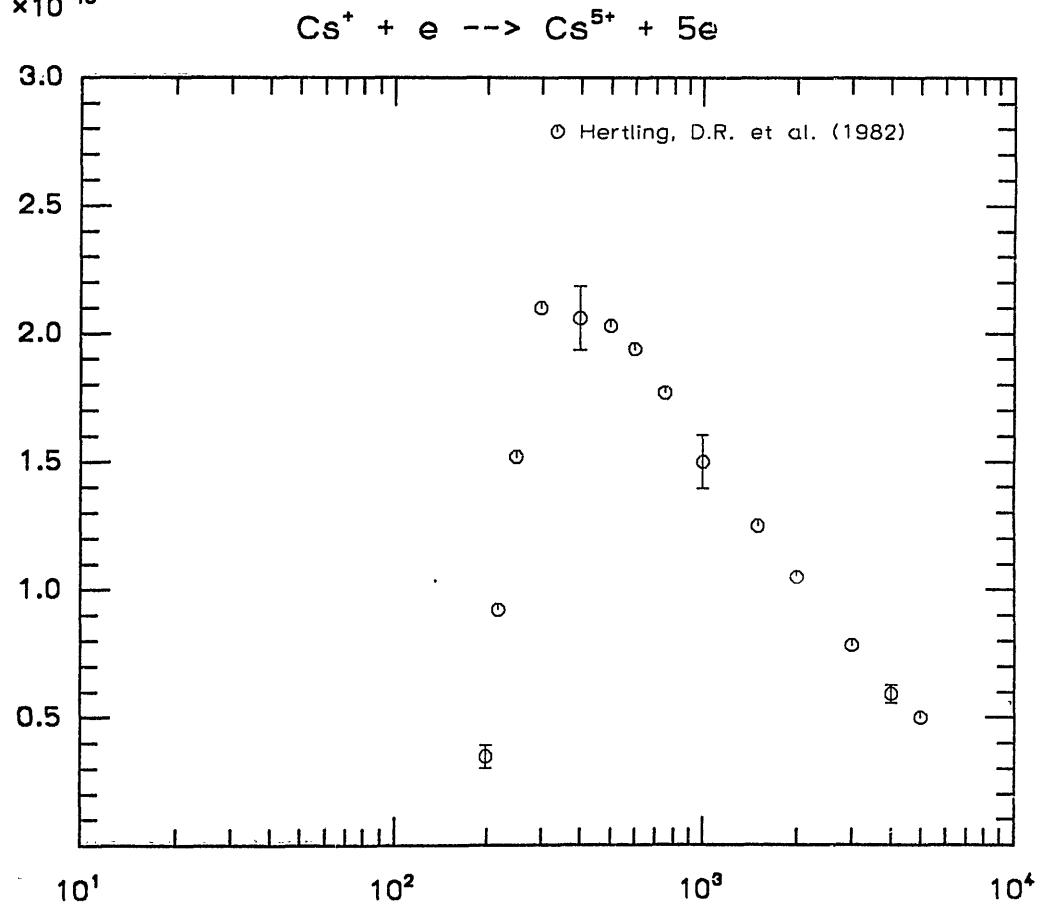


Fig. 261

Electron energy (eV)

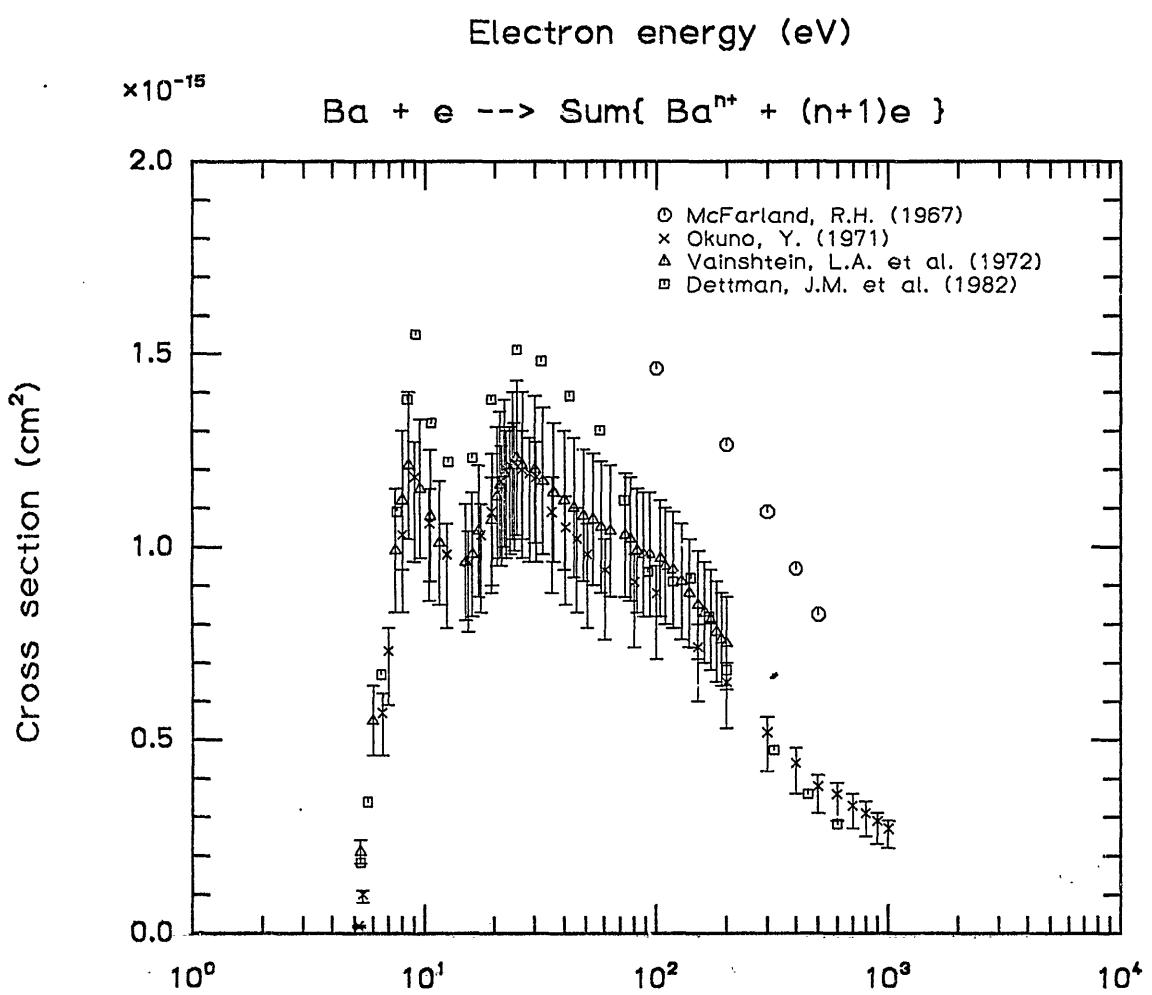
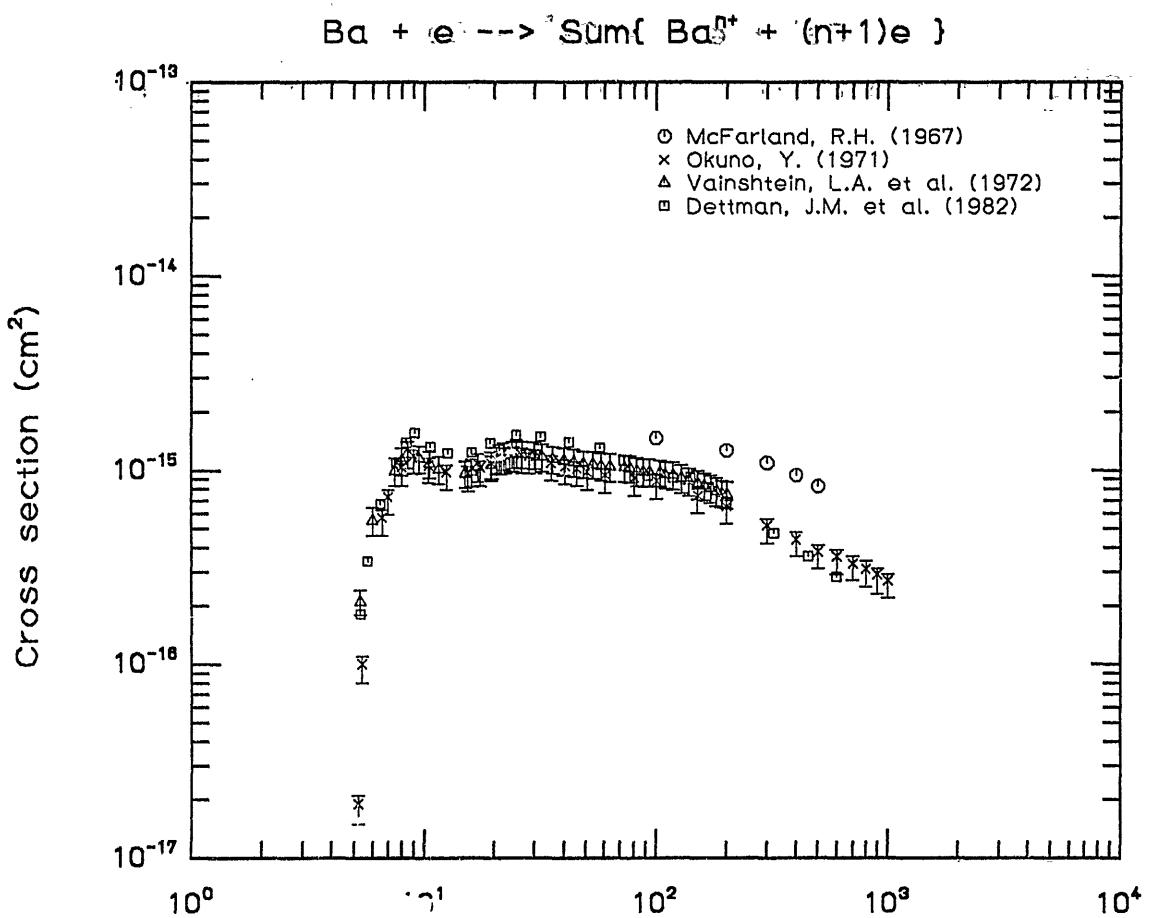
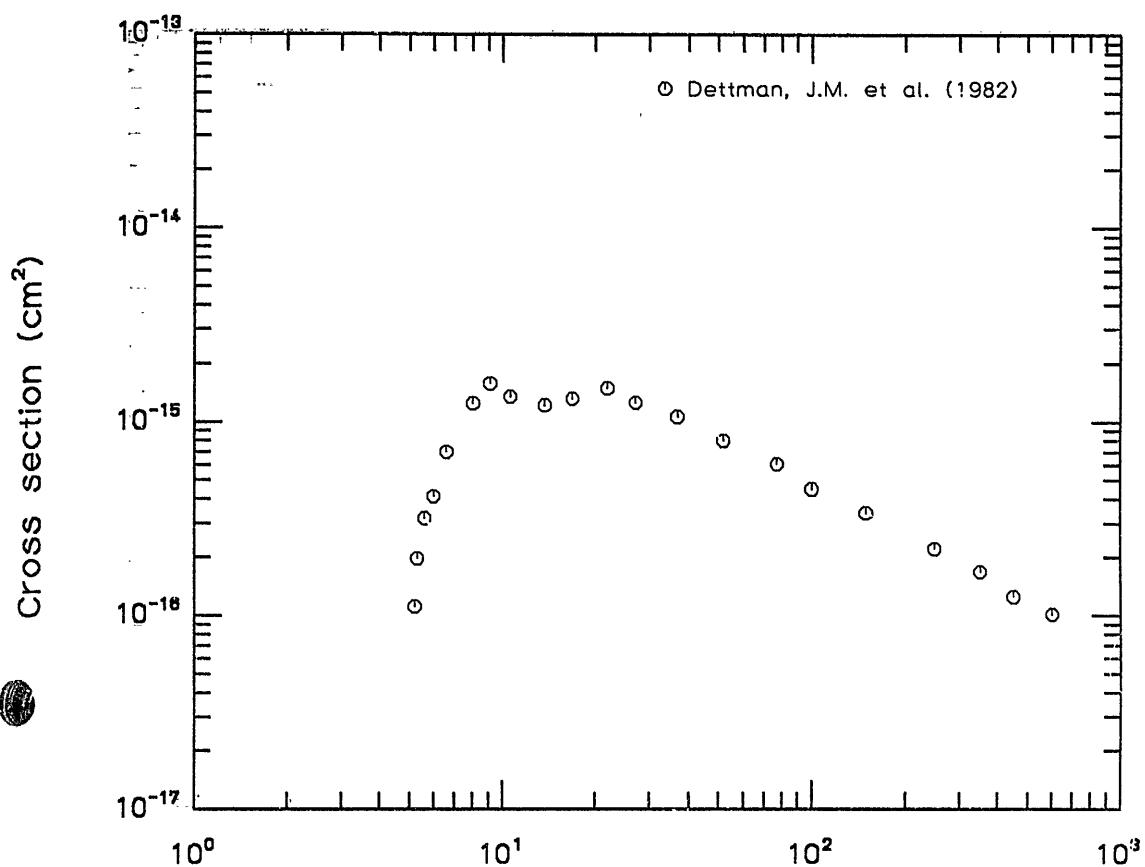
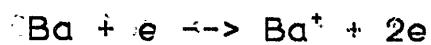


Fig. 262 Electron energy (eV) 10<sup>0</sup> to 10<sup>4</sup>



Electron energy (eV)

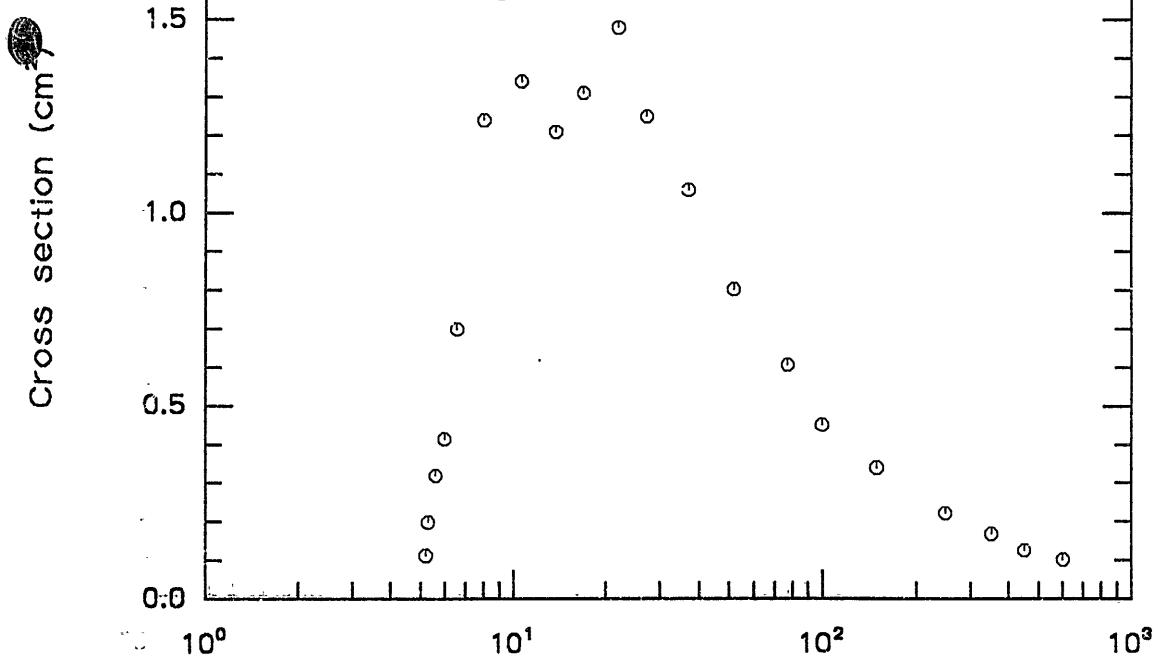
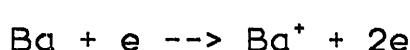


Fig. 263

Electron energy (eV)

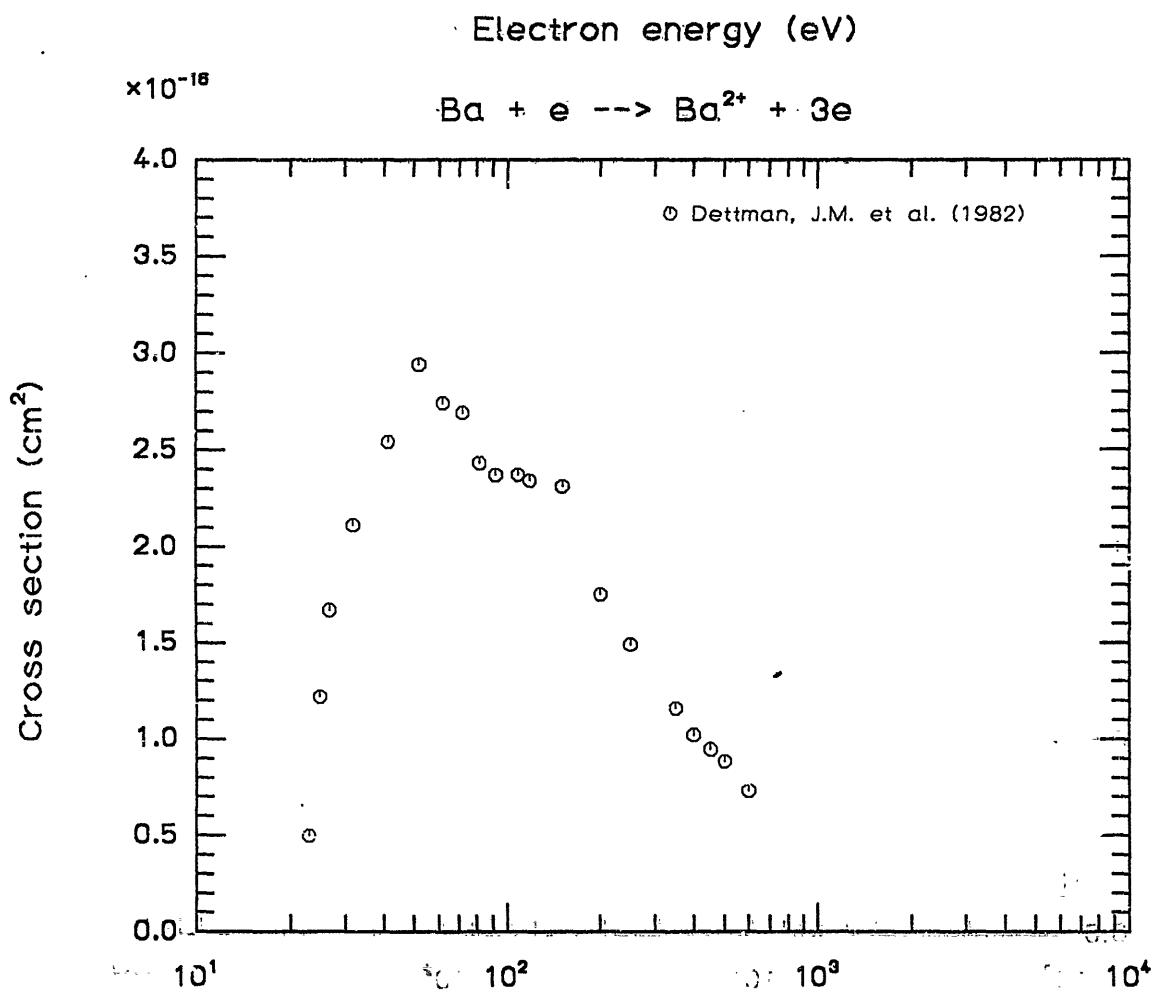
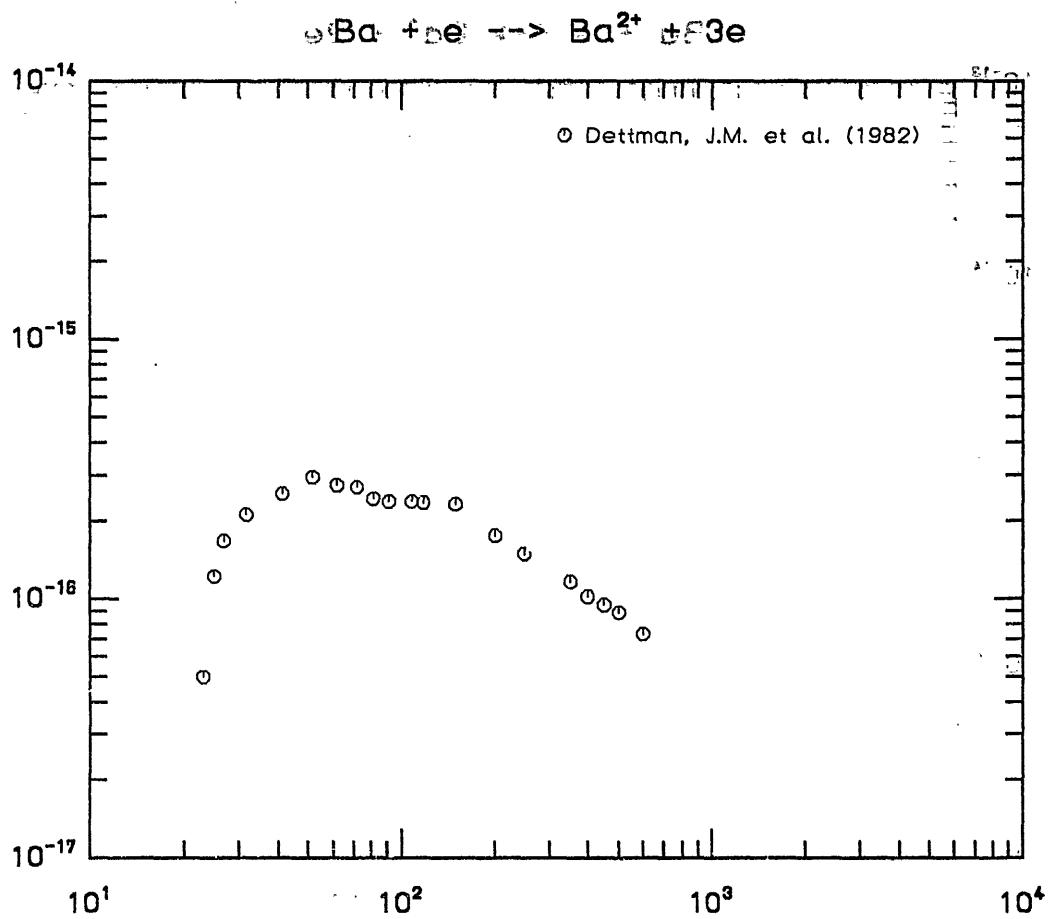


Fig. 264 (V) Electron energy (eV) Fig. 263

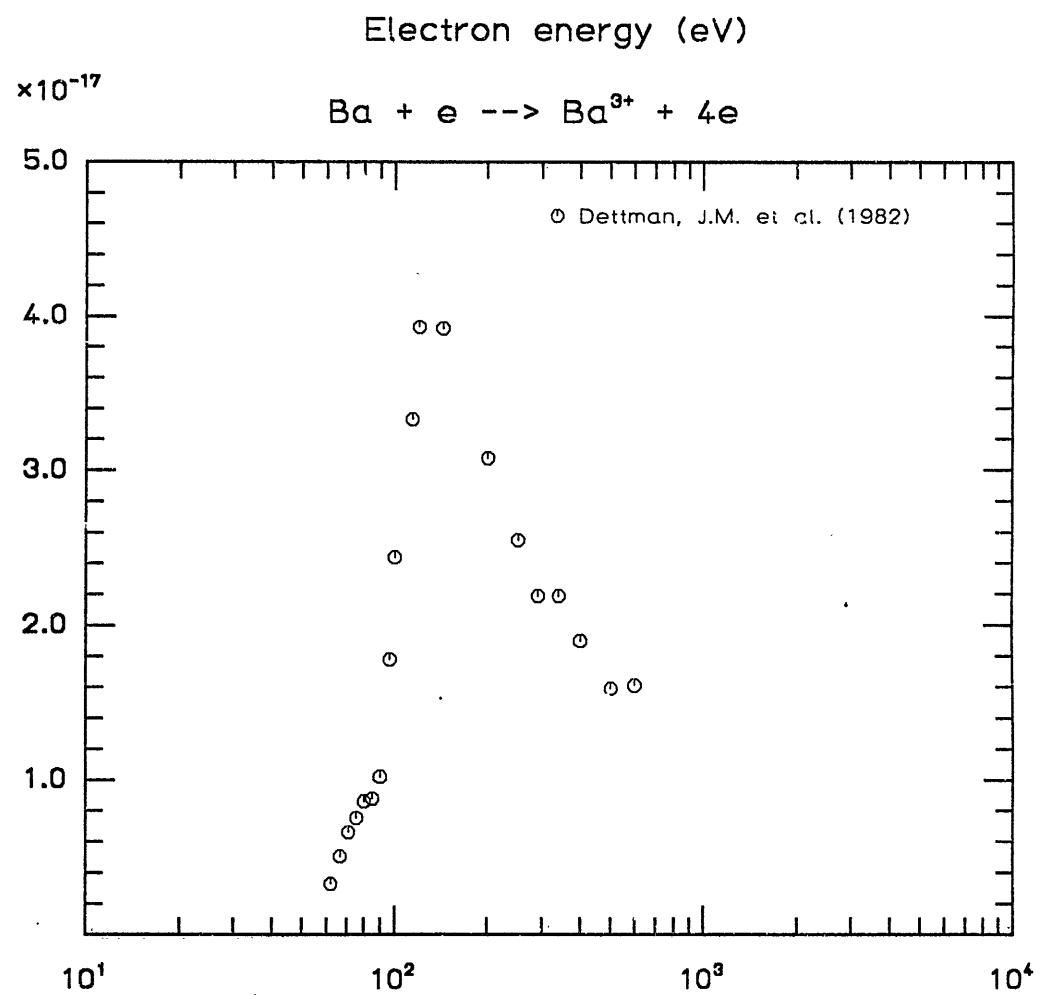
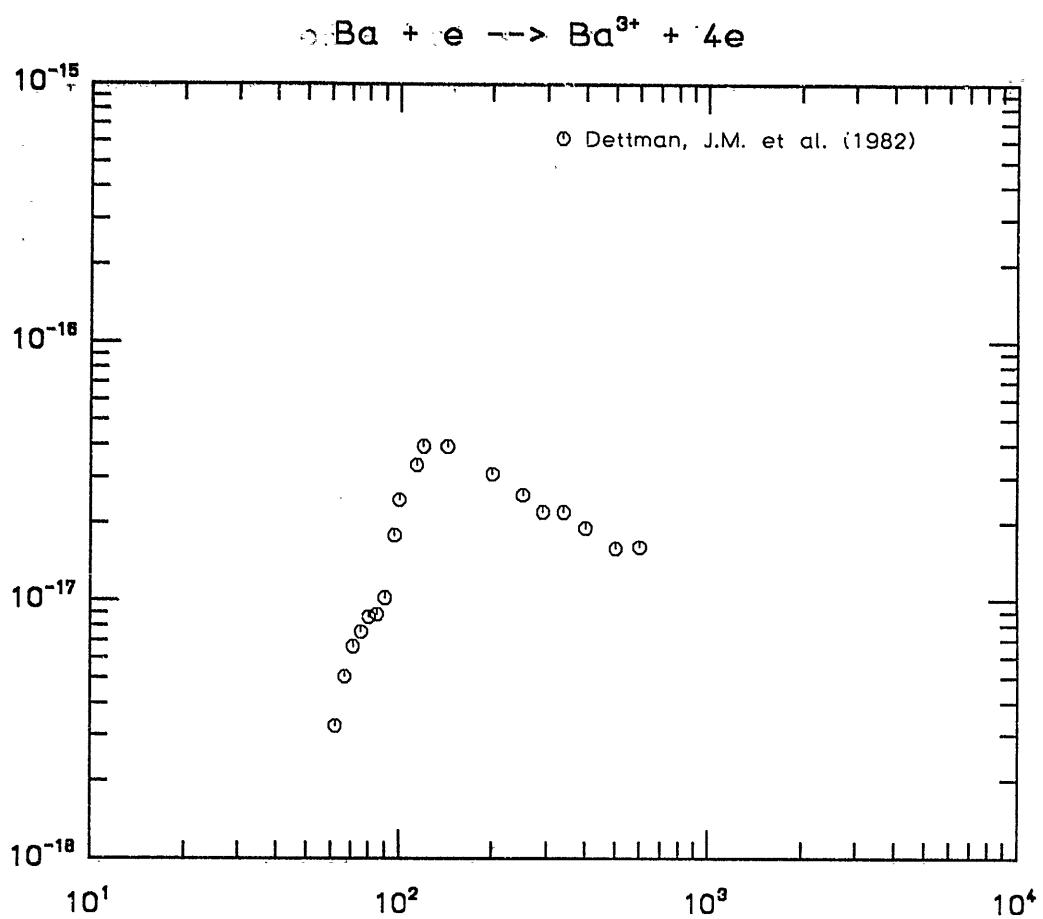


Fig. 265 · Electron energy (eV)

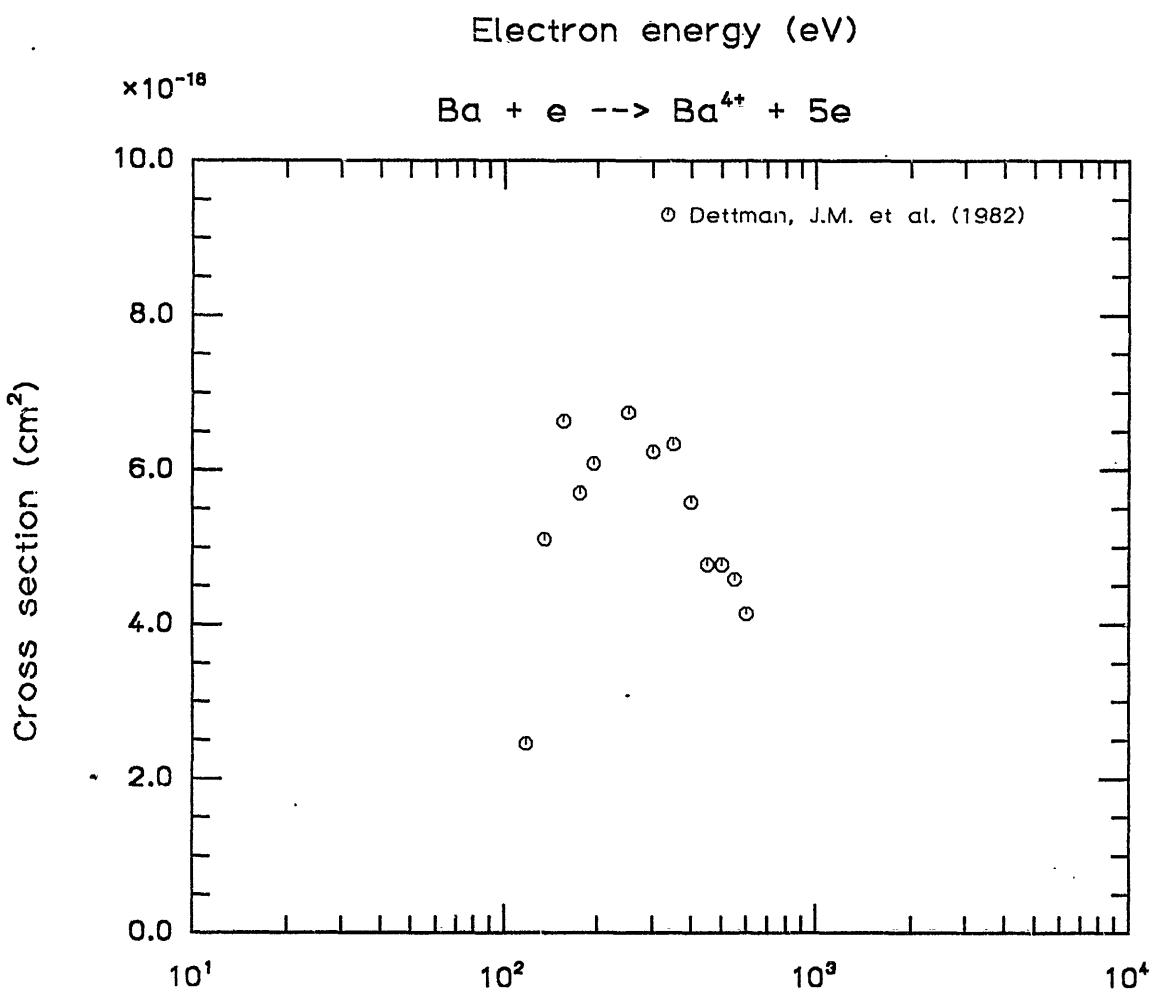
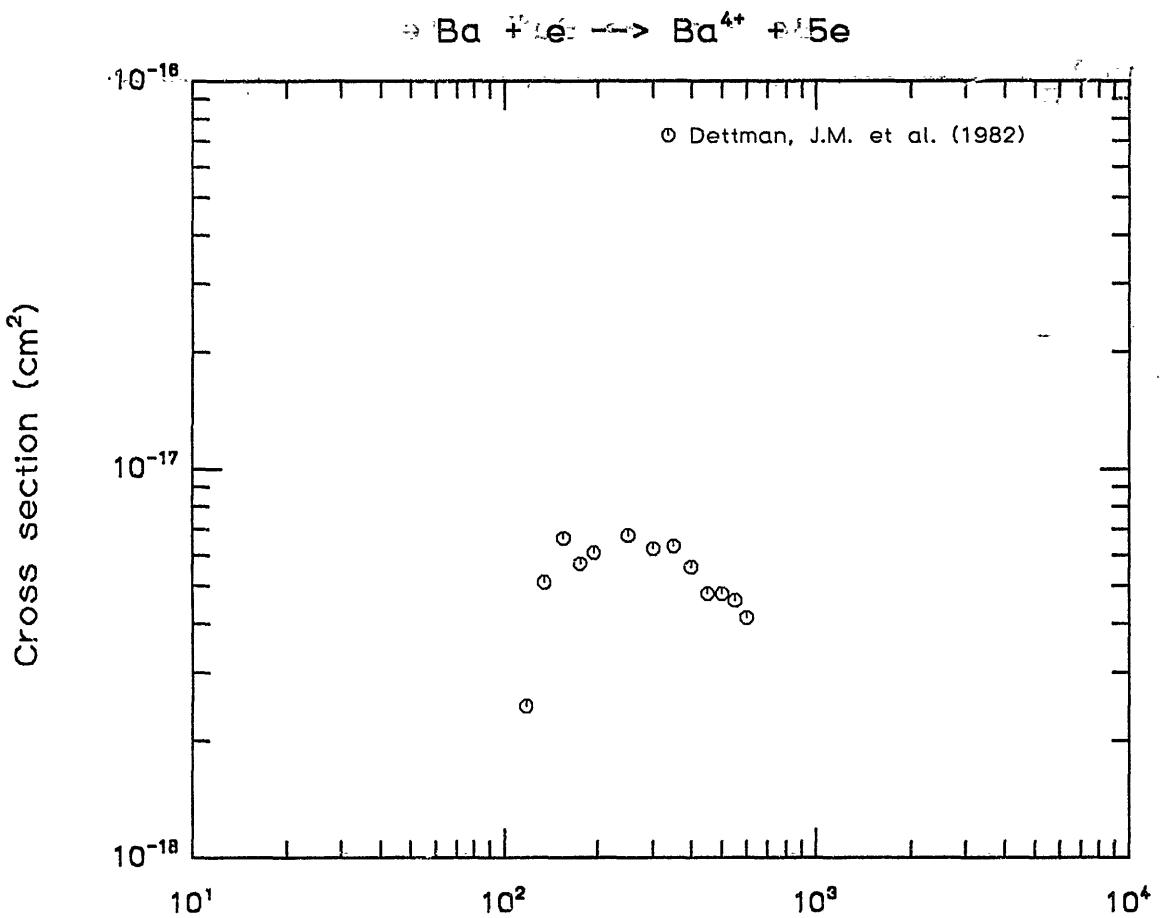


Fig. 266 Cross section vs Electron energy (eV)  $\text{Ba} + \text{e}^- \rightarrow \text{Ba}^{4+} + 5\text{e}$

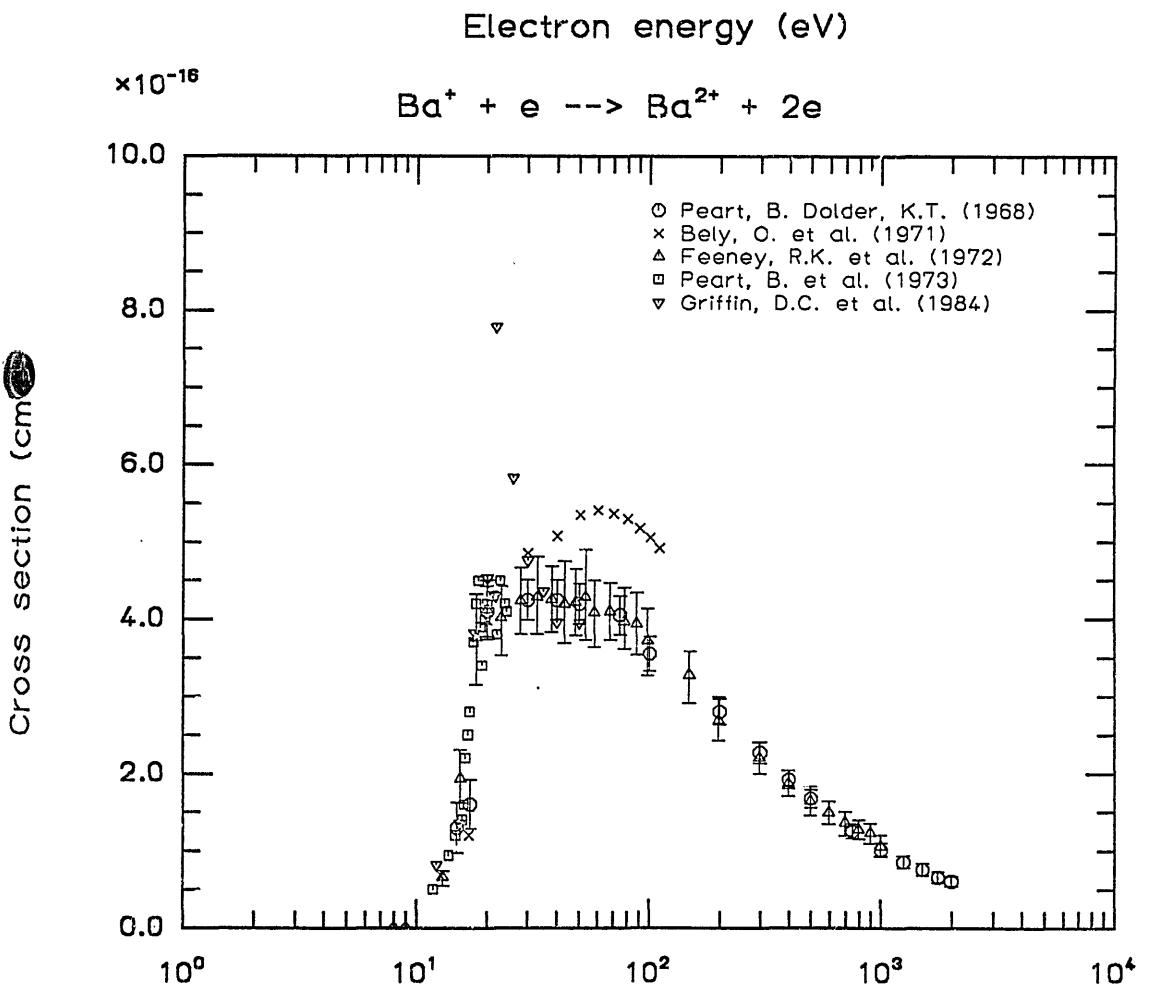
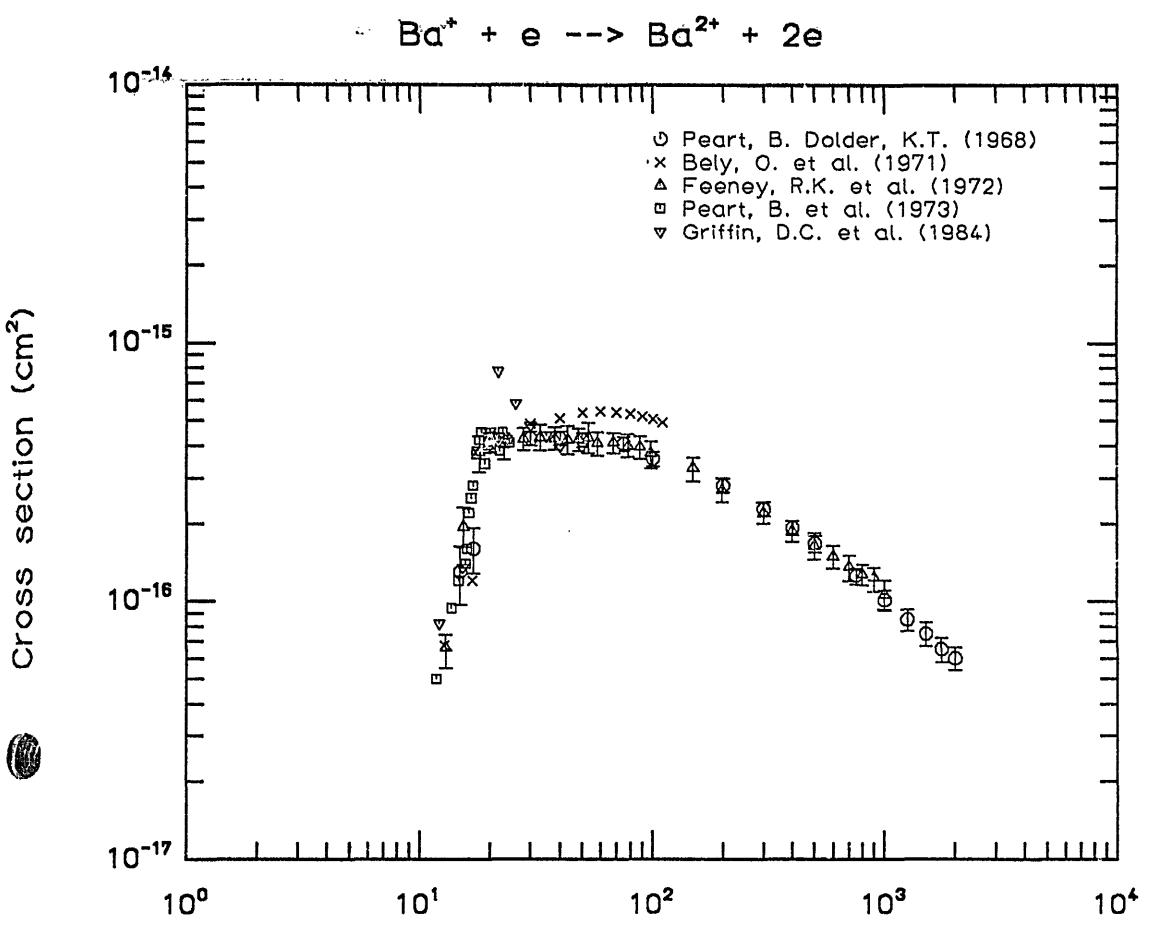


Fig. 267      Electron energy (eV)

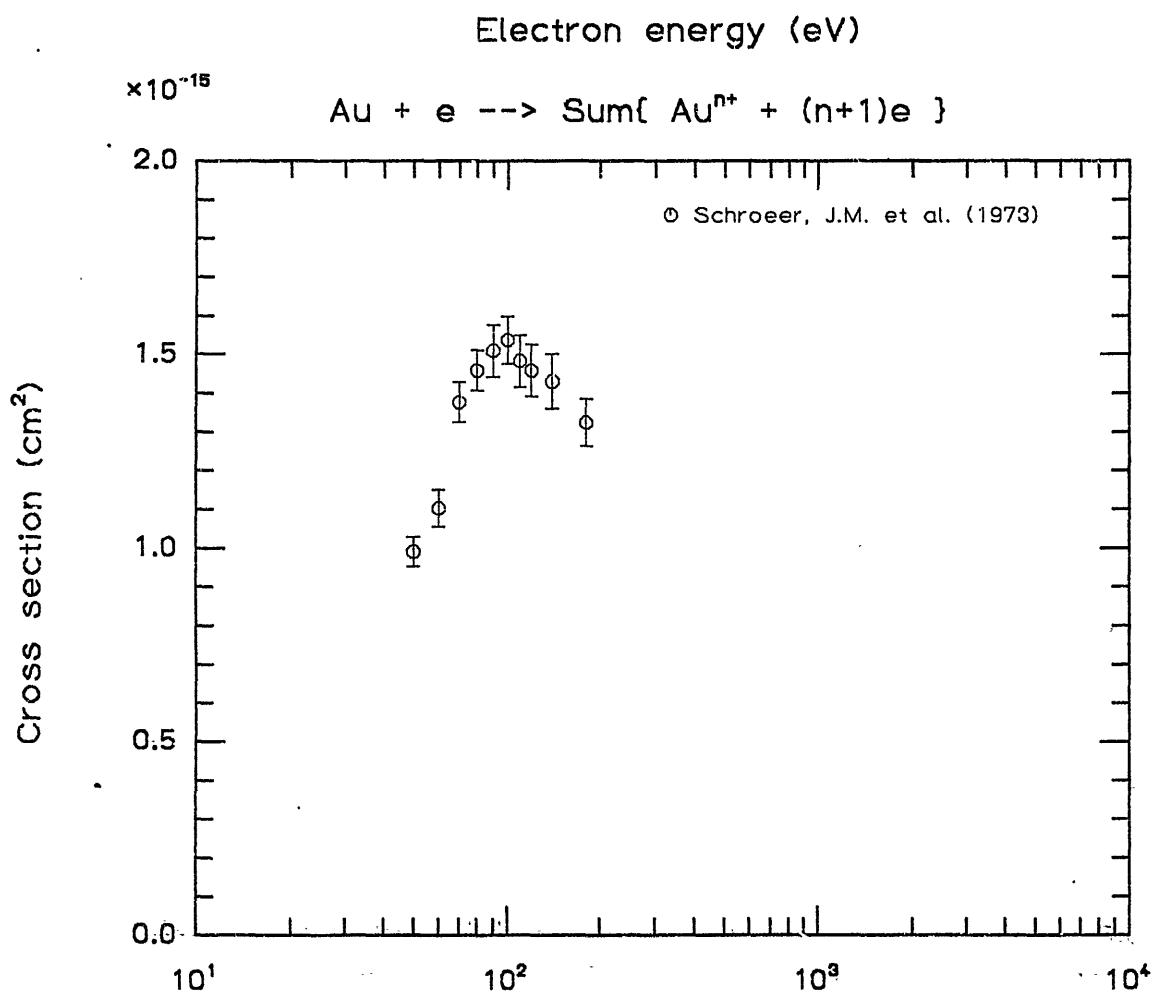
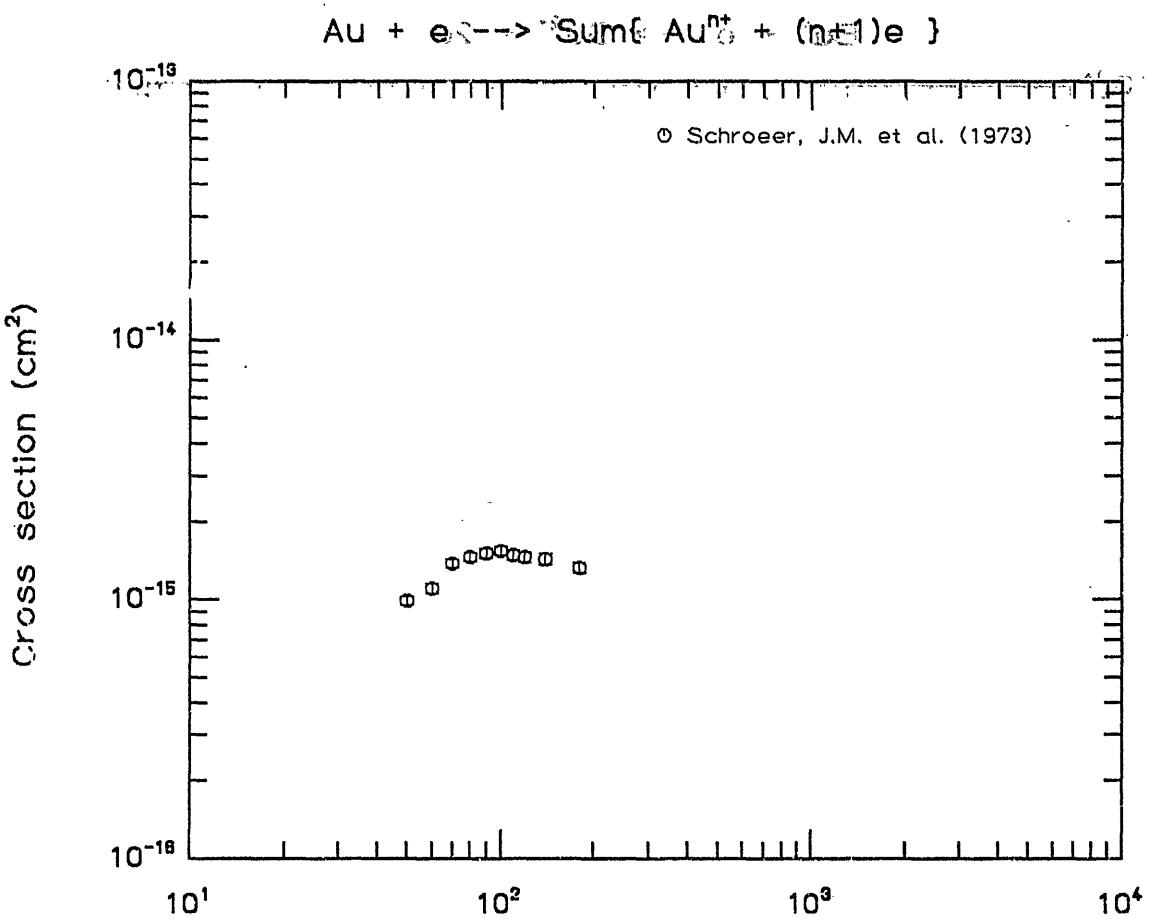


Fig. 268      Electron energy (eV)      Cross sec<sup>2</sup>

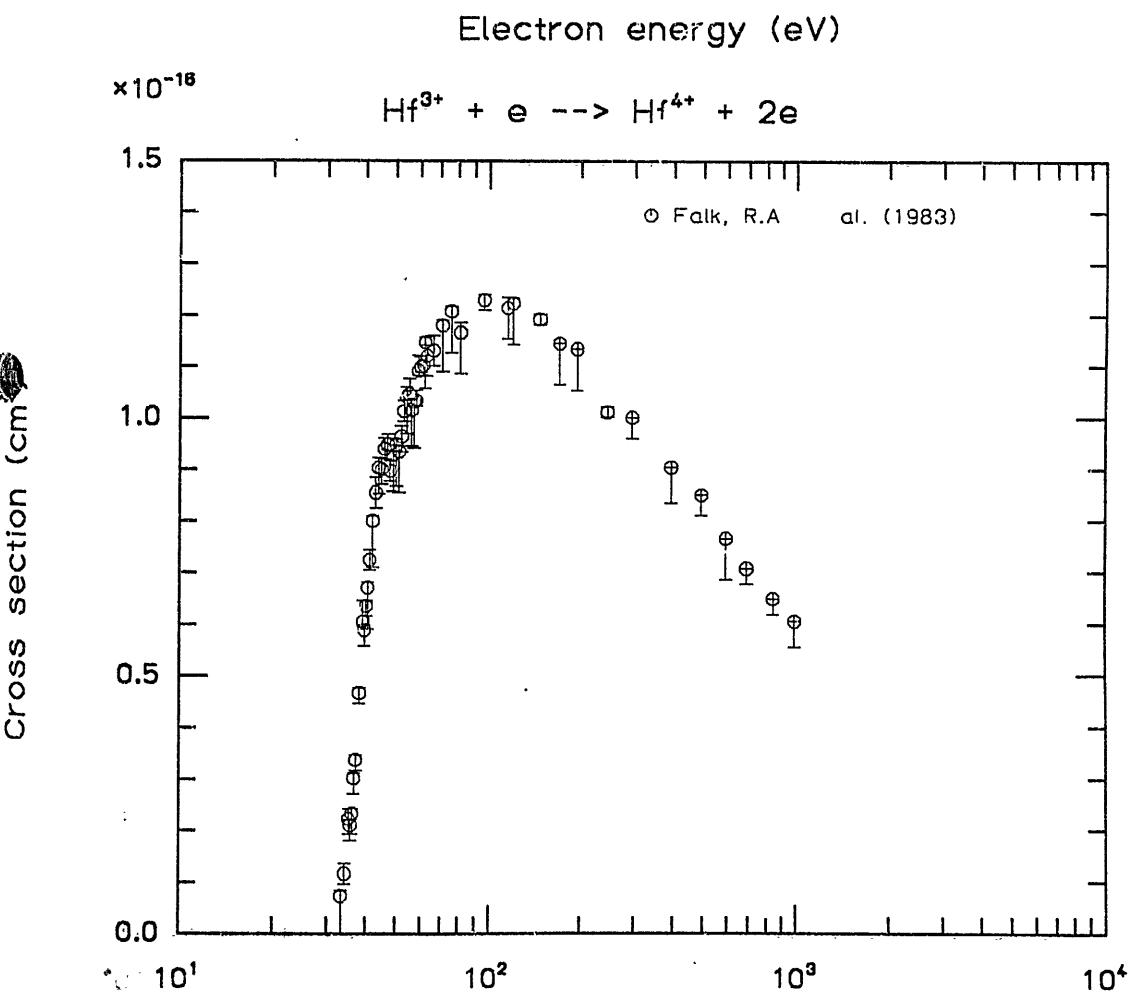
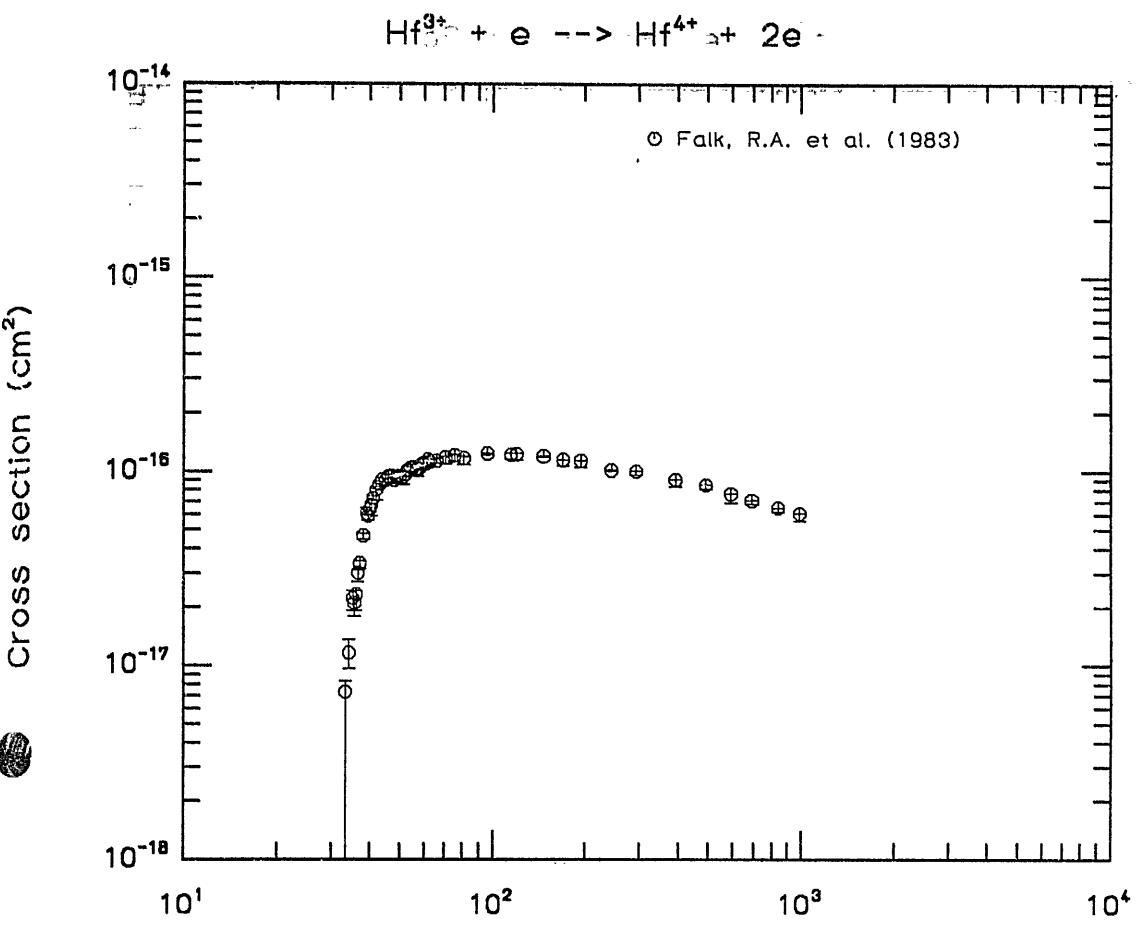


Fig. 269      Electron energy (eV)

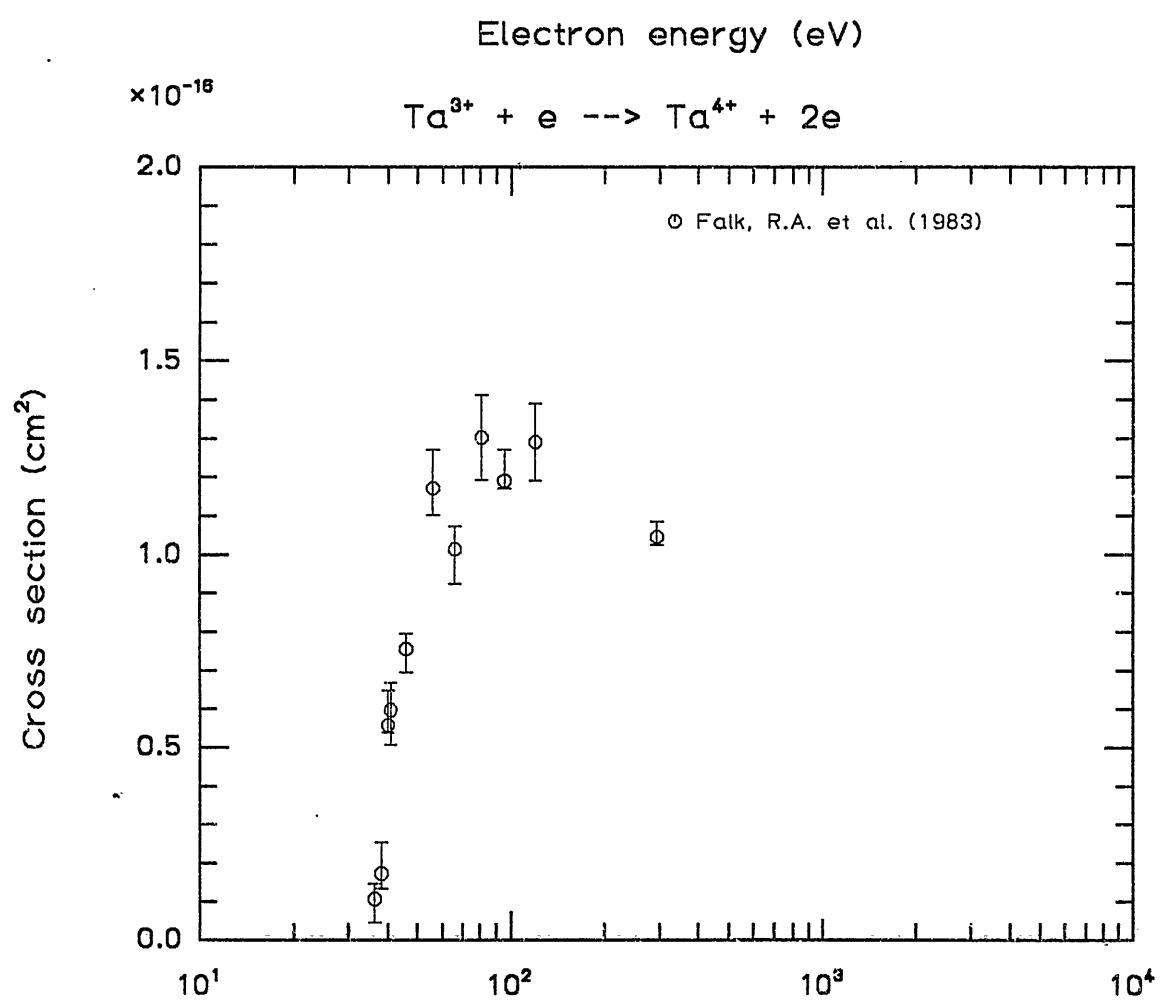
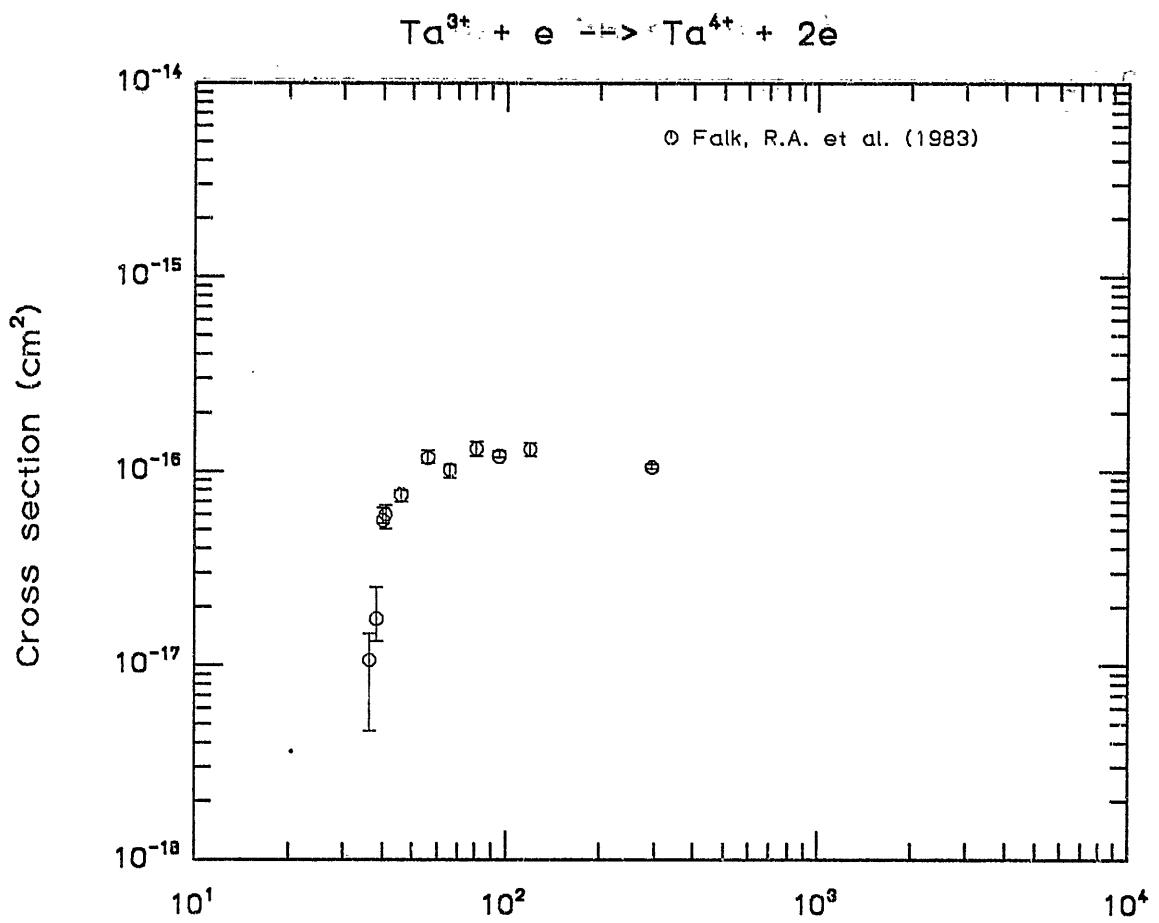


Fig. 270

Electron energy (eV)

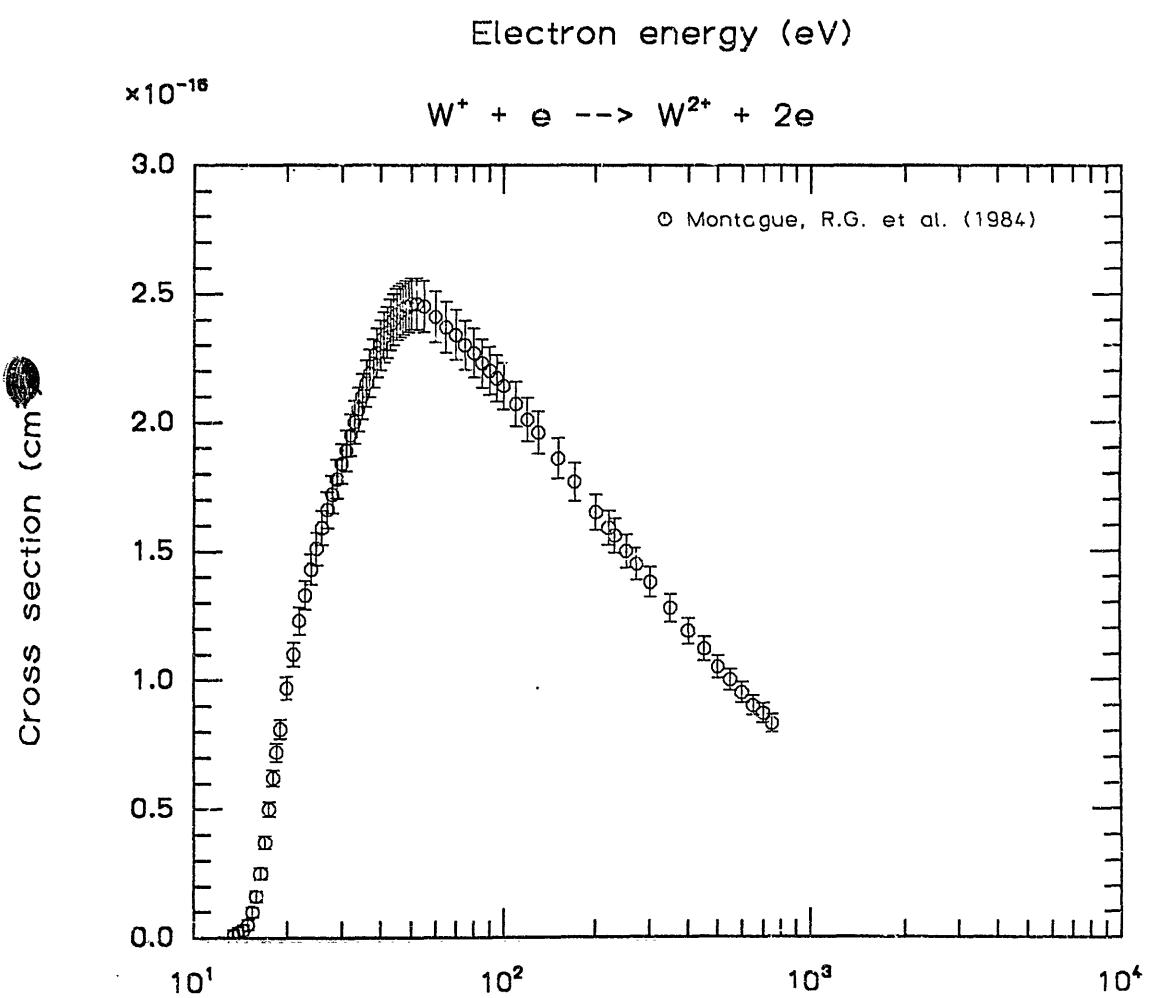
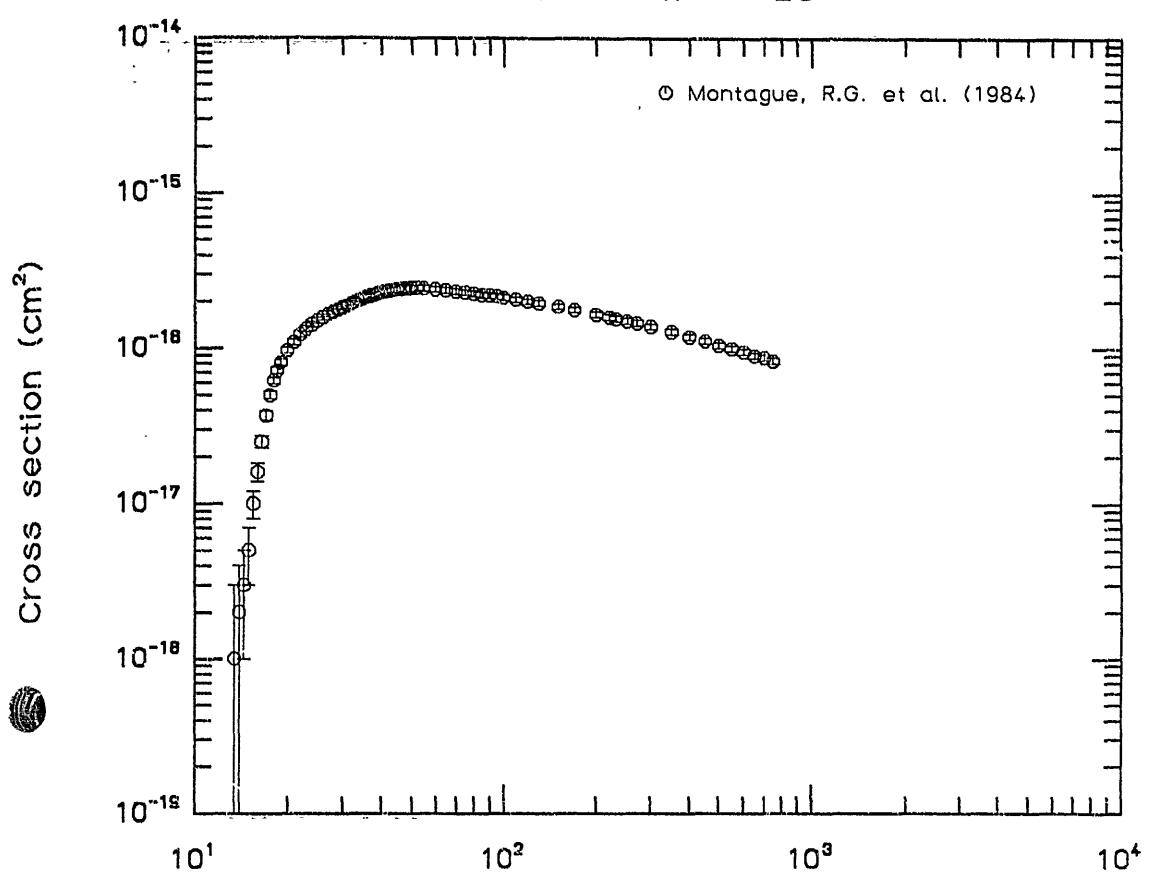
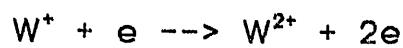


Fig. 271

Electron energy (eV)

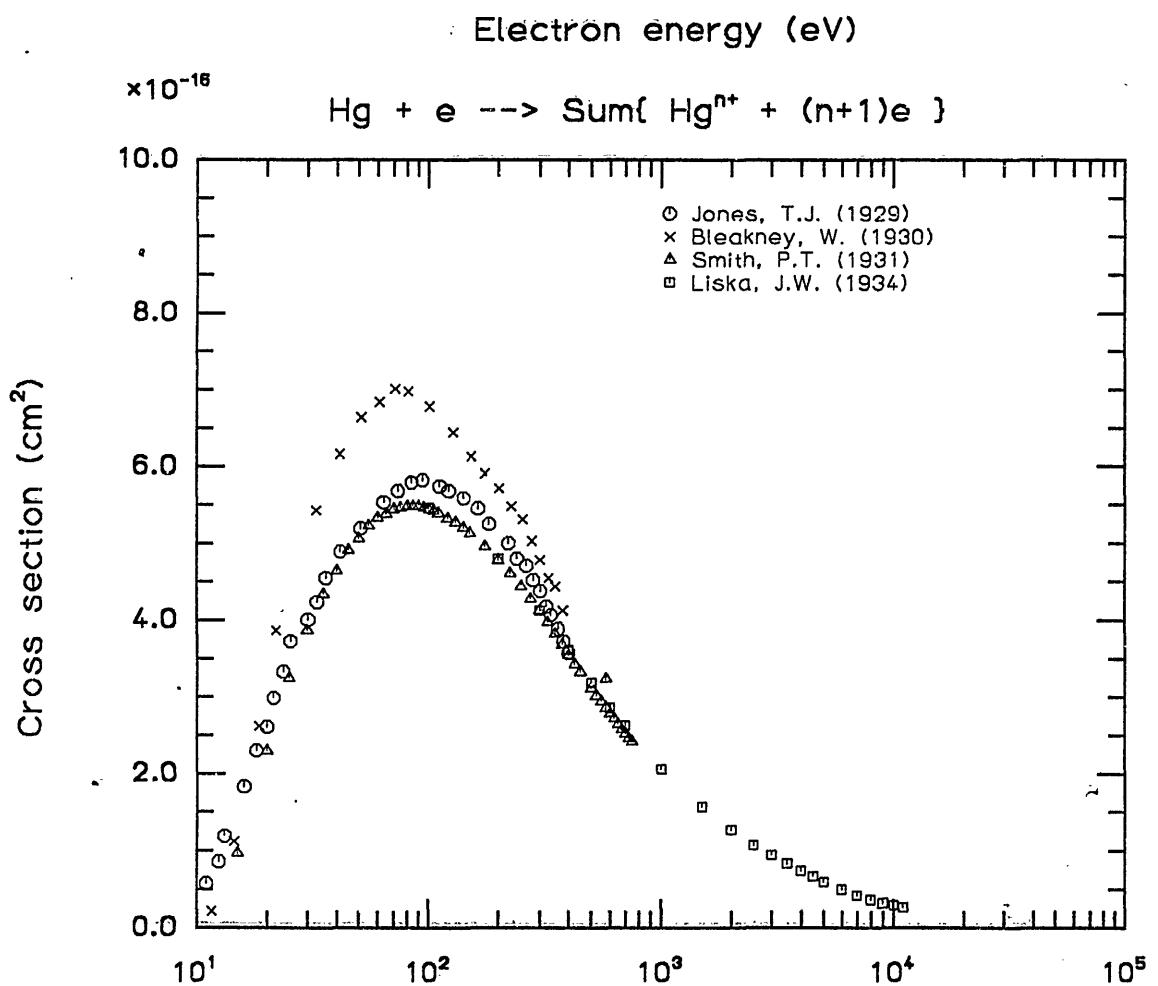
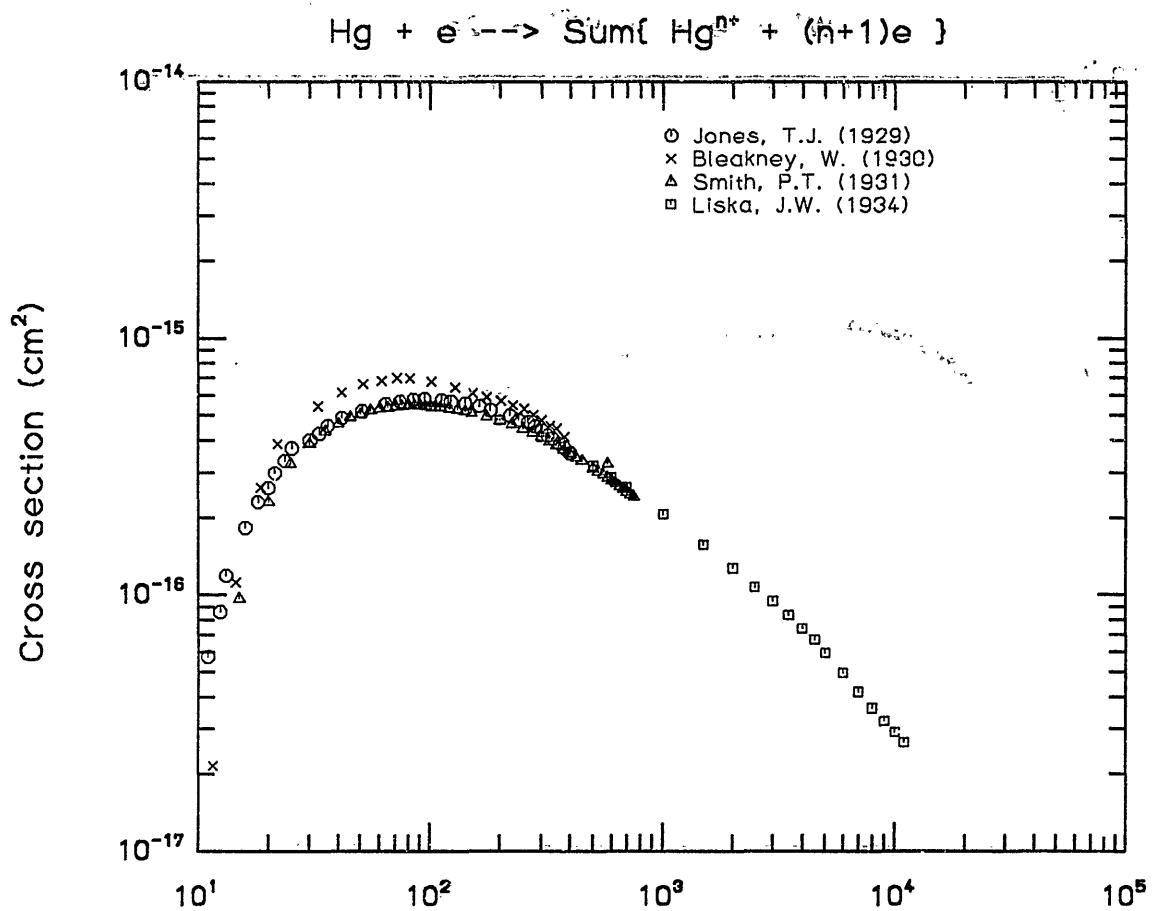
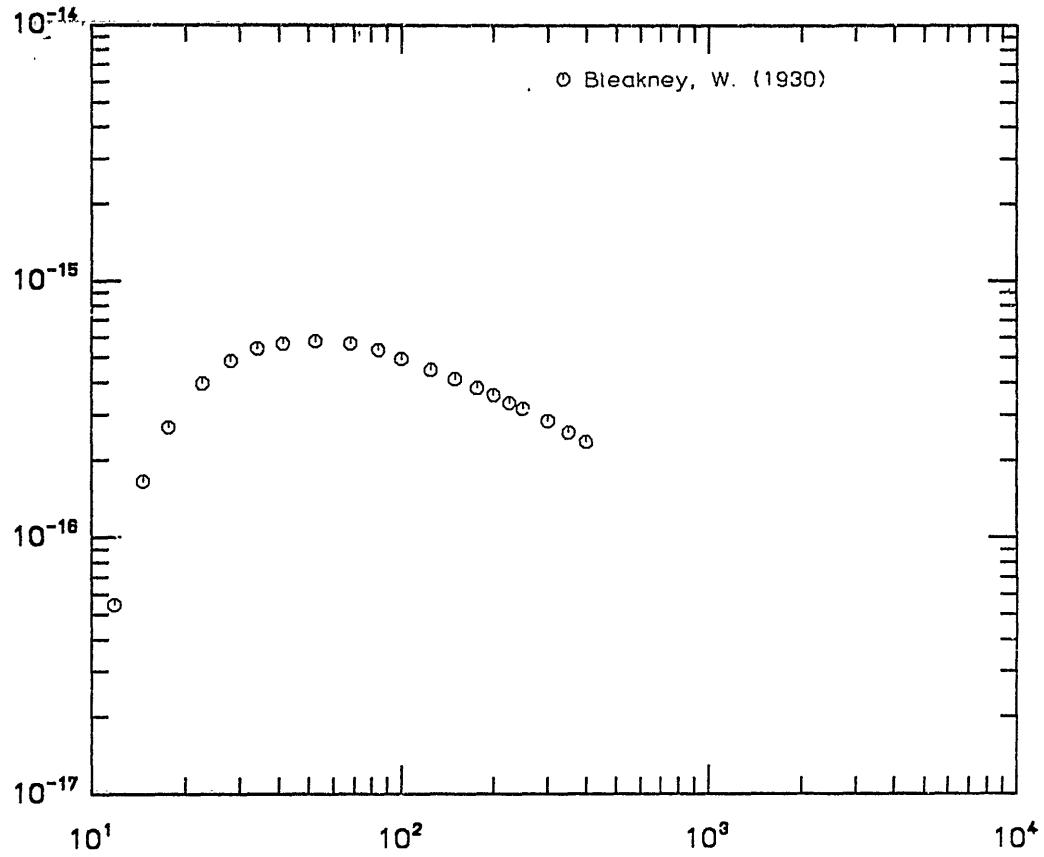


Fig. 272 Electron energy (eV)  $\times 10^{-18}$  vs.  $\sigma$



Electron energy (eV)

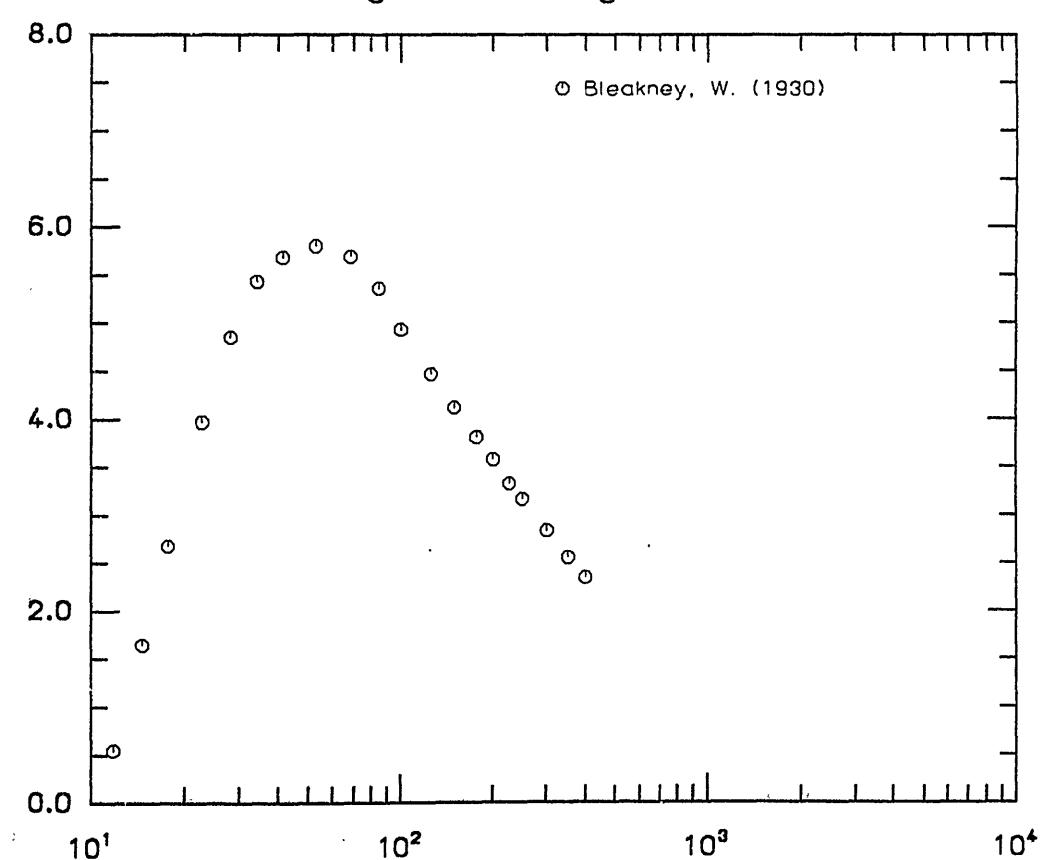
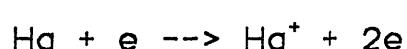


Fig. 273

Electron energy (eV)

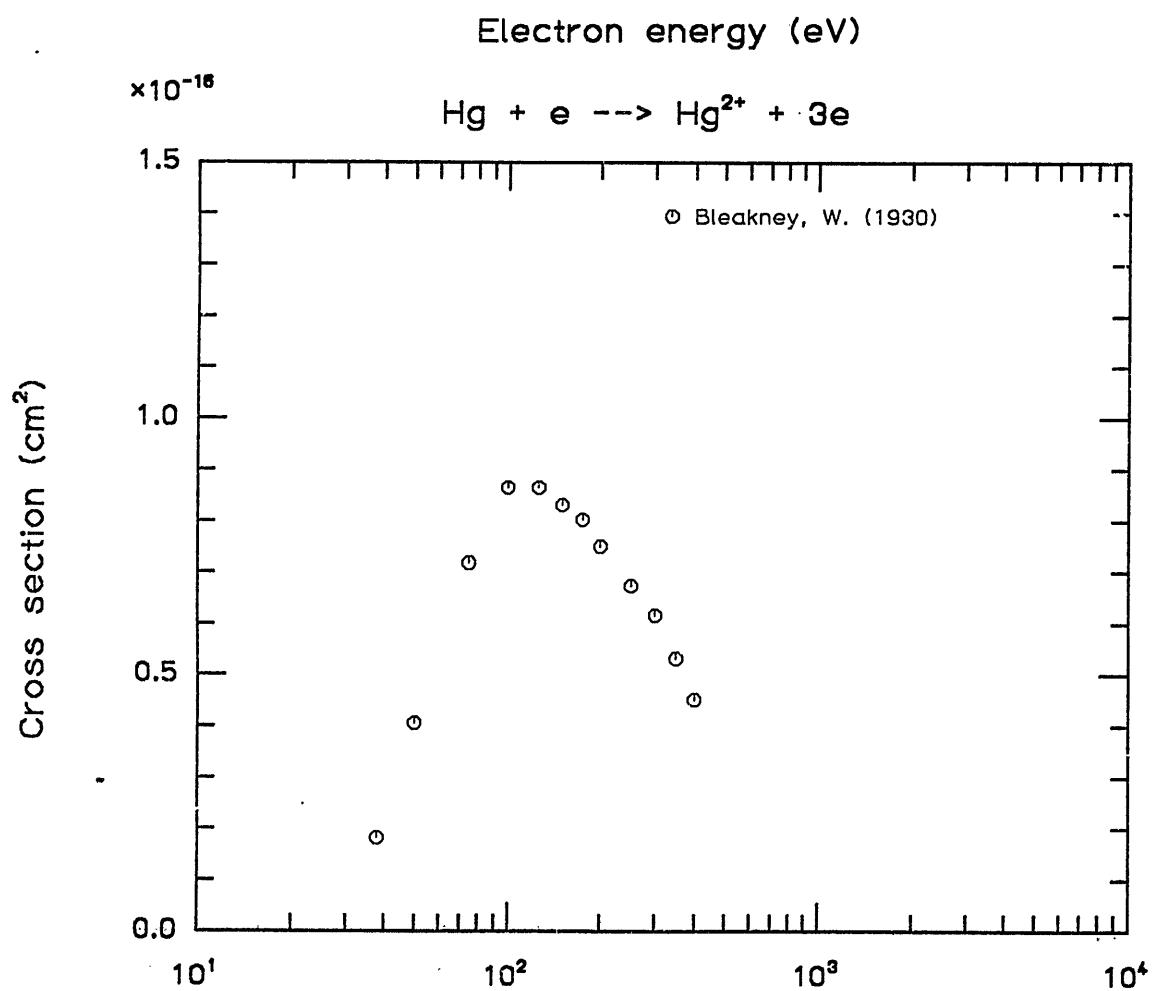
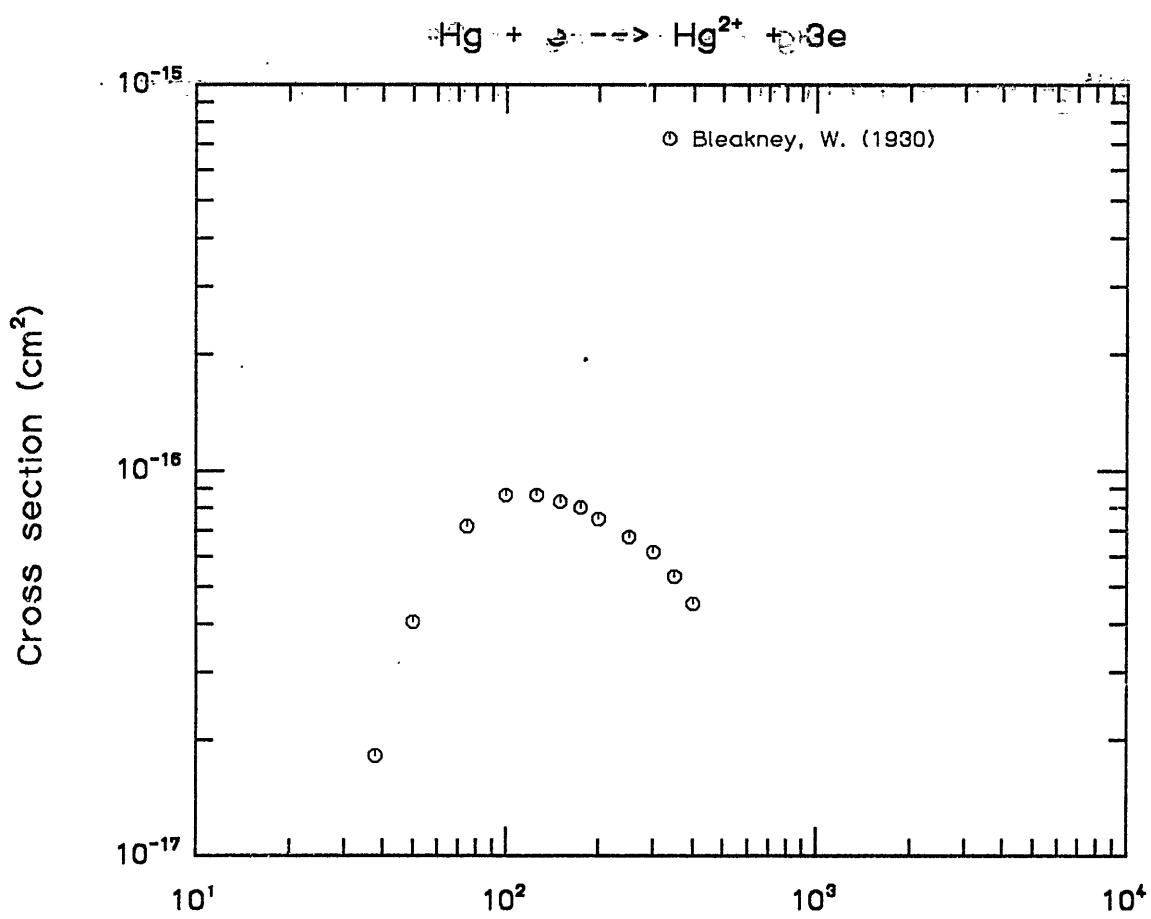
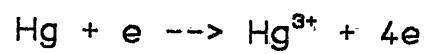
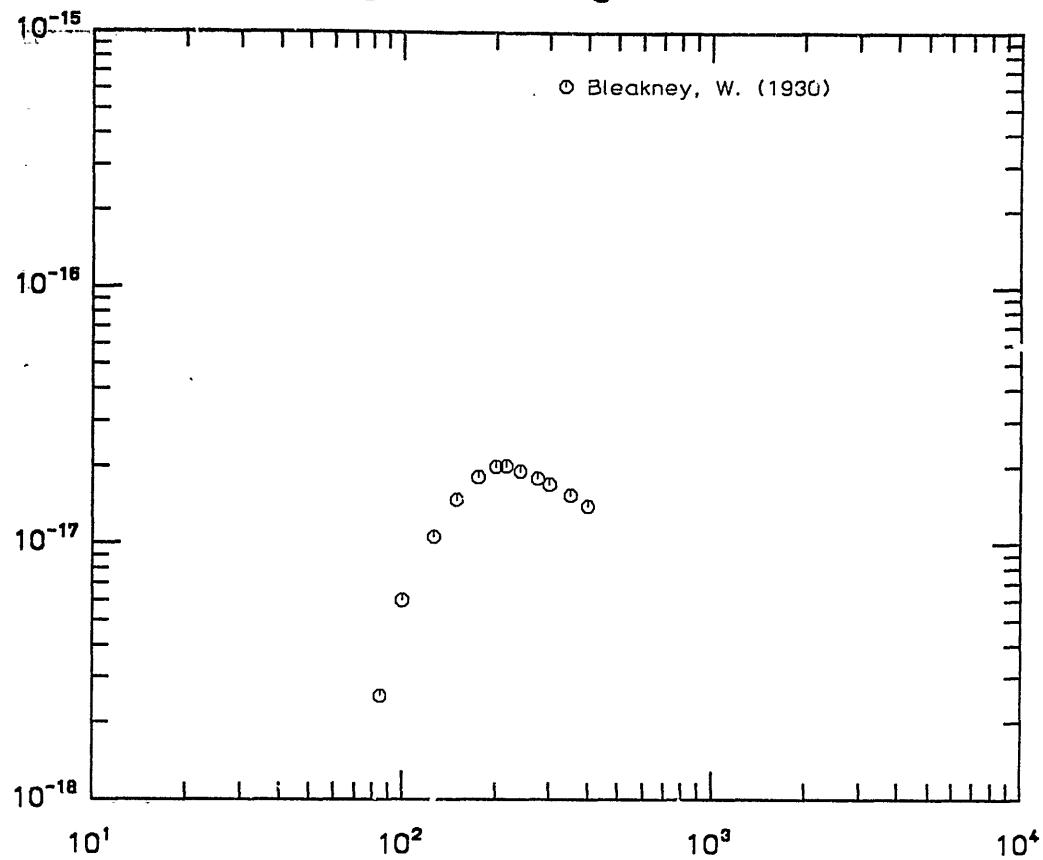


Fig. 274 Electron energy (eV) vs.  $\sigma$



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

Cross section ( $\text{cm}^2$ )

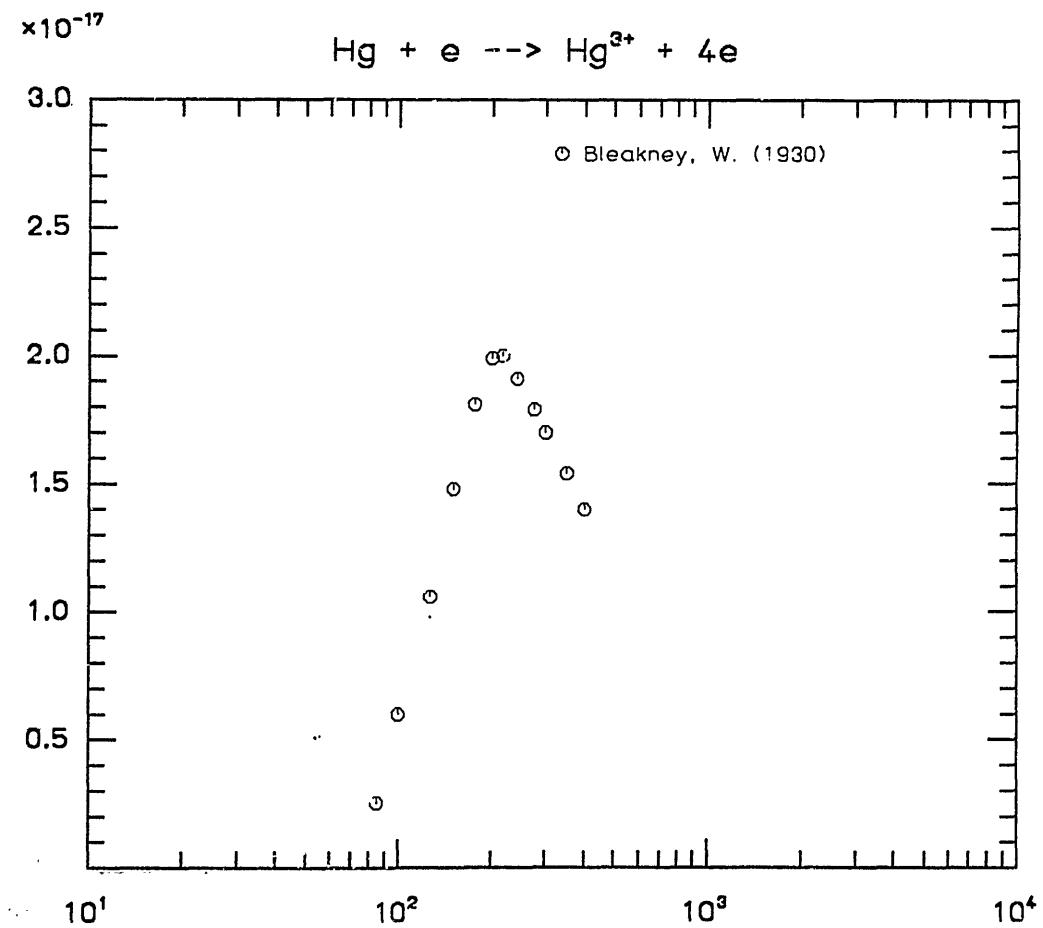


Fig. 275

Electron energy (eV)

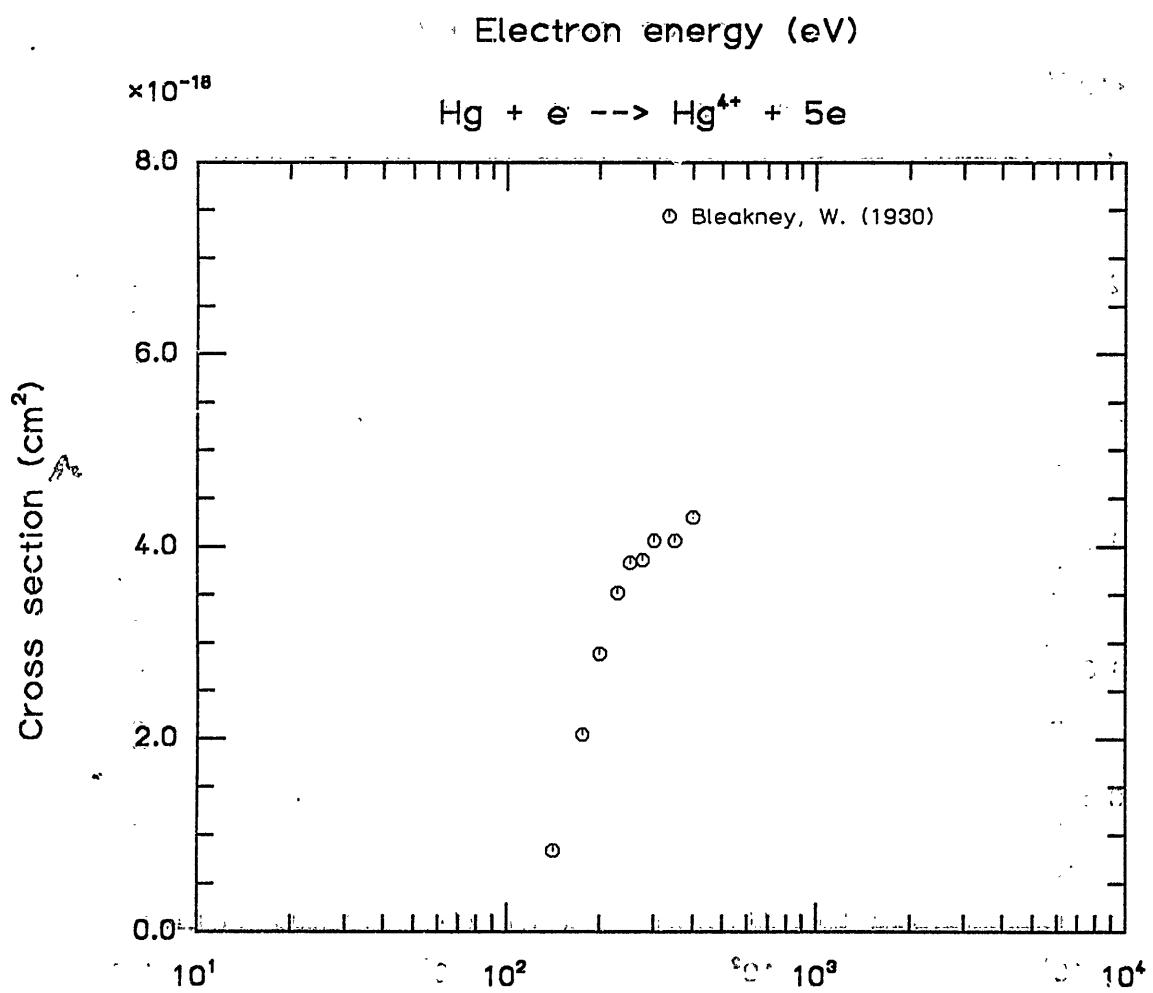
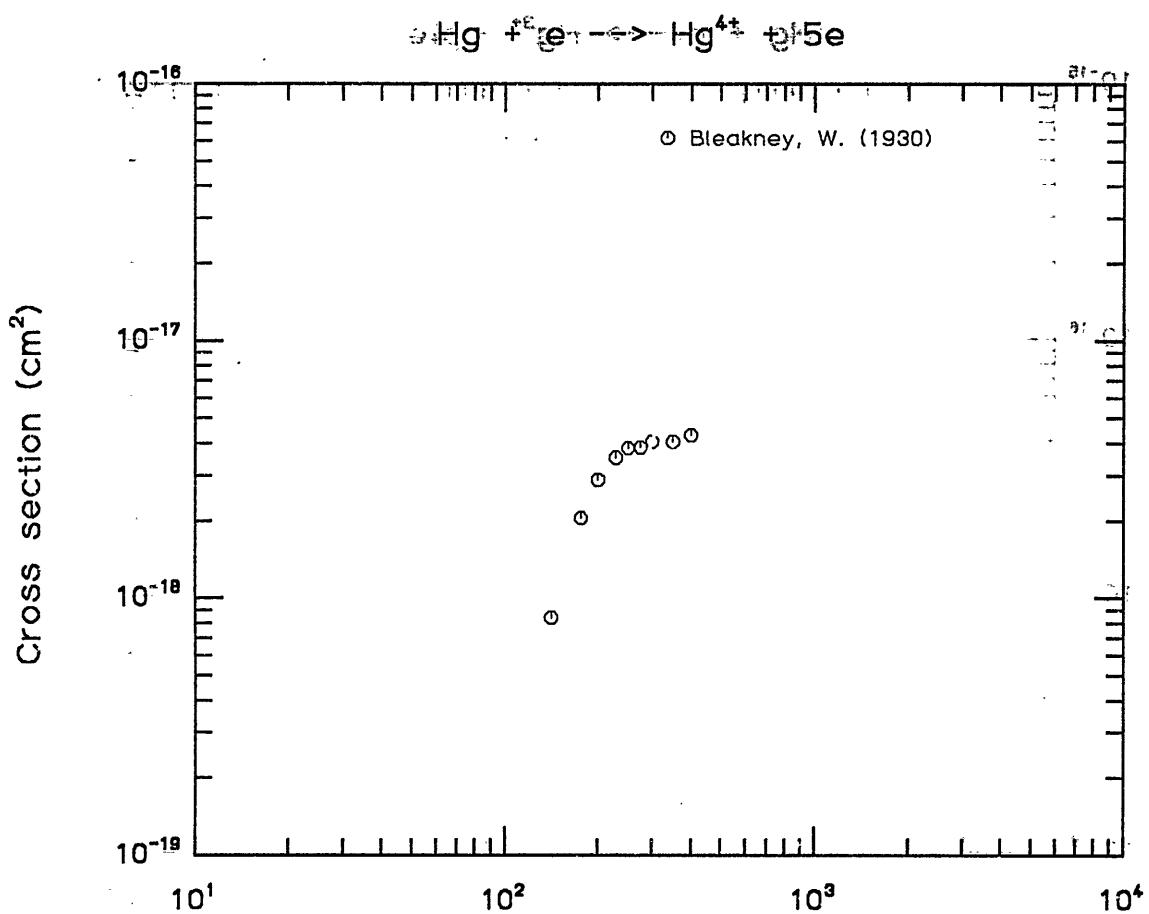
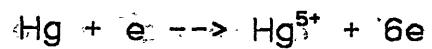
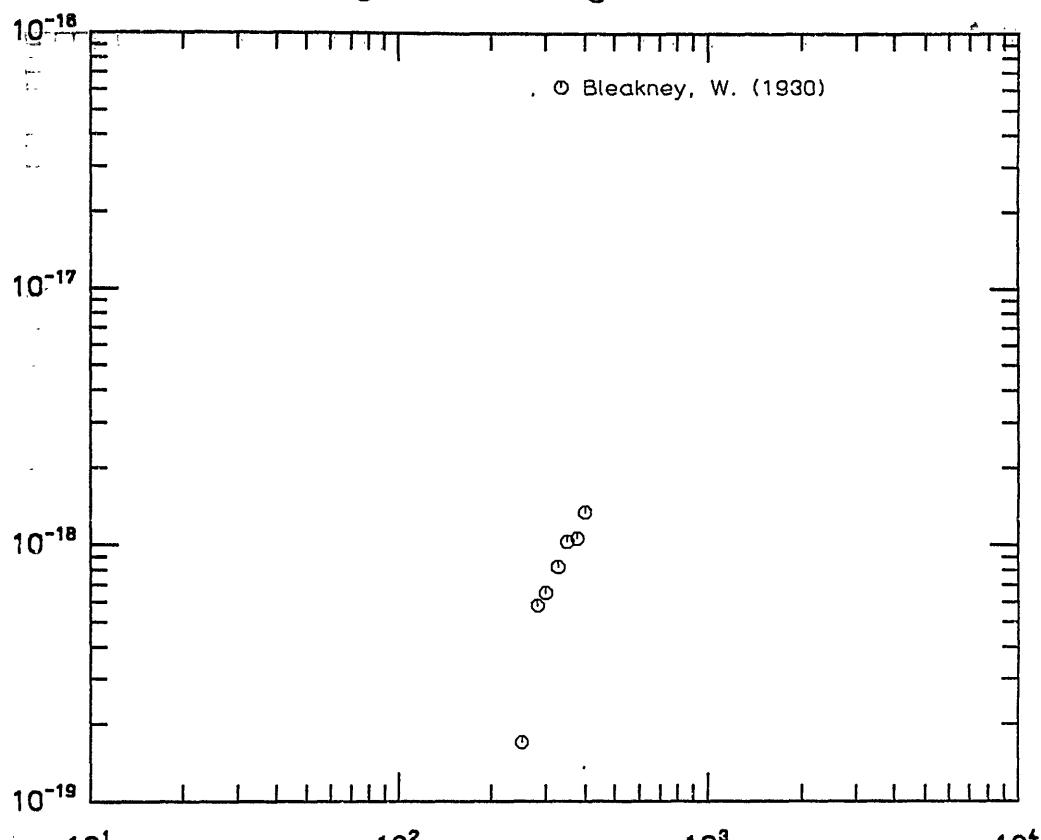


Fig. 276 (v) Electron energy (eV) 276 p17



(b)



(c)

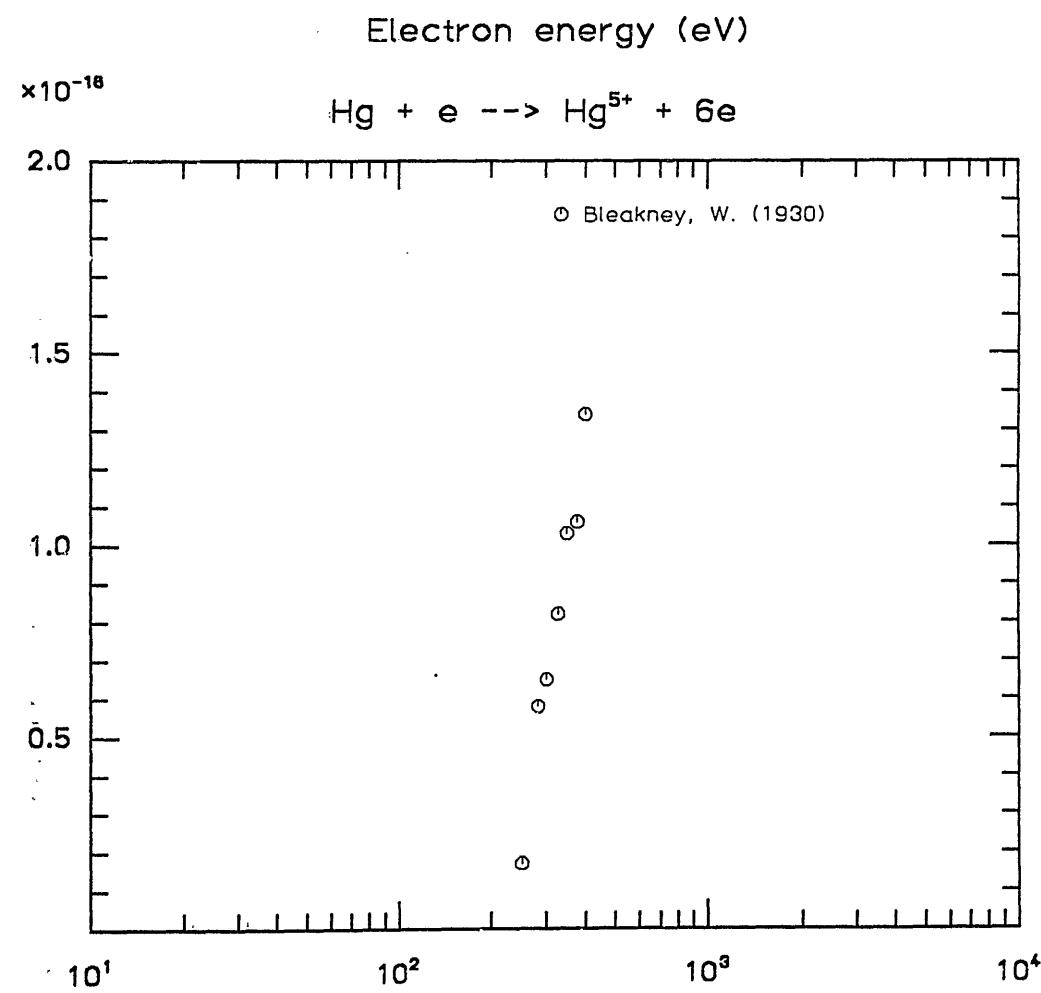


Fig. 277      Electron energy (eV)      (b) (c)

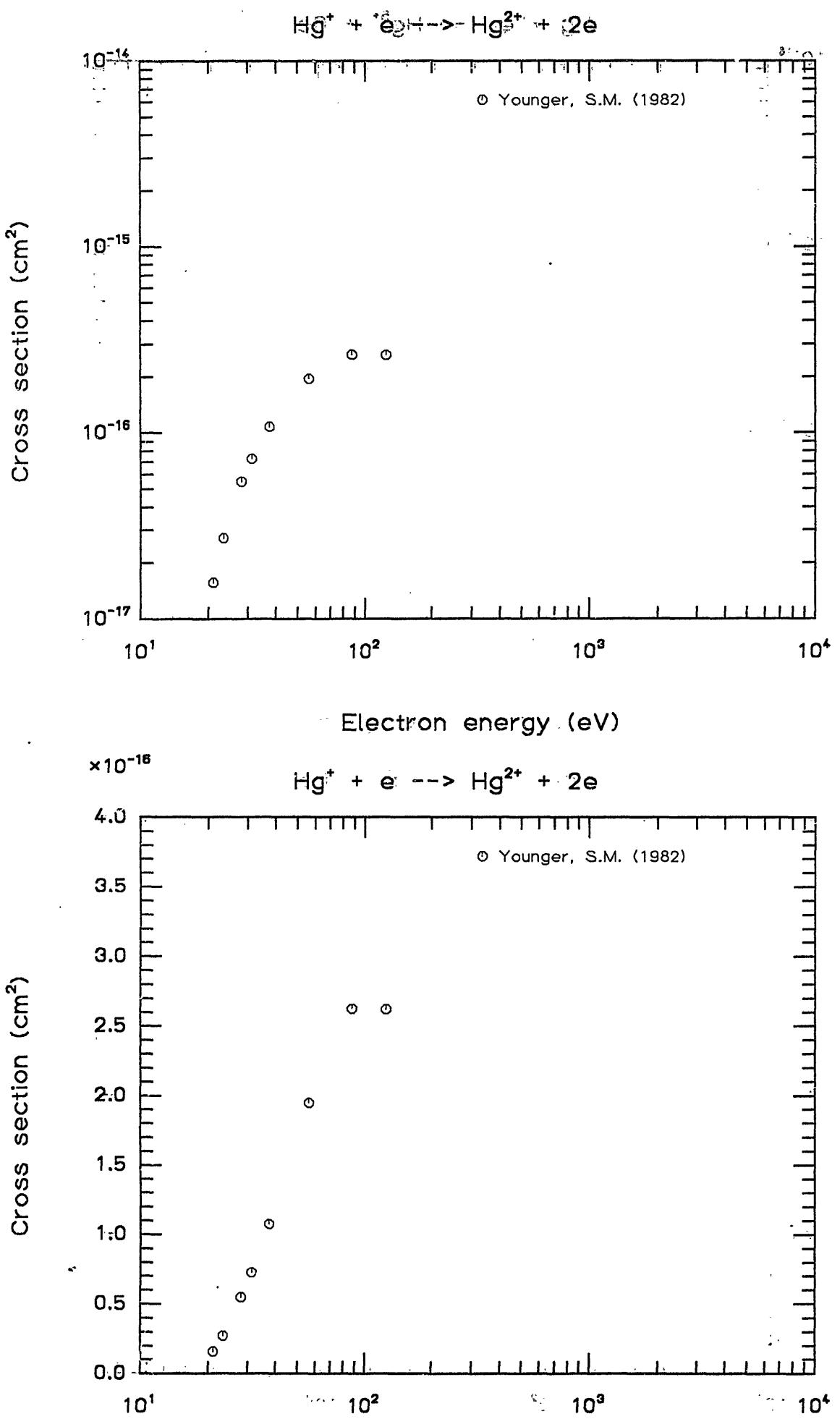
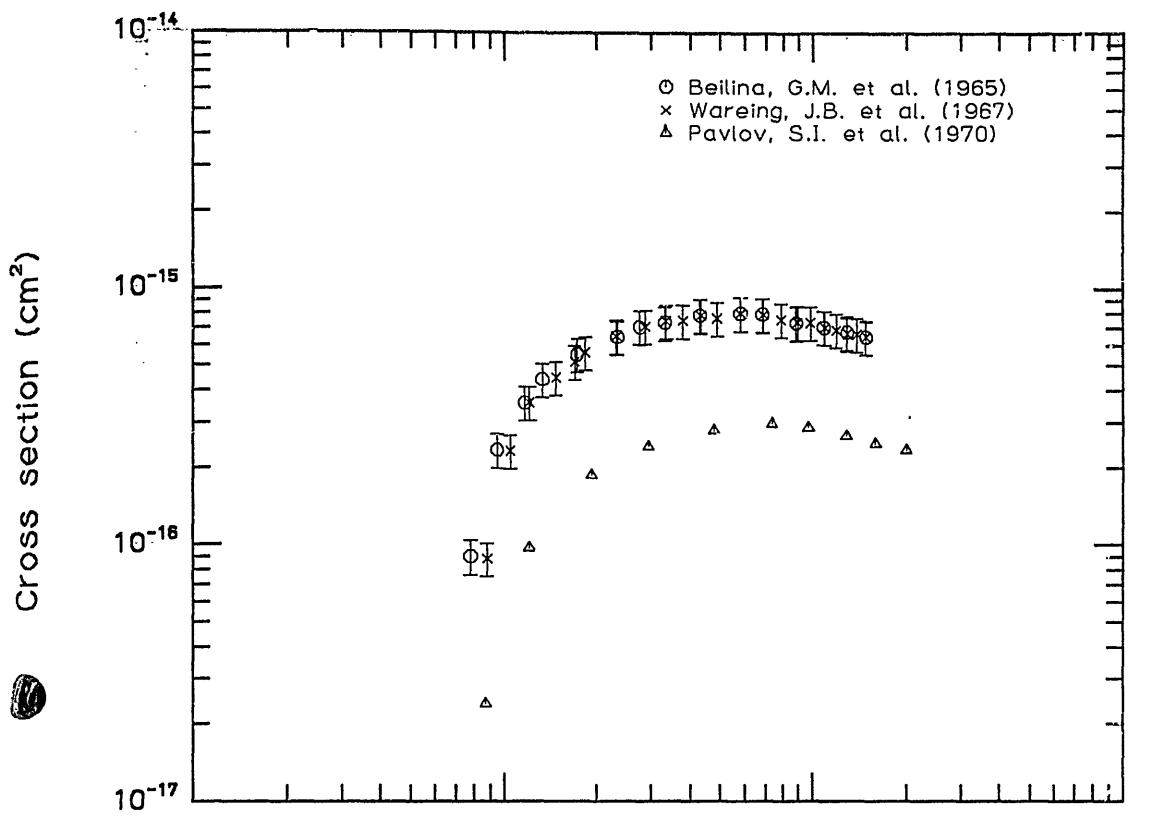
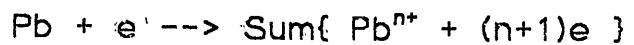


Fig. 278 (vs Electron energy (eV)) 772.gif



Electron energy (eV)

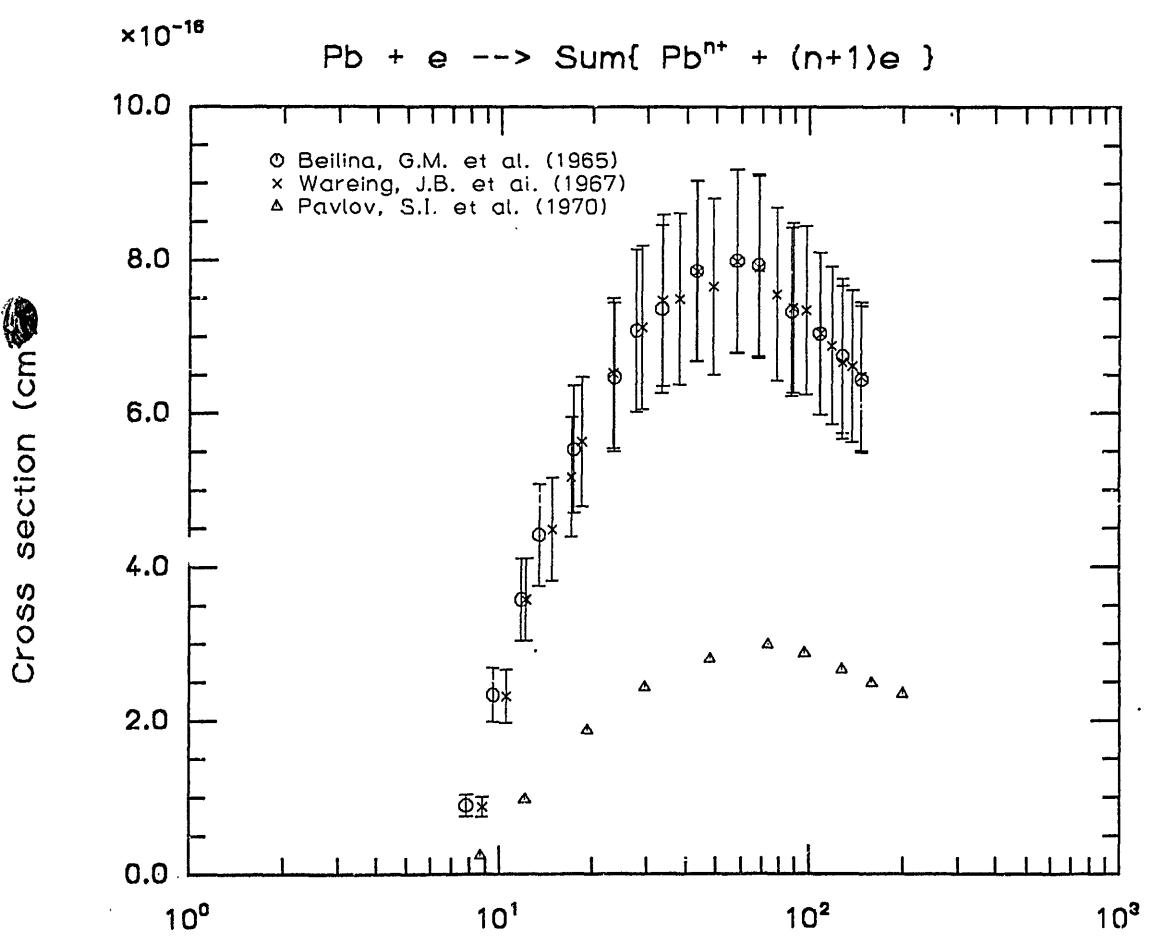


Fig. 279

Electron energy (eV)

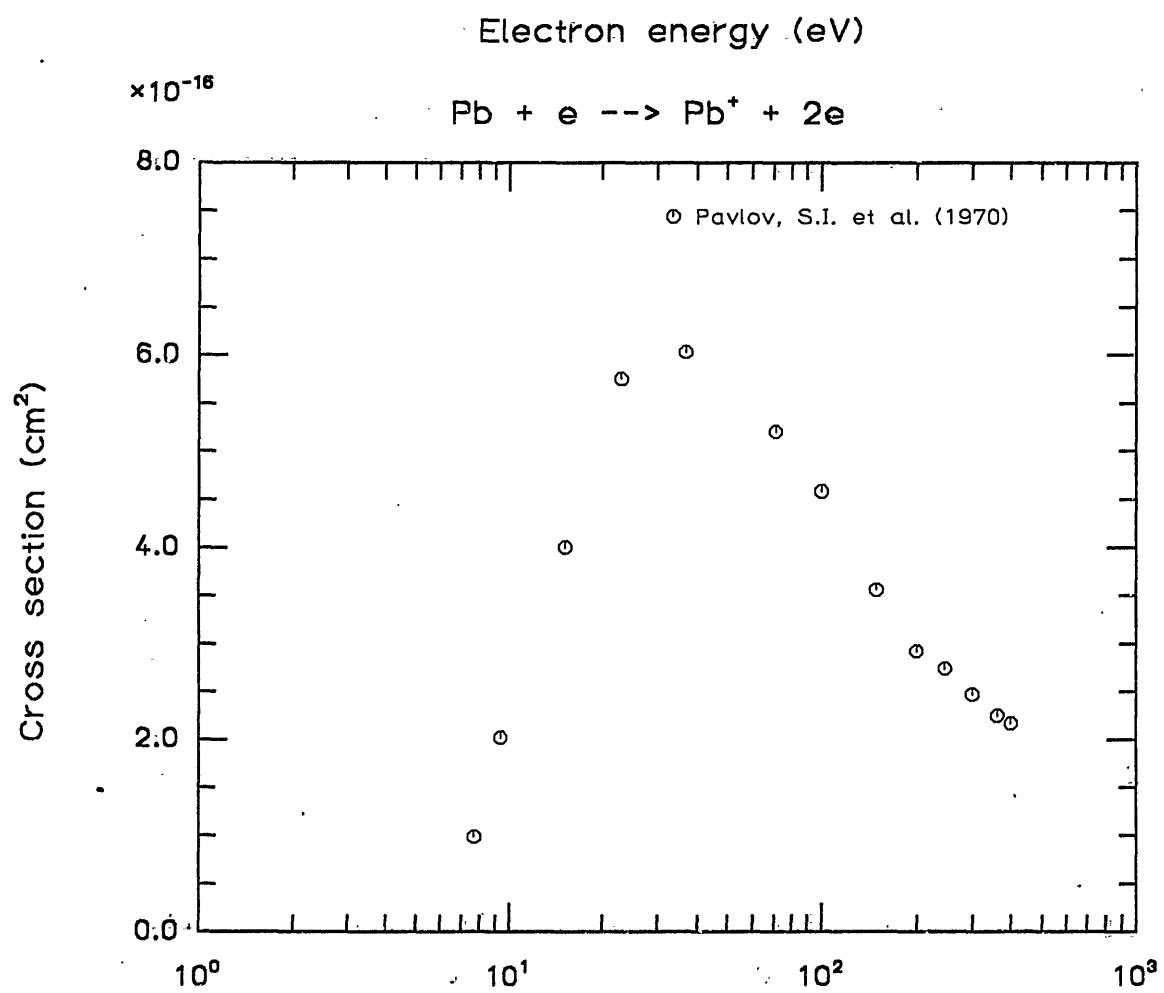
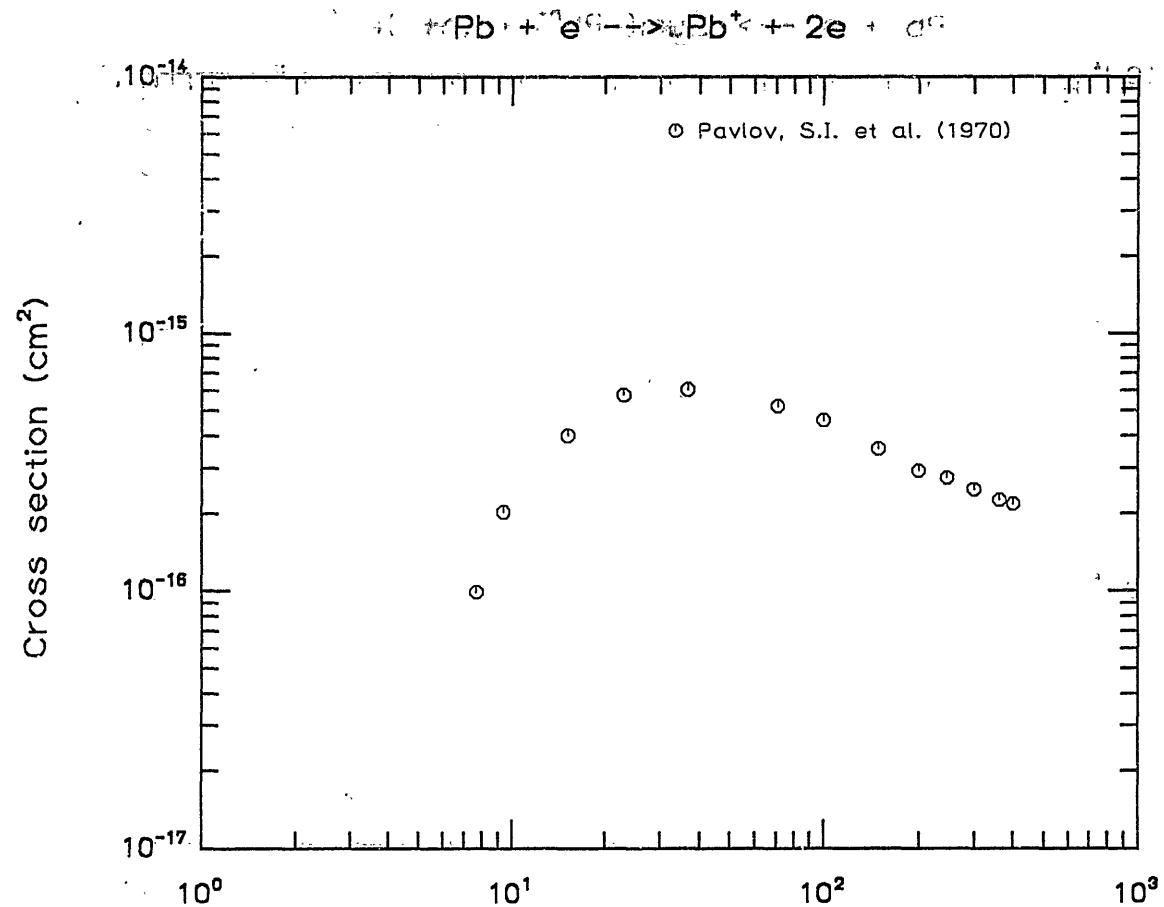
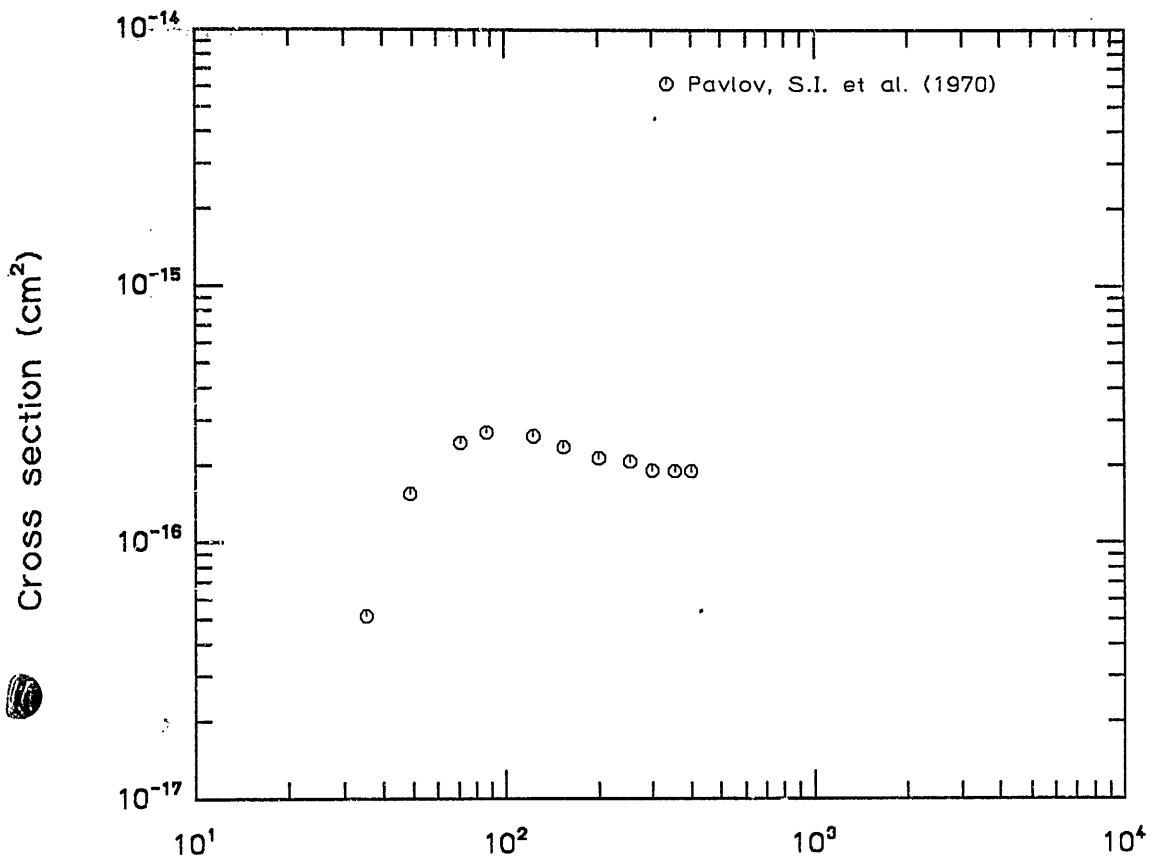


Fig. 280 Electron energy (eV) ESR p.11



Electron energy (eV)

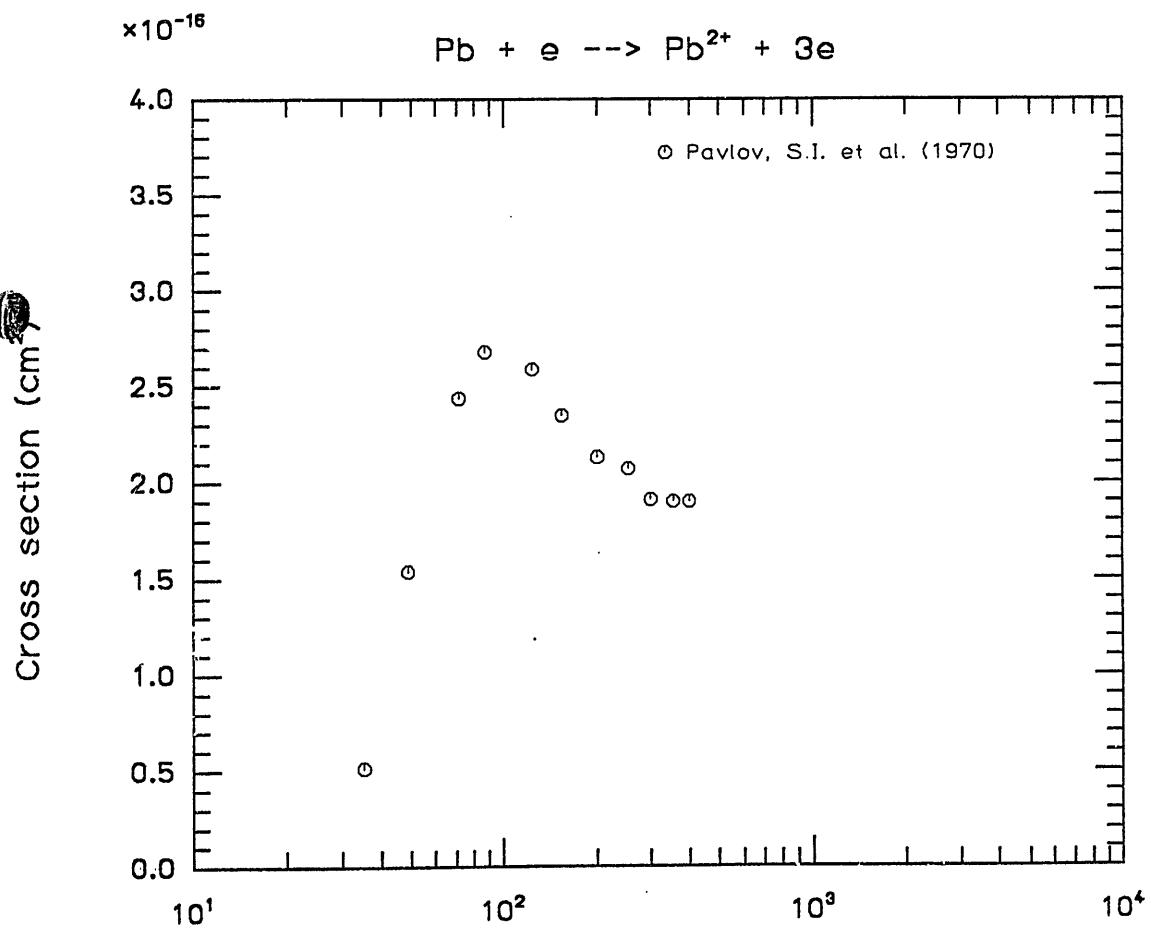


Fig. 281

Electron energy (eV)

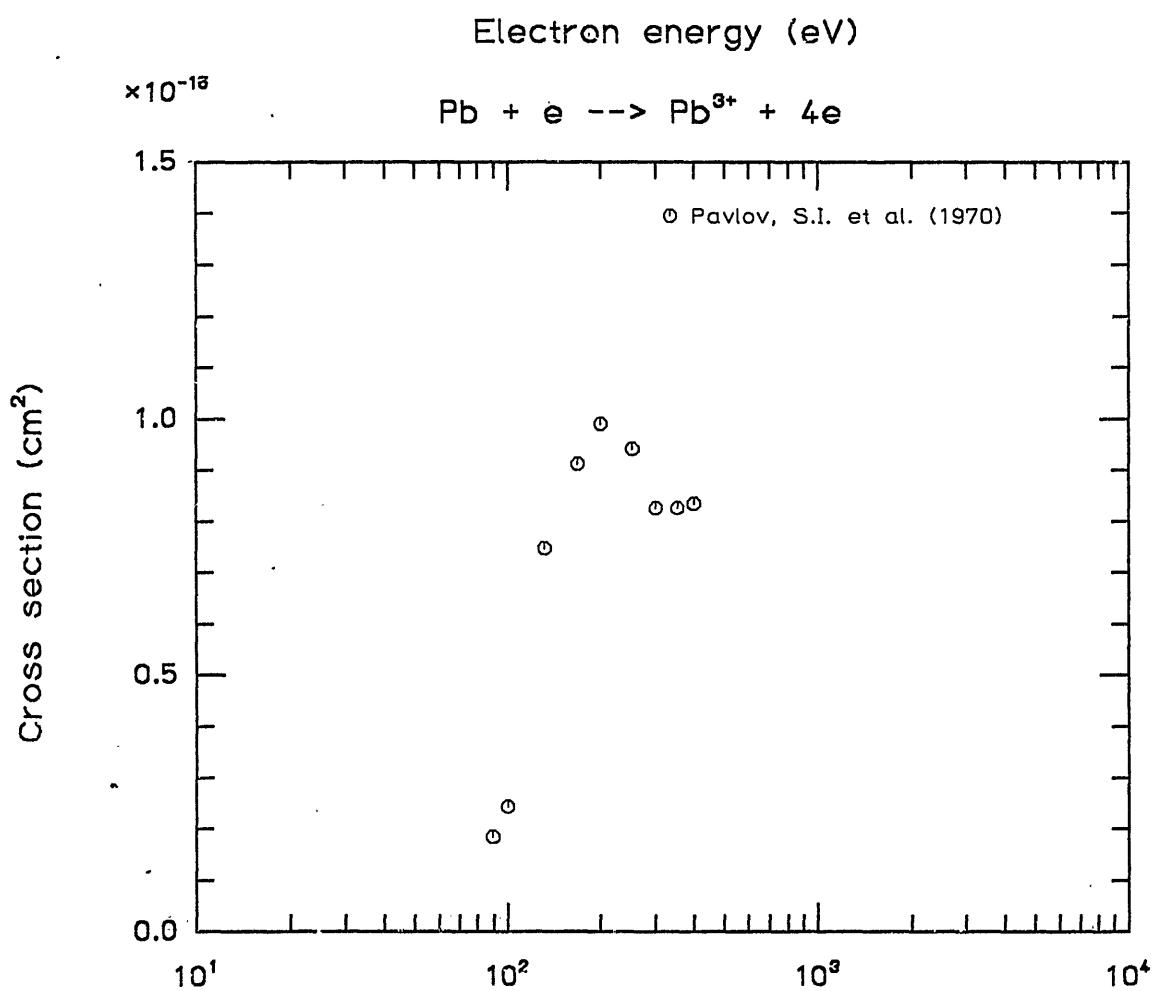
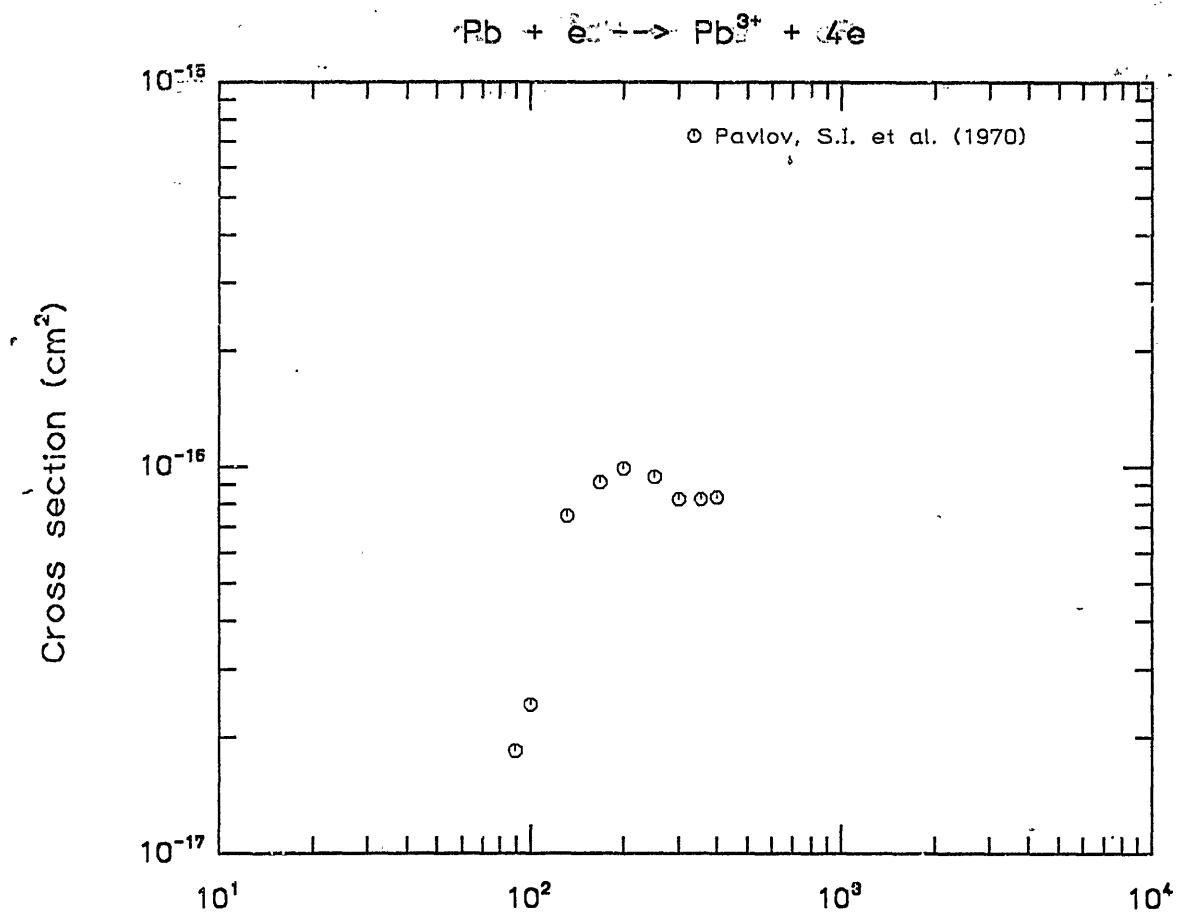
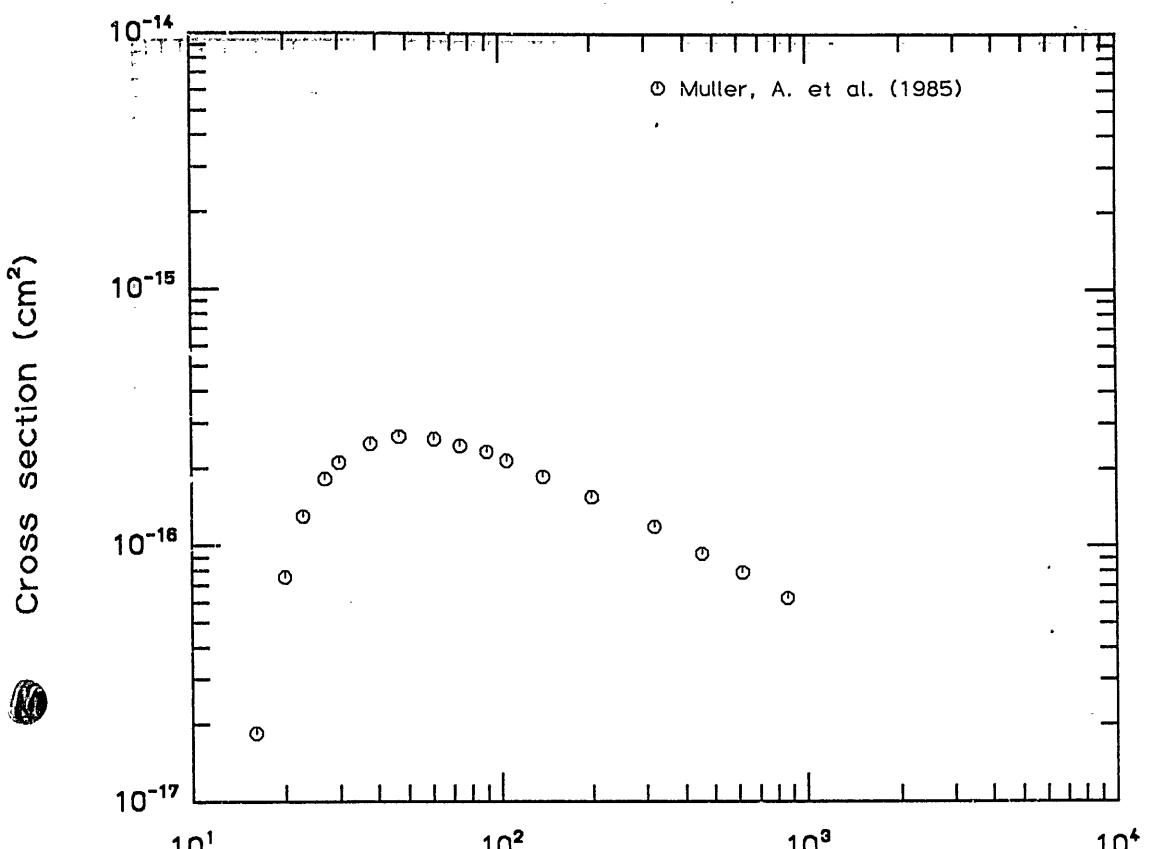
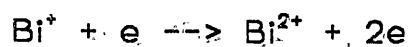


Fig. 282      Electron energy (eV)      [282]



Electron energy (eV)

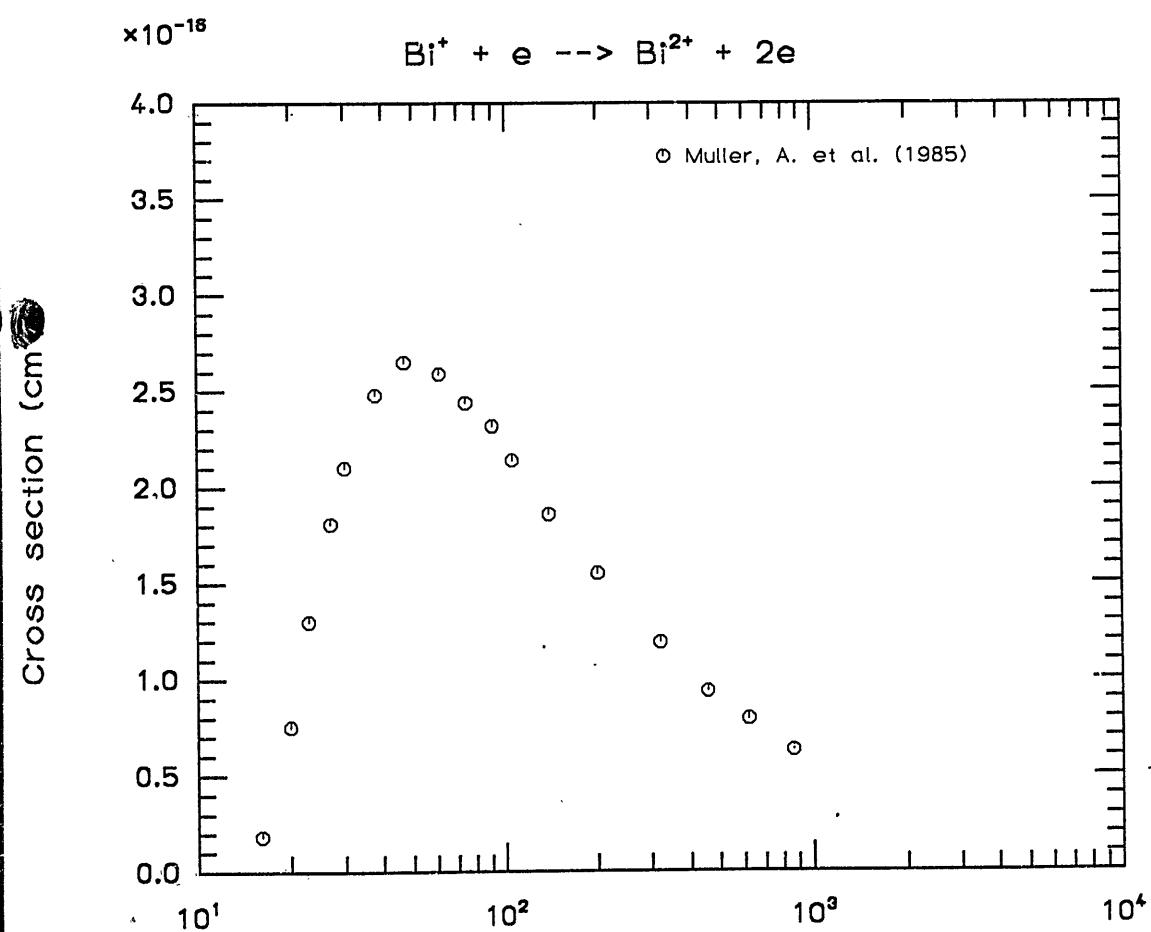


Fig. 283

Electron energy (eV)

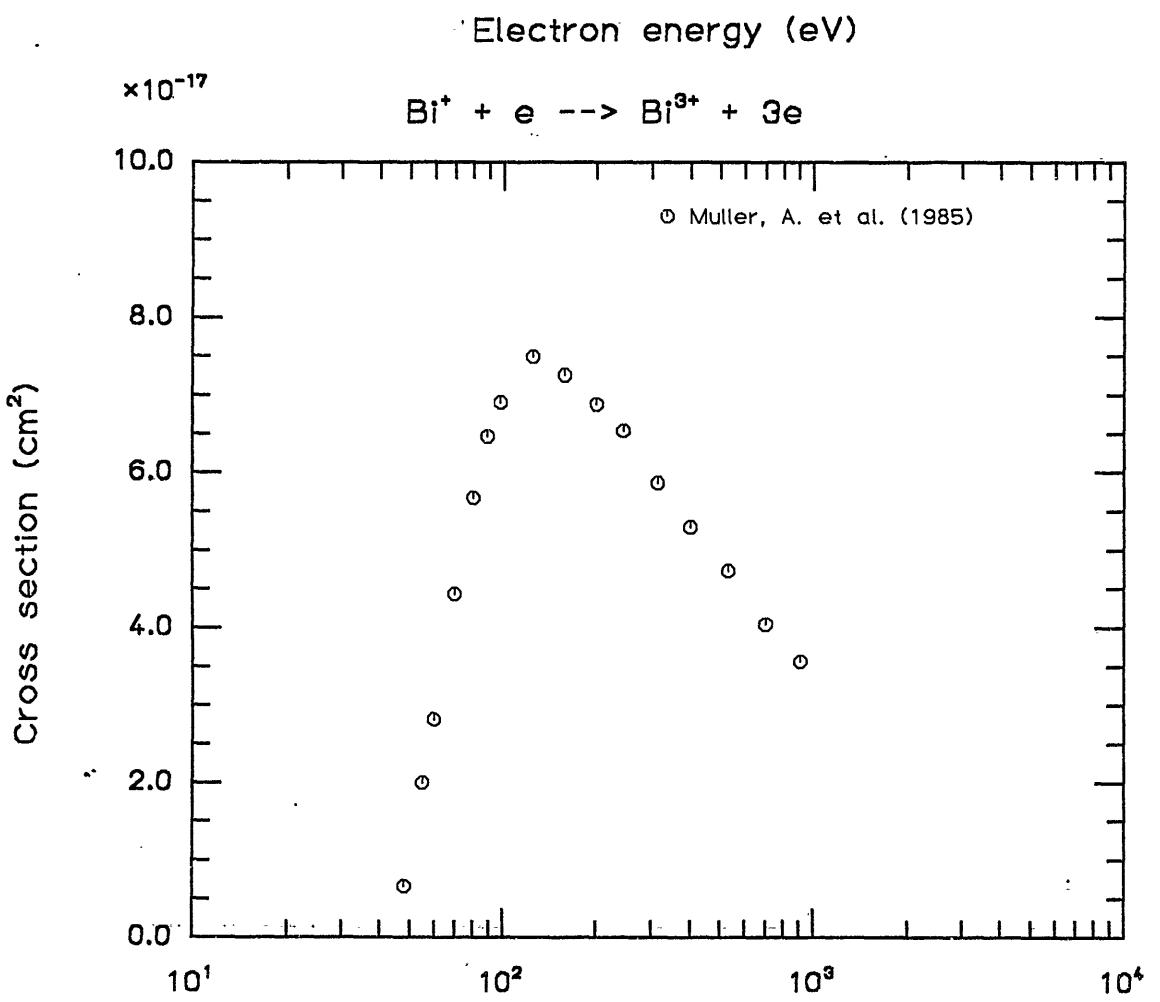
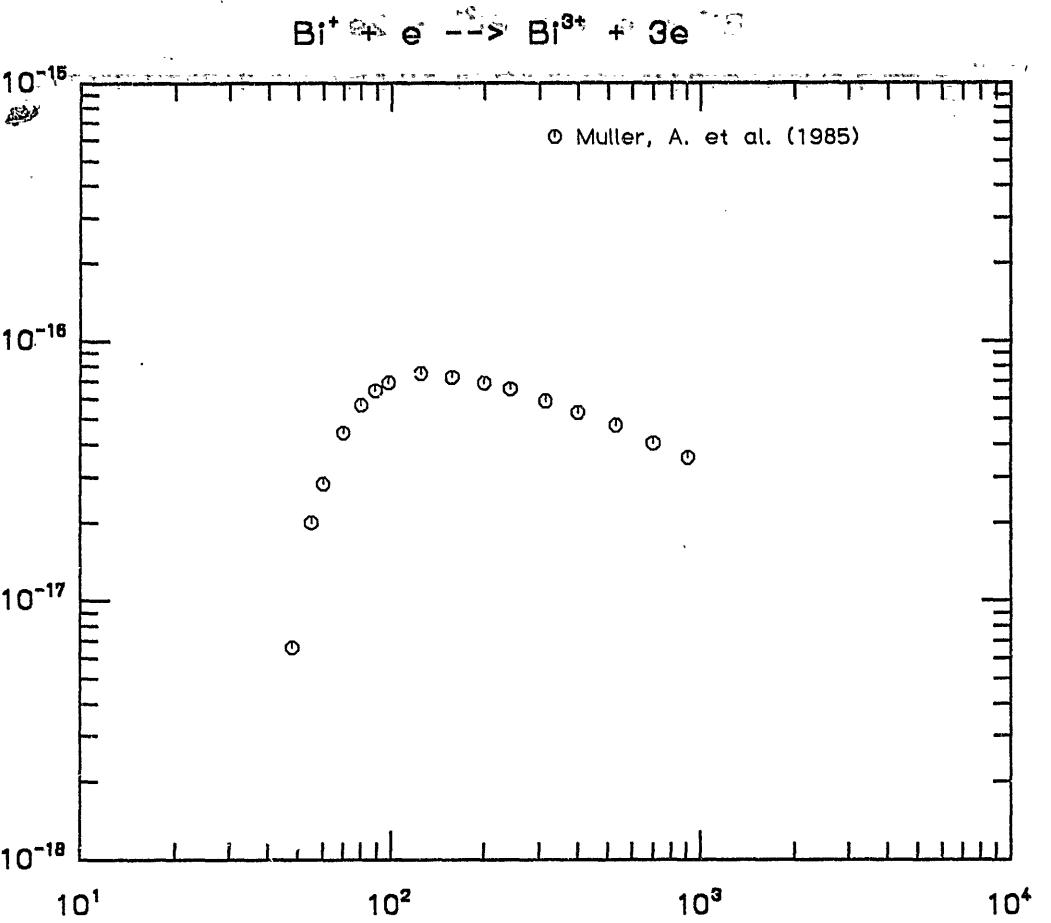
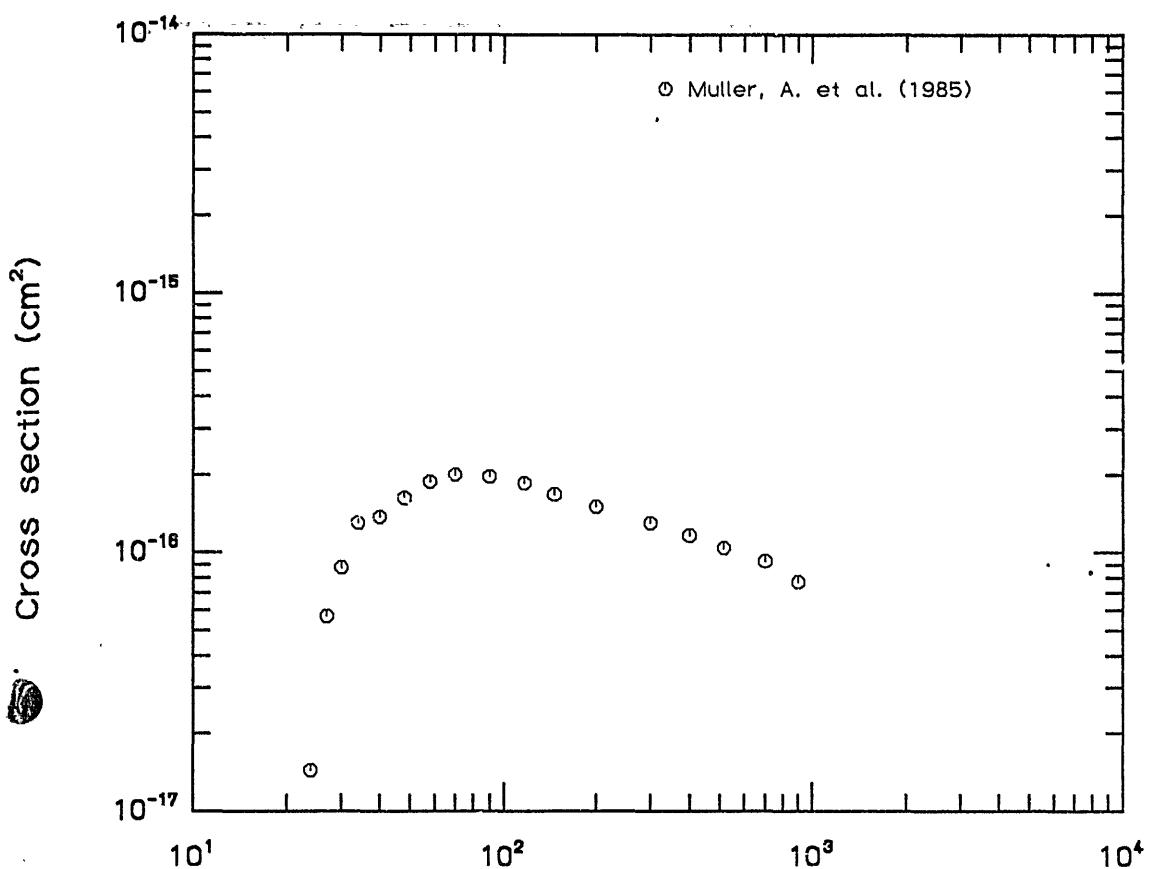
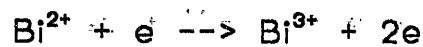


Fig. 284      Electron energy (eV)      284



Electron energy (eV)

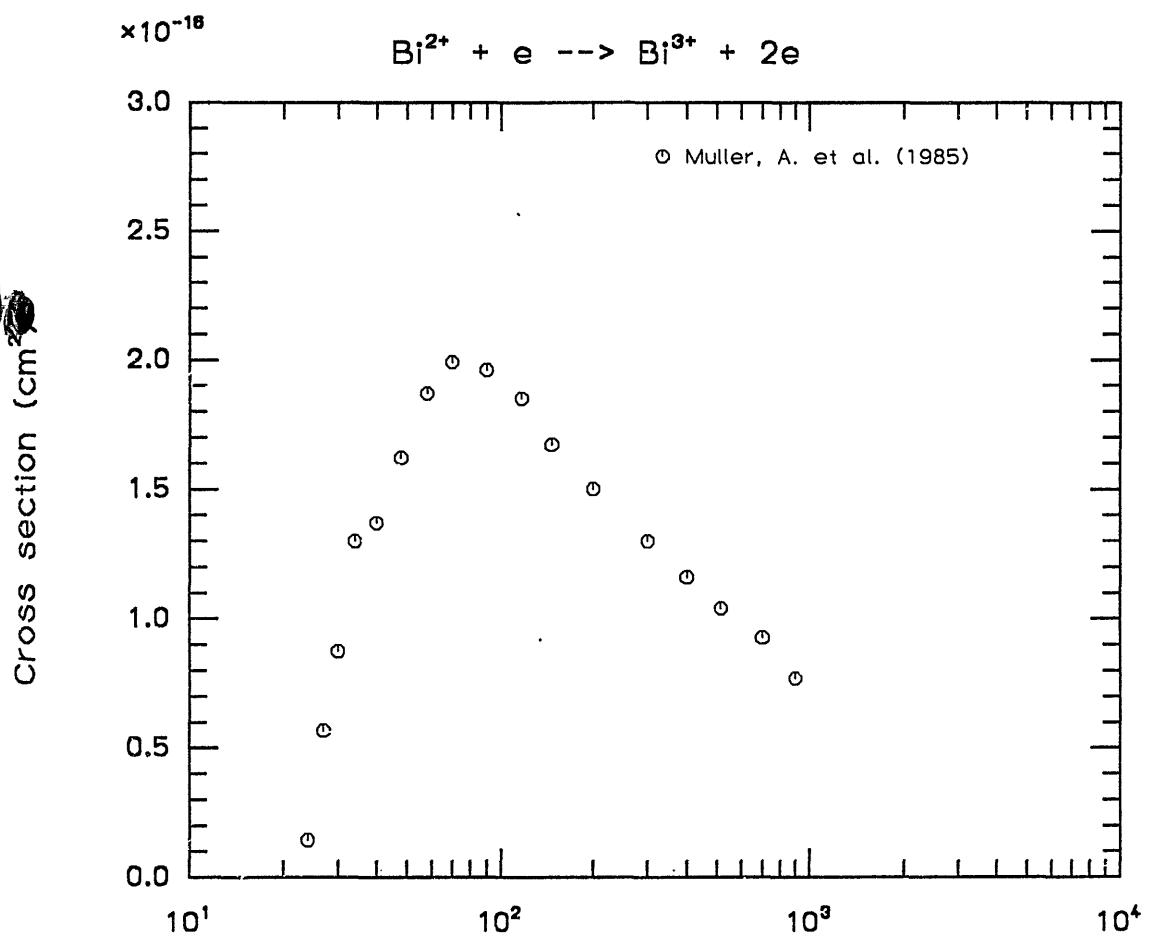


Fig. 285      Electron energy (eV)

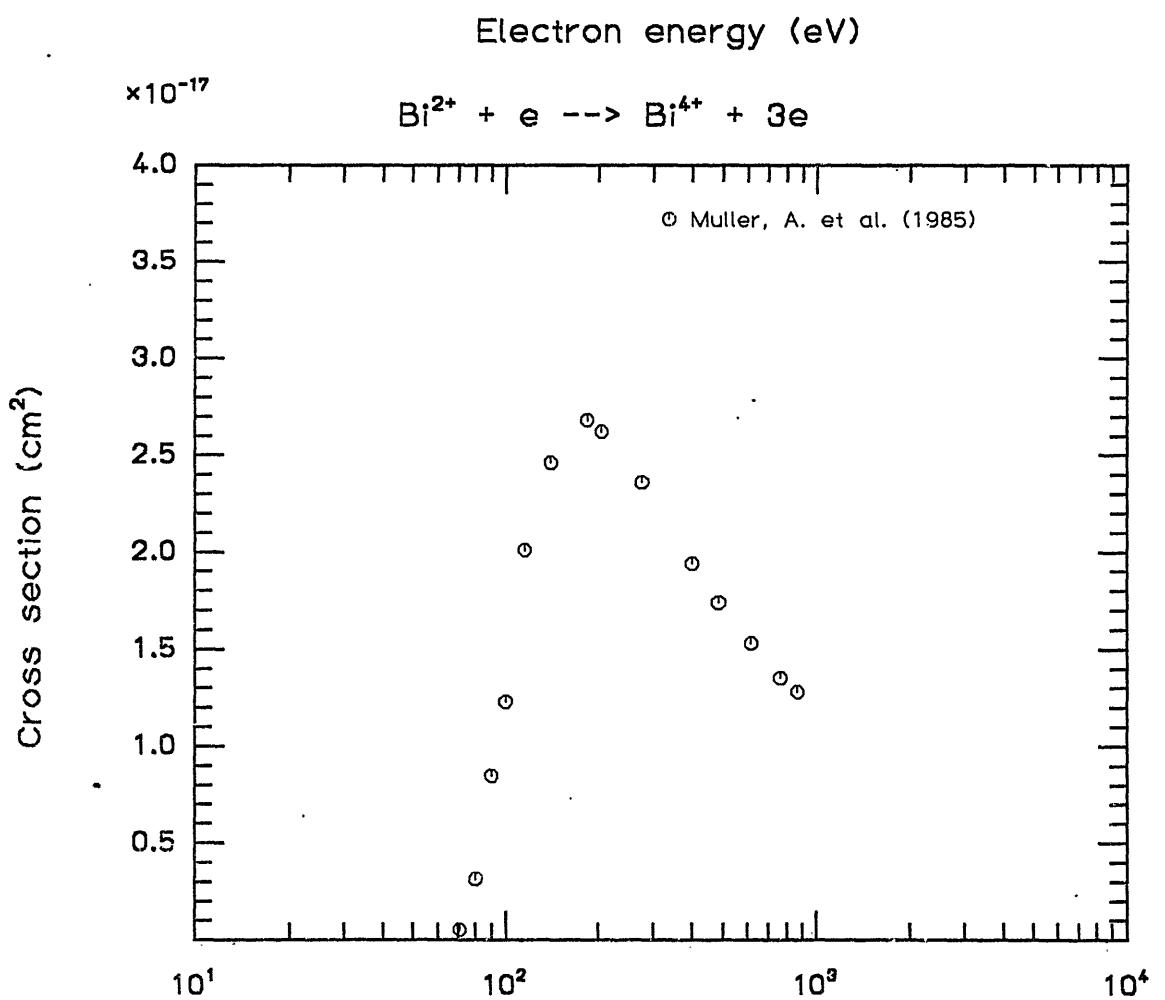
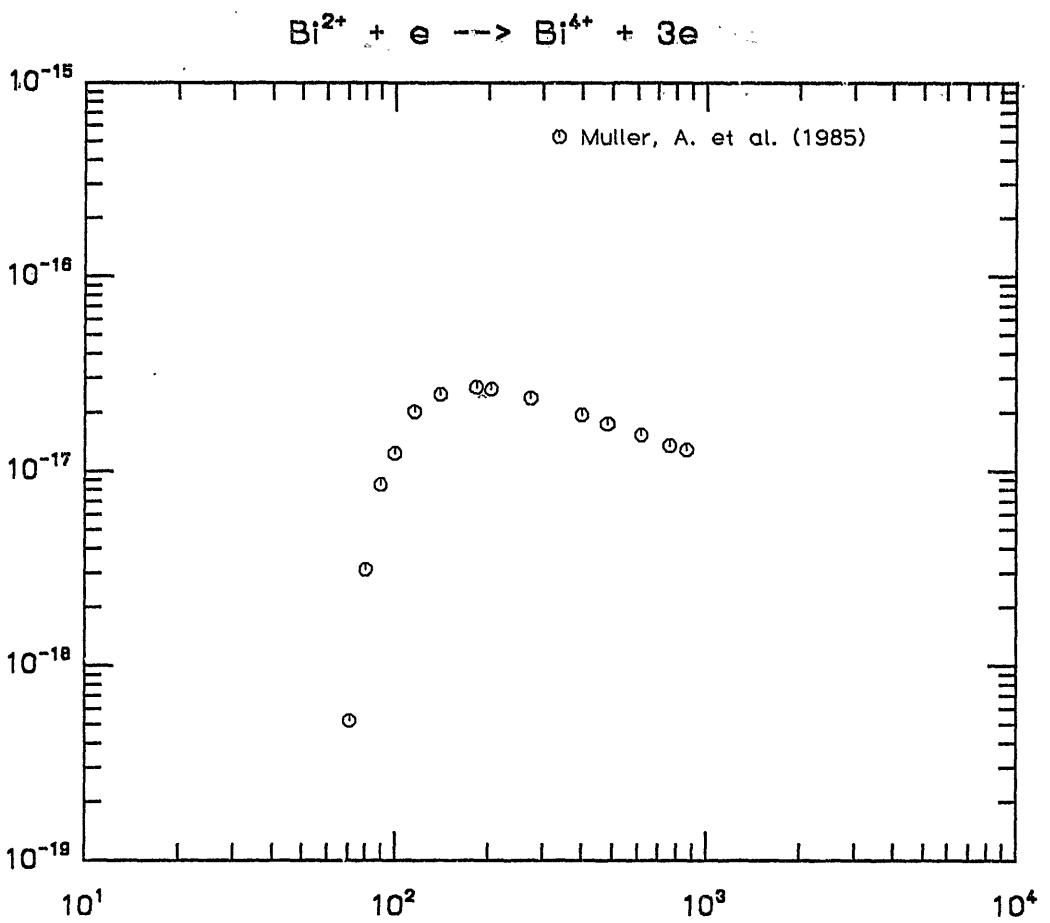


Fig. 286      Electron energy (eV)

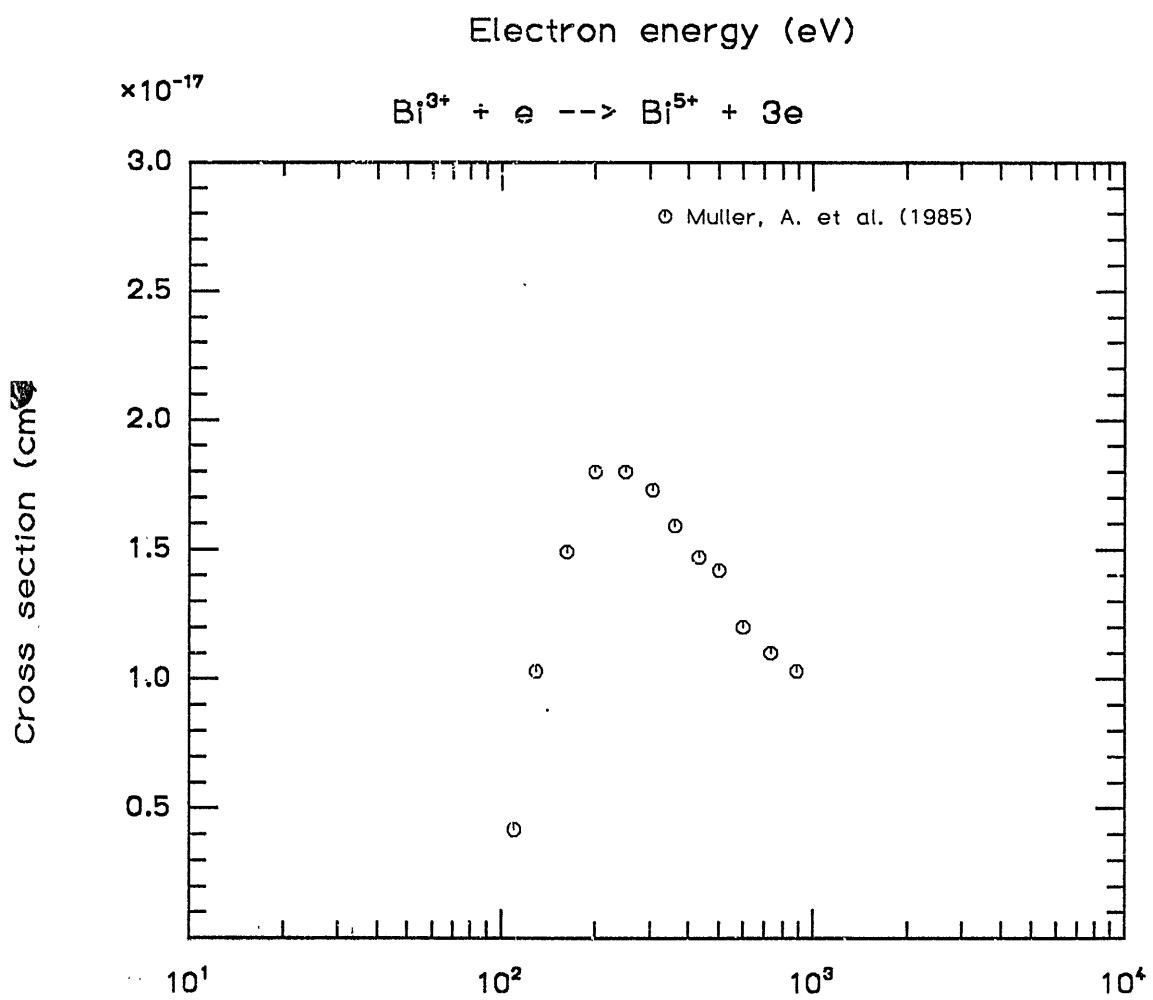
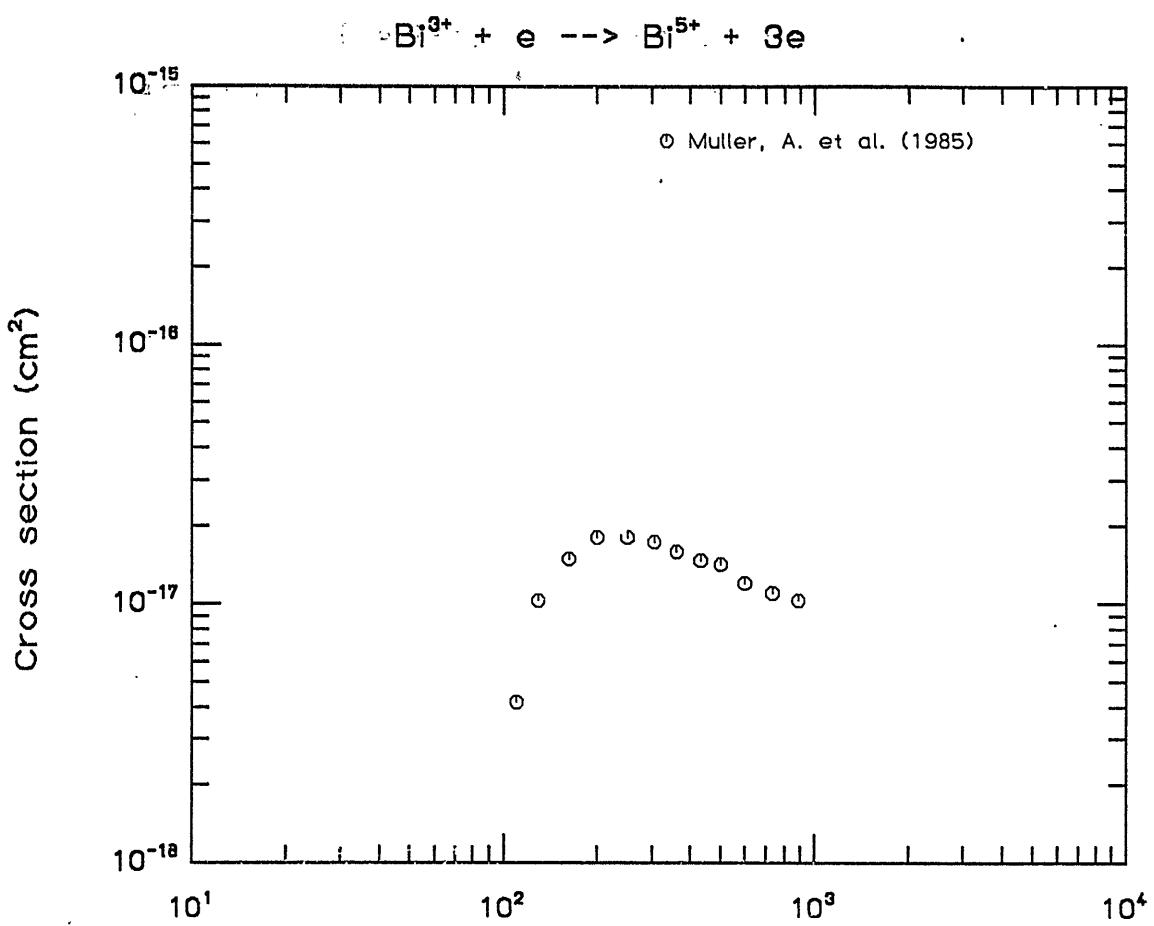


Fig. 287      Electron energy (eV)

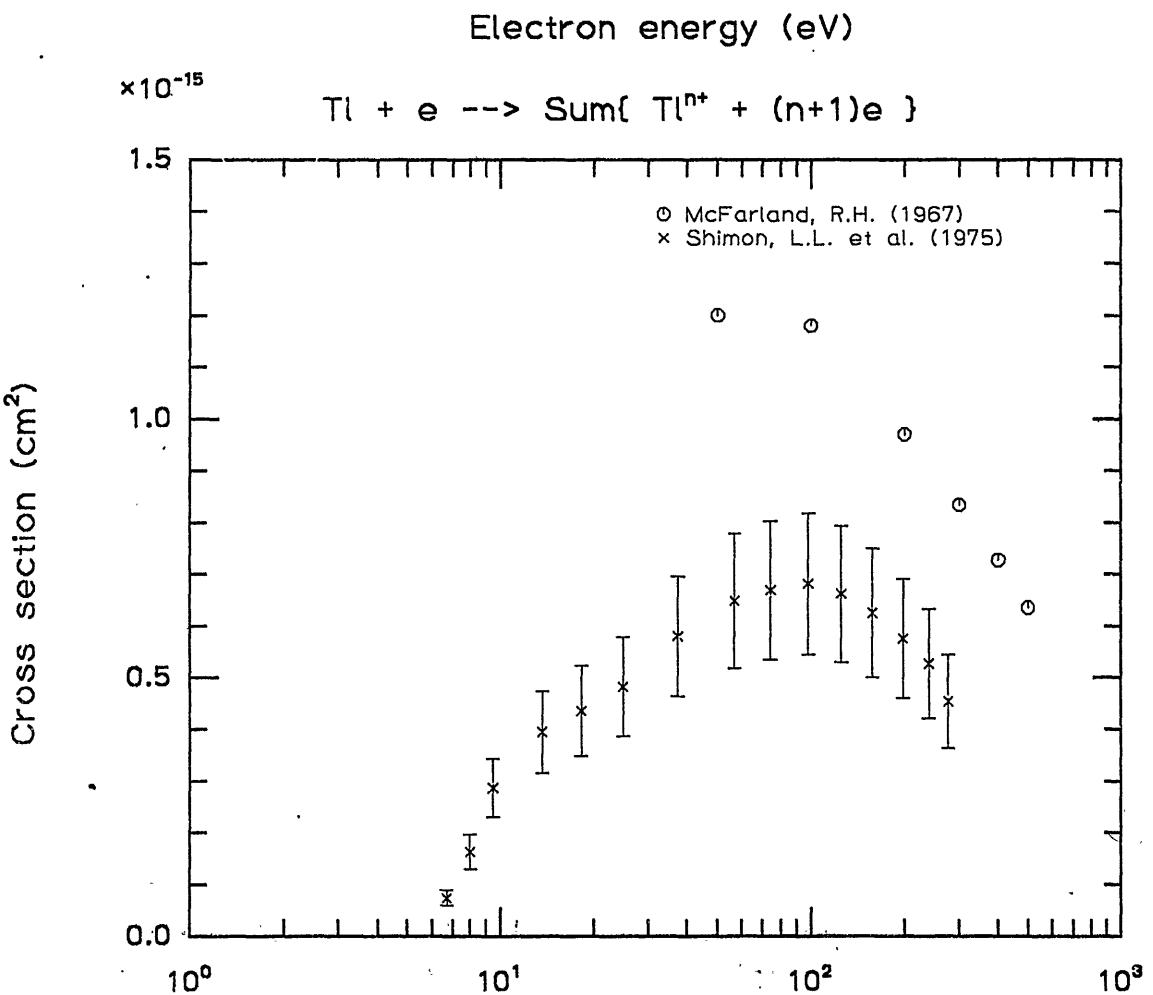
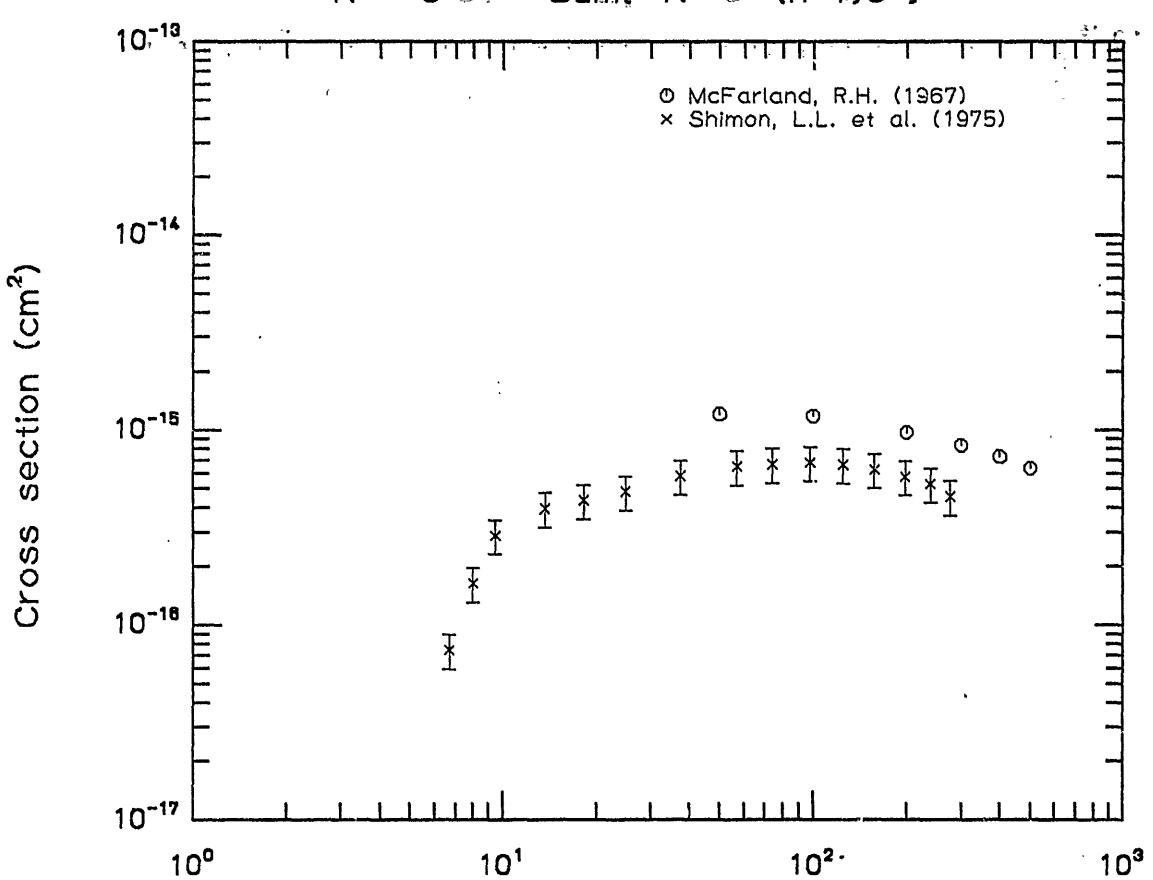
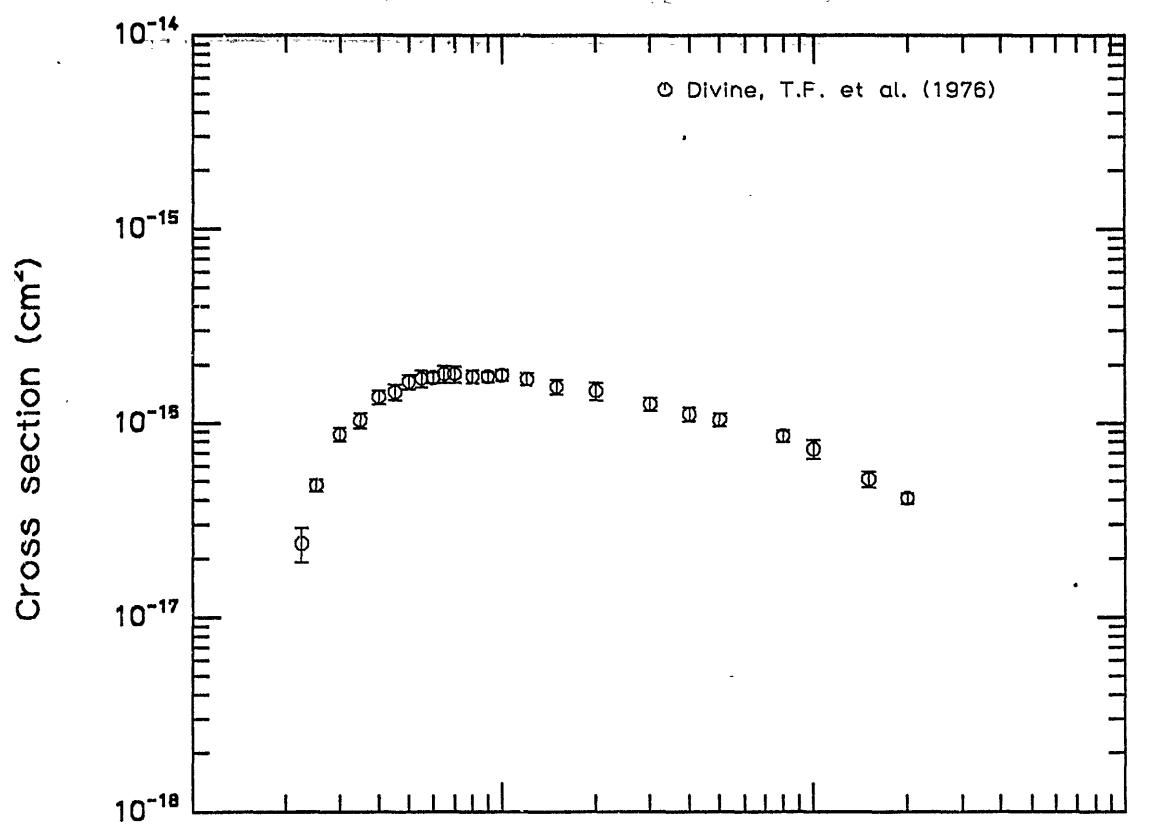
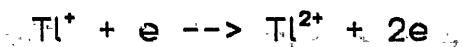


Fig. 288

Electron energy (eV)

188 pH



Electron energy (eV)

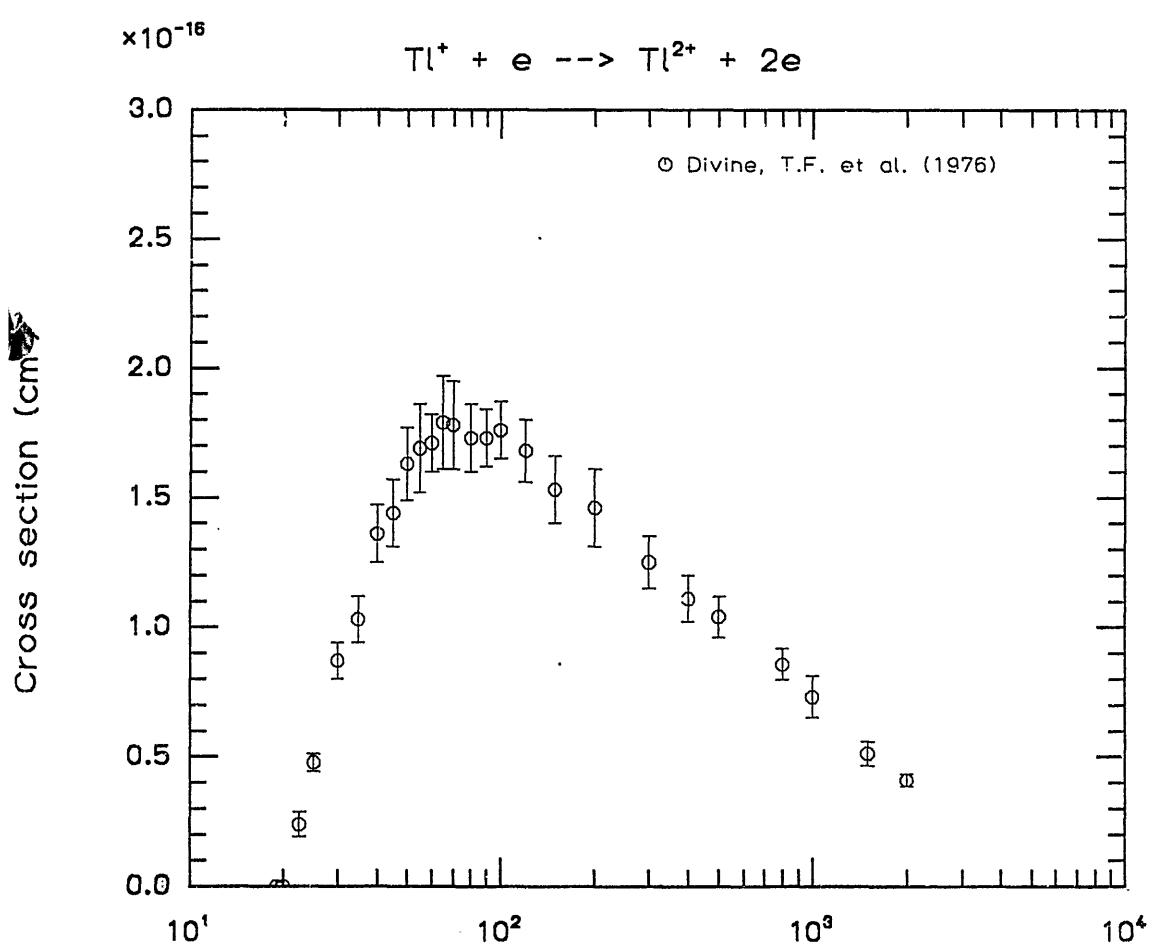


Fig. 289 Electron energy (eV)

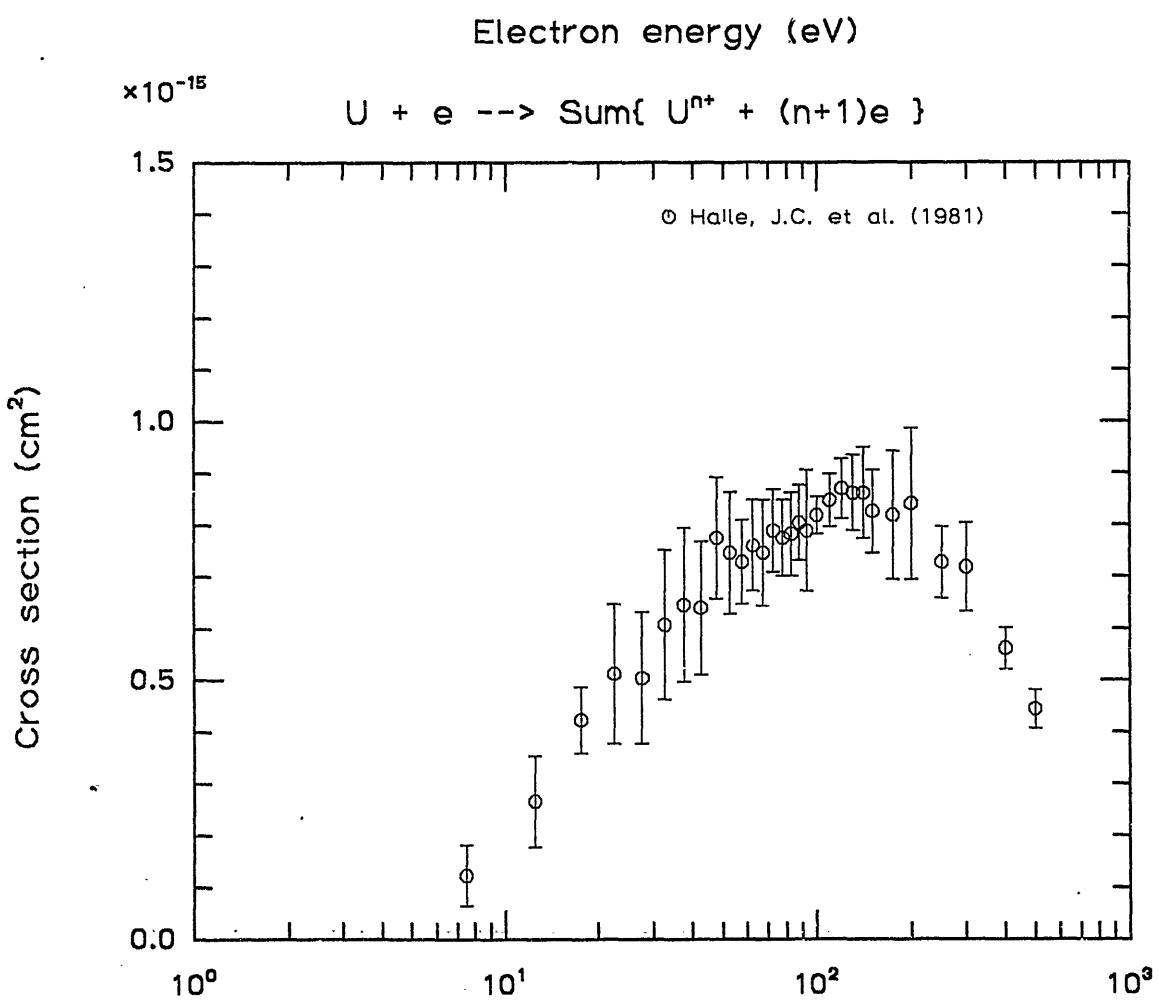
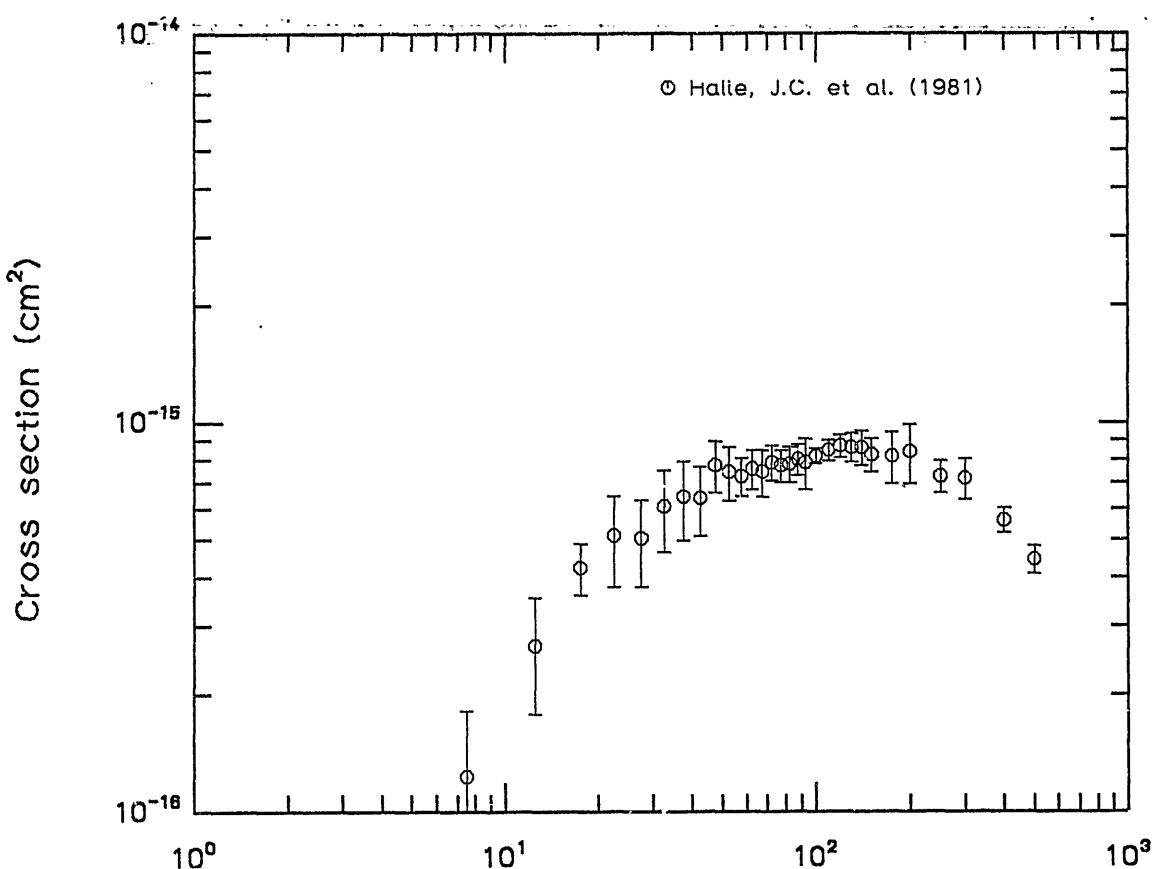
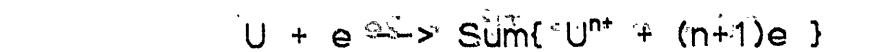


Fig. 290

Electron energy (eV)

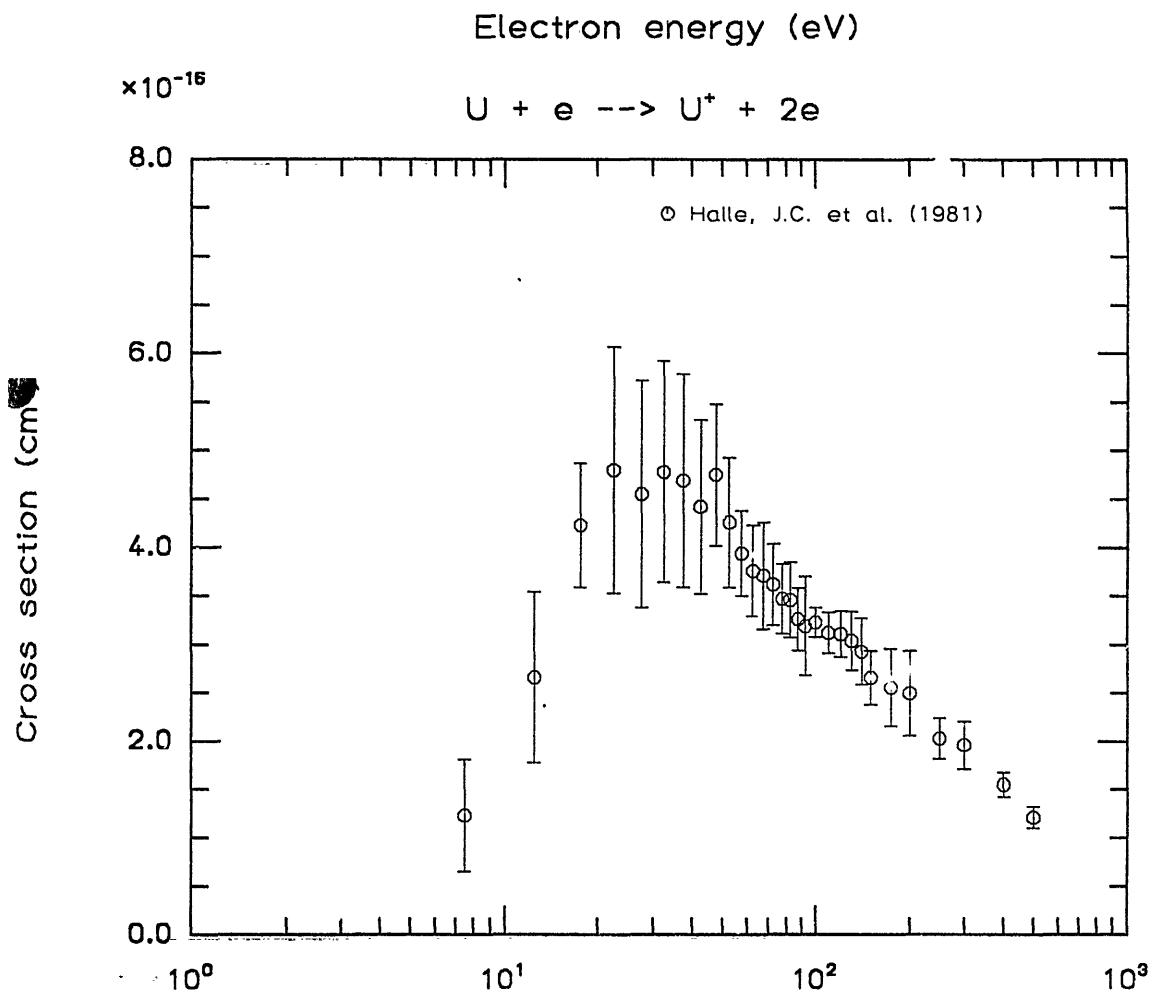
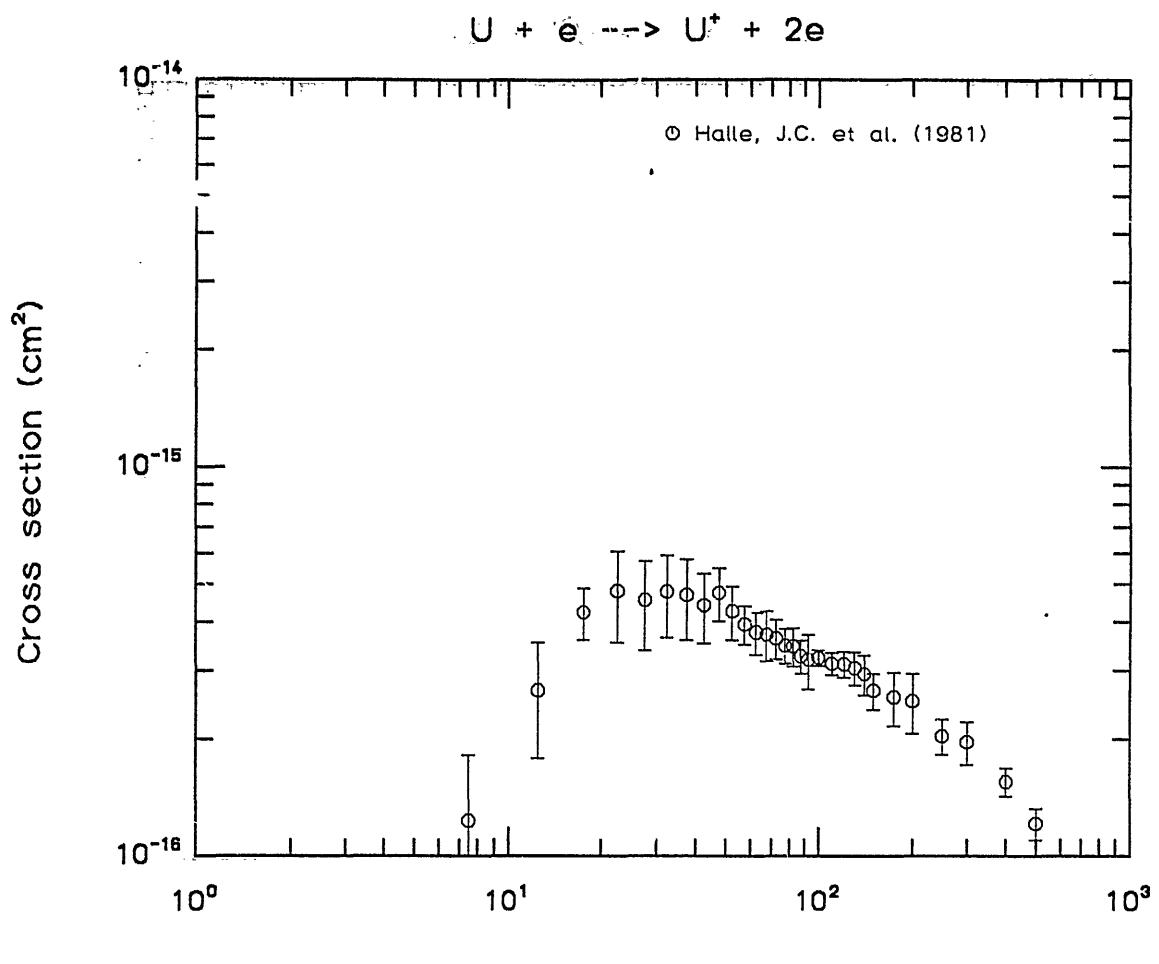


Fig. 291

Electron energy (eV)

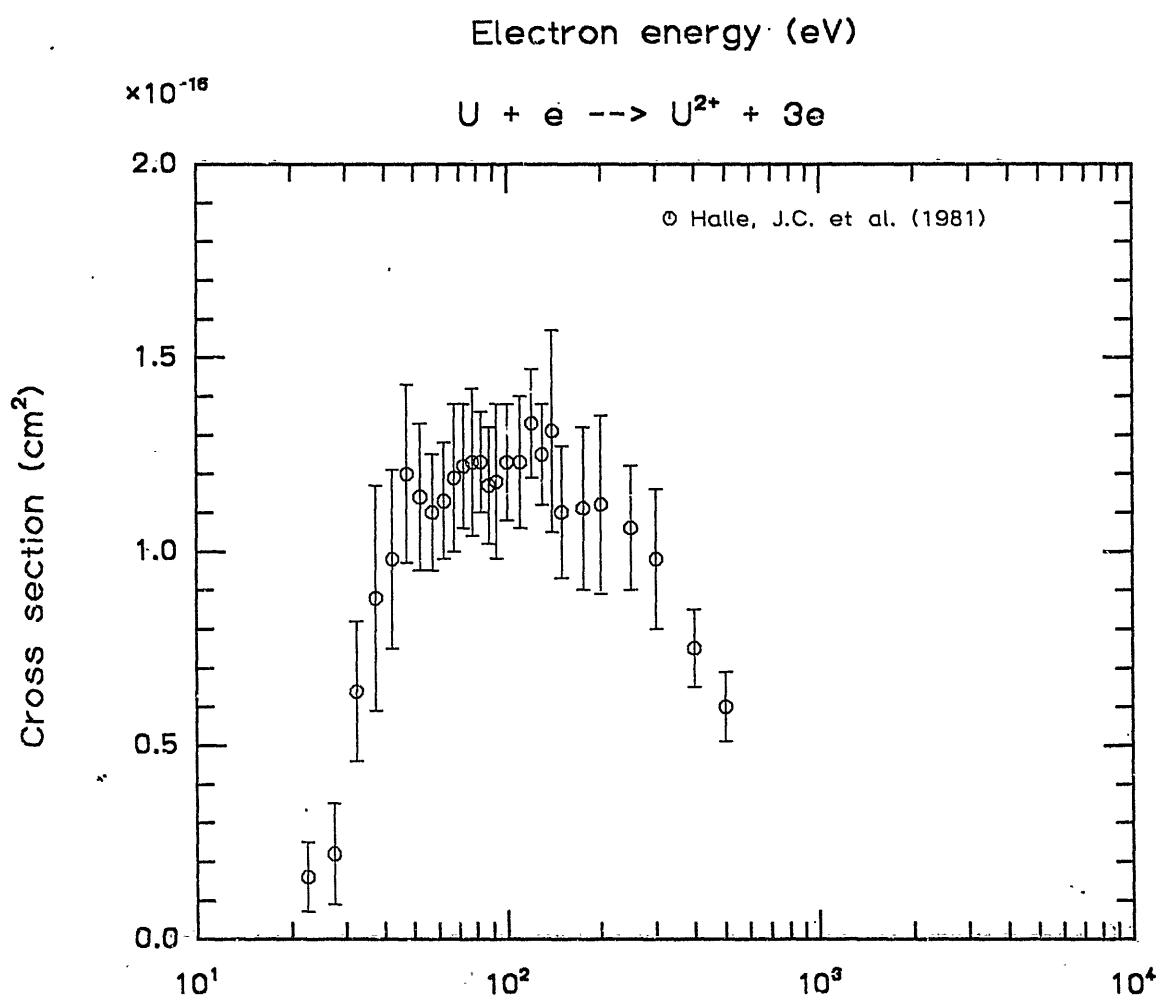
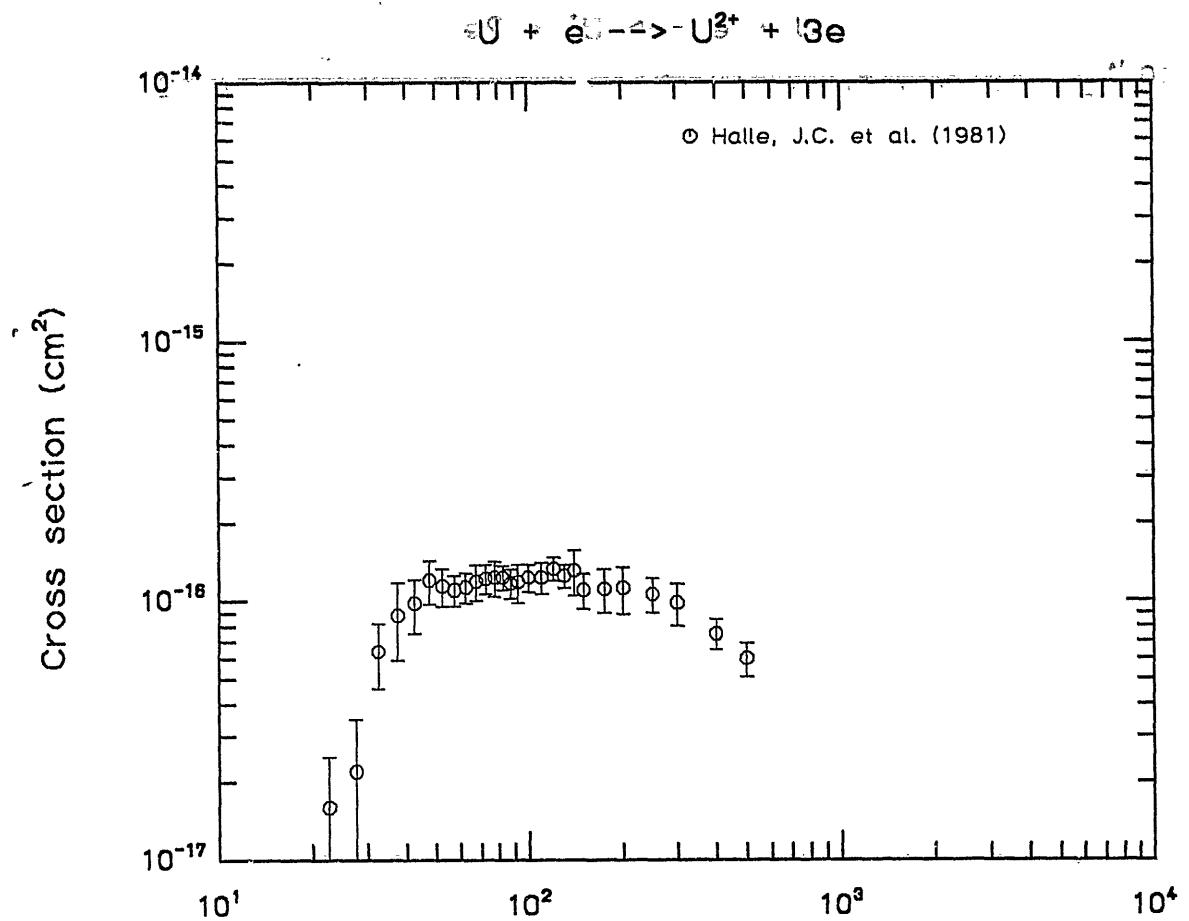
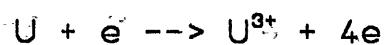
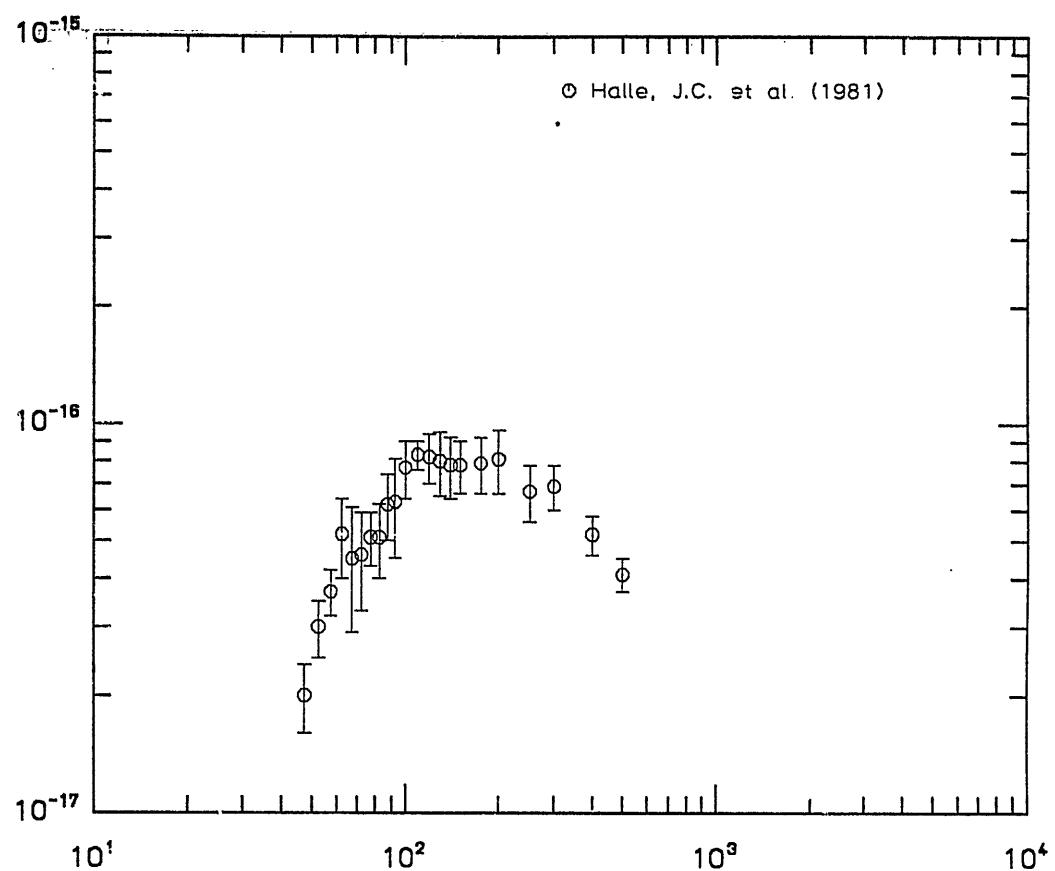


Fig. 292      Electron energy (eV)      res. up3



Cross section ( $\text{cm}^2$ )



Electron energy (eV)

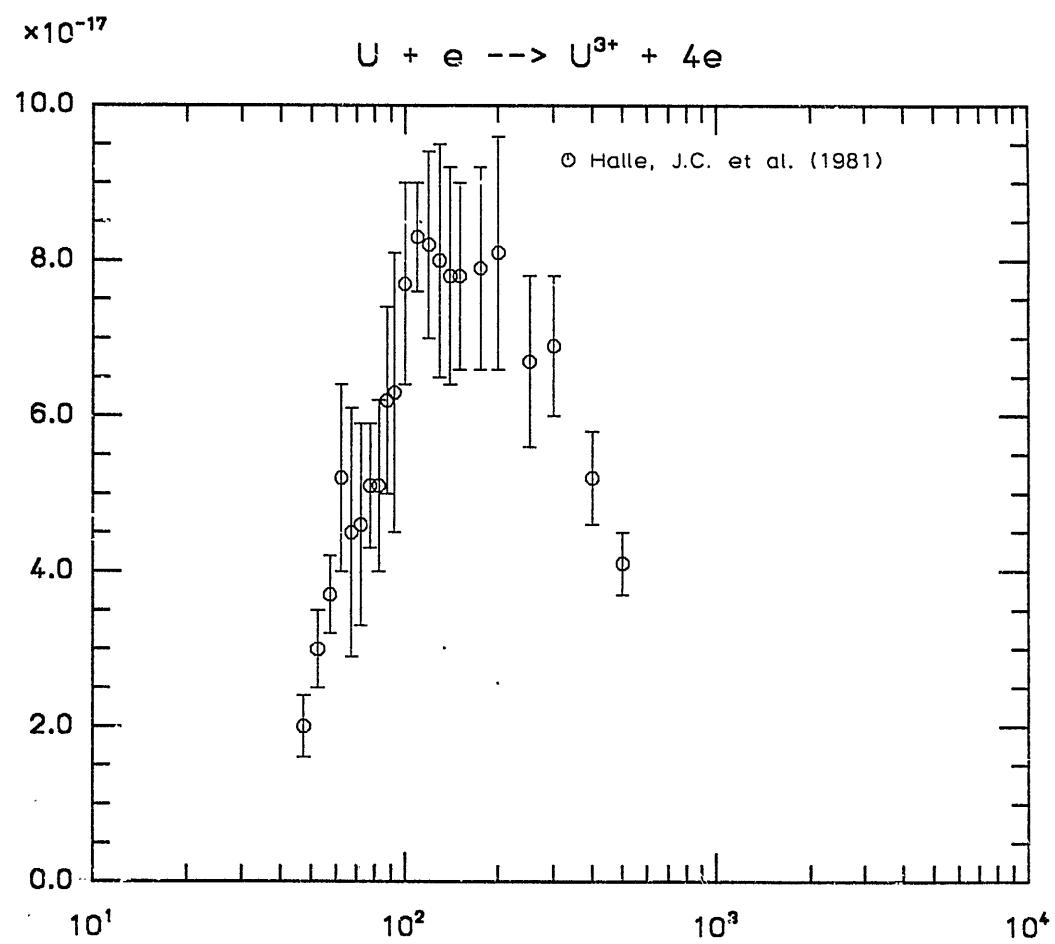


Fig. 293

Electron energy (eV)

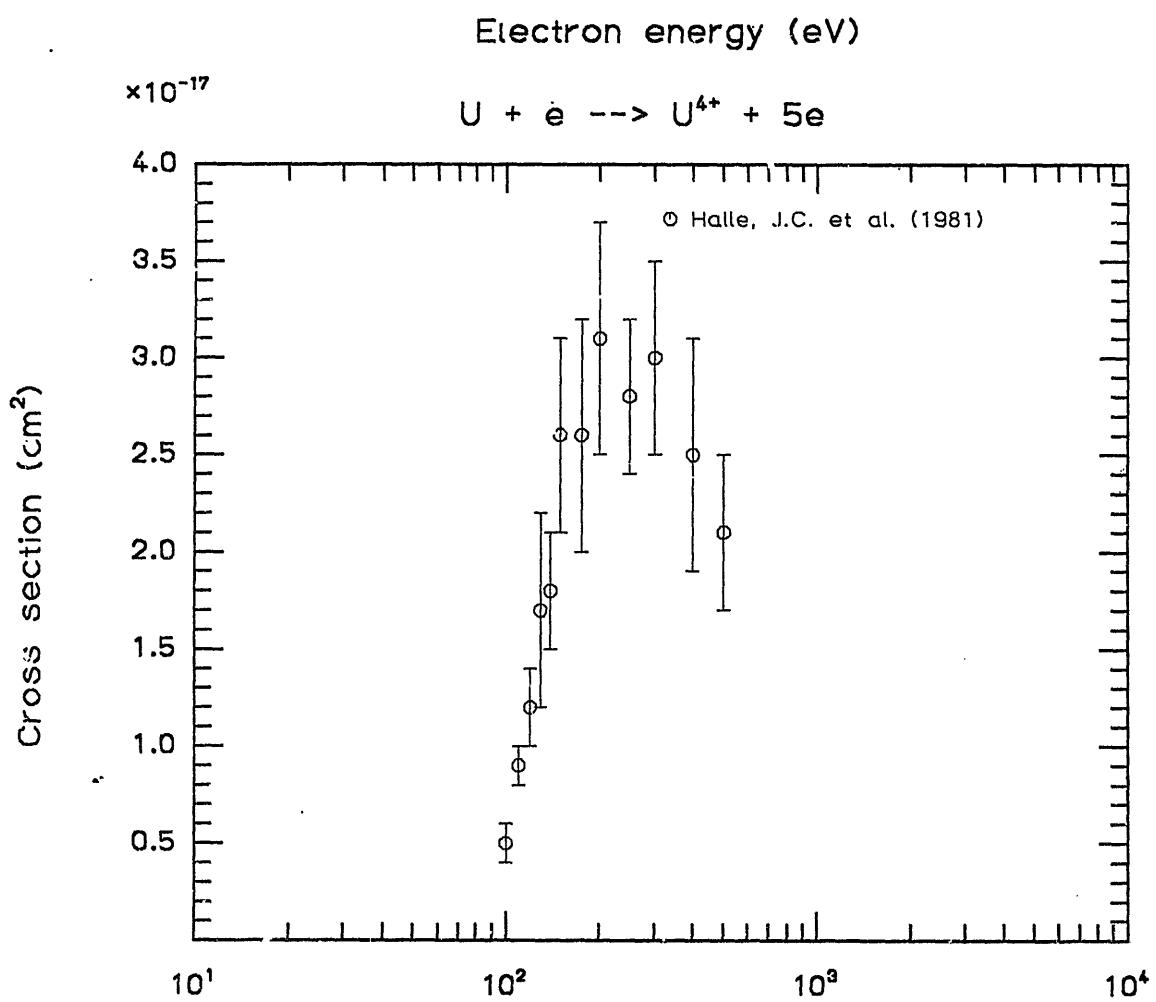
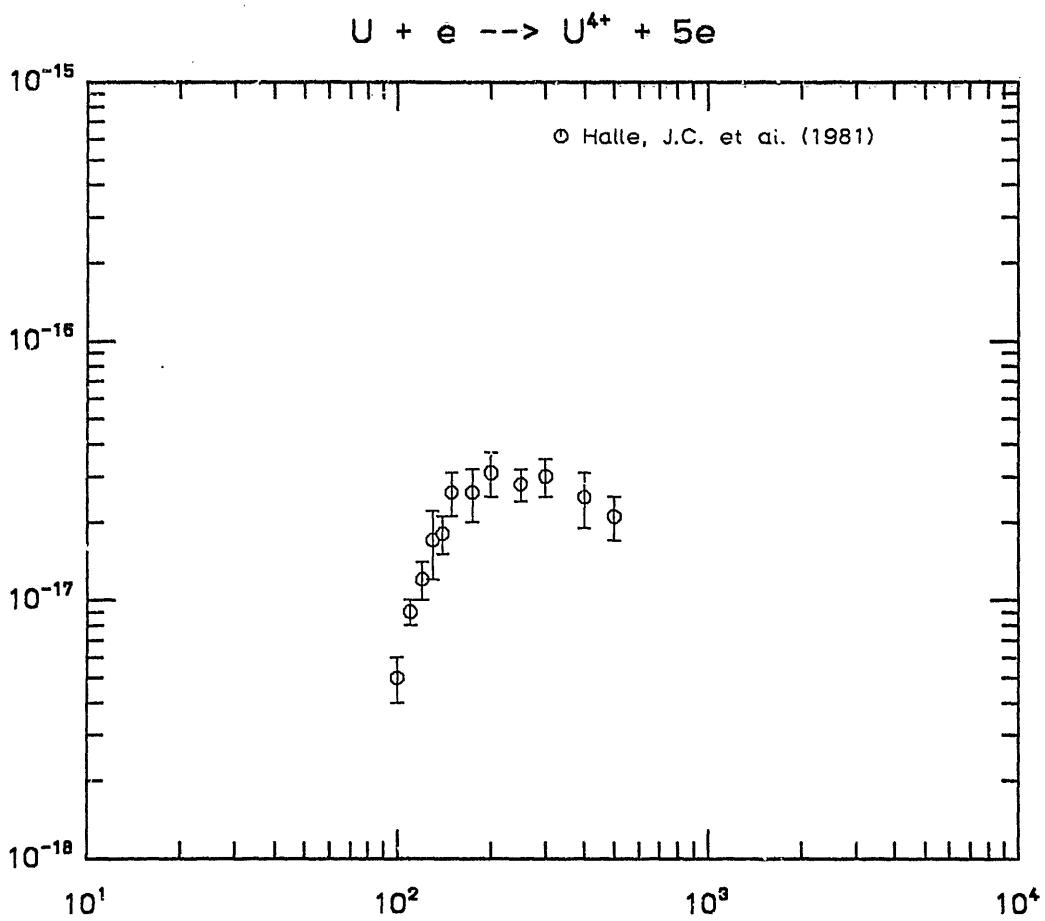


Fig. 294

Electron energy (eV)

## *References for Figures*

### **Experiment**

- Achenbach, C., Muller, A., Salzborn, E., Becker, R.**  
J. Phys. B 17, 1405 (1984)  
Single ionization of multiply charged xenon ions by electron impact
- Adamczyk, B., Boerboom, A.J.H., Schram, B.L., Kistemaker, J.**  
J. Chem. Phys. 44, 4640 (1966)  
Partial ionization cross sections of He, Ne, H<sub>2</sub> and CH<sub>4</sub> for electrons from 20 to 500 eV
- Al'tken, K.L., Harrison, M.F.A.**  
J. Phys. B 4, 1176 (1971)  
Measurements of the cross sections for electron impact ionization of multi-electron ions. I. O<sup>+</sup> to O<sup>2+</sup> and O<sup>2+</sup> to O<sup>3+</sup>
- Al'tken, K.L., Harrison, M.F.A., Runder, R.D.**  
J. Phys. B 4, 1189 (1971)  
Measurement of the cross sections for electron impact ionization of multi-electron ions. II. N<sup>2+</sup> to N<sup>3+</sup> and C<sup>+</sup> to C<sup>2+</sup>
- Asundi, R.K., Craggs, J.D., Kurepa, M.V.**  
Proc. Phys. Soc. 82, 967 (1963)  
Electron attachment and ionization in oxygen, carbon monoxide and carbon dioxide
- Asundi, R.K., Kurepa, M.V.**  
J. Electron. Control 15, 41 (1963)  
Ionization cross sections in He, Ne, Ar, Kr and Xe by electron impact
- Bellina, G.M., Pavlov, S.I., Rakhevskii, V.I., Sorokaletov, O.D.**  
J. Appl. Mechan. Tech. Phys. 2, 86 (1965)  
Measurement of electron impact ionization functions for metal atoms
- Bleakney, W.**  
Phys. Rev. 35, 139 (1930)  
Probability and critical potentials for the formation of multiply charged ions in Hg vapor by electron impact
- Bleakney, W.**  
Phys. Rev. 36, 1303 (1930)  
Ionization potentials and probabilities for the formation of multiply charged ions in helium, neon and argon
- Boyd, R.L., Green, G.W.**  
Proc. Phys. Soc. 71, 351 (1958)  
Electron ionization cross sections using chopped beams
- Brink, G.O.**  
Phys. Rev. 134, A345 (1964)  
Absolute ionization cross sections of the alkali metals

- Brook, E., Harrison, M.F.A., Smith, A.C.H.**  
 J. Phys. B 11, 3115 (1978)  
 Measurements of the electron impact ionization cross sections of He, C, O and N atoms
- Cowling, I.R., Fletcher, J.**  
 J. Phys. B 6, 665 (1973)  
 Electron-molecule collision ionization in hydrogen and deuterium
- Crandall, D.H., Phaneuf, R.A., Taylor, P.O.**  
 Phys. Rev. A 18, 1911 (1978)  
 Electron-impact ionization of C<sup>3+</sup> and N<sup>4+</sup>
- Crandall, D.H., Phaneuf, R.A., Hasselquist, B.E., Gregory, D.C.**  
 J. Phys. B 12, L249 (1979)  
 Measured cross sections for ionization of C<sup>3+</sup>, N<sup>4+</sup> and O<sup>5+</sup> ions with contribution due to excitation-autoionization
- Crandall, D.H., Phaneuf, R.A., Gregory, D.C.**  
 ORNL/TM Z020 (1979)  
 Electrom impact ionization of multicharged ions
- Crandall, D.H., Phaneuf, R.A., Falk, R.A., Belic, D.S., Dunn, G.H.**  
 Phys. Rev. A 25, 143 (1982)  
 Absolute cross-section measurements for electron-impact ionization of Na-like ions-Mg<sup>+</sup>, Al<sup>2+</sup>, and Si<sup>3+</sup>
- Crawford, C.K., Wang, K.I.**  
 J. Chem. Phys. 47, 4667 (1967)  
 Electron-impact ionization cross sections for silver
- Crowe, A., McConkey, J.W.**  
 J. Phys. B 6, 2088 (1973)  
 Dissociative ionization by electron impact I. protons from H<sub>2</sub>
- Crowe, A., McConkey, J.W.**  
 J. Phys. B 6, 2108 (1973)  
 Dissociative ionization by electron impact II. N<sup>+</sup> and N<sup>++</sup> from N<sub>2</sub>
- Daly, N.R., Powell, R.E.**  
 Proc. Phys. Soc. 89, 273 (1966)  
 Electron collisions in nitrogen
- Danjo, A., Matsumoto, A., Ohtani, S., Suzuki, H., Tawara, H., Wakuya, K., Yoshino, M.**  
 J. Phys. Soc. Jpn. 53, 4091 (1984)  
 Electron impact single ionization of Ne<sup>2+</sup>, Ar<sup>2+</sup>, Kr<sup>2+</sup> and Xe<sup>2+</sup> ions
- Defrance, P., Brouillard, F., Claeys, W., van Wassenhove, G.**  
 J. Phys. B 14, 103 (1981)  
 Crossed beam measurement of absolute cross sections on alternative method and its application to the electron impact ionisation of He<sup>+</sup>

- Defrance, P., Claeys, W., Cornet, A., Poulaert, G.**  
 J. Phys. B 14, 111 (1981)  
 Electron impact ionization of metastable atomic hydrogen
- Dettman, J.M., Karstensen, F.**  
 J. Phys. B 15, 287 (1982)  
 Absolute ionisation functions for electron impact with barium
- Diserens, M.J., Harrison, M.F.A., Smith, A.C.H.**  
 J. Phys. B 17, L621 (1984)  
 A measurement of the cross section for electron impact ionisation of  $\text{Ne}^+$
- Divine, T.F., Feeney, R.F., Sayle, II, W.E., Hooper, J.W.**  
 Phys. Rev. A 13, 54 (1976)  
 Absolute experimental cross section for the ionization of  $\text{Ti}^+$  ions by electron impact
- Dixon, A.J., von Engel, A., Harrison, M.F.A.**  
 Proc. R. Soc. London A 343, 333 (1975)  
 A measurement of the electron impact ionization cross section of atomic hydrogen in the metastable 2S state
- Dixon, A.J., Harrison, M.F.A., Smith, A.C.H.**  
 J. Phys. B 9, 2617 (1976)  
 A measurement of the electron impact ionization cross sections of helium atoms in metastable states
- Dolder, K.T., Harrison, M.F.A., Thonemann, P.C.**  
 Proc. R. Soc. London A 264, 367 (1961)  
 A measurement of the ionization cross sections of ions by electron impact
- Dolder, K.T., Harrison, M.F.A., Thonemann, P.C.**  
 Proc. R. Soc. London A 274, 546 (1963)  
 A measurement of the ionization cross section of  $\text{Ne}^+$  to  $\text{Ne}^{2+}$  by electron impact
- Donets, E.D., Ovsyannikov, V.P.**  
 Sov. Phys.-JETP 53, 466 (1981)  
 Investigation of ionization of positive ions by electron impact
- El-Sherbini, Th.M., Van der Wiel, M.J., de Heer, F.J.**  
 Physica 48, 157 (1970)  
 Multiple ionization of Kr and Xe by 2-14 keV electrons
- Falk, R.A., Dunn, G.H.**  
 Phys. Rev. A 27, 75 (1983)  
 Electron-impact ionization of  $\text{Be}^+$
- Falk, R.A., Dunn, G.H., Gregory, D.C., Crandall, D.H.**  
 Phys. Rev. A 27, 724 (1983)  
 Measurement of the contribution of excitation-autoionization to electron impact ionization of ions:  $\text{Ti}^{+3}$ ,  $\text{Zr}^{+3}$ ,  $\text{Hf}^{+3}$ , and  $\text{Ta}^{+3}$

- Falk, R.A., Stefani, G., Camilloni, R., Dunn, G.H., Phaneuf, R.A., Gregory, D.C., Crandall, D.H.**  
**Phys. Rev. A 28, 91 (1983)**  
**Measured electron-impact ionization of Be-like ions: B<sup>+</sup>, C<sup>2+</sup>, N<sup>3+</sup>, and O<sup>4+</sup>**
- Feeley, R.K., Hooper, J.W., Elford, M.T.**  
**Phys. Rev. A 5, 1469 (1972)**  
**Absolute experimental cross sections for the ionization of singly charged barium ions by electron impact**
- Feeley, R.K., Sayle, II, W.E., Divine, T.F.**  
**Phys. Rev. A 18, 82 (1978)**  
**Absolute experimental cross sections for the electron-impact ionization of Rb<sup>+</sup> ions**
- Fite, W.L., Brackmann, R.T.**  
**Phys. Rev. 112, 1141 (1958)**  
**Collisions of electron with hydrogen atoms. I. Ionization**
- Fite, W.L., Brackmann, R.T.**  
**Phys. Rev. 113, 815 (1959)**  
**Ionization of atomic oxygen on electron impact**
- Fletcher, J., Cowling, I.R.**  
**J. Phys. B 6, L258 (1973)**  
**Electron impact ionization of neon and argon**
- Gaudin, A., Hageman, R.**  
**J. Chim. Phys. 64, 1209 (1967)**  
**Determinations absolues des sections efficaces totales et partielles d'ionisation de l'hélium, du néon, de l'argon et de l'acétylène, pour des électrons de 100 à 200 eV**
- Gregory, D.C., Dittner, P.F., Crandall, D.H.**  
**Phys. Rev. A 27, 724 (1983)**  
**Absolute-cross-section measurements for electron-impact ionization of triply charged inert-gas ions: Ne<sup>3+</sup>, Ar<sup>3+</sup>, Kr<sup>3+</sup>, and Xe<sup>3+</sup>**
- Gregory, D.C., Crandall, D.H.**  
**Phys. Rev. A 27, 2338 (1983)**  
**Measurement of the cross section for electron-impact ionization of Xe<sup>6+</sup> ions**
- Gregory, D.C., Crandall, D.H., Phaneuf, R.A., Howald, A.M., Dunn, G.H., Falk, R.A., Mueller, D.W., Morgan, T.J.**  
**ORNL/TM 9501 (1985)**  
**Electron impact ionization of multicharged ions at ORNL:1980-1984**
- Griffin, D.C., Bottcher, C., Pindzola, M.S., Younger, S.M., Gregory, D.C., Crandall, D.H.**  
**Phys. Rev. A 29, 1729 (1984)**  
**Electron-impact ionization in the xenon isonuclear sequence**
- Halas, St., Adamczyk, B.**  
**Int. J. Mass Spectrometry and Ion Phys. 10, 157 (1972)**  
**Cross sections for the production of N<sub>2</sub><sup>+</sup>, N<sup>+</sup> and N<sup>2+</sup><sub>2</sub> from nitrogen by electrons in the energy range 16-600 eV**

- Halle, J.C., Lo, H.H., Fite, W.**  
**Phys. Rev. A 23, 1708 (1981)**  
**Ionization of uranium atoms by electron impact**
- Hamdan, M., Birkinshaw, K., Hasted, J.B.**  
**J. Phys. B 11, 331 (1978)**  
**Ionisation of positive ions by electrons in the hollow-beam trap.**
- Harrison, M.F.A., Dolder, K.T., Thonemann, P.C.**  
**Proc. Phys. Soc. 82, 368 (1963)**  
**A measurement of the cross section for the ionization of N<sup>+</sup> to N<sup>2+</sup> by electron impact**
- Hasted, J.B., Awad, G.L.**  
**J. Phys. B 5, 1719 (1972)**  
**Electron impact ionization of ions trapped in a hollow electron beam.**
- Heil, H., Scott, B.**  
**Phys. Rev. 145, 279 (1966)**  
**Cesium ionization cross section from threshold to 50 eV**
- Hertling, D.R., Feeney, R.K., Hughes, D.W., Sayle II, W.E.**  
**J. Appl. Phys. 53, 5427 (1982)**  
**Absolute experimental cross sections for the electron impact single, double, triple, and quadruple ionization of Cs<sup>+</sup> ions**
- Hooper, J.W., Lineberger, W.C., Bacon, F.M.**  
**Phys. Rev. 141, 165 (1966)**  
**Absolute cross sections for single ionization of alkali ions by electron impact.  
 II. Na<sup>+</sup> and K<sup>+</sup> results and comparison with theory**
- Hughes, D.W., Feeney, R.K.**  
**Phys. Rev. A 23, 2241 (1981)**  
**Absolute experimental cross sections for the electron-impact multiple ionization of singly charged rubidium ions**
- Jalin, R., Hageman, R., Botter, R.**  
**J. Chem. Phys. 59, 952 (1973)**  
**Absolute electron impact ionization cross sections of Li in the energy range from 100 to 2000 eV**
- Jones, T.J.**  
**Phys. Rev. 29, 822 (1929)**  
**Probability of ionization of mercury vapor by electron impact**
- Karstensen, F., Schneider, M.**  
**Z. Phys. A 273, 321 (1975)**  
**Ionization function of magnesium for electron impact**
- Karstensen, F., Schneider, M.**  
**J. Phys. B 11, 167 (1978)**  
**Absolute cross sections for single and double ionization of Mg atoms by electron impact**

- Korchevol, Yu.P., Przonski, A.M.**  
 Sov. Phys.-JETP 24, 1089 (1967)  
 Effective electron excitation and ionization cross sections for cesium, rubidium and potassium atoms in the pre-threshold region
- Krebarle, P., Godbole, E.W.**  
 J. Chem. Phys. 36, 302 (1962)  
 Ionization and fragmentation of some molecules by high energy electrons
- Lin, S.S., Stafford, F.E.**  
 J. Chem. Phys. 47, 4664 (1967)  
 Electron-impact ionization cross sections. IV. group IVb atoms
- Lineberger, W.C., Hooper, J.W., McDaniel, E.W.**  
 Phys. Rev. 141, 151 (1966)  
 Absolute cross sections for single ionization of alkali ions by electron impact.  
 I. Description of apparatus and  $\text{Li}^+$  results
- Liska, J.W.**  
 Phys. Rev. 46, 169 (1934)  
 Efficiencies of ionization of helium and mercury by electron impact at high voltage
- Long, D.R., Geballe, R.**  
 Phys. Rev. A 1, 260 (1970)  
 Electron-impact ionization of  $\text{He}(2s\ ^3S)$
- Lyubimov, A.P., Pavlov, S.I., Rakhevskii, V.I., Zaitseva, N.G.**  
 Bull. Acad. USSR. Phys. Ser. 17, 1033 (1963)  
 Procedure for measuring the ionization cross sections and ionization coefficients of metal atoms
- Muller, A., Salzborn, E., Frodl, R., Becker, R., Klein, H., Winter, H.**  
 J. Phys. B 13, 1877 (1980)  
 Absolute ionization cross sections for electron incident on  $\text{O}^+$ ,  $\text{Ne}^+$ ,  $\text{Xe}^+$ , and  $\text{Ar}^{i+}$  ( $i=1,\dots,5$ ) ions
- Muller, A., Frodl, R.**  
 Phys. Rev. Lett. 44, 29 (1980)  
 L-shell contributions to multiple ionization of  $\text{Ar}^{i+}$  ions ( $i=1,2,3$ ) by electron impact
- Mark, T.D.**  
 J. Chem. Phys. 63, 3731 (1975)  
 Cross section for single and double ionization of  $\text{N}_2$  and  $\text{O}_2$  molecules by electron impact from threshold up to 170 eV
- Martin, S.O., Peart, B., Dolder, K.T.**  
 J. Phys. B 1, 537 (1968)  
 Measurements of cross sections for the ionization of  $\text{Mg}^+$  to  $\text{Mg}^{2+}$  by electron impact
- McFarland, R.H., Kinney, J.D.**  
 Phys. Rev. 137, A1058 (1965)  
 Absolute cross sections of lithium and other alkali metal atoms for ionization by electrons

**McFarland, R.H.**

Phys. Rev. 159, 20 (1967)

Electron-impact ionization measurements of surface-ionizable atoms

**McGowan, J.W., Clarke, E.M.**

Phys. Rev. 167, 43 (1968)

Ionization of H(1s) near threshold

**McGowan, J.W., Fineman, R.A., Clarke, E.M., Hanson, H.P.**

Phys. Rev. 167, 52 (1968)

Direct and auto-ionization of H<sub>2</sub> near threshold

**Montague, R.G., Harrison, M.F.A.**

J. Phys. B 16, 3045 (1983)

A Measurement of the Cross Section for Electron Impact Ionization of Al<sup>+</sup>

**Montague, R.G., Diserens, M.J., Harrison, M.F.A.**

J. Phys. B 17, 2085 (1984)

A measurement of the cross section for electron impact ionisation of Fe<sup>+</sup>

**Montague, R.G., Harrison, M.F.A.**

J. Phys. B 17, 2707 (1984)

A measurement of the cross section for electron impact ionisation of singly charged tungsten ions

**Montague, R.G., Harrison, M.F.A., Smith, A.C.H.**

J. Phys. B 17, 3295 (1984)

A measurement of the cross section for ionisation of helium by electron impact using a fast crossed beam technique

**Muller, A., Achenbach, C., Salzborn, E., Becker, R.**

J. Phys. B 17, 1427 (1984)

Multiple ionization of multiply charged xenon ions by electron impact

**Muller, A., Tinschert, K., Achenbach, C., Becker, R., Salzborn, E.**

J. Phys. B (1984)

Electron-impact double ionisation of Ar<sup>1+</sup> and Ar<sup>2+</sup> ions

**Muller, A., Tinschert, K., Achenbach, C., Salzborn, E., Becker, R., Pindzola, M.S.**

Phys. Rev. Letters 54, 414 (1985)

Collision-strength shift in electron impact single and double ionization of Sb and Bi ions

**Muller, A.**

Private Communication (1985)

**Nagy, P., Skutlartz, A., Schmidt, V.**

J. Phys. B 13, 1249 (1980)

Absolute ionization cross sections for electron impact in rare gases

**Nygaard, K.J.**

J. Chem. Phys. 49, 1995 (1968)

Electron-impact ionization cross sections in cesium

**Okuno, Y., Okuno, K., Kaneko, Y., Kanomata, I.**

J. Phys. Soc. Jpn. **29**, 154 (1970)

Absolute measurement of total ionization cross section of Mg by electron impact

**Okuno, Y.**

J. Phys. Soc. Jpn. **31**, 1189 (1971)

Ionization cross sections of Ca, Sr and Ba by electron impact

**Pavlov, S.I., Rakhovskii, V.I., Fedorova, G.M.**

Sov. Phys.-JETP **25**, 12 (1967)

Measurement of cross sections for ionization by electron impact at low vapor pressures

**Pavlov, S.I., Stotskii, G.I.**

Sov. Phys.-JETP **31**, 61 (1970)

Single and multiple ionization of lead atoms by electrons

**Pearl, B., Dolder, K.T.**

J. Phys. B **1**, 240 (1968)

Measurements of cross sections for the ionization of  $\text{Na}^+$  to  $\text{Na}^{2+}$  and  $\text{K}^+$  to  $\text{K}^{2+}$  by electron impact

**Pearl, B., Dolder, K.T.**

J. Phys. B **1**, 872 (1968)

Measurements of cross sections for the ionization of  $\text{Li}^+$  and  $\text{Ba}^+$  ions by electron impact

**Pearl, B., Dolder, K.T.**

J. Phys. B **2**, 1169 (1969)

The ionization of  $\text{Li}^+$  to  $\text{Li}^{3+}$  by electron impact

**Pearl, B., Martin, S.O., Dolder, K.T.**

J. Phys. B **2**, 1176 (1969)

Measurements of electron impact ionization cross section of  $\text{Mg}^{2+}$  ions and a note on the classical scaling law

**Pearl, B., Walton, D.S., Dolder, K.T.**

J. Phys. B **2**, 1347 (1969)

The ranges of validity of the Born and Bethe approximations for the single ionization of  $\text{He}^+$  and  $\text{Li}^+$  ions by electron impact

**Pearl, B., Dolder, K.T.**

J. Phys. B **8**, 56 (1975)

Measurements of cross sections for inner- and outer- shell ionization of  $\text{Rb}^+$ ,  $\text{Cs}^+$ ,  $\text{Ca}^+$  and  $\text{Sr}^+$  ions by electron impact

**Pearl, B., Stevenson, J.G., Dolder, K.T.**

J. Phys. B **6**, 146 (1973)

Measurements of cross sections for the ionization of  $\text{Ba}^+$  by energy resolved electrons

**Pindzola, M.S., Griffin, D.C., Battcher, C., Crandall, D.H., Phaneuf, R.A., Gregory, D.C.**

Phys. Rev. A **29**, 1749 (1984)

Electron-impact double ionization of rare-gas ions

**Pottle, R.F.**

J. Chem. Phys. 44, 916 (1966)

Cross sections for ionization by electrons. I. absolute ionization cross sections of Zn, Cd and Te<sub>2</sub>. II. comparison of theoretical with experimental values for atoms and molecules

**Rakhovskii, V.I., Stepanov, A.M.**

High Temp. 7, 1001 (1969)

Absolute values of the apparent cross sections for calcium ionization by electrons

**Rapp, D., Englander-Golden, P.**

J. Chem. Phys. 43, 1464 (1965)

Total cross sections for ionization and attachment in gases by electron impact. I. positive ionization

**Rogers, W.T., Stefanl, G., Camilloni, R., Dunn, G.H., Msezane, A.Z., Henry, R.J.W.**

Phys. Rev. A 25, 737 (1982)

Electron-impact ionization of Zn<sup>+</sup> and Ga<sup>+</sup>

**Rothe, E.W., Marino, L.L., Neynaber, R.H., Trujillo, S.M.**

Phys. Rev. 125, 582 (1962)

Electron impact ionization of atomic hydrogen and atomic oxygen

**Schneider, M.**

J. Phys. D. 7, L83 (1974)

Measurement of absolute ionization cross sections for electron impact

**Schram, B.L., de Heer, F.J., Van der Wiel M.J., Kistemaker, J.**

Physica 31, 94 (1965)

Ionization cross sections for electrons (0.6-20 keV) in noble and diatomic gases

**Schram, B.L., Boerboom, A.J.H., Kistemaker, J.**

Physica 32, 185 (1966)

Partial ionization cross sections of noble gases for electrons with energy 0.5-16 keV. I helium and neon

**Schram, B.L.**

Physica 32, 197 (1966)

Partial ionization cross sections of noble gases for electrons with energy 0.5-18 keV. II argon, krypton and xenon

**Schram, B.L., Moustafa, H.R., Schutten, J., de Heer, F.J.**

Physica 32, 734 (1966)

Ionization cross sections for electrons (100-600 eV) in noble and diatomic gases

**Schroer, J.M., Gunduz, D.H., Livingston, S.**

J. Chem. Phys. 58, 5035 (1973)

Electron impact ionization cross sections of Cu and Au between 40 and 250 eV, and the velocity of evaporated atoms

**Shchemelinin, S.G., Andreev, E.P.**

Sov. Phys.-Tech. Phys. 20, 941 (1976)

Absolute cross sections for single and multiple electron-impact ionization of He, Ne and Ar

- Shimon, L.L., Nepiirov, E.I., Zapesochnyi, I.P.**  
Sov. Phys.-Tech. Phys. 20, 434 (1975)  
Effective total electron-impact ionization cross sections for aluminum, gallium, indium and thallium
- Smith, A.C.H., Caplinger, E., Neynaber, R.H., Rothe, E.W., Trujillo, S.M.**  
Phys. Rev. 127, 1647 (1962)  
Electron impact ionization of atomic nitrogen
- Smith, P.T.**  
Phys. Rev. 36, 1293 (1930)  
The ionization of helium, neon, and argon by electron impact
- Smith, P.T.**  
Phys. Rev. 37, 808 (1931)  
The ionization of mercury vapor by electron impact
- Srinivasan, V., Rees, J.A.**  
J. Appl. Phys. 18, 59 (1967)  
A note on the total ionization cross section for electrons in the inert gases and carbon monoxide
- Stephan, K., Helm, H., Mark, T.D.**  
J. Chem. Phys. 72, 3763 (1980)  
Mass spectrometric determination of partial electron impact ionization cross sections of He, Ne, Ar and Kr from threshold up to 180 eV
- Tate, J.T., Smith, P.T.**  
Phys. Rev. 39, 270 (1932)  
The efficiencies of ionization and ionization potentials of various gases under electron impact
- Tate, J.T., Smith, P.T.**  
Phys. Rev. 46, 773 (1934)  
Ionization potentials and probabilities for the formation of multiply charged ions in the alkali vapors and in krypton and xenon
- Tozer, B.A., Craggs, J.D.**  
J. Electron. Control 8, 103 (1960)  
Cross sections for ionization of the inert gases by electron impact
- Vainshtein, L.A., Ochkur, V.I., Rakhevskii, V.I., Stepanov, A.M.**  
Sov. Phys.-JETP 34, 271 (1972)  
Absolute values of electron impact ionization cross sections for magnesium, calcium, strontium and barium
- Van der Wiel, M.J., El-Sherbini, Th.M., Vriens, L.**  
Physica 42, 411 (1969)  
Multiple ionization of He, Ne and Ar by 2-16 keV electrons
- Wareing, J.B., Dolder, K.T.**  
Proc. Phys. Soc. 91, 887 (1967)  
A measurement of the cross section for ionization of  $\text{Li}^+$  to  $\text{Li}^{2+}$  by electron impact

- Woodruff, P.R., Hublet, M.-C., Harrison, M.F.A.**  
J. Phys. B 11, L305 (1978)  
A measurement of the cross section for electron impact ionization of Ar<sup>+</sup>
- Woodruff, P.R., Hublet, M.-C., Harrison, M.F.A., Brook, E.**  
J. Phys. B 11, L679 (1978)  
A measurement of the cross section for electron impact ionization of C<sup>2+</sup>
- Zapesochnyi, I.P., Aleksakhin, I.S.**  
Sov. Phys.-JETP 28, 41 (1969)  
Ionization of alkali metal atoms by slow electrons
- Ziegler, D.L., Newman, J.H., Smith, K.A., Stebbings, R.F.**  
Planet. & Space Sci. 30, 451 (1982)  
Double ionization of atomic oxygen by electron impact
- Ziegler, D.L., Newman, J.H., Goeller, L.N., Smith, K.A., Stebbings, R.F.**  
Planet. & Space Sci. 30, 1269 (1982)  
Single and multiple ionization of sulfur atoms by electron impact

## References for Figures

### Theory

- Banks, D., Boesten, L.G.J.**  
J. Phys. B 11, 2209 (1978)  
Ionisation of He<sup>+</sup> ions by electron impact
- Bely, O., Schwartz, S.B., Val, J.L.**  
J. Phys. B 4, 1482 (1971)  
Autoionization structure in the ionization of Ba<sup>+</sup> by electron impact.
- Blaha, M., Davis, J.**  
NRL Memo. Rept. 4245 (1980)  
Electron ionization cross sections in the distorted-wave approximation
- Burgess, A., Summers, H.P., Cochrane, D.M., McWhirter, R.W.P.**  
Mon. Not. R. Astron. Soc. 179, 275 (1977)  
Cross-sections for ionization of positive ions by electron impact
- Burke, P.G., Kingston, A.E., Thompson, A.**  
J. Phys. B 16, L385 (1983)  
Electron impact ionisation of Ca<sup>+</sup>
- Burke, P.G., Fon, W.C., Kingston, A.E.**  
J. Phys. B 17, L733 (1984)  
Electron impact ionisation of Ti<sup>3+</sup>
- Burnett, T., Rountree, S.P.**  
Phys. Rev. A 20, 1468 (1979)  
Differential and total cross sections for electron-impact ionization of atomic oxygen
- Griffin, D.C., Bottcher, C., Pindzola, M.S.**  
Phys. Rev. A 25, 154 (1982)  
Contributions of excitation autoionization to the electron-impact ionization of Mg<sup>+</sup>, Al<sup>2+</sup>, and Si<sup>3+</sup> in the distorted-wave approximation with exchange
- Griffin, D.C., Pindzola, M.S., Bottcher, C.**  
J. Phys. B 17, 3183 (1984)  
Calculations of the contributions of excitation-autoionisation to the electron impact ionisation of Ca<sup>+</sup> and Ba<sup>+</sup> in the distorted-wave approximation
- Hahn, Y.**  
Phys. Rev. A 18, 1028 (1978)  
Distorted-wave theory of electron-ion collisions.II. Auger ionization and excitation fluorescence
- Jakubowicz, H., Moores, D.L.**  
J. Phys. B 14, 3733 (1981)  
Electron impact ionisation of Li-like and Be-like ions
- Kumar, A., Roy, B.N.**  
Phys. Lett. A 66, 362 (1978)  
Electron impact ionization of positive ions

- Kumar, A., Roy, B.N.**  
J. Phys. B 12, 3979 (1979)  
Electron impact ionisation of alkali metal ions
- Kunc, J.A.**  
J. Phys. B 13, 587 (1980)  
Electron ionisation cross sections of excited atoms and ions
- Mathur, K.C., Tripathi, A.N., Joshi, S.K.**  
Phys. Rev. 184, 242 (1969)  
Electron-impact ionization cross section of ions.
- Mathur, K.C., Tripathi, A.N., Joshi, S.K.**  
Astrophys. J. 165, 425 (1971)  
Cross-sections and reaction rates for electron impact ionization of lithiumlike and sodiumlike positive ions.
- McGuire, E.J.**  
Phys. Rev. A 26, 125 (1982)  
Born-approximation electron ionization cross sections for  $\text{Al}^{n+}$  and some ions of the Na isoelectronic sequence
- Moores, D.L.**  
J. Phys. B 5, 286 (1972)  
Electron impact ionization of positive ions with configuration  $1s^2 2s^2 2p^q$ .
- Moores, D.L.**  
J. Phys. B 11, L403 (1978)  
Electron impact ionization of Li-like and Be-like carbon and nitrogen ions
- Moores, D.L.**  
J. Phys. B 12, 4171 (1979)  
The autoionisation contribution to the electron impact ionisation cross sections of  $\text{C}^+$  and  $\text{N}^{2+}$  at low energies
- Pindzola, M.S., Griffin, D.C., Bottcher, C.**  
Phys. Rev. A 25, 211 (1982)  
Electron-impact excitation autoionization of Ga II
- Pindzola, M.S., Griffin, D.C., Bottcher, C.**  
J. Phys. B 16, L355 (1983)  
Electron impact double-ionisation of Xe II
- Rudge, M.R.H., Schwartz, S.B.**  
Proc. Phys. Soc. 88, 579 (1966)  
The ionization of complex ions by electron impact : I.- Ionization cross section for Fe XV and Fe XVI.
- Salop, A.**  
Phys. Rev. A 14, 2905 (1976)  
Electron impact ionization of multicharged ions.

- Sampson, D.H., Golden, L.B.**  
J. Phys. B 12, L785 (1979)  
Ionisation cross sections for Li-like ions including excitation-autoionisation contributions
- Sampson, D.H.**  
J. Phys. B 15, 2087 (1982)  
Ionisation of Na-like ions including excitation-autoionisation
- Sato, S., Kobayashi, K., Takebe, H.**  
Publ. Astron. Soc. Jpn. 19, 290 (1967)  
Collisional ionization cross-sections for Fe ions and the temperature of the solar corona.
- Thomas, B.K., Garcia, J.D.**  
Phys. Rev. 179, 94 (1969)  
Ionization of positive ions.
- Tsuji, A., Kotegawa, H., Narumi, H.**  
J. Phys. Soc. Jpn. 48, 2062 (1980)  
Eikonal approach to break-up processes of hydrogenic ions
- Younger, S.M.**  
Phys. Rev. A 22, 111 (1980)  
Electron-impact ionization cross sections for highly ionized hydrogen- and lithium-like atoms.
- Younger, S.M.**  
Phys. Rev. A 22, 1425 (1980)  
Electron-impact-ionization cross sections for highly ionized heliumlike atoms
- Younger, S.M.**  
Phys. Rev. A 23, 1138 (1981)  
Distorted-wave electron-impact-ionization cross sections for highly ionized neonlike atoms
- Younger, S.M.**  
Phys. Rev. A 24, 1272 (1981)  
Cross sections and rates for direct electron-impact ionization of sodiumlike ions
- Younger, S.M.**  
Phys. Rev. A 24, 1278 (1981)  
Electron-impact-ionization cross sections and rates for highly ionized berylliumlike ions
- Younger, S.M.**  
J. Quant. Spectrosc. & Radiat. Transfer 27, 541 (1982)  
Electron impact ionization rate coefficients and cross sections for highly ionized iron
- Younger, S.M.**  
Private Communication (1982)  
Analytic fits to distorted wave Born-exchange cross sections for argon ions.

Younger, S.M.  
Private Communication (1982)

Younger, S.M.  
Phys. Rev. A 25, 3396 (1982)  
Electron impact ionization cross sections for highly ionized chlorine-like ions

Younger, S.M.  
Phys. Rev. A 26, 3177 (1982)  
Distorted-electron impact ionization cross sections for the argon isoelectronic sequence

Younger, S.M.  
J. of Res. of the National Bureau of Standards 87, 49 (1982)  
Electron impact ionization of lithium

Younger, S.M.  
Private Communication (1982)  
Distorted wave Born-exchange cross sections for electron impact ionization

Younger, S.M.  
J. Quant. Spectrosc. & Radiat. Transfer 29, 61 (1982)  
Electron ionization rate coefficients for highly ionized iron and scandium

## LIST OF IPPJ-AM REPORTS

- IPPJ-AM-1\* "Cross Sections for Charge Transfer of Hydrogen Beams in Gases and Vapors in the Energy Range 10 eV–10 keV"  
H. Tawara (1977) [Published in Atomic Data and Nuclear Data Tables 22, 491 (1978)]
- IPPJ-AM-2\* "Ionization and Excitation of Ions by Electron Impact –Review of Empirical Formulae—" T. Kato (1977)
- IPPJ-AM-3 "Grotrian Diagrams of Highly Ionized Iron FeVIII-FeXXVI"  
K. Mori, M. Otsuka and T. Kato (1977) [Published in Atomic Data and Nuclear Data Tables 23, 196 (1979)]
- IPPJ-AM-4 "Atomic Processes in Hot Plasmas and X-Ray Emission"  
T. Kato (1978)
- IPPJ-AM-5\* "Charge Transfer between a Proton and a Heavy Metal Atom"  
S. Hiraide, Y. Kigoshi and M. Matsuzawa (1978)
- IPPJ-AM-6\* "Free-Free Transition in a Plasma –Review of Cross Sections and Spectra—" T. Kato and H. Narumi (1978)
- IPPJ-AM-7\* "Bibliography on Electron Collisions with Atomic Positive Ions: 1940 Through 1977"  
K. Takayanagi and T. Iwai (1978)
- IPPJ-AM-8 "Semi-Empirical Cross Sections and Rate Coefficients for Excitation and Ionization by Electron Collision and Photoionization of Helium"  
T. Fujimoto (1978)
- IPPJ-AM-9 "Charge Changing Cross Sections for Heavy-Particle Collisions in the Energy Range from 0.1 eV to 10 MeV I. Incidence of He, Li, Be, B and Their Ions"  
Kazuhiko Okuno (1978)
- IPPJ-AM-10 "Charge Changing Cross Sections for Heavy-Particle Collisions in the Energy Range from 0.1 eV to 10 MeV II. Incidence of C, N, O and Their Ions"  
Kazuhiko Okuno (1978)
- IPPJ-AM-11 "Charge Changing Cross Sections for Heavy-Particle Collisions in the Energy Range from 0.1 eV to 10 Mev III. Incidence of F, Ne, Na and Their Ions"  
Kazuhiko Okuno (1978)
- IPPJ-AM-12\* "Electron Impact Excitation of Positive Ions Calculated in the Coulomb-Born Approximation –A Data List and Comparative Survey—" S: Nakazaki and T. Hashino (1979)
- IPPJ-AM-13 "Atomic Processes in Fusion Plasmas – Proceedings of the Nagoya Seminar on Atomic Processes in Fusion Plasmas Sept. 5-7, 1979"  
Ed. by Y. Itikawa and T. Kato (1979)
- IPPJ-AM-14 "Energy Dependence of Sputtering Yields of Monatomic Solids"  
N. Matsunami, Y. Yamamura, Y. Itikawa, N. Itoh, Y. Kazumata, S. Miyagawa, K. Morita and R. Shimizu (1980)

- IPPJ-AM-15 "Cross Sections for Charge Transfer Collisions Involving Hydrogen Atoms"  
Y. Kaneko, T. Arikawa, Y. Itikawa, T. Iwai, T. Kato, M. Matsuzawa,  
Y. Nakai, K. Okuno, H. Ryufuku, H. Tawara and T. Watanabe (1980)
- IPPJ-AM-16 "Two-Centre Coulomb Phaseshifts and Radial Functions"  
H. Nakamura and H. Takagi (1980)
- IPPJ-AM-17 "Empirical Formulas for Ionization Cross Section of Atomic Ions for  
Electron Collisions -Critical Review with Compilation of Experimental  
Data-"  
Y. Itikawa and T. Kato (1981)
- IPPJ-AM-18 "Data on the Backscattering Coefficients of Light Ions from Solids"  
T. Tabata, R. Ito, Y. Itikawa, N. Itoh and K. Morita (1981)
- IPPJ-AM-19 "Recommended Values of Transport Cross Sections for Elastic Collision and  
Total Collision Cross Section for Electrons in Atomic and Molecular Gases"  
M. Hayashi (1981)
- IPPJ-AM-20 "Electron Capture and Loss Cross Sections for Collisions between Heavy  
Ions and Hydrogen Molecules"  
Y. Kaneko, Y. Itikawa, T. Iwai, T. Kato, Y. Nakai, K. Okuno and H. Tawara  
(1981)
- IPPJ-AM-21 "Surface Data for Fusion Devices - Proceedings of the U.S-Japan Work-  
shop on Surface Data Review Dec. 14-18, 1981"  
Ed. by N. Itoh and E.W. Thomas (1982)
- IPPJ-AM-22 "Desorption and Related Phenomena Relevant to Fusion Devices"  
Ed. by A. Koma (1982)
- IPPJ-AM-23 "Dielectronic Recombination of Hydrogenic Ions"  
T. Fujimoto, T. Kato and Y. Nakamura (1982)
- IPPJ-AM-24 "Bibliography on Electron Collisions with Atomic Positive Ions: 1978  
Through 1982 (Supplement to IPPJ-AM-7)"  
Y. Itikawa (1982)
- IPPJ-AM-25 "Bibliography on Ionization and Charge Transfer Processes in Ion-Ion  
Collision"  
H. Tawara (1983)
- IPPJ-AM-26 "Angular Dependence of Sputtering Yields of Monatomic Solids"  
Y. Yamamura, Y. Itikawa and N. Itoh (1983)
- IPPJ-AM-27 "Recommended Data on Excitation of Carbon and Oxygen Ions by Electron  
Collisions"  
Y. Itikawa, S. Hara, T. Kato, S. Nakazaki, M.S. Pindzola and D.H. Crandall  
(1983)
- IPPJ-AM-28 "Electron Capture and Loss Cross Sections for Collisions Between Heavy  
Ions and Hydrogen Molecules (Up-dated version of IPPJ-AM-20)  
H. Tawara, T. Kato and Y. Nakai (1983)

- IPPJ-AM-29 "Bibliography on Atomic Processes in Hot Dense Plasmas"  
T. Kato, J. Hama, T. Kagawa, S. Karashima, N. Miyanaga, H. Tawara, N. Yamaguchi, K. Yamamoto and K. Yonei (1983)
- IPPJ-AM-30 "Cross Sections for Charge Transfers of Highly Ionized Ions in Hydrogen Atoms (Up-dated version of IPPJ-AM-15)"  
H. Tawara, T. Kato and Y. Nakai (1983)
- IPPJ-AM-31 "Atomic Processes in Hot Dense Plasmas"  
T. Kagawa, T. Kato, T. Watanabe and S. Karashima (1983)
- IPPJ-AM-32 "Energy Dependence of the Yields of Ion-Induced Sputtering of Monatomic Solids"  
N. Matsunami, Y. Yamamura, Y. Itikawa, N. Itoh, Y. Kazumata, S. Miyagawa, K. Morita, R. Shimizu and H. Tawara (1983)
- IPPJ-AM-33 "Proceedings on Symposium on Atomic Collision Data for Diagnostics and Modelling of Fusion Plasmas, Aug. 29 – 30, 1983"  
Ed. by H. Tawara (1983)
- IPPJ-AM-34 "Dependence of the Backscattering Coefficients of Light Ions upon Angle of Incidence"  
T. Tabata, R. Ito , Y. Itikawa, N. Itoh, K. Morita and H. Tawara (1984)
- IPPJ-AM-35 "Proceedings of Workshop on Synergistic Effects in Surface Phenomena Related to Plasma-Wall Interactions, May 21 – 23, 1984"  
Ed. by N. Itoh, K. Kamada and H. Tawara (1984)
- IPPJ-AM-36 "Equilibrium Charge State Distributions of Ions ( $Z_1 \geq 4$ ) after Passage through Foils – Compilation of Data after 1972"  
K. Shima, T. Mikumo and H. Tawara (1985)
- IPPJ-AM-37 "Ionization Cross Sections of Atoms and Ions by Electron Impact"  
H. Tawara, T. Kato and M. Ohnishi (1985)

---

Available upon request to Research Information Center, Institute of Plasma Physics, Nagoya University, Nagoya 464, Japan, except for the reports noted with\*.

